



DOE Hydrogen Program

# New Polymer/Inorganic Proton Conductive Composite Membranes for PEMFC

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PENNSSTATE



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**The Pennsylvania State University**  
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Project ID#:  
FC\_03\_Ivov

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## Timeline

- Project start date: *May 1<sup>st</sup>, 2006*
- Project end date: *April 30<sup>th</sup>, 2009*
- Percent complete: *100%*

## Budget

- Total project funding:
  - >>DOE share: *\$900K*
  - >>Contractor share: *\$325K*
- Funding received in FY08: *\$300,000*

## Barriers

- **Durability**  
Chemical and mechanical stability of PEMs
- **Performance**  
High temperature, low RH proton conductivity
- **Cost**

## Partners

- **Prof. S. Lvov's group** –  
*The Energy Institute's Electrochemical Lab, PSU*
- **Prof. M. Chung's group** –  
*Department of Materials Science and Engineering, PSU*
- **Prof. S. Komarneni's group** –  
*Materials Research Institute, PSU*
- **BekkTech LLC** –  
*Fuel Cell Testing & Diagnostic Services*

# Project Objectives and Tasks

## Objectives

### *The main project objectives of the last year:*

- Development of a proton conductive inorganic/polymeric membrane for PEMFC operating at elevated temperatures up to 120 °C and significantly reduced relative humidity
- Durability, improved fuel cell performance

### Task 1

**Synthesis and characterization of functionalized polymeric materials suitable for the desirable composite membranes.**

### Task 2

**Synthesis and characterization of inorganic proton conductive materials suitable for the desirable composite membranes.**

### Task 3

**Fabrication and characterization of composite membranes and MEAs for PEMFC in automotive applications.**

# Project Relevance to DOE Objectives

This membrane development effort addresses the DOE research and development objective to provide membranes that allow operation at high temperatures. High-temperature polymer electrolyte membranes (PEMs) are desirable for transportation applications (to facilitate thermal management) as well as for stationary applications (higher value heat in combined heat and power and/or CO-tolerance).

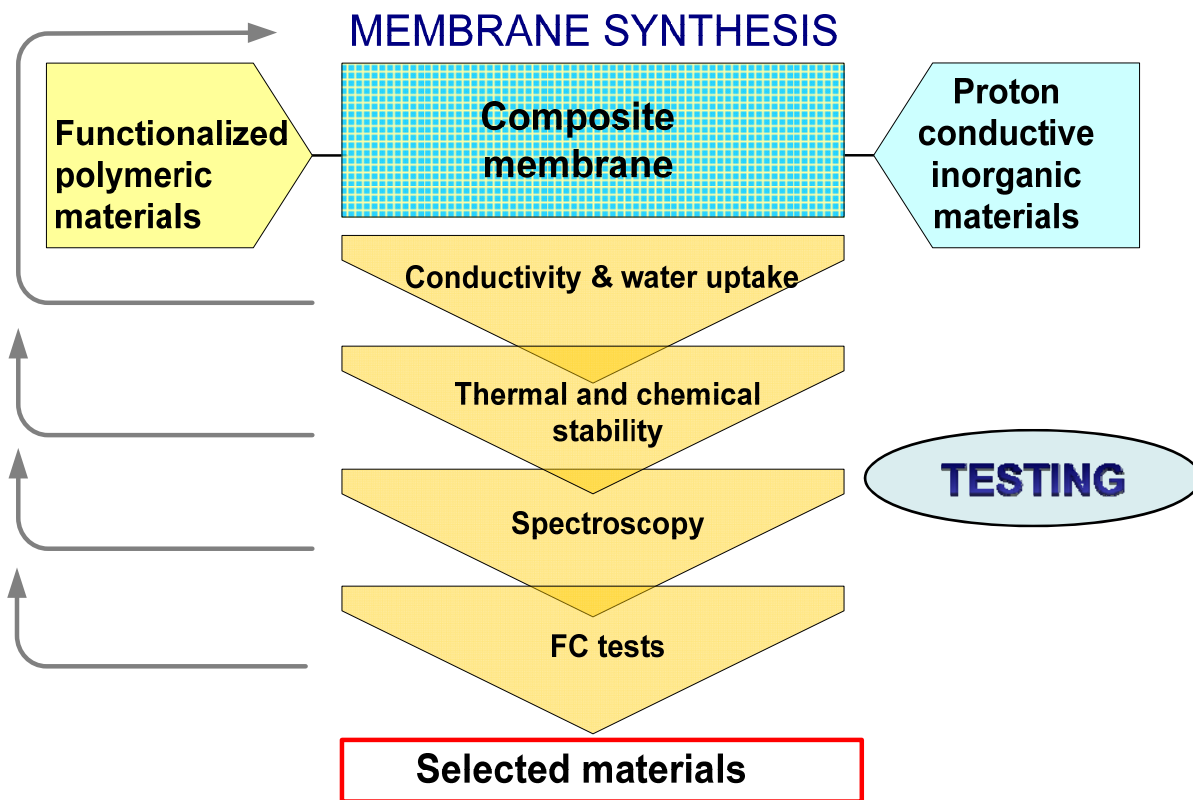
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The project addresses DOE conductivity milestones: 70 mS/cm at 30°C, 80% RH and 100 mS/cm at 120°C, 50%RH and DOE high-temperature membrane 2010 and 2015 targets: a membrane capable of operating at inlet water vapor partial pressure <1.5 kPa.

