

Dimensionally Stable Membranes™

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Project ID # fc_02_mittelsteadt

Dimensionally Stable Membranes for High Temperature Applications

Timeline

- Begin 4/3/2006
- Review 4/2/2009
- 65% Complete

Budget

- Total project funding (to 2009)
 - \$949K DOE Funding
 - \$589K Recipient
 - 37% Cost Share
 - \$350K received FY 2008
 - \$65K Remaining (through Feb '09)
 - \$871K DOE funds spent to date

Barriers addressed

- A. Durability
- B. Cost

Technical Targets (DOE 2010 Targets)

- 0.10 S/cm at 1.5 kPa H₂O Air inlet
- <\$40/m²
- > 5000 h lifetime
- Stability in Condensing conditions

Partners

- General Motors
- SUNY-ESF

OVERVIEW

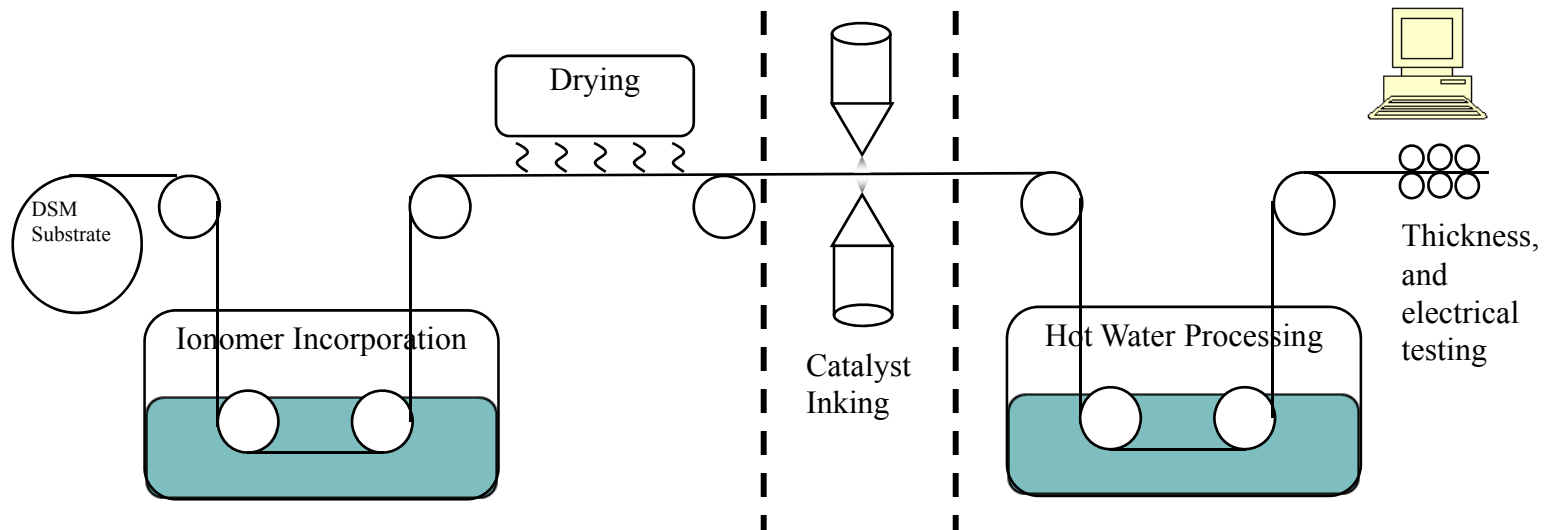
- OBJECTIVES
- APPROACH and ACCOMPLISHMENTS
 - Rationale
 - Two Dimensionally Stable Membranes™
 - Three Dimensionally Stable Membranes™
- CHALLENGES

OBJECTIVES

YEAR	OBJECTIVE
2006	<p>Determine the effect of pore size and substrate thickness on conductivity and water uptake.</p> <p>Demonstrate polymerization conditions suitable for bulk polymerization of the PFSA.</p>
2007	<p>Demonstrate, by the 3rd Quarter, membrane conductivity of 0.07 S/cm at 80% relative humidity at room temperature using non-Nafion materials. Samples will be prepared and delivered to the Topic 2 Awardee.</p>
2008	<p>Go/No-Go Decision: Demonstrate, by the 3rd Quarter, membrane conductivity > 0.1 S/cm at 25% relative humidity at 120°C using non-Nafion® materials. Samples will be prepared and delivered to the Topic 2 Awardee.</p>

OBJECTIVES: Ultimate Goal

Meet performance targets with film that can be generated in roll at DOE cost targets



OVERVIEW

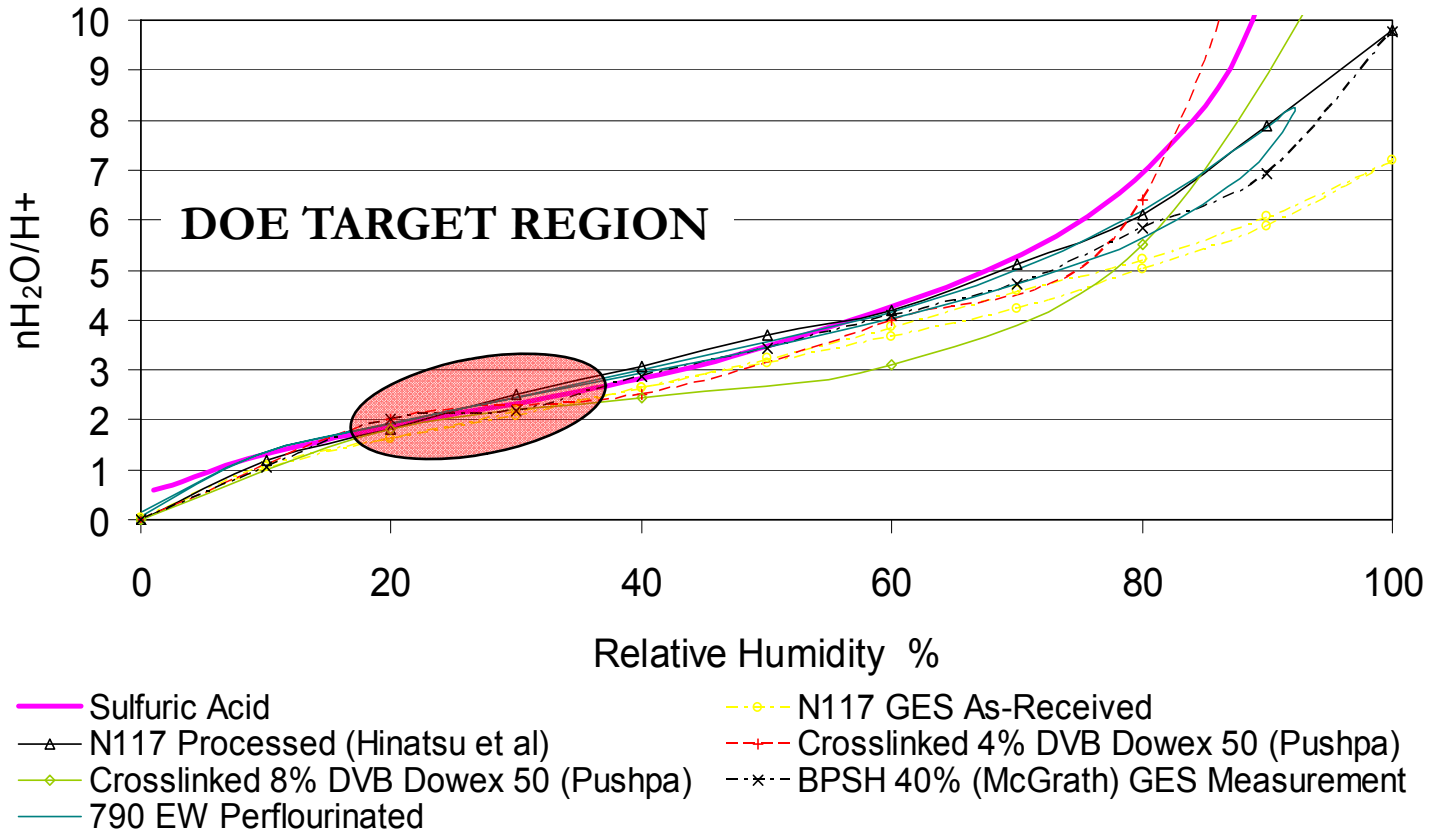
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APPROACH: Rationale

