A Multifunctional Isostructural Bilayer Oxygen Evolution Electrode for Durable Intermediate-Temperature Electrochemical Water Splitting

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> > Project ID P190

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Project Overview

 Timeline Start date: 10/1/2019 End date: 9/30/2022 Actual starting date: 2/10/2020 	 Barriers High degradation rate Oxygen/electrolyte interfacial delamination Cr-poisoning
 Budget Total Project Budget: \$1.25M Total Recipient Share: \$0.25M Total Federal Share: \$1.0M Total DOE Funds Spent: \$17K as of March 31, 2020 	 Prof. Xinfang Jin, University of Massachusetts at Lowell INL for testing nodes usage NREL for modeling nodes usage

Relevance

• Objectives:

To develop a highly active and Cr-resistant oxygen electrode for high-durability, high-efficiency and high-rate hydrogen production via high temperature solid oxide electrolysis cells (HT-SOECs). HT-SOEC is currently an integrated part of HydroGEN Consortium R&D program

Key Milestones:

Metric (in stack)	State of the Art	Expected Advance
Button cell @ j= 0.9A/cm², 700ºC	η _{OE} >0.20V	η _{ΟΕ} <0.15V
Planar 2-cells @ j= 0.8A/cm ² and 700ºC	η _{0E} >0.25V >15 mV/kh 1000h	η _{0E} <0.15V <10 mV/kh 1000h
Tubular cells @ j= 0.9A/cm ² and 700ºC	η _{0E} >0.20V >10 mV/kh 5000h	η _{0E} <0.15V <8 mV/kh 5000h

• Impact

Accelerating commercialization of SOEC technology for low-cost and high-efficiency hydrogen production

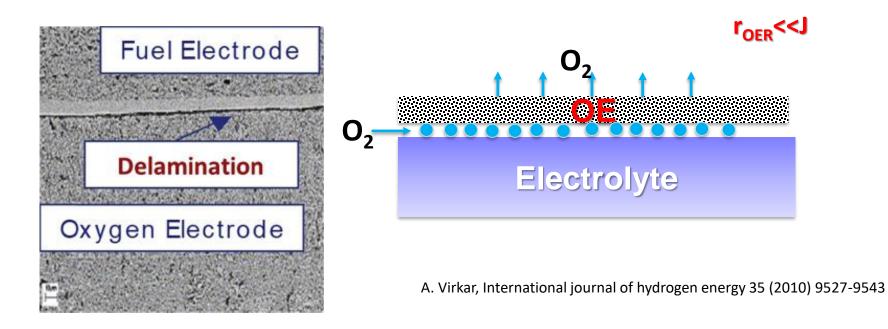
Approach

- Huang group at UofSC
 - Process optimization of isostructural bilayer OE for scaleup fabrication
 - Fast screening of the bilayer OE and performance evaluations at button-cell scale
 - Cell performance demonstrations at bench-scale and stack level
- Jin group at UMass Lowell
 - Development of a combined chemical and mechanical model
 - Identification of failure modes
 - Establishment of mitigation strategies

Milestone Status (FY19-20)

Date	Milestones	Completion
5/31/2020	Finalize bilayer OE loading	60%
8/31/2020	Finalize bilayer OE calcination condition	60%
11/30/2020	Finalize the design and fabrication of STEHC testing fixture	10%

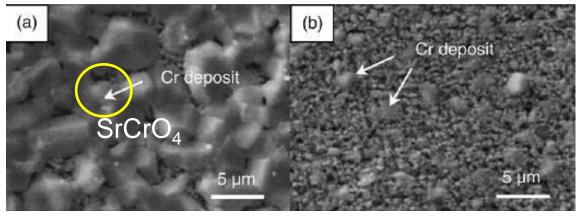
Project Background- Barrier #1



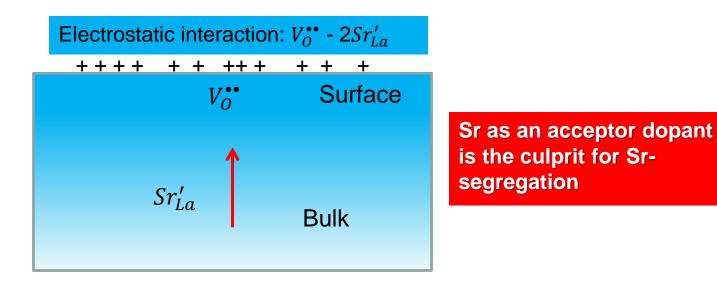
Coralli, et al. Science and Engineering of Hydrogen-Based Energy Technologies, Hydrogen Production and Practical Applications in Energy Generation, 2019, p39-122.