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The project focuses on the barriers in the development of PEMs such as high temperature low RH conductivity and durability.

# Approach



The unique aspect of our approach is the use of hydrophilic highly proton conductive additives to hold on to water more tightly than the ionomer and increase membrane conductivity at higher temperature and low RH. These additives provide water-rich surfaces inside the membrane to attract the hydrophilic portions of the ionomer.

The approach addresses the barriers in the development of PEMs such as high temperature low RH conductivity and durability and DOE conductivity milestone of 100 mS/cm at 120°C, 50%RH

## Task 1

Previously presented accomplishments include the synthesis of Sulfonated Styrene Grafted Fluoropolymer P(VDF-CTFE)-g-PS via atom transfer radical polymerization and membrane testing.

P(VDF-CTFE)-g-PS had high thermostability, high proton conductivity, exceeding Nafion® at 120°C and 50, 70% RH at high sulfonation degree. Substantial lowering of excessive swelling and improvement of membrane flexibility were reached by the introduction of hexafluoropropylene (HFP) and increase of molecular weight of VDF-CTFE copolymer (HMW) without of compromising conductivity (Fig. 1), but swelling was still very high in comparison with Nafion®

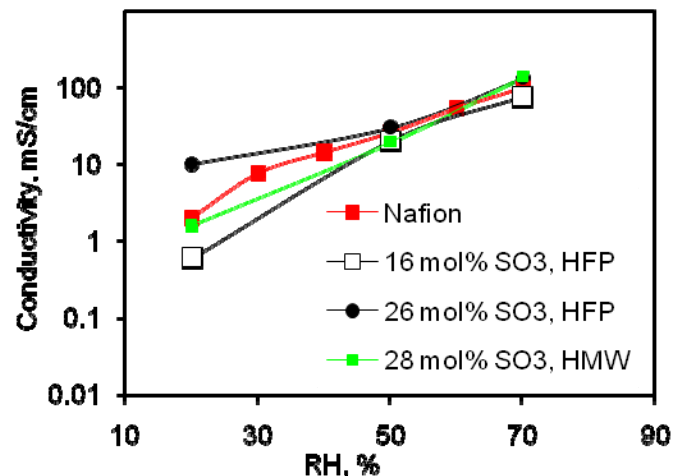
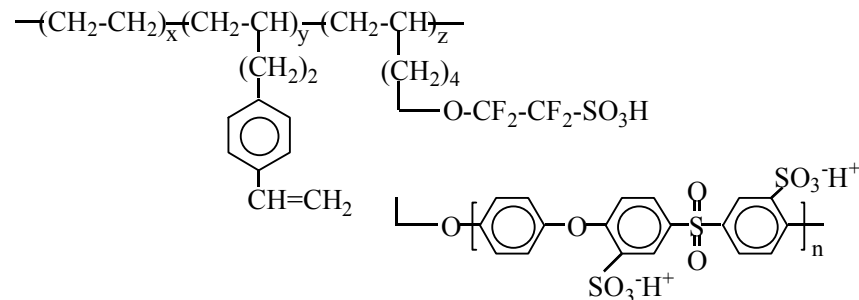


