

---

# **U.S. Department of Energy Hydrogen Program**

## ***Hydrogen Storage***

**Sunita Satyapal**

**2008 DOE Hydrogen Program  
Merit Review and Peer Evaluation Meeting**

**June 9, 2008**





# Goal and Objectives

**GOAL: On-board hydrogen storage for > 300 mile driving range across different vehicle platforms, *WITHOUT COMPROMISING* passenger/cargo space, performance (wt, vol, kinetics, safety, etc.) or cost**

Develop on-board storage systems to meet DOE targets, including:

- Capacity
- Operating temperature range (-40 to +85°C)
- Hydrogen supply rate/refueling rate
  - 0.02 g H<sub>2</sub> per sec. per kW of power
  - Refueling time <3 min. for 5 kg H<sub>2</sub>
- System cost
- Fuel cost
- Safety, C&S, reliability, cycle life, efficiency, etc.



# Challenges

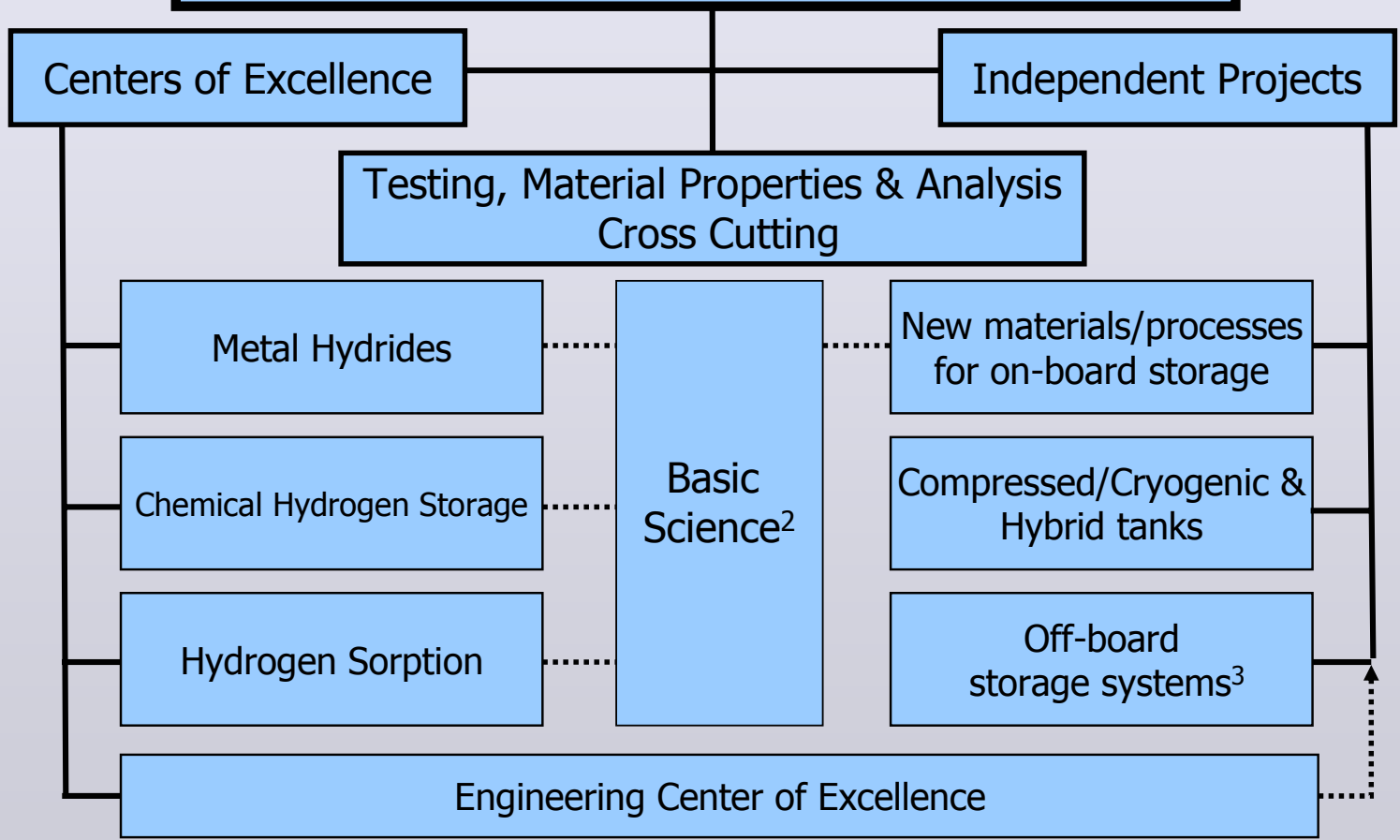
- **Vehicles are being designed by OEMs that can achieve > 300 miles**
  - 350 or 700 bar
  - 1 to 4 tanks
  - Specified range from ~200 to > 350 miles
- **But performance, space on-board and cost are still challenges for mass market penetration...**
- **Is there a low pressure alternative?**





# Strategy – Diverse, Balanced Portfolio

## National Hydrogen Storage Project<sup>1</sup>



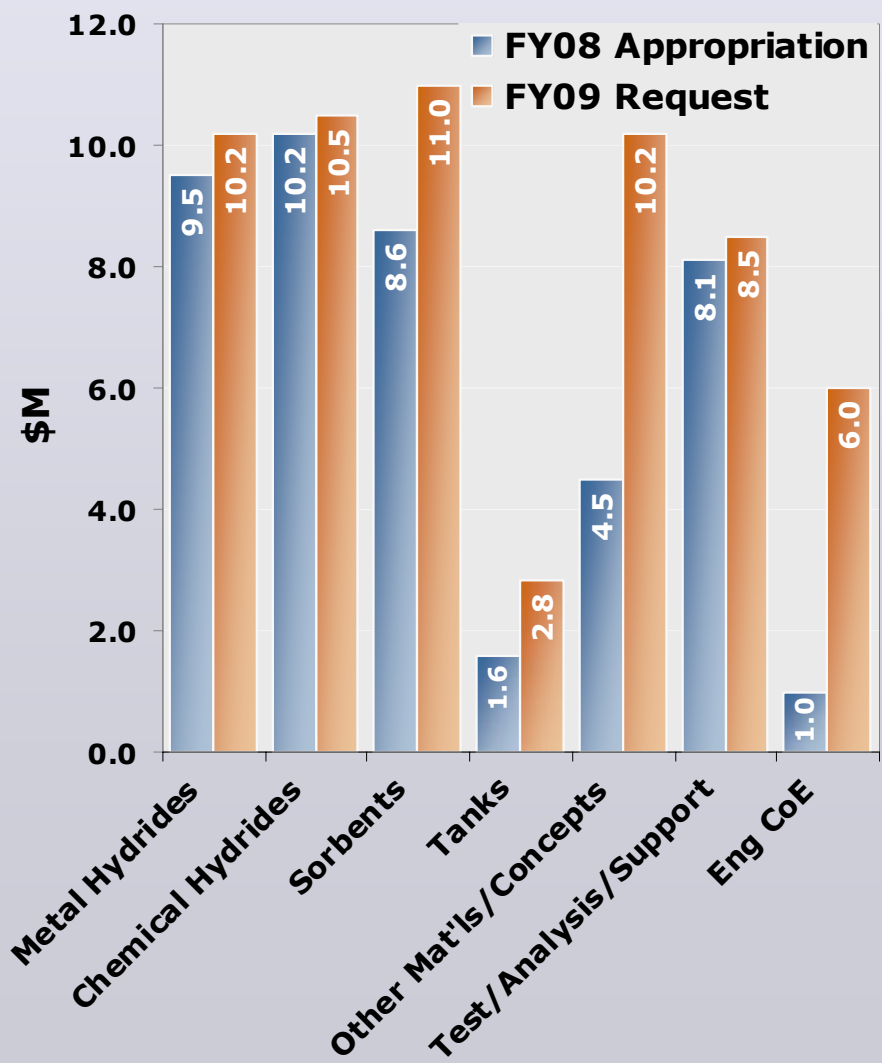
- 1. Coordinated by DOE Energy Efficiency and Renewable Energy, Office of Hydrogen, Fuel Cells and Infrastructure Technologies
- 2. Basic science for hydrogen storage conducted through DOE Office of Science, Basic Energy Sciences
- 3. Coordinated with Delivery Program element

