



U.S. DEPARTMENT OF  
**ENERGY**

# **Fuel Cell Subprogram Overview**

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**2007 DOE Hydrogen Program  
Merit Review and Peer Evaluation Meeting**

**May 15, 2007**



# Outline

- Goal and Objectives
- Budget
- Challenges
- Progress
  - Accomplishments/Status
- Future Plans



# Fuel Cell Subprogram

## Goal

Develop and demonstrate fuel cell power system technologies for transportation, stationary, and portable power applications.

## Objectives

For transportation applications:

- By 2010, develop a 60% peak-efficient, direct hydrogen fuel cell power system at a cost of \$45/kW with 5000 hours of durability (80°C); by 2015, a cost of \$30/kW.

For stationary power and other early market fuel cell applications:

- By 2011, develop a distributed generation PEM fuel cell system operating on natural gas or LPG that achieves 40% electrical efficiency and 40,000 hours durability at \$750/kW.
- By 2010, develop a fuel cell system for consumer electronics (<50 W) with an energy density of 1,000 Wh/L.
- By 2010, develop a fuel cell system for auxiliary power units (3-30 kW) with a specific power of 100 W/kg and a power density of 100 W/L.



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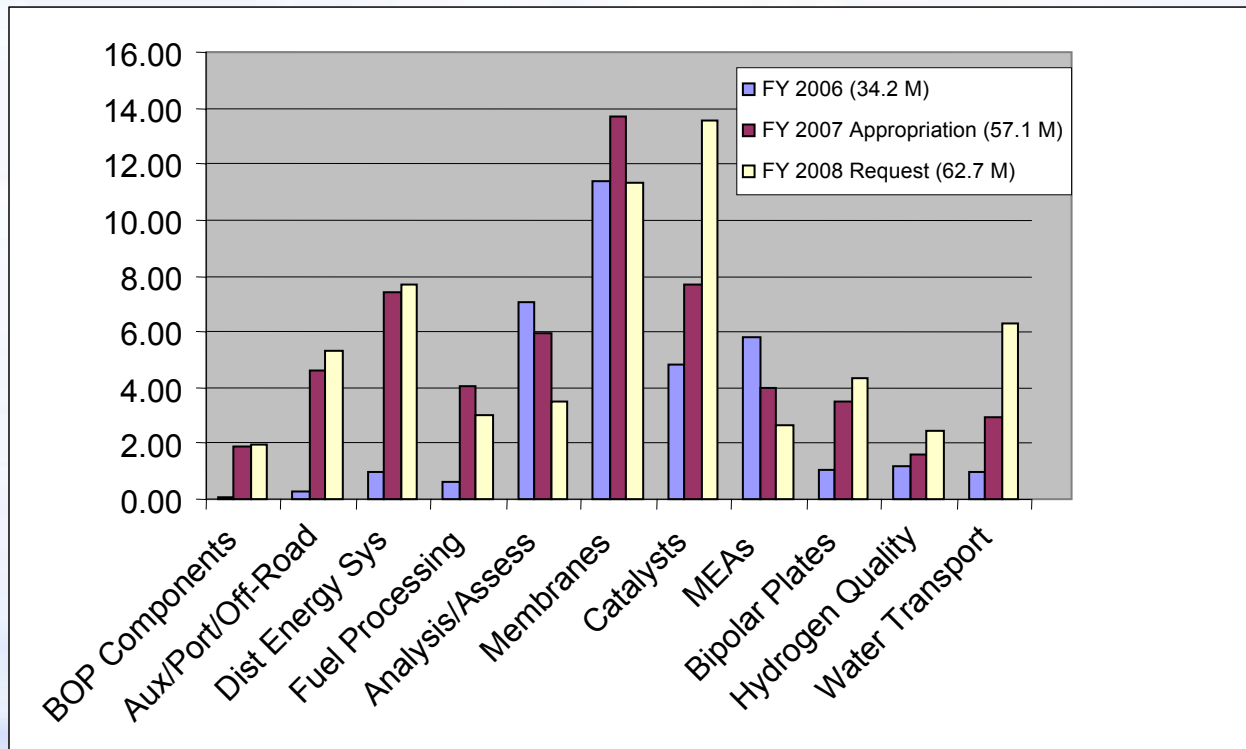


# Fuel Cell Budget

Budget Activity	Funding (\$ in thousands)	
	FY 2007 Appropriation	FY 2008 Request
Fuel Cell Stack Component R&D	38,082	44,000
Transportation Fuel Cell Systems	7,518	8,000
Distributed Energy Fuel Cell Systems	7,419	7,700
Fuel Processor R&D	4,056	3,000