Project Background- Barrier #2

Cr-poisoning (LSCF at 900°C for 50h with Fe-Cr alloy)

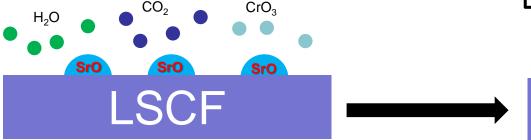


Solid State Ionics 179 (2008) 1459-1464



Surface Sr-Segregation: Consequences





• H_2O and CO_2 can compete with O_2 for adsorption on oxygen vacancy sites

- SrO + $H_2O \rightarrow Sr(OH)_2$
- SrO + $\overline{CO}_2 \rightarrow SrCO_3$
- SrO + CrO₃ \rightarrow SrCrO₄

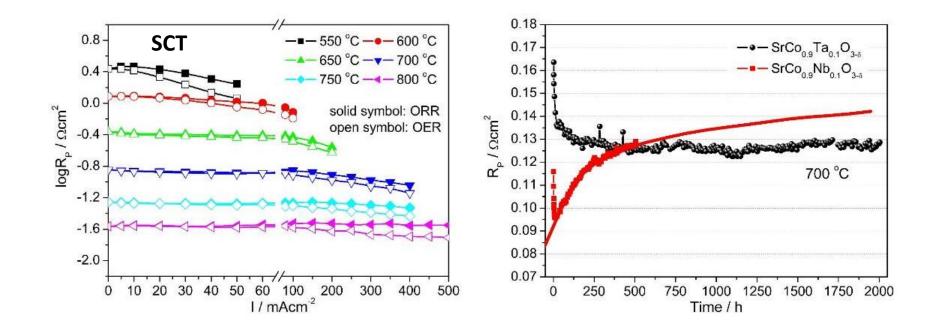


A New Bilayer OE Concept



Core: LSCF-GDC Shell: SrCo_{0.9}Ta_{0.1}O_{3-δ} (SCT)

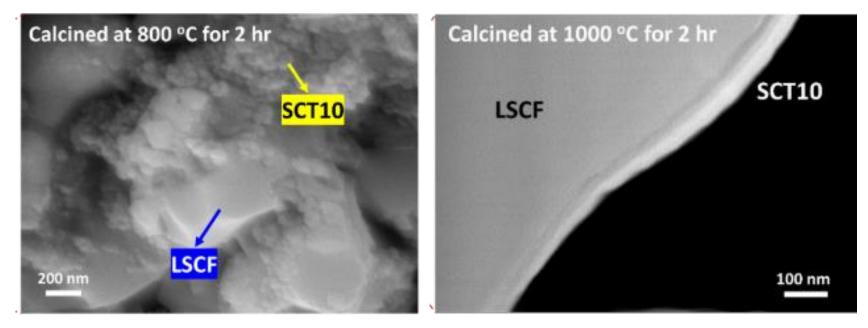
Prior Work on the Shell Material



Wang and Huang et al., Journal of Materials Chemistry A, 2017, 5, 8989-9002.