- Limitations of Ionomers based on $-\text{SO}_3\text{H}$ functionality
 - Water uptake/retention as a function of RH
 - Conductivity Limitations
 - Dependence on Water
 - Functionality

APPROACH: Rationale

Water Uptake of Ionomers based on $-\text{SO}_3\text{H}$ Moiety

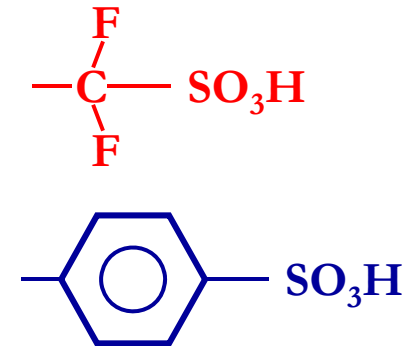
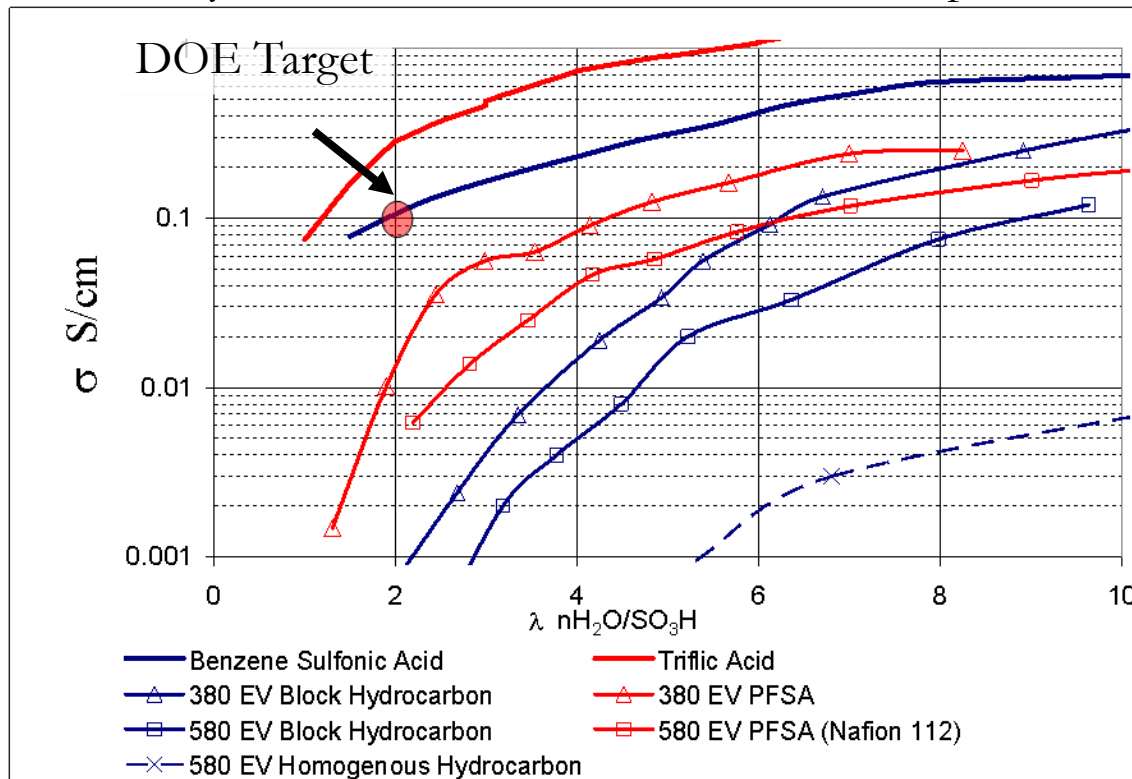


Water Content is the same regardless of pendant group

APPROACH: Rationale

Importance Of Ionic Functional Group, Morphology

Conductivity of Various Ionomers and Model Compounds at 80°C

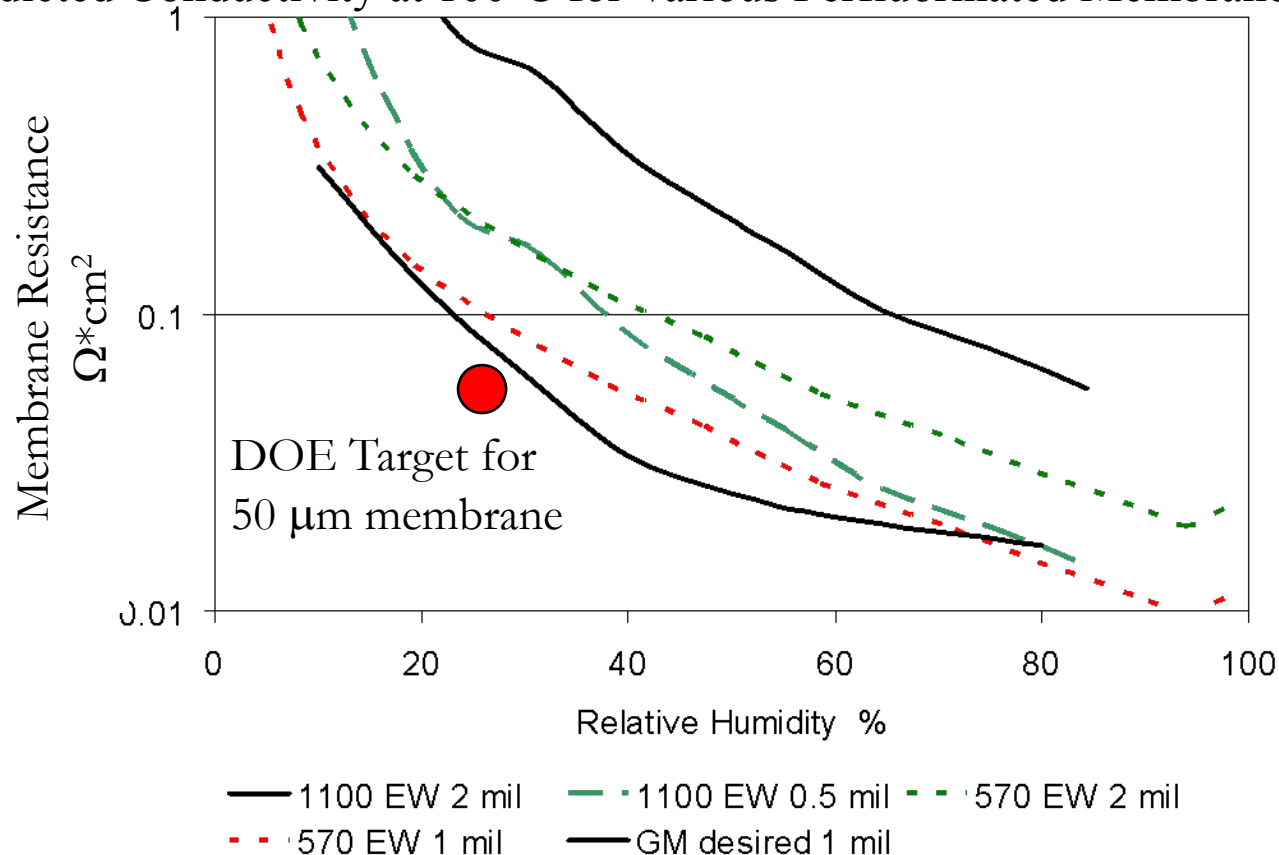


$$\text{EV} = \frac{\text{EW}}{\text{Density}}$$

APPROACH: Rationale

Limitations of Ionomers Based on $-SO_3H$ Moiety

Predicted Conductivity at 100°C for Various Perfluorinated Membranes



APPROACH: Rationale

CONCLUSIONS

SO₃H Polymers will need

- Very low EW
- Perfluorinated End Groups
- To be very thin

THESE THREE REQUIREMENTS LEAD TO POOR MECHANICALS

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 - Three Dimensionally Stable Membrane™
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Approach: Lower EW of perfluorosulfonic Acid ionomers to increase low RH conductivity and support the ionomer with two and three-dimensional non-ionic materials

- Two Dimensionally Stable Membrane™

- Generate Supports
 - Thickness and Pore Size
- Incorporate Ionomers
 - 700 to 1100 EW PFSA
- Characterize
 - Performance
 - Durability
 - Cost/Manufacturability