# Fuel Cell Budget by topic





# Outline

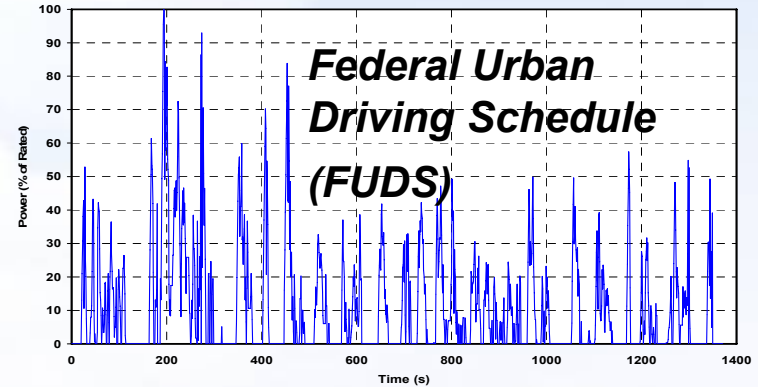
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# Fuel Cell Challenges

- A. Durability
- B. Cost
- C. Performance
- D. Air Management
- E. System Thermal and Water Management
- F. Water Transport within the Stack
- G. Start-up and Shut-down Time-and-Energy/Transient Operation







# Outline

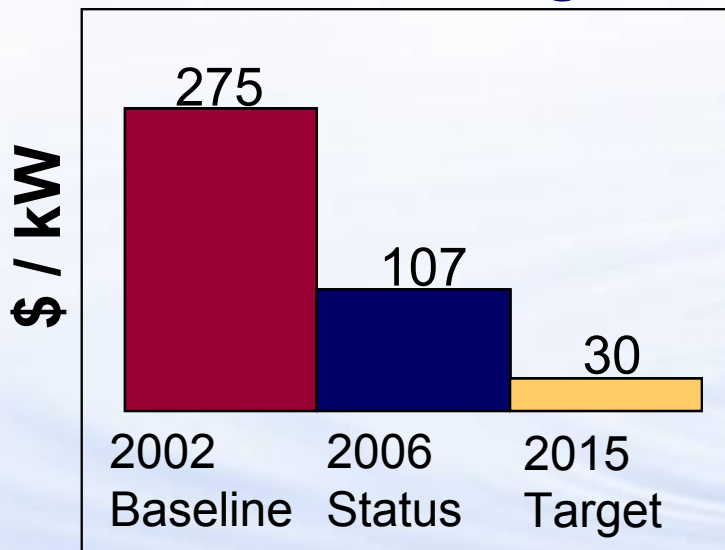
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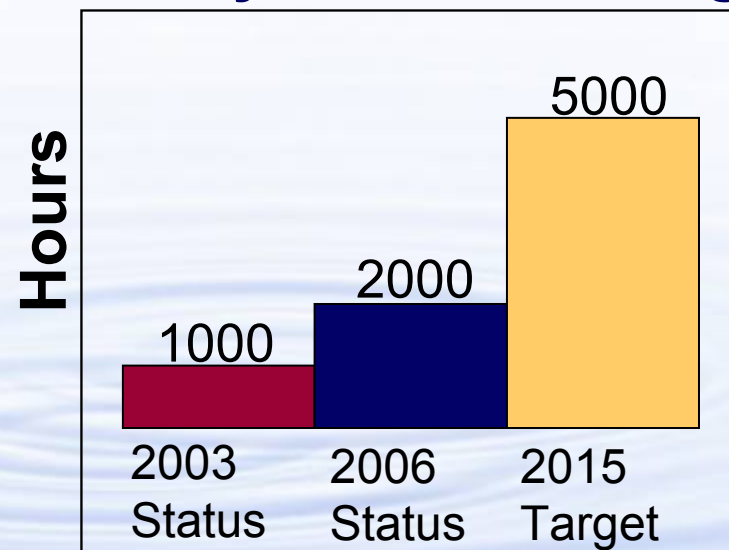
# Fuel Cells Progress

Program has reduced cost and improved durability.

