

High Purity Hydrogen from Coal-Derived Syngas

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Overview

Timeline

- Start: June 2005
- End: May 2008
- 75% Complete

Budget

- Total project funding
 - DOE: \$498k
 - Contractor: \$125k
- Funding received in FY06
 - \$231k
- Funding for FY07
 - \$201k

Barriers

- Hydrogen from Coal – Research, Development & Demonstration Program
 - WGS Reaction Barriers
 - D. Impurity Intolerance/Catalyst Durability
 - Hydrogen Separation Barriers
 - I. Poisoning of Catalytic Surfaces
 - Q. Impurities in Hydrogen from Coal



Overall Goal

Develop water gas shift (WGS) membrane reactor that cost-effectively produce pure hydrogen from coal-derived synthesis gas containing significant amounts of hydrogen sulfide and hydrogen chloride

- Replaces multiple process units (contaminant removal, high temperature shift, low temperature shift, pressure swing adsorption) with single catalytic membrane reactor
- High process efficiency realized by improving hydrogen yield at low steam to carbon ratios
- Produces high-pressure CO₂-rich stream potentially suitable for sequestration



Objectives

- Develop sulfur- and chloride-tolerant, highly active WGS catalyst that is able to operate at low steam to carbon ratios
- Develop low-cost, contaminant-tolerant H₂-selective membrane
- Demonstrate an integrated WGS membrane reactor operating on simulated coal-derived syngas
- Analyze cost benefits of the demonstrated WGS membrane reactor technology



Approach

Develop water gas shift catalyst and H₂-selective membrane that both operate at 300-500°C, at 300-500 psig and are tolerant of high concentrations of H₂S and HCl

