



2005 Annual DOE Hydrogen Program Merit Review

Hydrogen Storage

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Basic Science Research Needs presented by George Thomas

ST1

¹ Laboratory Fellow, change of station assignment to DOE HQ



- Overview & Approach
 - Challenges & Targets
 - Basis for Targets & Current Status
 - Research Portfolio & RD&D Plan
- Status & Key Accomplishments
 - Technology Progress
 - Key Activities & Outputs
 - Program Planning/Coordination
- Future Plans
 - Budget
 - Upcoming Solicitation & RD&D Needs



Hydrogen Storage: Challenges & Targets



Challenge: How to store hydrogen on-board to meet performance (wt, vol, kinetics, etc.) , safety and cost requirements and enable > 300 mile range, without compromising passenger/cargo space.

Targets: Developed through



These
Are
System
Targets

Material
capacities
must be
higher!

| | 2010 | 2015 |
|---|--|--|
| System Gravimetric Capacity= Specific Energy (net) | 2.0 kWh/kg (7.2 MJ/kg) (6 wt%) | 3.0 kWh/kg (10.8 MJ/kg) (9 wt%) |
| System Volumetric Capacity=Energy Density (net) | 1.5 kWh/L (5.4 MJ/L) (0.045 kg/L) | 2.7 kWh/L (9.7 MJ/L) (0.081 kg/L) |
| Storage system cost | \$4/kWh | \$2/kWh |



Focus is on capacity: but many other requirements...



Primary 2005 targets achieved with high P/LH₂

2007

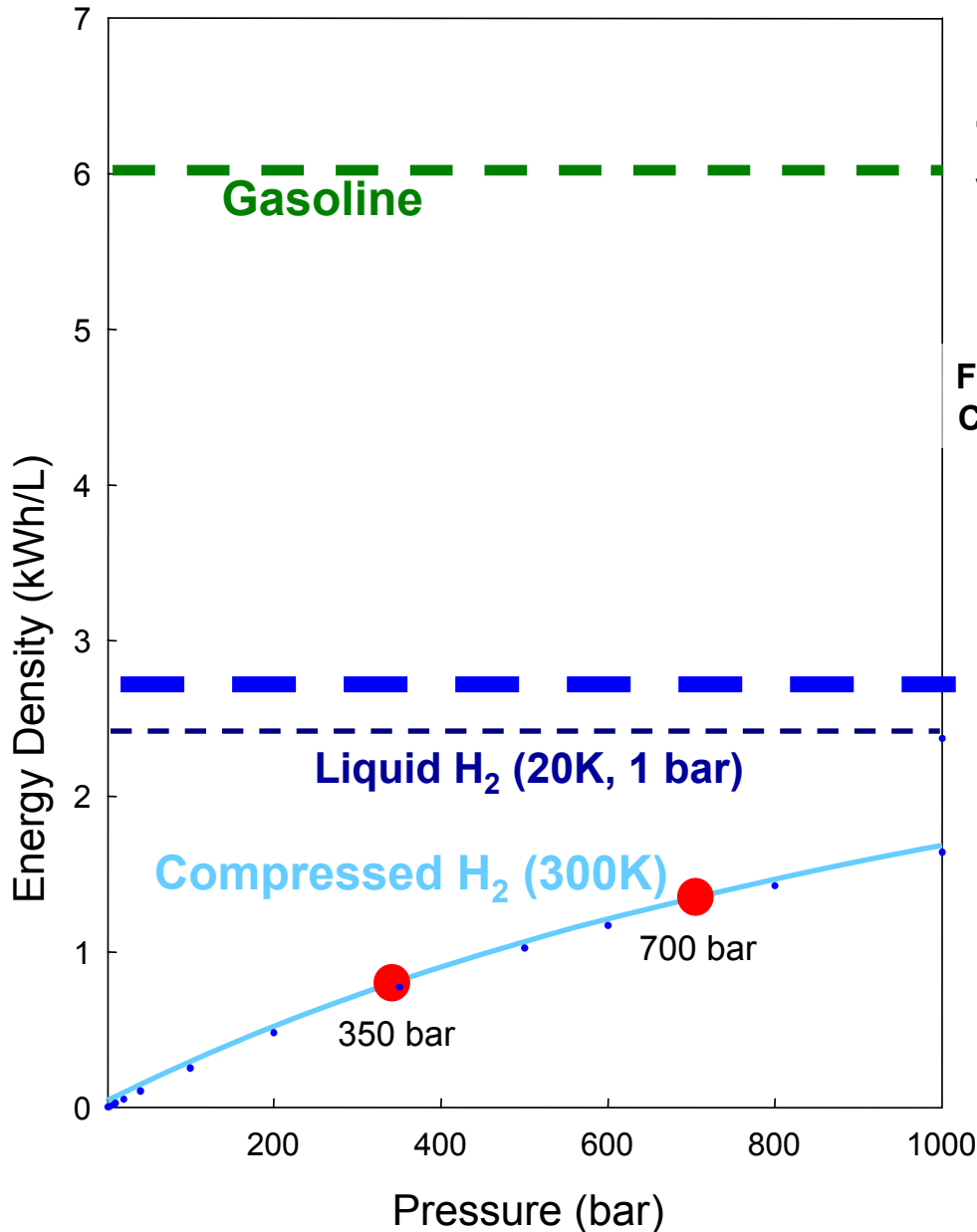
For materials-based systems

| Parameter | Units | 2005(7) | 2010 | 2015 |
|--|--------------------------------------|--|------------------------|------------------------|
| Specific energy net | kWh/kg | 1.5 | 2.0 | 3.0 |
| Energy density net | kWh/L | 1.2 | 1.5 | 2.7 |
| Storage system cost | \$/kWh | 6 | 4 | 2 |
| Cycle life (25-100%) | cycles | 500 | 1000 | 1500 |
| Cycle life variation | % of mean (min) @ % confidence | N/A | 90/90 | 99/90 |
| Max delivery temp. | °C | 85 | 85 | 85 |
| Minimum delivery pressure of H ₂ from tank, FC=fuel cell, I=ICE | atm (abs) | 8 FC 10 ICE | 4 FC 35 ICE | 3 FC 35 ICE |
| Start time to full flow @ 20 °C | sec | 4 | 4 | .5 |
| System fill time (for 5 kg) | min | 10 | 3 | 2.5 |
| Loss of useable hydrogen | (g/h)/kg H₂ stored | 1 | 0.1 | 0.05 |
| Permeation and leakage | Sc/h | Federal enclosed-area safety standard | | |
| Toxicity | | Meets or exceeds applicable standards | | |
| Safety | | Meets or exceeds applicable standards | | |

Current Focus



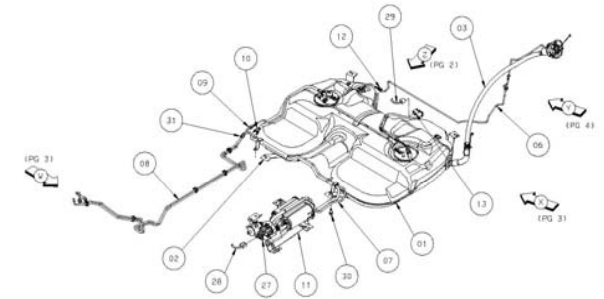
Energy Density is Critical



Today's Vehicles

Fuel Cell Efficiency, Conformable Tanks

2015 Target

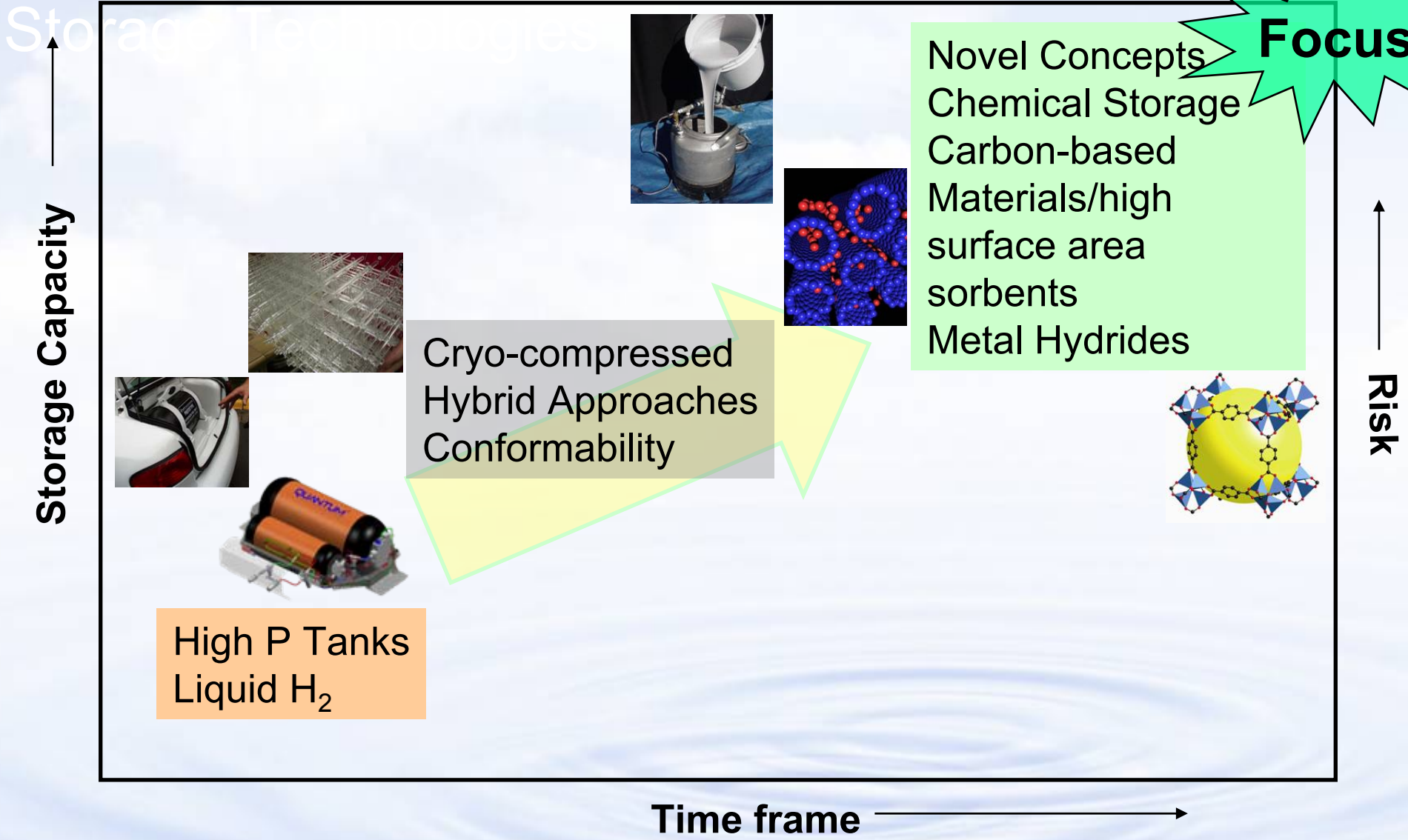


Today's gasoline "tank": fuel tank, fuel filler tubes, gas cap, hoses, fuel lines, fuel pump, fuel filter, carbon vapor canister, leak detection device, purge control solenoid, rollover check valve, tank hanger straps, clips, and other small fasteners

For Hydrogen Systems: Also include insulation, sensors, regulators, first charge, any byproducts/reactants, etc.



