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# Improved Accelerated Stress Tests Based on FCV Data

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UTC POWER

MAY 11, 2011



**UTC Power**

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Project ID: FC015

# Overview

## Timeline

- Start Date: December 2009
- Finish Date: November 2011
- Status: 65% Complete

## Budget

- Total funding \$3,847,218
- Cost share 20%
- Spend on-plan

GFY '09	\$778,015
GFY '10	\$1,638,508
GFY '11	\$1,430,694

## Barriers

(2007 RD&D for auto FC)

- ❑ >5,000 hr stack durability (with cycling)
  - Include all materials (e.g. membrane, seals)
  - UTC bus fleet target >15,000 h stack life
- ❑ <10% Performance decay
  - Start-stop / transient operation
- ❑ ASTs used to avoid costly durability testing

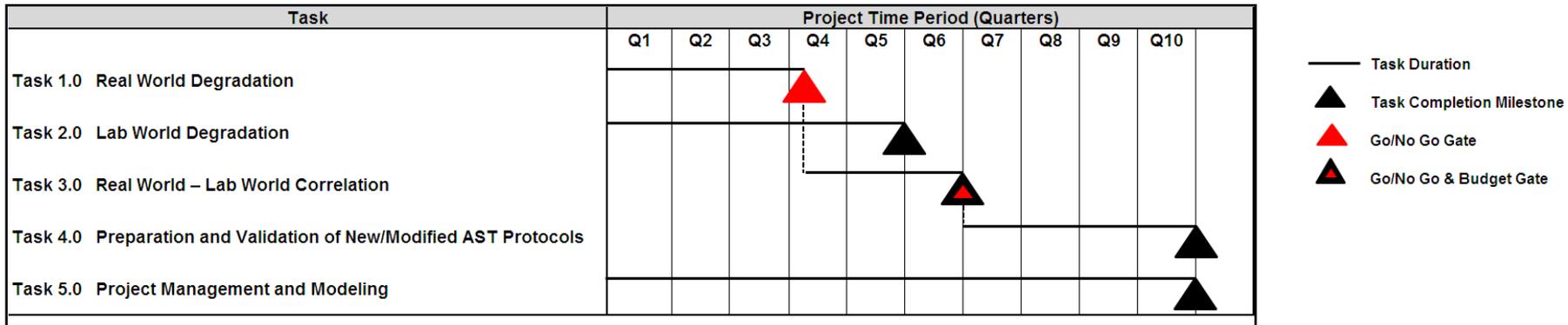
## Partners



# Relevance

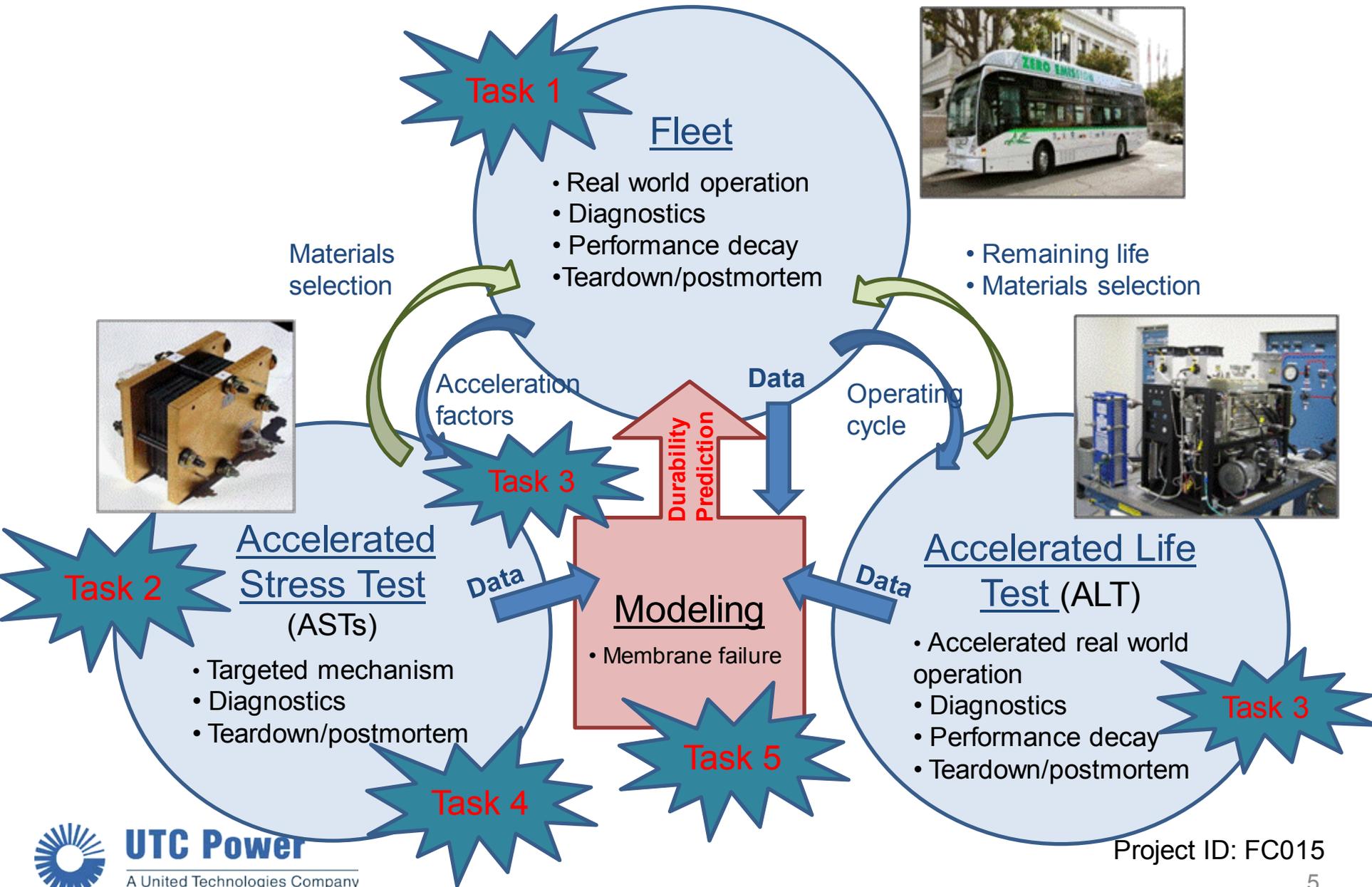
Program Objectives	Current Gaps	2009-2010 Objectives
Comparison of conditions & materials in bus field operation vs. DOE ASTs	DOE ASTs not calibrated with real world degradation	<p><b>Task 1</b> - Analyze performance data and characterize degraded materials from 2850 hr stacks in bus service</p> <p><b>Task 2</b> - Analyze data and degraded materials run in DOE ASTs (same as in bus stacks)</p>
Develop acceleration factors for DOE AST mechanisms → recommend modifications	DOE ASTs may over- or under-accelerate mechanisms → inadequate material selections	<p><b>Task 3</b> - Correlate results for all current DOE ASTs:</p> <ol style="list-style-type: none"> <li>1) PGM decay</li> <li>2) Carbon corrosion</li> <li>3) Membrane mechanical</li> <li>4) Membrane chemical</li> </ol>
Identify life-limiting mechanisms not addressed by DOE AST's → recommend new AST's	Validated GDL specific AST; Validated integrated membrane mechanical/chemical AST;	<p><b>Task 4</b> – Prepare and Validate New/Modified AST Protocols</p>

# Approach



Task	Progress	Status
1.0. Real world degradation	<ul style="list-style-type: none"> <li>Completed bus operating cycle analysis</li> <li>Completed characterization of field-operated bus stack (2850 h) for all 4 decay mechanisms covered by current AST's</li> </ul>	100% complete
2.0. Lab world degradation	<ul style="list-style-type: none"> <li>Completed all 4 DOE AST on the same materials as the bus stack</li> <li>Post-mortem characterization completed</li> </ul>	100% complete
Go / No Go Gate 1	<ul style="list-style-type: none"> <li>Correlate all observed degradation to field operating conditions → sufficient degradation in field conditions</li> </ul>	100% complete
3.0. Real – Lab correlation	<ul style="list-style-type: none"> <li>Acceleration factors determined for all existing DOE mechanisms</li> <li>New degradation mode found and accelerated test proposed</li> <li>New breadboard unit developed and running accelerated life test (ALT)</li> </ul>	100% complete
Go/No Go Gate 2	<ul style="list-style-type: none"> <li>Gap in DOE AST's identified for isolation of GDL oxidation effects</li> <li>Modeling and DMA used to understand chemical + mechanical membrane degradation</li> </ul>	100% complete
4.0 Preparation and Validation of New/Modified AST Protocols	<ul style="list-style-type: none"> <li>Integrated membrane chemical/mechanical AST underway</li> <li>Ex-situ GDL oxidation AST development proposed</li> </ul>	10% complete
5.0 Project Management and Modeling	<ul style="list-style-type: none"> <li>Further development of membrane hydration strain model ongoing</li> </ul>	25% complete

