

Large Area Cell for Hybrid Solid Oxide Fuel Cell Hydrogen Co-Generation Process

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Ceramatec, Inc.

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This presentation does not contain any
proprietary or confidential information

Project ID #PDP12

Overview

Timeline

- Project start date
 - May 2005
- Project end date
 - May 2007
- Percent complete: 0%

Budget

- Total project funding
 - DOE share
 - Contractor share
- No FY04 Funding
- Funding for FY05
 - None YTD

Barriers Addressed

- H₂ from natural gas or renewable liquids
 - A – Capital Cost
 - E – CO₂ emissions
- H₂ generation by water electrolysis
 - G – Capital Cost
 - H – System Efficiency
 - I – Grid Electricity Emissions
 - J – Renewable Integration
 - K – Electricity Cost

Partners

- Ceramatec, Inc.
- Hoeganaes
- Idaho National Laboratory
- University of Washington

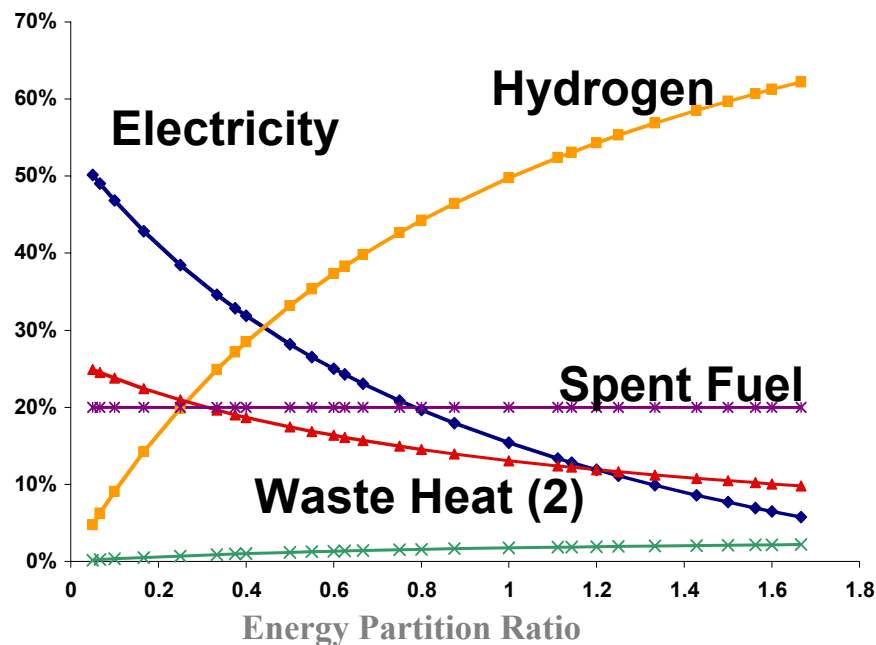
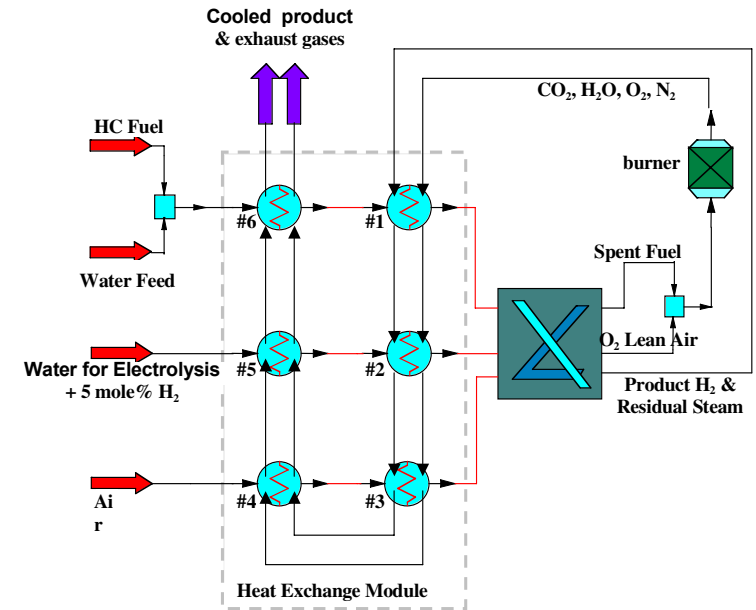
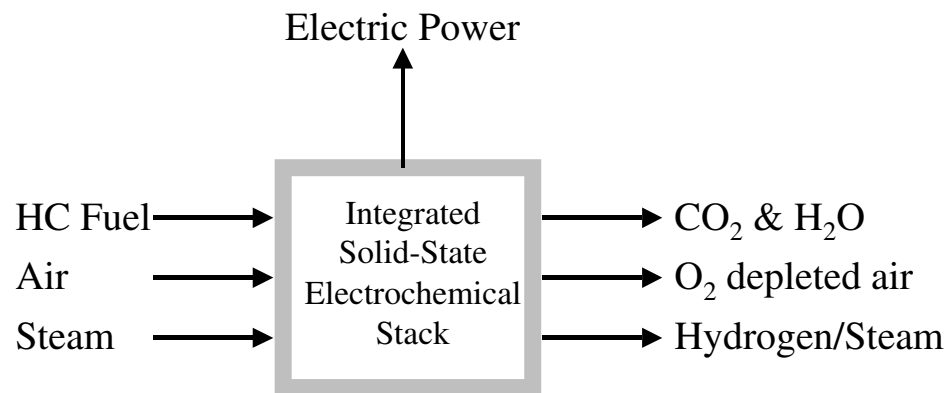
Project Objectives

Project Objectives	Challenges	Overall Concept	Specific Activity	Team Members
* Low cost hydrogen generation * Cogeneration of hydrogen and electricity	* Thermal Management (operational limit) * Cell size (fabrication limit)	* Physical, chemical and thermal integration of fuel cell/electrolysis functions – allows operation at near thermal neutral condition * Large area cell fabrication by the use of porous metal substrate	Substrate alloy selection / fabrication	Ceramatec/ Hoeganaes
			Layer Deposition - Thermal Spray	INEEL
			Slurry coating and constrained sintering	Ceramatec/ U of WA
			Materials Selection Stack Test Process Model /Data Analysis	Ceramatec
			Cost Analysis	All

Process Objectives

Process objectives	Hybrid design features	Key benefits
Direct PEM grade H ₂	Steam electrolysis	No shift or CO cleanup required
Electric power	Interleaved SOFC cells	Co-generation of hydrogen and electricity
Eliminate POx penalty	Electrochemical process	High Faraday and Nernst efficiencies
Thermal management	Thermal integration	Temperature/resistance uniformity, reduced thermal stress and air preheat duty, large area cells
Design flexibility	Cell function ratio	Selective energy partitioning, H ₂ :electric power ratio
Carbon sequestration	Nitrogen free reformat	Non-condensable free exhaust

Technical Approach



- Extension of NASA SBIR
- Leverage SOFC Development
- Integrated Hybrid Stack
- Continuous H₂ Cogeneration
- Natural Gas Fueled
- Optimized Thermal Management
- Enables Large Area Cell ⁵

Technical Accomplishments/ Progress

- Demonstrated >100 nlph hydrogen production in stack at INL (DOE NE NHI program)
- Operated advanced high performance cathode supported LSGM electrolyte in electrolysis mode (DOE FE SBIR)
- Addressed seal accelerated corrosion issue under electrolysis conditions (DOE Membrane Seal SBIR)
- Characterized metal interconnect scale growth in dual (reducing/oxidizing) atmosphere conditions (DOE SECA CTP project)

