

Development and Evaluation of Advanced Hydride Systems for Reversible Hydrogen Storage

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– A Participant in the DOE Metal Hydride Center of Excellence –

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This presentation does not contain any proprietary or confidential information

Project ID
ST-17

Timeline

- Project start date: FY05
- Project end date: FY09
- 25 % complete

Budget

- Expected total project funding:
 - \$1.78M (DOE)
- Funding received in FY05
 - \$150K (DOE)
- Funding received for FY06:
 - \$261K (DOE)

Barriers/System Targets

- A. System Weight and Volume
 - 2010 Targets: 6 wt.% & 45 gH/L
- D. Durability/Operability
 - 2010 Target: Life of 1000 cycles
- E. Charging/Discharging Rates
 - 2010 Target: Fill time of 3 min for 5 kg H₂
- P. Lack of Understanding of Hydrogen Physisorption and Chemisorption

Partners

- Participant in DOE MHCoe – collaborations with partners in five sub-group Tasks [primarily with Caltech, HRL, NIST, BNL, SNL, LLNL in FY-06]
- Washington U. and Caltech to support a new [FY-06] BES Project on solid state NMR of light element hydrides
- International: IFE (Norway), Philips (Netherlands), CNRS (France), and AIST (Japan)

Develop and demonstrate light-metal hydride systems that meets or exceeds the 2010/2015 DOE goals for on-board hydrogen storage

- (1) Validation of initial storage properties and reversibility in light element metal hydrides and assess their aging durability during extended cycling**
 - Nanophase, destabilized hydrides based upon LiH, MgH₂, & LiBH₄ produced at HRL, Caltech, & other MHCoe partners.
 - Complex hydrides (e.g., amides/imides, borohydrides, & AlH₃-hydrides) provided by SNL, NIST, BNL. & other MHCoe partners
- (2) Support developing lighter weight and thermally efficient hydride storage vessels and experimentally demonstrating their compatibility with appropriate complex and destabilized nanophase hydrides.**

FY05/FY06 Objectives:

- **Determine reversibility of the destabilized LiH/Si & LiH/Ge systems**
- **Evaluate behavior of destabilized MgH₂/LiBH₄ & MgH₂/Si systems to assess reversibility, kinetics, & H₂ storage parameters against targets**
- **Characterize phases & chemical bonding via MAS-NMR for Li amides/imides, AlH₃, & selected other hydrides provided by MHCoe partners to better understand basic chemisorption processes**
- **Start extended cycling tests on at least one destabilized hydride (i.e., Li-Mg-N-H) to assess lifetime potential & durability.**

Perform Analysis and Characterization of Selected Hydrides:

- Volumetric measurements hydrogen storage capacities and pressures on destabilized nanophase and complex metal hydrides.
- Magic Angle Spinning - Nuclear Magnetic Resonance (MAS-NMR) measurements performed at Caltech Solid State NMR Facility to assess the phase compositions and chemical bonding parameters.
- Examinations by neutron scattering and diffraction, etc. in collaboration with MHCoE partners NIST and Caltech.

Prototype Hydride Beds Development and Life Testing:

- Evaluate the performance and robustness of candidate hydrides using well-characterized experimental test-beds during many cycles of hydrogen absorption and desorption.
- Support development of more efficient hydride storage vessels to reduce storage system mass and demonstrate their compatibility with appropriate complex and destabilized nanophase hydrides.
- Support system design and analyses using methods established at JPL for sorption cryocooler hydride compressor beds.

DOE/EERE

Metal Hydride Center of Excellence (MHCoE)
MHCoE Coordinating Council ('06)
Sandia National Laboratory (SNL) - Lead

Project Groups

A

Destabilized Hydrides

- HRL(POC)
- Caltech
- **JPL**
- Stanford
- U. Hawaii
- U. Pitt
- UIUC
- U. Utah
- Intematix
- NIST

B

Complex Anionic Materials

- SNL(POC)
- U. Hawaii
- UIUC
- **JPL**
- ORNL
- NIST
- U. Pitt
- UNR

C

Amides/ Imides (M-N-H System)

- SNL(POC)
- GE
- U. Utah
- UNR
- ORNL
- Intematix
- U. Hawaii
- **JPL**

D

Alanes (AlH₃)

- BNL(POC)
- SRNL
- **JPL**
- U. Hawaii
- SNL

E

Engineering Analysis & Design

- SRNL(POC)
- SNL
- NIST
- **JPL**
- GE

F

Novel or Cross Cutting Concepts

To be Determined Later.

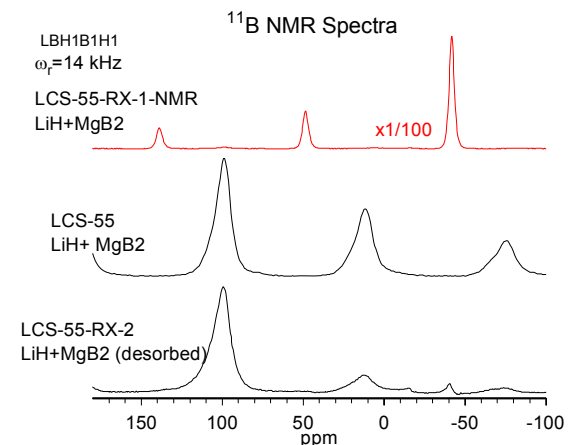
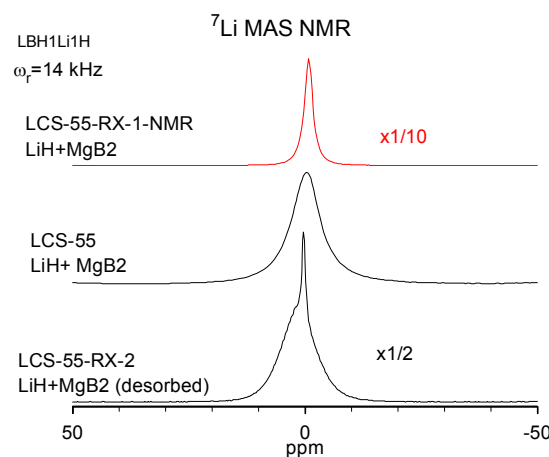
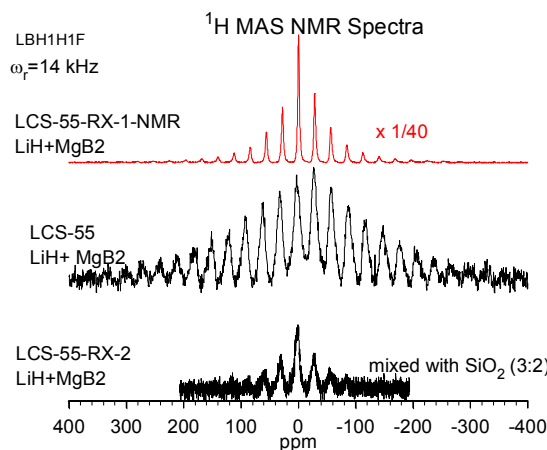
• JPL Objectives in FY05/FY06:

- Validation of initial storage properties and reversibility in nanophase, destabilized hydrides based upon LiH, MgH₂, LiBH₄ & others and also to assess their aging durability during extended cycling

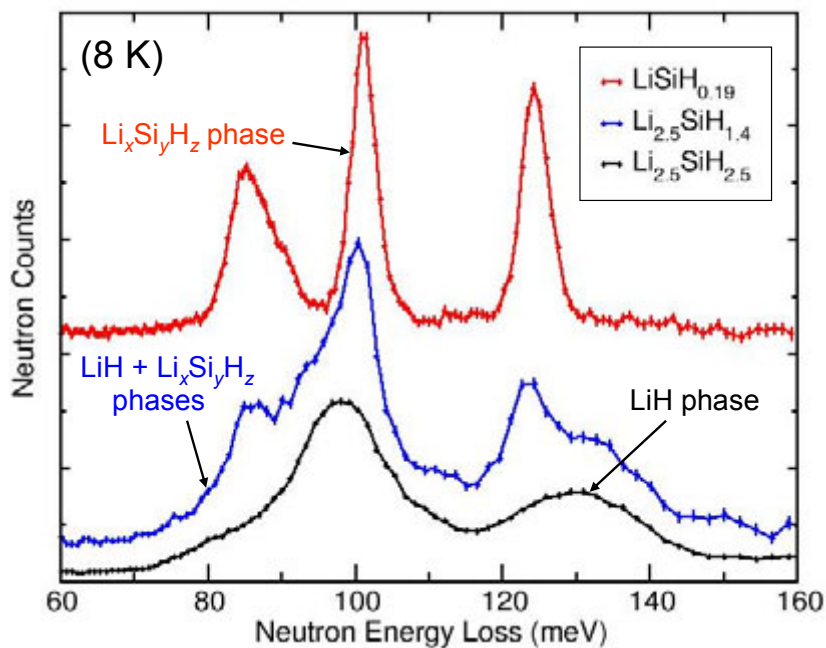
• Accomplishments:

- MAS-NMR determined phase formation and reversibility in LiBH₄/MgH₂ (SNL):
 - ⁷Li, ¹¹B and ¹H MAS-NMR spectra showed expected different phases with variation in hydrogen contents – foundation for more systematic studies of phase conversion, reversibility, catalytic effects, & degradation behavior.

Samples	Code	Treatment	Comments
LiH+MgB ₂	LCS-55	As ball milled	From J. Vajo [HRL]
LiBH _x +MgH ₂	LCS-55: RX-1	Absorbed H ₂	Saturated hydrides
LiH+MgB ₂	LCS-55: RX-2	Desorbed H ₂	very metallic, max spinning = 6 kHz
MgB ₂ + LiH + LiBH _x	LCS-55: RX-2 + SiO ₂ Powder	Desorbed H ₂ : diluted for better MAS-NMR	Incomplete desorb reaction noted



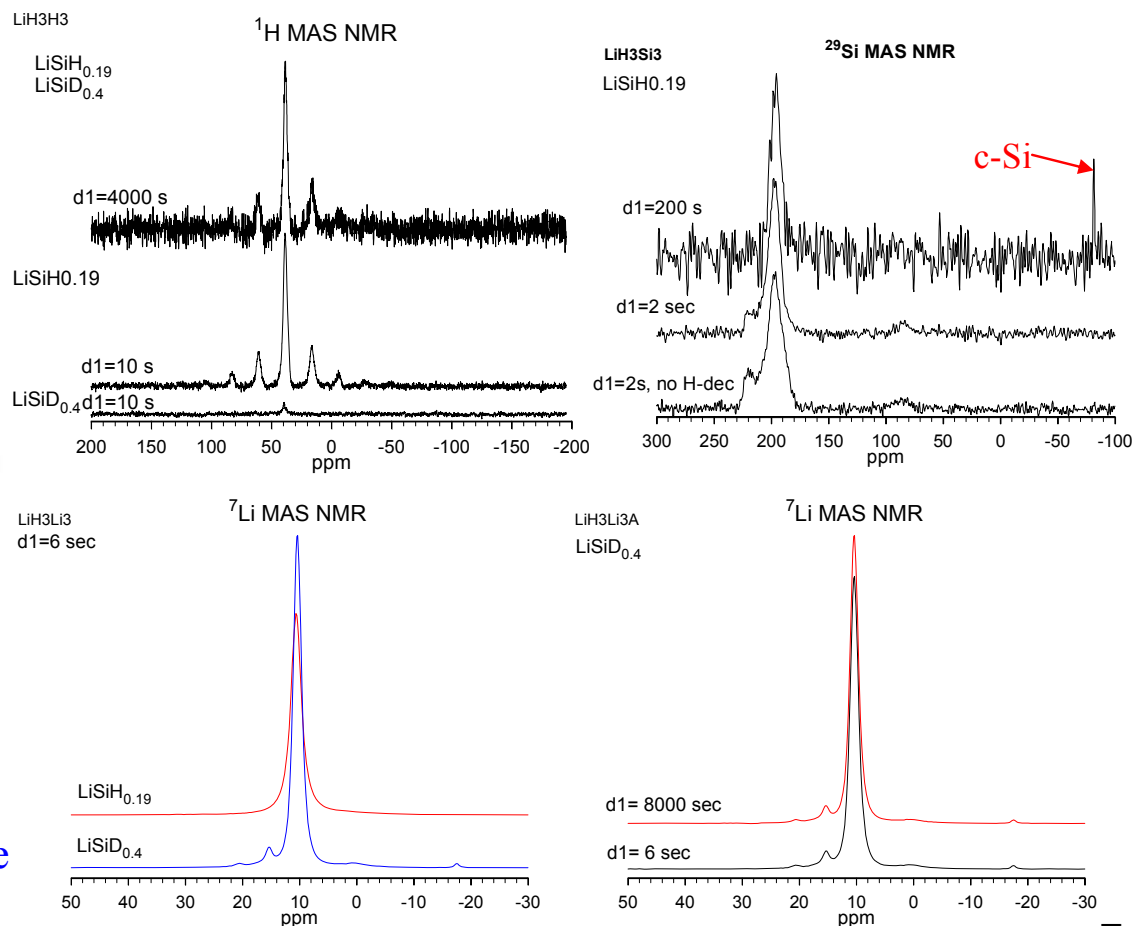
LiH/Si samples – neutron vibrational spectra (NIST), XRD & neutron diffraction patterns, & proton contents verify formation of a ternary Li-Si-H phase initially detected as impurity component by NMR and x-ray diffraction in cycled $\text{Li}_{2.5}\text{SiH}_x$