# Approach



# Collaborations

## Partners



### United Technology Research Center (Industry)

- Membrane hydration strain modeling
- Material characterization



### Los Alamos National Laboratory (Federal)

- AST development
- Subscale fuel cell and electrochemical testing



### Oak Ridge National Laboratory (Federal)

- Material characterization



# Technical Accomplishments

## Real World Performance Degradation

### Real World Conditions

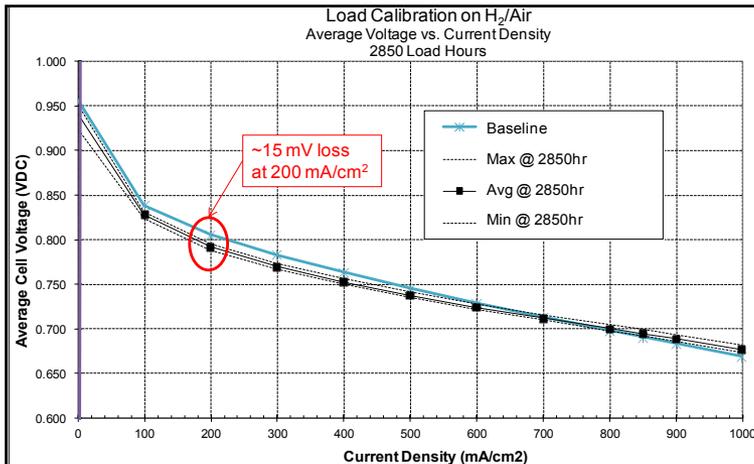
Reactants: H<sub>2</sub>(dry)/Air(ambient)

Average Temperature: 63 C

Cycle Frequency: 100-120 cycles/hr

Operating time: 2800 hours

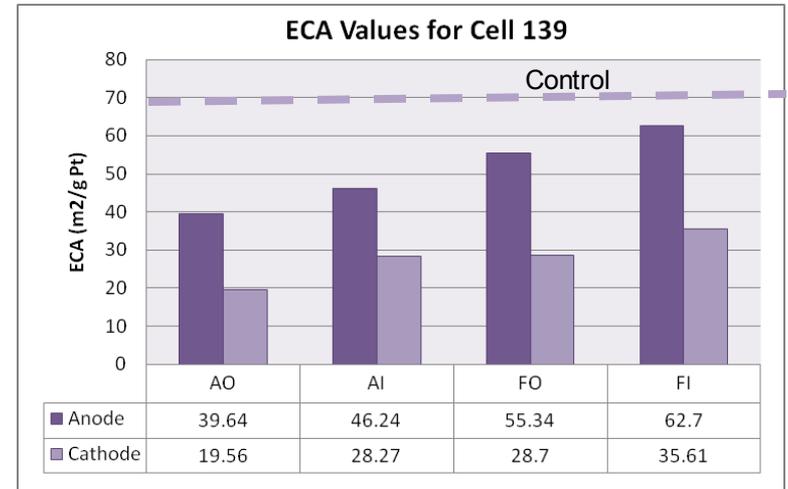
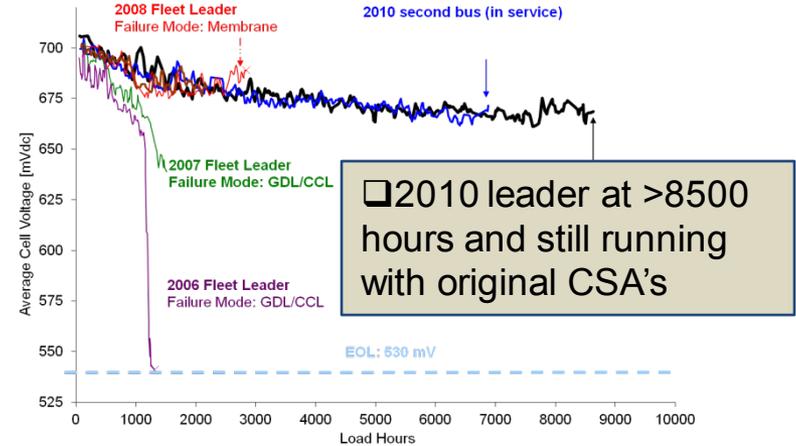
Cumulative cycles: 300,000 cycles



UTC 2008 fleet leader at BOL and EOL

~ 15 mV loss @ 0.2 A/cm<sup>2</sup>

~50% Platinum area loss



ECA measurements performed on subscale sections at indicated locations (calculated particle size shown in red)

# Technical Accomplishments

## PGM AST

Lab ASTs by LANL on UTC 2008 fleet leader BOL components in 50 cm<sup>2</sup> cell

### AST Conditions

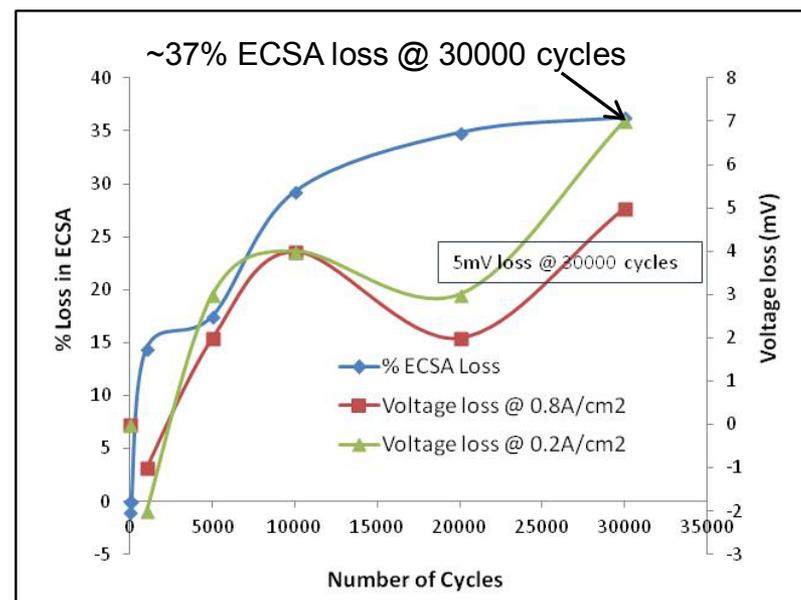
Reactants: H<sub>2</sub>(100%)/N<sub>2</sub>(100%)

Average Temperature: 80 C

Cycle Frequency: 112 cycles/h

Cycle Range: 0.6 – 1.0 V (Triangle)

Observed @ 30k cycles	DOE Target @ 30k cycles
~37% ECA loss	≤40% loss of initial area
~7 mV loss @ 0.2 A/cm <sup>2</sup>	≤30 mV loss at 0.8 A/cm <sup>2</sup>
~5 mV loss @ 0.8 A/cm <sup>2</sup>	
~20% loss of activity	≤40% loss of activity