Future Work

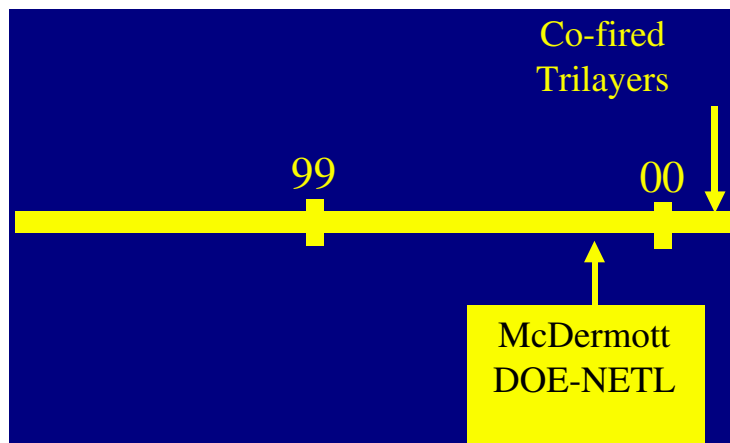
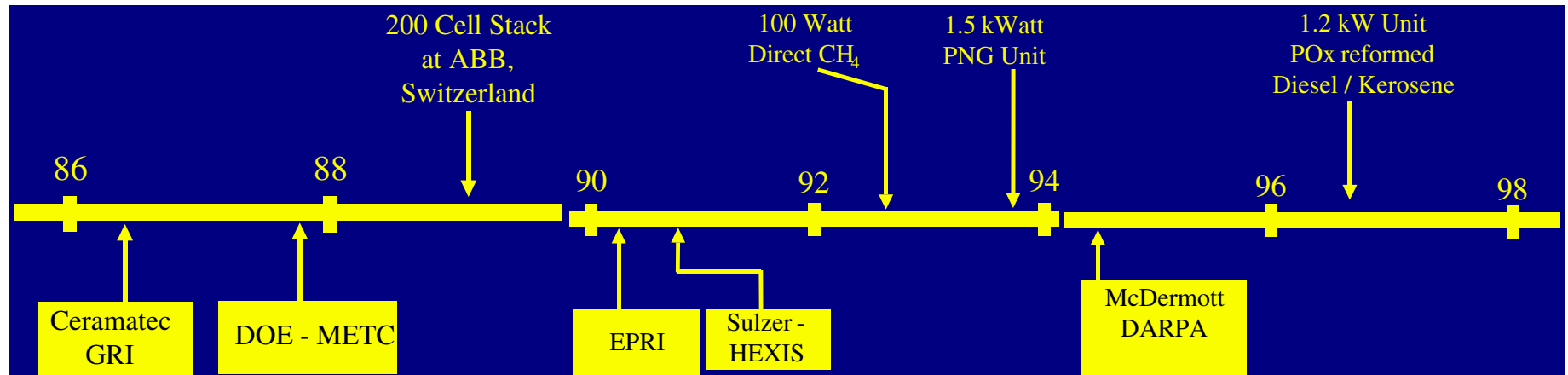
- Begin project per proposed work scope
 - Review objectives with team members
 - Produce powder metal heats
 - Form sintered porous metal plate substrate
 - Pre-treat PM substrate for low resistance scale
 - Develop thermal spray deposition of cell layers
 - Develop thermal crack healing for gas tight membrane
 - Begin electrochemical characterization of PM supported cells in electrolysis mode

SOFC Electrolysis Technology Fit

- High temperature Solid Oxide Fuel Cells (SOFC)
 - Generation of electricity and heat
 - Hydrogen or hydrocarbon reformat fuels
- High Temperature Electrolysis (HTE)
 - Reversed SOFC current generates hydrogen from steam
- Commonality of SOFC & HTE
 - Material sets, fabrication methods, stack design, modeling
 - Performance
 - Seamless transition between operating modes
- Multi-mode technology for transition to hydrogen economy
 - **Transitional technology**
 - Distributed power generation using hydrocarbon fuels
 - **End point technology**
 - Hydrogen fuel production from renewable energy or nuclear energy

Ceramatec SOFC History

- 19 years of SOFC R&D



SBIR & SECA Contracts

Recent SOFC-derived Contracts

- **DOE-FE - SBIR**

- | | | |
|-------------------------------------|----------|------------|
| – High Temp. Heat Exchanger | Phase II | current |
| – Hydrogen Separation Membrane | Phase II | current |
| – Intermediate Temp. SOFC | Phase II | Aug 04 end |
| – SOFC Insulation Material | Phase II | awarded |
| – Glass composite seals | Phase I | awarded |
| – Improved Cathodes for SOFC | Phase I | complete |
| – Pre-ceramic polymer seal for SOFC | Phase I | complete |

- **DOE-FE (non-SBIR)**

- | | | |
|---|----------|---------|
| – Metal Interconnects for SOFC (SECA) | Phase II | current |
| Initial work done under an SBIR-Phase I | | |
| – SECA industrial team participation | Phase I | 4 Years |
| Cummins-SOFCo team subcontract | | |

Recent SOFC-related Contracts

- **NASA - SBIR**

- Integrated SOFC System Phase II current
 - Electrolysis/SOFC hybrid cogeneration of H₂ & Power

- **Air Force SBIR**

- Integration of JP-8/diesel reformer and SOFC
Phase II awarded

- **DOE-FE SBIR**

- Environmental Barrier Coating Phase II
Application of metal interconnect technology from Phase I
SBIR and Phase I SECA

Investments in R&D Infrastructure

- 90,000 ft² Manufacturing and R&D Facility
- Start to Finish Ceramic Processing
 - Lab-Scale to Pilot-Scale to Production
 - Class 10,000 Clean room
- Well equipped Materials R&D Characterization Laboratory



Lamination



Inspection & Analysis



Tape Casting



Ceramic Powder



Laser Cutting



Sintering

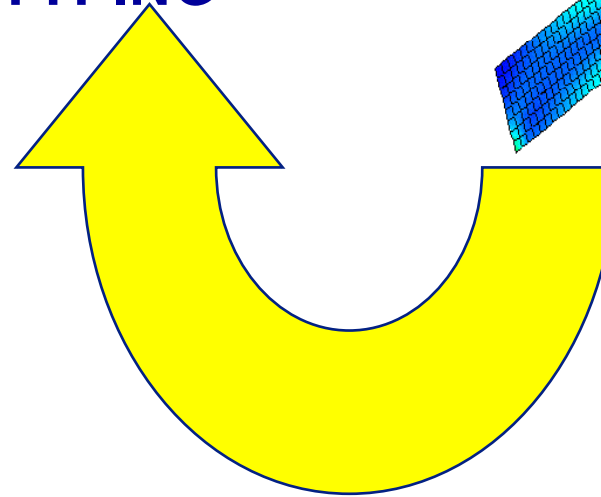
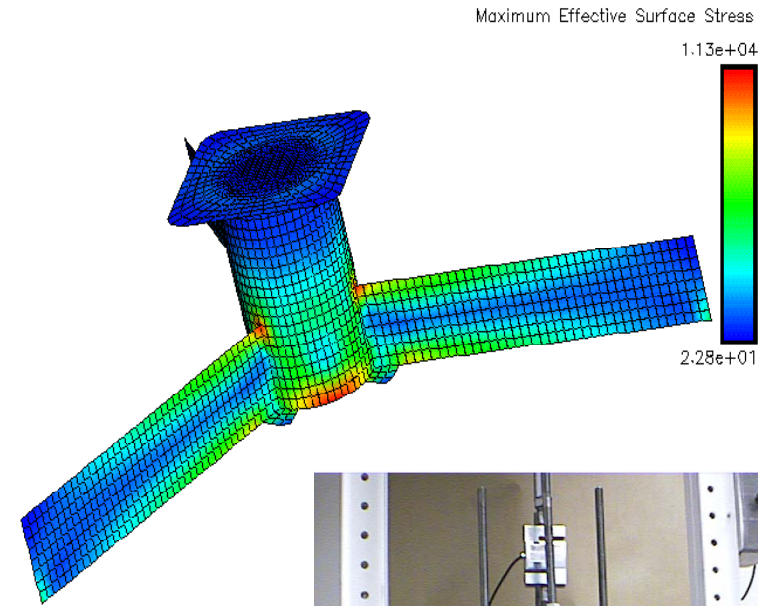
Investments in Background Technology

- Nearly \$100 million invested over 20 years
 - DOE & DOD funding
 - Utility R&D Groups
 - EPRI & GRI
 - Industry consortium
 - NorCell - Norsk Hydro, Saga Petroleum, Elkem, NTNF
 - Ceramatec partners
 - McDermott/SOFCo
 - Air Products
 - Ceramatec
- Technology and facilities available to this project.

