**Innovation for Our Energy Future** 

# Photoelectrochemical Water Systems for H<sub>2</sub> Production

2007 DOE Hydrogen, Fuel Cells, and Infrastructure Technologies Program Review

May 17, 2007

John A. Turner, Todd Deutsch, Jeff Head, and Paul Vallett
National Renewable Energy Laboratory

jturner@nrel.gov 303-275-4270

This presentation does not contain any proprietary or confidential information.

## **Overview**

#### **Timeline**

- Project start date: 1991
- Project end date: tbd
- Percent complete: tbd

## **Budget**

- Total project funding to date
  - DOE share: \$5.9M (~0.75 FTE + postdoc, average)
- Funding received in FY 2006: \$140k
- Funding for FY 2007: \$800k

#### **Barriers**

#### **Barriers addressed**

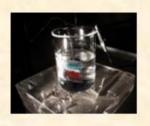
- ✓ M. Materials durability
- ✓ O. Materials efficiency
- ✓ N. Device configuration designs

### **Partners**

#### Interactions/collaborations

- UNLV-SHGR
- University of Nevada, Reno
- Colorado School of Mines
- University of Colorado
- Program production solicitation
  - MVSystems, Inc
  - Midwest Optoelectronics











## Photoelectrochemical Hydrogen Production: **UNLV-SHGR Program Subtask**

Robert Perret UNLVRF

15 May 2007

#PDP37

This presentation does not contain any proprietary or confidential information



## **Photoelectrochemical Conversion**

### **Goals and Objectives**

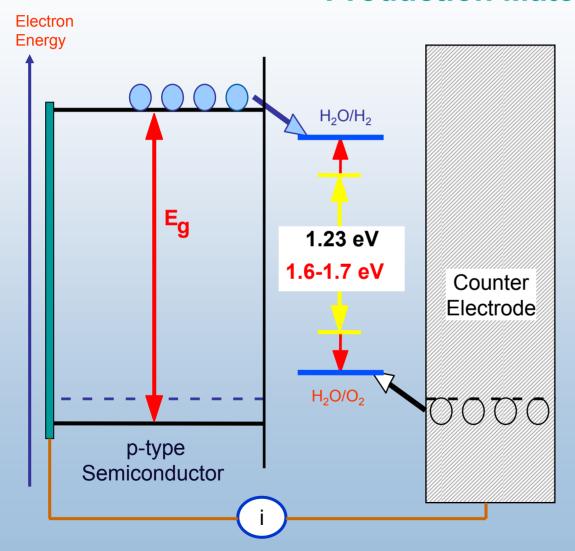
The goal of this research is to develop a stable, cost effective, photoelectrochemical-based system that will split water using sunlight as the only energy input. Our objectives are:

- 1. Identify and characterize <u>new semiconductor materials</u> that have appropriate bandgaps and are stable in aqueous solutions.
- 2. Study <u>multijunction semiconductor systems</u> for higher efficiency water splitting.
- 3. Develop techniques for the <u>energetic control</u> of the semiconductor electrolyte interface and for the preparation of transparent <u>catalytic</u> <u>coatings</u> and their application to semiconductor surfaces.
- 4. Identify environmental factors (e.g., pH, ionic strength, solution composition, etc.) that affect the energetics of the semiconductor, the properties of the catalysts, and the stability of the semiconductor.
- 5. Develop <u>database</u> to house a library of the material properties discovered by the DOE program.



## Material Challenges (the big three)

Characteristics for Ideal Photoelectrochemical Hydrogen Production Material



- Efficiency band gap (E<sub>g</sub>) must be at least 1.6-1.7 eV, but not over 2.2 eV; must have high photon to electron conversion efficiency
- Material Durability semiconductor must be stable in aqueous solution
- ➤ Energetics band edges must straddle H<sub>2</sub>O redox potentials (Grand Challenge)

All must be satisfied simultaneously.

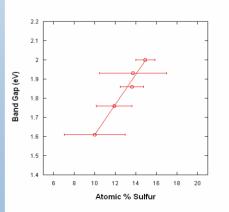


# Approach: High Efficiency Materials & Low-Cost Manufacturing

PEC devices must have the same internal photon-to-electron conversion efficiency as commercial PV devices.