Figure 1. Conductivity of P(VDF-CTFE)-g-PS membranes with addition of HFP and using VDF-CTFE of HMW at 120°C and different RH

Synthesis of new polyolefin based proton conductors:

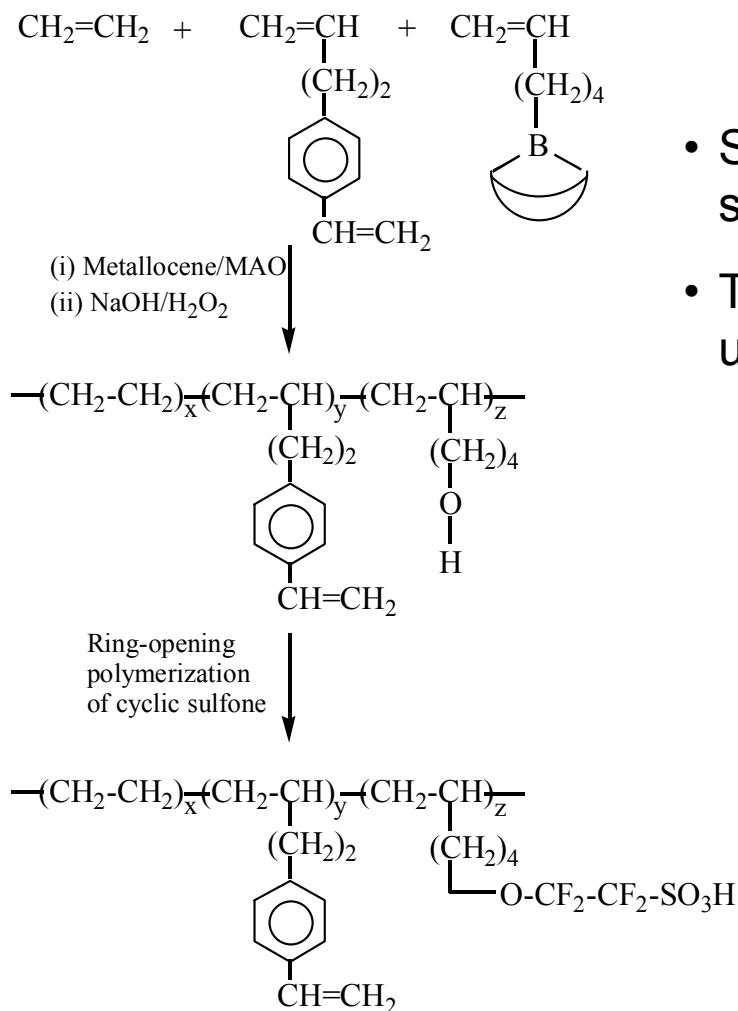
- very stable under free radical conditions
- semicrystalline thermoplastics with excellent mechanical strength
- have co-continuous hydrophobic and hydrophilic phase separated morphology
- cross-linkable



New polyethylene based proton conductor (PE)

## Task 1

## Synthesis of Sulfonated PE PEMs



- Sulfonated PE polymers successfully synthesized
- The major hurdle is in the processing of uniform PEM films

*New approach to prepare the hydroxylated and crosslinked PE film before sulfonation*

## Task 2

### Synthesis and Characterization of Proton Conductive Inorganic Materials

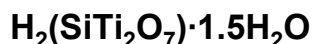
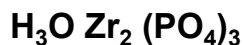
Synthesis of new inorganics with different structures was developed and a number of inorganics including sulfonic acid functionalized porous materials were synthesized and characterized by X-ray diffraction (phase purity) and SEM (morphology and particle size).