**~40 Universities, ~20 Companies, ~15 Federal Laboratories**



# Hydrogen Storage Budget

**FY 2009 REQUEST = \$59.2M**  
**FY 2008 APPROPRIATION = \$43.5M**

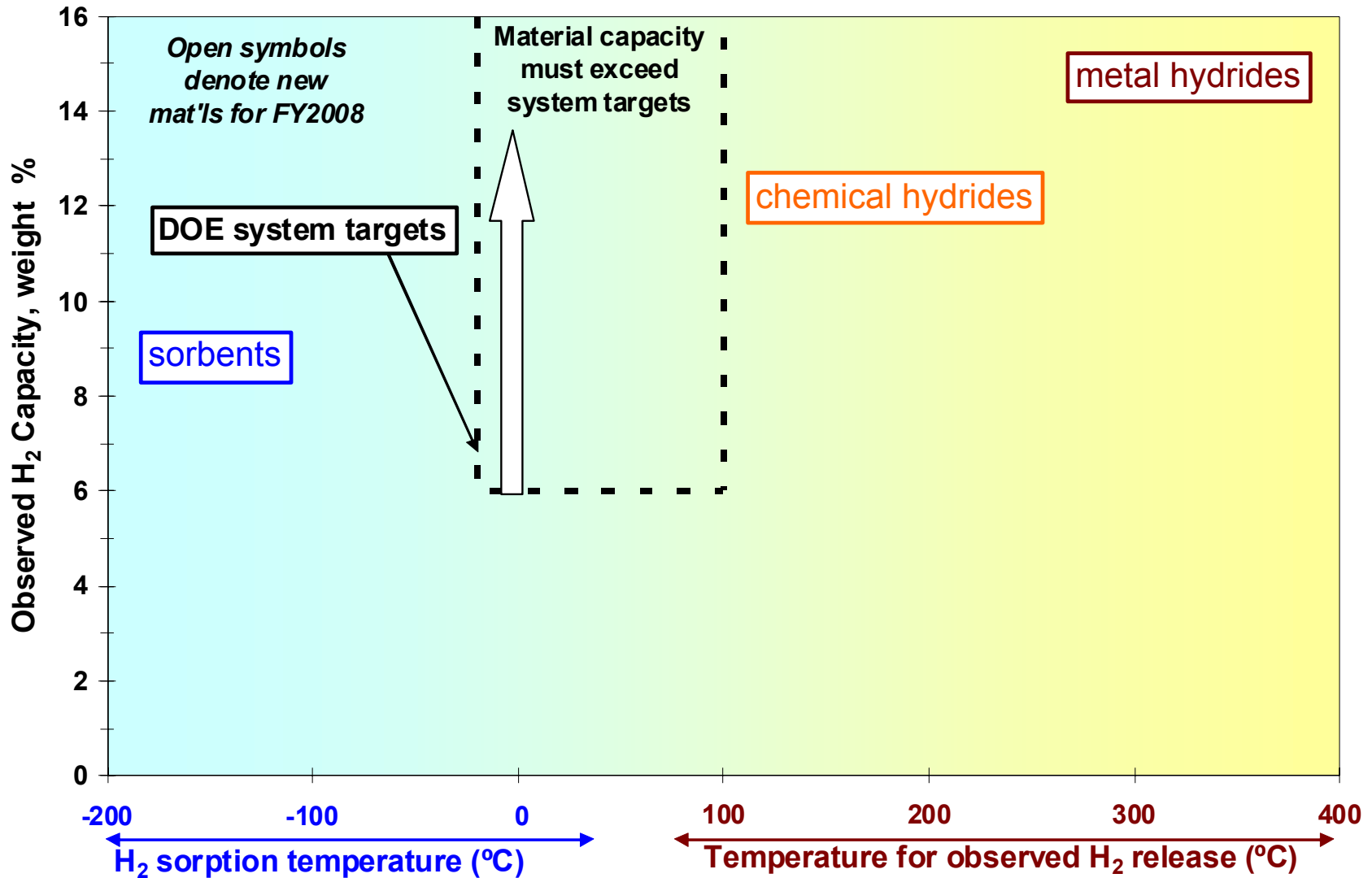


- ### FY2009 Emphasis
- Increase engineering in addition to materials R&D through Centers of Excellence and independent projects to enable system targets.
  - Focus on kinetics, temperature, pressure, cycle life, spent fuel regeneration, etc. *in addition* to capacity
  - Strengthen tank R&D to address NAS recommendations. Focus on cost reduction and advanced concepts. Also applicable to materials-based approaches.
  - Continue close coordination with Basic Science



