Integrated Short Contact Time Hydrogen Generator (SCPO) 2006 DOE Project Review Meeting

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This presentation does not contain any proprietary or confidential information

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imagination at work

Project ID # PD3

Overview

Timeline

Project start date: 05/30/2005 Project end date: 05/30/2008 Percent complete: 38%

Budget

- Total project funding
 - A DOE share: \$2.6M
- A Contractor share: \$1.4M
- Funding received in FY05: \$490K Funding for FY06: \$400K

Barriers

- Technical Barriers Addressed:
 - A. Cost of Fuel Processor
 - C. Operation and Maintenance (O&M)
 - D. Feedstock Issues
 - E. Catalyst sulfur tolerance & durability
- Technical Targets (2010):
 - Total Energy Efficiency (%LHV) > 75%
 - Total $H_2 \text{ Cost} < \$3.00/\text{gge H2}$

Partners

- University of Minnesota
- Argonne National Lab







SCPO Project Objectives & Highlights

Project Team

GE Global Research

- •System & Economic Analysis
- •Catalyst Dev. & High P Valid.
- •Prototype Reactor Design
- •Overall Project Management

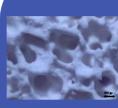
University of Minnesota

- •CPO Catalyst Discovery
- •Parametric Testing & Modeling
- •Catalyst Characterization

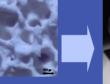
Argonne National Laboratory •SMR Catalyst Discovery •Catalyst Durability

•Catalyst Characterization

Project Objective: To Develop a Compact Hydrogen Generator that can Deliver H₂ at a Cost of <\$3.00/kg



Lab Screen





Pilot

Technical Approach

•Develop Robust Short Contact Time Catalysts •Develop & Design Comp. CPO, SMR, WGS & HE •Demonstrate Critical Components

Prototype Scale

Anticipated Benefits

•Compact, Low Cost, H2 Generation Technology •Applications in Refueling Stations, NGCC NOx Reduction & CO2 Capture

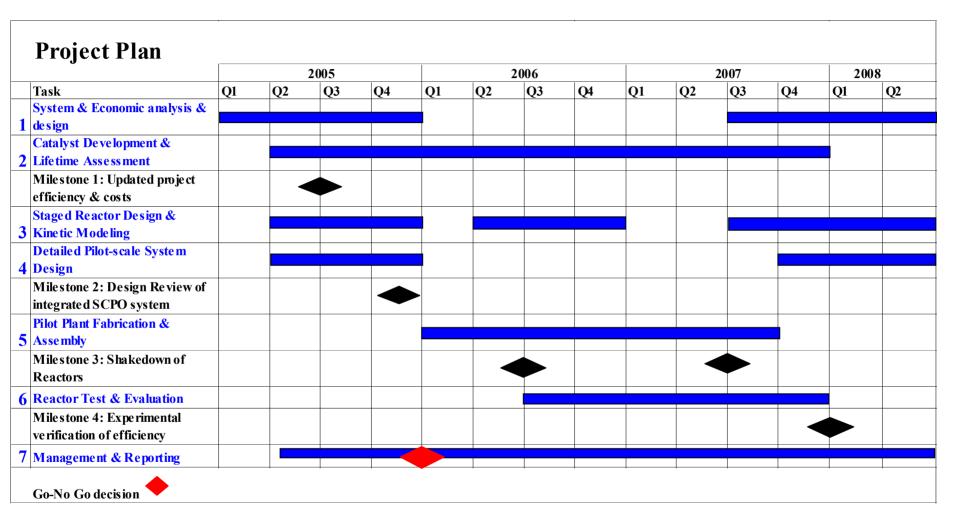
Deliverables

•Short Contact Time Catalysts •Demonstration of Critical Technology & Components •Reactor & Process Model for Scale-up