- III-V materials have the highest solar conversion efficiency of any semiconductor material
  - Large range of available bandgaps
  - ....but
    - Stability an issue nitrides show promise for increased lifetime
    - Band-edge mismatch with known materials tandems an answer
- I-III-VI materials offer high photon conversion efficiency and possible low-cost manufacturing
  - Synthesis procedures for desired bandgap unknown
  - ....but
    - Stability in aqueous solution?
    - Band-edge mismatch?
- Other thin-film materials with good characteristics
  - SiC: low-cost synthesis, stability
  - SiN: emerging material





# **Approach: Materials Summary**

The primary task is to synthesize the semiconducting material or the semiconductor structure with the necessary properties. This involves material research issues (material discovery), multi-layer design and fabrication, and surface chemistry. Activities are divided into the task areas below – focus areas in black:

- GaPN NREL (high efficiency, stability)
- CuInGa(Se,S)<sub>2</sub> UNAM (Mexico), NREL (low cost)
- Silicon Nitride NREL (protective coating, new material)
- GaInP<sub>2</sub> NREL (fundamental materials understanding)
- Energetics
  - Band edge control
  - Catalysis
  - Surface studies

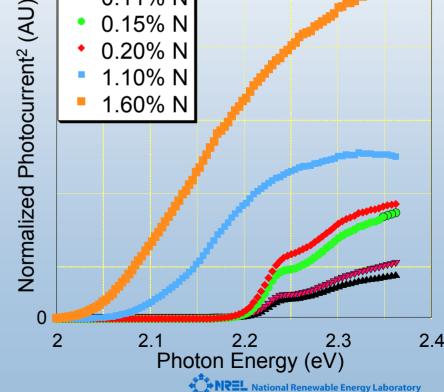


# III-V Nitrides: GaP<sub>1-x</sub>N<sub>x</sub>

Previous work with III-V nitrides for PEC water splitting showed improved stability with very small amounts of nitrogen, but problems with morphology at higher nitrogen content

- $GaP_{1-x}N_x$ 
  - Addition of small amounts of N causes GaP band gap to narrow (bowing) and transition to become direct
  - Nitrogen enhances stability





