

# Design and Development of High-Performance Polymer Fuel Cell Membranes

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**GE Global Research**

DOE Hydrogen Program Review  
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GE imagination at work

*This presentation does not  
contain any proprietary or  
confidential information*

**Project ID #**  
**FCP 11**

# Overview

## Timeline

**Project start date:** 4/2006  
**Project end date:** 4/2011  
**Percent complete:** New Project

## Barriers

### Barriers addressed

- Membrane cost
- Membrane durability

## Budget

### Total project funding

- DOE share: \$1.5M
- GE share: \$0.5M

### Funding for FY06

- DOE: \$150K
- GE: \$50K

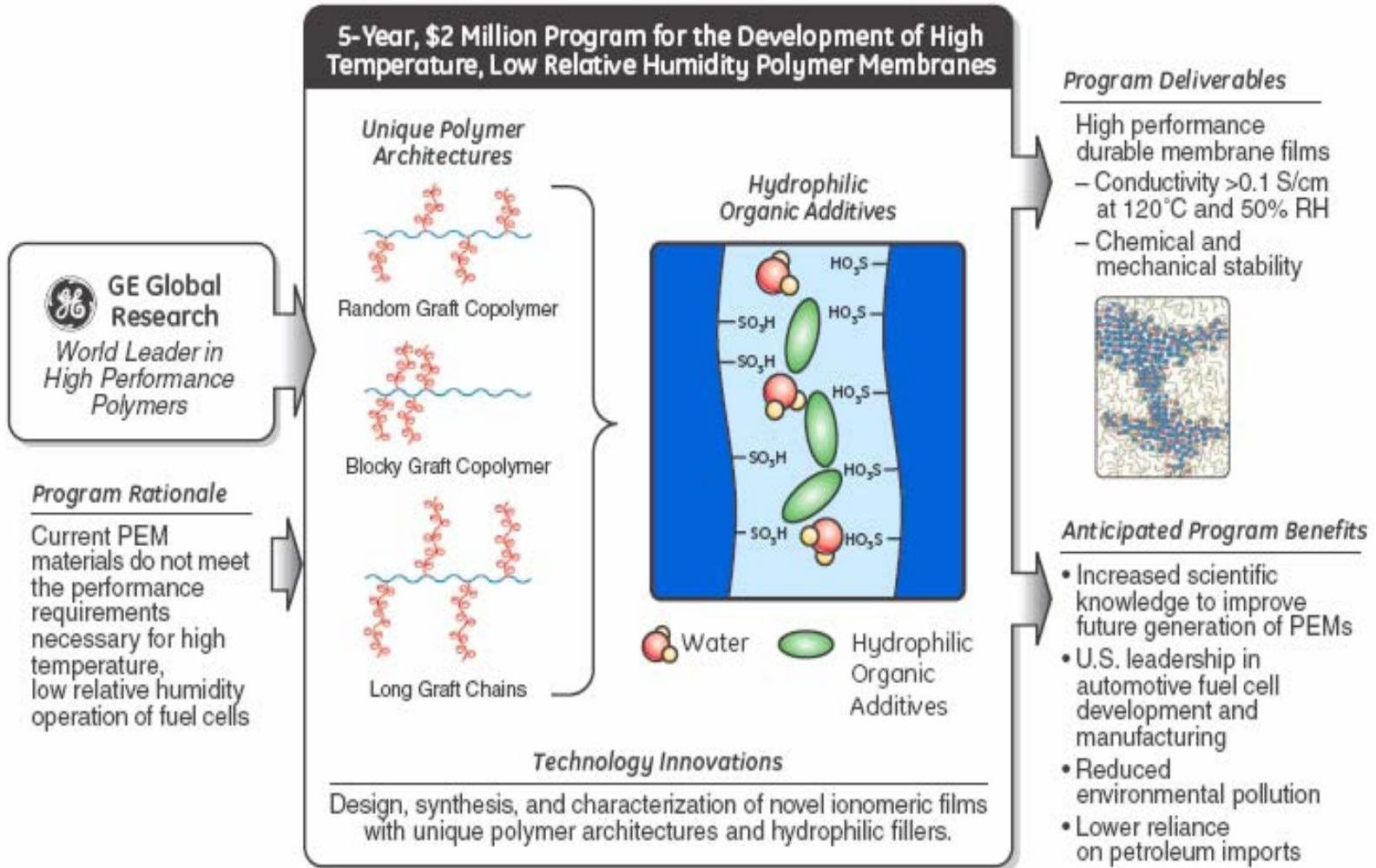
# GE Program Objective

Design and develop novel polymer electrolyte membrane materials for fuel cell operation at high temperature (up to 120 °C) and low relative humidity (25-50 %RH)

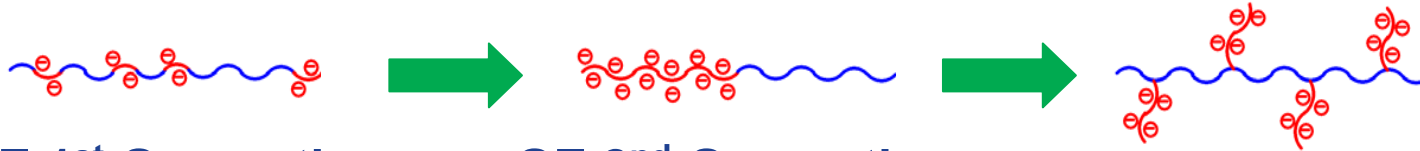
## Goals for FY06

- Design and synthesize new high performance polymer structures
- Design and synthesize hydrophilic organic additives
- Evaluate membrane performance with and without additives

# Approach



# GE Progress in New Membrane Development

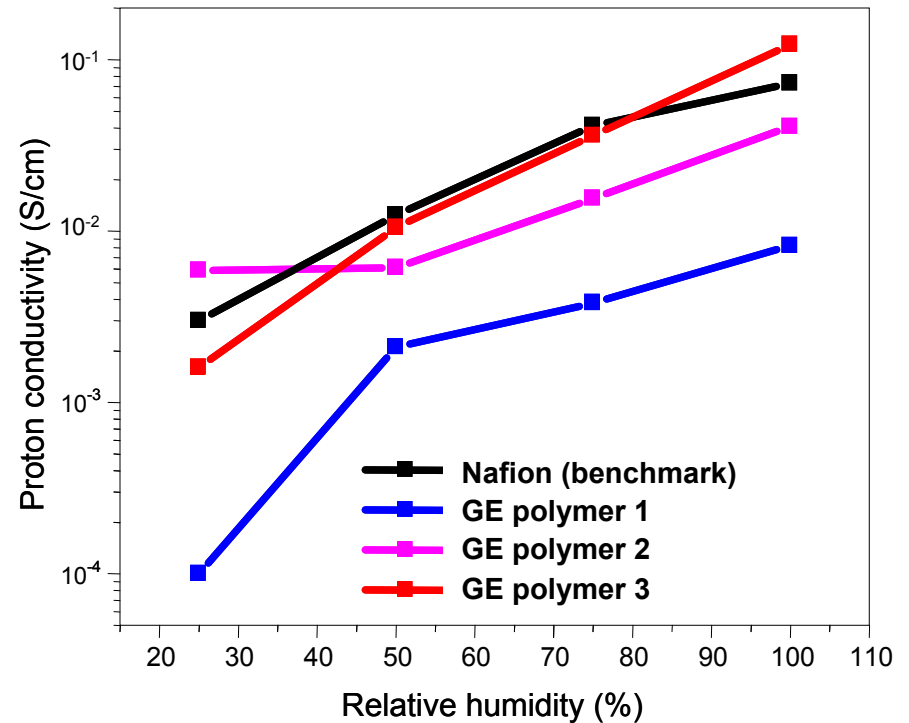
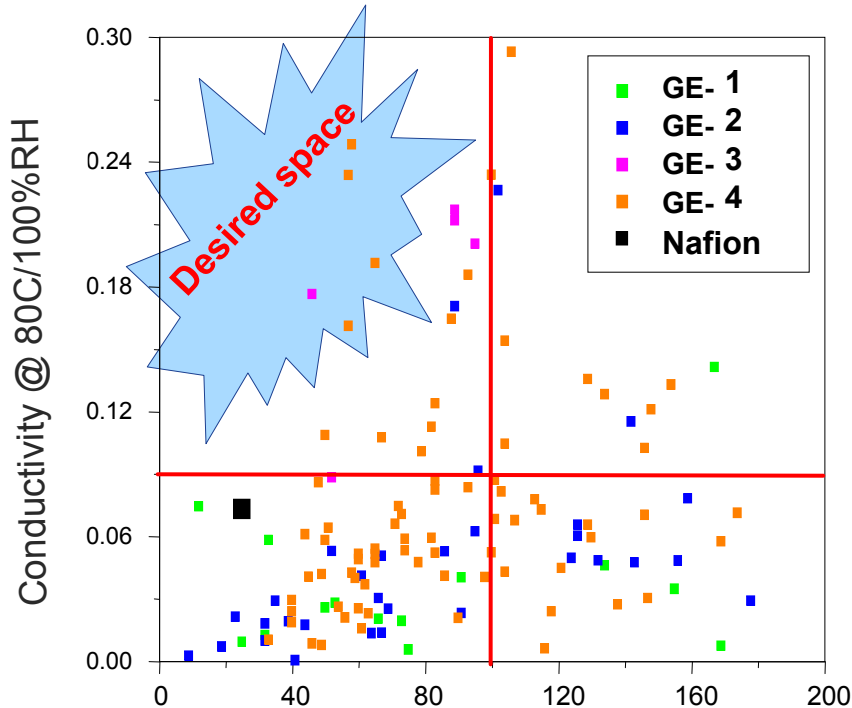