#### **Inorganic:**

##### Layered



##### 3-dimensional



##### 2-dimensional

SBA-15

MCM-41

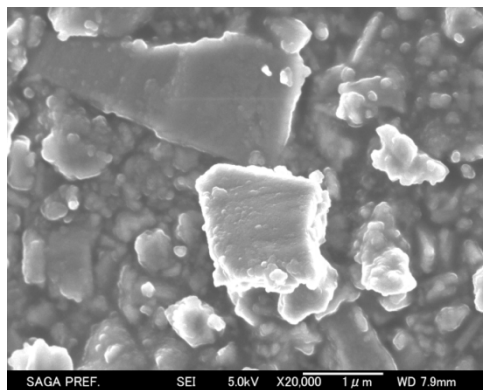
##### Sulfated

S-SiO<sub>2</sub>

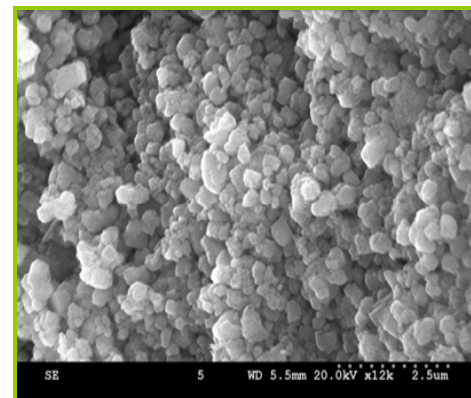
S-ZrO<sub>2</sub>

##### Phosphated

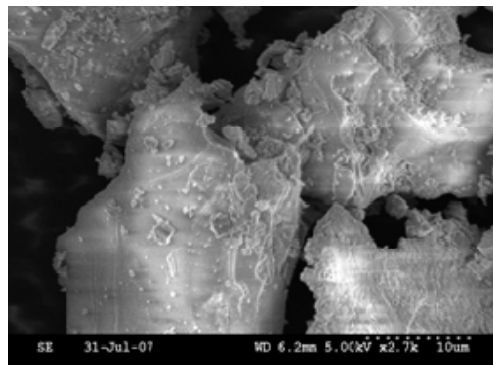
P<sub>2</sub>O<sub>5</sub>-SiO<sub>2</sub> gel



SEM Image of  $\text{Sn}(\text{HPO}_4)_2 \cdot \text{H}_2\text{O}$ .