- DOE award announced in November 2004
- Initiated in January 2005

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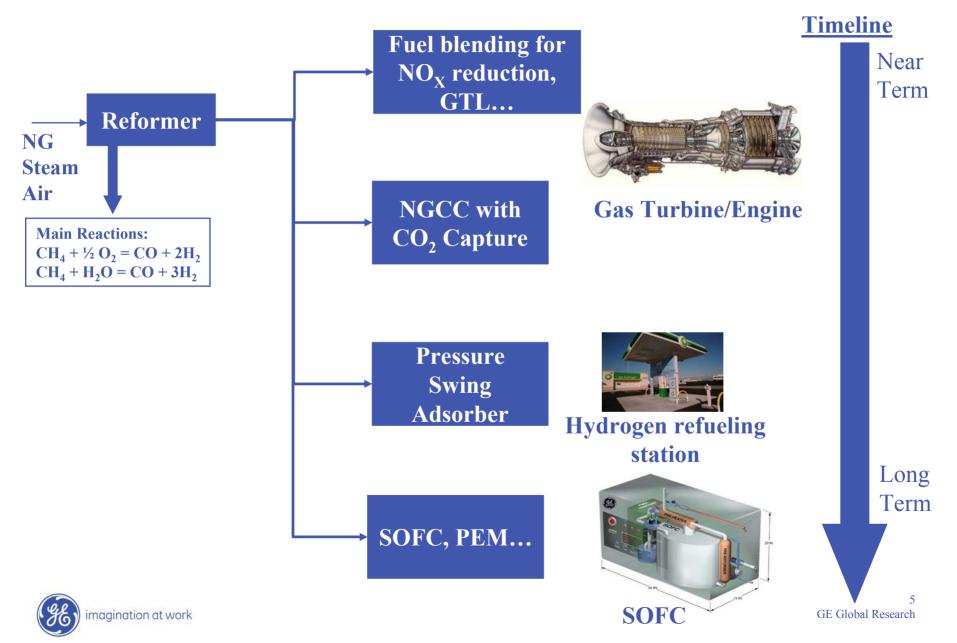
Overall Project Plan



YE '05 Go/No Go



Reforming & Potential Applications

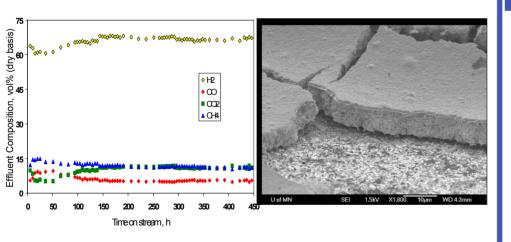


Approach: Leverage Partners' Facilities & Knowledge Base

Integrated Activities

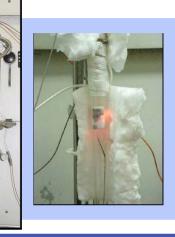
- GE design the overall system & define system conditions
- ANL & U of Mn have created catalysts and tested them at conditions GE identified
- U of Mn: New CPO catalyst & kinetics model
- ANL: New SMR catalyst & long term stability of catalysts
- •Goal: Increase GHSV & S tolerance to lower capital cost.

Example Data



U of Mn Invented the CPO





ANL Has the State-of-Art Cat Testing Facilities





Argonne National Laboratory – Activities and Objectives

Evaluate the performance and durability of the "best available" CPO, SMR, and WGS catalysts from commercial catalyst manufacturers

A Test matrix

-CPO (2 vendors, 5 different formulations)

-SMR (2 vendors, 3 different formulations)

-WGS (1 vendor, formulation selection not finalized)

A Objectives

-Measure conversion and selectivity as a function of operating parameters $(O_2: C \text{ and } H_2O: C \text{ ratios, temperature, GHSV, sulfur content})$

– Determine sulfur tolerance and durability

-Determine failure modes/deactivation mechanisms (if applicable)