# Progress

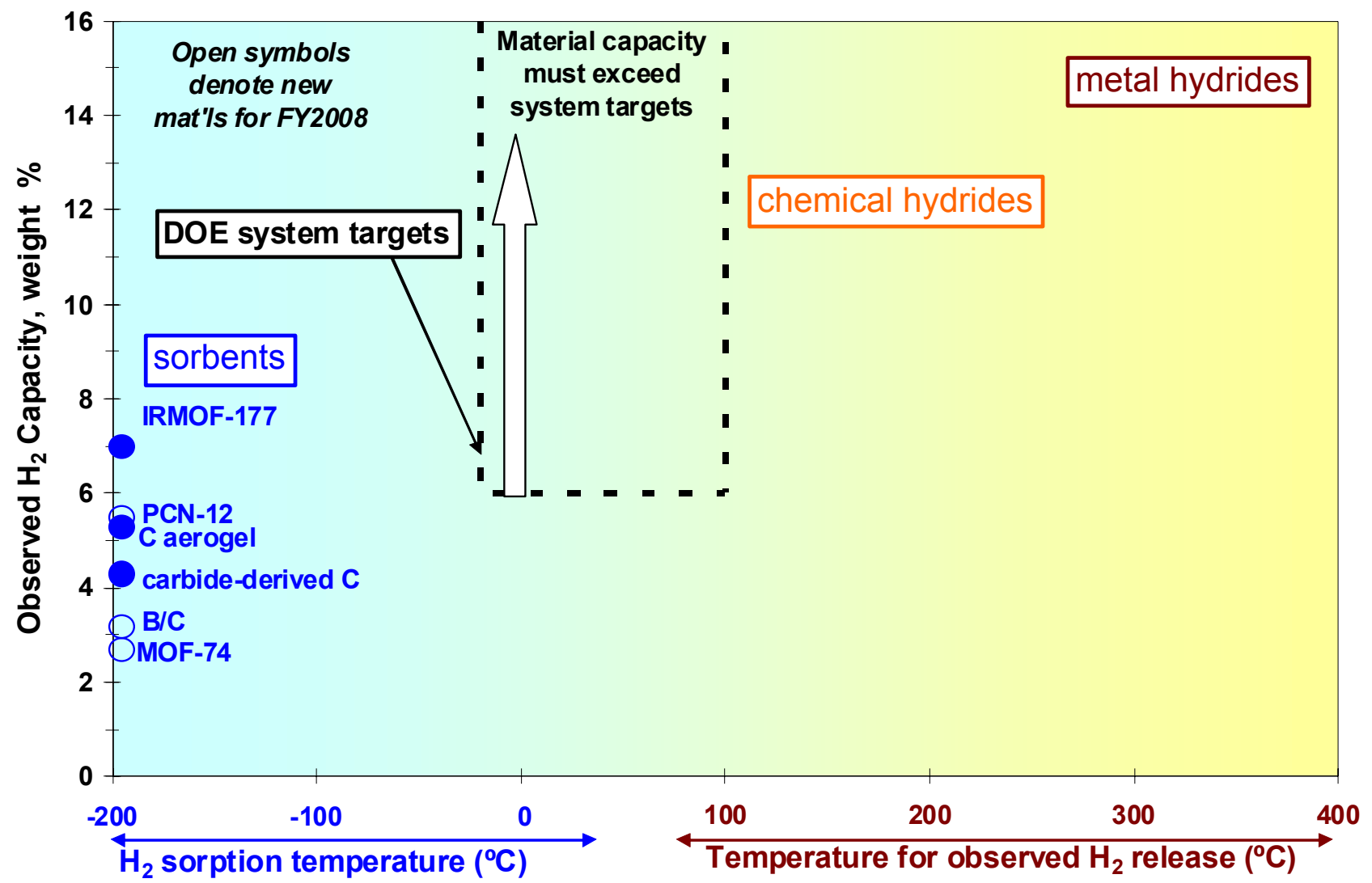
## Material Capacity vs. Temperature





# Progress

## Material Capacity vs. Temperature

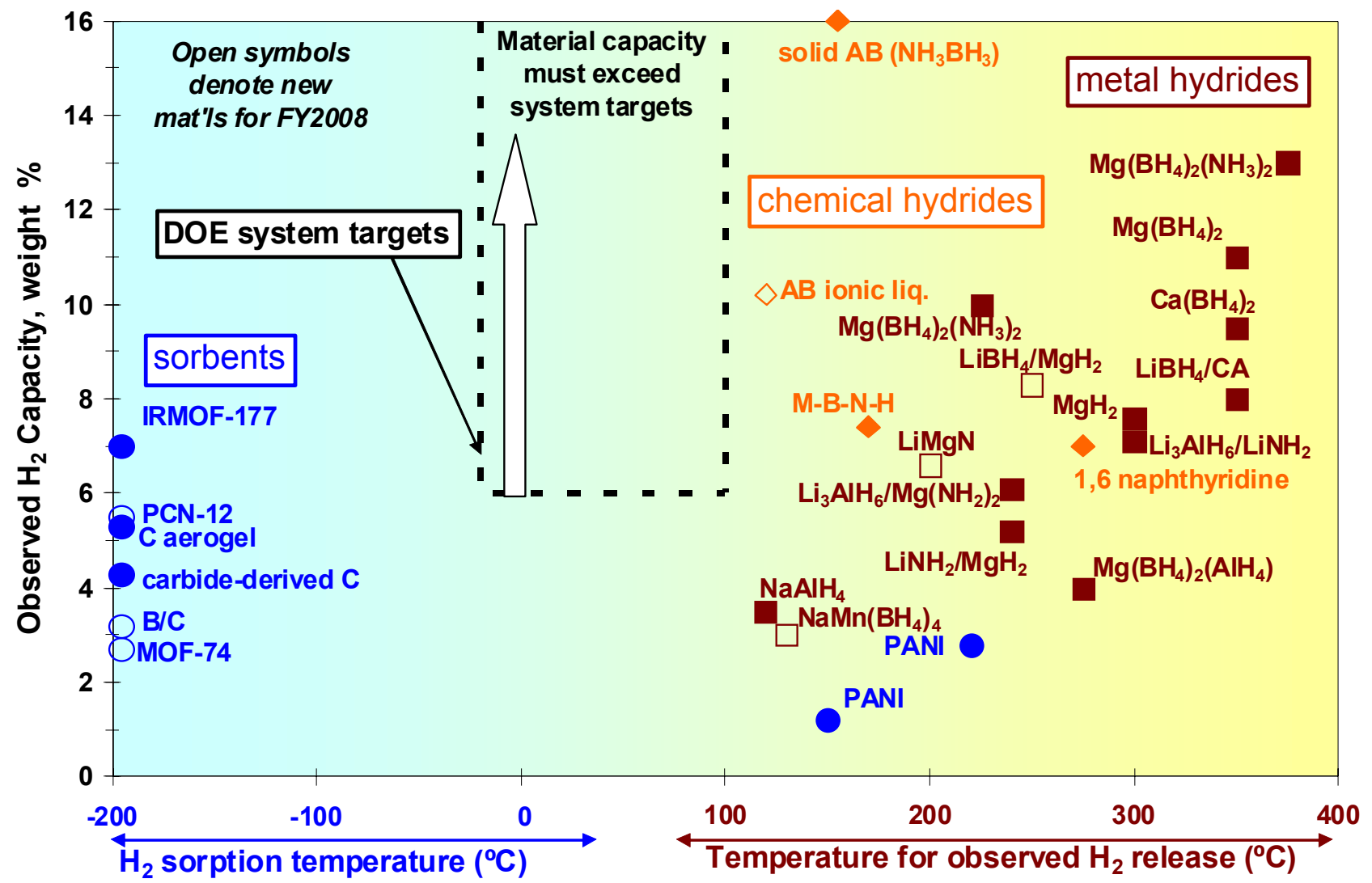


DOE: G. Thomas (2007), G. Sandrock (2008)



# Progress

## Material Capacity vs. Temperature



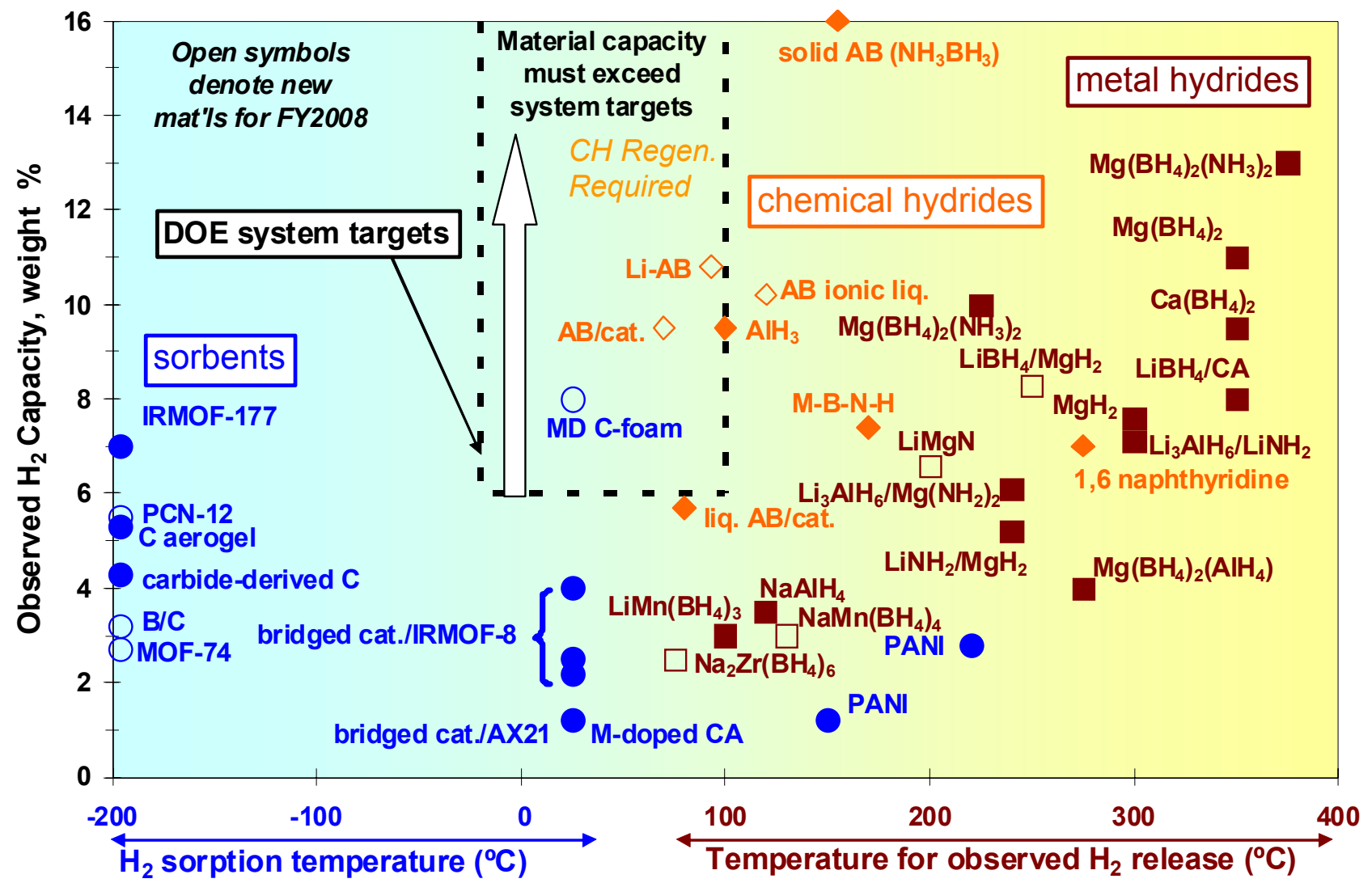
DOE: G. Thomas (2007), G. Sandrock (2008)