## Fuel Cell System Costs Status vs. Targets



## Fuel Cell System Durability Status vs. Targets





# Major Project - Fuel Cell Membranes

## Importance to goals:

- Fuel cell stack performance and durability depend on membrane properties
- Membranes are a significant contributor to the fuel cell cost
- Membrane limitations add complexity to the fuel cell system

## Three Strategies for High-Temperature Membrane Research:

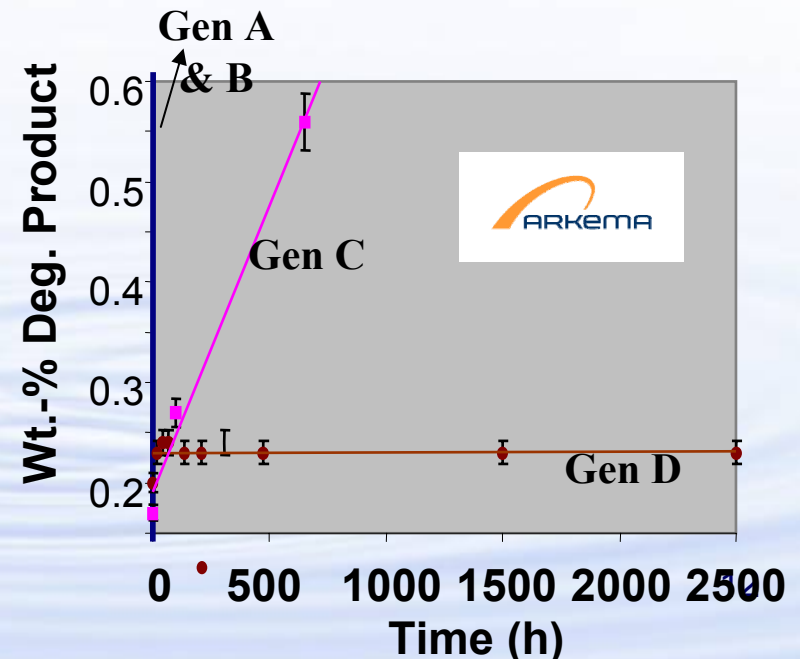
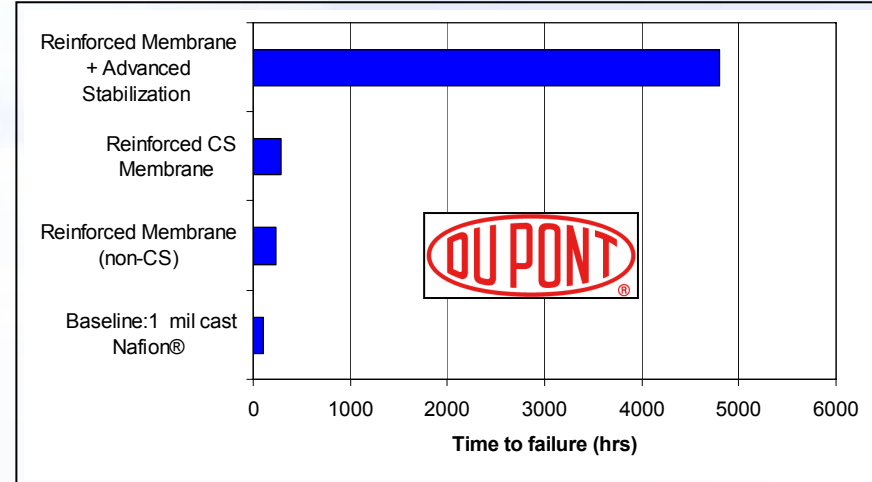
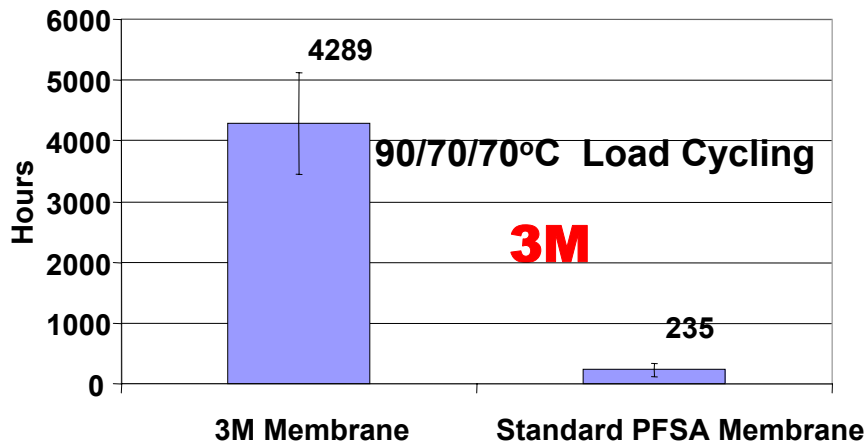
- **Strategy 1 – Phase segregation control** (*polymer & membrane*)  
*Polymer* - Separate blocks of hydrophobic and hydrophilic functionality incorporated within the same polymer molecule (CWRU, Clemson, GE, Penn State, University of Tennessee, Virginia Tech, University of Central Florida).
- *Membrane* – Two-polymer composites. One polymer provides mechanical strength while the other polymer enables proton conduction (Arkema, CWRU, Fuel Cell Energy, Giner, 3M).
- **Strategy 2 – Non-aqueous proton conductors**  
Membranes that use inorganic oxides, heteropolyacids, or ionic liquids, rather than water, for proton conductivity (Arizona State, Colorado School of Mines, Penn State, LBNL).
- **Strategy 3 – Hydrophilic additives**  
Membranes with additives that maintain water content and conductivity at higher temperature (CWRU, Fuel Cell Energy, GE, University of Central Florida, 3M).