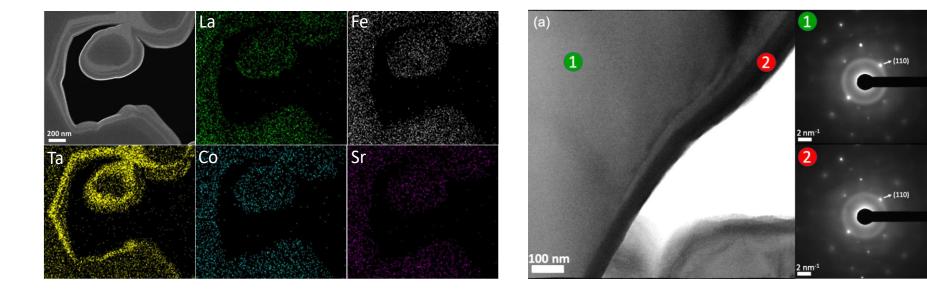
The Bilayer Structure

The calcination temperature effect

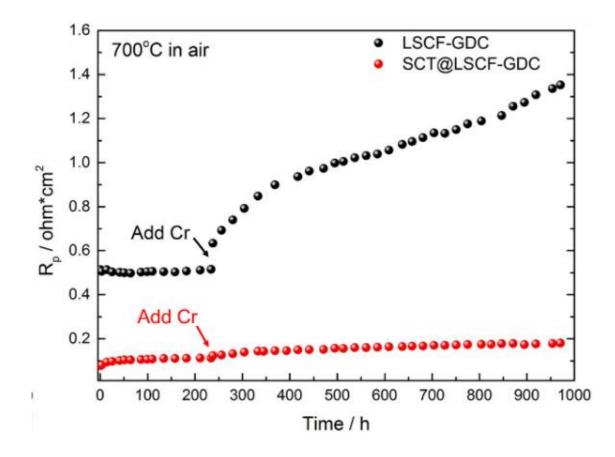


Yang and Huang et al. Journal of Materials Chemistry A, 2020, 8, 82-86

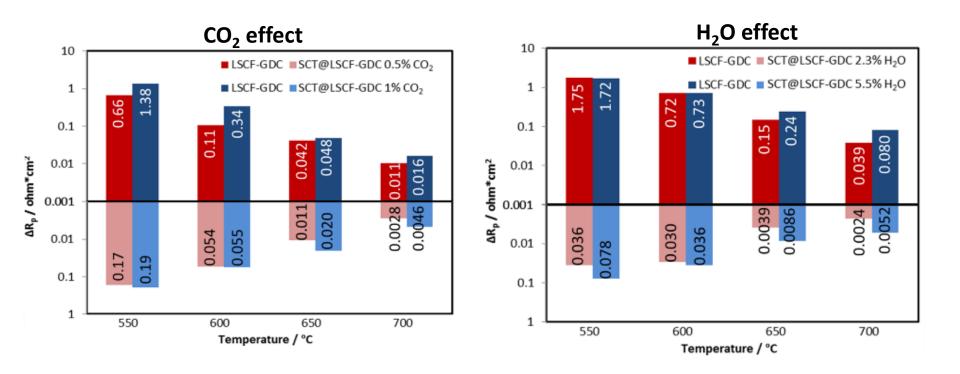
The Isostructural Bilayer Structure



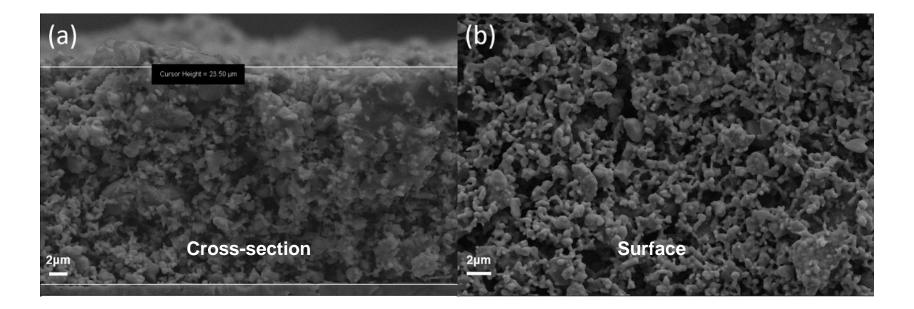
The Cr-Tolerance