Mag:700 kV:20 plasma clean, bottom surface 10 μm

- Three Dimensionally Stable Membrane™

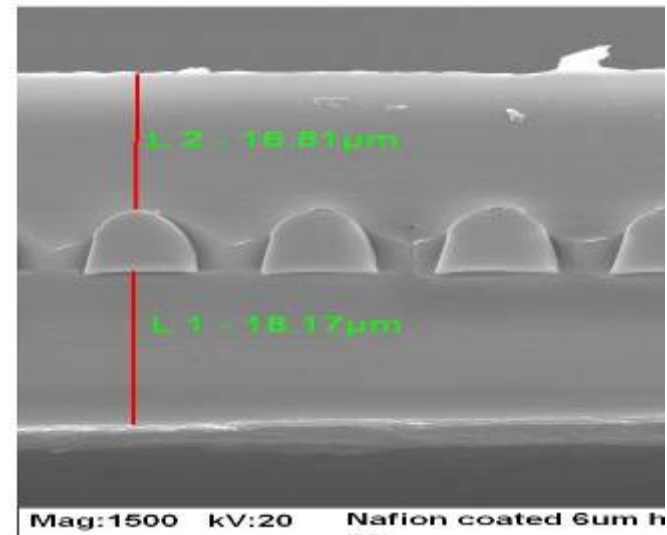
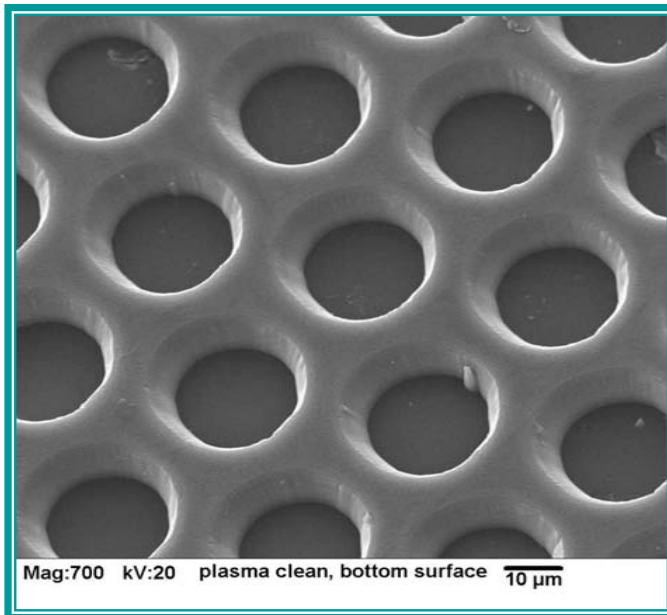
- Develop Bulk Polymerization Methods
- Polymerize in Selected Supports
- Characterize
 - Performance
 - Durability
 - Cost/Manufacturability

APPROACH: 2DSM™:

- COMPOSITE POLYMER ELECTROLYTE MEMBRANES

DSM™ is high acid content membrane reinforced with high strength polymer support

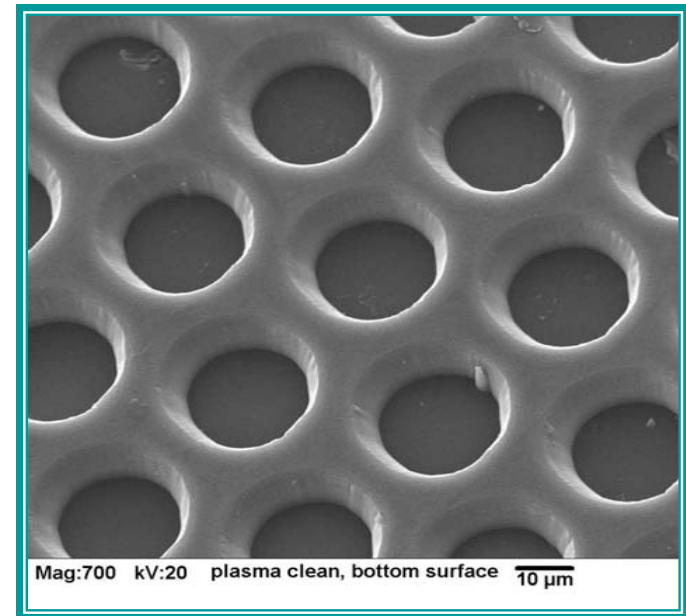
< 1 Mil Diameter hole
Nearly 1,000,000 holes/in²



APPROACH: 2DSMTM

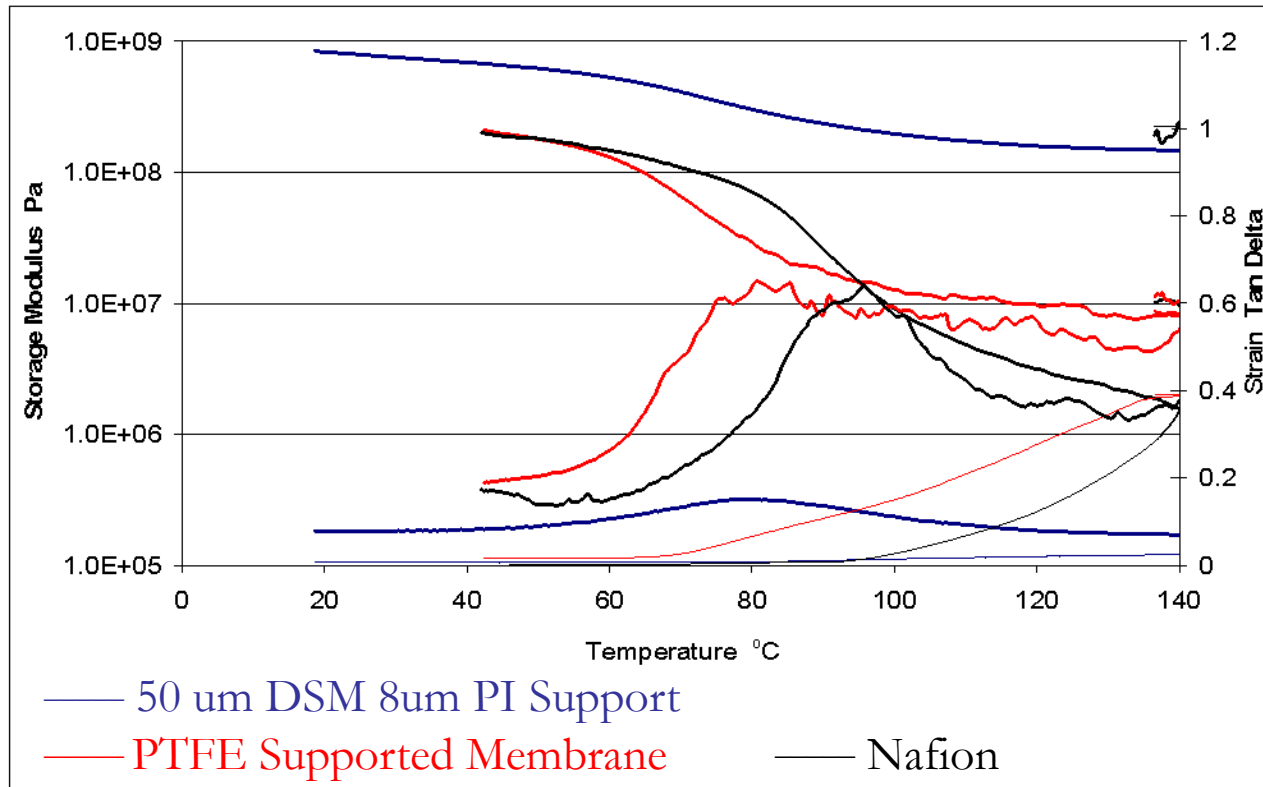
Dimensionally Stable MembraneTM Support Structures Used to Date

Material, condition	Young's Modulus (Mpa)
Nafion [®] 112 Dry 20°C	300
Nafion [®] 112 Wet 80°C	70
Poly(tetrafluoroethylene) (PTFE)	400
Polysulfone	2600
Poly(etherether-ketone) (PEEK)	2700
Polyimide (PI), e.g., Kapton [®]	2900