SEM Image of  $\text{H}_2(\text{SiTi}_2\text{O}_7) \cdot 1.5\text{H}_2\text{O}$



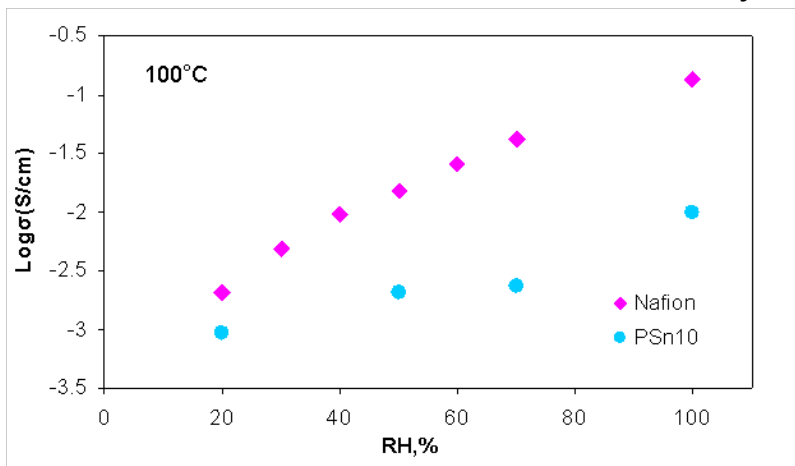
SEM Image of S-ZrO<sub>2</sub>



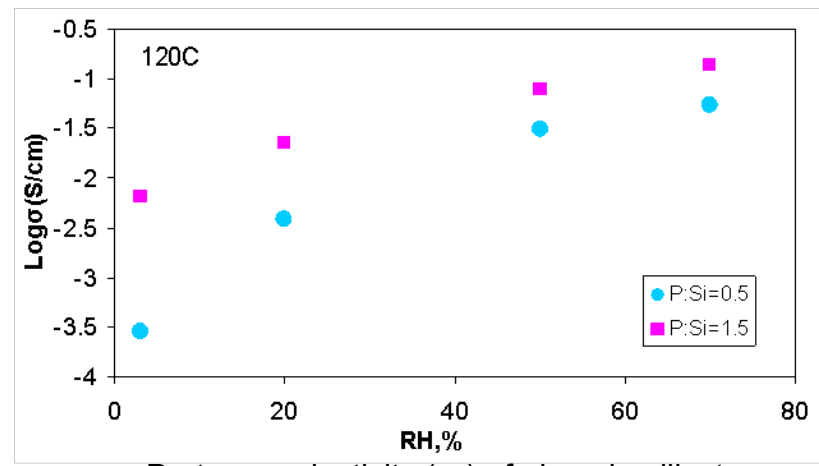
## Task 2

### Conductivity of New Inorganic Materials

- Two-electrode through-plane measurements, EIS
- Inorganic pellet sample preparation
- Characterization for porosity and BET surface area
- BekkTech conductivity hardware



Proton conductivity ( $\sigma$ ) of  $\text{Sn}(\text{HPO}_4)_2$  vs. Nafion<sup>®</sup> at 100°C and different RH.



Proton conductivity ( $\sigma$ ) of phosphosilicate gels at 120°C and 3% to 70% RH.

- $\sigma$  reached 0.01 S/cm at 120 °C
- low dependence on RH ( $\sigma$  changed within one order of magnitude a range of RH from 20% to 70%)
- the structure of stannic phosphate can provide water retention at 120 °C and reduced RH.

- $\sigma$  of gel with a P/Si mole ratio of 1.5 reached 0.1 S/cm at 120 °C 70%RH, which is higher than for Nafion<sup>®</sup>.

**The inorganics can be used in composite membranes to achieve DOE technical targets and milestone of 100 mS/cm at 120°C and 50% RH.**

## Task 3

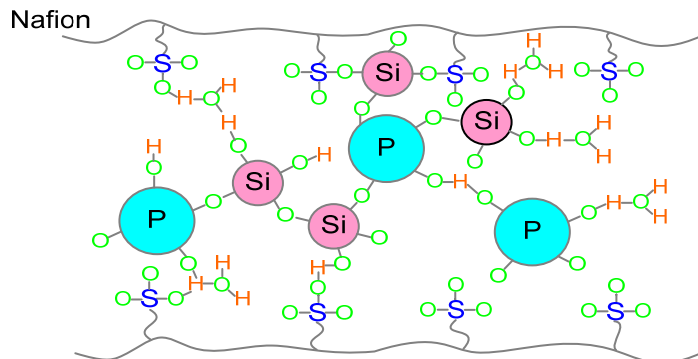
### Polymer/Inorganic Composite Membranes

In the last year, the effect of newly synthesized inorganic materials and membrane fabrication techniques on composite membrane conductivity was tested in Nafion®/Inorganic composite membranes to select the best candidates for future sulfonated PE/inorganic composite membranes to meet DOE conductivity milestone 100 mS/cm at 120°C, 50% RH and fuel cell needs on durability, mechanical and chemical stability

#### □ Inorganic tested

- Sulfated SiO<sub>2</sub>
- Sulfated ZrO<sub>2</sub>
- Phosphosilicate gel
- SBA - 15
- MCM - 41