# Results: R&D Highlights - Membranes

## Membrane Durability

- Developed membrane with nearly 5,000 hours (DOE target) durability with humidity and voltage cycling (DuPont)
- Sulfur loss issue resolved in PVDF composite membrane (Arkema)
- Initial fluoride release in voltage cycle testing correlated to accelerated lifetime (3M)
- DOE Accelerated Stress Test protocols developed for membranes/MEAs





# Major Project - Catalysts & Supports

## Importance to objectives:

- Platinum cost is ~80% of total stack cost
- Platinum availability is an issue at high loading
- Catalyst durability needs improvement

## Four Strategies for Catalysts & Supports R&D:

### Strategy 1 – Lower PGM content

Improved Pt catalyst utilization along with durability (Brookhaven, ANL, LANL)

### Strategy 2 – Pt alloys

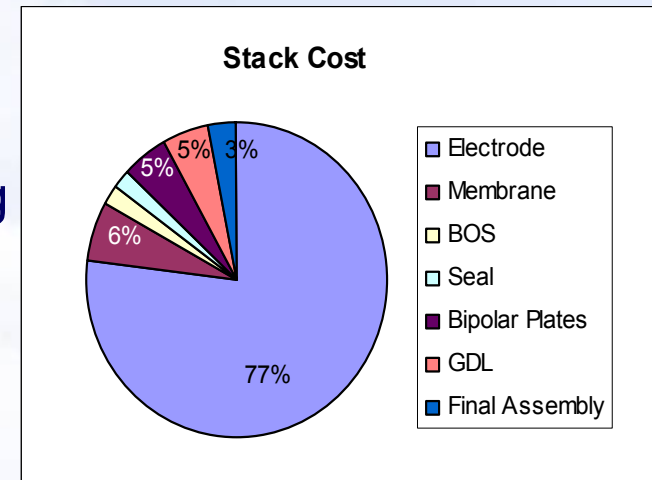
Pt-based alloys that maintain performance compared to Pt and cost less (3M, Cabot, UTC, Brookhaven)

### Strategy 3 – Novel support structures

Non-carbon supports and alternative carbon structures (3M, ANL)

### Strategy 4 – Non-Pt catalysts

Non-precious metal catalysts that maintain performance and durability compared to Pt (3M, Los Alamos NL, Ballard, U. of S. Carolina).