GaP<sub>1-x</sub>N<sub>x</sub> Epilayer Direct Band Gap

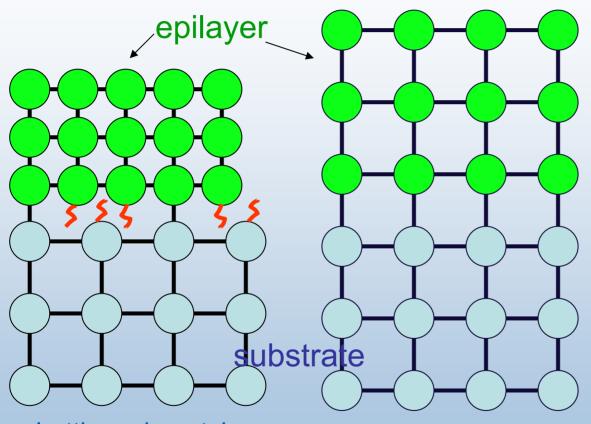
0.06% N 0.11% N



Dr. Todd Deutsch

## **Increasing the Nitrogen Content:**

**Lattice Matching Using Indium as a Lattice Expander** 



Lattice mismatch from nitrogen addition

Lattice matched

Results: Latticed matched with more than 2% nitrogen

 $Ga_{.96}In_{.04}N_{.024}P_{.976}$ 

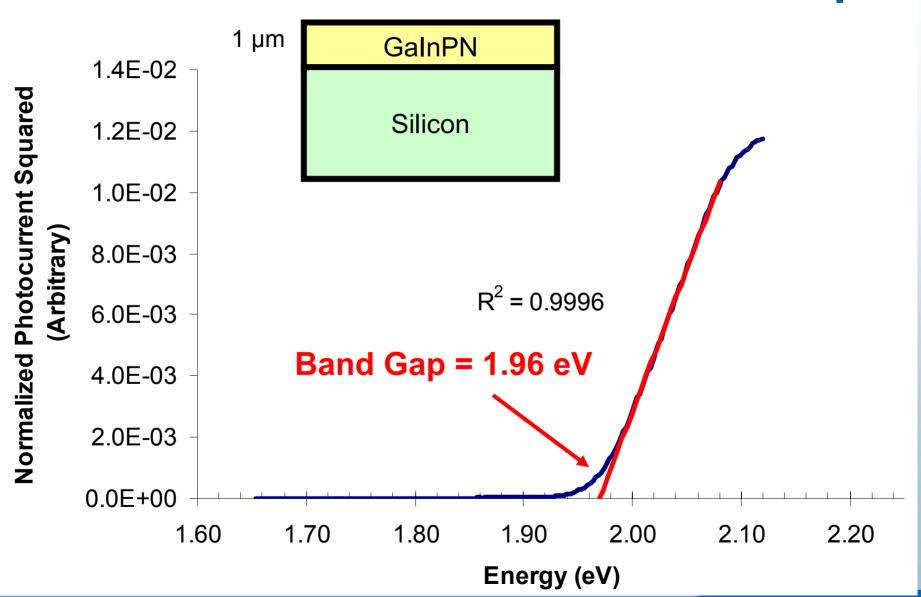
 $Ga_{.95}In_{.05}N_{.025}P_{.975}$ 



Jeff Head, Paul Vallett

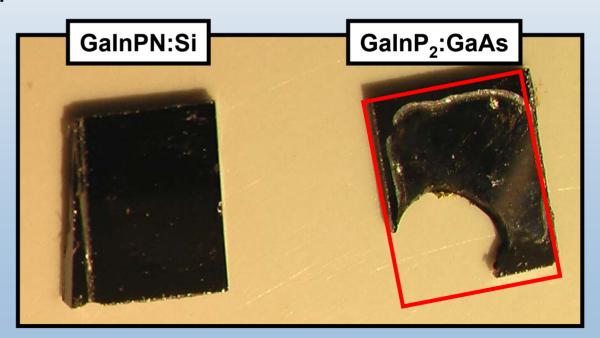
NREL National Renewable Energy Laboratory

# Photocurrent Spectroscopy To Determine the GalnPN Band Gap



# Durability Analysis Showed Improved Corrosion Resistance

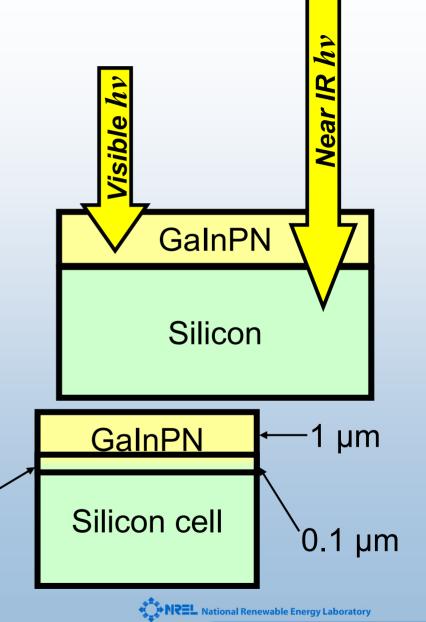
- 24-hour corrosion test
  - Constant 5 mA/cm<sup>2</sup> applied current
  - Sample illuminated at 1 sun
- Profilometry etch depth
  - 0.1 µm average



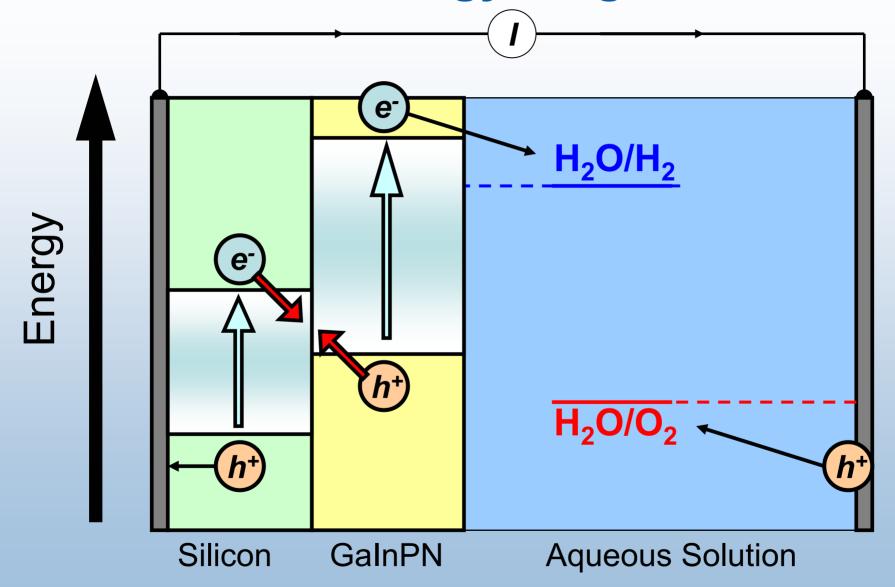
# **GalnPN on Silicon Tandem Cell**

- GaInPN top cell
  - Visible light conversion
  - Corrosion protection
- Si-bottom cell
  - Absorbs IR radiation passing through the top cell
  - Provides energy to lower h<sup>+</sup>
     below O<sub>2</sub> redox potential
- Also applied back surface field for field-aided charge collection