Catalyst

- MoS₂-based catalyst with modifiers to promote impurity tolerance

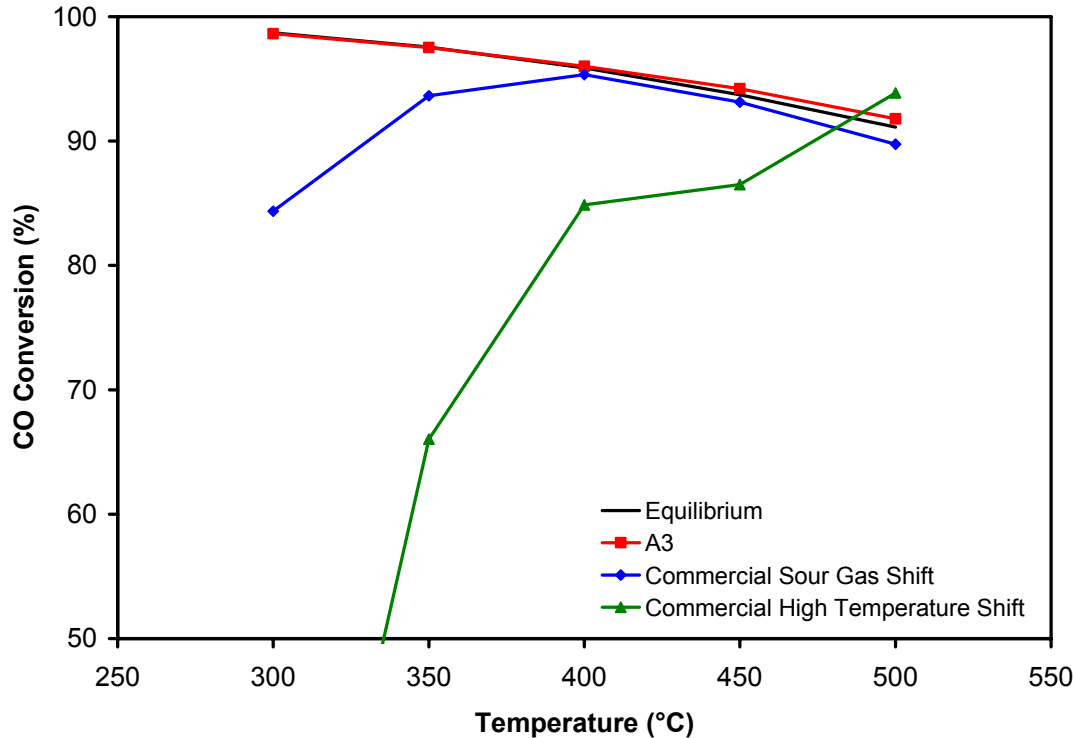
Membrane

- Dense Ta-based membrane with surface modifications to promote H₂ dissociation/association reactions in the presence of S and Cl



WGS Catalyst Development

60% CO, 25% H₂, 10% CO₂, 5% N₂, 3000 ppm H₂S, H₂O/CO=4, 400 psig, GHSV=3000 h⁻¹

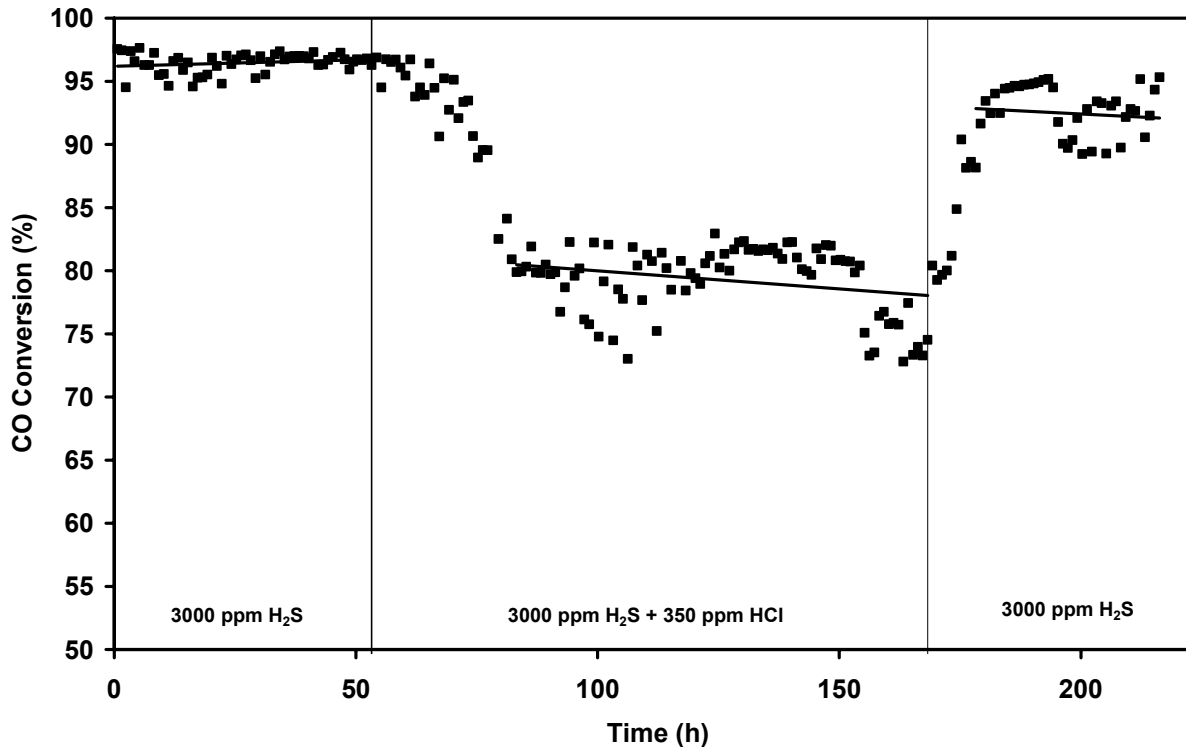


- Equilibrium yields realized with catalyst A3 at 300-500°C
- Tolerant to 3000 ppm_v (dry gas) H₂S