Research Areas



Portfolio stresses longer-term solutions but continues some R&D on viable options for the transition phase

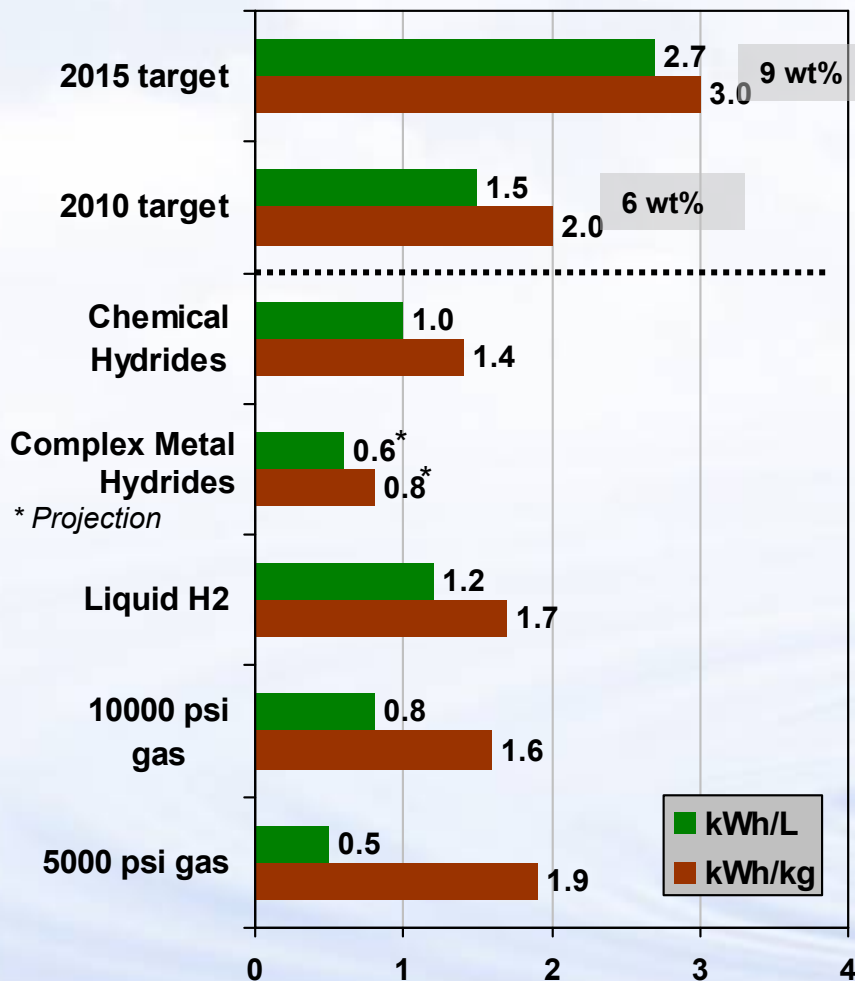


Status Relative to Targets

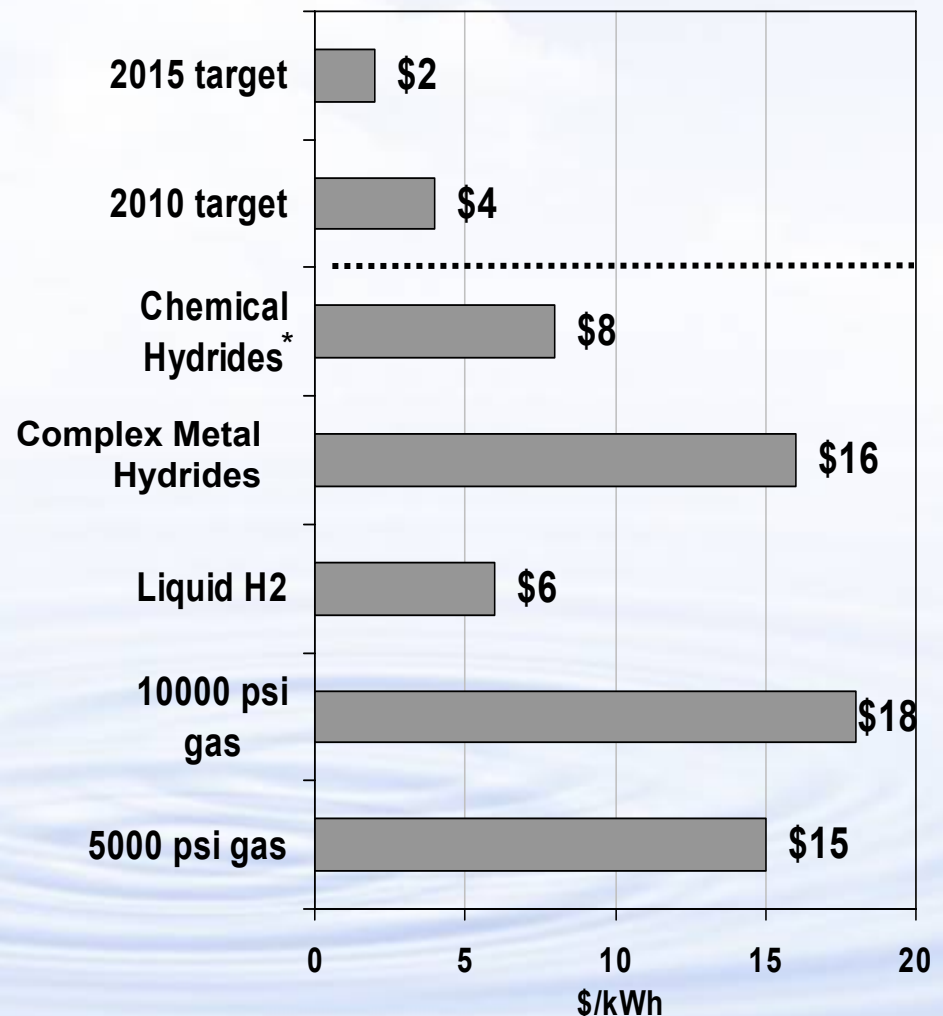


No current hydrogen storage technology meets the targets.

Volumetric & Gravimetric Energy Capacity



System Cost, \$/kWh

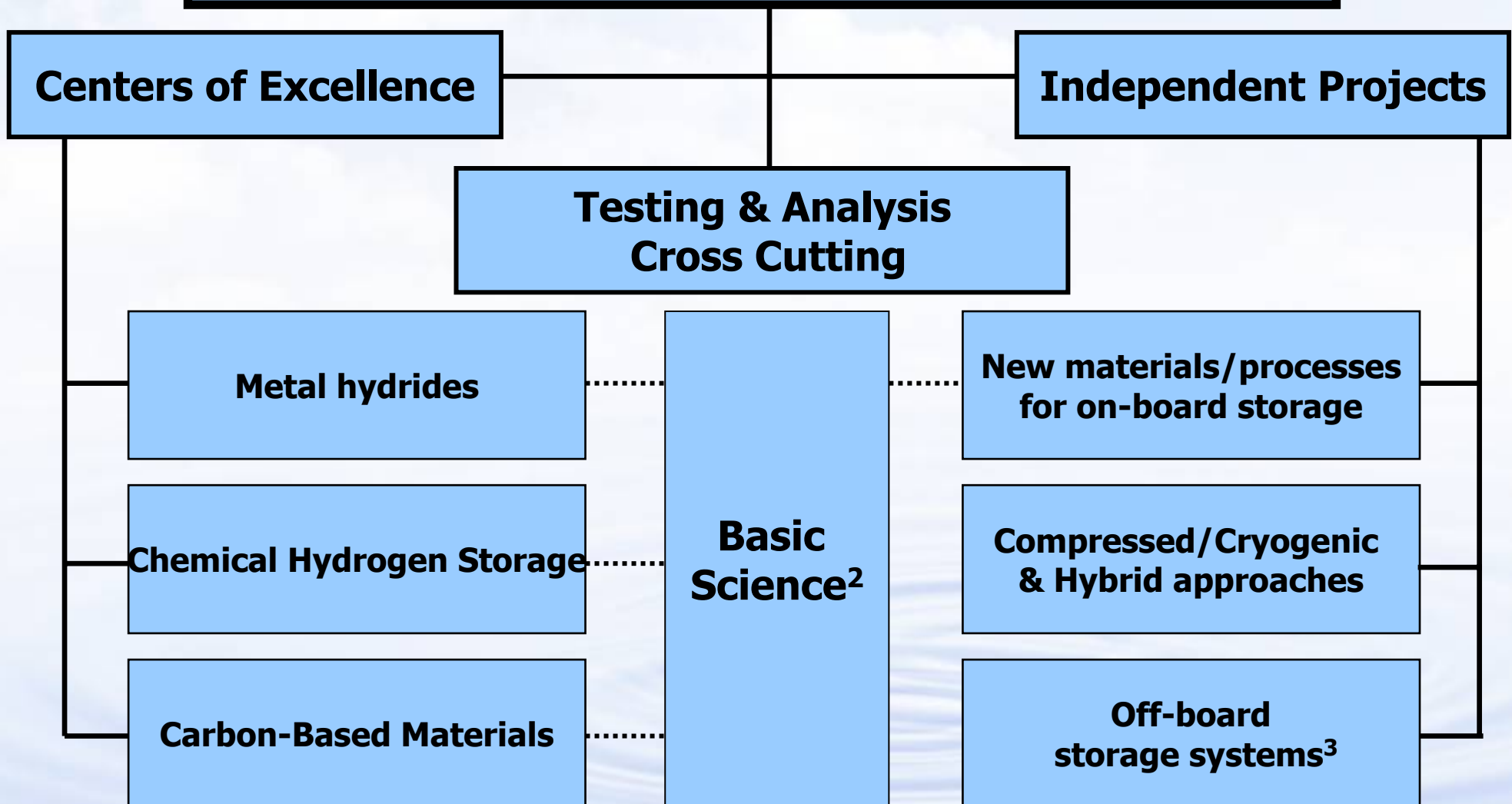


Estimates from developers- to be continuously updated

* Regeneration costs excluded



National Hydrogen Storage Project¹



1. Coordinated by DOE Energy Efficiency and Renewable Energy, Office of Hydrogen, Fuel Cells and Infrastructure Technologies
2. Basic science for hydrogen storage conducted through DOE Office of Science, Basic Energy Sciences
3. Coordinated with Delivery program element



Hydrogen Storage “Grand Challenge” Partners



Centers of Excellence

Metal Hydride Center

National Laboratory:
Sandia-Livermore

Industrial partners:
General Electric
HRL Laboratories
Intematix Corp.