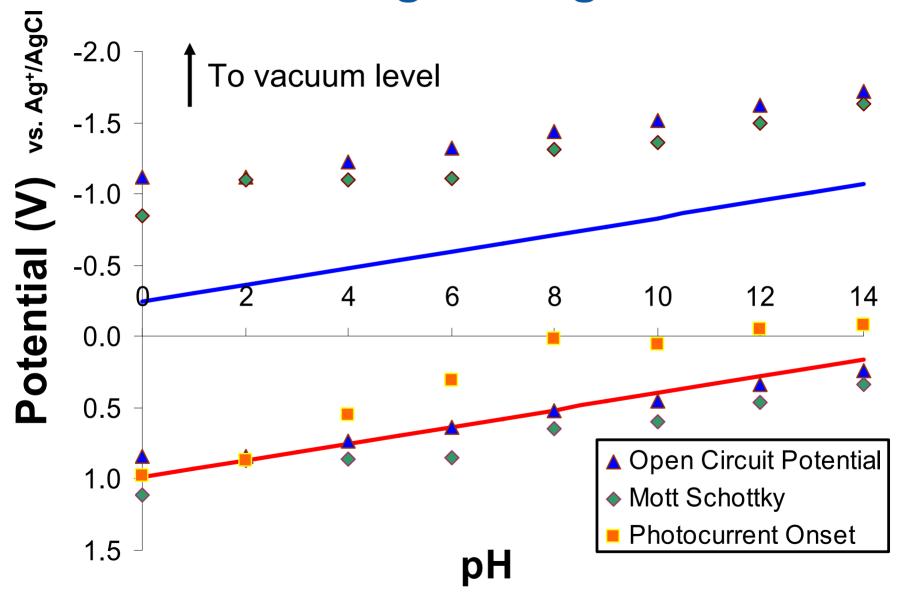
Zn-doped GaInPN



# **PEC Energy Diagram**



## **Band Edge Energetics**



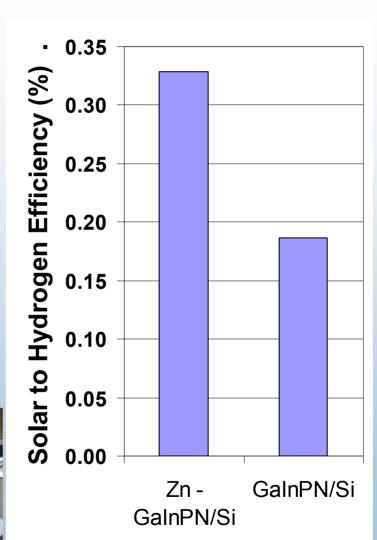
## Water Splitting

- Zn-doped layer provides back surface field and improves efficiency
- Nitrogen negatively impacts the electronic properties of material
- Photon to chemical energy conversion efficiency
  - GaInPN: ~25%
  - GalnP<sub>2</sub>: ~60%

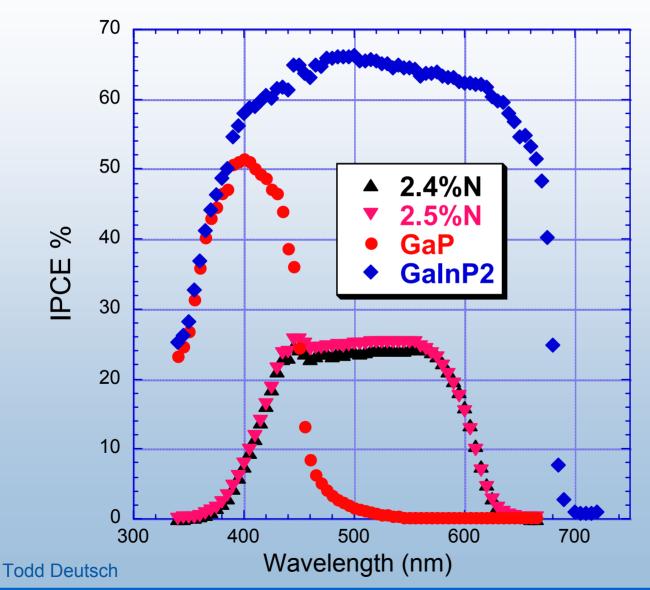




Jeff Head, Paul Vallett



# Integrated Photon-to-Electron Current Efficiency (IPCE)

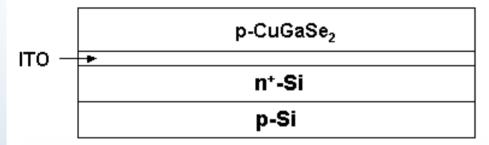


PEC devices must have the same internal photon-to-electron conversion efficiency as commercial PV devices.