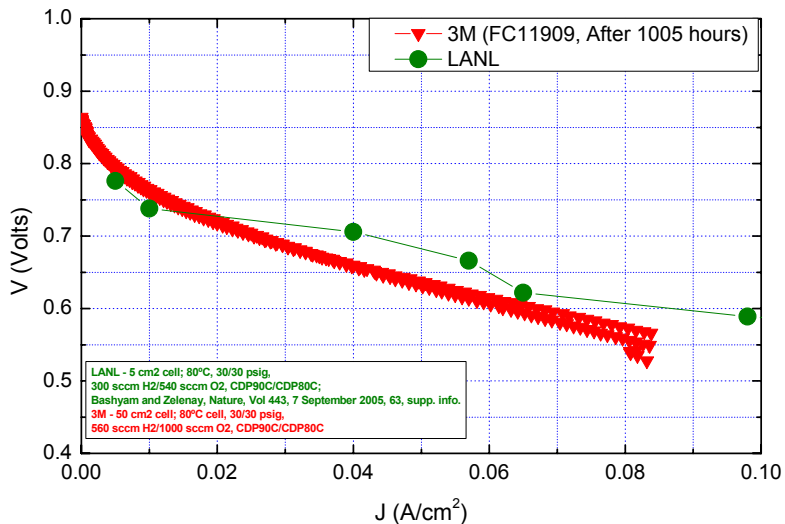




# Results: R&D Highlights - Catalysts

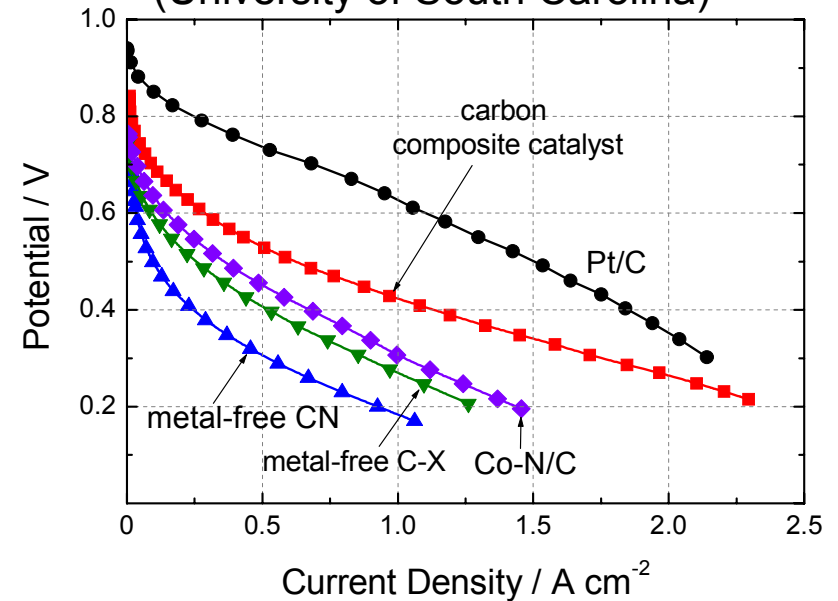
## Non-PM Catalysts

- Increased durability of non-PM catalysts, achieved 1,000 h with practically no irreversible degradation losses (3M)



## Non-PM Catalysts

- Increased activity of non-PM catalysts (University of South Carolina)



- **Anode:** 2 mg/cm<sup>2</sup> of ETEK 20% Pt/C
- **Cathode:** 6 mg/cm<sup>2</sup> of cathode catalyst
- **Membrane:** Nafion® 112
- **Operating temperature:** 77°C (H<sub>2</sub>); 75 °C (O<sub>2</sub>); 75°C (cell)
- **Back pressure:** 30 psi (H<sub>2</sub>)/40 psi (O<sub>2</sub>)





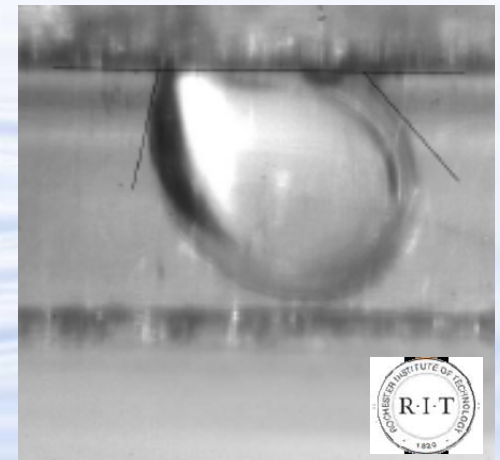
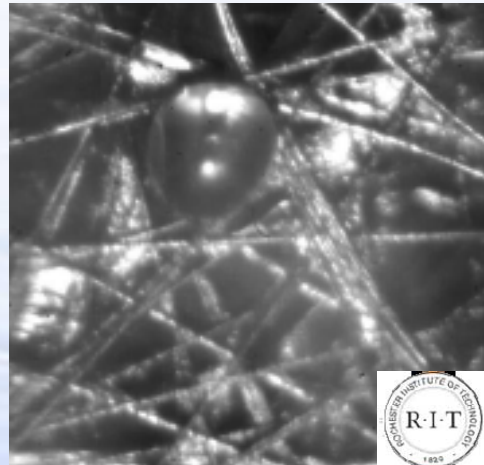
# Major Project - Water Transport

## Importance to goals:

- Understanding water transport key to operation in cold climates
- Water management at high power to prevent flooding
- Water management to prevent membrane drying out

## Strategy for Water transport Research:

- Optical and neutron imaging of water movement in fuel cells and theoretical modeling (NIST, LANL, CFD Research, Nuvera, RIT, GM, ANL)
- New projects beginning