Universities:
CalTech
Stanford
Pitt/Carnegie
Mellon
Hawaii
Illinois
Nevada-Reno
Utah

Federal Lab Partners:
Brookhaven
JPL
NIST
Oak Ridge
Savannah River

Carbon Materials Center

National Laboratory:
NREL

Industrial partners:
Air Products &
Chemicals

Universities:
CalTech
Duke
Penn State
Rice
Michigan
North Carolina
Pennsylvania

Federal Lab Partners:
Lawrence Livermore
NIST
Oak Ridge

Chemical Hydrogen Center

National Laboratories:
Los Alamos
Pacific Northwest

Industrial partners:
Intematix Corp.
Millennium Cell
Rohm & Haas
US Borax

Universities:
Northern Arizona
Penn State
Alabama
California-Davis
UCLA
Pennsylvania
Washington

Independent Projects

New Materials & Concepts

Alfred University
Carnegie Institute of Washington
Cleveland State University
Michigan Technological University
TOFTEC
UC-Berkeley
UC-Santa Barbara
University of Connecticut
University of Michigan
University of Missouri

High-Capacity Hydrides

UTRC
UOP
Savannah River NL

Carbon-based Materials

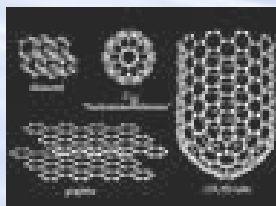
State University of New York
Gas Technology Institute
UPenn & Drexel Univ.

Chemical Hydrogen Storage

Air Products & Chemicals
RTI
Millennium Cell
Safe Hydrogen LLC

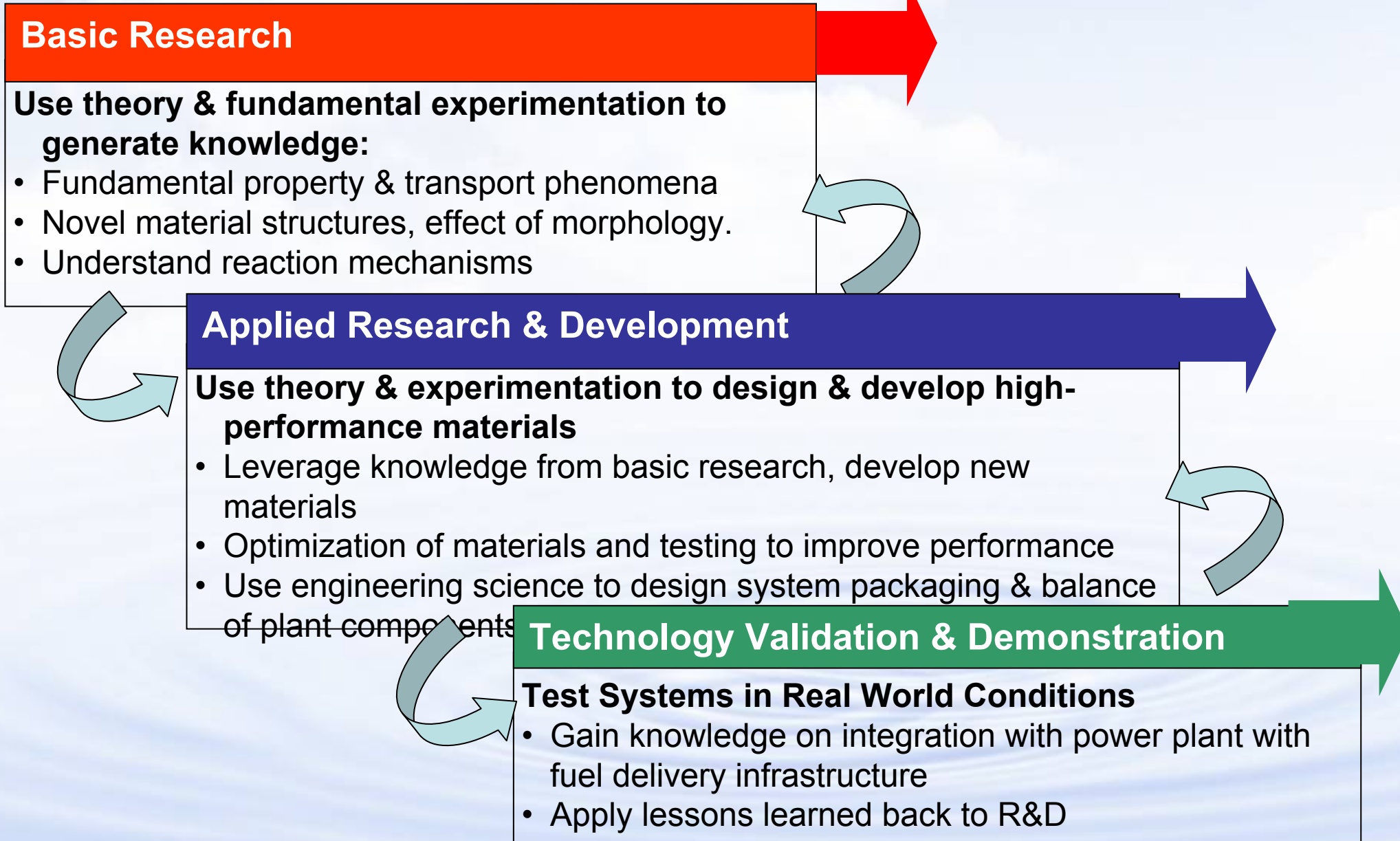
OffBoard, Tanks, Analysis & Testing

Gas Technology Institute
Lawrence Livermore
Quantum
Argonne Nat'l Lab & TIAX LLC
SwRI





Continuum of Knowledge Transfer Across Stages of Development





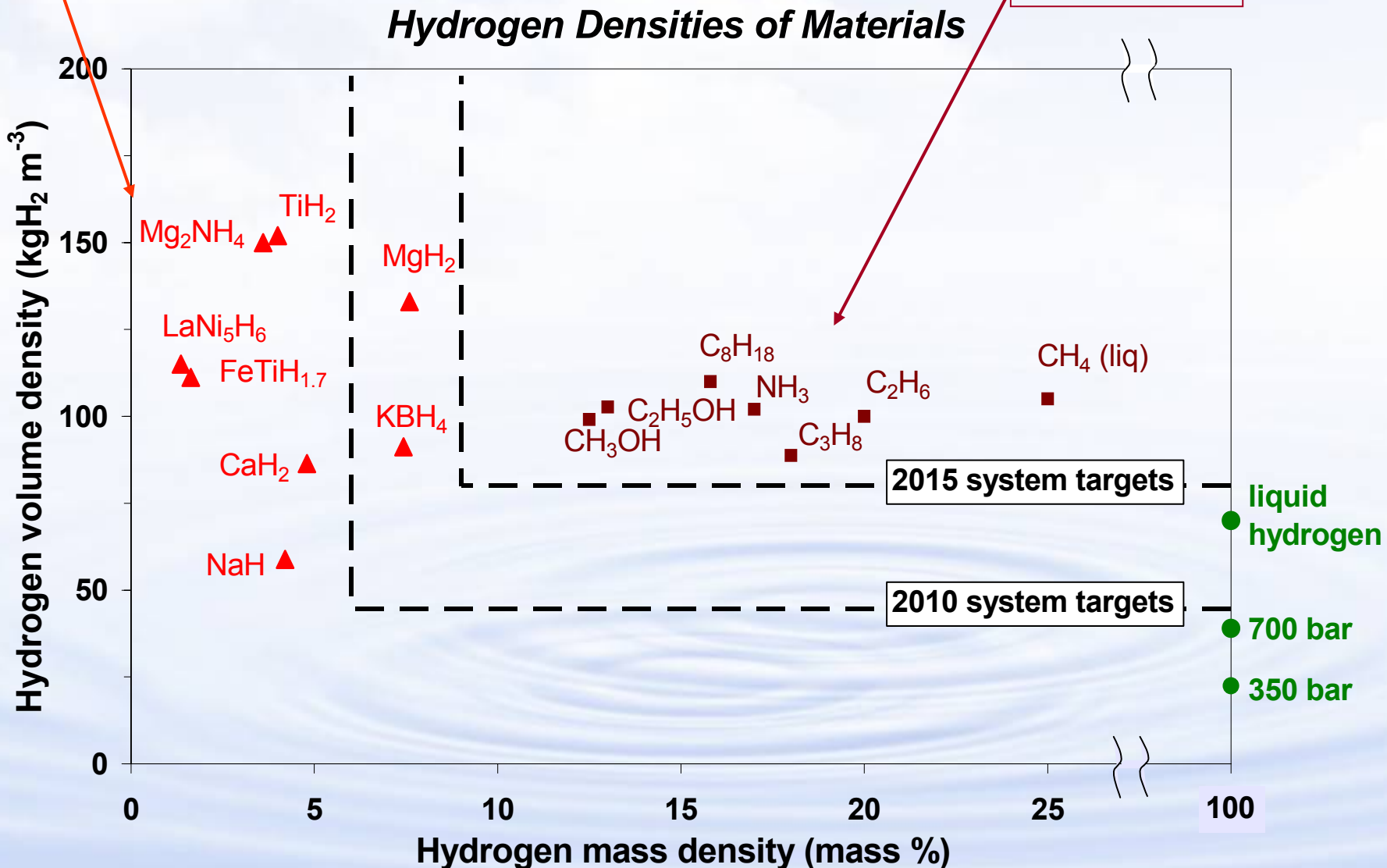
Basic Science for Hydrogen Storage



Basic material properties are a key issue in hydrogen storage.