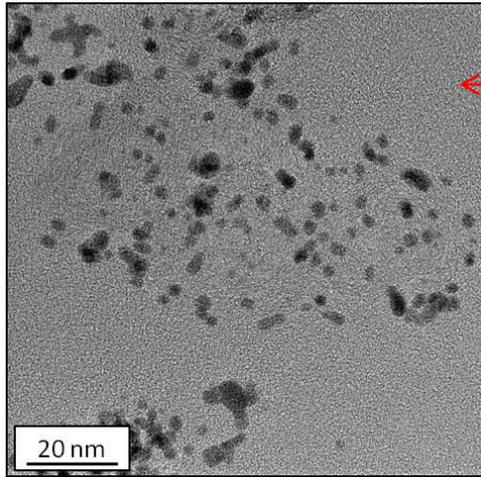


Voltage and ECA loss during potential cycling AST

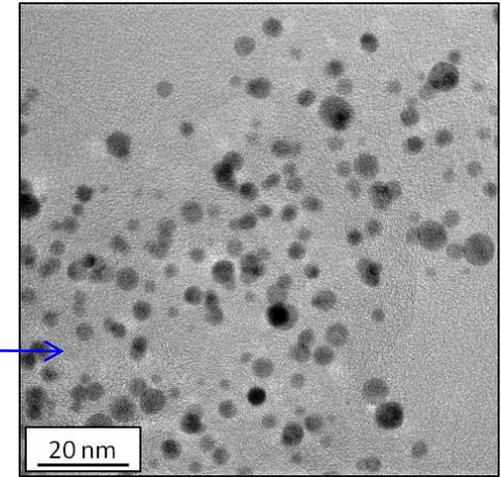
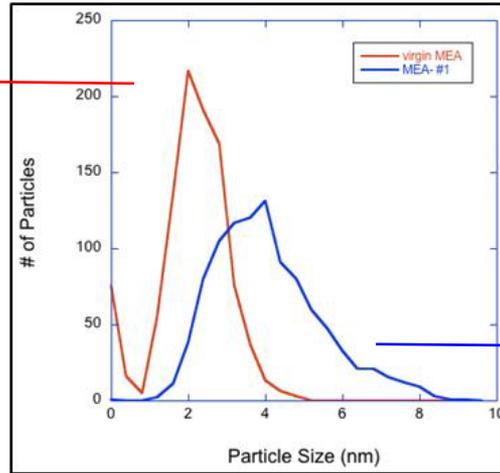
□ Potential cycling AST has acceleration factor of ~8X vs. fleet operation

# Technical Accomplishments

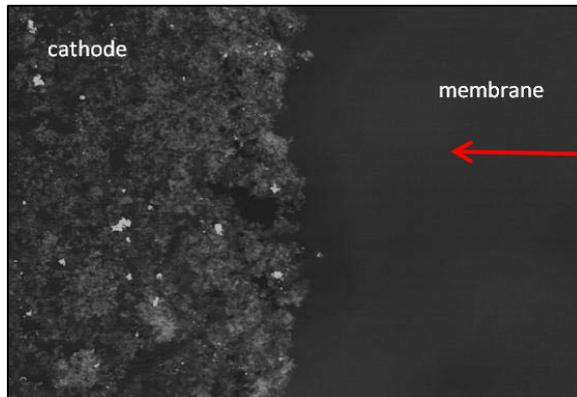
## TEM Analysis on UTC 2008 fleet leader



Baseline

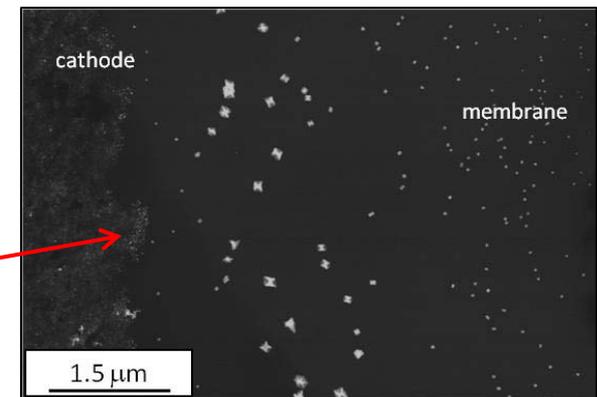


Air Inlet - 2850 hr



Baseline cathode = ~2nm  
2850 hr cathode = ~4nm

No Pt observed in  
membrane of virgin MEA

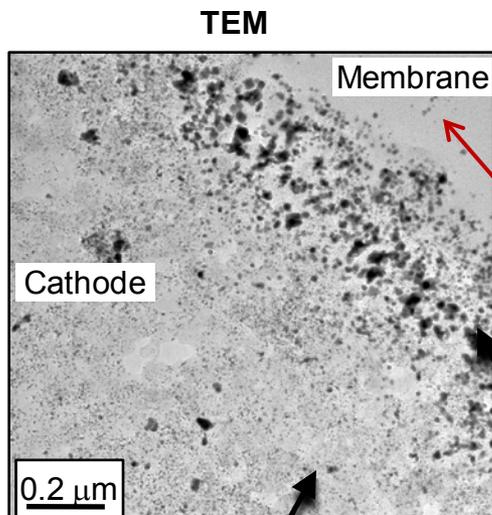


Pt precipitation in  
membrane adjacent to  
cathode

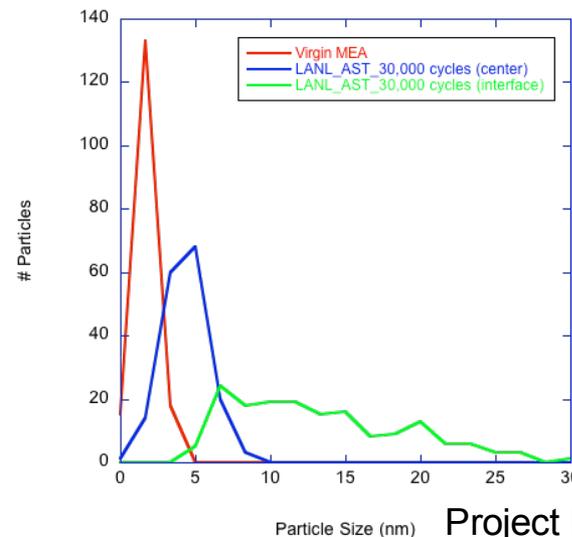
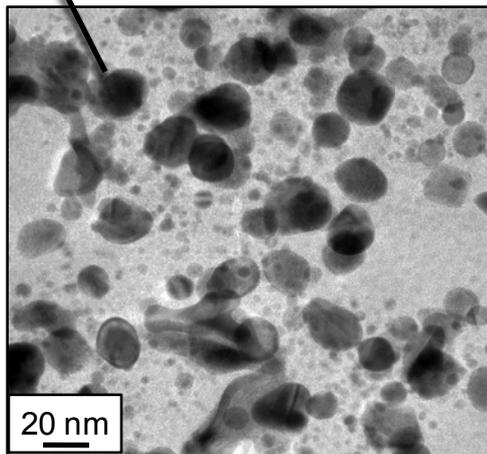
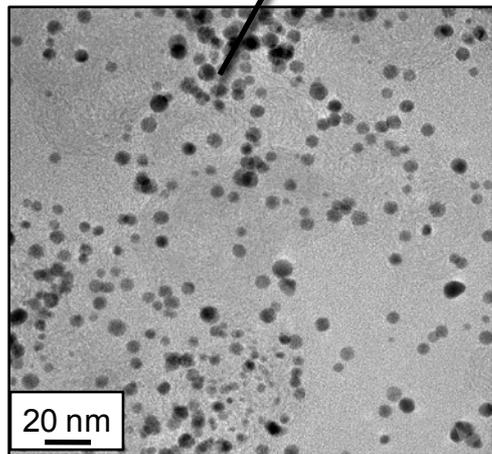
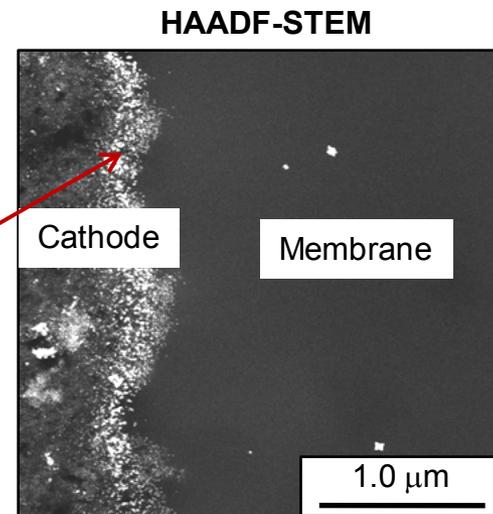
□ TEM analysis shows particle agglomeration and platinum precipitation within the membrane adjacent to cathode catalyst layer

# Technical Accomplishments

## TEM analysis of MEA subjected to PGM AST



- ❑ After AST, formation of large Pt single crystals at cathode - membrane interface observed with little Pt precipitation observed within membrane
- ❑ This can be explained by difference in gas compositions; Platinum will precipitate in membrane under H<sub>2</sub>/Air and just outside of membrane under H<sub>2</sub>/N<sub>2</sub>