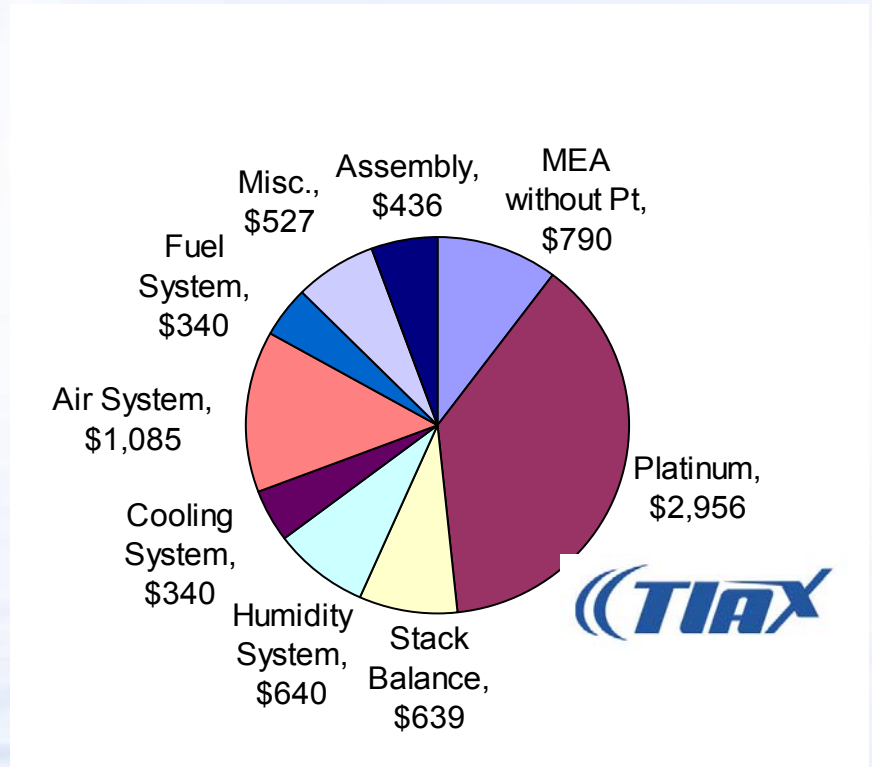
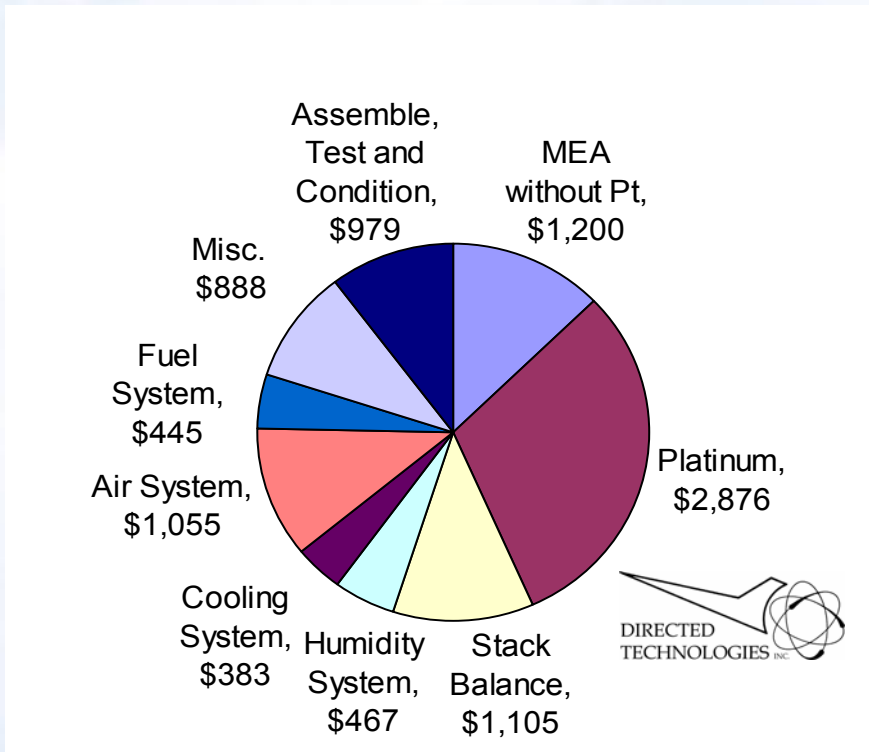




# Two High-volume Cost Analyses in 2006 - Directed Technologies and TIAX

**DTI Fuel Cell System 80 kW Direct H<sub>2</sub>**  
**Cost = \$118/kW (net), \$9412**

**TIAX Fuel Cell System 80 kW Direct H<sub>2</sub>**  
**Cost = \$97/kW (net), \$7760**



- The differences between the DTI and TIAX estimates are: the cost of the MEA and seals in stack balance and DTI included Test & Conditioning
- The 2015 cost target is \$30/kW, \$3200.