intermetallic hydrides
too heavy

liquid fuels

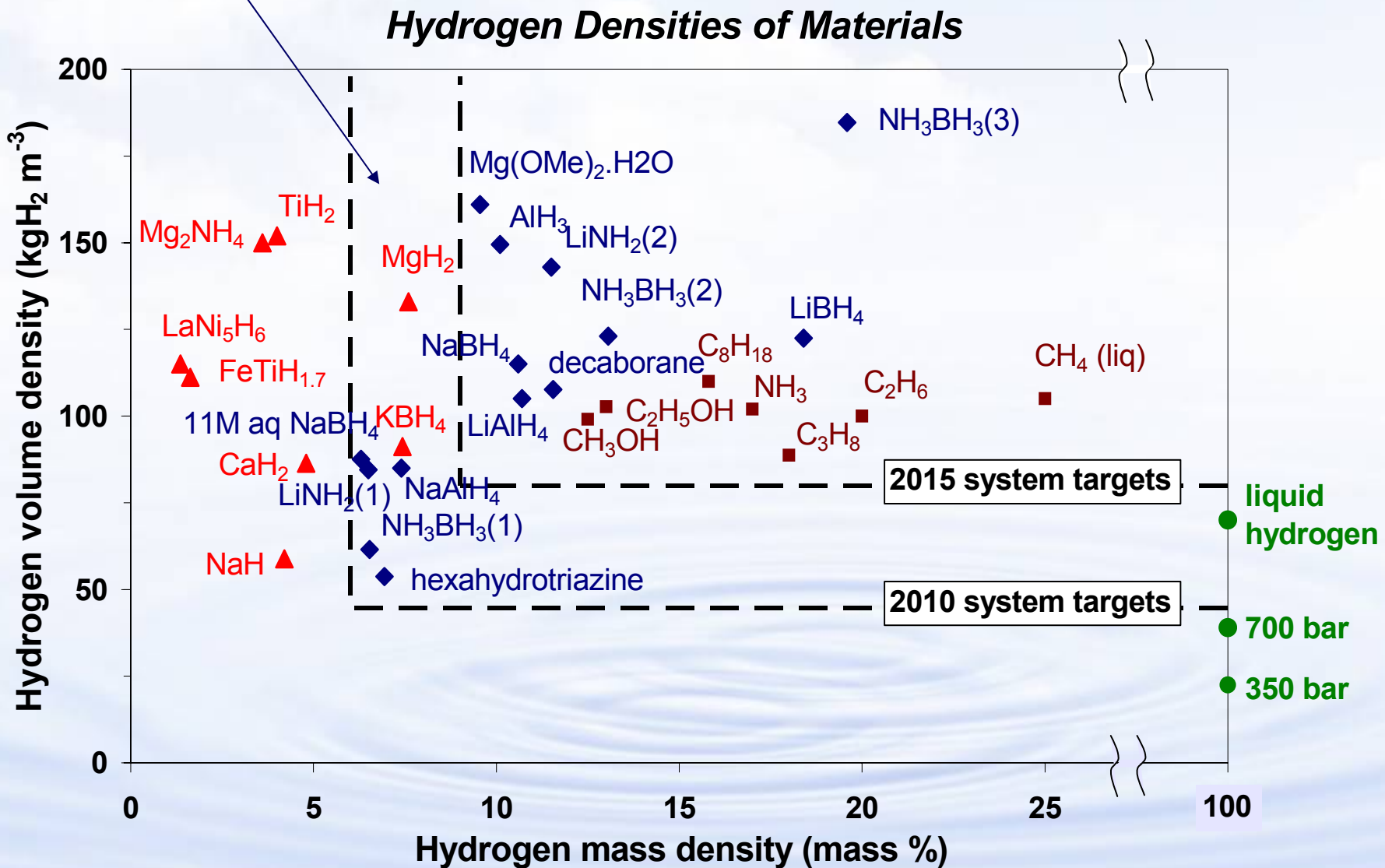




Program focus is on high energy density materials.



Some of the materials under study in CoE's

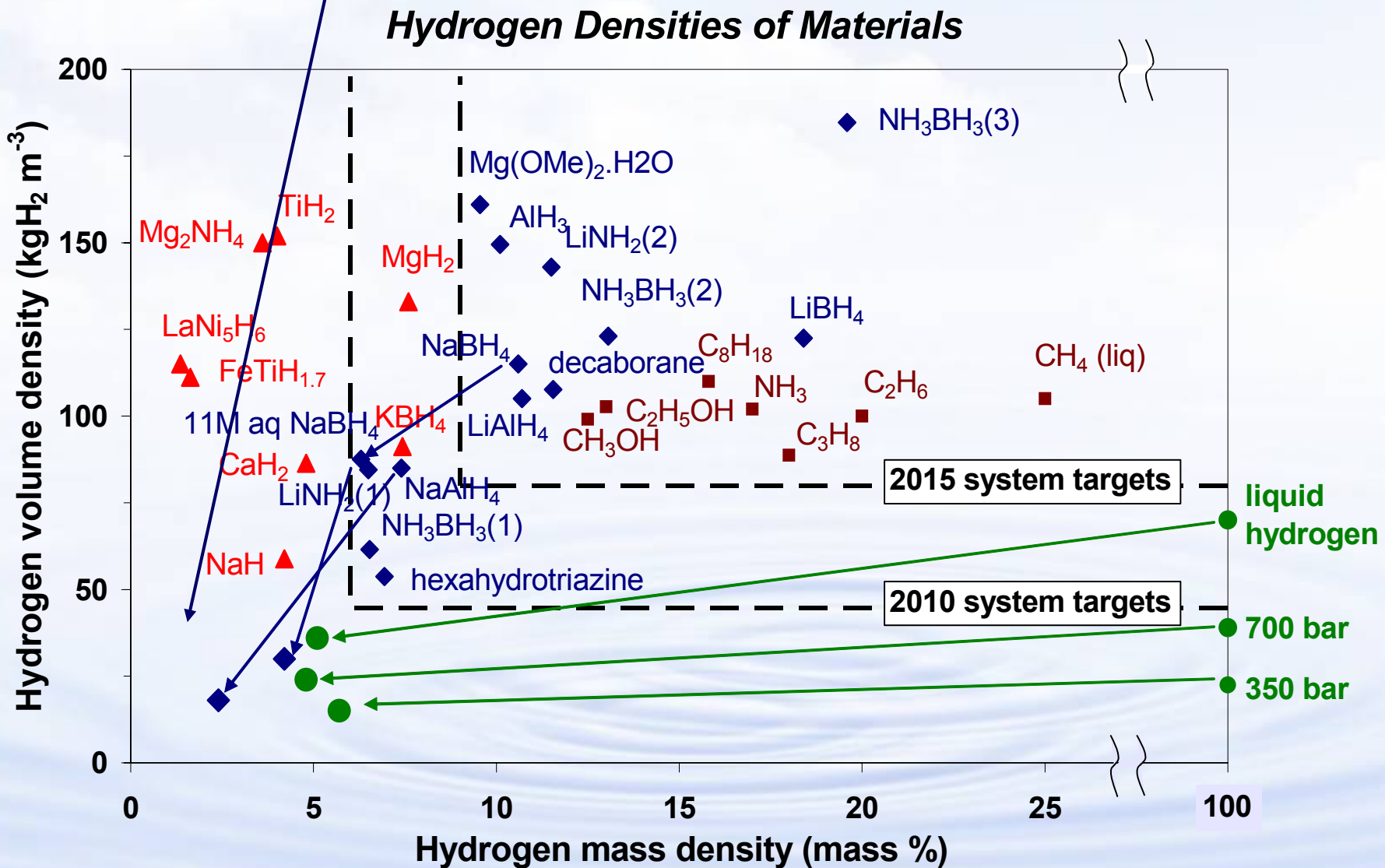




Storage system adds weight and volume



No current material meets system requirements





Need to determine & tune material properties



Other material property issues include thermodynamic properties.

For reversible systems, equilibrium between gas and solid given by:

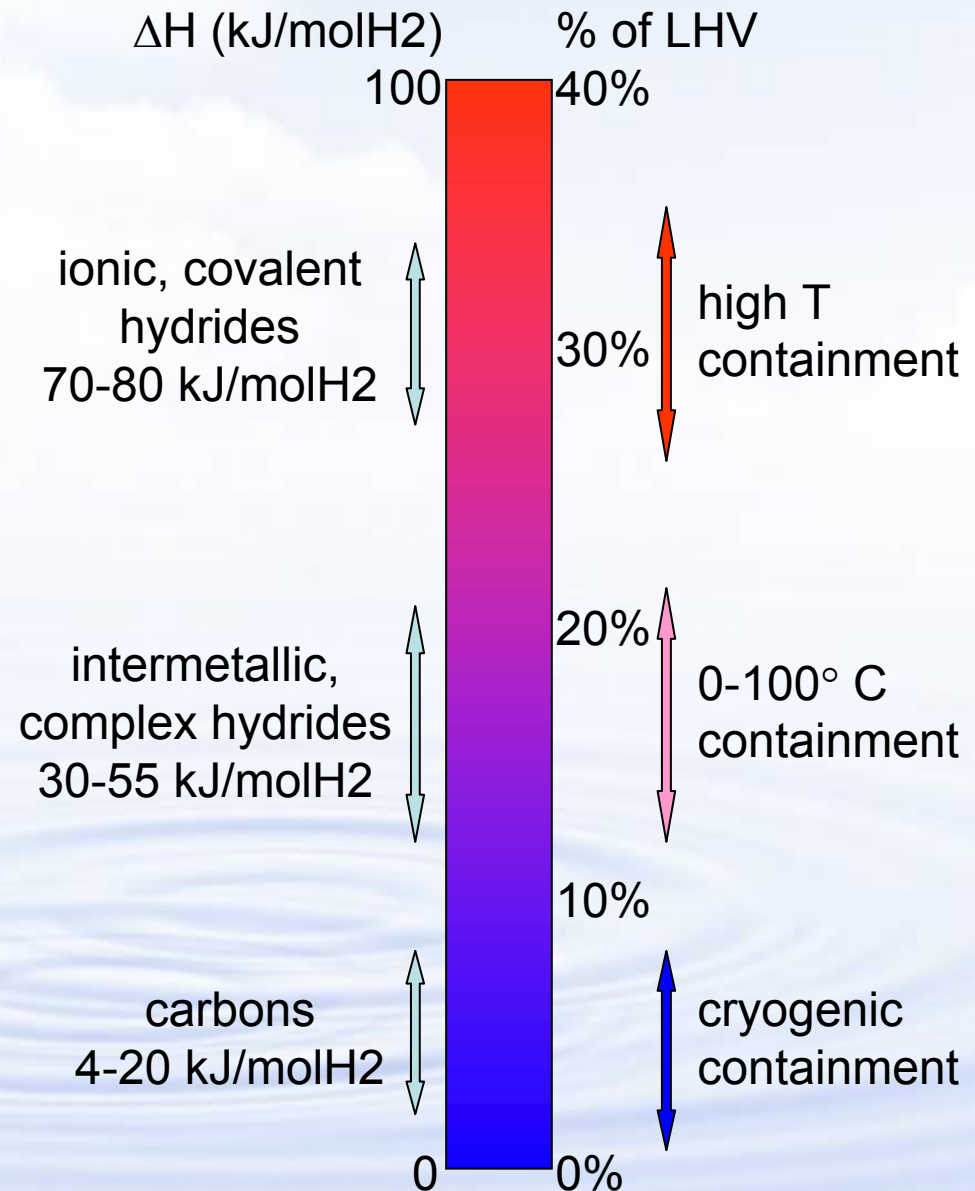
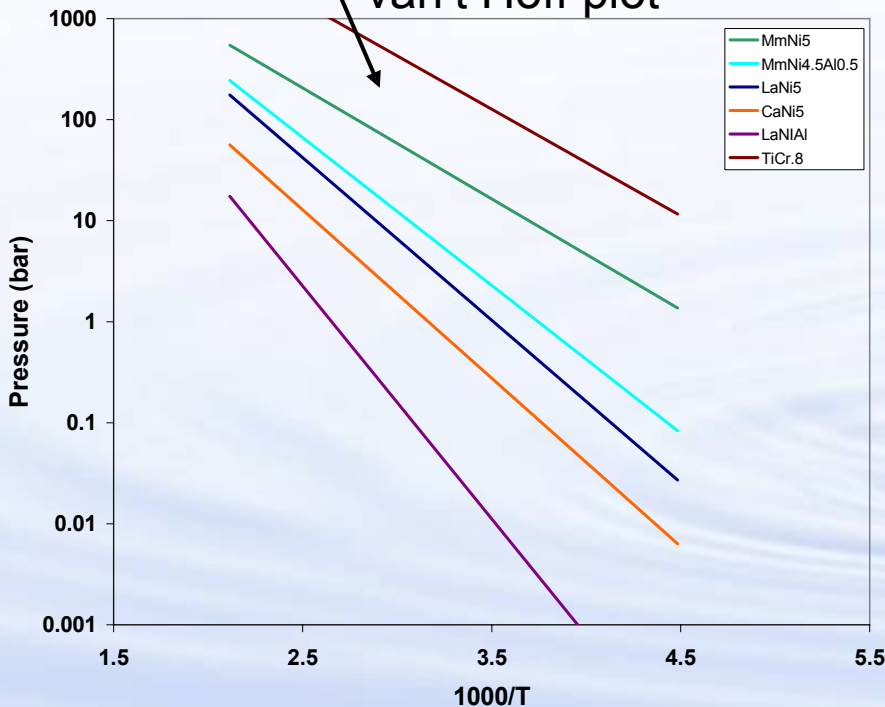
$$P = \exp(-\Delta H/RT + \Delta S/R)$$

$$\text{or } \ln P = -\Delta H/RT + \ln P_{\text{Tinf}}$$

ΔH =enthalpy (kJ/mol H₂),

$$d(\ln P)/d(1/T) = \Delta H/R$$

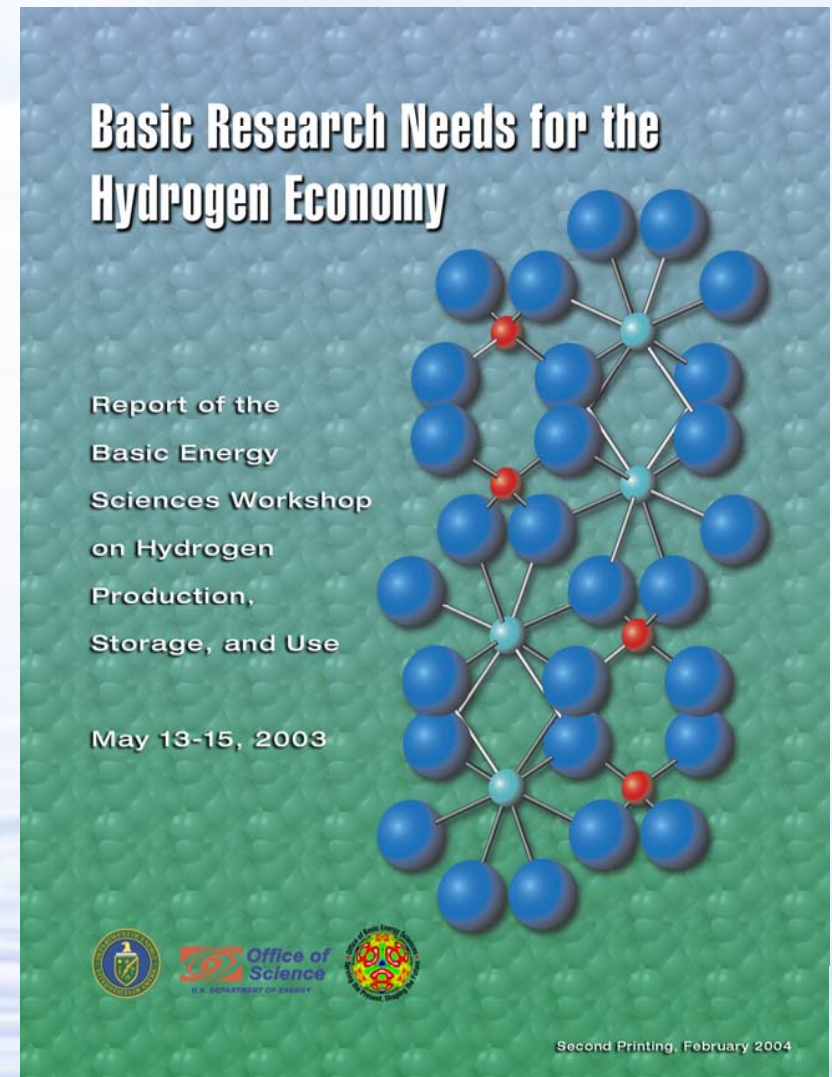
van't Hoff plot





BES workshop was held to identify basic research needs.

- Understanding the fundamental factors governing material behavior
- Applying these principles to modify material performance
- Identifying new materials and new classes of materials
- Designing of new materials at the nanoscale
- Theory, modeling and simulation of materials and molecular processes





Workshop laid foundation for 2005 BES solicitation.

- 227 full proposals were received
- Five technical focus areas:
 - Novel materials for hydrogen storage (50 proposals)
 - Membranes for separation, purification and ion transport (52 proposals)
 - Design of catalysts at the nanoscale (56 proposals)
 - Solar hydrogen production (49 proposals)
 - Bio-inspired materials and processes (20 proposals)

Stay tuned for further announcements!



Tanks- Accomplishments & Status

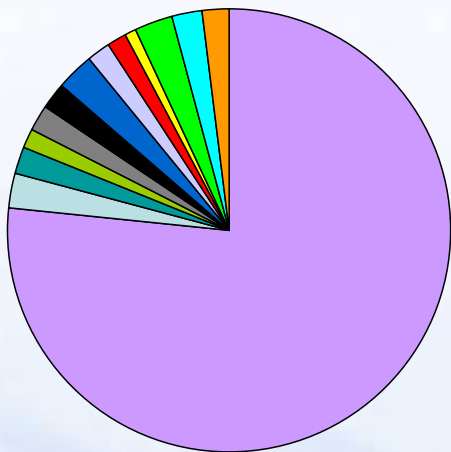


Tanks: Performance close to targets - Next steps are cost and hybrid approaches.

| | Specific Energy | Energy Density | Cost |
|----------------------------|-------------------------|------------------------|--------------------|
| 2015 Targets (2010) | 3.0 kwh/kg (2.0) | 2.7 kwh/L (1.5) | \$2/kWh (4) |
| 5,000 psi System | 1.9 kwh/kg | 0.5 kwh/L | \$15/kWh |
| 10,000 psi System | 1.6 kwh/kg | 0.8 kwh/L | \$18/kWh |

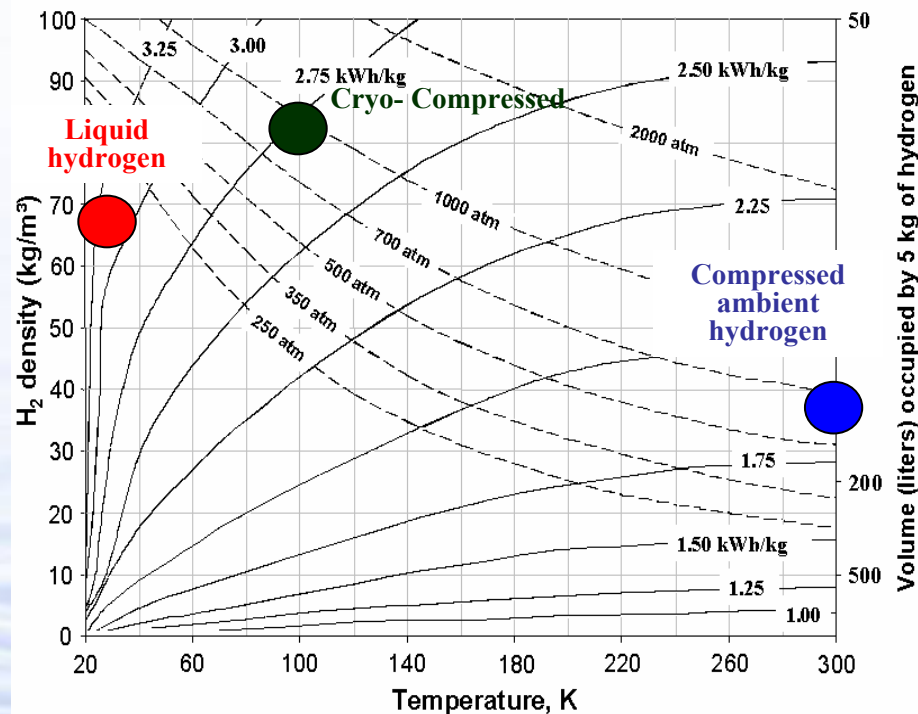
Status*
(Quantum)

Primary driver is material cost
40 - 80% is carbon fiber cost



- Carbon Fiber
- Glass Fiber
- Epoxy
- Curatives
- Liner Polymer
- Foam Dome
- Front Boss
- Aft Boss
- 1-1/8 Adapter
- Seals
- Valve
- PRD
- Miscellaneous