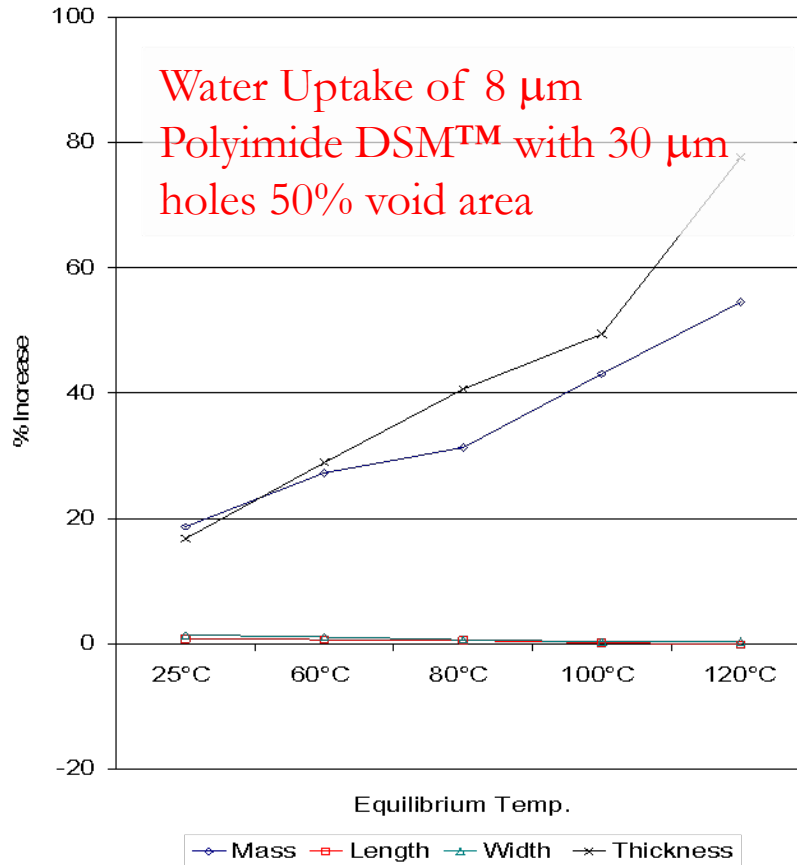
ACCOMPLISHMENTS: 2DSM™

Increased Modulus of Support Material Leads to Proportional Increase in Composite Material



ACCOMPLISHMENTS: 2DSM™

Effect of Pore Size on Water Uptake

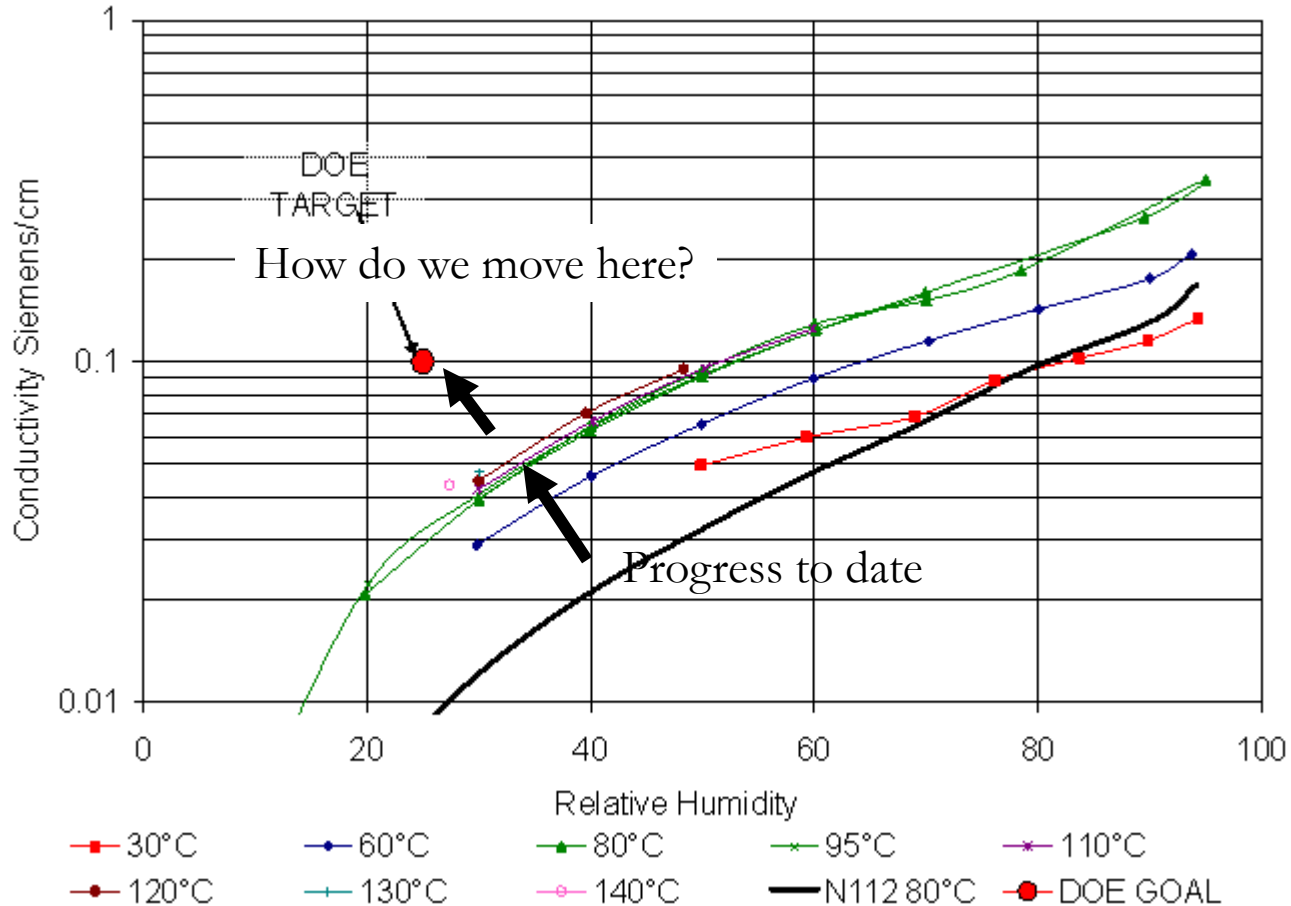


- No macroscopic swelling on the X-Y plane
- *Effect is the same for*
 - *pore sizes of 10-40 μm*
 - *Support thickness of 8-25 μm*
 - *EW of fill of 700-1100 EW*
- Membrane retains integrity even after tested at 120°C in water.
- *SIGNIFICANTLY Improves, handling and storage of membranes*
- *We were not able to generate 75% void area materials*