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# New Fuel Cell projects from 2006 Solicitation/LabCall

## Improved Fuel Cell Membranes

- Semi-interpenetrating networks of PVDF & sulfonated polyelectrolyte (Arkema \$6.3M)
- Membranes/MEAs for dry, hot operating conditions (3M - \$8.9M)
- New Polyelectrolyte Materials (LBNL- \$6M)

## Water Transport within the Stack

- Characterization under Freezing Conditions (RIT - \$2.7M)
- Subfreezing Start/Stop Protocol (Nuvera - \$5.0M)
- Water Transport Exploratory Studies (LANL - \$5.5M)
- Modeling, Material Selection, Testing & Optimization (CFD Research - \$4.7M)

## Catalysts and Supports

- Advanced Cathode Catalysts and Supports (3M - \$8.4M)
- Highly Dispersed Alloy Cathode Catalyst for Durability (UTC - \$6.4M)
- Non-Platinum Bi-metallic Cathode Electrocatalysts (LANL - \$6.8M)
- Alternative and Durable High Performance Cathode (ANL - \$5.4M)
- Supports for PEM Fuel Cells (PNNL - \$4.6M)



# New Fuel Cell Projects (Continued)

## Cell Hardware

- Next Gen. Bipolar Plates for Automotive PEMFCs (GrafTech - \$2.3M)
- Nitrided Metallic Bipolar Plates (ORNL - \$4.5M)
- Low Cost, Durable Seals (UTC - \$2.0M)

## Innovative Fuel Cell Concepts

- Adaptive Stack with Subdivided Cells (Plug Power - \$1.0M)
- Light-weight, Low-cost PEM Fuel Cell Stacks (CWRU - \$0.8M)
- Low-Cost, Microchannel Humidification Systems (PNNL - \$1.0M )
- Stack Using Patterned, Aligned Carbon Nanotubes as Electrodes (ANL - \$1.0M)

## Effects of Impurities on Fuel Cell Performance and Durability

- University of Connecticut - \$1.9M
- Clemson University - \$2.0M
- Los Alamos National Laboratory (LANL - \$3.6M)

## Stationary Fuel Cell Demonstration

- International: Combined Heat & Power Unit (Intelligent Energy - \$2.2M)
- IPHE: Development & Demo of High Efficiency (Plug Power - \$3.6M)
- Intergovernmental: Stationary Fuel Cell System Demo (Plug Power - \$4.0M)



# 2010-2015 Research Goals

## Performance

- Develop membranes and catalysts that operate at high temperature (120°C) with little or no humidification
- Develop water management strategies that prevent the membrane from either flooding or dehydrating
- Develop materials and strategies that enable start-up from -40°C

## Durability

- Develop stable catalyst supports that do not corrode during fuel cell operation
- Develop membranes and catalysts that withstand the chemical and mechanical stresses of fuel cell operation
- Quantify effects of impurities on fuel cell performance, develop mitigation strategies, and establish H<sub>2</sub>/air(?) purity requirements for fuel cells