Ceramatec's Core Activities – Integrated Engineering, Processing and Prototyping

**DESIGN &
ENGINEERING**

**DESIGN
VALIDATION
&
PROTOTYPING**



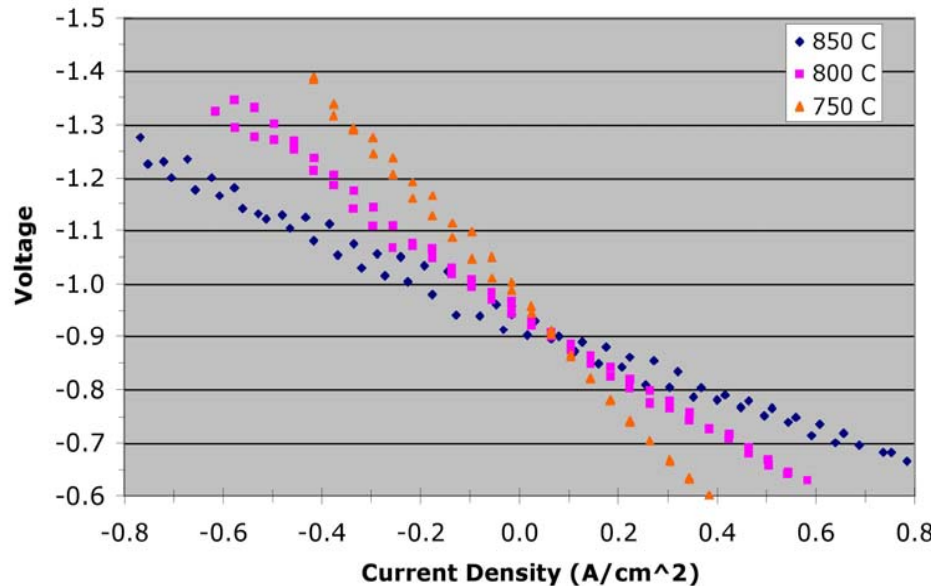
**PROCESSING
DEVELOPMENT
(Lab to Pilot Production)**

TESTING



SOFC Performance in Reversible Mode

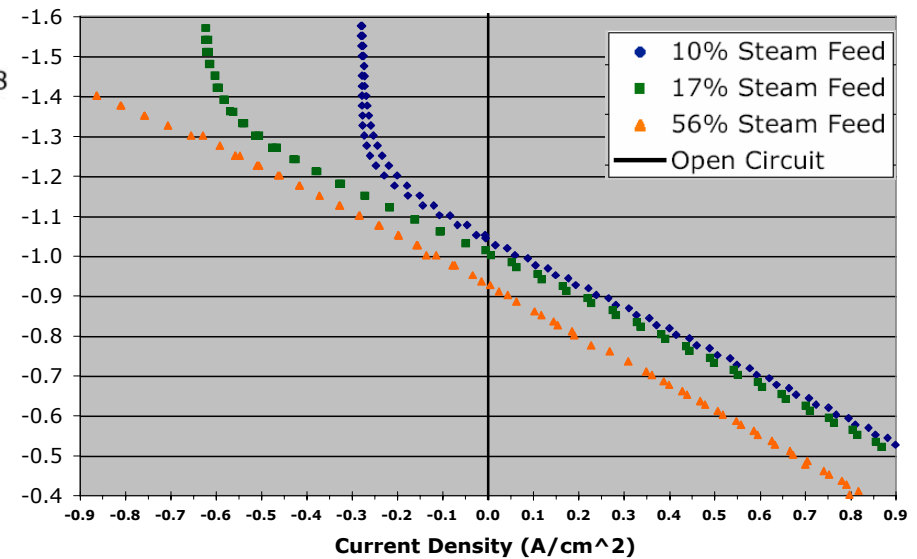
Reversible SOFC Electrolysis



Performance symmetry about OCV not changed by operating temperature

Performance symmetry is limited by reactant availability

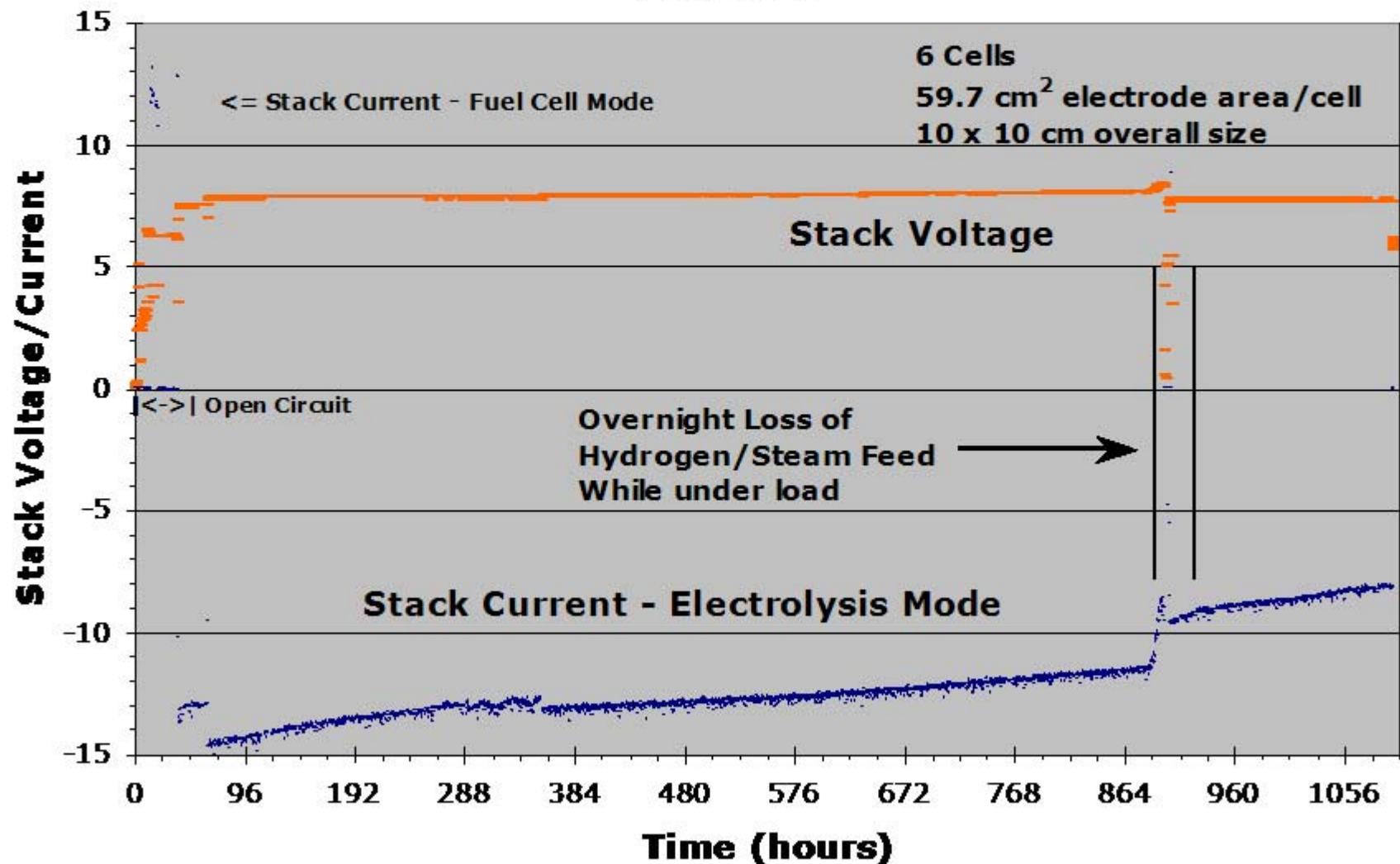
LSGM Reversible Fuel Cell & Hydrogen Generator



