2008 DOE Hydrogen Program Review Sulfur-lodine Thermochemical Cycle

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Project PD25

Sulfur-Iodine Thermochemical Cycle Project Overview

Timeline

- Start 4/2006
- Finish 4/2009
- ~ 80% complete

Barriers

- Materials high temperature, corrosive environments
- Process chemistry, catalysts, diagnostics
- Reactor-process coupling

Budget

- Funding
 - FY06- 5.5 M\$
 - FY07 4.3 M\$
 - FY08 2.9 M\$
 - French CEA in kind

Partners

- INERI Project with CEA
- Process CEA, SNL, General Atomics
- Supporting Technologies INL, UNLV

Sulfur-Iodine Thermochemical Cycle Objectives

- Evaluate the potential of the Sulfur-Iodine cycle for hydrogen production using nuclear energy
 - DOE/NHI CEA INERI Project (CEA, SNL, General Atomics)
 - Sulfur cycles potential for high efficiency, relative development level
 - Approach construct, operate an Integrated Lab Scale (ILS) experiment to investigate the key technical issues
 - Provide basis for nuclear hydrogen technology decisions

Phase 1 (Cycle Evaluation)

FY03 – 05 - Process chemistry options, flowsheets, materials, small scale experiments

Phase 2 -- (Integrated Lab Scale Experiment - ILS)

- FY06 Develop the 3 major reaction sections for S-I at 3 locations
- FY07 Integrate 3 sections at General Atomics experiment site
- FY08 Conduct integrated lab scale, closed loop experiments

NHI Sulfur Based Thermochemical Cycles Sulfur-Iodine



Sulfur Iodine

- (1) $H_2SO_4 \rightarrow H_2O + SO_2 + 1/2O_2$
- $(2) \qquad 2HI \rightarrow I_2 + H_2$
- (3) $2H_2O + SO_2 + I_2 \rightarrow H_2SO_4 + 2HI$

Sulfur-Iodine Integrated Lab Scale Experiment ILS Approach





Sulfur-Iodine ILS Experiment

Status

S-I Integrated Lab Scale Experiment at GA Site



- Interface skid allows independent operation of each section, facilitates startup, shutdown
- Lexan enclosures, separate ventilation system for each section
- Initial integrated operations underway March 2008

Technical Accomplishments/ Progress Overview

- HI decomposition (GA)
 - First hydrogen production experiment using CEA Bunsen section feed material ~50 liters/hr - April 2008
 - Iodine extraction, HI distillation, decomposition, recycle demonstrated
 - Diagnostics, control, flow systems developed and tested, reliable

• Bunsen reactor section testing (CEA)

- First Bunsen reaction test completed produced and separated HIx (heavy) phase and H₂SO₄ (light) phase acids
- Subsequently processed by HI section to produce H₂
- SO₂/O₂ separation section operated reliably in multiple operations
- H₂SO₄ decomposition experiments (SNL)
 - Acid vaporization, decomposition, and recuperation in one integrated SiC bayonet unit, no corrosion issues identified in multiple test series
 - Experiments completed at 850 C, ambient to 5 bar, ~300 l/hr SO₂ at 40 mole %, SO₂ conversion at ILS flow rates ~90% of theoretical
 - Operated in integrated mode with CEA Bunsen section March, 2008
- Next steps diagnostics and equipment modifications, and complete FY08 testing program

Sulfuric Acid Decomposition Section SiC Integrated Decomposer



Sulfuric Acid Decomposition Section Results

SiC Bayonet Decomposer (ILS)

- ~ 300 l/hr SO₂ production rate @ 850 C (10 • moles/hr, 40 mole% concentration)
- Conversion rates ~90% of theoretical •
- SO₂ production rate limited by heat transfer to • catalyst region
- Pressure, flow rate dependence evaluated ٠
- Acid decomposer operation reproducible ٠ through ~20 cycles
- Catalyst durability requires further • development





Sulfuric Acid Decomposition Catalysts INL H₂SO₄ Catalyst Studies

- Catalyst stability for extended operation remains a key issue
- Supports studied: SiO₂, γ-Al₂O₃, ZrO₂, α-Al₂O₃ and TiO₂, Pt/TiO₂ most stable in short term tests
- Some complex metal oxides show high activity above 825°C
- Stability of some complex metal oxides appeared promising
- Further investigation of complex metal oxides anticipated in FY09