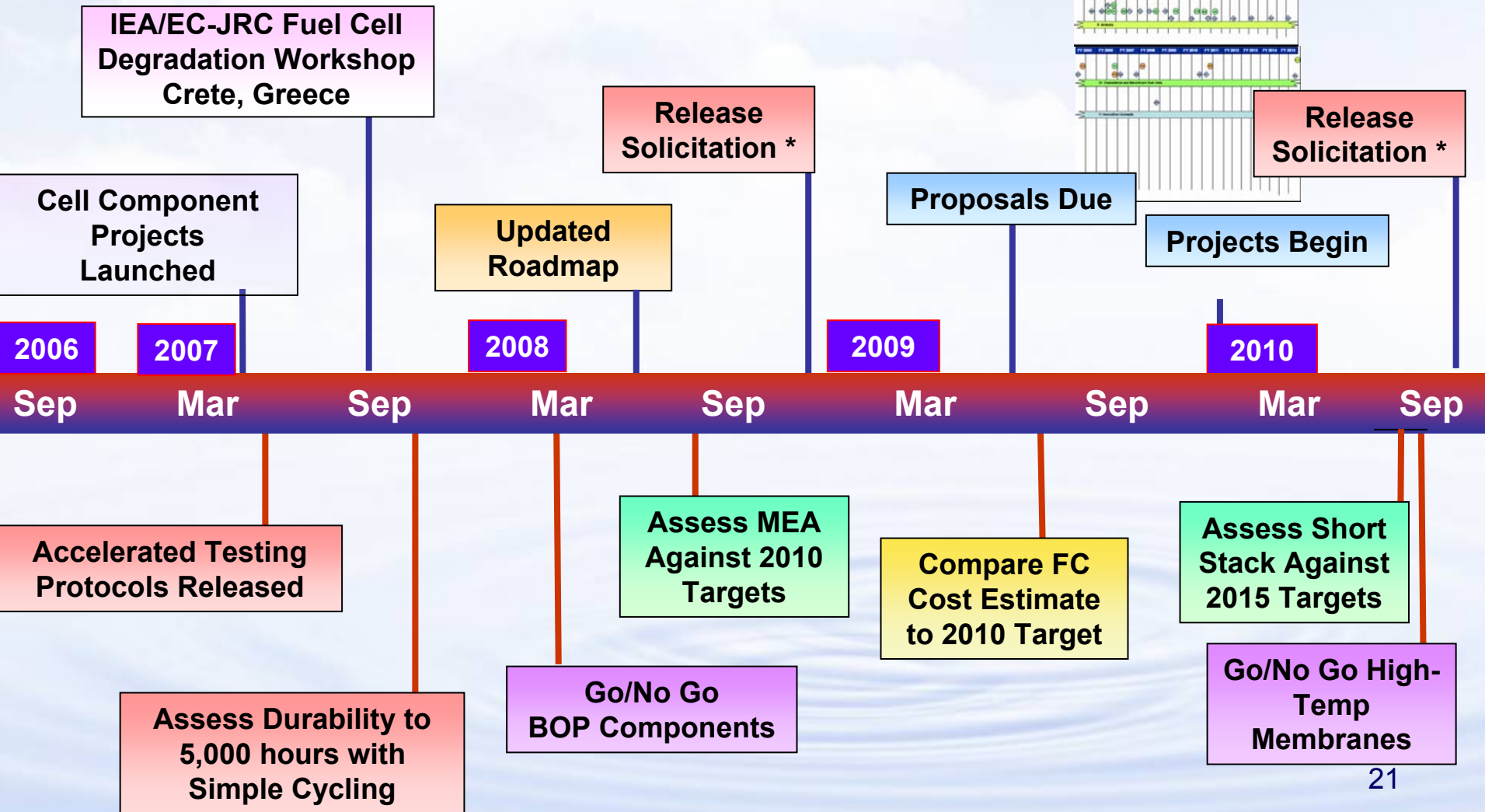
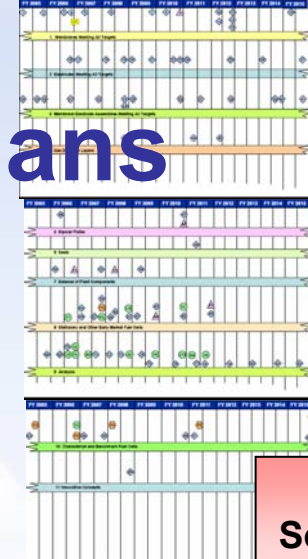
## Cost

- Develop high performance precious and non-precious metal catalysts that meet catalyst loading targets (0.3 mg Pt/cm<sup>2</sup>: 2010; 0.2 mg Pt/cm<sup>2</sup>:2015)
- Investigate non-traditional stack and system designs that enable the use of less expensive materials and components





# Fuel Cells – Future Plans



\*Subject to appropriations



# For More Information

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The screenshot shows the homepage of the Hydrogen Program website. The header includes the U.S. Department of Energy logo and the text "hydrogen.energy.gov". A navigation menu contains links for Home, About, DOE Participants, International, Library, and News/Events. A search bar is located in the top right. The main content area features several sections: "INCREASE YOUR H<sub>2</sub>IQ" with an "Announcement" for a "Peer Evaluation Report Focuses on Merit of DOE Hydrogen and Fuel Cell Projects"; "News" with a headline "Independent Review Panels Assess Progress Towards Technical Targets" dated October 5, 2006; "DOE Announces Hydrogen Funding Opportunity for Small Businesses" dated September 27, 2006; and "DOE Loan Guarantee Program Promotes Innovative Technologies" dated August 23, 2006. A sidebar on the left lists categories like Hydrogen Production, Delivery, Storage, Manufacturing, Conversion/Fuel Cells, Applications/Technology Validation, Safety, Codes & Standards, Education, Basic Research, Systems Analysis, and Systems Integration. The right sidebar includes a "Features" section with a "President's Hydrogen Fuel Initiative" and logos for "ADVANCED ENERGY INITIATIVE", "Hydrogen.gov", and "FreedomCAR Fuel Partnership".

[www.hydrogen.energy.gov](http://www.hydrogen.energy.gov)