Inaugural INL Project Stack Test

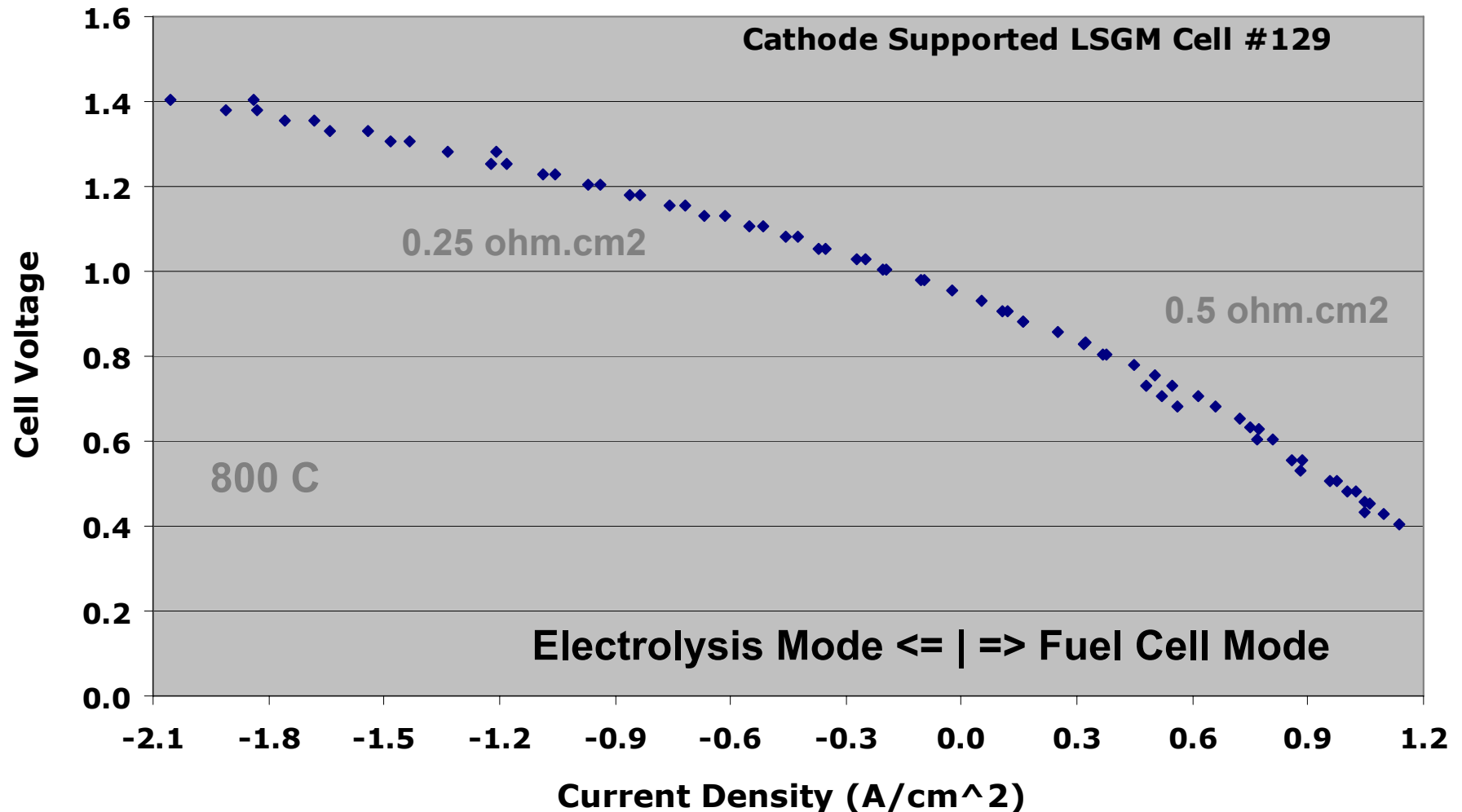
Electrolysis Stack #410

Fall 2003



Cross Program Derived HTE Advance

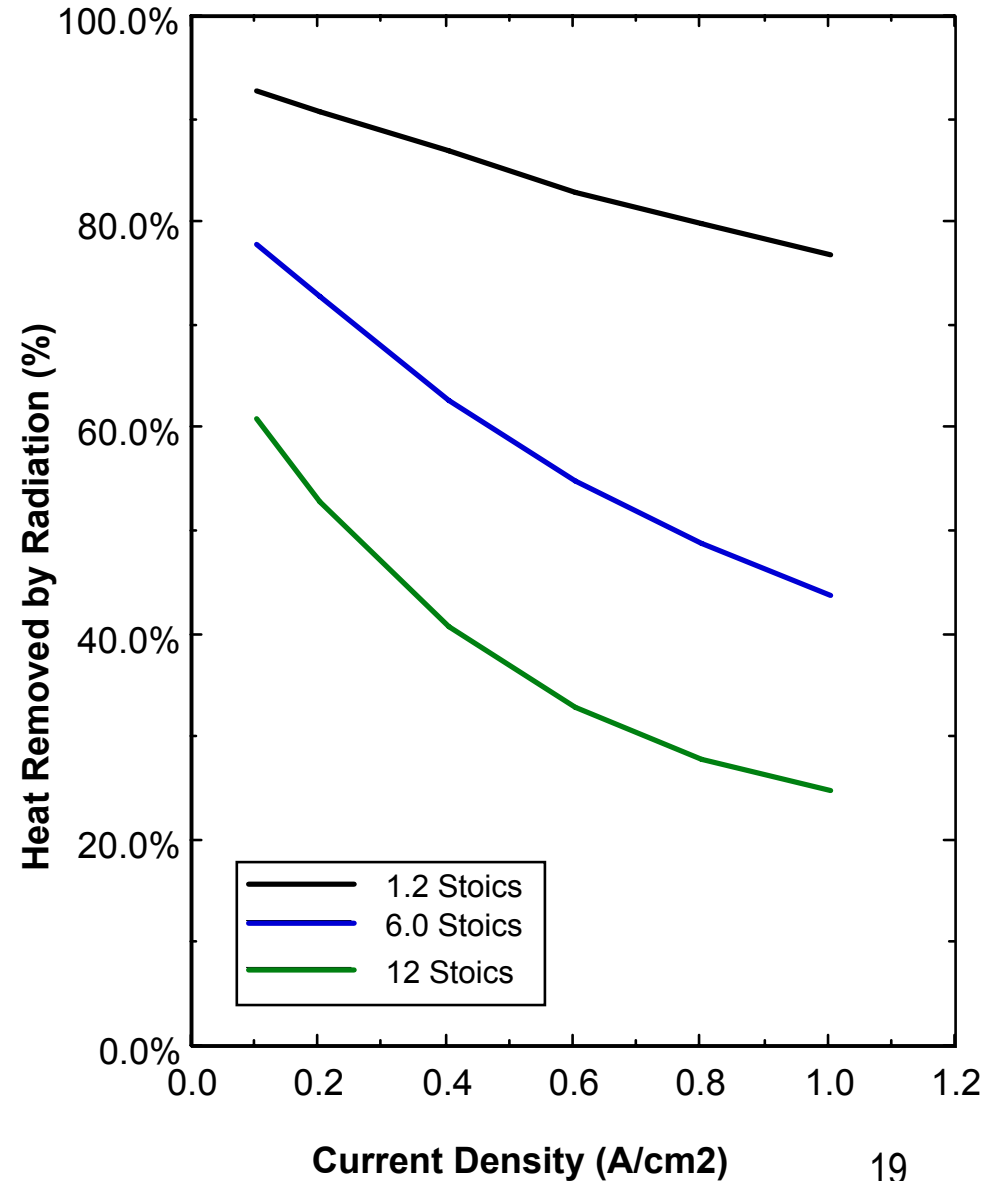
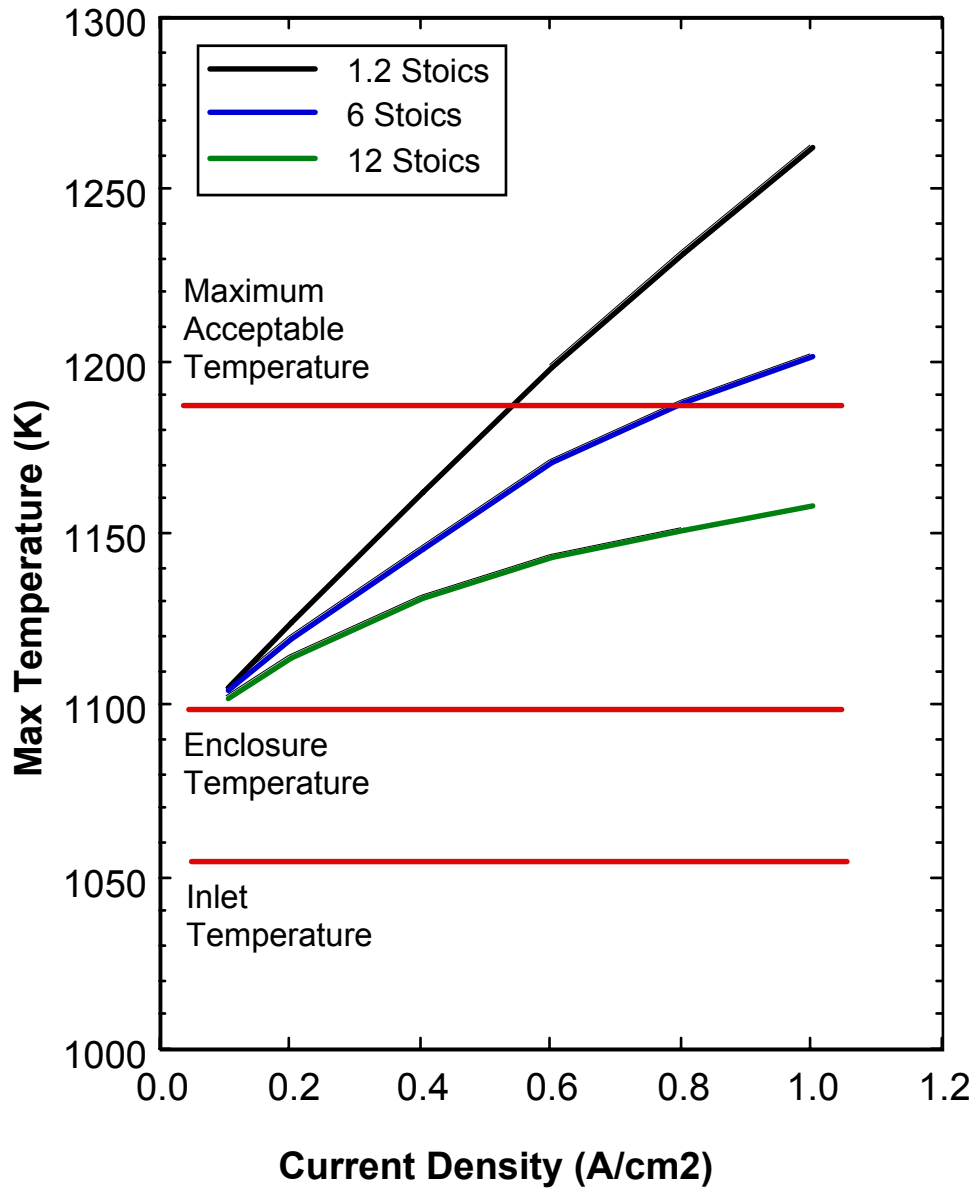
DOE-FE SBIR: Intermediate Temperature Fuel Cell (LSGM)



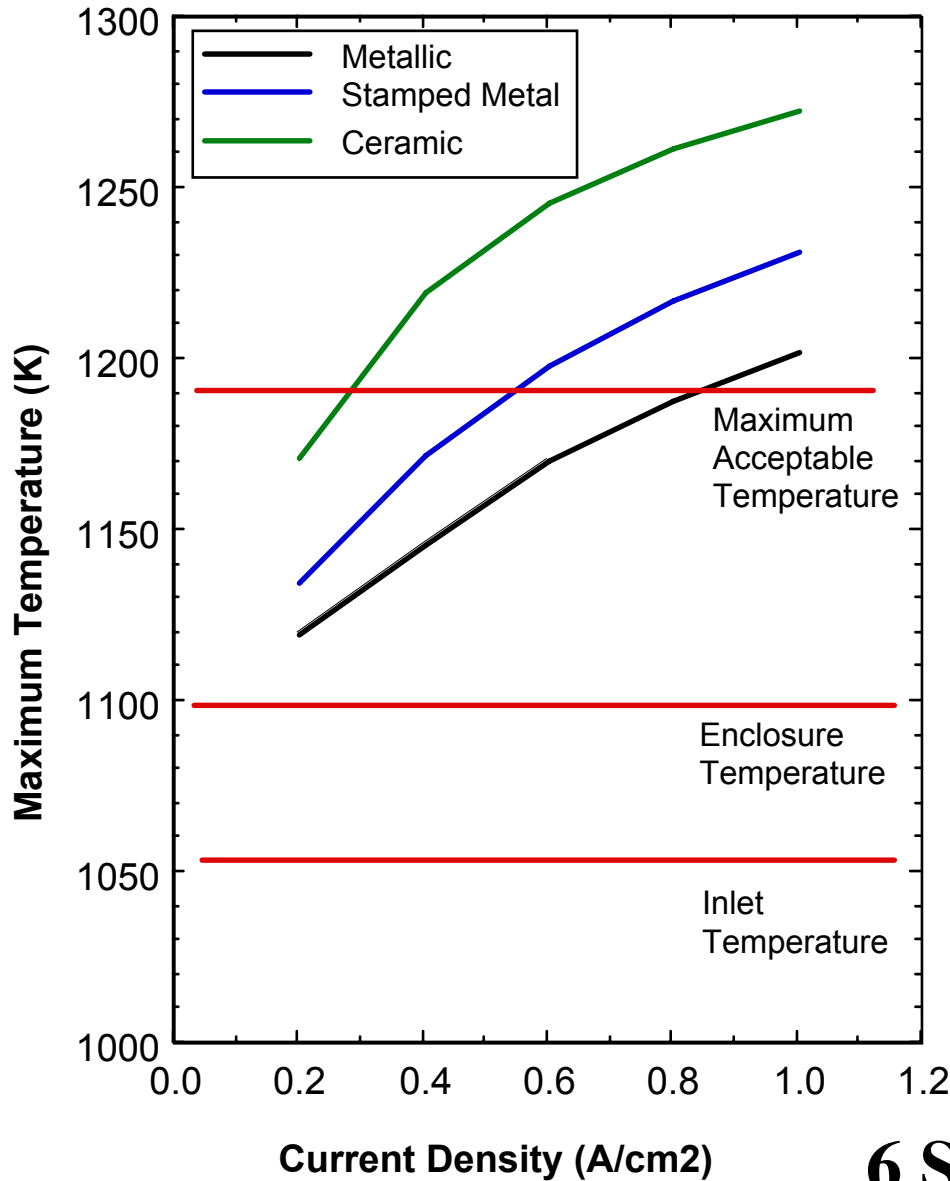
Thermal Management Impacts Cell Size

- Heat rejection limits size of SOFC
 - $T\Delta S$ term is exothermic in SOFC mode
 - $T\Delta S$ term is endothermic in electrolysis mode
 - I^2R is exothermic for both modes
 - Models show counteracting $T\Delta S$ and I^2R terms simplifies operation in electrolysis mode
- Current fabrication methods also limit size
- Large area fabrication route needed for electrolysis cells (e.g. 1 m² active area cells)