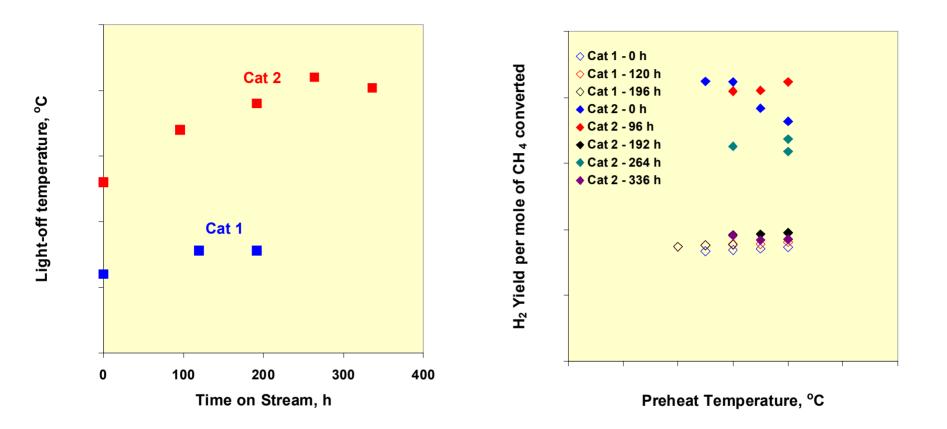
Argonne steam methane reforming (SMR) catalyst development

A Improved activity and stability under demanding SMR conditions $-low CH_4$ and high H_2 , CO, and CO₂ concentrations

-short contact time



Results: Evaluating the durability of CPO catalysts



Cat 2 initially displayed higher activity than Cat 1; however after about 200 h on stream the activities were comparable



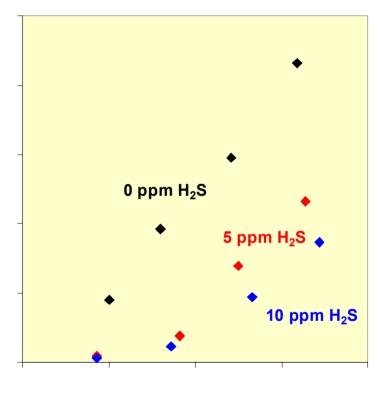
Results: Evaluating the sulfur-tolerance of commercial **SMR** catalysts

Sulfur tolerance of SMR catalyst may be more critical than that of CPO catalyst

Sulfur poisoning of SMR catalysts

- A rapid decrease in activity when exposed to a few ppm of H₂S
- CH4 Conversion, % A activity stabilizes after initial decrease with only a very slight decrease in activity observed over next 8-24 h
- A essentially complete recovery of activity when H₂S is removed

Long-term effect of sulfur on catalyst performance and stability is unknown and being investigated



Results: Argonne SMR catalysts

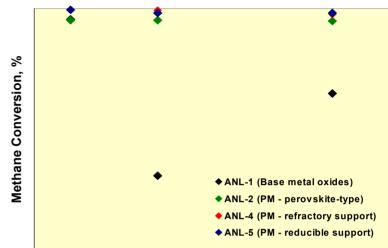
Six different formulations are being evaluated

- A Base metal mixed oxide (ANL-1)
- A Precious metal perovskite-type oxide (ANL-2)
- A Precious metal-modified hexaaluminate (ANL-3)
- A Oxide-supported precious metals (ANL-4, -5, -6)

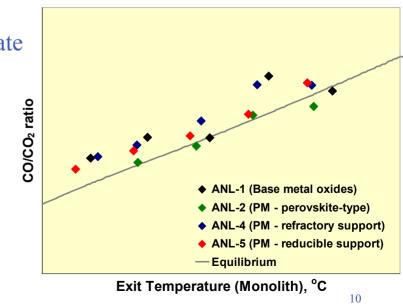
Base metal oxide (ANL-1) - less active and temperature cycling accelerates loss in activity.

PM perovskite-type oxide (ANL-2) and PM hexaluminate catalysts (ANL-3) - good activity but stability is questionable, especially for the perovskite.

Oxide-supported PM catalysts (ANL-4, -5, -6) – good activity and good temperature cycling stability.