WGS Catalyst Stability

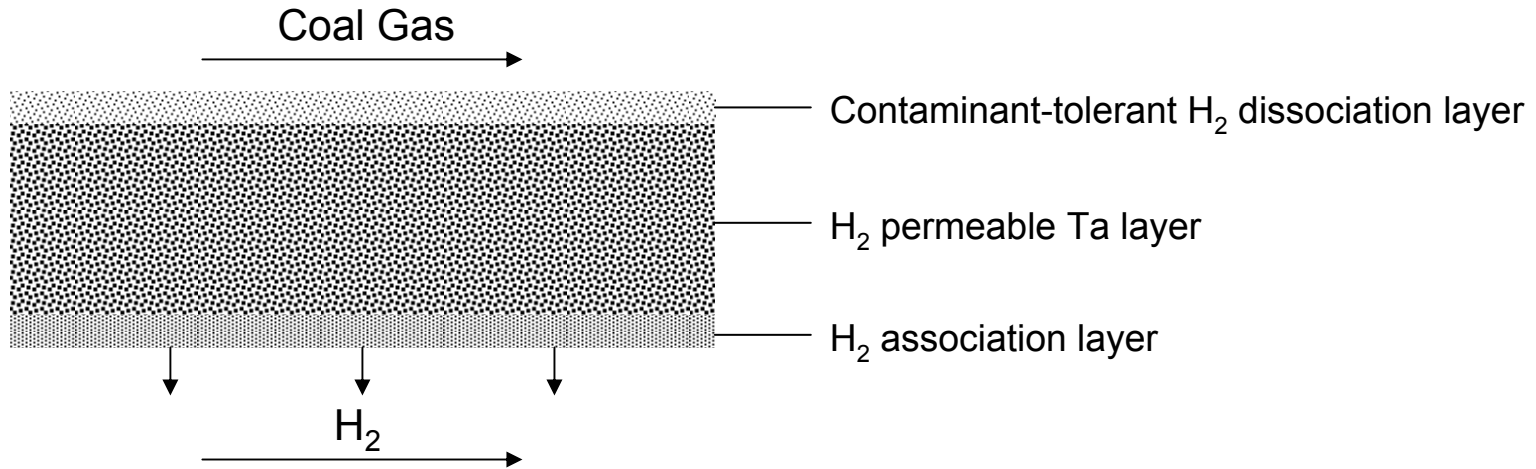
Catalyst A3, 60% CO, 25% H₂, 10% CO₂, 5% N₂, H₂O/CO=4, 400°C, 400 psig, GHSV=3000 h⁻¹



- Stable in the presence of 3000 ppm_v (dry gas) H₂S
- Mild reversible deactivation with 350 ppm_v (dry gas) HCl