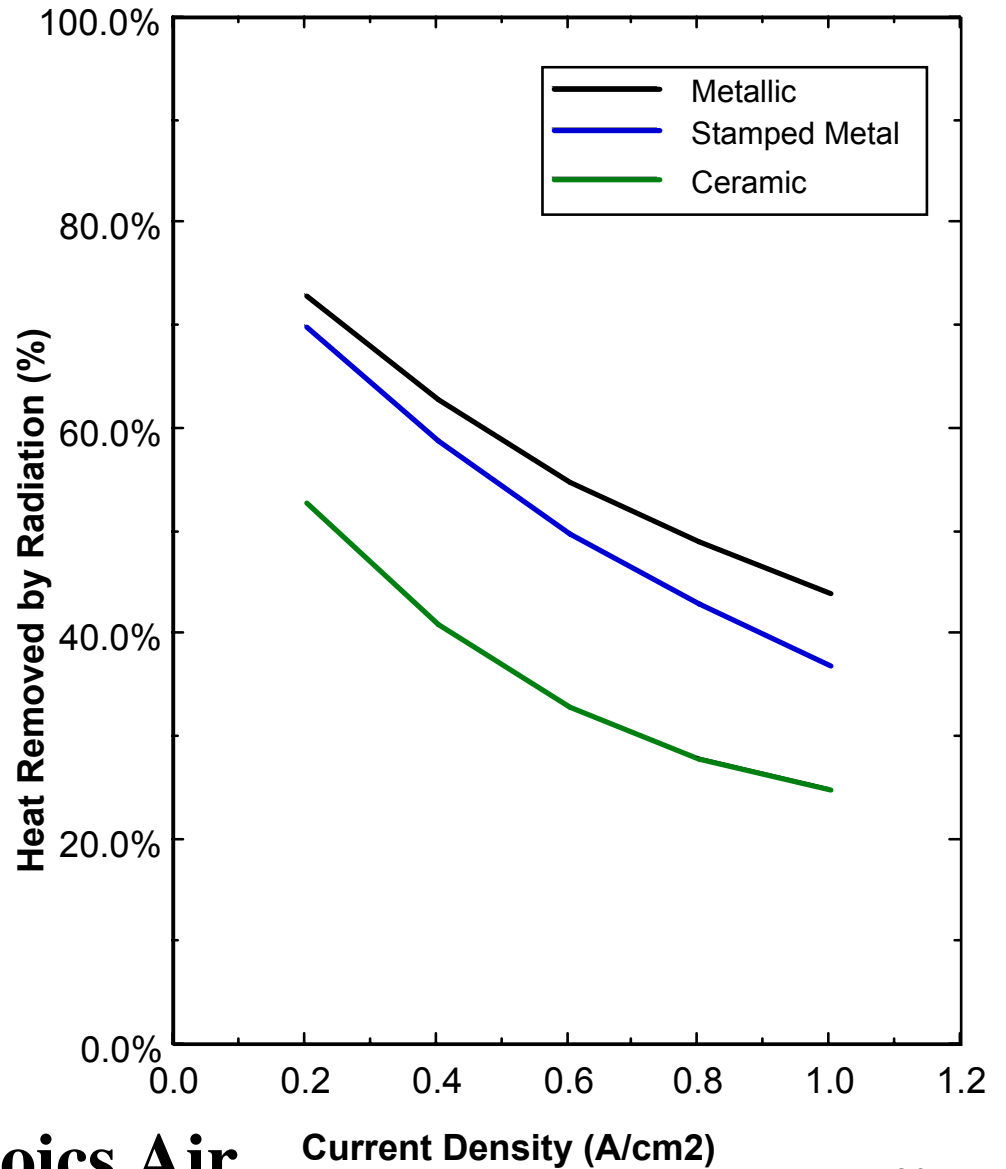
Excess Air Required to Cool 10cm Metal Icon Stack