ACCOMPLISHMENTS: 2DSM™

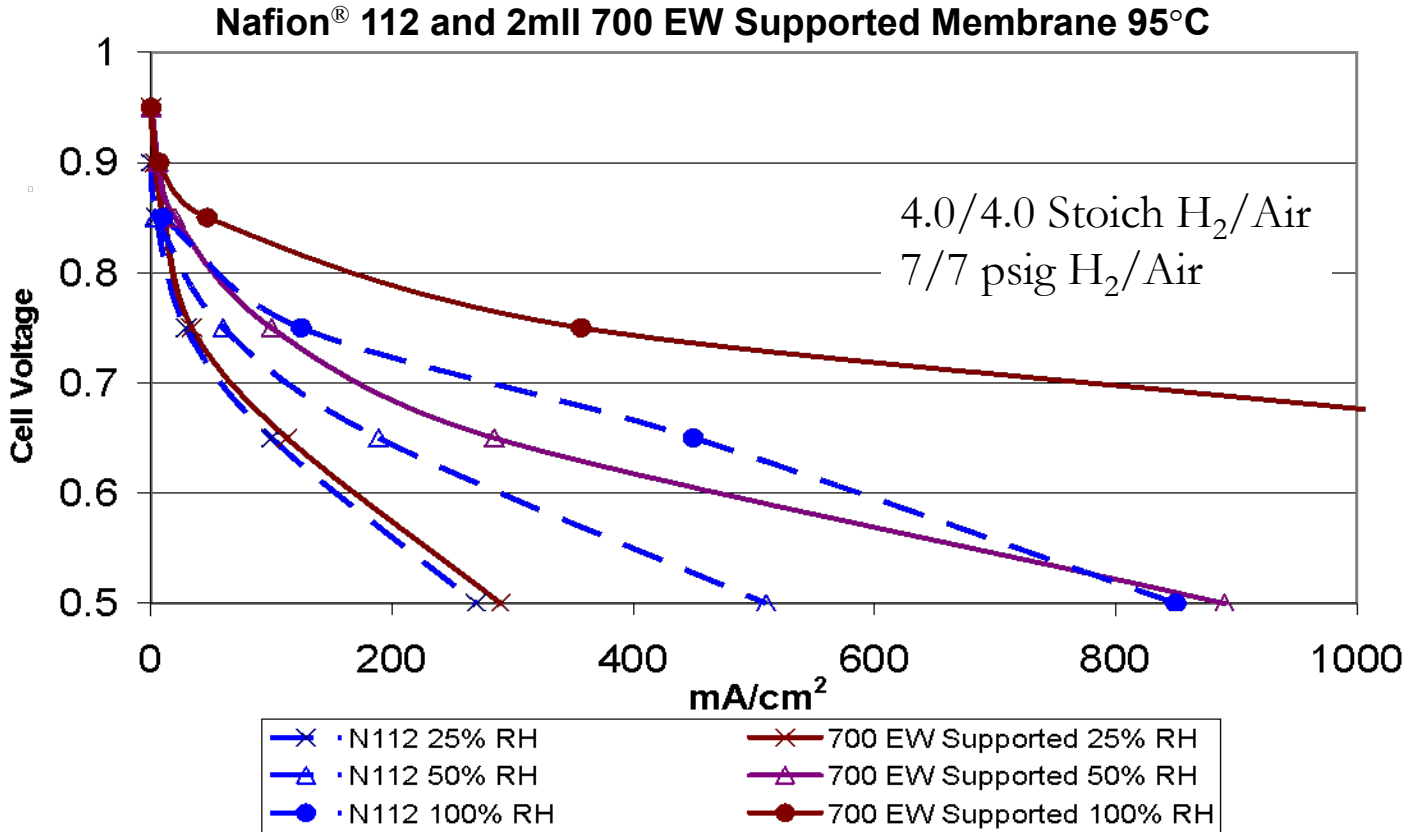
In-Plane Conductivity

700 EW Membrane with DSM™ Support, Conductivity as a function of RH



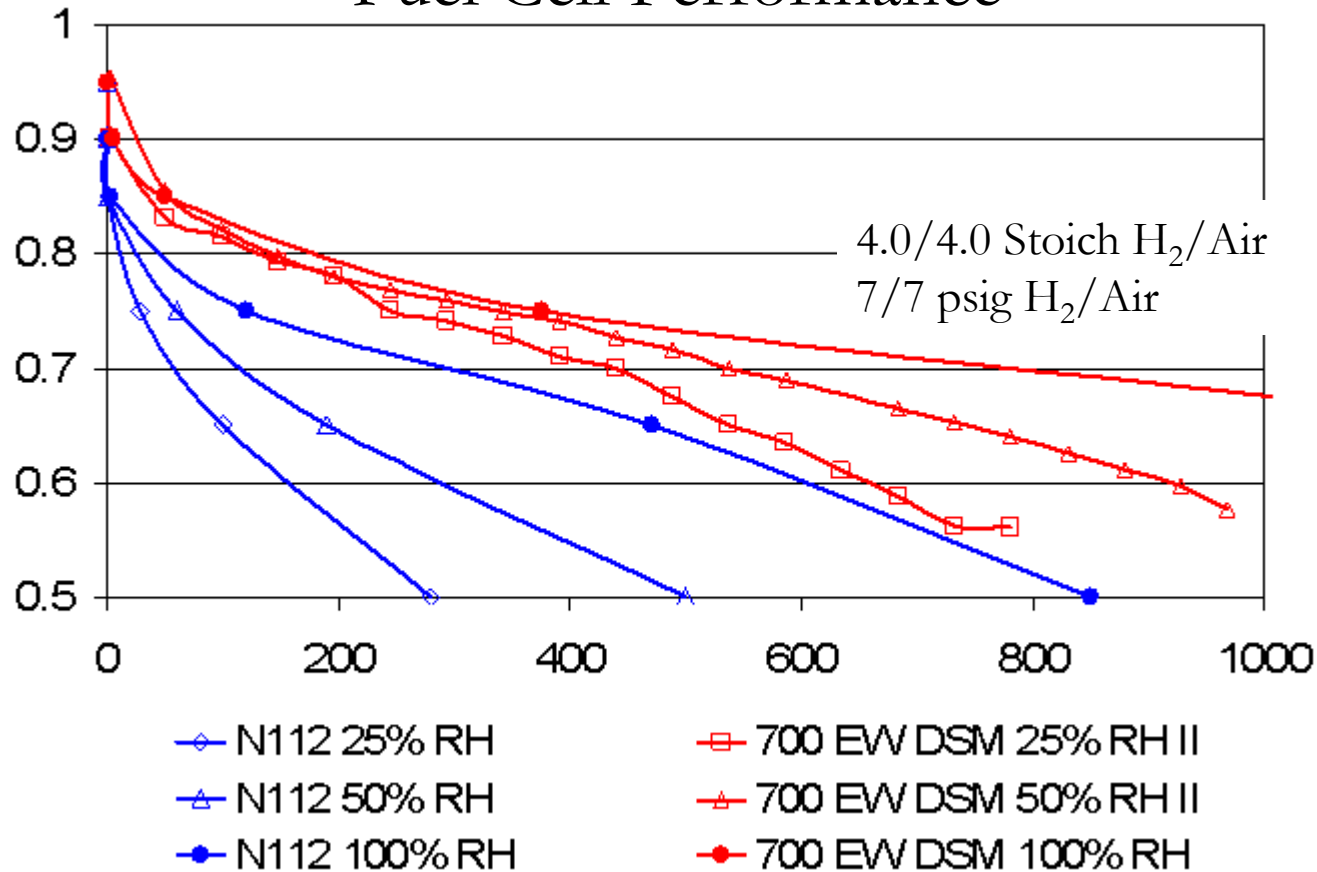
ACCOMPLISHMENTS: 2DSM™

Fuel Cell Performance



25% RH data much worse than predicted based on DSM™ conductivity and high-frequency resistance: Work to do on the catalyst layer

ACCOMPLISHMENTS: 2DSM™ Fuel Cell Performance

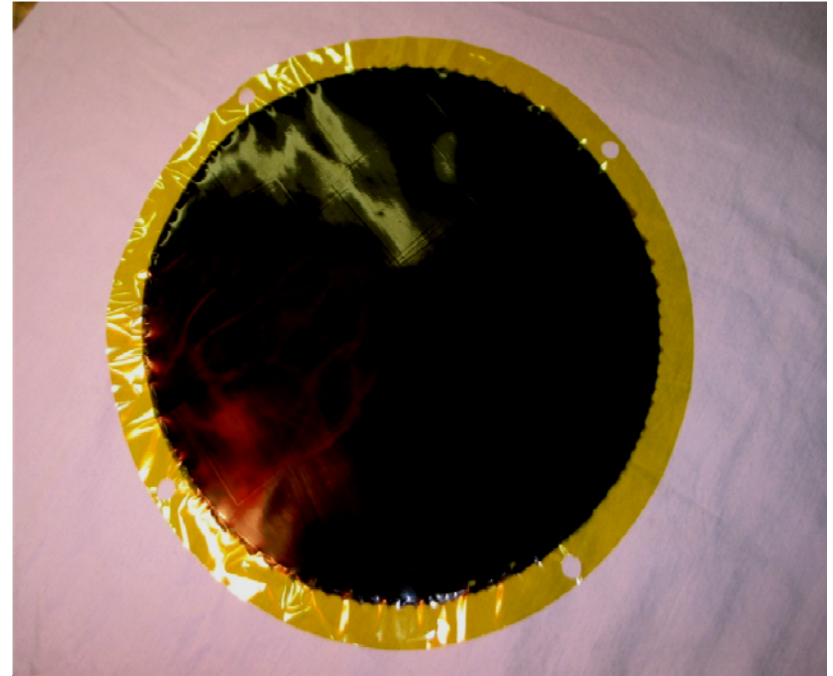


25% RH data much worse than predicted based on DSM™ conductivity and high-frequency resistance: Work to do on the catalyst layer

ACCOMPLISHMENTS: 2DSM™

MEA Fabrication

- Generated 200 cm² MEA
- Catalyst Applied Directly To Membrane
- **No Decal Transfer**