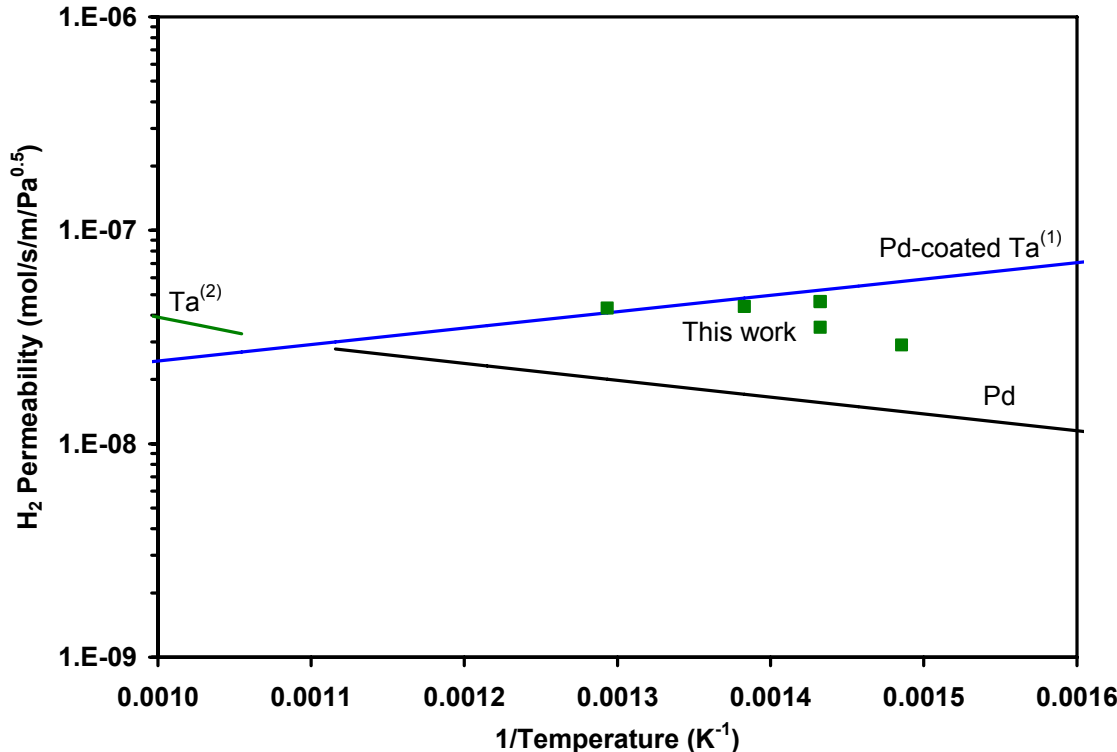
Membrane Design



- Developed substrate cleaning and coating technique
- Examined effects of substrate thickness, coating composition, coating thickness and annealing conditions on pure H₂ permeability
- Examined effects of H₂O, H₂S, CO, and CO₂ on membrane performance