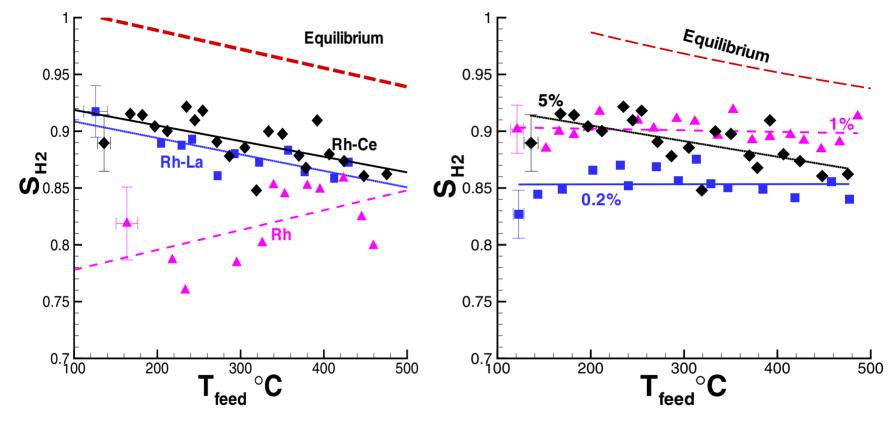
Gas-Hourly Space Velocity, h⁻¹



GE Global Research



Results: Investigation of Catalyst system and catalyst loading



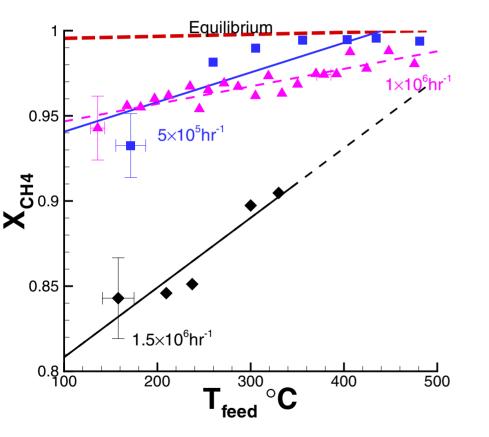
1×10⁶ hr⁻¹, 5% Rh, 80 ppi, 2% Ce/La

1×10⁶ hr⁻¹, 80 ppi, 2% Ce/La





Results: Effect of Flow rate

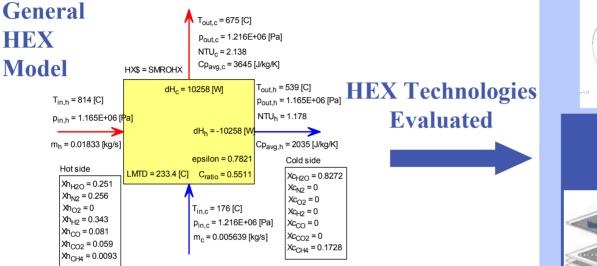


- The reaction zone is pushed to the back with increasing flow
- Experiments with 3mm monoliths show that at 1×10⁶hr⁻¹ the reaction is not complete
- The support determines the effect of mass & heat transfer which are the limiting factors