# Progress

## Material Capacity vs. Temperature

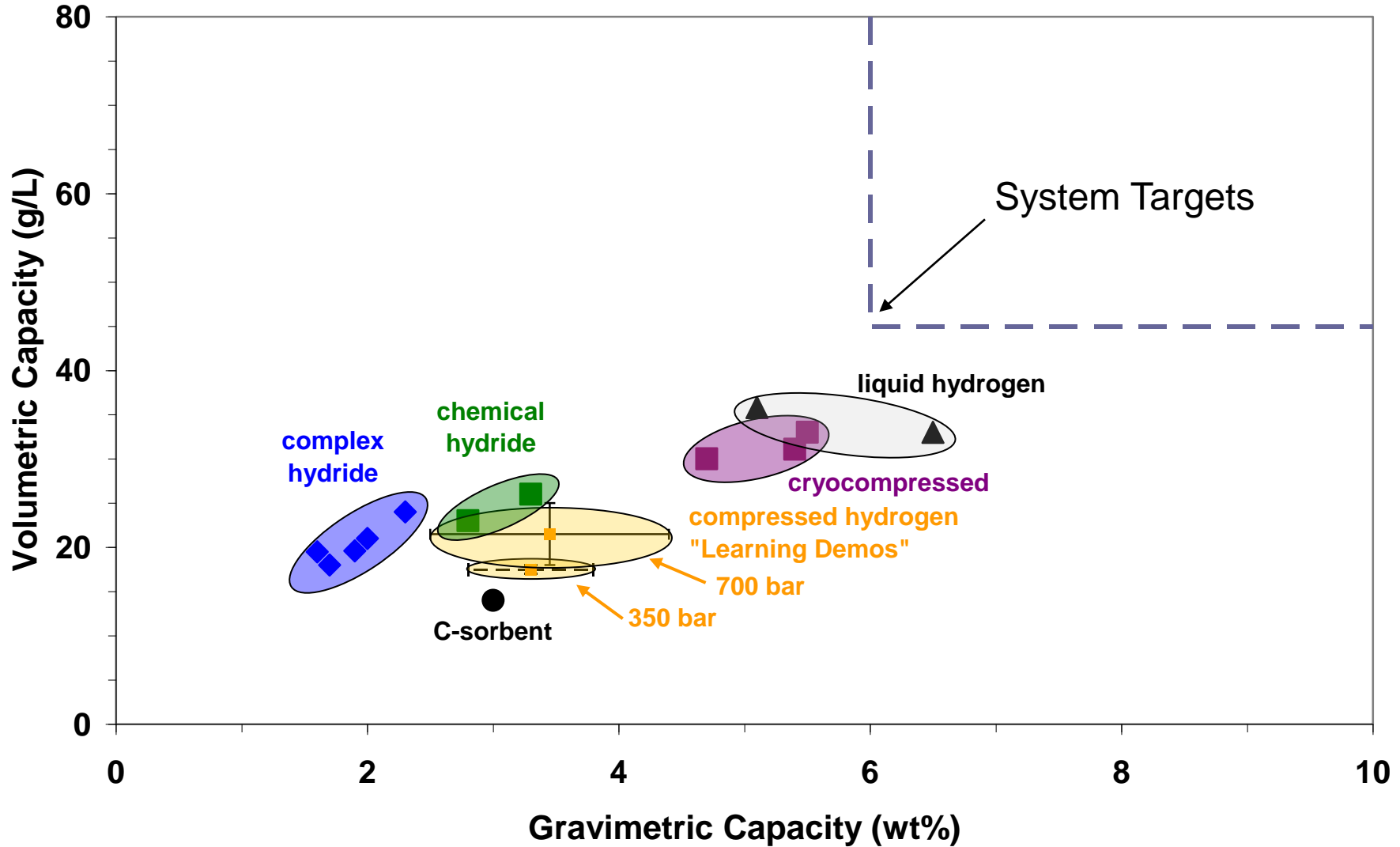


DOE: G. Thomas (2007), G. Sandrock (2008)



# Current Status

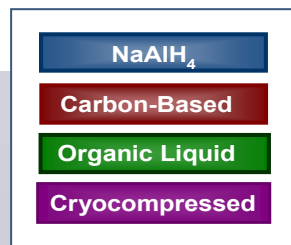
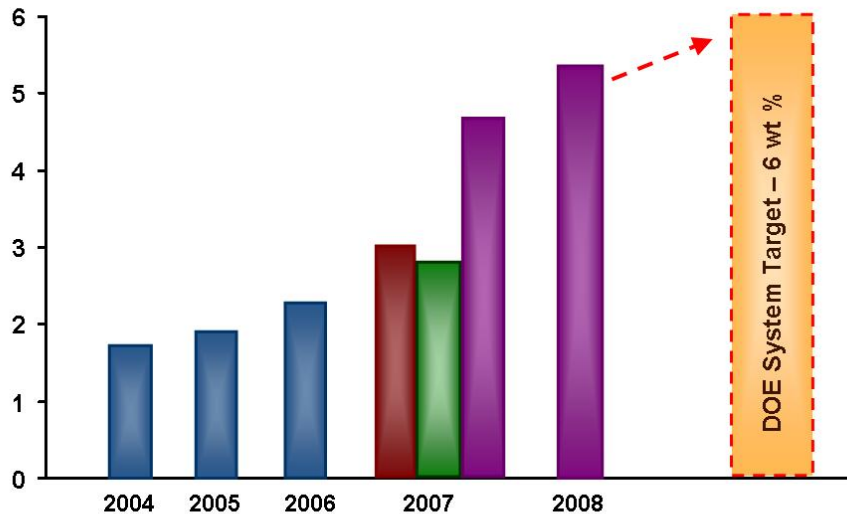
*No technology meets targets*