Membrane H₂ Permeability

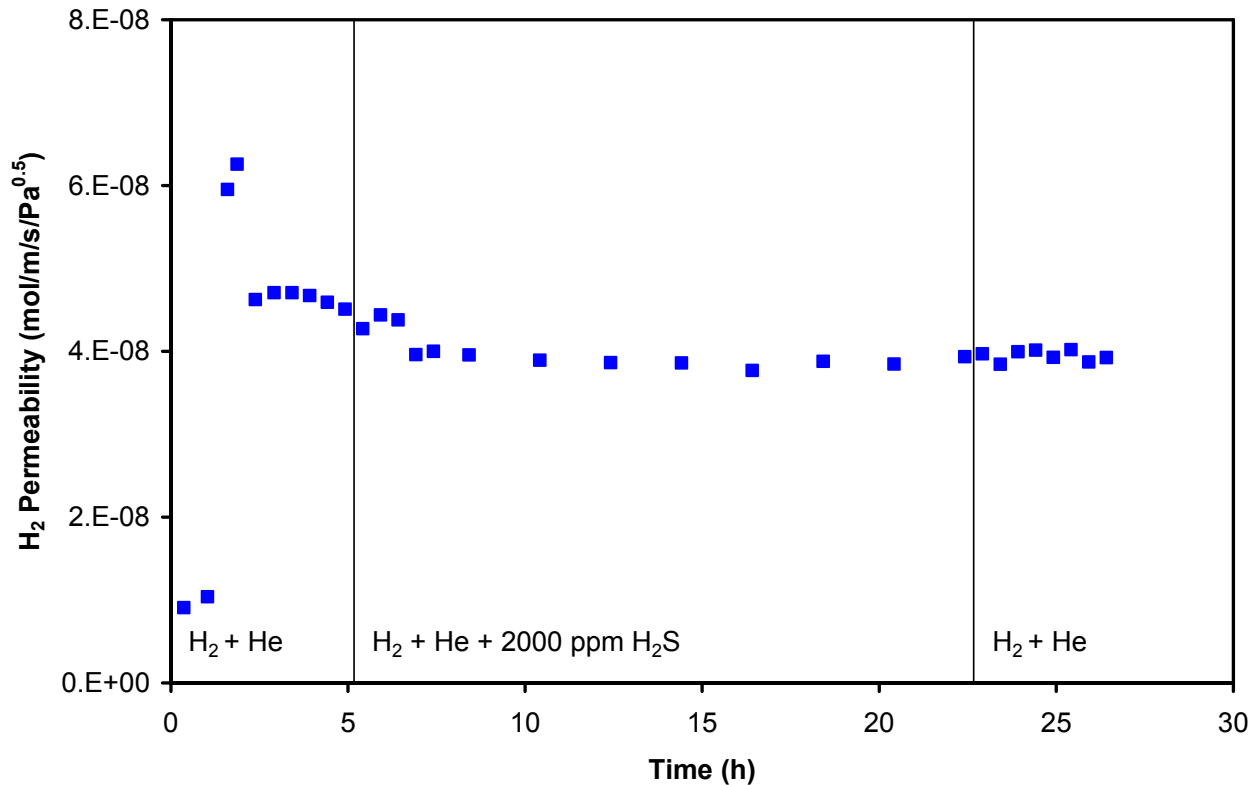


Test Conditions
95% H₂ – 5% He
350-550°C
200 psig H₂

- H₂ permeability greater than dense Pd-based membranes
- High selectivity → H₂ purity >99.9%
- Good mechanical strength - tested up to 550°C & 400 psi