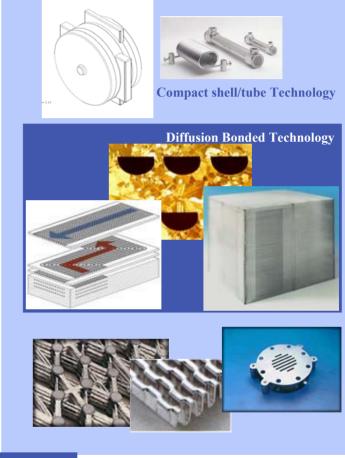




Progress & Results: Heat Exchanger Design Process



Design of Modular, Manufacturability, Compact Heat Exchangers for SCPO





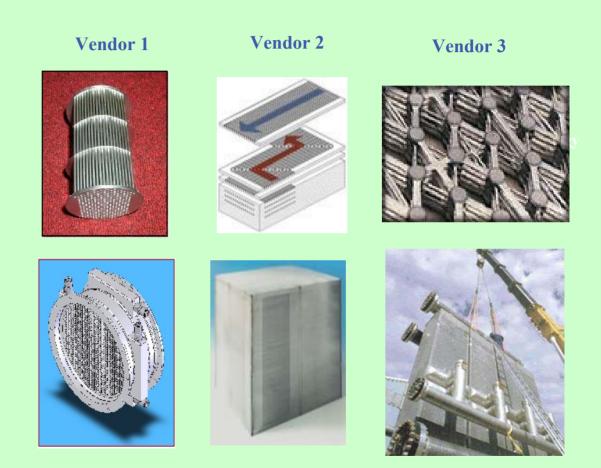
- Vendor database for GRC
- Leverage experience
- Visited vendors
- 6-Sigma tradeoff analysis





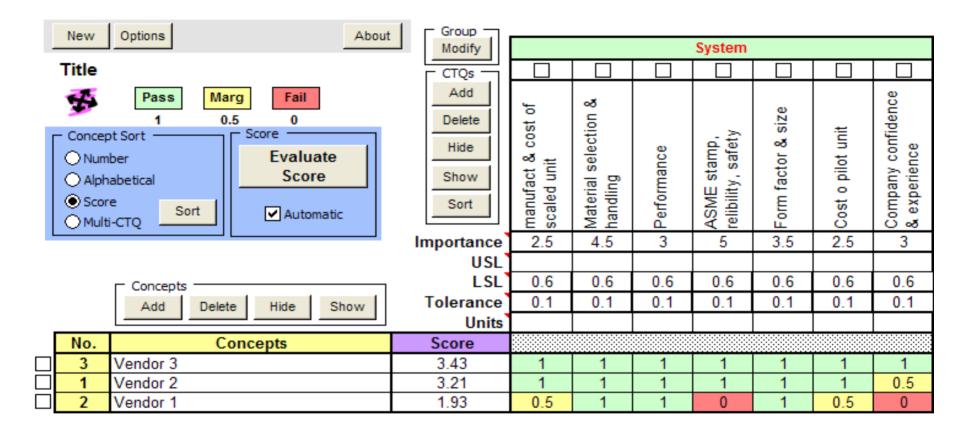
Results: Identified HEX Technology Among >50 Vendors

- GRC created HEX specifications
- Built Hex vendor database (>100 vendors)
- Vendors dropped off:
 - Missing form factor
 - No high T designs
 - Slow responses
 - Expensive
- [~] Identified 3 vendors
 - -Designs and quotes
 - -Visited all of them





HEX Technology Trade-Off Matrix



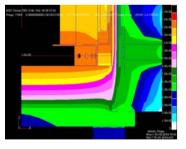
• Used DFSS 6-sigma tool to identify HEX vendor & technologies

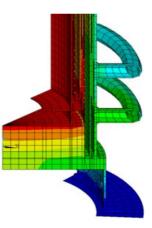


Modeling & Analysis Tools will be Used to Ensure Proper Design

Material/Criterias	Importance	SS304	Inconel 600	Inconel 601	Inconel 690	Inconel 693	Inconel 625	Inconel 617	Inconel 800H	Inconel 800HT	RA330	RA333	RA602CA
Maximum Temperature (F)			2000	2100	2000	2100	2000	2000	1900	1900	2200	2200	2200
Material Properties and Resistance													
Thermal Coefficient of Expansion	3		4	4	4	4	5	5	3	3	3	4	4
Oxidation Resistance	4	1	3	4	4	5	5	4	2	2	3	4	5
Carburization Resistance	4	1	4	4	4	5	4	5	3	3	3	4	4
Rupture Life test	4		4	4	2	4	5	5	4	4	3	4	5
Metal Dusting in a Reduced Enviroment	4		3	2	4	5	3	4	1	1	3	3	3
Costs and Availability													
Cost per ft	5		5	5	1	1	4	1	5	5	5	1	1
Lead time	5		5	5	2	1	5	2	4	4	4	2	1
	Total		118	118	83	98	128	102	94	94	102	87	90

Material of construction will be selected after careful evaluation





Thermal & stress modeling will be performed to ensure expected vessel lifetime. The design will satisfy ASME codes.