# Hydrogen Storage System Progress

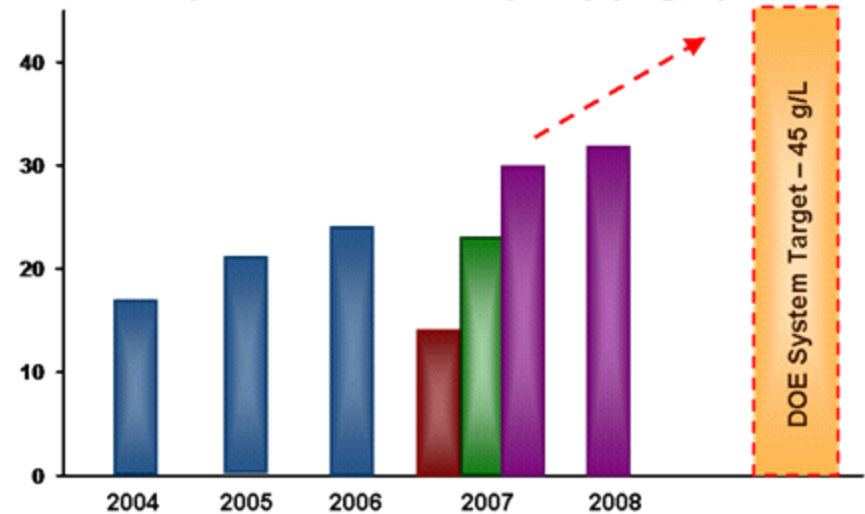
System Gravimetric Capacity (in weight %)



- Projected system capacities based on modeling and material data.
- Subscale prototype developed for NaAlH<sub>4</sub>
- Full scale prototype developed for cryocompressed tank

- Preliminary designs developed and improvements made
- But no technology meets targets
- Need to focus on volumetric capacities

System Volumetric Capacity (in g/L)





# Progress

## Metal Hydride Down-Selects

**More than 50 Metal Hydride Approaches Assessed**

~ 50% discontinued

~ 50% show some promise

### Examples of Approaches Discontinued:

- $\text{MgH}_2/\text{Si}$ : not reversible **X**
- $2\text{LiNH}_2 + \text{MgH}_2$  : wt. % limited **X**
- $\text{Li}_2\text{Zn}(\text{BH}_4)_4$ : high diborane **X**

### Examples of Approaches Continued:

- $\text{LiBH}_4/\text{MgH}_2$  in aerogels ✓
- $\text{LiMgN}$ ,  $\text{Li}_3\text{AlH}_6/3\text{LiNH}_2$  ✓
- $\text{A}_x\text{Mn}(\text{BH}_4)_y$  [A=Li, Na, K] ✓
- $\text{AlH}_3$  , and 22 other systems ✓

***In fulfillment of materials down-select milestone***

**Materials Go/No-Go Decisions Made Within the Department of Energy Metal Hydride Center of Excellence (MHCoe)**

In fulfillment of the end of Fiscal Year 2007 Project Milestone on Materials Down-selection

Lennie Klebanoff, Director  
Sandia National Laboratories  
Livermore, CA 94551

September/October 2007



[http://www1.eere.energy.gov/hydrogenandfuelcells/hydrogen\\_publications.html#h2\\_storage](http://www1.eere.energy.gov/hydrogenandfuelcells/hydrogen_publications.html#h2_storage)



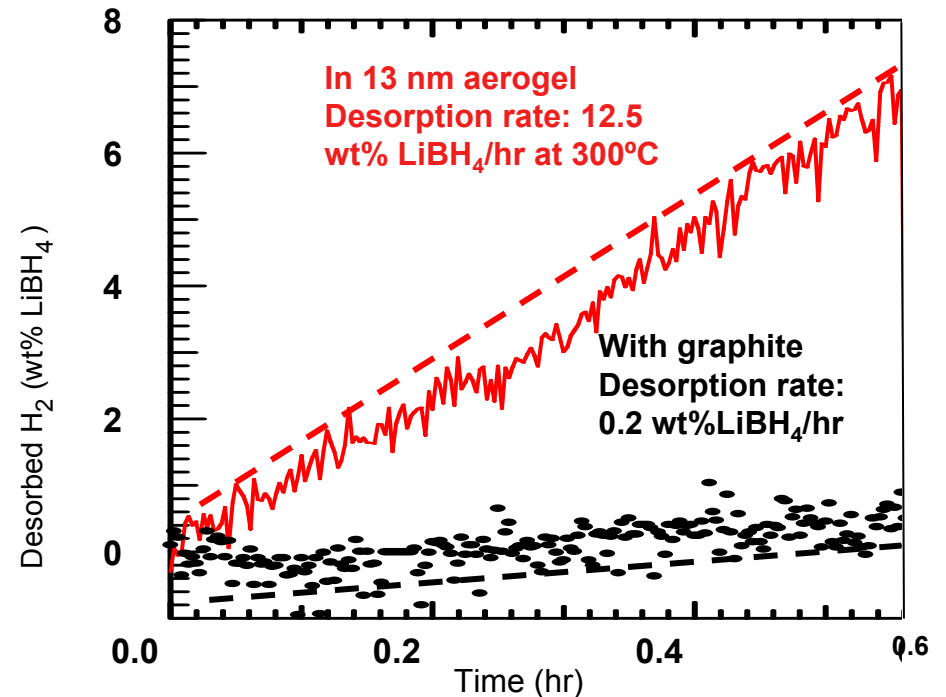
# Accomplishments

## Metal Hydride Examples

- Developed theory tool and screened > 16 million compositions
- Identified > 40 single step reactions as promising so far (with > 6 wt% H<sub>2</sub> and 15 < ΔU<sub>0</sub> < 75 kJ/mol H<sub>2</sub>)

Alapati, Johnson and Sholl,  
*J. Phys. Chem. C*, 112, 5258 (2008)

**Increased kinetics**  
**60-fold using LiBH<sub>4</sub>**  
**in aerogel**



Vajo, et al, HRL & Baumann, et al, LLNL



# 2008 Progress

## Chemical Hydrogen Storage Down-Selects

### More than 60 Materials Assessed

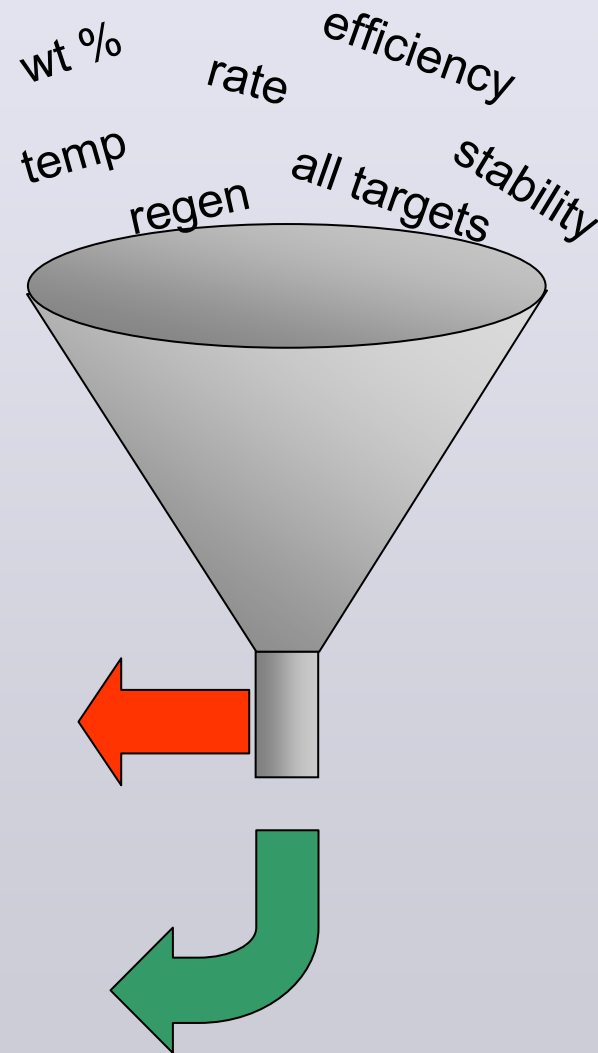
- ~ 50% discontinued
- ~ 30% show some promise but have issues
- ~ 20% show some potential to meet targets