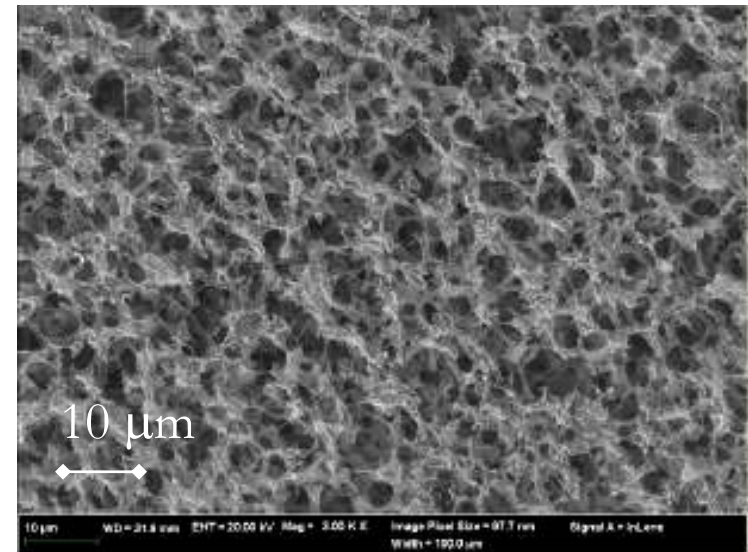


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- CHALLENGES

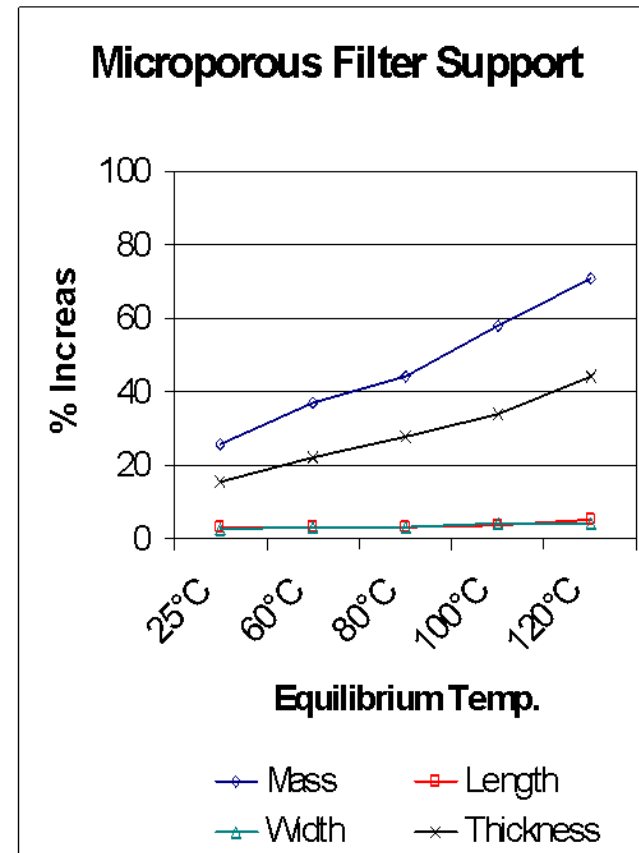
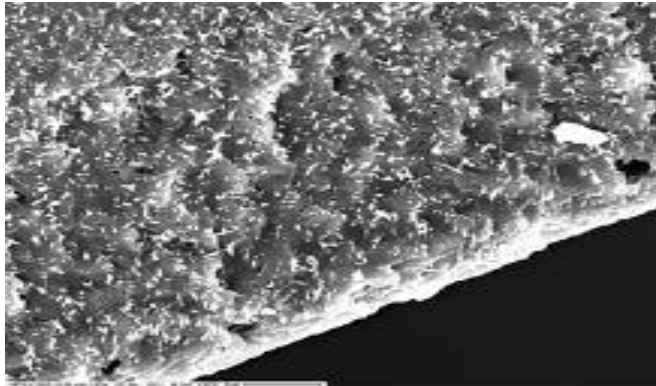
APPROACH: 3DSM™

- Cost of Laser Drilling Prohibitively High
- Three-Dimensional Supports Commercially Available
- Using Conventional PFSA Ionomers to Meet Cost Targets
- Synthesizing New Low EW Ionomers to Meet Performance Targets
 - *Work being done by Israel Cabasso's Group at SUNY Syracuse Polymer Research Institute*

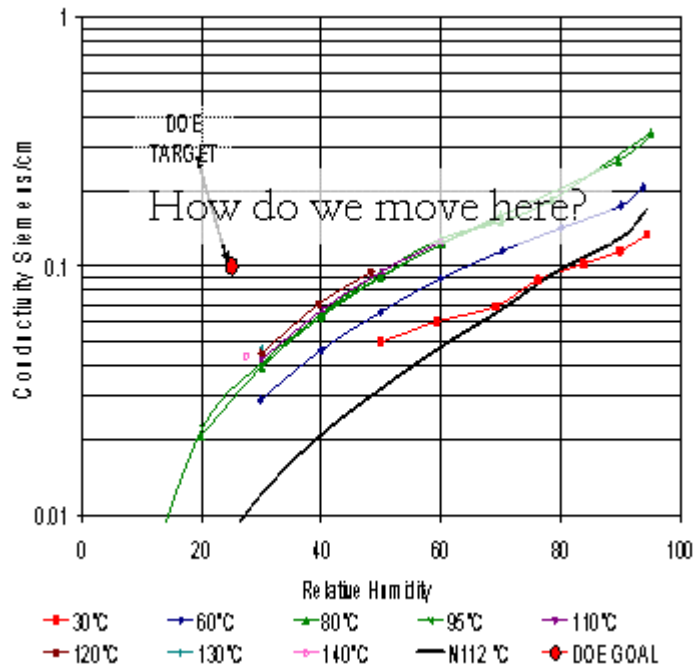


-Commercial PFSA

- Filled Supports With Commercial PFSA Material
- Swelling is Comparable to 2DSM™
- Completely Filling Support was a Major Challenge
- ~33% Conductivity Penalty



APPROACH: 3DSM™ Alternative Synthesis



Synthesis of new PFSA Monomers
Working with SUNY-ESF to make new
copolymers

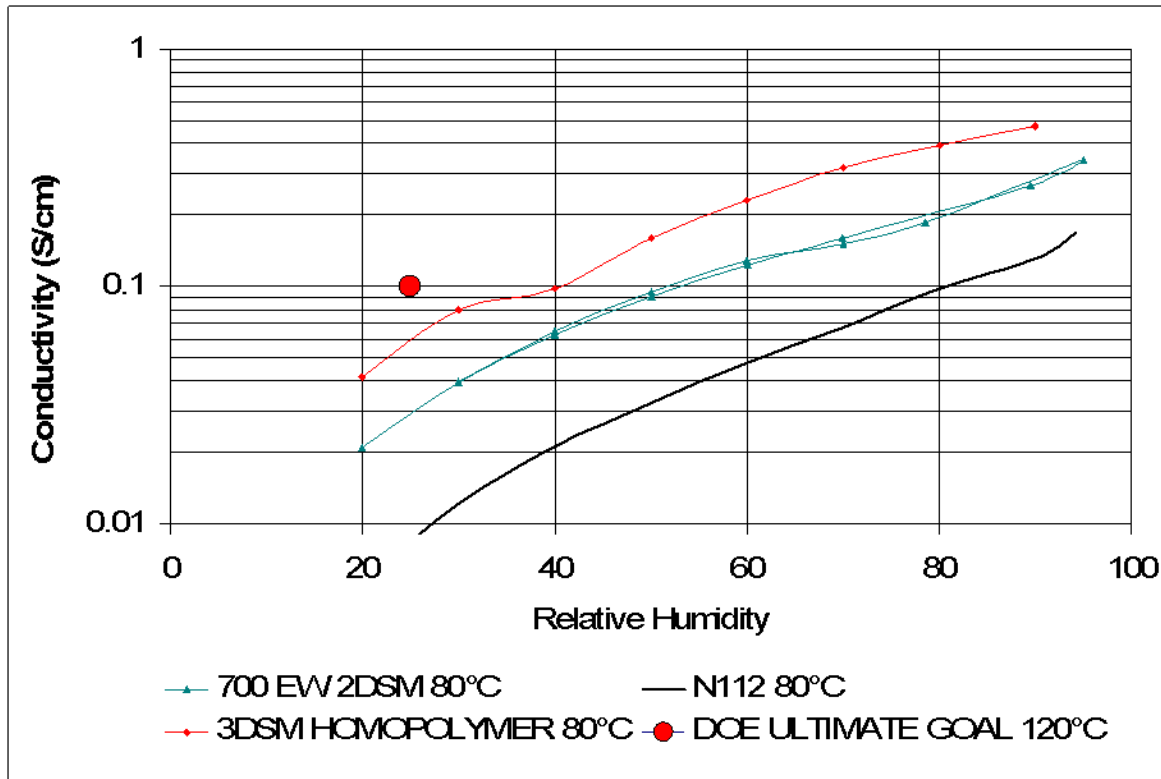


+ ?