Leverage prior experience in ACR development



Progress & Results: PSA System Evaluation



Separation Summary

•2 leading PSA technologies evaluated, none currently reach DOE goal of 90 % recovery

•Compressor systems for both storage and feed composition have been found and quotations received

•Alternative technologies for feed compositions (oxygen membranes) were evaluated.

•Pressure of SCPO system becomes important for purification

•Storage systems for hydrogen have been studied

	DOE	CGA		
	Proposal	Industrial H2,	Vendor A	Vendor B
Species	Specification	Grade B	(120 PSI)	(120 PSI)
H2	98%	99.95%	99.996	99.95
CO	<1 ppm	<10ppm	<1.0 ppm	ND
CO2	<100 ppm	<10 ppm	<1.0 ppm	ND
N2	balance	<400 ppm	44.4 ppm	500 ppm
CH4	<100 ppm	<10 ppm	<1.0 ppm	<1 ppm
Recovery			75%	79%



Assess "Cost of Hydrogen": DOE H2A

Model output

Hydrogen cost for 10 year life of refueling station

Key inputs

- Detailed installed capital costs
- Process operating efficiencies
- Feedstock costs
- O&M

Enables

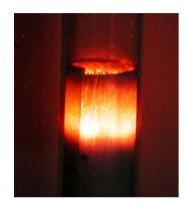
Comparison across alternative reforming technologies

Calculated Base Case Cost of Hydrogen is \$3.05/kg. We believe the SCPO capacity factor >70%, thus we can reach DOE target of <\$3.00/kg



(SCPO) Staged Catalytic Partial Oxidation Reformer Benefits

Short Contact Time CPO Catalysts





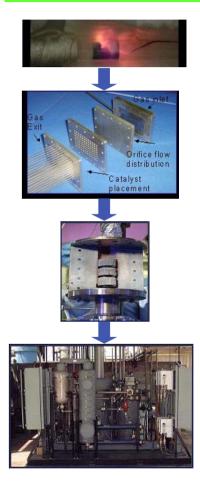
Most cost effective hydrogen production unit

- Compact (5 times smaller than ACR)
- Modular design for mass production
- Balance efficiency & cost

Ease of ownership

- "Maintenance free" Reformer life >10 y; Catalyst life>5 y
- ➤ Fuel flexibility
- **Environmentally sound**
 - ➢ Higher efficiency
 - No emissions

Patented Staging of CPO & SMR Catalysts



Cheap Compact Reformer

SCPO Accomplishments So far

- □ System analysis/design completed.
- □ System pressure trade off analysis completed
- □ Base case catalysts identified
- **Reactor sizing / design completed**
- □ HEX technology tradeoff completed
- □ HEX technology selected, design completed
- Control strategy, start-up & shut-down procedure developed
- Completed cost analysis using GE's process model & DOE's H2A model
- **Prototype & Lab unit design / budgeting completed.**
- □ ANL & U of Mn worked closely with GE and delivered catalyst test results
- **Conducted preliminary FMEA analysis, major risks identified**



Conclusions & Recommendations

□ SCPO will be the leading technology for H2 production from NG. It is the most cost-effective H2 production technology based on the economic analysis of different H2 production technologies. With minor modification, we can extend the feed to gasoline, diesel, ethanol & methanol.

□ The technologies developed in this program has good synergies with application in fuel blending, NGCC with CO2 capture, SOFC & syngas production for GTL....

Design and build CPO & SMR reactors in 2006 and conduct testing in 2006 & 2007.



Supplemental Slides

The following three slides are for the purposes of the reviewers only – they are not to be presented as part of your oral or poster presentation. They will be included in the hardcopies of your presentation that might be made for review purposes.



Publications and Presentations

• No publications so far, except last year's DOE H2 conference presentation & DOE quarterly reports.

• Two patent applications filed into US patent office.