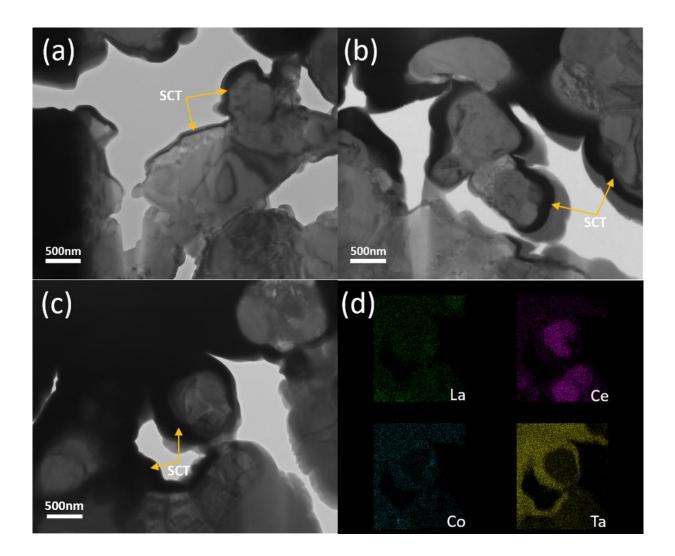
The CO₂ and H₂O Tolerance



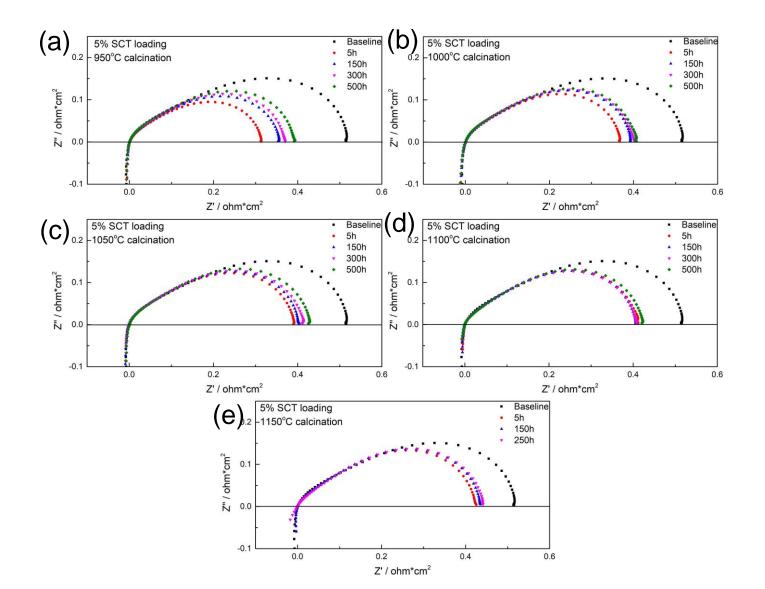
Accomplishment and Progress: LSCF-GDC Skeleton



SCT@LSCF-GDC

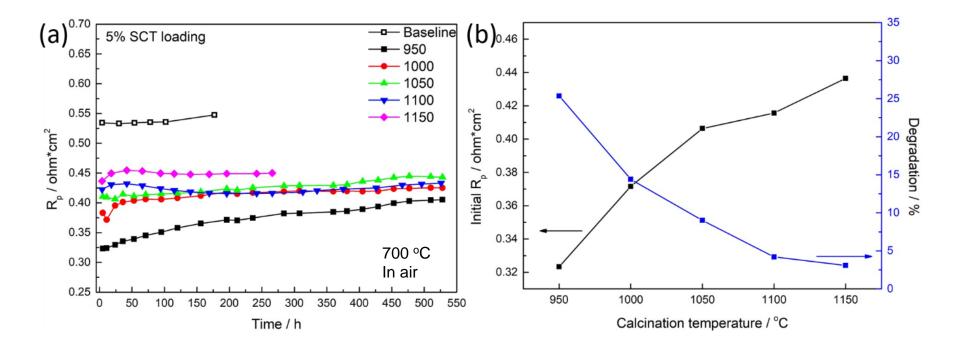


EIS Spectra of 5%SCT@LSCF-GDC

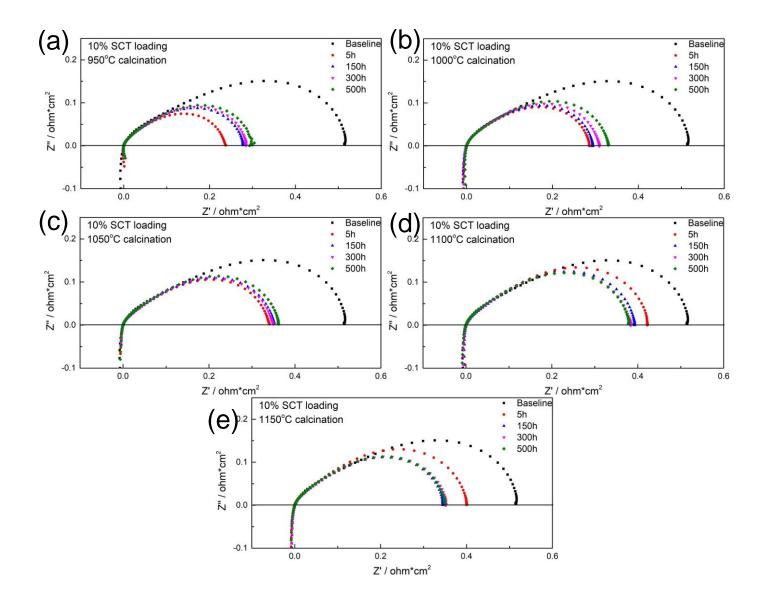


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Long-term stability of 5%SCT@LSCF-GDC

