



# Improved, Low-Cost, Durable Fuel Cell Membranes

**2008 Hydrogen Program Annual Review** 

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#### Timeline

- Start Date: Sept. 30, 2007 A) Durability
- End Date: Sept. 30, 2010 •
- % Complete: 20%

### **Budget**

- **Total Funding** 
  - DOE: \$6,278k
  - Partners: \$1,569k
- FY2007 Funding Received

• \$0k

 FY2008 Funding Received •\$392k

#### **Barriers Addressed**

- B) Cost

#### **Partners**

- **Project Lead** 
  - Arkema Inc.
- Partners
  - Virginia Tech
  - Oak Ridge National Lab
  - Johnson Matthey Fuel Cells
  - University of Hawai'i
    - Hawai'i Natural Energy Institute (HNEI)





- To develop a membrane capable of operating at 80°C at low relative humidity (25-50%).
- To develop a membrane capable of operating at temperatures up to 120°C and ultra-low relative humidity of inlet gases (< 1.5 kPa).</li>
- To elucidate ionomer and membrane failure and degradation mechanisms via ex-situ and in-situ accelerated testing.
  - Develop mitigation strategies for any identified degradation mechanism.



# Milestones – Low RH Membranes

Task #	Milestone #	Title	Project Months	Project Deliverable	Go/No- Go Decision
1	1	Membrane meets 80°C requirements	6		X
	2	80°C membrane delivered to JMFC	8	Х	
	3	Membrane meets > 100°C requirements	18		Х
	4	> 100°C membrane delivered to JMFC	21	Х	
	5	Membrane manufacturing plan complete	36	Х	
2	6	MEA meets 80°C requirements	15		Х
	7	Large MEAs 80°C delivered to HNEI	18	Х	
	8	MEA meets $> 100^{\circ}$ C requirements	27		Х
	9	Large MEAs > 100°C delivered to HNEI	30	Х	
3	10	80°C MEAs perform per DOE requirements	27		Х
	11	> 100°C MEAs perform per DOE	36		Х
		requirements			
4	12	Post-mortem analysis of MEAs complete	36	Х	
5	N/A	Reporting, Planning, Administration	As Required		

#### • Task 1 work in progress

• Develop new membrane candidates



# Background

#### Polymer blend

- Decouples conductivity from other requirements
- Kynar<sup>®</sup> PVDF
  - Chemical and electrochemical stability
  - Mechanical strength
- Polyelectrolyte
  - H<sup>+</sup> conduction and water uptake
- Robust blending process
  - Compatible with various polyelectrolytes
  - Morphology and physical property control
- Lower cost approach compared to PFSA
  - Kynar<sup>®</sup> PVDF commercial product
  - Polyelectrolyte hydrocarbon based
- M41 highly sulfonated polyelectrolyte
  - Maximize conductivity at high RH





# M41 Physical Properties

	PFSA	M41
Dry Thickness (µm)	25	25
Equivalent Weight (g/(mol H <sup>+</sup> ))	1100	800
Specific gravity (g/cm <sup>3</sup> )	1.8	1.5
Water Uptake (%) <sup>a</sup>	37	60
X,Y Swell (%)	15	20
Thickness Swell (%)	14	10-15
Tensile Stress Break (MPa) <sup>b</sup>	19	27
Elongation (%)	103	95
Tear Strength(Ib <sub>f</sub> /in) <sup>c</sup>	404	934
Tear Propagation (Ib <sub>f</sub> ) <sup>d</sup>	0.004	0.018

M41 shows equal/better mechanical properties than PFSA



D882 D1004 D1938

### Background – M41 Creep Testing



 M41 shows a significantly larger resistance to flow compared to PFSA (9% vs. 140%)



# **Background - M41 Transport Properties**

#### Equivalent proton conductivity compared to Nafion<sup>®</sup>



#### Superior gas barrier property than Nafion<sup>®</sup> membranes



by 4-point in-plane AC measurements in water at 70°C

by electrochemical method at 80°C with 100% RH



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# Fuel Cell Testing: BOL Performance