# Technical Accomplishments

## Carbon Corrosion AST

### AST Conditions

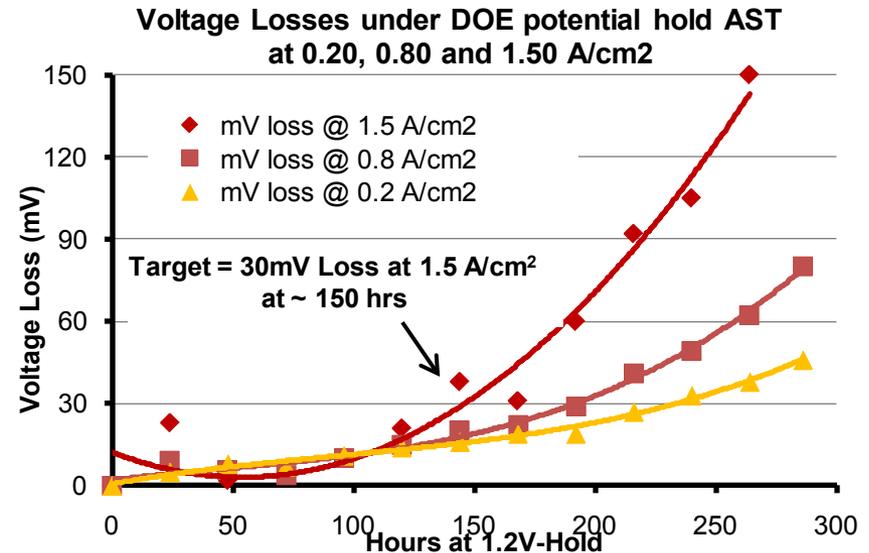
Reactants: H<sub>2</sub>(100%)/N<sub>2</sub>(100%)

Average Temperature: 80 C

Cycle Period: 24 h

Cycle Range: 1.2 V

Observed @ 285 hr	DOE Target @ 400 hr
~40% mass activity	≤60% mass activity
>150mV @ 1.5 A/cm <sup>2</sup>	≤30mV @ 1.5 A/cm <sup>2</sup>
~50% ECA loss	≤40% ECA loss



### Summary

Severe carbon corrosion under Lab AST

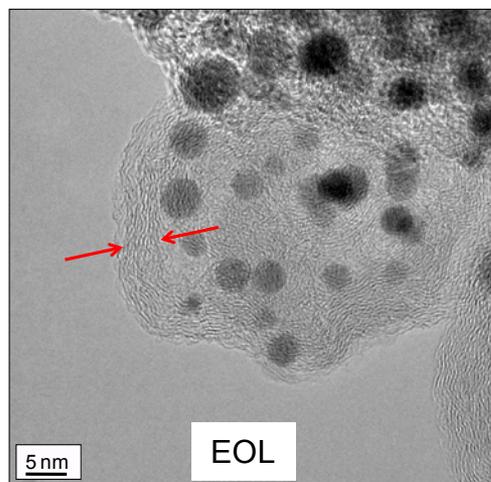
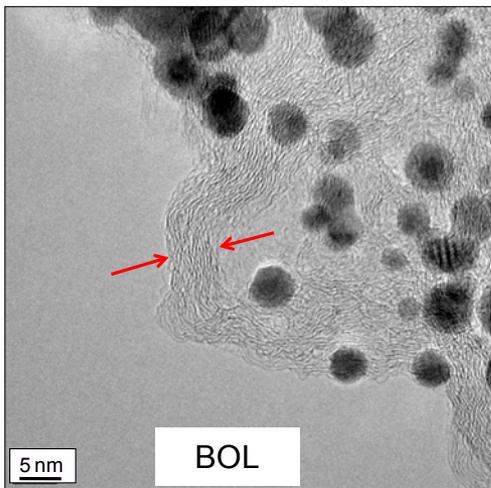
Damaging potentials avoided in 2008 fleet leader

2006 and 2007 fleet leaders suffered from carbon corrosion

# Technical Accomplishments

## Real World and AST Carbon Morphology

UTC 2008 Fleet Leader

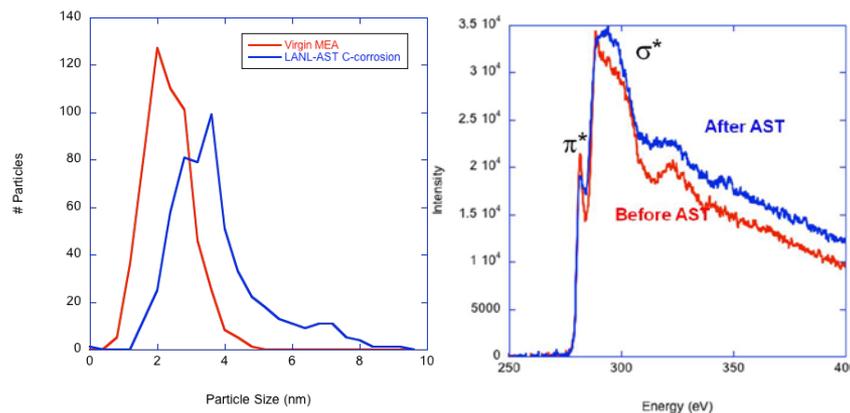


### UTC 2008 fleet leader at BOL and EOL

- Small voltage loss @ 1.0 A/cm<sup>2</sup>
- Effective system mitigations for start-stop decay → Minor carbon corrosion in real-world operation

### AST

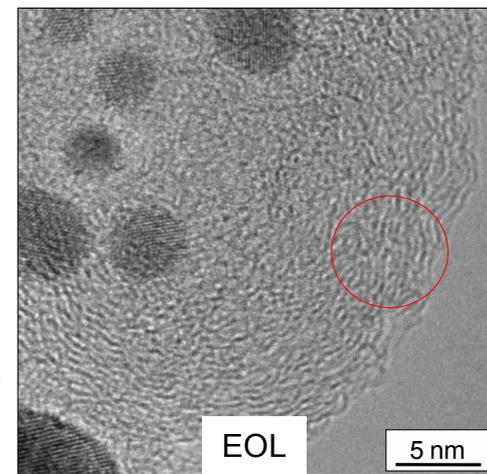
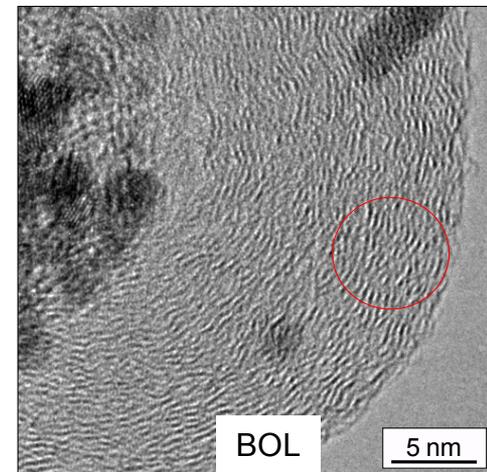
- Cathode electrode thinning (20-70%)
- Platinum particle growth
- Increased  $\sigma$  carbon bonds
- AST effective for C corrosion



Pt particle growth: ~2.0nm in baseline to ~3.8nm after AST

Increased  $\sigma$  carbon bonds after AST

AST



Graphitic structure disorganized

Project ID: FC015

# Technical Accomplishments

## Real World Membrane Mechanical Damage

### Real World Conditions

Reactants: H<sub>2</sub> / Air

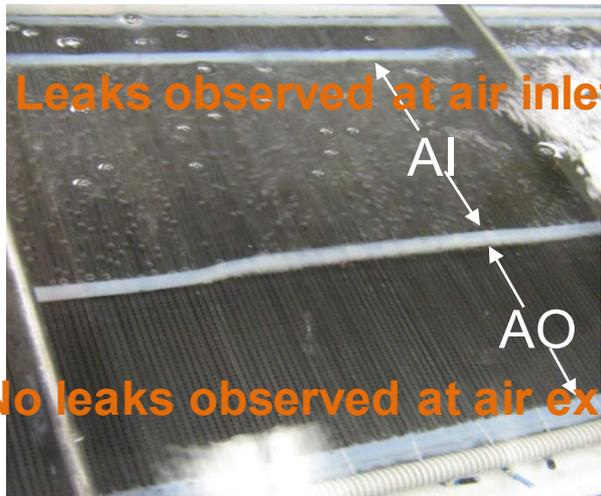
Average Temperature: 63 C

Cycle Frequency: 100-120 cycles/hr

Humidity range: 50-100% ( $\Delta$ RH=50%)