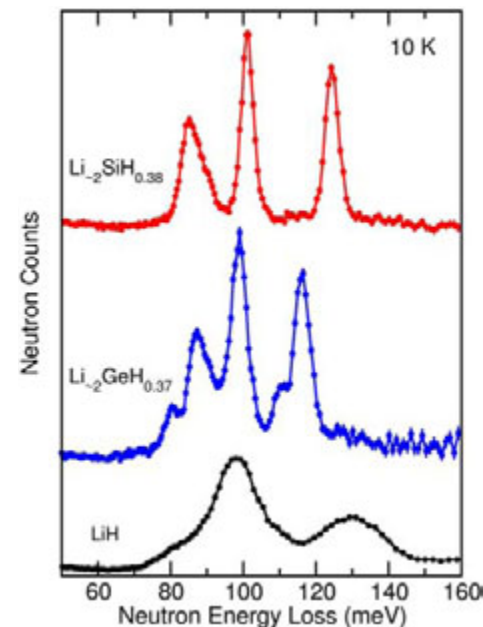
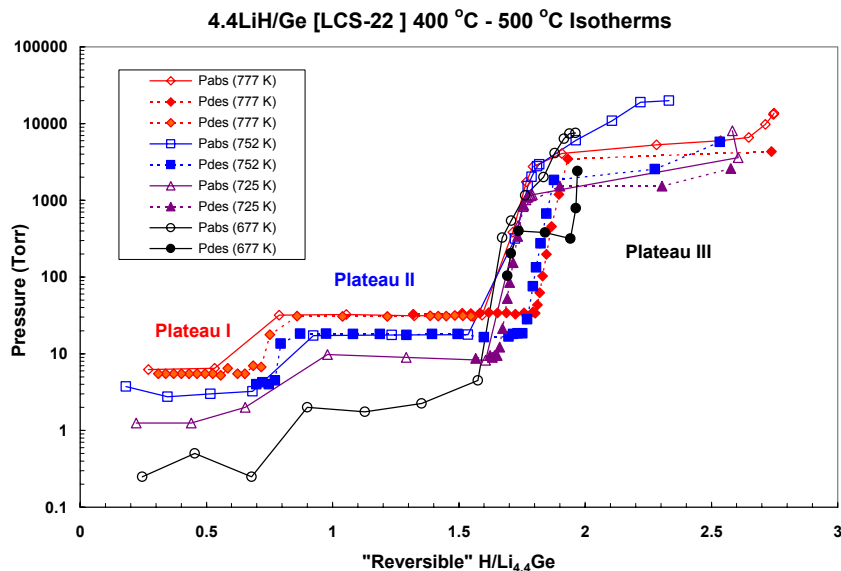
Neutron Vibrational Spectrum of Unknown $\text{Li}_x\text{Si}_y\text{H}_z$ Ternary Phase in $\text{Li}_{2.5}\text{SiH}_{1.4}$ at 8 K.

Conclusion: LiH/Si system illustrates both destabilizing and formation of previously unexpected ternary (i.e., $\text{Li}_y\text{Si}_z\text{H}_x$) phases that limits of complete reversibility – an issue to be looked for with other combinations.

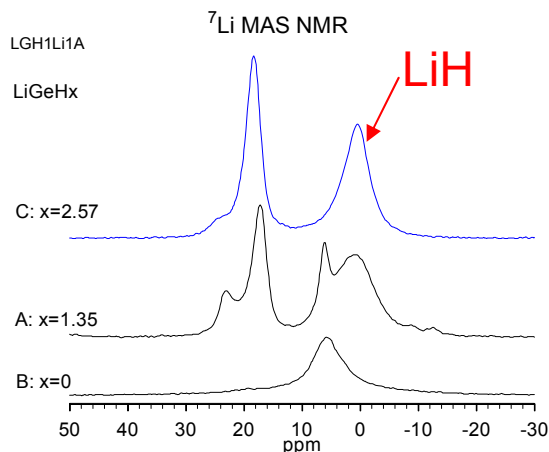
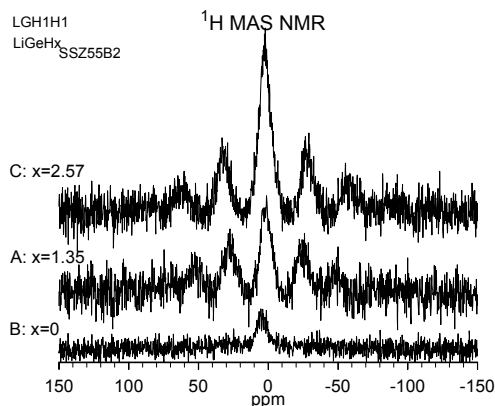
MAS-NMR confirm the LiSiH phase [Li_ySiH_x] and some c-Si phase, **but no LiH!**



LiH/Ge samples – Similar destabilization behavior to LiH/Si system without a ternary phase a higher LiH/Ge ratio but a $\text{Li}_y\text{Ge}_z\text{H}_x$ phase @ LiH/Ge ~ 2 (NIST)