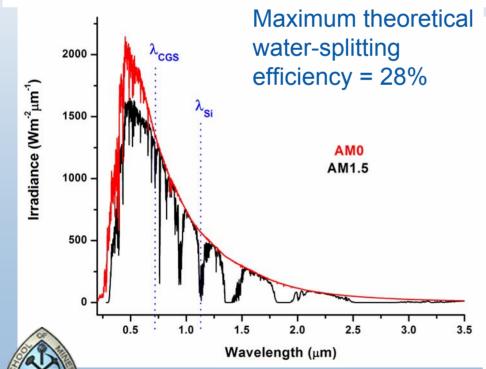


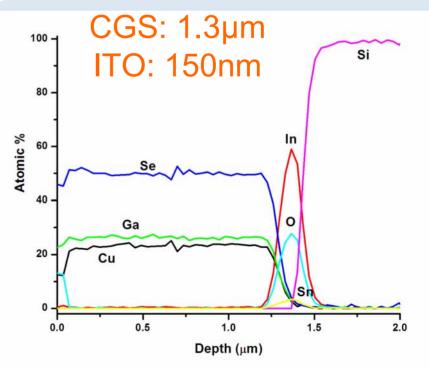
## **CuGaSe<sub>2</sub> Tandem Cell Configuration:**

Possible High Efficiency, But a New Deposition Approach is Required







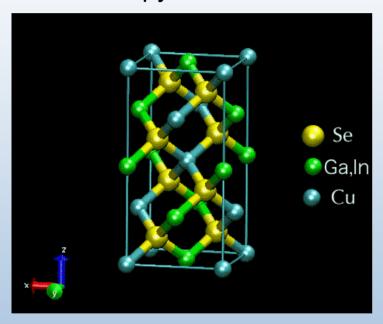


Dr. Jennifer E. Leisch

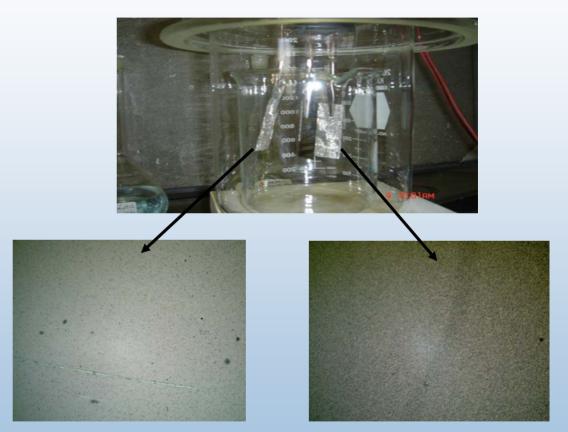
### **Electrodeposited CGS - Deposition of Ga Difficult**

### Deposition Bath Compositions Have Been Developed To Produce Good Quality Films

#### Chalcopyrite Structure



CuGaSe<sub>2</sub> (CGS)



Fair quality electrodeposited CGS film on Mo substrate

Good quality electrodeposited CIGS film on Mo substrate



## **Conclusions**

#### III-V nitrides

- Higher nitrogen content realized with In as lattice expander
- PEC water splitting demonstrated on GaInPN:Si tandem cell
- Nitrogen enhanced stability, but nitrides grown by MOCVD have poor performance
- p+ layer field aided charge collection, increased efficiency
- Nitride epilayer needs improvement

#### CIGSSe thin films

- Theoretical 23% water-splitting efficiency with Si-CuGaSe<sub>2</sub>
   tandem cell, but configuration difficult to realize
- Possible to electrodeposit good quality films with high Ga
- Annealing system is being constructed and trials will soon be underway



### **Future Plans**

#### Remainder of FY 2007:

- Continue understanding and improving nitride-based material: III-V nitrides and SiN
- Develop new electrosynthesis approaches and lowtemperature annealing processes for CIGSSe films
- Improve tandem cell design with thin-film CIGSSe

#### • For FY 2008:

- Look at possible new materials with UNLV, CSM ...
- Coatings: SiN, SiC ...
- Multijunction structures