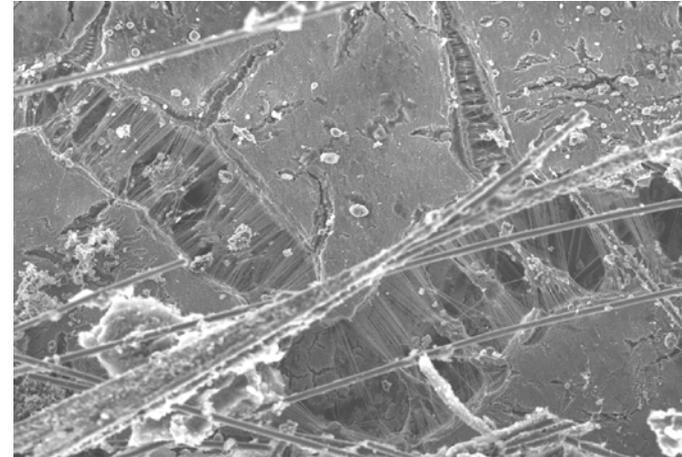
Total Time: 2800 hours

Total Cycles: 300,000 cycles

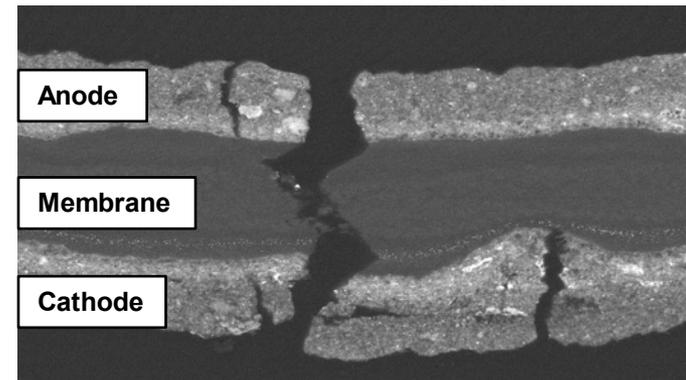


- ❑ In fleet operation, membrane failed after 2800 hours at air inlet due to membrane hydration strain cycling induced by load-flow cycling

### Surface image



### Cross-section



SEM micrographs confirming membrane failure at the air inlet

# Technical Accomplishments

## AST Membrane Mechanical Damage

### Lab AST Conditions

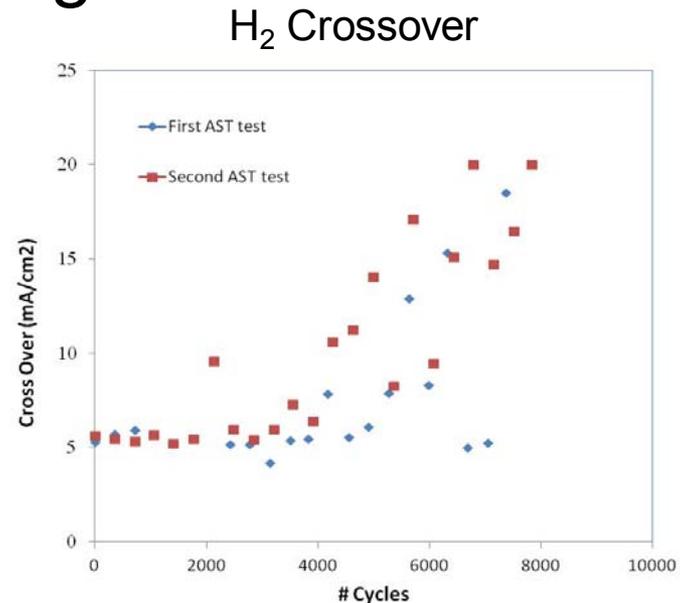
Reactants: Air / Air

Average Temperature: 80 C

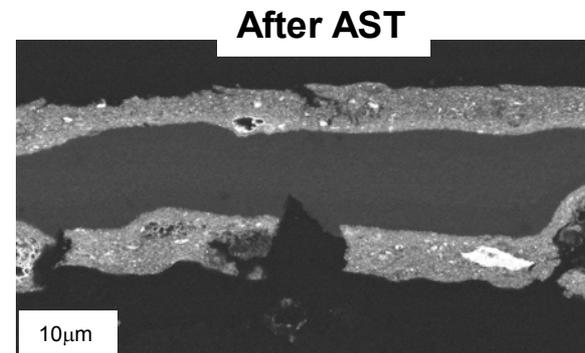
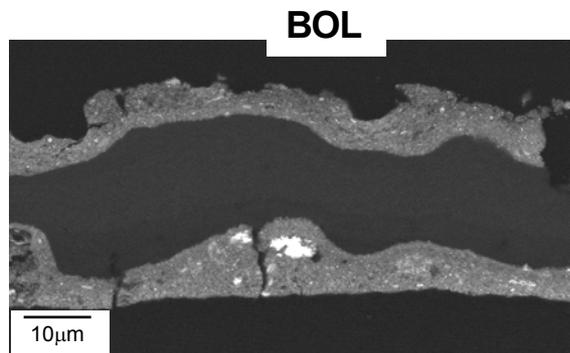
Cycle Frequency: 15 cycles/h

Humidity range: 0-100% ( $\Delta RH=100\%$ )

	DOE AST Target @ 20000 cycles	Observed @ 7000 cycles
Crossover	<2 mA/cm <sup>2</sup>	20 mA/cm <sup>2</sup>
Shorting Resistance	> 1000 ohm cm <sup>2</sup>	~1400 ohm cm <sup>2</sup>



- ❑ Membrane mechanical AST resulted in acceleration factor of ~6X vs. fleet operation with same failure mode



Structural damage observed in membrane and catalyst layers

# Technical Accomplishments

## Real World Membrane Chemical Damage

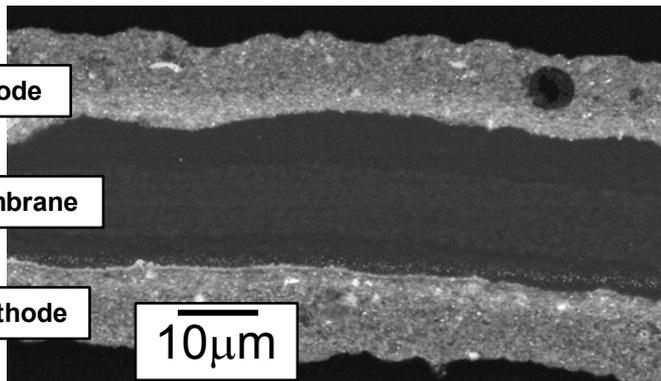
### Real World Conditions

Reactants: H<sub>2</sub> / Air

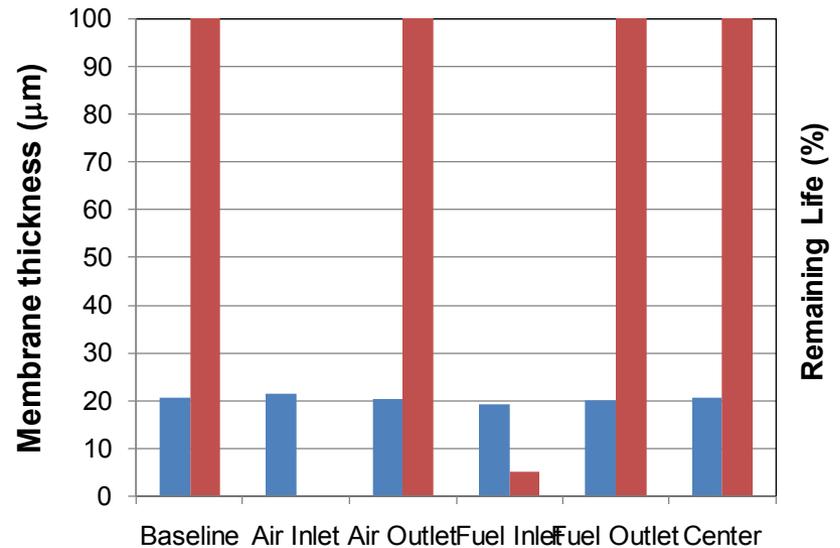
Average Temperature: 63 C

Time at high voltage: 30%

Humidity: 100%



No detectable chemical attack



Membrane thickness measurements and DMA testing of remaining life results of BOL and EOL at various locations in the active area