### Examples of Approaches Discontinued:

- Hydrolysis of polyhedral boranes, &  $\text{NaBH}_4$ : inefficient spent fuel regen **X**
- $\text{Mg}(\text{OMe})_2/\text{H}_2\text{O}$ :  $>200\text{C}$  release T **X**

### Examples of Approaches Continued:

- **Exothermic release:** e.g. Ammonia borane (AB), AB-scaffolds, AB - ionic liquid mixtures, liquid amine boranes ✓
- **New materials, thermoneutral or coupled release:** E.g., Metal-boron-nitrogen materials ✓





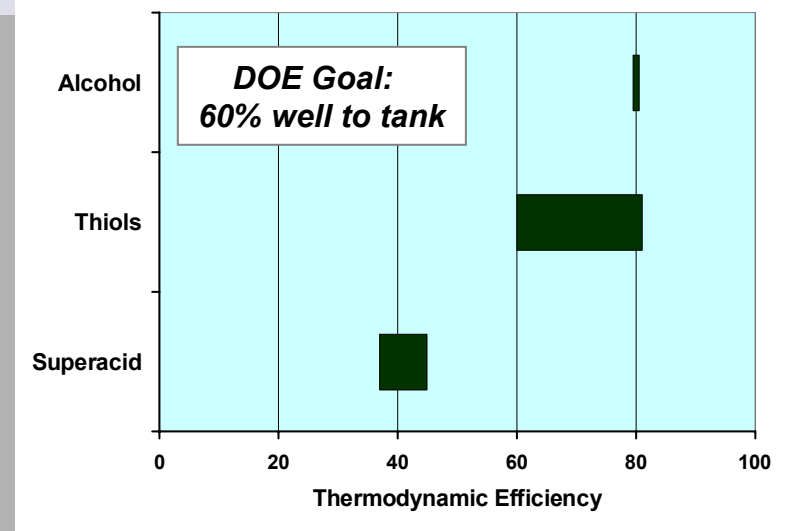
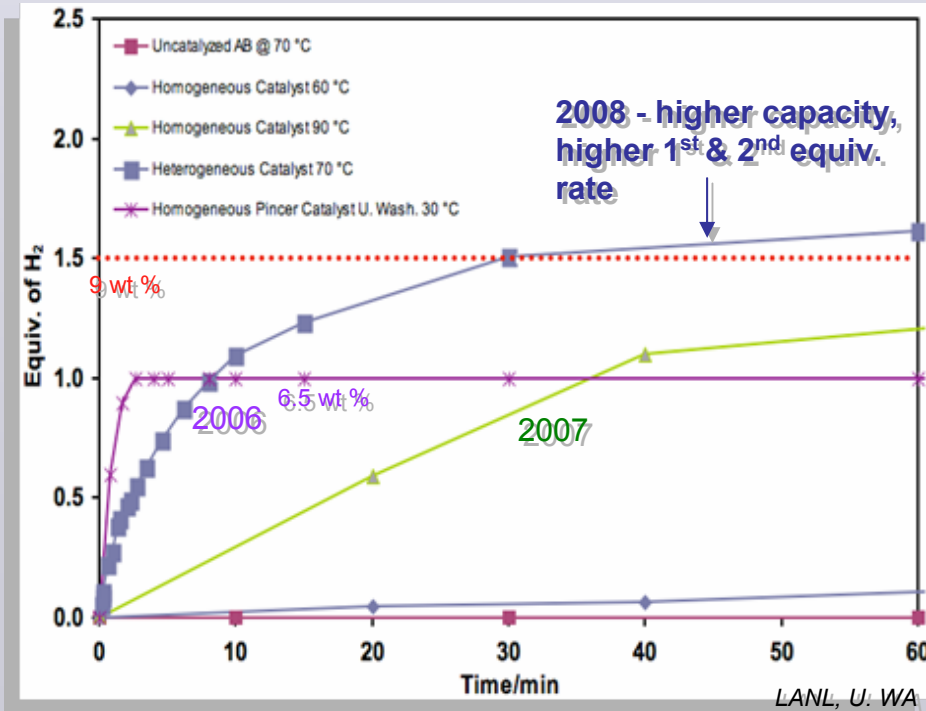
# Accomplishments

## Ammonia borane

$\text{NH}_3\text{BH}_3 \rightarrow \text{BNH}_x + 3\text{H}_2$  **19.4 wt.%, 160 g/L** (theoretical material capacity)

- Increased  $\text{H}_2$  release rates by 4X compared to 2007. Can meet DOE rate targets
- Improved  $\text{H}_2$  capacity by > 50% since 2006

- Improved regeneration efficiencies by 22-35%



- Increased efficiency via design of optimum digesting agent and reduction strategy
- Improved yields for all steps in the 3 regen schemes

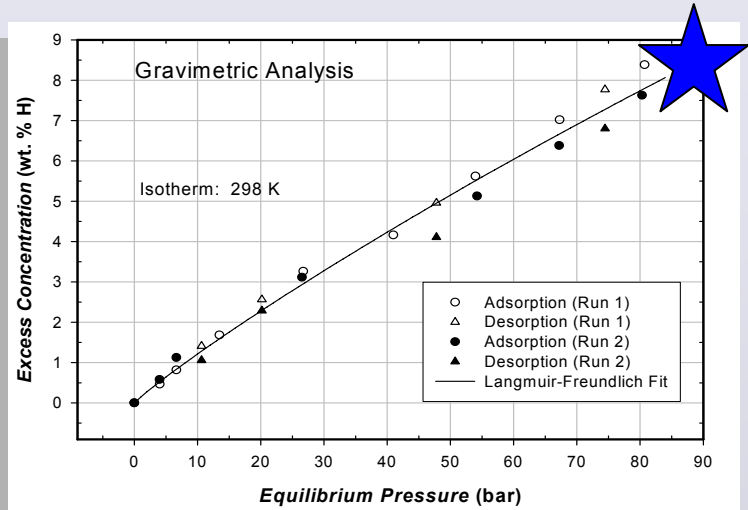


# Progress

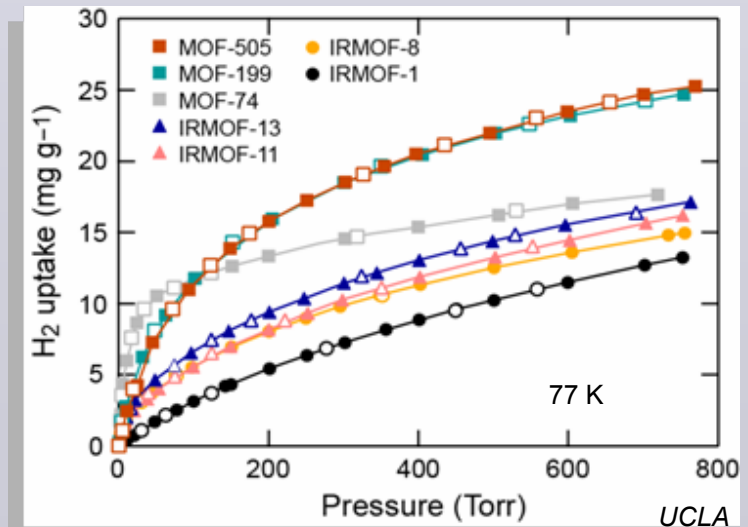
## Hydrogen Sorption Examples

- DOE work on “spillover” catalyzed worldwide R&D
- Led to 8 wt.% at room temp

- Tailored binding energies
  - ✓ PCN-12  $\Delta H_i \sim 12$  kJ/mol
  - ✓ MOF-74  $\sim 8$  kJ/mol
  - ✓ Activated C-fiber  $\sim 10$  kJ/mol (compared to  $<6$  kJ/mol)
- Increased  $H_2$  uptake by 75% using open metal sites
- Modeling of sorbents and spillover identified thermodynamically favorable approaches