NMR & XRD sees only mixture of LiH & Li_yGe_z phases for LiH/Ge ≥ 4



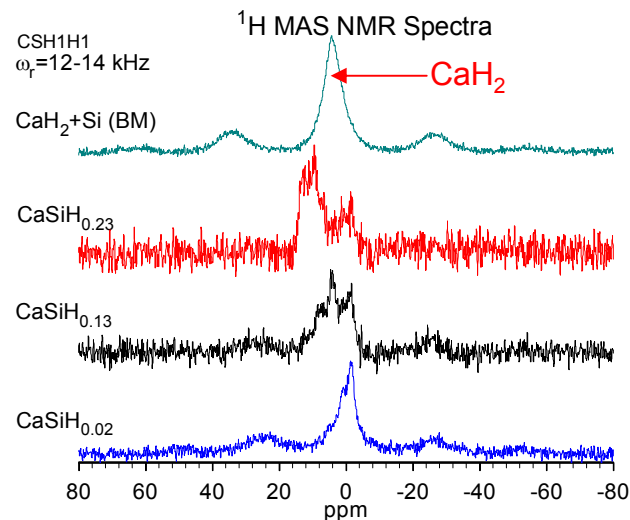
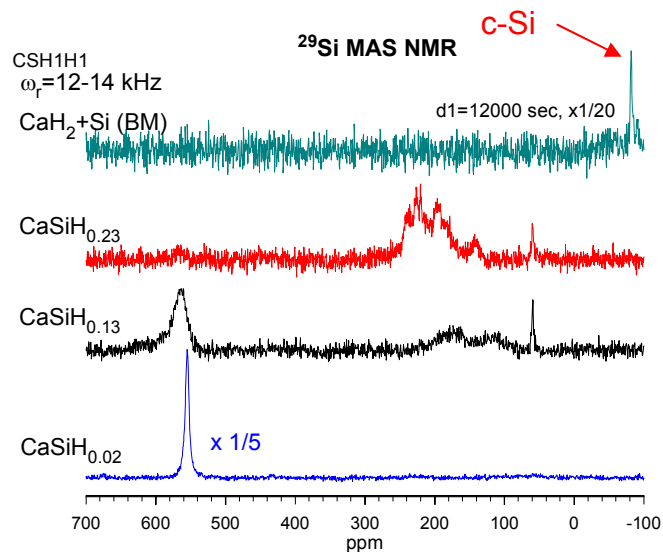
- NPD and NVS measurements indicate that the new $\text{Li}_x\text{Ge}_y\text{H}_z$ ternary phase is similar in structure to the previously discovered $\text{Li}_x\text{Si}_y\text{H}_z$ ternary phase.
- NPD, NVS, and Prompt-Gamma Activation Analysis (PGAA) suggest that the Li/Ge and Li/Si ratios for the ternary phases are equal to or slightly greater than 2.
- The ternary-phase structures for both ternary phases are still unknown.

•JPL Objectives:

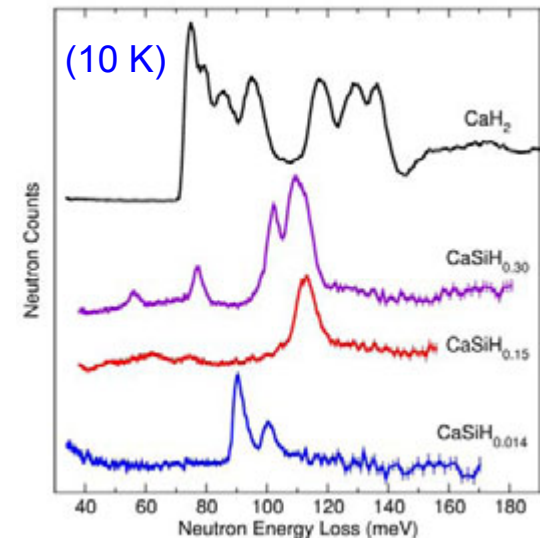
- Support phase characterizations (i.e., structure & bonding properties) for new borohydrides [i.e., $\text{Ca}(\text{BH}_4)_2$] and silicide hydrides [i.e., Na-Si-H and Ca-Si-H] with NMR in collaboration with SNL, NIST, Caltech, & LLNL
- Improve understanding of catalysts, dopants, and processing on alanates, borohydrides, and other hydrides.
- Provide novel insights on the phase compositions and local chemical bonding parameters for crystalline and highly disordered (i.e., amorphous) phases at various stages of reactions.
- NMR results will test and complement theoretical modeling of nanophase formation and transitions

•Accomplishments:

- Obtained MAS-NMR results on samples from MHCoE partners:
 - Ca-Si-H (NIST) – ^{29}Si and ^1H MAS-NMR spectra showed changes with different hydrogen contents similar to hydrogen vibration spectra from NIST due to site occupancy and local structures
 - Supporting analyses of NMR data from LLNL on Ca-B-H and Na-Si-H samples

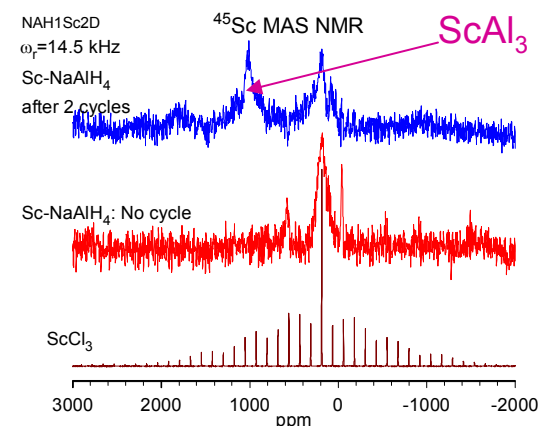
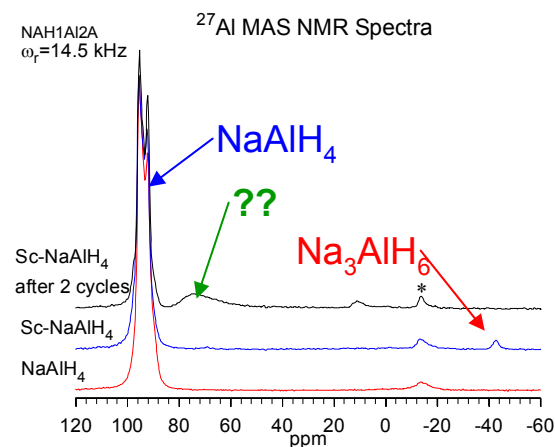
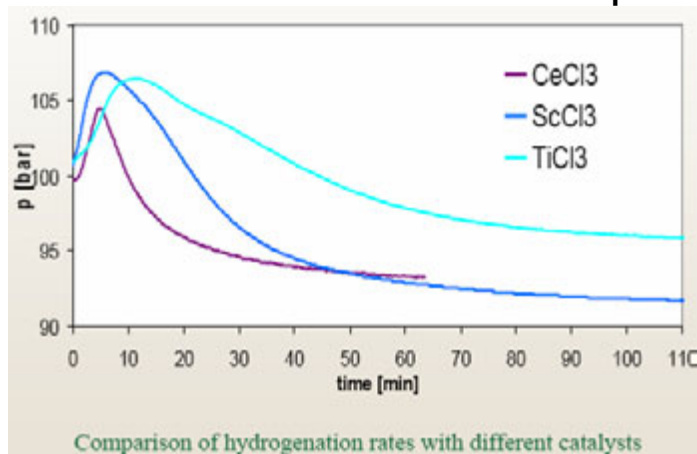