Future focus: Cost reduction,
advanced concepts, conformability



Density improvements possible with cryo-compressed H₂

Aceves, et al, LLNL

* 5 kg storage using one tank; volume of 500,000 tanks/year



Storage Accomplishments- Metal Hydrides

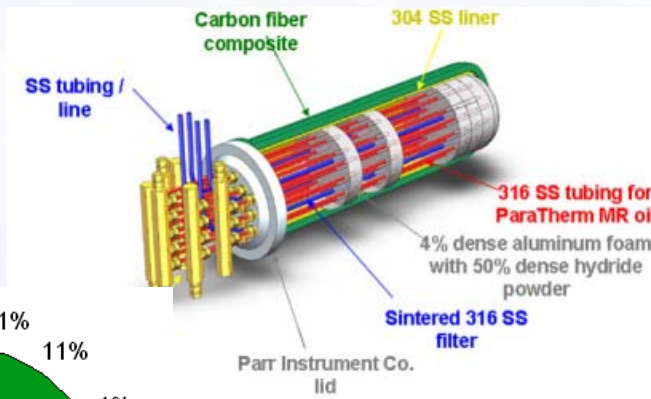


System Engineering:

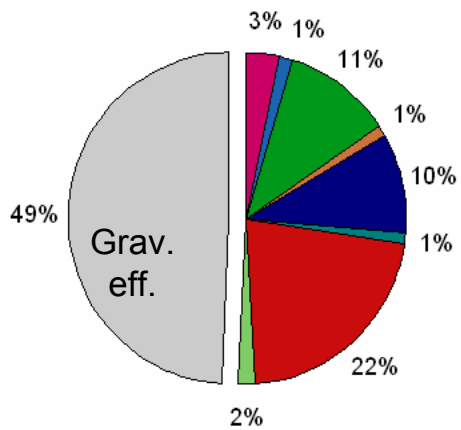
- Preliminary 1-kg hydrogen system prototype developed (*Anton, et al, UTRC*)
- With composite vessel, ~50% of system is balance of plant

Materials Development:

- Mg modified Li-amides demonstrate 5 wt% materials-based capacity, reversible with potential up to 10 wt% (*Luo, Wang, et al, SNL*)
- Absorption demonstrated down to 180C with Mg substitution
- Over 100 cycles demonstrated for Li/Mg amide system (*Luo, Gross, et al, SNL*)



UTRC



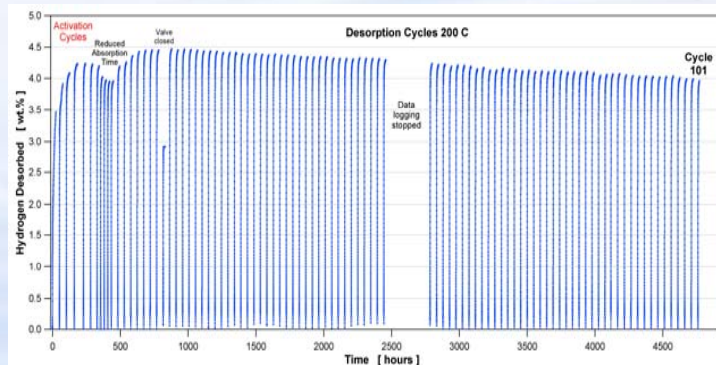
660 Wh / kg 530 Wh / L



H wt.% (material only)

SNL

← 0.005 wt% loss/cycle



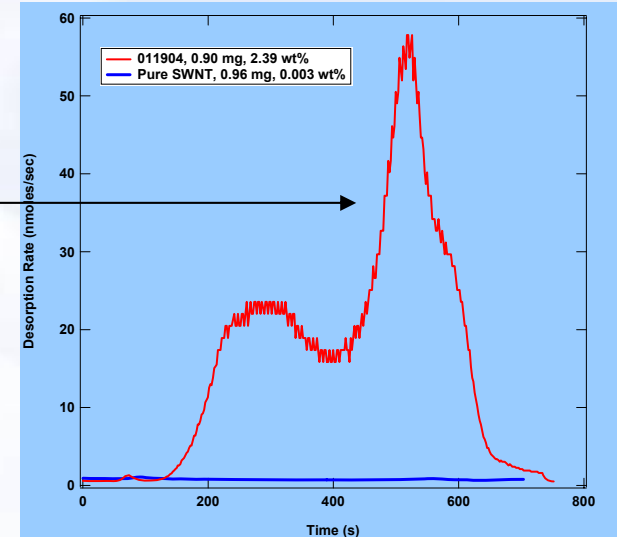
SNL



Storage Accomplishments- Carbon Materials

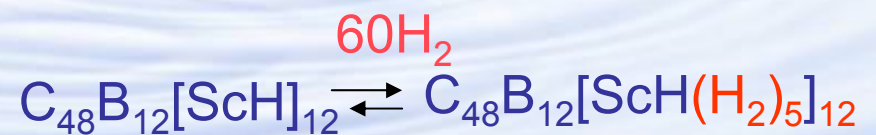
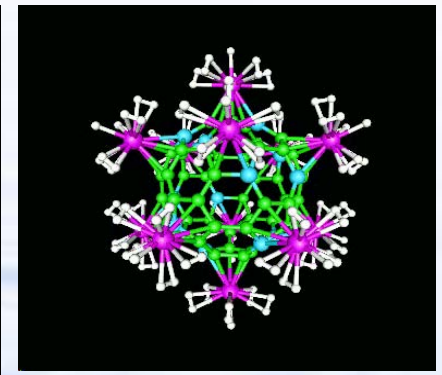
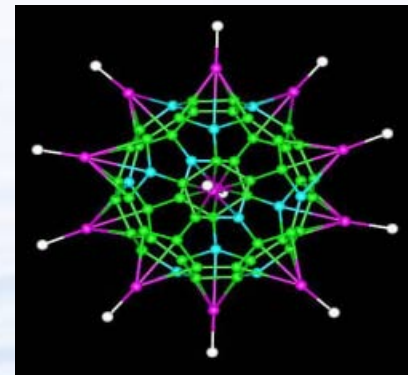
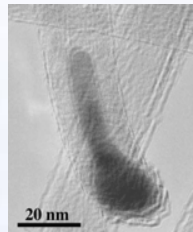
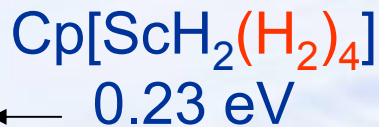
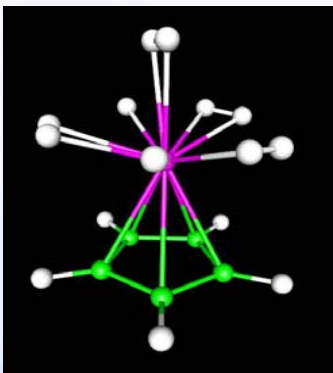


- Doped single-wall nanotubes (SWNTs) synthesized and capacity measured to be ~2.5 - 3 wt.% hydrogen storage
- Binding energies calculated and optimum compounds theoretically predicted for potential storage materials



H₂ Desorption

Iron in nanotube



Potential for 8.8 wt%

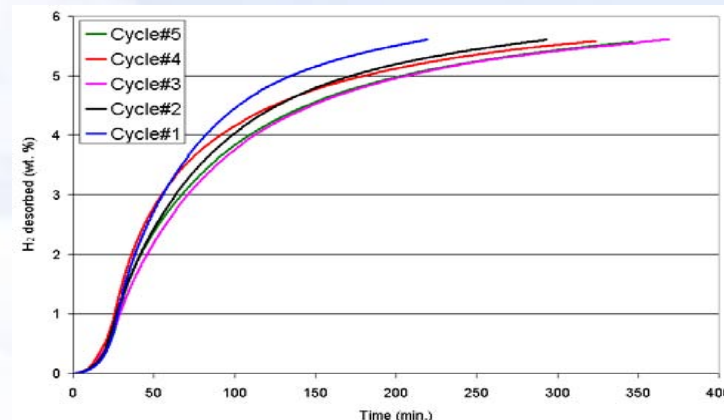
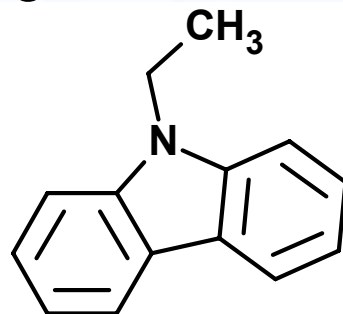


Storage Accomplishments- Chemical Hydrides



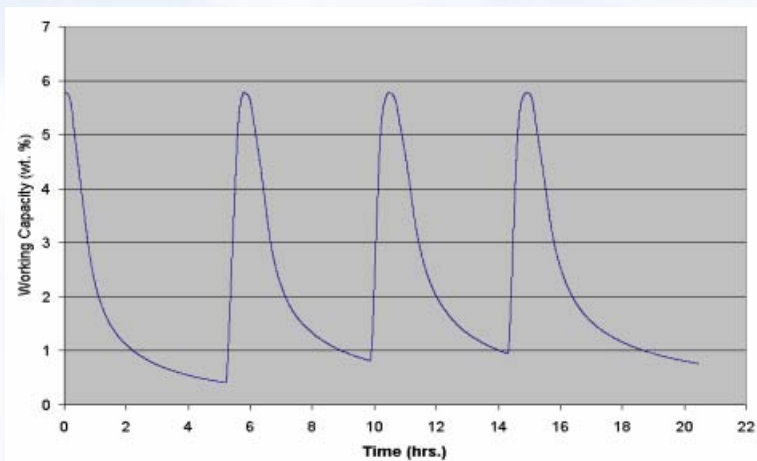
- Identified chemical hydride with 5.5 wt% materials-based H₂ storage capacity

N-ethylcarbazole

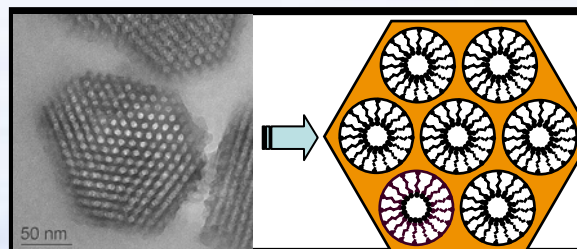


Cooper, Pez, et al, Air Products

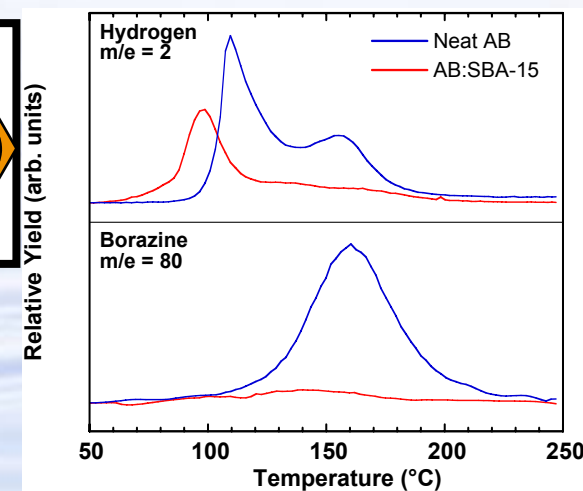
- Demonstrated several cycles
- 40 catalysts screened for dehydrogenation