WHSV = 50 g acid/g cat./hr

WHSV = 2,000 g acid/g cat./hr, 850°C

SO₂ yields with temperature



Section 3- HI Decomposition Overview



HI Decomposition section (section 3) installed at GA ILS site



Extractive distillation method

- H_3PO_4 separates HI/H_2O from I_2
- HI distilled from H₃PO₄,
- HI decomposed at ~ 400 C to H_2 and I_2
- Undecomposed HI recycled to HI reactor
- I₂ returned to Bunsen reactor section
 Key Issues
- Uncertainty in HI/I₂/H₂O VLE
- Materials corrosion, catalysts

Section 3- HI Decomposition Section First HI section H2 run with Bunsen section feed



HI Recycle Effect (5 bar)

Bunsen Section Status (CEA)



CEA Counter-current Bunsen Reactor Section



CEA Bunsen Section assembled at GA experiment site.

Bunsen Section Status (CEA)

Counter-current Bunsen reactor – initial run



System Analyses

FY07 NHI Baseline Process Cost Projection Update

NHI Baseline Cycle	H2 Cost (\$/kg)	Efficiency (H2A Analysis)
Sulfur-Iodine (extractive distillation)	3.41	40
Sulfur-Iodine (reactive distillation)	3.05	39
Hybrid Sulfur	2.94	43
High Temperature Electrolysis	3.22	44

• Efficiencies based on available flowsheets

Cost of energy

- Electricity = 0.06 \$/kWhr

- Thermal = 0.02 \$/kWhr

Other assumptions

- Low temp electrolysis 660 \$/kWe
- High temp electrolysis 500 \$/kWe

Cost Analyses (H2A)

- Development of H2A cost framework for NHI processes (TI, MPR)
- Update capital, operating cost estimates, performance based on updates from process teams
 - Consistency, uncertainty review at MPR (FY08), H2A analyses for updated cost estimate
 - FY07 update (preliminary draft)

Conventional Atmospheric Alkaline Electrolysis
 Conventional SMR (w/o CO2 removal)
 Advanced SMR (with CO2 removal)

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Sulfur Iodine Thermochemical Cycle Planned Activities - FY08 - FY09 (tbd)

• FY08 – Complete Phase 2 test program in ILS apparatus

- Process improvements, equipment modifications
 - Level diagnostics for Bunsen reactor achieve flowsheet light and heavy acid phase characteristics,
 - Smaller HI reactor to facilitate section 3 flow control,
 - Interface unit diagnostics for extended operation
- Establish baseline operational data for integrated operations, HI extractive distillation, and counter current Bunsen reactor

FY09 – TBD – examine improved /alternate process options

- Complete investigation of operational characteristics and performance for current configuration, extended operations
- Consider reactive or co-current options
- Complete assessment of S-I for nuclear hydrogen

Sulfur-Iodine ILS Experiment Project Summary

Relevance: This project is providing the technical information needed to assess the potential of the Sulfur lodine thermochemical cycle for large scale production of hydrogen using Generation IV reactors. Results from this project will support the DOE FY2011 technology decision for the NGNP hydrogen production technology.

Approach: Flowsheet analysis identified process options for the S-I cycle. Laboratory experiments evaluated process options and provided the basis for the design and testing of the 3 major reaction sections of the S-I cycle.

Technical Accomplishments: The 3 major reaction sections of the S-I cycle have been assembled at the integrated lab scale experiment site and initial testing has been initiated. SNL has completed testing of a SiC bayonet sulfuric acid decomposer section at the GA site and conducted initial integrated operations with the CEA. The CEA has completed the first Bunsen reactor test, producing and separating the heavy (HI) and light (H2SO4) phase acids for subsequent processing. GA has completed the initial H2 production runs on the HI extractive distillation and decomposition section.

Tech Transfer/Collaboration: The S-I cycle research is conducted as an INERI project with the French CEA. There is also collaboration with Universities (chemical analyses, materials). The DOE sponsored work will be a major component in the Generation IV International Forum (GIF) nuclear hydrogen collaboration signed in March FY2008.

Future Research: The focus in FY08 will be to complete the initial series of ILS test in the current configuration. FY09 is TBD but the possibility of examining advanced process options is a primary consideration. Research on improved catalysts will also be conducted.