**GE 1<sup>st</sup> Generation**  
Random copolymers

**GE 2<sup>nd</sup> Generation**  
Block copolymers

**New Concepts**  
Unique polymer architectures

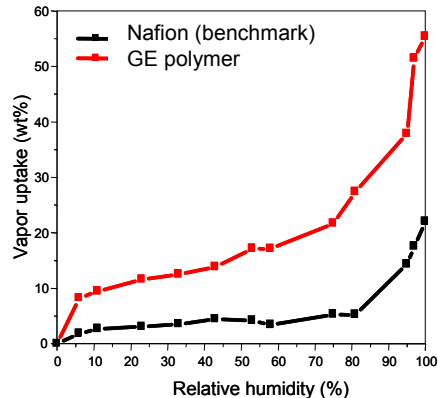
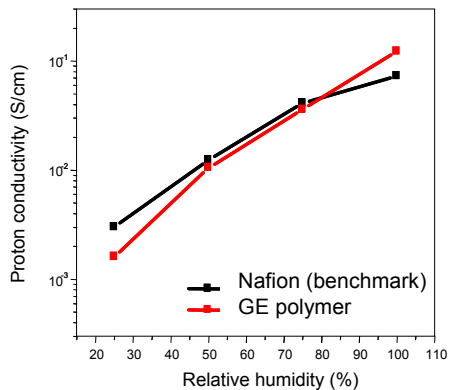
**Over 150 new materials and compositions**



# GE Membrane Characterization

## All samples

- Proton conductivity (9-point)
  - 20°C, 100 %RH; 100°C, 75 %RH
  - Humidity scan from 25-100 %RH at 80°C
  - Temperature scan from 60-120°C at 50 %RH
- Vapor uptake (12-point)
  - Room temperature, 6-100 %RH
- Water uptake (4-point)
  - Weight and volume change in liquid water at room temperature, 60, 90, and 100°C



## Select samples

- Microscopy (TEM, AFM)
  - Membrane morphology under dry and wet conditions
- Thermal-hydro-mechanical (TMA, DMA)
  - Dynamic, creep, and stress relaxation tests under dry and wet conditions
- Fuel cell performance
  - Gas crossover, polarization curves, durability at various temperature and RH conditions
- Accelerated durability testing
  - Mechanical and/or chemical stability under accelerated cycling conditions
- State of water (DSC, TGA)
  - Free, slightly bound, bound water, and their effect on polymers
- Small angle X-ray scattering (SAXS)
  - Membrane morphology under dry and wet conditions