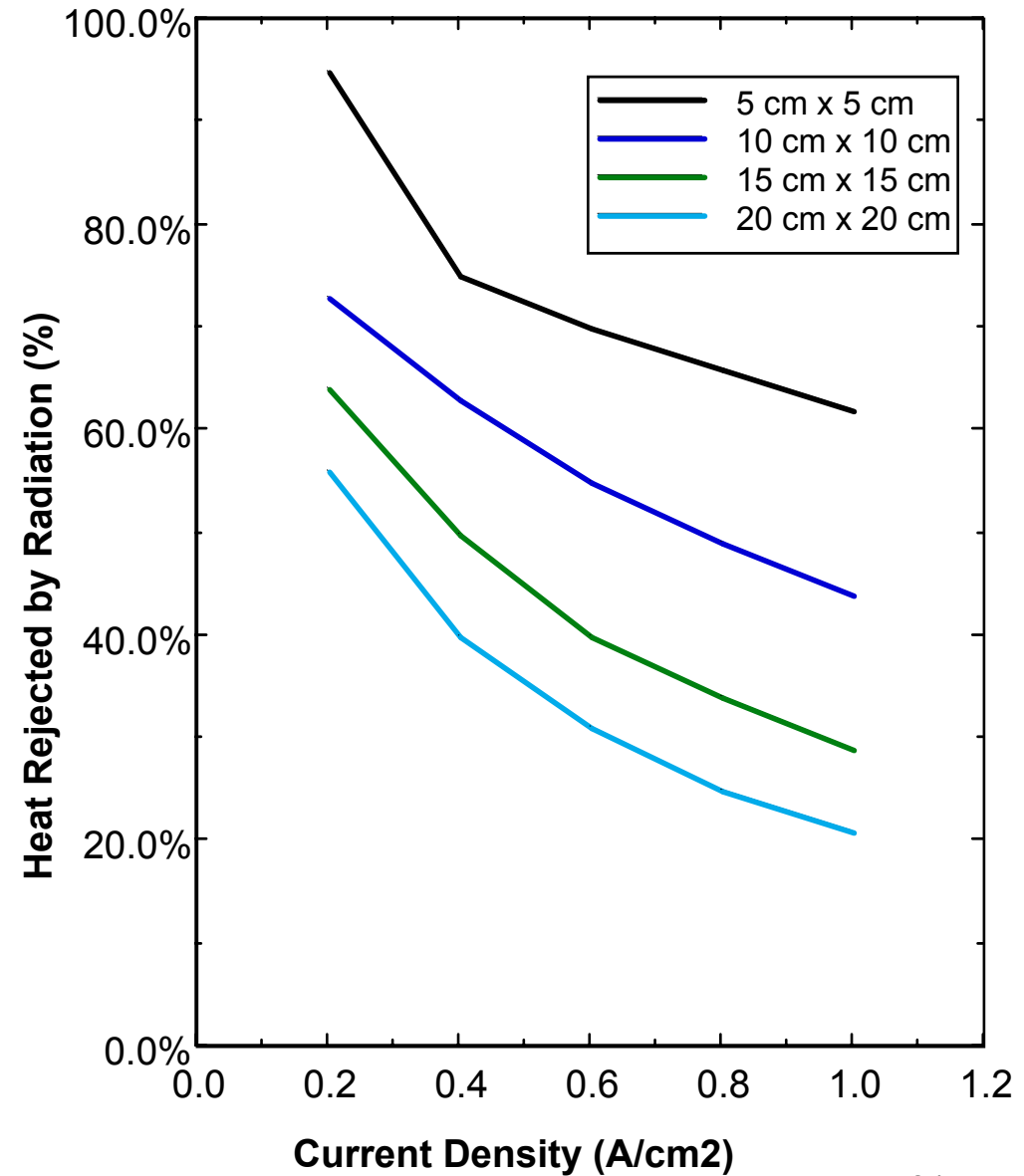
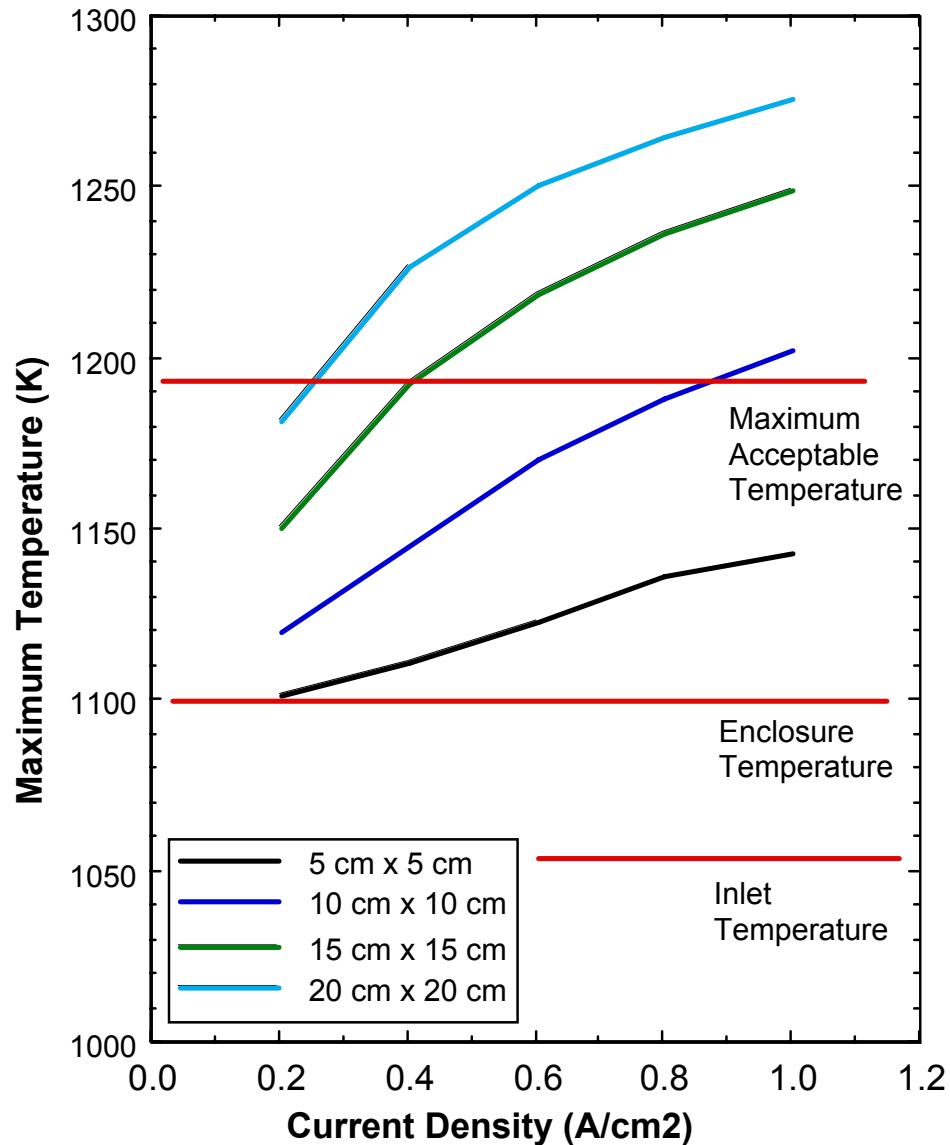
Effect of Interconnect Material on Stack Cooling



6 Stoics Air



Difficulty in Cooling Stacks of Large Area Cells



High Temperature Electrolysis Operation At V_{tn}

min: 1.10e+03, node 147497
max: 1.10e+03, node 114234

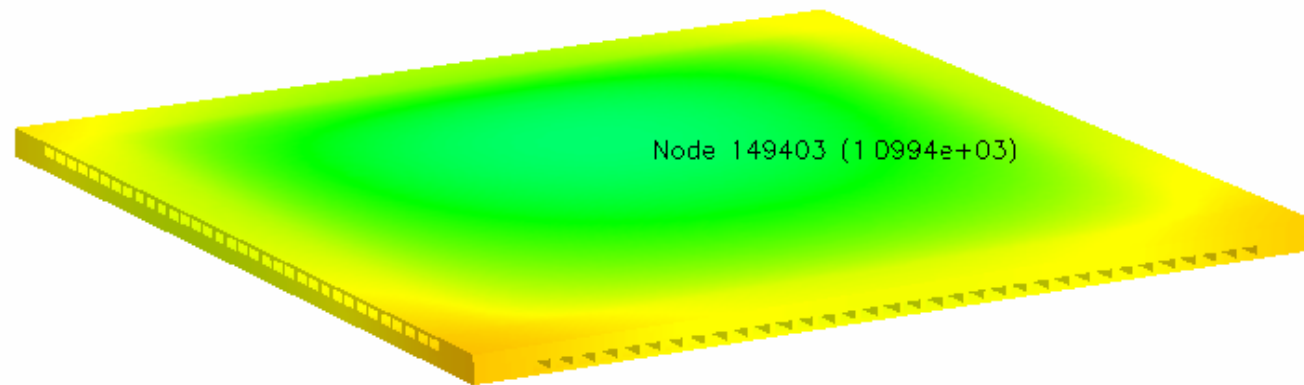
$$V_{op} = 1.288 \text{ V}$$

$$I = 21.37 \text{ A}$$

$$T = 1100 \text{ K}$$

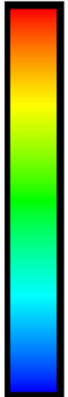
Feed: $\text{H}_2\text{O}:\text{H}_2$ 90:10 $4.39\text{e-}6$ mol/sec-channel