Membrane Sulfur Tolerance



Test Conditions

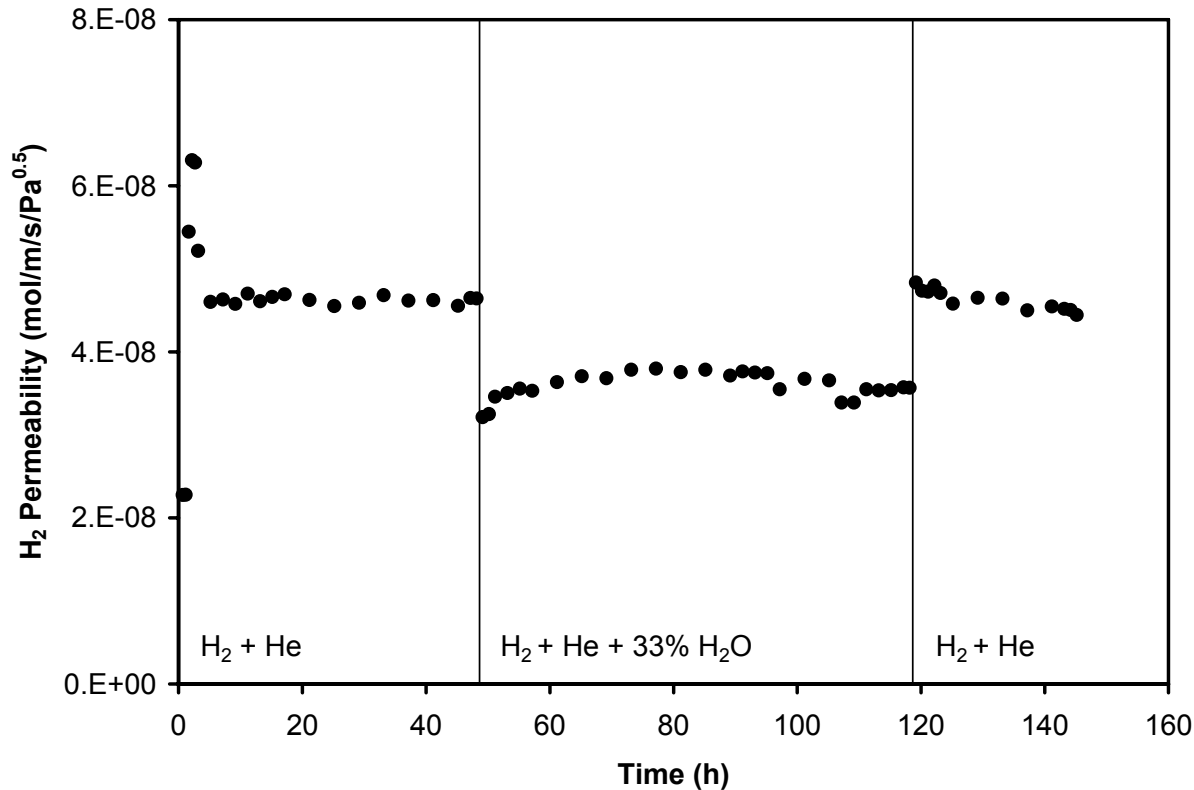
425°C

150 psig H₂

- Membrane tolerant of high concentration of H₂S
- 14% reduction in permeability upon exposure to 2000 ppm H₂S



Membrane Moisture Tolerance



Test Conditions

425°C

150 psig H₂

- Membrane tolerant of high concentration of steam
- 22% reduction in permeability upon exposure to 33% H₂O



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Future Work

FY 2007

- Quantify performance and stability of membrane in contaminant-laden syngas
- Examine membrane microstructure in order to relate permeability to structure and synthetic parameters
- Fabricate and demonstrate integrated water gas shift membrane reactor

FY 2008

- Conduct cost analysis



Summary

Water Gas Shift

- Contaminant-tolerant water gas shift catalyst development and testing completed
- Equilibrium yields demonstrated at practical operating conditions
- Catalyst maintains high activity over wide temperature range (300-500°C)
- Excellent catalyst durability in the presence of 3000 ppm H₂S and 350 ppm HCl observed over 220 hours

H₂-Permeable Membrane

- Dense metal membranes with high H₂ permeability and H₂ selectivity demonstrated
- Membrane H₂ permeability greater than that of Pd-based membranes
- Membrane maintains high permeability in the presence of 2000 ppm H₂S and 33% H₂O



Hydrogen From Coal Program Goals

Water Gas Shift Catalyst

Performance Criteria	Current Status	Targets		This Work
		2010	2015	
Catalyst Form	Pellets	Advanced configurations – TBD		Pellets
Active Metal	Cu/Zn or Fe/Cr or Co/Mo	Advanced configurations – TBD		Mo
Feed Temperature (°C)	200-300	>250	>400	>250
Feed Pressure (psia)	450-1150	>450	>750	415
Approach to Equilibrium (°C)	8-10	<6	<4	<4
Min. H ₂ O/CO Ratio	2.6	<2.6	<2	<2
Sulfur Tolerance (ppm _v)	varies	>20	>100	3000
COS Conversion	varies	Partial	Total	-
Chloride Tolerance (ppm _v)	varies	>3	>100	350
Durability (y)	3-7	>7	>10	-
Catalyst Cost (\$/lb)	~5	<5	<5	-

From “2006 Hydrogen from Coal Program RD&D Plan”

Hydrogen From Coal Program Goals

Hydrogen Separation

Performance Criteria	Current Status			Targets			This Work
	Microporous	Cermet	Dense Ceramic	2007	2010	2015	
Flux (scfh/ft ²)	20-100	~300	2	100	200	300	20
Temperature (°C)	300-600	300-400	900	400-700	300-600	250-500	350-550
S Tolerance (ppm _v)	Yes	~20	-	-	20	>100	2000
Cost (\$/ft ²)	150-200	<200	-	150	100	<100	-
WGS Activity	No	-	-	Yes	Yes	Yes	Yes
ΔP Capability (psig)	100	1000	-	100	400	800	400
CO Tolerance	Yes	Yes	-	Yes	Yes	Yes	-
H ₂ Purity (%)	90-98	>99.999	-	95	99.5	99.99	>99.9
Durability (y)	-	0.9	-	1	3	5	-

From “2005 Hydrogen from Coal Program RD&D Plan” & “2006 Hydrogen from Coal Program RD&D Plan”



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