# GE Material Design

## Thermally stable aromatic hydrocarbon polymers

- Build on GE's strength and expertise in engineering polymers

## No perfluorinated polymers

- Lower cost
- Benefit environment

## Balance proton conductivity, water uptake, and mechanical properties via material design

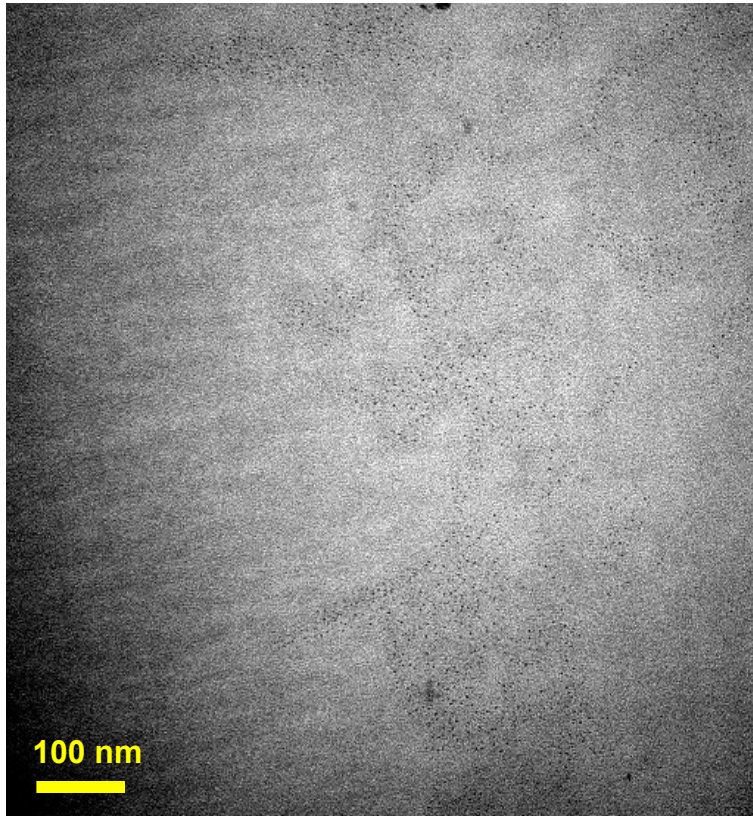
- Direct polymer synthesis from monomer building blocks
- Functionalization with acidic/basic groups, additives
- Control of membrane morphology through polymer architecture