Three non-PTFE copolymers
successfully synthesized. EW from
500-1300.

ACCOMPLISHMENTS: 3DSM™

Alternative Synthesis: Homopolymer



Synthesis of new PFSA Monomers
Working with SUNY-ESF to make
new copolymers and homopolymer

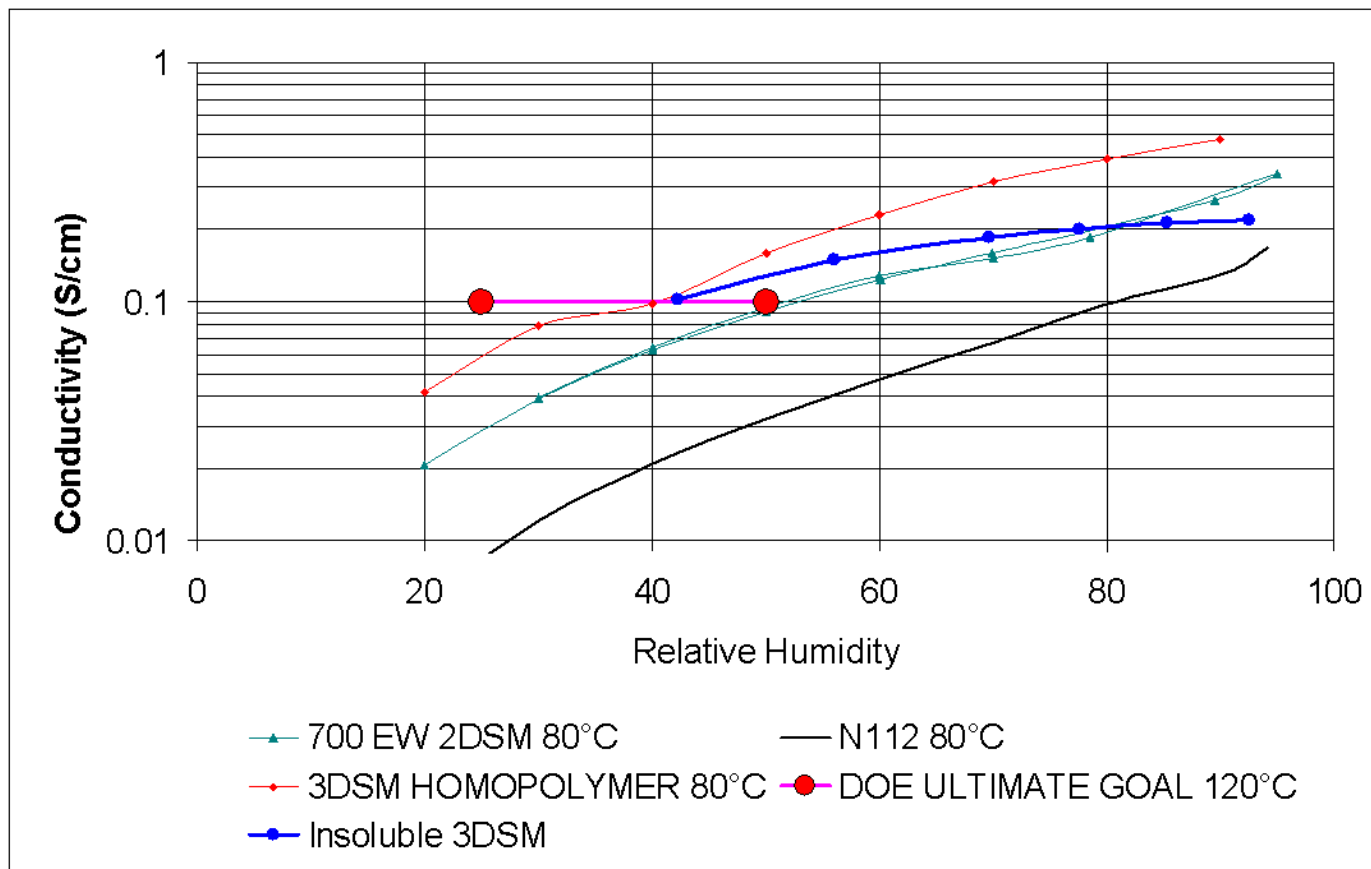


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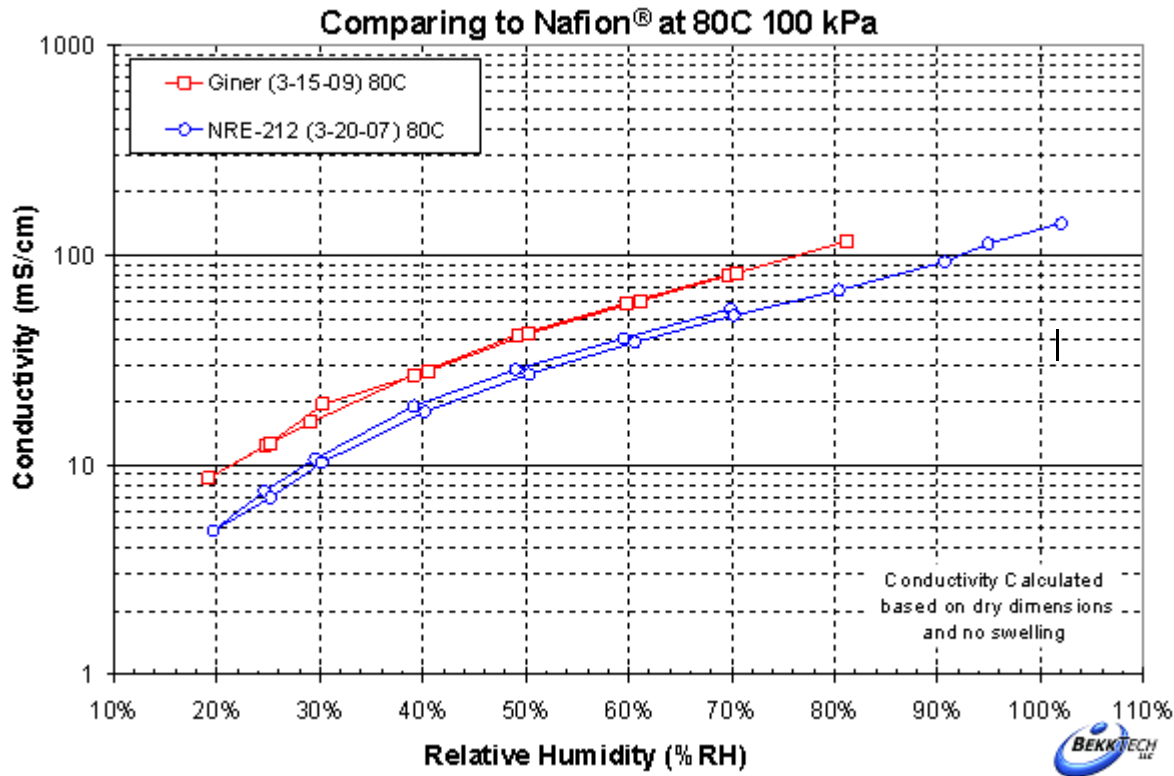
Getting closer to DOE target! This polymer is water soluble

Accomplishments 3DSM™: New Insoluble low EW PFSA



Insoluble High Performing 3DSM™ Generated

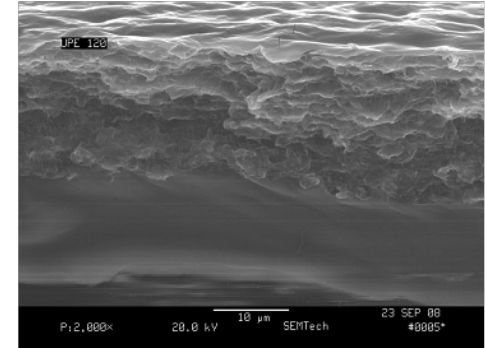
Accomplishments 3DSM™: New Insoluble low EW PFSA



Bekktch testing considerably under our own
Material is inconsistent
Our data matched well for past samples

How to make thinner?