- Mesoporous scaffolds internally coated with ammonia borane show hydrogen release at < 100 C and reduce borazine formation



Autrey, et al, PNNL

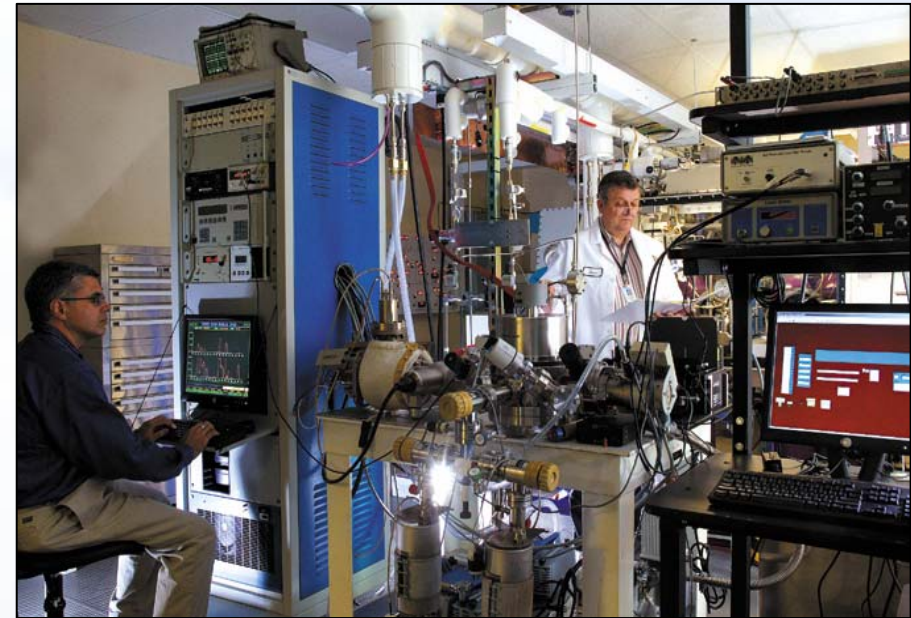




Independent Testing and Analysis

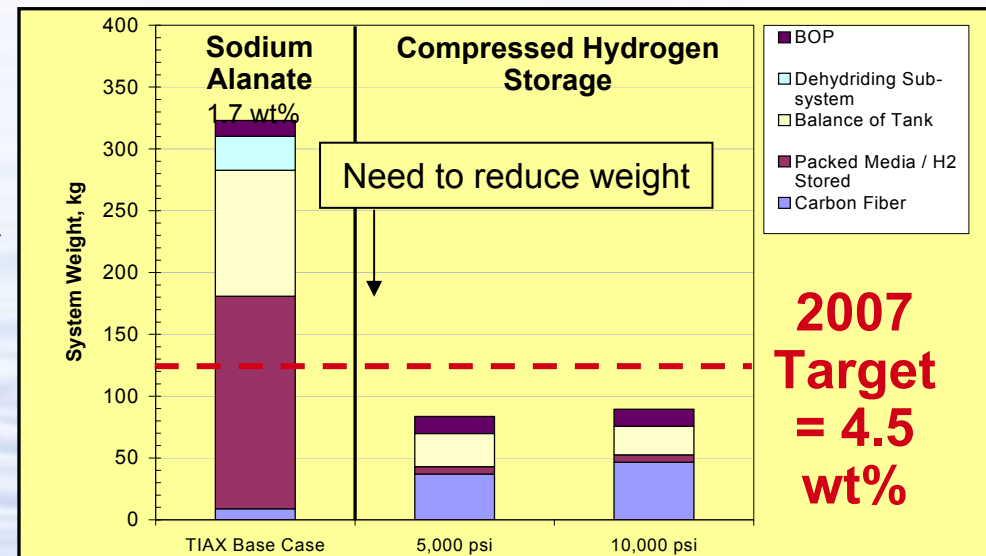


- Standardized Test Facility
 - Independent laboratory, SwRI
 - Construction complete
 - Test protocols developed
 - Website planned
 - Validation underway (double-blind testing)



SwRI

- Storage Systems Analyses
 - Storage Systems Analysis Working Group formed (March, '05)
 - Need for independent analyses: Performance, cost, life cycle energy & environmental impact, toxicity, safety
 - Breakdown of system into components (TIAX, ANL, Quantum, UTRC, Safe H2, Millennium Cell)



TIAX



Key Activities and Outputs (2005-2006)



IPHE Hydrogen Storage Technology Conference June 20-22, 2005, Barga, Lucca, Italy



- Forum for Technical experts in hydrogen storage worldwide
- Track global progress- inventory of projects & expertise
- Objectives: Promote leveraging of global R&D
 - Identify and reduce duplication
 - Create and strengthen partnerships
 - Initiate process, identifying clear guidelines for IPHE projects
 - Gather feedback from global technical community on issues to be addressed by IPHE
 - Accelerate collaborative research projects
 - Recommendations to IPHE Committees



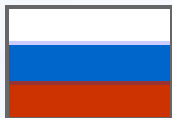
Italy



U.S.
DOE



European
Commission



Russian
Federation

Key Organizers:

P. Garibaldi, S. Malysenko,
M. Steen, G. Sandrock, M.
Conte, C. Filiou, S. Satyapal

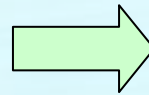
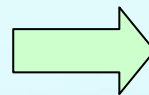
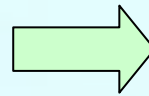
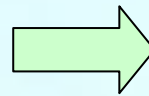
Limited Attendance: www.iphe.net or <http://www.engconfintl.org/5ar.html>



Key Activities and Outputs (2005-2006)



- Water Availability Model
- Position Paper on Ammonia
- Review Targets & Update
- Purity Specification Guidelines
 - Input to storage systems
 - Output from storage systems

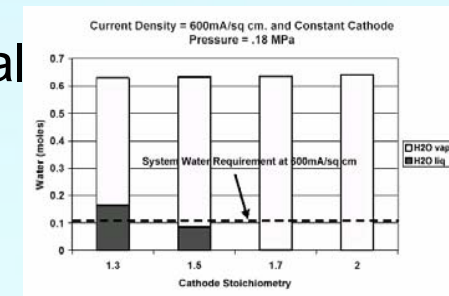


Help guide chemical hydrogen storage researchers

Draft for public comment, June 2005

June-August 2005

Tolerance of storage materials to various contaminants
Coordination with Fuel Cell, Production & Delivery Program Elements