10% of SOFC Air $4.2\text{e-}6$ mol/sec-channel



Temperature

1.10e+03

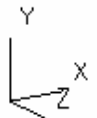


1.10e+03

Isothermal

HTE Operation at V_{tn}

t = 1.00000e+00



HTE Operation Above V_{tn}

min: 1.11e+03, node 114234
max: 1.12e+03, node 150661

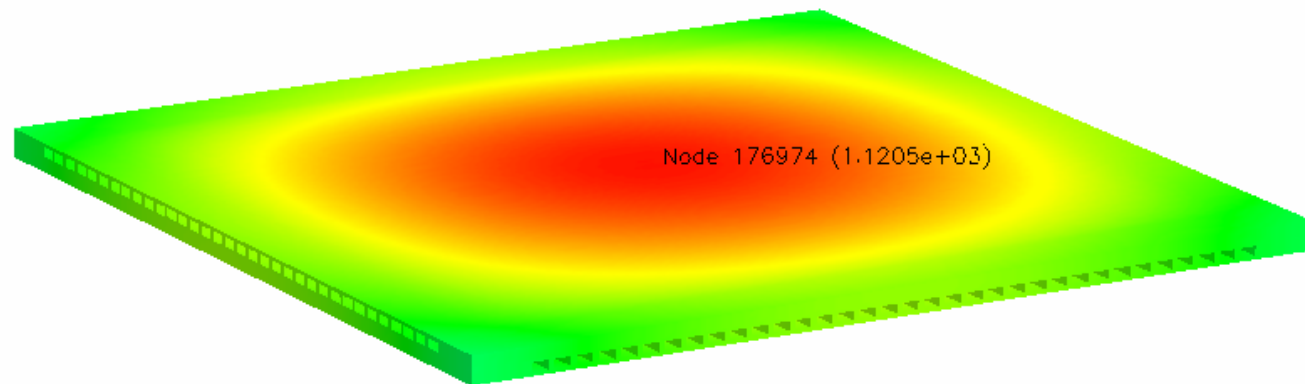
$$V_{op} = 1.45 \text{ V}$$

$$I = 33.87 \text{ A}$$

$$T = 1100 \text{ K}$$

Feed: $\text{H}_2\text{O}:\text{H}_2$ 90:10 $6.60\text{e-}6$ mol/sec-channel

10% of SOFC Air $4.2\text{e-}6$ mol/sec-channel



Exotherm produces $\sim 10^\circ\text{C}$ temperature rise

HTE Operation Below V_{tn}

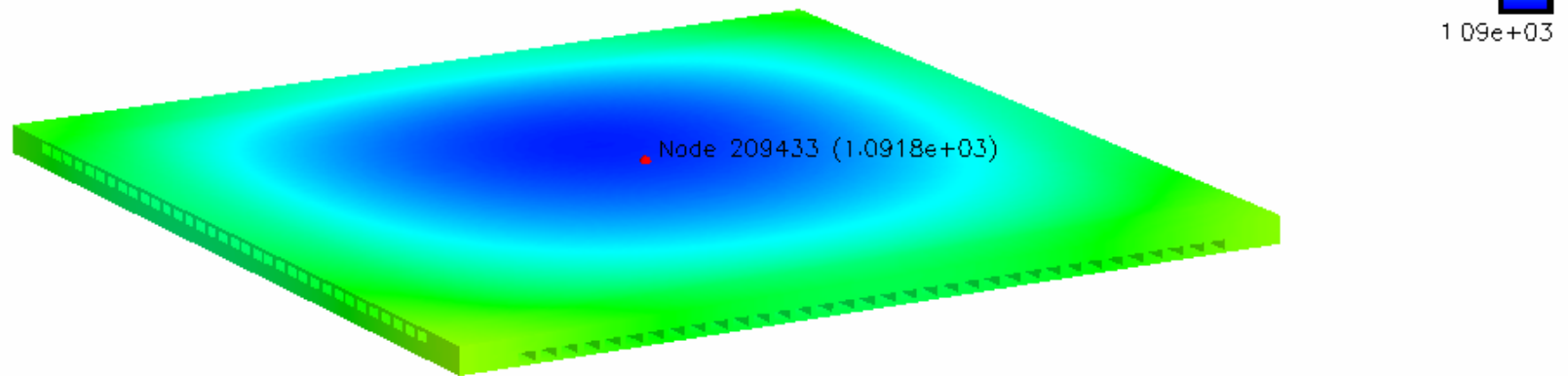
min: 1.09e+03, node 148765
max: 1.10e+03, node 114234

$$V_{op} = 1.15 \text{ V}$$

$$I = 11.49 \text{ A}$$

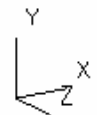
$$T = 1100 \text{ K}$$

Feed: $\text{H}_2\text{O}:\text{H}_2$ 90:10 $3.00\text{e-}6$ mol/sec-channel
10% of SOFC Air $4.2\text{e-}6$ mol/sec-channel



Endotherm produces $\sim 8^\circ\text{C}$ temperature drop

HTE Operation at 1.115V
t = 1.00000e+00



SOFC Operating Point

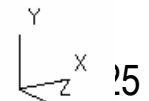
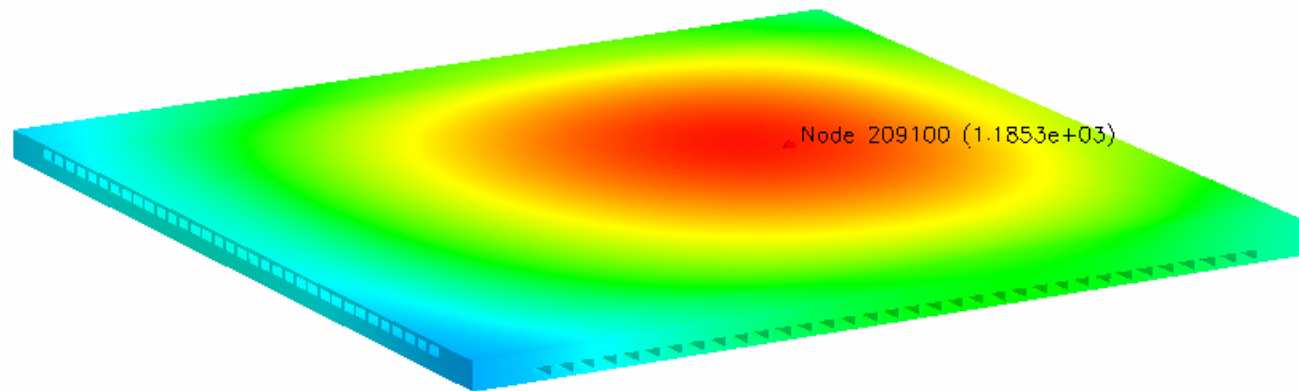
min: 1.12e+03, node 114234
max: 1.19e+03, node 150088

$$V_{\text{op}} = 0.65 \text{ V}$$

$$I = 21.02 \text{ A}$$

$$T = 1100 \text{ K}$$

Feed: H₂O:H₂ 10:90 4.39e-6 mol/sec-channel
Full SOFC Air 4.2e-5 mol/sec-channel



SOFC Operation at 0.65V – no end heat loss
t = 1.00000e+00

Project Summary

- DOE Hydrogen Production & Delivery Research
 - 4 Team Members
 - Ceramatec, lead, cell testing, metal coating
 - Hoeganaes, metal powder and foams
 - INEEL, thermal spray processing
 - Univ. of Washington, constrained sintering
 - Develop processes scalable to 1m² active area
 - Cell design based on thermal spray process using porous powder metal substrate
 - Industry Cost Share > 20%
 - Non-nuclear power based electrolysis
 - Distributed co-generation of hydrogen and electric power
 - DOE Power Park concept

Publications and Presentations

Related Electrolysis Programs (None yet on this program)

- National Hydrogen Association SBIR Workshop April 1, 2005
- ASME 3rd Int. Conf. On Fuel Cell Sci. Ypsilanti, MI, May 2005
- University of Utah Graduate Seminar – Feb 2005
- Joint IEA/AIE Workshop, San Antonio Nov 2004
- European SOFC Forum, Lucerne Switzerland, Jul 2004
- NURETH-11, Avignon France, Oct 2005

Hydrogen Safety

The most significant hydrogen hazard associated with this project is:

Fire.

Hydrogen which has leaked into air is easily ignited by any hot or clean metal surface. This is a concern in and around the room temperature piping. Leaks in the high temperature portions of the process will burn of course, but in areas designed for high temperatures. In addition, leaks there cannot build up in concentration as they react in a diffusion flame sheet as quickly as the hydrogen can diffuse to oxygen. A potentially greater hazard is a cold leak that spreads to a large area before igniting, which could overpressure parts of the structure.

Hydrogen Safety

Our approach to deal with this hazard is:

- Outside cylinder storage & pressure relief
- Metal piping – leak test
- Point of use flow restriction orifices
- High capacity ventilation system
- Fusible link valve closure
- Hydrogen/Combustible gas sensor/alarm
- Power failure gas cutoff
- Sprinkler system

Questions?

- Contact: Rich Bechtold at 301-429-4566, richard.bechtold@qssgroupinc.com or Melissa Lott at 301-560-2214, Melott@qssgroupinc.com