II.I.6 Robust Low-Cost Water-Gas Shift Membrane Reactor for High-Purity Hydrogen Production from Coal-Derived Syngas

Jim Torkelson, Neng Ye, Zhijiang Li, Mark Fokema (Primary Contact) Aspen Group Inc. 184 Cedar Hill Street Marlborough, MA 01752 Phone: (508) 481-5058 E-mail: fokema@aspensystems.com

DOE Technology Development Manager: Dan Cicero Phone: (412) 386-4826 E-mail: Daniel.Cicero@netl.doe.gov

DOE Project Officer: Arun Bose Phone: (412) 386-4467; Fax: (412) 386-4604 E-mail: arun.bose@netl.doe.gov

Contract Number: DE-FC26-05NT42452

Start Date: May 20, 2005 Projected End Date: May 31, 2007

Objectives

- Develop a water-gas shift (WGS) membrane (Figure 1) that replaces multiple process units with a single catalytic membrane reactor, has high process efficiency realized by improving hydrogen yield at low steam to carbon rations, and produces a highpressure CO₂-rich stream potentially suitable for sequestration.
- Develop a lower cost, robust WGS membrane reactor that can be used to process coal-derived syngas for the production of high-purity hydrogen.
- Develop and demonstrate a contaminant-tolerant, highly active WGS catalyst.



FIGURE 1. Membrane Design

- Develop and demonstrate a selected Pd/Cu coated Ta membrane with improved durability.
- Construct and demonstrate a bench-scale WGS membrane reactor.
- Demonstrate the scalability of the technology by constructing a 500 L hydrogen/hr production capacity unit.
- Perform a study on the economic feasibility of the WGS membrane reactor.
- Analyze cost benefits of the demonstrated WGS membrane reactor technology.



Introduction

There exists a need to develop a lower cost, robust WGS membrane reactor that can be used to process coal-derived syngas for the production of high-purity hydrogen. The required characteristics of this WGS membrane reactor are a contaminant-tolerant, highly active WGS catalyst and a selective Pd/Cu coated Ta membrane with improved durability. Such a membrane reactor should be constructed and demonstrated at the bench-scale to demonstrate the scalability of the technology, preferably at 500 L hydrogen/hr production capacity. WGS membrane reactors incorporating a highly selective H₂-permeable membrane offer the advantages of: 1) generating more hydrogen in a single reactor, instead of conventional two-stage (high-temperature and low-temperature) shift process, 2) simplifying or completely eliminating downstream hydrogen separation and purification processes, and 3) producing a high-pressure stream of predominantly CO₂ for cost-effective capture and sequestration. However, despite these obvious advantages, H₂ generation from syngas employing WGS membrane reactors has not been commercialized in any significant scale, primarily due to the unavailability of reliable, highperformance and economical membrane reactors and WGS catalysts. This project will address these technical barrier issues.

Approach

Initial work will be directed towards the development of a contaminant-tolerant, highly active WGS catalyst (MoS_2 -based with added modifiers). A suitable Pd/Cu coated Ta membrane, capable of promoting H_2 dissociation/association reactions in the presence of S and Cl, with improved durability, will

then be selected for development and demonstration. Both the WGS catalyst and the H_2 -selective membrane must be capable of operating at 300-500°C, 300-500 psig and remain tolerant of high concentrations of H_2S and HCl. Using data from these tests, a bench-scale WGS membrane reactor will then be constructed and demonstrated. Finally, to show that the WGS membrane reactor technology is scalable, a 500 L hydrogen/hr system will be built and an economic feasibility study of the WGS membrane reactor will then be performed.

Accomplishments

- Completed preparation of 10-16 samples of both alumina-supported and unsupported nanosized Mo₂C and sulfided Mo, MoCo and CoCr catalysts and completed characterization of the catalysts by investigating the structural properties and determining their surface area, pore size, agglomerate size, composition, and crystallinity.
- Completed evaluation of the catalysts prepared in Q1 and compared with commercial hightemperature WGS catalysts for their WGS activity and stability in a high-pressure fixed-bed catalyst testing system. Identified and selected the most active and stable WGS catalysts and the best operating conditions.
- Completed the development of Pd/Cu-coated Ta tubular H₂ membrane elements and studied the use of an electroless plating method to fabricate Pd/Cu-coated Ta membrane tubes with different wall thicknesses.
- Identified contaminant-tolerant WGS catalysts with high activity over a wide temperature range (300-500°C). The catalysts demonstrated excellent durability in the presence of 3,000 ppm H₂S and 350 ppm HCl for over 220 hours (see Table 1).

Portormanco Oritoria	Current Statue	Targ	This Work		
renonnance ontena	Guileni Sidius	2010	2015	THIS WORK	
Catalyst Form	Pellets	Advanced config	Pellets		
Active Metal	Cu/Zn or Fe/Cr or Co/Mo	Advanced config	Mo		
Feed Temperature (°C)	200-300	>250	>400	>250	
Feed Pressure (psia)	450-1150	>450	>750	415	
Approach to Equilibrium (°C)	8-10	<6	<4	<4	
Min. H ₂ O/CO Ratio	2.6	<2.6	<2	<2	
Sulfur Tolerance (ppm _v)	varies	>20	>100	3000	
COS Conversion	varies	Partial	Total		
Chloride Tolerance (ppm _v)	varies	>3	>100	350	
Durability (y)	3-7	>7	>10		
Catalyst Cost (\$/lb)	~5	<5	<5		

TABLE 1. Result Summary – WGS Catalyst

• Identified dense metal membranes with high H_2 permeability and selectivity. The membranes maintained high permeability in the presence of 2,000 ppm H_2S and 33% H_2O .

Future Directions

• Complete characterization and testing for H₂ separation (see Table 2) in order to identify and select the most stable and efficient membranes and associated membrane preparation parameters for building a WGS reactor and scale-up.

	Current Status			Targets			
Performance Criteria	Microporous	Cermet	Dense Ceramic	2007	2010	2015	This Work
Flux (scfh/ft ²)	20-100	~300	2	100	200	300	20
Temperature (°C)	300-600	300-400	900	400-700	300-600	250-500	350-550
S Tolerance (ppm _v)	Yes	~20	-	-	20	>100	2000
Cost (\$/ft ²)	150-200	<200	-	150	100	<100	
WGS Activity	No	•	-	Yes	Yes	Yes	Yes
∆P Capability (psig)	100	1000	-	100	400	800	400
CO Tolerance	Yes	Yes		Yes	Yes	Yes	
H ₂ Purity (%)	90-98	>99.999	-	95	99.5	99.99	>99.9
Durability (y)		0.9	-	1	3	5	

TABLE 2. Result Summary – Hydrogen Separation

- Quantify performance and stability of membrane in contaminant-laden syngas.
- Examine membrane microstructure in order to relate permeability to structure and synthetic parameters.
- Fabricate and demonstrate integrated WGS membrane reactor.
- Conduct cost analysis.