- ❑ No membrane thinning observed in real world application; Significant mechanical degradation observed at reactant inlets due to membrane hydration strain cycling

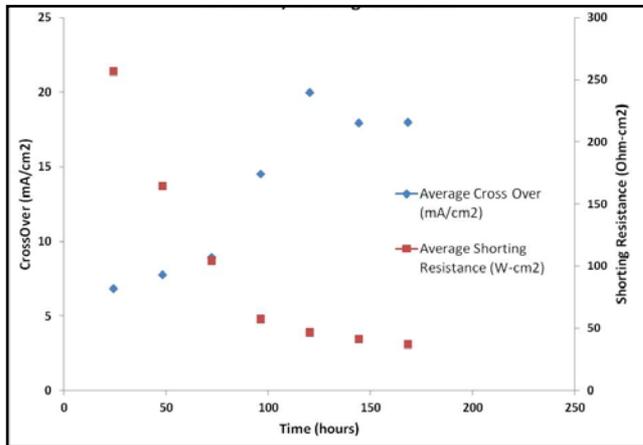
# Technical Accomplishments

## AST Membrane Chemical Decay

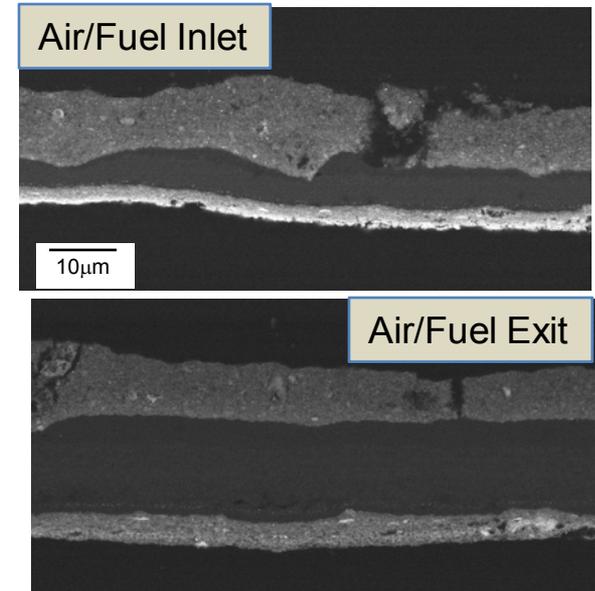
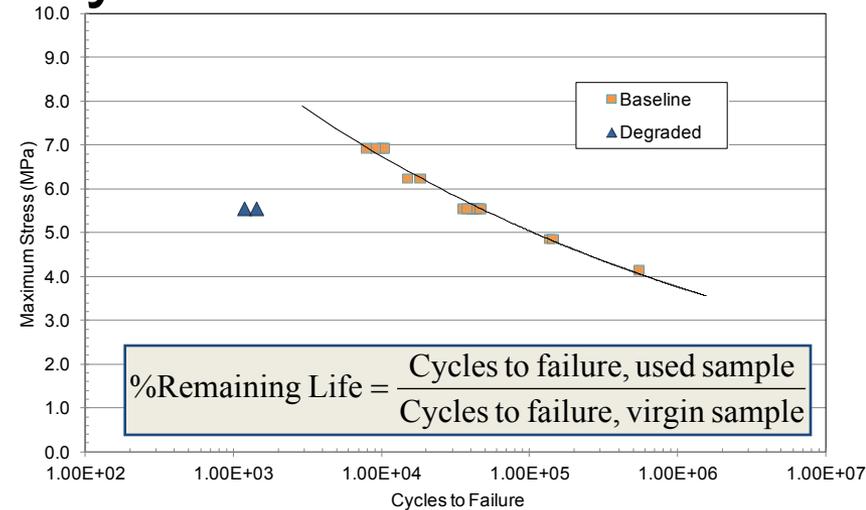
**AST Conditions**  
 Reactants: H<sub>2</sub> / Air  
 Average Temperature: 90 C  
 Voltage: Open Circuit  
 Humidity: 30%

	DOE Target @ 500 hr	Observed @ 200 hr
Crossover	≤ 2 mA/cm <sup>2</sup>	~ 20 mA/cm <sup>2</sup>
OCV loss	≤ 20%	~ 20%
Shorting Resistance	< 1000 Ω-cm <sup>2</sup>	~ 250 Ω-cm <sup>2</sup>

### Crossover/Shorting Resistance



❑ Significant membrane thinning observed after membrane chemical decay AST; DMA technique useful for estimating remaining life

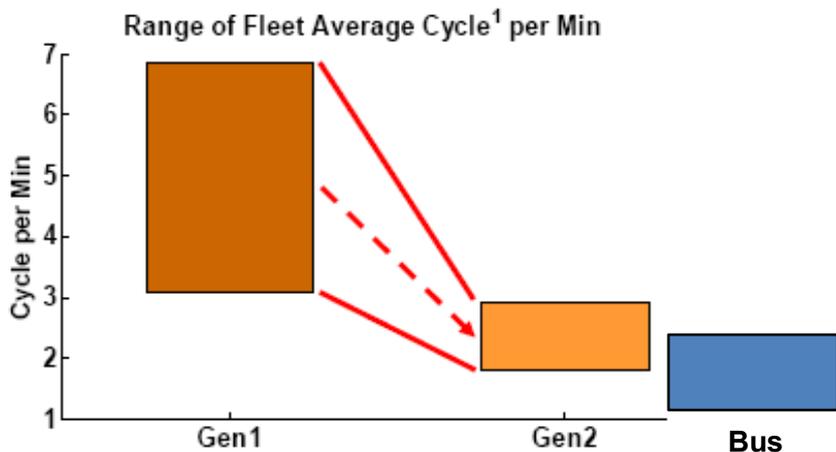


# Technical Accomplishments

## ALT Rig

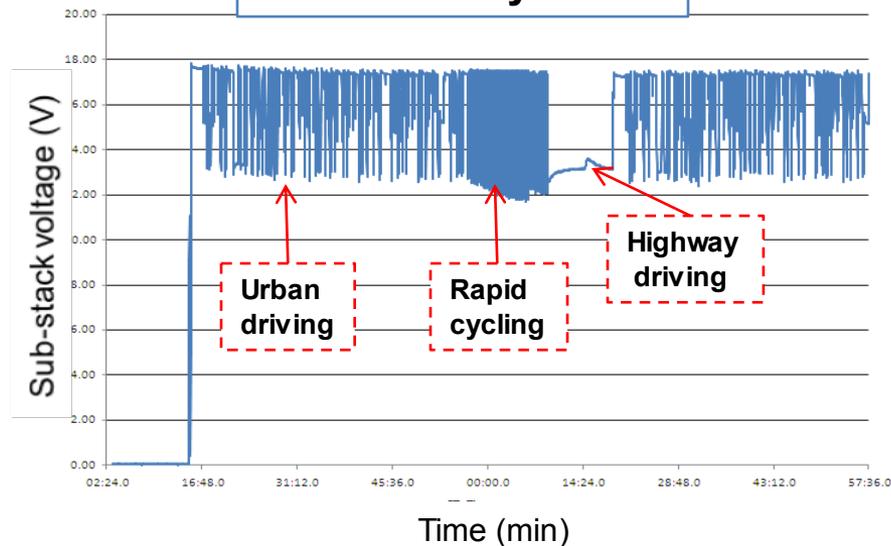
Stack format	Failure mode	Expected completion
2008 leader	Membrane	May 2011
2010 leader + advanced UEAs	Unknown	Nov 2011