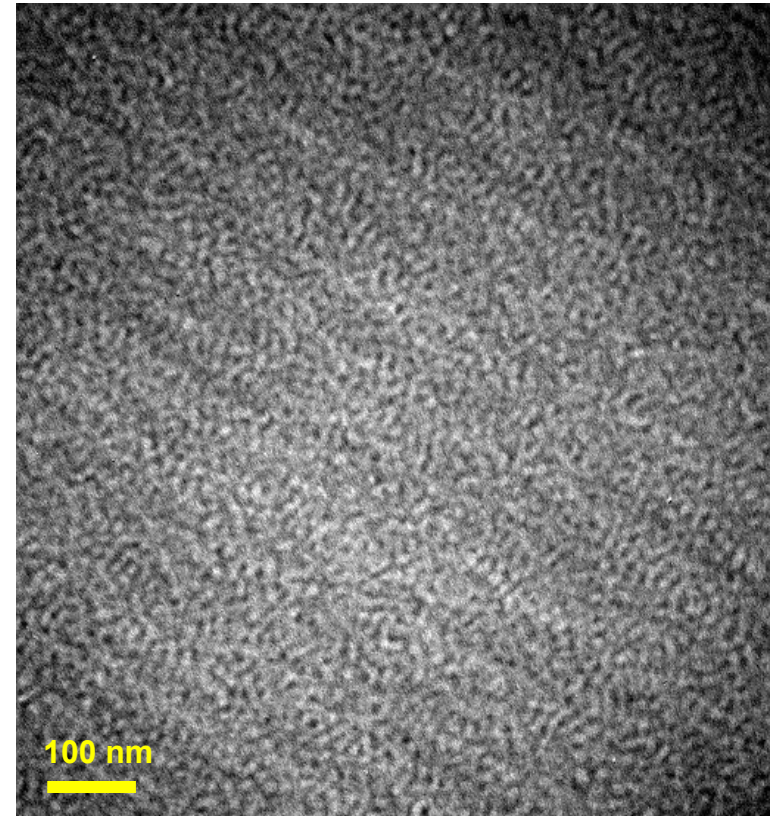
# Increasing Proton Conductivity

More open, connected membrane morphology  $\rightarrow$  higher proton conductivity



**Random Copolymer**

$\sigma \sim 10^{-3}$  S/cm (50 %RH)



**Block Copolymer**

$\sigma \sim 10^{-2}$  S/cm (50 %RH)



# Improving Performance, Cost, and Durability

## Optimize membrane morphology: new concepts in polymer chain design

- Promote phase separation and ionic aggregation
- Higher proton conductivity with good balance of water uptake properties

- **Design: Graft copolymers**

- Higher concentration of acidic groups on pendant chains
- More efficient proton conductivity due to enhanced phase separation
- Neutral polymer backbone provides better mechanical support when hydrated

Performance,  
Durability

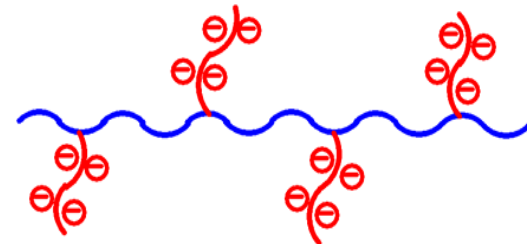
- **Materials: Aromatic hydrocarbon polymers**

- Unprecedented architectures in aromatic hydrocarbon polymers
- Synthesis is non-trivial

Lower  
Cost

- **Current status:**

- Several synthetic approaches developed
- Explored two approaches
- Synthesis in progress



# Improving High Temperature, Low RH Performance

## Additives to maintain high T, low RH performance:

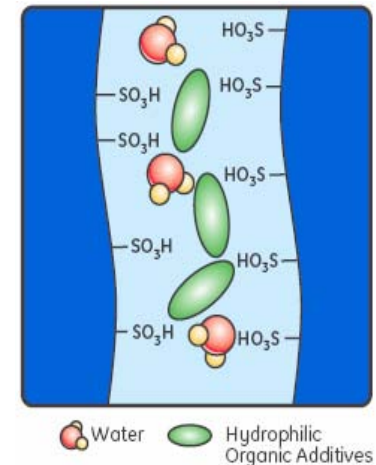
- Water retention at low RH
- Water supplement/replacement (with participation in proton conduction)

- **Design: Additives**

- Hydrophilic/hygroscopic
- Thermally, hydrolytically stable
- May participate in proton conduction

- **Materials: Organic compounds**

- Simple and versatile synthesis



- **Current status**

- Designed and synthesized several additive candidates

# Future Work

## FY06/FY07

### Materials synthesis

- Demonstrate feasibility of synthetic approaches to making new aromatic hydrocarbon polymer structures
- Design, synthesize, and characterize new monomers, polymers, and hydrophilic organic additives

### Membrane evaluation

- Evaluate membrane properties (proton conductivity, water uptake, mechanical properties)
- Study membrane morphology to understand the effect of variations in polymer architecture on membrane performance



# Project Summary

## Relevance

Apply new concepts in polymer membrane design to resolve challenging technical issues related to membrane performance over a wide range of temperatures and humidities.

## Approach

Design and synthesize new polymer architectures that promote membrane phase separation and ionic aggregation. Explore hydrophilic organic additives to improve performance at high temperature, low RH.

## Progress

Developed synthetic approaches. Synthesis and evaluation in progress.

## Future research

Continue design, synthesis, and evaluation of new materials. Develop further understanding of the effect of polymer architecture on membrane morphology and performance.