National Center of Scientific Research “Demokritos” (NESSHY, EC) and SwRI (DOE)



UCLA



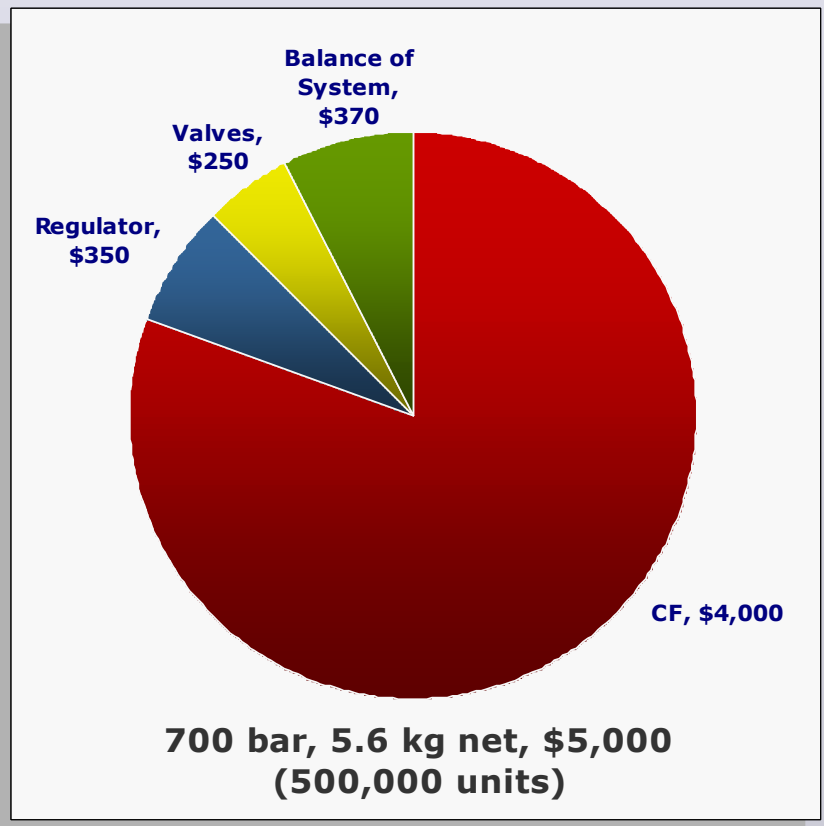


# Progress Tanks

- Demonstrated 103 to 190 mile range across 92 vehicles (Gen 1) through Technology Validation activity
- Demonstrated ~ 2X increase in dormancy using cryocompressed tanks (LLNL)
- Assessed high P tank cost (TIAX)
  - High volume cost projections:
    - ~ \$27/kWh (700 bar)
  - Assessed cryo-compressed tank cost & sensitivity analysis

**System Gravimetric Capacity**  
350 bar: 2.8-3.8 wt.%  
700 bar: 2.5-4.4 wt%

**System Volumetric Capacity**  
350 bar: 17-18 g/L  
700 bar: 18-25 g/L





# Progress

## Materials Properties, Testing & Analyses

- **Best Practices developed for hydrogen storage equilibrium & kinetics measurements**

– **Draft online for public comment**

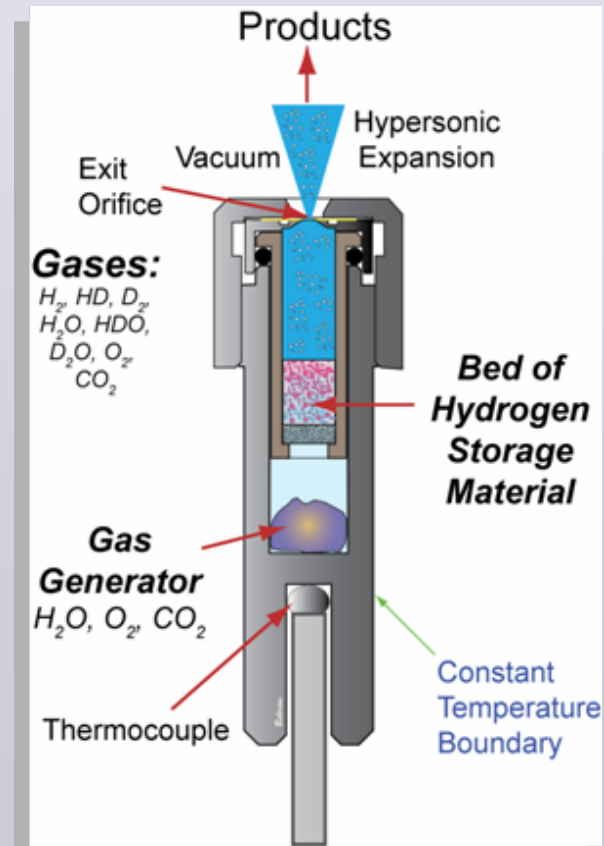
K. Gross, H2 Technology Solutions/HyEnergy, LLC/NREL

### Systems Analysis

- **Preliminary well to tank efficiency analysis conducted**
- **System capacity and cost analysis conducted for multiple approaches**

Argonne, TIAX

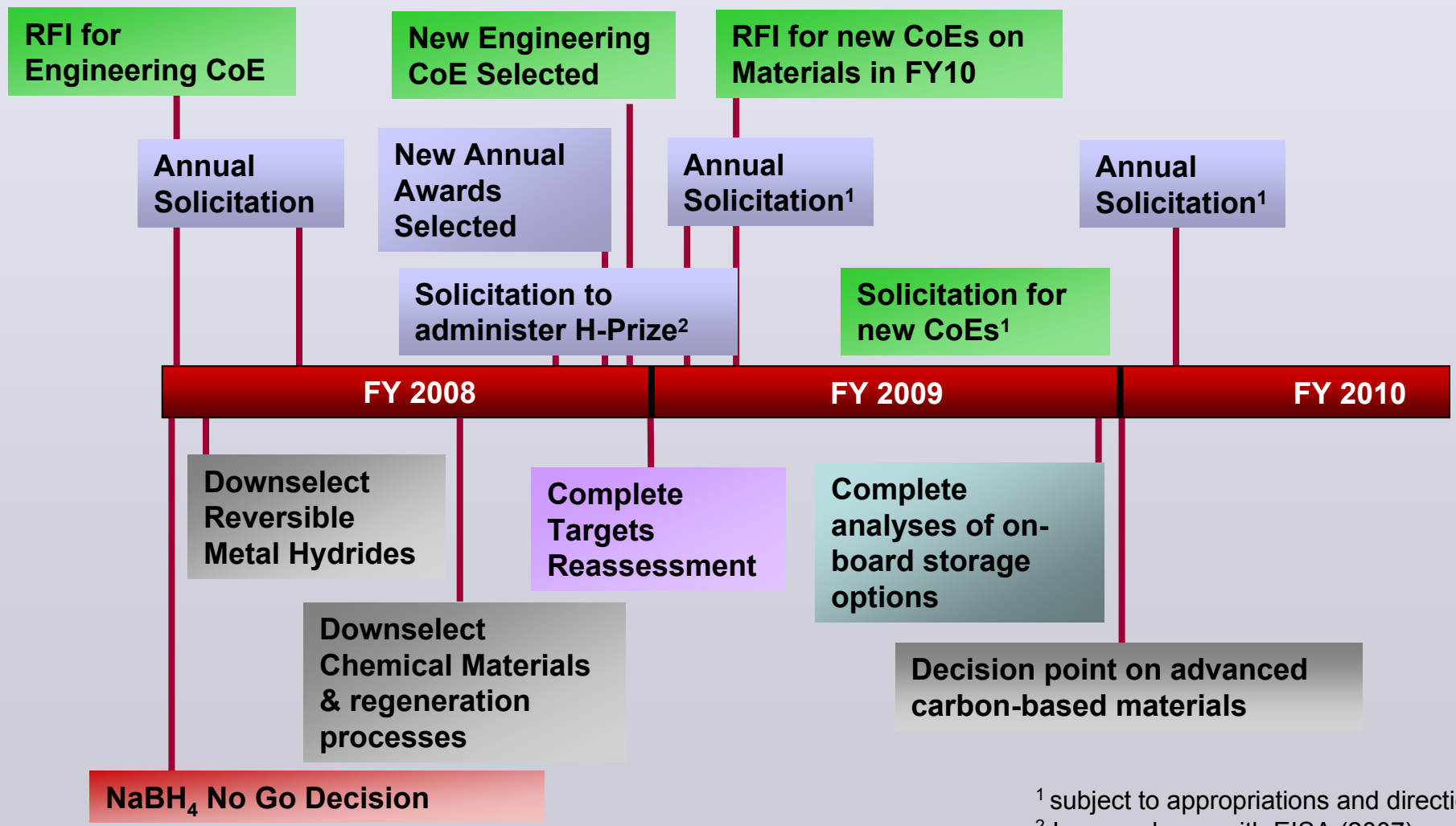
- **Reactivity of hydrogen storage materials assessed under various exposures – An IPHE Collaboration**