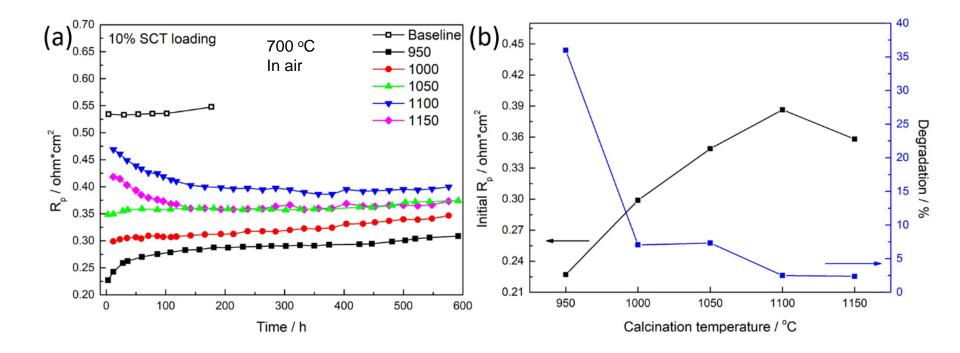


EIS Spectra of 10%SCT@LSCF-GDC

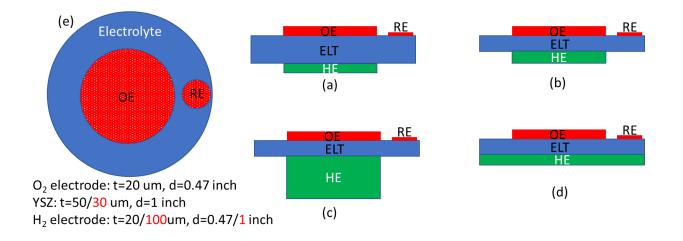


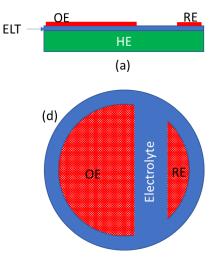
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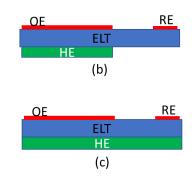
Long-Term Stability of 10%SCT@LSCF-GDC



Theoretical Analysis on 3electrode Configuration







 O_2 electrode: t=20 um, d=0.79 inch YSZ: t=30/50 um, d=1 inch H₂ electrode: t=500/20 um, d=0.79/1 inch

Modeling Results of Three-Electrode Overpotential Measurements

Top View	Cross-Section	Potential @ RE	Potential @ middle of ELT@0.4A/cm ²	Validity for 3- Electrode Measurement
Electrolyte OF RE	OE RE ELT HE	-0.310V	-0.310V	Valid
	OE RE ELT HE	-0.307V	-0.307V	Valid
	OE RE ELT HE	-0.240V	-0.240V	Valid
	ELT HE	0V	-0.307V	Invalid
Blettrolyte	, OE RE HE	0V	-0.250V	Invalid
	OE RE ELT HE	-0.307V	-0.307V	Valid
	OE RE ELT HE	-0.270V	-0.307V	Error

Collaboration and Coordination

- University of Massachusetts at Lowell
 - subcontractor, focusing on multiphysics modeling
- Idaho National lab
 - Nodes usage, to reproduce and validate bilayer OE at a larger scale
- National Renewable Energy Lab
 - Nodes usage, to develop thermochemical and electrochemical multiphysics models

Remaining Challenges and Barriers (FY20-21)

- Demonstration of bilayer OE milestone performance at button-cell level
- Demonstration of bilayer OE milestone performance larger-cell level.
- Establishing multiphysics models for degradation understanding and mitigation.

Proposed Future Work (FY20-21)

Milestone	Date	Description	Status
2.2	02/28/21	Meet button-cell performance: $\leq 0.15 \text{ V OER}$ overpotential at 1 A/cm ² for 1 kh @ 700 °C	Not started
4.1	02/28/21	Establish a Multiphysics model to describe OER degradation mechanisms	Not started
4.2	02/28/21	Demonstrate a combined thermal/electrochemical model	Not started
Go/No-Go	02/28/21	Demonstrate the best OER performance specified in Milestone 2.2	Not started
3.1	05/31/21	Successfully transfer the bilayer OE fabrication process to INL for scale-up fabrication	Not started

Technology Transfer Activities

 Officially filed a US utility patent on February 27, 2020, "Method to Make Isostructural Bilayer Oxygen Electrode", Serial No. 16/802,608.

Summary- Progress and Accomplishments

- Completed the loading-calcination-morphology-R_P relationship study for 5% and 10% SCT added LSCF-GDC OEs. A clear trending is observed.
- Built a new test rig (including steam generator, high temperature reactor, on-line MS gas analysis system) for button and tubular cells.
- Started anode-supported thin film electrolyte button cell fabrication
- Computationally studied seven 3-electrode configurations for possible overpotential measurement of the OE.