Neutron vibrational spectra of CaH₂ as well as CaSiH_x after various hydrogen treatments (NIST Data)

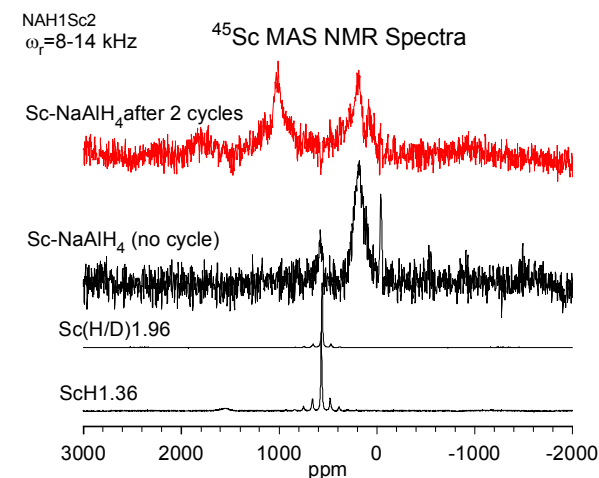
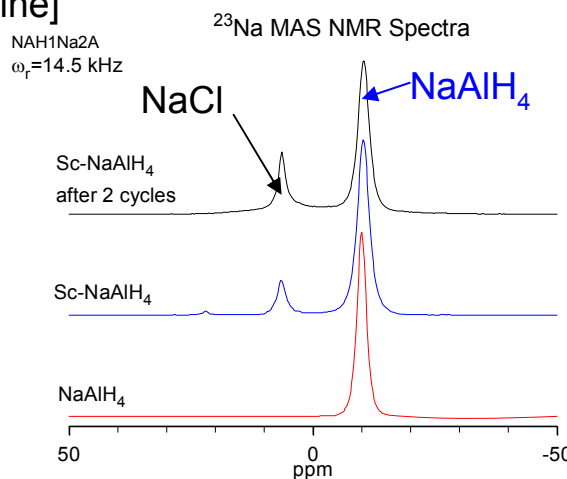
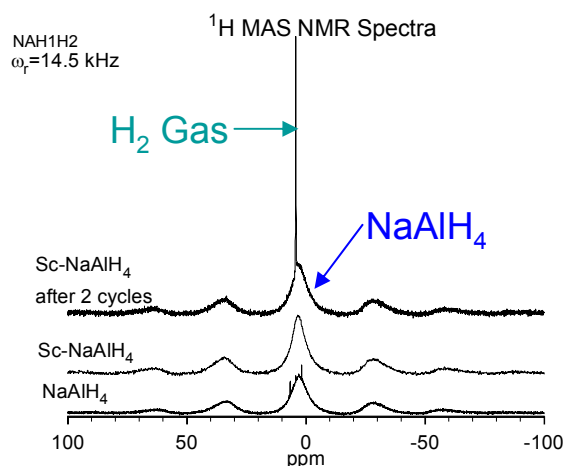


Investigating catalyst mechanisms in alanates & borohydrides – combined study with Washington U./Caltech [Joint effort with BES funding] & IFE [Norway].

- Ti roles in alanates (i.e., NaAlH_4) remains unresolved in spite of extensive studies.
- *In-Situ* NMR measurements of diffusion & phase transformation at Wash.U. are starting.
- Ti is extremely difficult nuclei for NMR, but ^{45}Sc is excellent candidate & is also good catalyst!
- Obtained initial MAS-NMR spectra [^1H , ^{23}Na , ^{27}Al , & ^{45}Sc] on Sc-doped NaAlH_4 samples.



B. Bogdanovic, Adv. Mater. [2006 on-line]

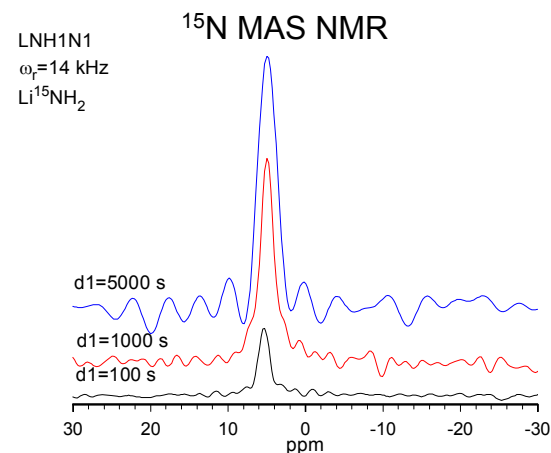
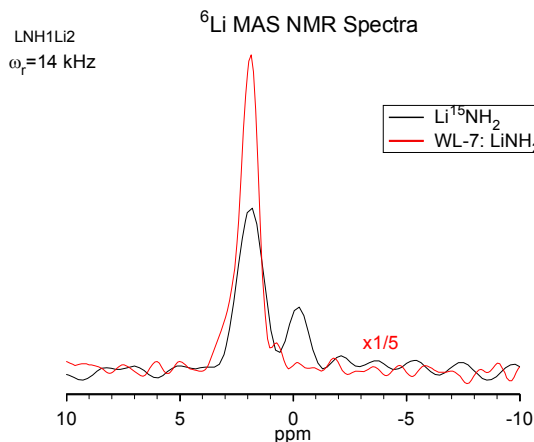
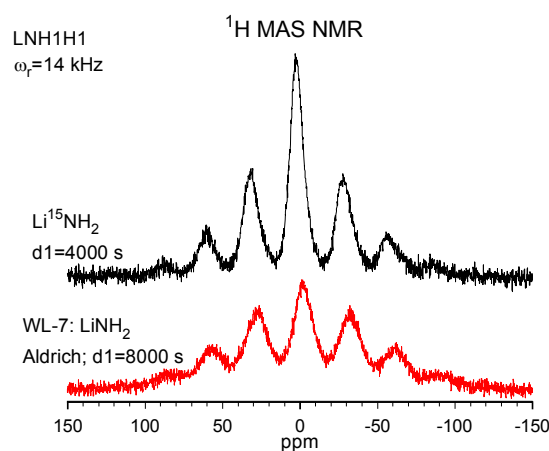