- Commercially available supports are ~ 2 mils in thickness after filled and sintered.
- Commercial supports have “skin” from casting process, which is a resistive barrier
- Working with Millipore on thinner, skinless supports



Material/ Pore Size	Ionomer EW	Thickness mils	Thru-Plane σ N1100 = 1
A/1.2 μ m	1100	2.6	0.6
B/1.2 μ m	1100	3.9	0.9
C/0.45 μ m	1100	1.0	0.5
D/0.6 μ m	1100	0.9	0.1
E/0.6μm	1100	0.45	0.7

Accomplishments:

2DSM™ and 3DSM™ RH Cycling

- Three Membranes Tested (all 2 mils)
 - 3DSM™ with 1100 EW
 - 2DSM™ with 700 EW
 - 2DSM™ with 700 EW
 - Extruded Nafion® 112
- All 4 passed 10,000 Cycles
 - 80°C
 - 2 min 95°C dew point
 - 2 min bone-dry
 - 1SLPM

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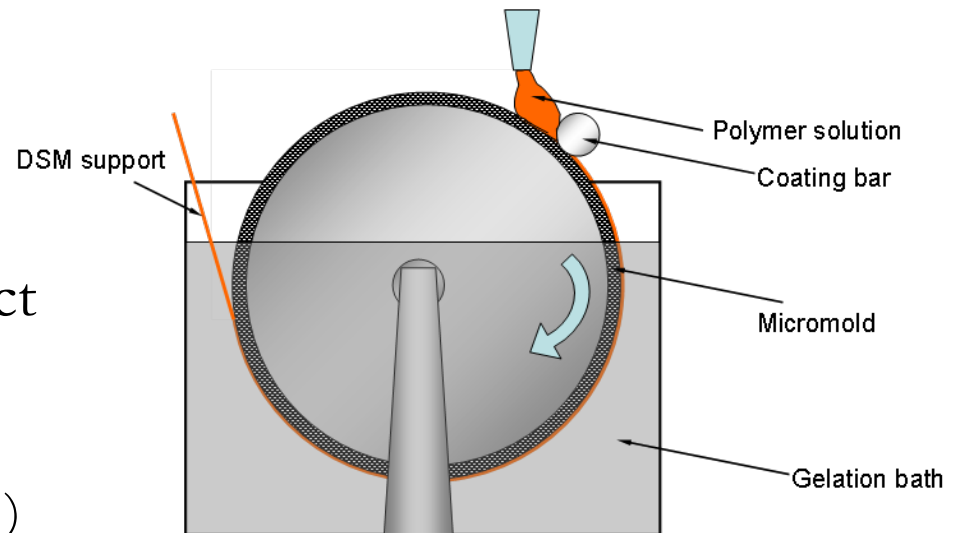
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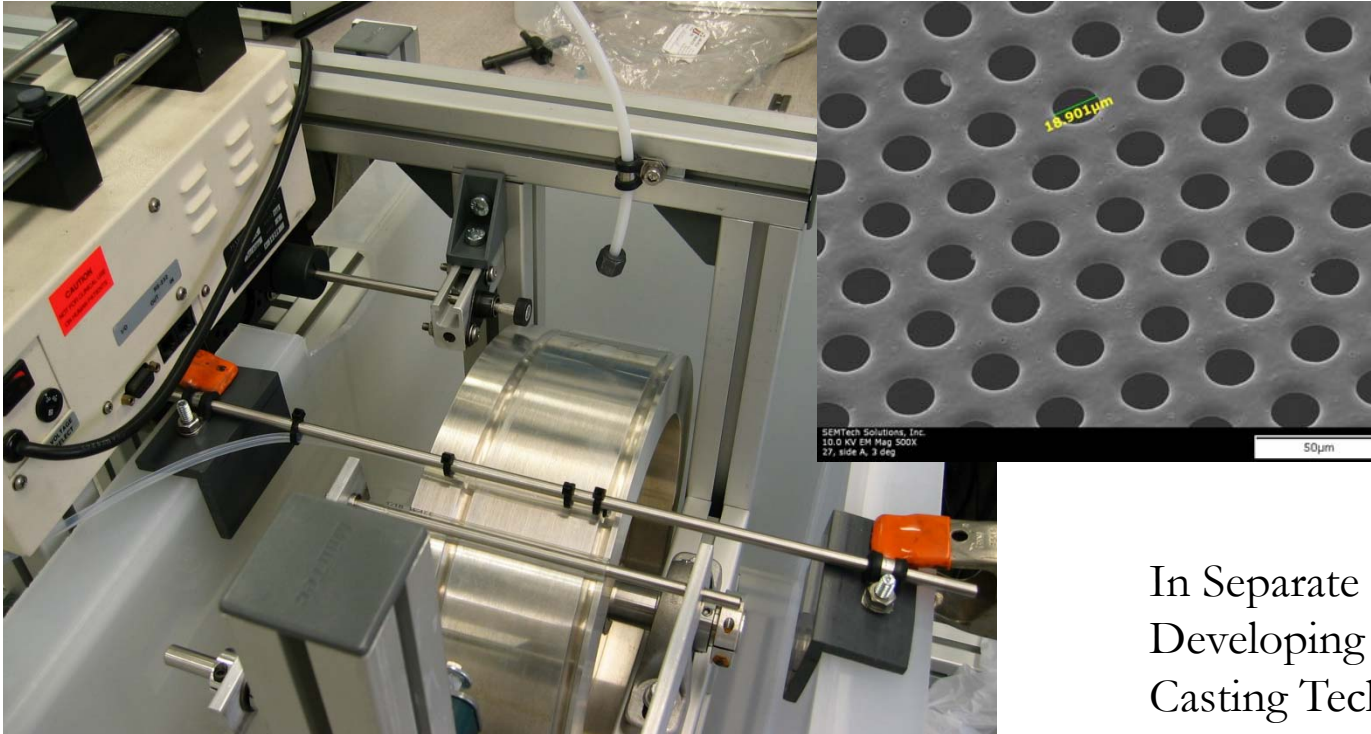
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- OTHER OPPORTUNITIES
- CHALLENGES

Challenges: 2DSM™

- Biggest Challenge is Cost
 - Laser Drilling
 - Currently $\sim \$1/\text{cm}^2$
 - Projected $\sim \$0.02/\text{cm}^2$
 - $\$200/\text{m}^2$
 - In Separate DOE Project Developing Low-Cost Casting Technique
 - (DE-FG02-05ER84322)
- *Currently testing continuous roll process*



Challenges: 2DSM™



Automatic solution dispensing
Automatic speed control
Manual support collection
Solution recycling can be implemented

In Separate DOE Project
Developing Low-Cost
Casting Technique
(DE-FG02-
05ER84322)
PI: Han Liu

Challenges: 3DSM™

- Incorporate lower EW ionomers into support
- Minimize resistance penalty due to “skin”
- Demonstrate improved fuel cell performance

Proposed Future Work: Thinner, Faster, Stronger

- Need to down-select for last two years
 - 2DSM™
 - 3DSM™
- Develop along that pathway and demonstrate
 - Cost feasibility
 - Fuel Cell Performance
 - RH Cycling Durability

SUMMARY

- Year 1 Milestones Achieved
 - DSMs with a wide range of pore size and thickness restrain x-y swelling
 - Polyimide and polysulfone both shown to be effective supports
 - Effective methods of generating new PFSA polymers have been generated
- Year 2 Milestones Achieved
 - Conductivity targets have been met
 - Discrepancy between Bekktech and GES results
 - Fuel Cell Performance Improvements Shown
 - Electrode Improvements
- Realistic Pathways for Meeting Cost Targets Seen for both Paths
- To reach ultimate DOE Goals we will need to continue improving the low EW materials that have been developed at SUNY

OBJECTIVES: Ultimate Goal

Meet performance targets with film that can be generated in roll at DOE cost targets