#### □ Schematic of a Nafion®/phosphosilicate membrane



#### □ Preparation

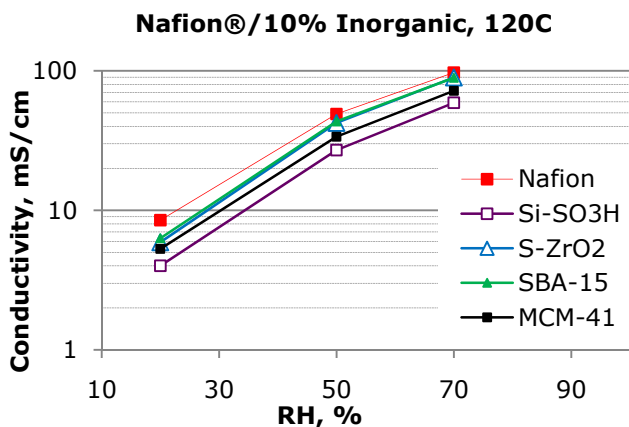
- Blending inorganic particles with 5% Nafion® solution followed by casting and annealing
- In situ formation of inorganic particles from precursor in a Nafion® film
- In situ formation of inorganic particles from precursor in 5% Nafion® solution followed by casting and heat treatment

## Task 3

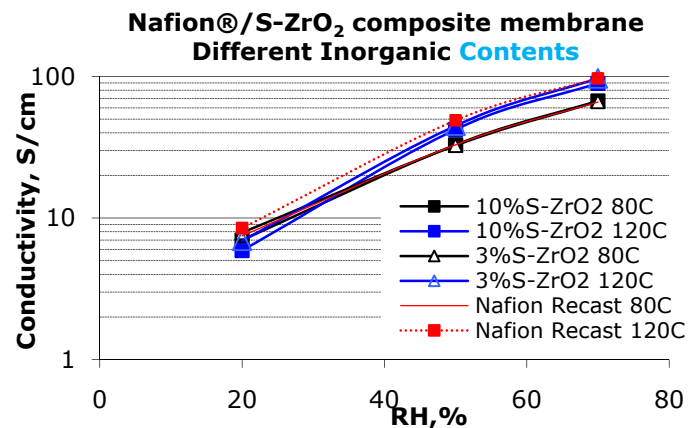
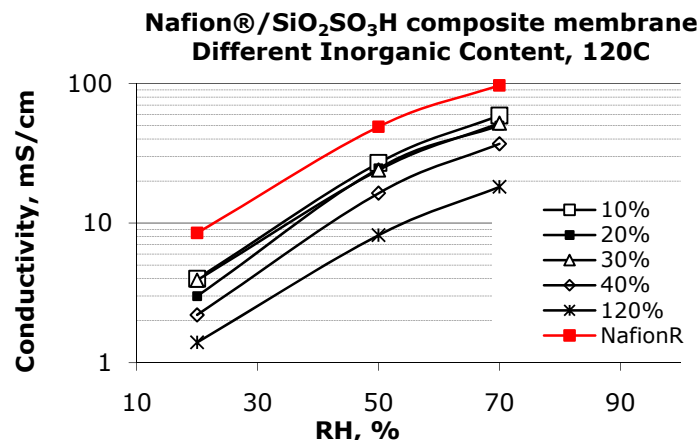
### Polymer/Inorganic Composite Membranes Blending technique Conductivity

- The nature of inorganic additive affects the conductivity of the composite membrane
- No enhancement in conductivity of polymer matrix was reached by addition of inorganic by blending technique.
- High doping level (>20%) lowered the conductivity.