• JPL Objectives:

- Improve understanding on formation, processing, and degradation of amides/imides.
- Provide novel insights on the phase compositions and local chemical bonding parameters for crystalline and highly disordered (i.e., amorphous) phases at various stages of reactions.
- NMR results will test and complement theoretical modeling of mechanisms for phase transformation including assessing role of ammonia on reaction & degradation

• Accomplishments:

- Produce the first ^{15}N -enriched sample of LiNH_2 (see MAS-NMR spectra) to allow ^{15}N MAS-NMR measurements of chemical bonding in various phases of Li-Mg-N-H.
- Filled a reactor with Li-Mg-N-H [SNL] for initial volumetric characterization before starting cycling measurements.



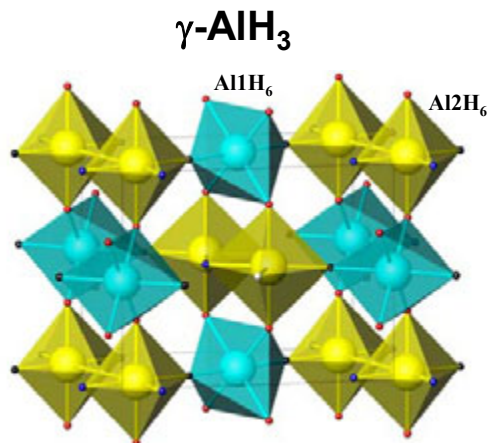
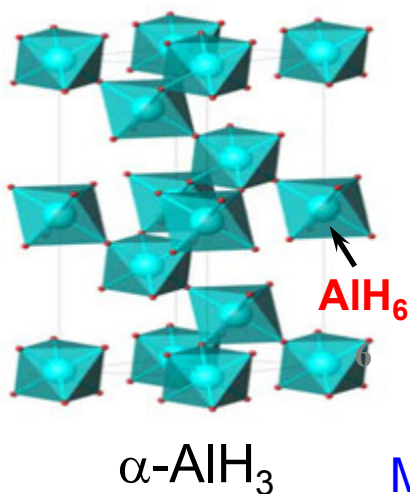
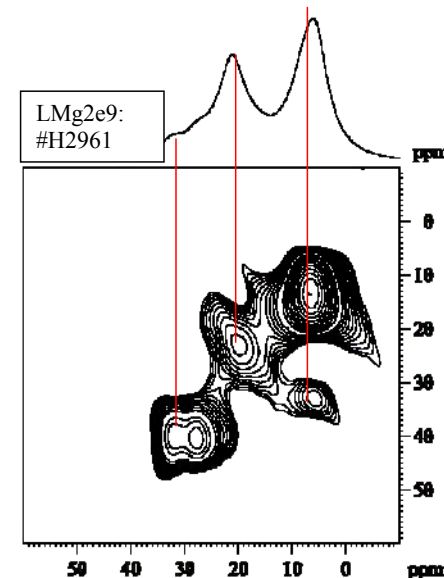
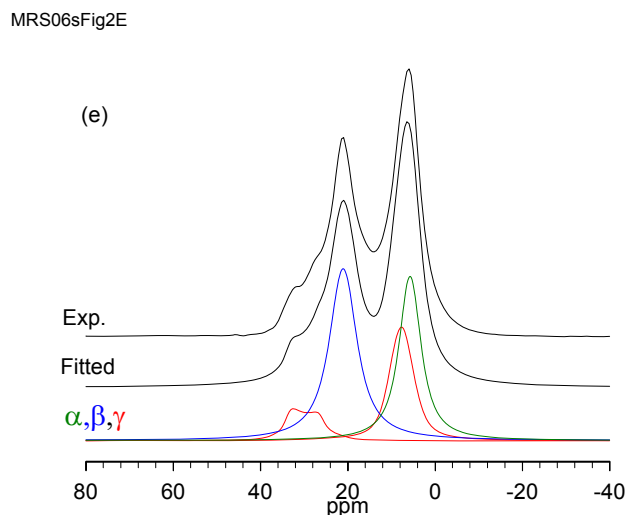
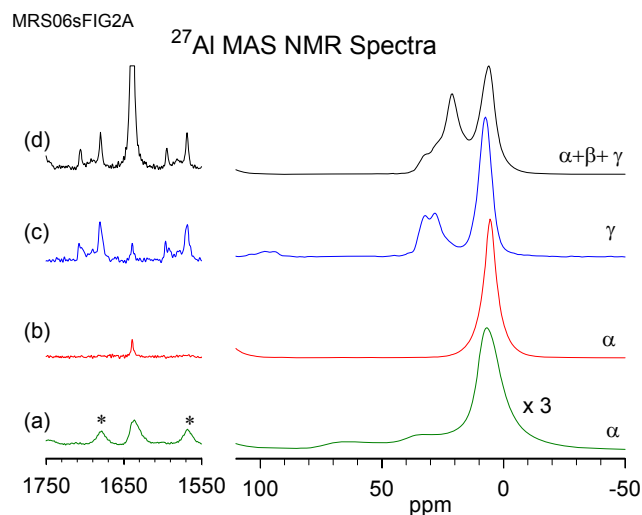
Comment: First ^{15}N -enriched amide sample is mixture of LiH & LiNH_2

•Objectives:

— Use NMR to provide novel insights on the phase compositions, decomposition, and local chemical bonding parameters for crystalline and highly disordered (i.e., amorphous) phases of AlH_3 .