Comparable in-cell performance to Nafion<sup>®</sup> 111 demonstrated



# Background – OCV Durability



- Short resistance decreases for PFSA and M41 membranes
- No fluoride and low sulfate emission from M41
- H<sub>2</sub> cross-over remains very low at failure for M41
- Fluoride emission and H<sub>2</sub> cross-over from PFSAs



# RH Cycling Durability: Gas Crossover



- Nafion<sup>®</sup> NRE-211 failed at approximately 6,000 cycles
- M41 and PFSA 111-IP MEAs met target of 20,000 cycles



# Voltage Cycling Durability: OCV to 0.4 V



- M41 membrane exhibits longer voltage cycling lifetime
- PFSA membranes allow higher current at 0.4V
- H<sub>2</sub> cross-over for PFSA <u>and</u> M41 MEAs at failure



# 400cm<sup>2</sup> M41 Testing – Preliminary Data

- 400 cm<sup>2</sup> (active area) MEAs fabricated by JMFC
- Preliminary data obtained in UTC Power hardware



- M41 MEA H<sub>2</sub>/air, 65 °C, 80/60% utilization, ambient pressure
- Pt loading: Anode: 0.2 mg/cm<sup>2;</sup> Cathode: 0.4 mg/cm<sup>2</sup>



### **Development of Low RH Membranes**

- Project initiated Sept 30, 2007
- Objectives
  - To develop a membrane capable of operating at 80°C at low relative humidity (25-50%).
  - To develop a membrane capable of operating at temperatures up to 120°C and ultra-low relative humidity of inlet gases (< 1.5 kPa).</li>
  - To elucidate ionomer and membrane failure and degradation mechanisms via *ex-situ* and *in-situ* accelerated testing.
    - Develop mitigation strategies for any identified degradation mechanism.



# M41 Low RH *Ex-situ* Testing

FSEC





Additional improvement required to meet low RH targets



1E+03

# Low RH Performance of M41-Based MEA

#### • Inlet RH



Significant performance loss at <65% RH on anode <u>and</u> cathode



# Low RH Performance of M41-Based MEA



• With only one-side humidified, reasonable performance is obtained (at ≥25% RH)

ARKeme

### **Thickness Effect**

• 0.7 mil M41 – better performance at low RH



### Low RH Membranes: Approach

- M41 good scaffold
  - Good MEA performance at >65% RH
  - 'Bridge the gap' @ lower RH operation
- Blending is transparent to the polyelectrolyte
- 1) Improved M41 production process: M43 membrane
- 2) Analogous approach to phosphoric acid-imbibed membranes
  - Hypothesis: Incorporating bound phosphonic acid groups will increase water retention at low RH
- 3) New polyelectrolytes
- 4) Control morphology of Kynar<sup>®</sup> blends
  - Vary the hydrophobicity/hydrophilicity of Kynar/polyelectrolyte blends
  - Process control
- 5) Additives



### M43 Initial Results





# **New Membrane Generation**

- Polyelectrolyte with phosphonic acid groups
  - M41: Highly sulfonated polyelectrolyte

• M51

• 1/4 of sulfonates replaced with phosphonate

• M52

½ of sulfonates replaced with phosphonate

• M53

- ¾ of sulfonates replaced with phosphonate
- Reoptimized PVDF blending parameters
- Produced new membranes (lab-scale)
- Collaboration initiated
  - Prof. V. Ramani (Ilinois Institute of Technology)



# MEA Testing – Phosphonated Membranes





# Summary

- M41 shows superior durability in accelerated in-situ testing
- M41 MEAs shown to operate down to 65% RH (inlet)
- M41 architecture is a good platform for low RH membranes
- New grant targets low RH performance
  - Process improvements show initial gain
  - Phosphonated materials did not show improved performance
- New membrane production, screening, and testing is underway
  - Varying membrane chemistry and/or morphology



# **Future Work**

- Investigation of structure/property/RH relationships
- Approaches
  - Improved blending process
  - Collaboration with Prof. J. McGrath (Virginia Tech)
  - New Arkema polyelectrolyte / Kynar blends
- *Ex-situ* and *in-situ* testing of new membranes
- Validation and optimization of *in-situ*, low RH performance
- Durability testing
- Elucidation of failure mechanisms



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