### Auto cycle

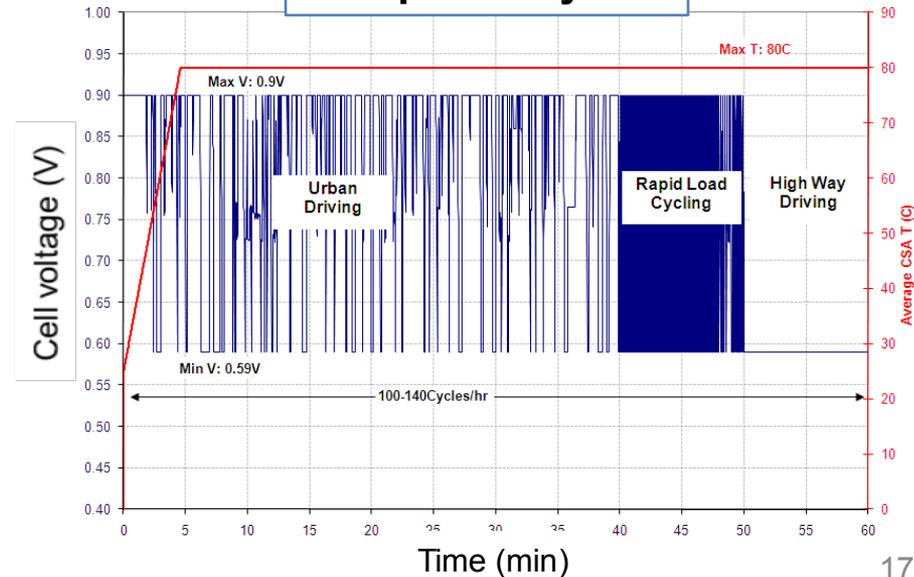


from "Controlled Hydrogen Fleet and Infrastructure Analysis" TV001, K. Wipke of NREL, 2010 AMR

### Transit cycle



### Proposed cycle



# Technical Accomplishments

## Summary of Real World to AST Comparisons

Mechanism	Real Cycle	Real Diagnostics	AST Cycle	AST Diagnostics	Correlation
Pt Dissolution/ sintering	<u>2850 hrs</u> 100 cycles/h 0.6 to 0.9 V	~2% V loss @ low power ~60% ECA loss	<u>30 kcyc</u> 112.5 cycles/h 0.6 – 1.0 V <sub>nhe</sub>	~1% V loss @ low power ~37% ECA loss	-High AST-Real correlation  -Low impact on Real World perf  -May depend on 0.4 mg Pt/cm <sup>2</sup>
Carbon Corrosion	<u>~500 cycles</u> H <sub>2</sub> / Air fronts from S/S cycles (mitigated)	Negligible performance loss @ 1.0 A/cm <sup>2</sup>	<u>250 hrs</u> Continuous hold @1.2 V <sub>nhe</sub>	90 mV performance loss @ 1.0 A/cm <sup>2</sup>	-No AST-Real correlation  -AST effective for C corrosion  -High impact on performance if S/S mitigations not employed
Membrane Mechanical	<u>2850 hrs</u> 100 cycles/h ΔRH: 40-50%	High RPS response & visible leakage across stack	<u>6000 cyc</u> ΔRH: 0- >100% 15 cycles/h	>15 mA/cm <sup>2</sup> (vs 2 mA/cm <sup>2</sup> BOL)	-High AST-Real correlation  -High impact on Real World perf
Membrane Chemical	<u>~850 hrs</u> H <sub>2</sub> /Air @ V <sub>idle</sub> 63°C 100%RH	Minimal membrane thinning	<u>100 hr</u> H <sub>2</sub> /air 90°C 30%RH OCV	>15 mA/cm <sup>2</sup> (vs 2 mA/cm <sup>2</sup> BOL)	-Low AST-Real correlation  -AST effective for membrane chemical degradation  -Combined chemical-mechanical

# Future Work

Task	Description	Status	Owner	Comments
Task 4	4.1 Develop new AST protocols	Underway	UTC Power / LANL	<ul style="list-style-type: none"> <li>Identified following opportunities:               <ol style="list-style-type: none"> <li>Integrated membrane chemical / mechanical AST[Q2/2011]</li> <li>Validated failure time prediction using DMA[Q3/2011]</li> <li>GDL oxidation AST[Q3/2011]</li> <li>Validate the ALT breadboard degradation[Q4/2011]</li> <li>Evaluate proposed FCTT load / RH cycle AST[Q2/2011]</li> </ol> </li> </ul>
	4.2 Test BOL cells with new AST's	Underway	UTC Power / LANL	<ul style="list-style-type: none"> <li>ALT currently testing BOL materials at UTCP[Q3/2011]</li> </ul>
	4.3 Validate new AST protocols	Not started	All	<ul style="list-style-type: none"> <li>LANL testing of developed protocols[Q3/2011]</li> <li>Employ analyses and techniques already developed for new AST's[Q3/2011]</li> </ul>
Task 5	5.1 Further development of membrane hydration strain model	Underway	UTRC	<ul style="list-style-type: none"> <li>Include modeling of composite structures[Q3/2011]</li> <li>Include effect of flow field design on membrane hydration strain[Q4/2011]</li> </ul>

# PROJECT SUMMARY

**Relevance:** Development of validated accelerated test protocols for all identified failure modes will decrease need for expensive, time consuming durability testing

**Approach:** Perform fuel cell diagnostics and materials characterization on real world samples and samples that have been subjected to accelerated test protocols; Identify any failure modes not being addressed by current DOE AST protocols and develop and validate new AST's for those failure modes;

## **Technical Accomplishments and Progress:**

- Completed characterization of field-operated bus stack (2850 h) for all 4 decay mechanisms covered by current AST's;
- New breadboard unit developed and running accelerated life test (ALT)
- Gap in DOE AST's identified for isolation of GDL oxidation effects
- DMA used to estimate remaining life

**Technology Transfer/Collaborations:** Active partnerships with LANL and ORNL in AST validation, development of new AST's, and material characterization . Technology transfer through team meetings, presentations and publications.

**Proposed Future Research:** Develop validated ex-situ GDL oxidation test; Validate combined membrane chemical/mechanical AST; Validate use of DMA as lifetime prediction tool;

# Technical Back-up Slides

# Technical Accomplishments

## Real World versus AST

Decay Mechanism	Task 1 – Real-World			Task 2 – Lab-World		
	Real-World Cycle	Real-World Diagnostics	Real-World Post-test Analysis	Lab AST Cycle	Lab AST Diagnostics	Lab AST Post-test Analysis
Pt Dissolution/sintering	H <sub>2</sub> /Air 63°C 100%RH 100 cycles/h 0.6 – 0.9 V	Performance ΔV @ 0.2A/cm <sup>2</sup> ECA loss	SEM XRD HRTEM	H <sub>2</sub> /N <sub>2</sub> 80°C 100%RH 112.5 cycles/h 0.6 – 1.0 V <sub>nhe</sub>	Performance ΔV @ 0.2A/cm <sup>2</sup> Mass activity & ECA loss	SEM XRD HRTEM
Carbon Corrosion	H <sub>2</sub> / Air fronts during start-up (25 C) and shutdown (65 C) (mitigated); Air-Air Time	Performance ΔV@1.0 A/cm <sup>2</sup>	SEM HRTEM EELS	H <sub>2</sub> /N <sub>2</sub> 80°C 100%RH 1.2 V <sub>nhe</sub>	Performance ΔV@1.5A/cm <sup>2</sup> CO <sub>2</sub> release Mass activity loss	SEM HRTEM EELS
Membrane Mechanical	H <sub>2</sub> /Air 63°C ΔRH: 40-50% 100 cycles/h	Crossover diagnostics (RPS)	SEM, DMA (BOL & EOL)	Air/Air 80°C ΔRH: 0→100% 15 cycles/h	H <sub>2</sub> cross-over Shorting	SEM, DMA (BOL & EOL)
Membrane Chemical	H <sub>2</sub> /Air 63°C 100%RH V <sub>idle</sub> for ~30% load time	Crossover diagnostics (RPS)	SEM, DMA (BOL & EOL)	H <sub>2</sub> /air 90°C 30%RH OCV	FER H <sub>2</sub> cross-over Shorting	SEM, DMA (BOL & EOL)



# Experimental – dynamic mechanical analysis (DMA)

Sample: MEA

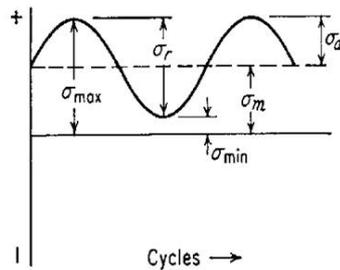
Conditions:

60°C

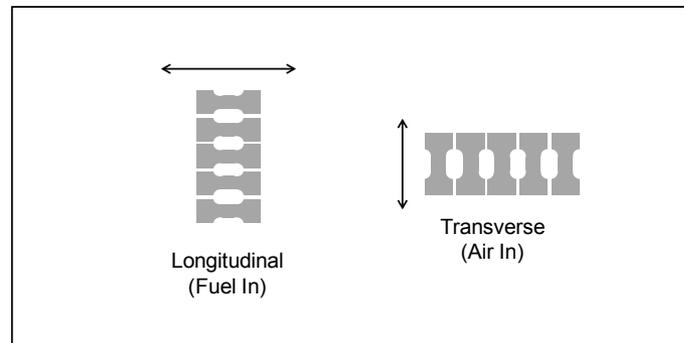
90%RH

10Hz

$$\sigma_{\text{Min}} = 20\% \sigma_{\text{max}}$$



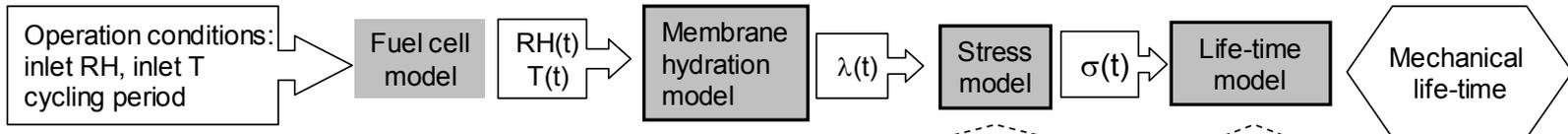
Sampling configuration:



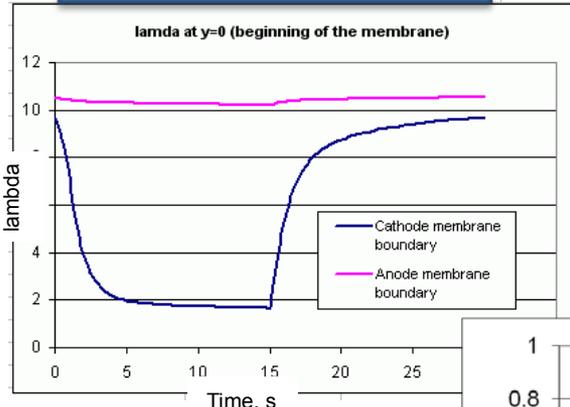
# Technical Accomplishments

## Membrane Hydration-induced Stress Model

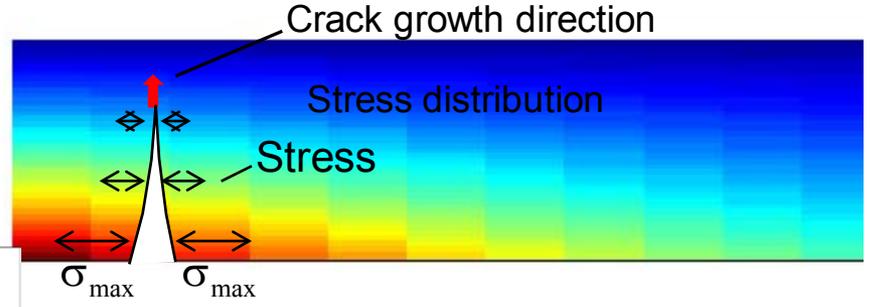
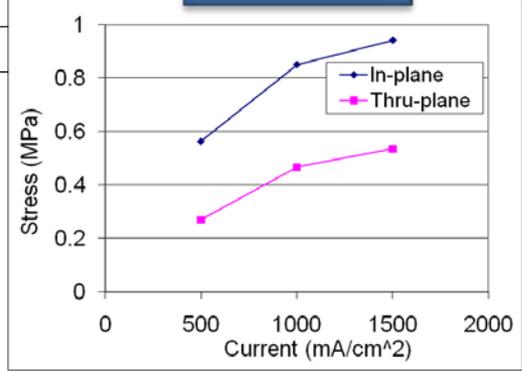
Membrane Mechanical Decay Model:



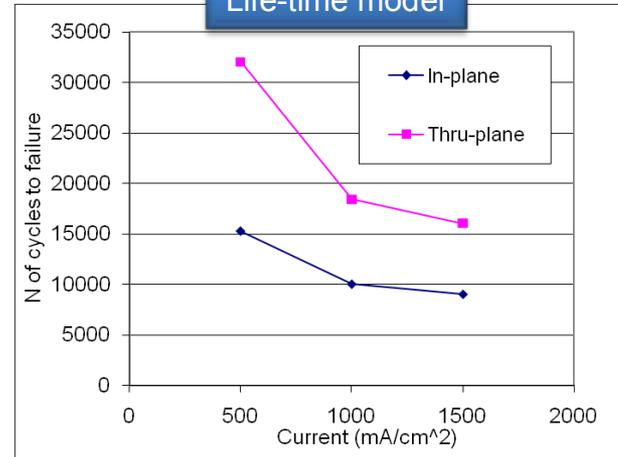
Membrane hydration model



Stress model



Life-time model

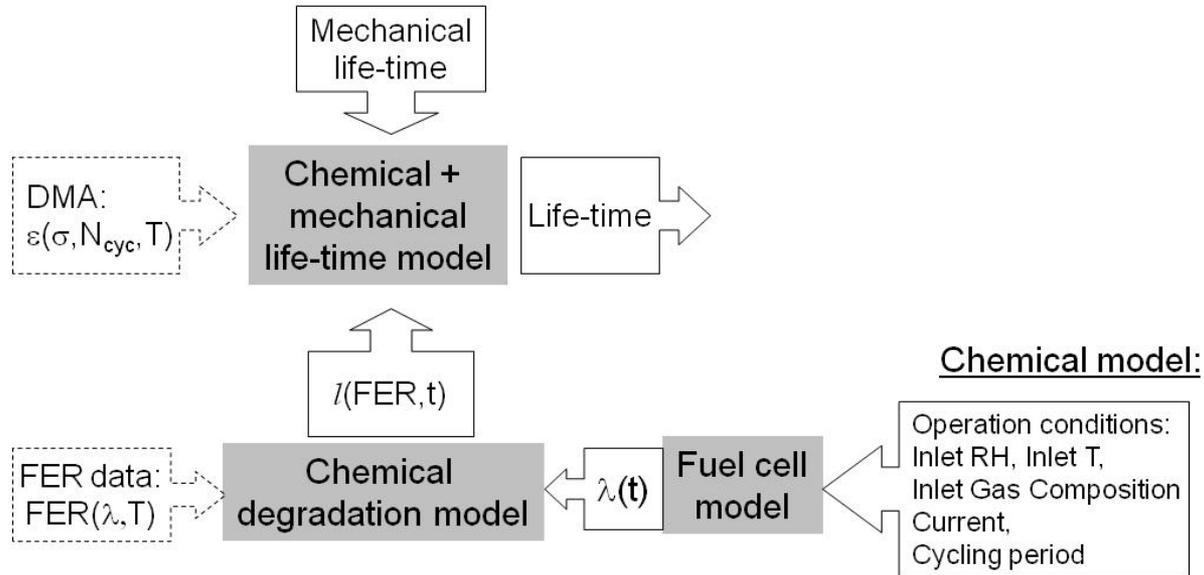


Developed modeling framework to relate fuel cell operation to membrane property changes and predict mechanical failure



st ID: FC015

# Motivation for Membrane Chemical + Mechanical



	Through plane	In plane
<b>Maximum stress, Mpa</b>	<b>0.41</b>	<b>0.81</b>
<b>Cycles to failure: Mechanical</b>	<b>142463.76</b>	<b>70840.79</b>
<b>Cycles to failure: Chemical + Mechanical</b>	<b>45208.29</b>	<b>30679.42</b>

“Model for Prediction of Abrupt Failure of PEMFC Membranes” in preparation to **Journal of the Electrochemical Society**, November 2010.

# Motivation for Diffusion Layer AST

## GDL Corrosion

- Occurs at all normal cathode potentials
- Can be partially oxidized to form surface oxide groups
$$C + H_2O \rightarrow C-OH + H^+ + e^-$$
  - Results in change in surface hydrophobicity
- Can be fully oxidized to form gas
$$C + 2H_2O \rightarrow CO_2 + 4H^+ + 4e^-$$
  - Results in change in structure
- Cycling vs holding at fixed potentials leads to different results