•Accomplishments:

– Obtained MAS-NMR results on AlH_3 samples with α -, β - and γ -phases and monitored Al metal & H_2 gas formation during spontaneous decomposition (BNL)



Conclusions:

- Two types of Al atoms in $\gamma\text{-AlH}_3$ but only one site in α & β phases – will help solve these structures from PND & PXRD
- ^{27}Al NMR shifts consistent with AlH_6 sites will assess differences in bonding to relate to decomposition of these AlH_3 phases

Models by Y. Yartys (IFE) from a TMS paper [March 2006]

Task 5: Engineering Analysis & Design (POC = SRNL) - Prototype Hydride Storage Bed Analysis, Design and Testing

- **JPL Objectives:**

- Support developing lighter weight and thermally efficient hydride storage vessels and experimentally demonstrating their compatibility with appropriate complex and destabilized nanophase hydrides.

- **Status:** Just started these activities in FY-06

- **Accomplishments:**

- Participated in Metal Hydride Systems “kick-off” meeting of the systems working group
- Participated in Storage System Analysis Working Group Meeting held November 2005 in Palm springs, CA
- Participated in the first MHCoe ES&D Project kick-off meeting (March 2006)

- **FY-06 Planned Activities:**

- Evaluate MH System models from ANL “MH-tool” analysis computer model for known hydrides (i.e., $\text{LaNi}_{4.8}\text{Sn}_{0.2}$) and light complex hydride (e.g., alanates, etc.)
- Work on storage vessel conceptual designs, etc. with SRNL, SNL & MHCoe partners.

Task 1. [Destablized Hydrides]

- Complete phase formation & reversibility studies on model LiH-Si, LiH-Ge, Li-B-Mg-H, and Mg-Si-H systems (w/HRL, NIST, Caltech).
 - Go/No-Go on viability of LiH-Si & Mg-Si-H systems by end of FY-06

Task 2. [Complex Anionic Materials]

- Continue the characterization of H bonding in the Ca-Si-H system (w/NIST).
- Pursue possibilities of aiding MHCoe partners and others associated with such systems as Na-Si-H (SNL), $\text{Ca}(\text{BH}_4)_2$ (SNL), catalyzed alanates (IFE), Mg-Ti-H & Mg-Sc-H (Washington U., Philips Research).

Task 3. [Evaluations of Amides/Imides]

- Perform systematic ^{15}N , ^6Li , ^7Li , and ^1H MAS-NMR studies of Li-Mg-N-H phases
- Evaluate accelerated degradation behavior of Li-Mg-N-H samples from SNL.
- Investigate impact of catalysts on reactions kinetics, diffusion, and reversibility for these materials.

Task 4: [Evaluations of Alanes]

- Continue assessments of AlH_3 phases and decomposition processes (BNL, U. Hawaii, IFE-Norway)

FY 2006 Milestones

Due

- Complete an initial phase cycling tests on at least one destabilized hydride system (e.g. $\text{LiBH}_4/\text{MgH}_2$ or Li-Mg-N-H) August, 2006
- Provide a top-level estimate of mass, volume, & thermal parameters for a prototype destabilized/complex hydride storage vessel September, 2006

FY 2007 Milestones

Milestones for Destabilized Hydride Systems:

- Complete initial degradation study on first destabilized systems (Li-Mg-N-H & Li-B-Mg-H)
- Initiate life cycle testing on best destabilized nanophase candidates

Milestone for Prototype Hydride Storage Bed Design & Testing

- Conceptual design for a prototype complex hydride sorbent bed

Milestones for NMR Evaluations of Advanced Complex Hydrides

- Complete 2nd phase NMR studies on Li-Mg-N-H samples (SNL)
- Complete 2nd phase NMR studies on AlH_3 samples (BNL)
- Perform NMR studies on alanates, borohydrides, etc. from MHCoE team



MHCoE Master Schedule & Milestones: Summary of JPL Tasks



Organization	Task #	Task Description	10/1	4/1	10/1	4/1	10/1	4/1	10/1	4/1	10/1	4/1	10/1
			FY05	FY05	FY06	FY06	FY07	FY07	FY08	FY08	FY09	FY09	FY10
A. Destabilized Hydrides													
	1	Destabilization Strategies and New Systems	[Blue bar with red triangles at 10/1 FY05, 4/1 FY06, 10/1 FY07, 4/1 FY08, 10/1 FY09]										
JPL	1.1.1	Conduct cycling studies on Li-Mg-Ni-H and Li-B-Mg-H systems	[Blue bar with yellow diamond at 4/1 FY08]										
JPL	1.1.2	Use MAS-NMR to measure phases in mixtures of LiBH ₄ with various hydrides (e.g. MgH ₂ , LiNH ₂); extend to other candidate materials	[Blue bar with yellow diamonds at 10/1 FY06 and 4/1 FY08]										
JPL (Lead)	2.2	Phase formation and reversibility studies	[Blue bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08]										
JPL	2.2.1	Complete phase formation and reversibility studies on model LiH-Si, LiH-Ge, Li-B-Mg-H, and Mg-Si-H systems	[Blue bar with red triangles and yellow diamonds for LiH-Si, LiH-Ge, Li-B-Mg-H, Mg-Si-H]										
B. Complex Anionic Materials (Borohydrides & Alanates)													
	2	New Hydrogen Storage Materials	[Orange bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08, 4/1 FY09]										
	2.1	Synthesis of new materials in the ternary Si-system	[Orange bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08]										
	2.2	Characterization of the new materials	[Orange bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08]										
JPL	2.2.1	NMR analysis of samples in the Na-Si-H system provided by SNL	[Orange bar with yellow diamond at 10/1 FY06]										
C. Amide/Imides (M-N-H Systems)													
	1	Improve current system	[Purple bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08, 4/1 FY09]										
	2	Understand reaction mechanism	[Purple bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08]										
JPL	2.2	NMR study the nature of N-H bond	[Purple bar with yellow diamond at 4/1 FY08]										
	5	Cycle stability and contamination test	[Purple bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08, 4/1 FY09]										
JPL, SNL	5.1	Cycle stability test	[Purple bar with yellow diamond at 4/1 FY09]										
D. Alanes (AlH₃)													
	1	Aluminum hydride synthesis: Synthesis of alanes that meet 2010 storage targets	[Teal bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08, 4/1 FY09]										
	2	Aluminum hydride properties: Down select alane materials that meet 2010 storage targets	[Teal bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08]										
JPL	2.4	Complete nuclear magnetic resonance (NMR) characterization on selected alanes	[Teal bar with yellow diamond at 10/1 FY06]										
	3	Alane Regeneration: Demonstrate alane regeneration laboratory process	[Teal bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08]										
	4	AlH ₃ tank study: Complete prototype alane tank system meeting 2010 targets	[Teal bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08]										
E. Engineering Analysis & Design													
	1	Component Design & Testing	[Maroon bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08, 4/1 FY09]										
JPL	1.1	Vessel Modeling (Detailed Component Fidelity)	[Maroon bar with yellow diamonds at 4/1 FY07 and 4/1 FY09]										
SNL/JPL	1.2	Materials Compatibility	[Maroon bar with yellow diamonds at 10/1 FY07 and 4/1 FY09]										
JPL	1.3	Vessel Testing (Detailed Component Fidelity)	[Maroon bar with yellow diamonds at 10/1 FY07 and 4/1 FY09]										
	2	Media Engineering Development	[Maroon bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08]										
	3	Safety Strategies	[Maroon bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08]										
All	3.1	Risk Assessment	[Maroon bar with yellow diamonds at 10/1 FY06 and 4/1 FY09]										
	4	Prototype Vessels	[Maroon bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08]										
	5	System Modeling	[Maroon bar with red triangles at 10/1 FY06, 4/1 FY07, 10/1 FY08]										
	▲	Portfolio Relocation Decision Point (Project bars)											
	◆	Milestone (Subtask bars)											
	●	Output (Task bars)											