Savannah River, Sandia, UTRC (US) & Japan, Germany, Canada



# Key Hydrogen Storage Milestones & Future Plans



<sup>1</sup> subject to appropriations and direction  
<sup>2</sup> In accordance with EISA (2007)



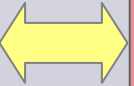
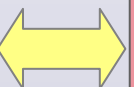
# Hydrogen Storage Collaborations

**Applied R&D under the President's Hydrogen Fuel and Advanced Energy Initiatives is coordinated among national and international organizations**

**INTERNATIONAL ACTIVITIES**

Examples

- IEA HIA Task 22  
~ 15 countries  
~ 50 projects
- IPHE  
5 projects  
(EU, Russia, Canada, US, New Zealand, Singapore, China)

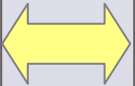


**DOE**

(Energy Efficiency & Renewable Energy- EERE)  
**National Hydrogen Storage Project**

*H<sub>2</sub> Storage Applied R&D*

- 3 Centers of Excellence
- Independent projects
- Total of ~80 projects



**INDUSTRY**

- FreedomCAR & Fuel Partnership  
*Tech teams:*
  - H<sub>2</sub> Storage
  - Fuel Cells, Delivery
  - Fuel Pathways
  - Vehicle Systems
- Codes & Standards Organizations



**TECHNOLOGY VALIDATION (DOE EERE)**

~92 vehicles & 15 stations

**National Collaboration (inter- and intra-agency efforts)**

<p><b>DOE – Basic Energy Sciences</b></p> <p>~30 Projects</p>	<p><b>NSF</b></p> <p>New projects in basic science</p>	<p><b>NIST</b></p> <ul style="list-style-type: none"> <li>• Neutron scattering</li> <li>• Measurements</li> </ul>	<p><b>DOT</b></p> <p>Material handling/transport</p>	<p><b>DOD – Defense Logistics Agency</b></p> <p>6 projects &amp; prototypes</p>
---------------------------------------------------------------	--------------------------------------------------------	-------------------------------------------------------------------------------------------------------------------	------------------------------------------------------	---------------------------------------------------------------------------------



# For More Information

## **Hydrogen Storage Team**

**Sunita Satyapal, Team Leader**

Overall Storage/ FreedomCAR Tech  
Team/International  
202-586-2336  
sunita.satyapal@ee.doe.gov

**Monterey Gardiner**

Tanks, Sorbents, Delivery  
202-586-1758  
monterey.gardiner@ee.doe.gov

**Grace Ordaz**

Chemical Hydrides, Chemical Hydrogen  
Storage Center of Excellence  
202-586-8350  
grace.ordaz@ee.doe.gov

**Carole Read**

Sorbents & Carbon, Hydrogen Sorption Center of  
Excellence/FreedomCAR Tech Team  
202-586-3152  
carole.read@ee.doe.gov

**Gary Sandrock**

On Assignment to DOE  
Oak Ridge Nat'l Lab  
202-586-8707  
gary.sandrock@ee.doe.gov

**Ned Stetson**

Metal Hydrides, Metal Hydride Center of  
Excellence, SC&S interface  
202-586-9995  
ned.stetson@ee.doe.gov

**Field Office Project Officers:**

Jesse Adams  
James Alkire  
Paul Bakke

**Support:**

Kristin Deason (Sentech)



# Applied R&D Hydrogen Storage “Grand Challenge” Partners

## Diverse Portfolio with University, Industry & National Labs

### Centers of Excellence

**Metal Hydride Center**  
**National Laboratory:**  
 Sandia-Livermore  
**Industrial partners:**  
 General Electric  
 HRL Laboratories  
 Intematix Corp.  
**Universities:**  
 CalTech  
 Stanford  
 Pitt / GATech  
 Hawai'i / UNB  
 Illinois  
 Ohio State  
 Nevada-Reno  
 Utah  
**Federal Lab Partners:**  
 Brookhaven  
 JPL, NIST  
 Oak Ridge  
 Savannah River

**Hydrogen Sorption Center**  
**National Laboratory:**  
 NREL  
**Industrial partners:**  
 Air Products & Chemicals  
**Universities:**  
 CalTech  
 Duke  
 Miami Univ.-OH  
 Michigan  
 North Carolina  
 Penn State  
 Rice  
 Univ. of Chicago  
**Federal Lab Partners:**  
 Argonne  
 Lawrence Livermore  
 NIST  
 Oak Ridge

**Chemical Hydrogen Storage Center**  
**National Laboratories:**  
 Los Alamos  
 Pacific Northwest  
**Industrial partners:**  
 Intematix Corp.  
 Millennium Cell  
 Rohm & Haas  
 US Borax  
**Universities:**  
 Northern Arizona  
 Penn State  
 Alabama  
 California-Davis  
 Univ. of Missouri  
 Pennsylvania  
 Washington  
**Federal Lab Partners:**  
 INL

### Independent Projects

**Advanced Metal Hydrides**  
 UOP  
 Univ. of Connecticut  
 Delaware State  
**Sorbent/Carbon-based Materials**  
 UCLA  
 State University of New York  
 Gas Technology Institute  
 UPenn & Drexel Univ.  
**Chemical Hydrogen Storage**  
 Air Products & Chemicals  
 RTI  
 Millennium Cell  
 Safe Hydrogen LLC  
**Other New Materials & Concepts**  
 Alfred University  
 Michigan Technological University  
 UC-Berkeley/LBL  
 UC-Santa Barbara  
 Univ. of Arkansas  
 Purdue  
 UNLV  
**Tanks, Safety, Analysis & Testing**  
 Lawrence Livermore Nat'l Lab  
 Quantum  
 Argonne Nat'l Lab, TIAX LLC  
 SwRI, UTRC, Sandia Nat'l Lab  
 Savannah River Nat'l Lab

**Coordination with: Basic Science (Office of Science, BES)**  
 MIT, U.WA, U. Penn., CO School of Mines, Georgia Tech, Louisiana Tech,  
 U.Georgia, Missouri-Rolla, Tulane, Southern Illinois, Rutgers, Stonybrook, UC  
 Davis, UC Santa Barbara, Sth Florida, Missouri-Columbia; Labs: Ames, BNL,  
 LBNL, ORNL, PNNL, SRNL