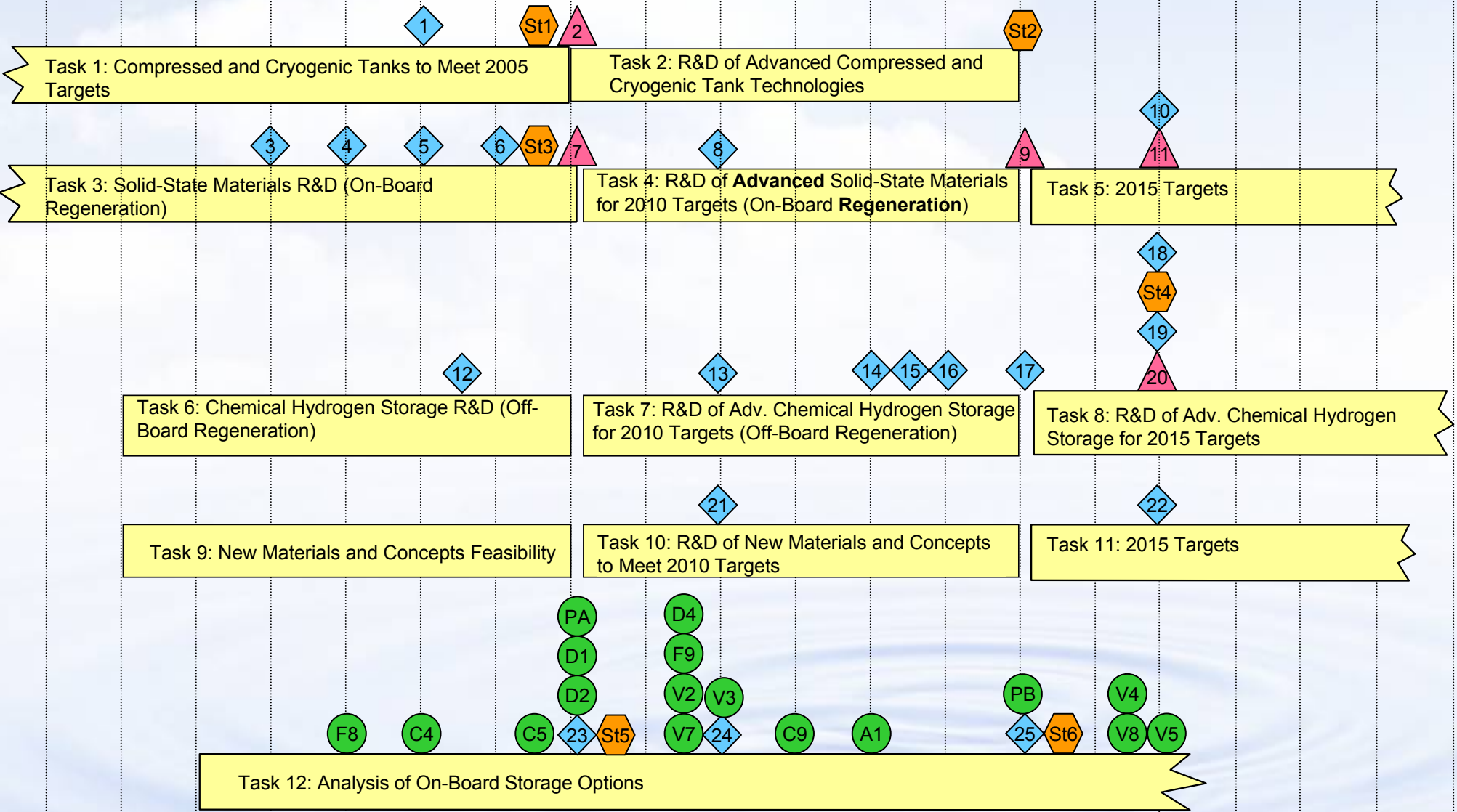




RD&D Plan- Tasks, Milestones



FY 2003 FY 2004 FY 2005 FY 2006 FY 2007 FY 2008 FY 2009 FY 2010 FY 2011 FY 2012



◆ Milestone
 ● Input
 ⬡ Output
 ▲ Go/No-go

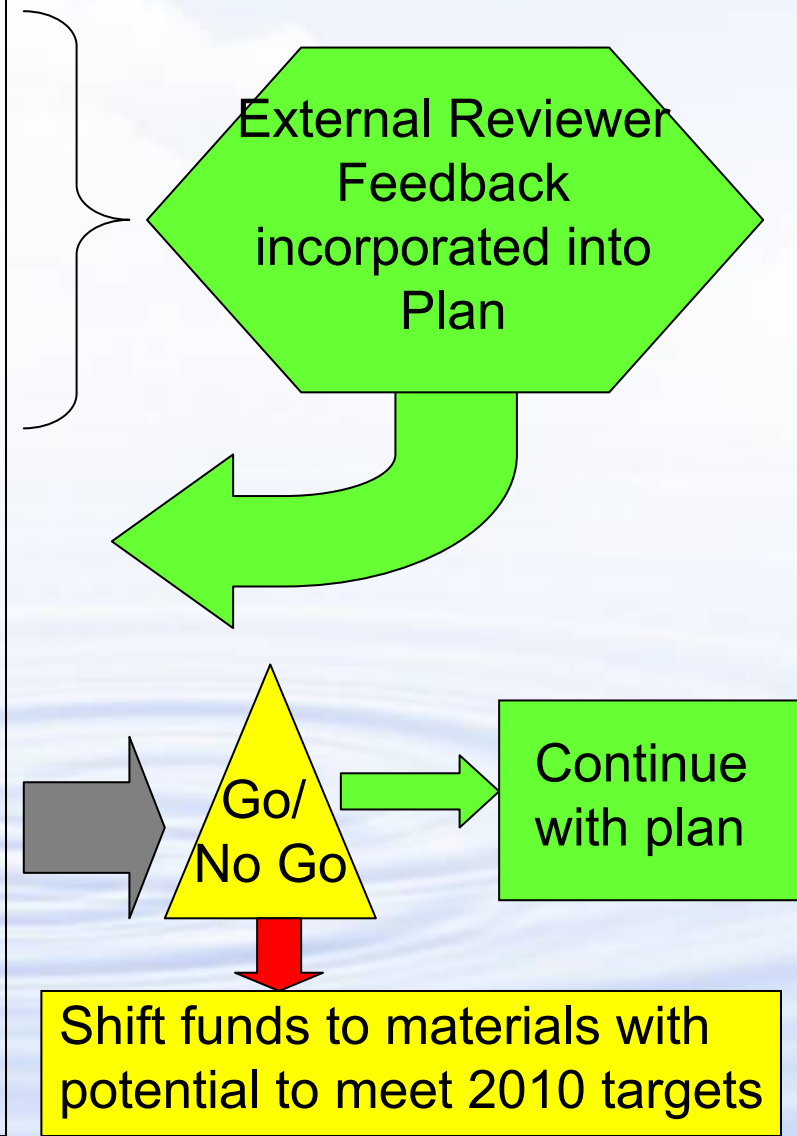


Go/No-Go Decision on Carbon Nanotubes (4Q FY 06)



Decision Planning & Implementation

- **Measurement Technique Validation**
 - ✓ Verify storage capacity measurement with 2 different techniques
 - ✓ Verify accuracy with standards
 - ✓ On-site peer review to inspect & verify measurement techniques (Jan. '04)
- **Interim Milestone:**
 - ✓ Demonstrate 3 wt % (materials-based storage capacity - Aug. '04)
- **Interim Milestone:**
 - o Demonstrate 4 wt % (materials-based storage capacity)
- **Milestone:**
 - o Reproducibly demonstrate 4 wt% in external laboratory (4Q FY 2005)
- **Go/no go Point:**
 - o Reproducibly demonstrate 6 wt% (materials-based storage capacity) in external laboratory (4Q FY 2006)

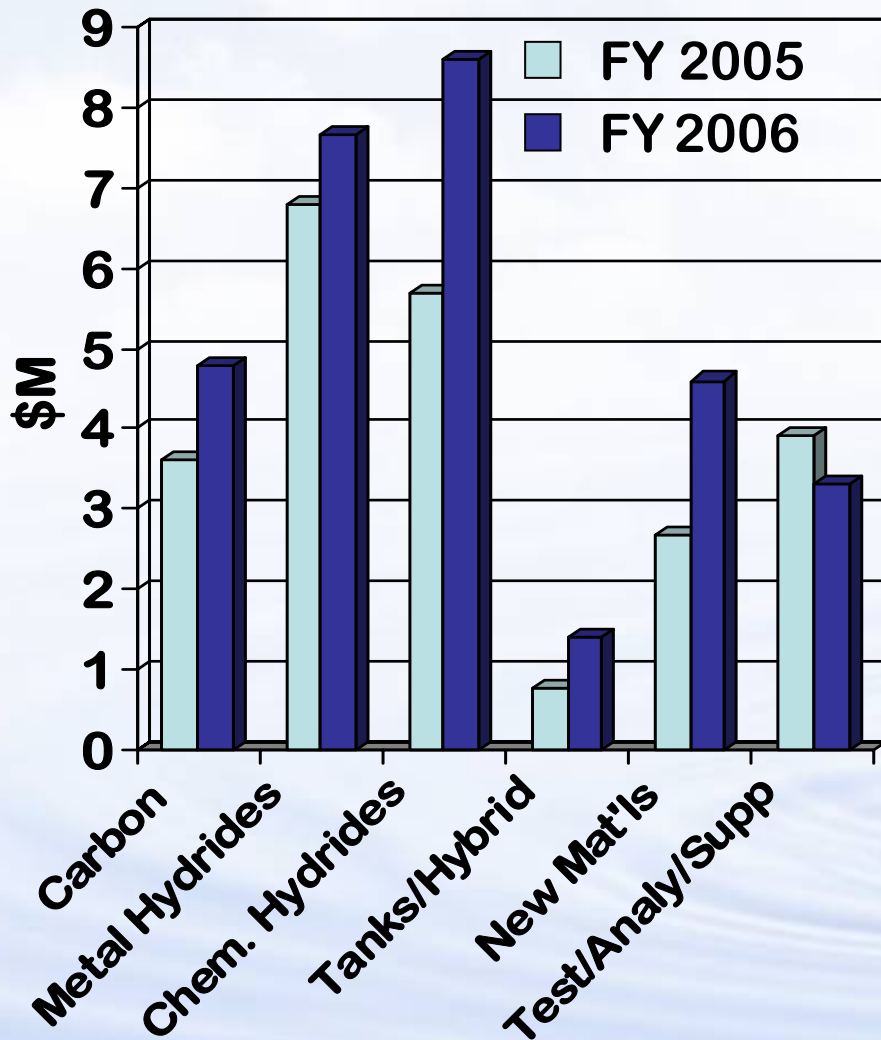




Budget: FY 2005 vs. FY 2006 Request



FY2006 Budget Request = \$29.9M
FY2005 Appropriation = \$24.4M



- **Emphasis:**
 - Continue Centers of Excellence and new materials projects to focus on 2010 hydrogen storage goals of 2.0 kWh/kg, 1.5 kWh/liter, \$4/kWh
 - Independent Testing & Analysis

- **Budget Distribution:**

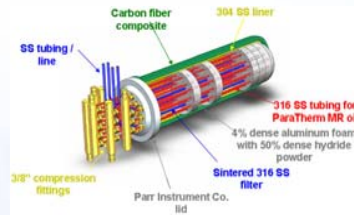
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|--------------------------------|----------------|
| Centers of Excellence | \$16.7M |
| Independent Projects & Support | <u>\$13.2M</u> |
| Total | \$29.9M |



Key Milestones



Complete construction of materials test facility (4Q, 2004)



Complete prototype complex metal hydride system (2Q, 2006)

Go/no go based on 6 wt% storage capacity on carbon nanotubes (4Q, 2006)

Down-select reversible metal hydrides (4Q, 2007)

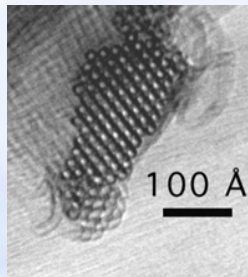
2005

2006

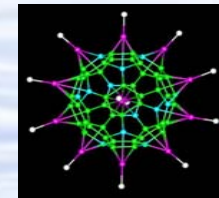
2007

2008

Demonstrate 4 wt% storage capacity on carbon nanotubes (4Q, 2005)



Down-select regeneration processes for chemical hydrides (4Q, 2007)



Down-select new materials / concepts (4Q, 2006)



Hydrogen Storage – Future Plans



Centers of Excellence
Kick-off Meetings

IPHE Hydrogen Storage Conference
Lucca, Italy

Storage Systems Analysis WG
Fuel Cell Seminar, Palm Springs

Announce new Solicitation Plans
("Grand Challenge" Solicitation*
cont'd)

Release Solicitation*

Full Proposals Due

2005

2006

Jan Mar Jun Sep Nov Jan Mar Jun Oct Dec

Storage Systems Analysis Working Group Launched

Annual Program Review

Testing Workshop

Complete Safety Plans Storage Projects

Preproposals Due

Annual Program Review

Basic Science Theory/Modeling Workshop

*Subject to appropriations



For More Information, Contact:

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- **Carole J. Read:** *Metal Hydrides, Metal Hydride Center of Excellence*
 - Phone 202-586-3152; carole.read@ee.doe.gov
- **John Petrovic:** *New concepts and materials, Laboratory Fellow, Change of Station assignment from LANL*
 - Phone 202 586-8058; john.petrovic@ee.doe.gov
- **New Hire:** *Materials science/engineering expertise junior/mid-level scientist/engineer- to be posted*
- **Sunita Satyapal:** *Hydrogen Storage Team Leader, Carbon Center of Excellence, Tanks, Analysis, IPHE activities*
 - Phone 202-586-2336; sunita.satyapal@ee.doe.gov

(Basic Science Contact: Harriet Kung, harriet.kung@science.doe.gov)