Back-up Slides (Not presented)

Responses to Previous Year Reviewers' Comments

- NA – This Project was new in FY-05 and was not reviewed.

- **Published Papers**

- R. C. Bowman, Jr., S.-J. Hwang, C. C. Ahn, and J. J. Vajo, “NMR and X-ray Diffraction Studies of Phases in the Destabilized LiH-Si System”, Mater. Res. Soc. Symp. Proc. **837** (2005) paper N3.6.1.

- **Papers Submitted for Publication**

- S.-J. Hwang, R. C. Bowman, Jr., Jason Graetz, and J. J. Reilly, “Solid State NMR Studies of the Aluminum Hydride Phases” Submitted to Proc. 2006 Spring Meeting, Symposium EE.

- **Manuscripts in Preparation or Planned**

- R. C. Bowman, Jr., S.-J. Hwang, C. C. Ahn, A. Dailly, and J. J. Vajo, “Studies of Phase Transformations in the Destabilized LiH-Si System”, J. Mater. Res. [Invited to submit].
- R. C. Bowman, Jr., S.-J. Hwang, C. C. Ahn, M. R. Hartman, T. J. Udovic, J. J. Rush, and J. J. Vajo, “Destabilization Behavior and Phase Compositions for LiH-Ge” In Preparation.
- Son-Jong Hwang, R. C. Bowman, Jr., Jason Graetz and J. J. Reilly, “NMR Studies of the Aluminum Hydride Phases and their Stabilities”, in Preparation.

•Presentations made in 2005/2006

- R.C. Bowman, Jr., S.-J. Hwang, C. C. Ahn, A. Dailly, J. J. Vajo, T. J. Udovic, M. Hartman, and J. J. Rush, “Studies of Thermodynamics and Phases Produced in the Destabilized LiH-Si System” invited talk at the Nordic Energy Research Meeting, Krusenberg, Sweden, 17-18 June 2005.
- R. C. Bowman, Jr., “Development of Metal Hydrides for Space Applications and Energy Storage” invited talk at the Materials Science & Technology 2005 Conference, Pittsburgh, PA, 25-28 September 2005.
- R. C. Bowman, Jr., “Behavior and Properties of Metal Hydrides for Reversible Hydrogen Storage” invited presentation at the WE-Heraeus-Seminar on Hydrogen Storage with Novel Nanomaterials, Bad Honnef, Germany, 24-27 October 2005.
- M. R. Hartman, T. J. Udovic, J. J. Rush, R. C. Bowman, Jr., J. J. Vajo, and C. C. Ahn, “Neutron Scattering Investigations of a Destabilized LiH:Si System for Hydrogen Storage Applications” 2005 Fall MRS Meeting, Boston, MA, 1 December, 2005.
- R. C. Bowman, Jr., S.-J. Hwang, C. C. Ahn, A. Dailly, M. R. Hartman, T. J. Udovic, J. J. Rush, and J. J. Vajo, “Reversibility and Phase Compositions of Destabilized Hydrides Formed from LiH”, invited presentation at the Spring 2006 MRS Meeting, 17-20 April, 2006.
- Son-Jong Hwang, R. C. Bowman, Jr., Jason Graetz and J. J. Reilly, “NMR Studies of the Aluminum Hydride Phases and their Stabilities”, presented at the Spring 2006 MRS Meeting, 17-20 April, 2006.
- M. R. Hartman, T. J. Udovic, J. J. Rush, R. C. Bowman, Jr., J. J. Vajo, and C. C. Ahn, “Neutron Scattering Investigations of a Destabilized LiH:Si System for Hydrogen Storage Applications” presented at the Spring 2006 MRS Meeting, 17-20 April, 2006.
- R. C. Bowman, Jr., “Current Status in Developing Advanced Metal Hydrides for Reversible Hydrogen Storage”, Invited presentation at 209th Meeting of Electrochemical Society, Denver, CO, May 10, 2006.

•Presentations planned for remainder FY-06

- R. C. Bowman, Jr., J. G. Kulleck, S.-J. Hwang, M. R. Hartman, T. J. Udovic, and J. J. Rush, “Characterization of Phase Compositions and Structures for Metal Hydrides Used in Hydrogen Storage” invited presentation at Amer. Crystallographic Soc., Honolulu, HI, 22 July, 2006.