#### Performance of Different Inorganics



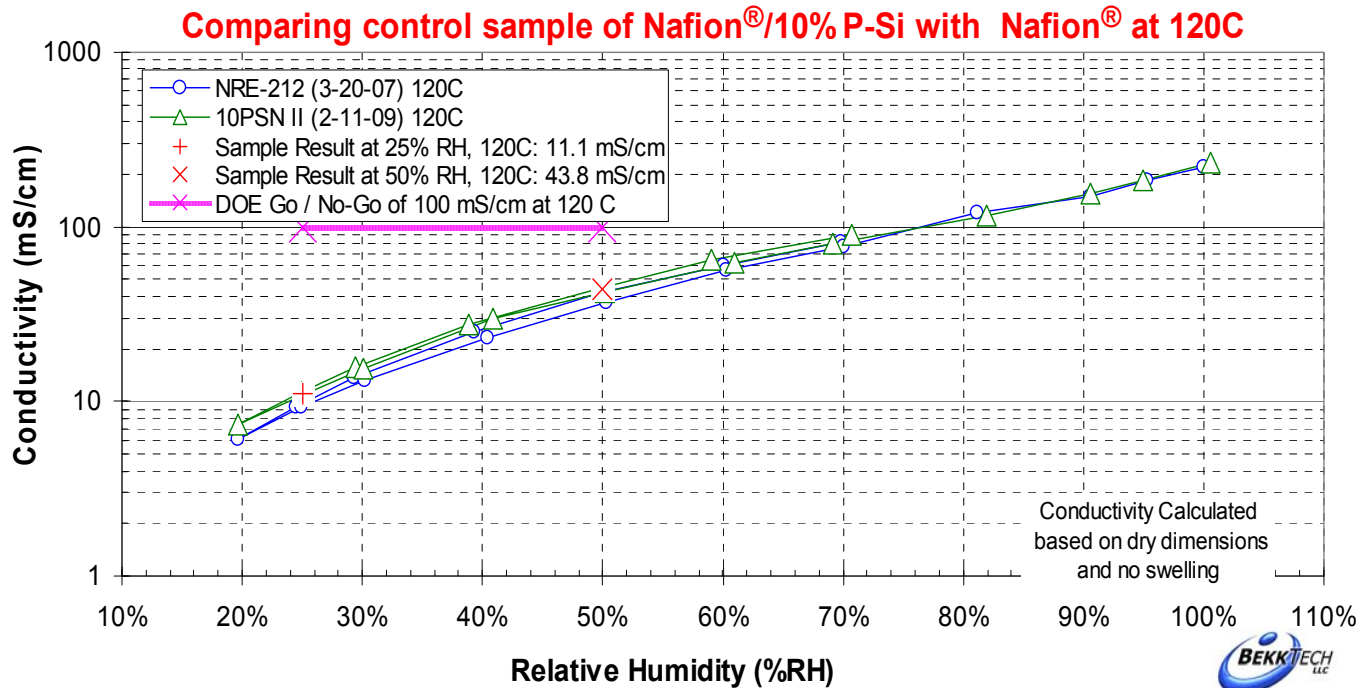
#### Different Inorganic Content



## Task 3

### Polymer/Phosphosilicate Gel Composite Membranes In situ formation technique *Conductivity*

- The best performance demonstrated the membrane prepared using in situ formation in Nafion® solution.
- P:Si ratio ( from 0.2 to 1.5) did not affect membrane performance.
- Conductivity of Nafion®/10% P-Si gel membrane was slightly higher than Nafion® at all conditions.

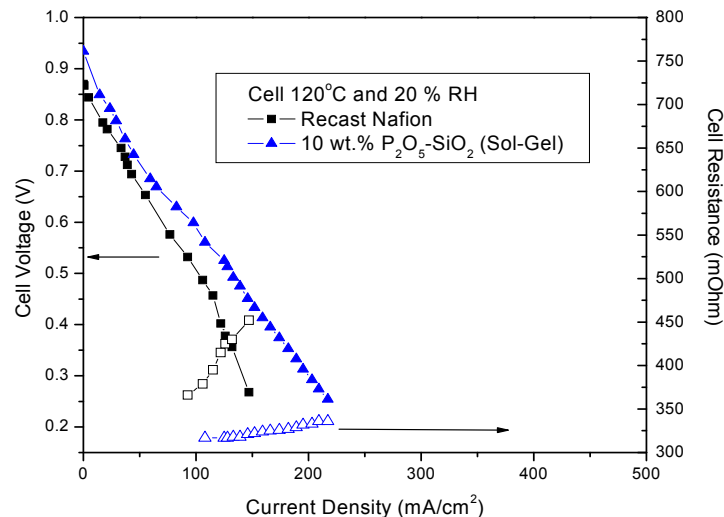
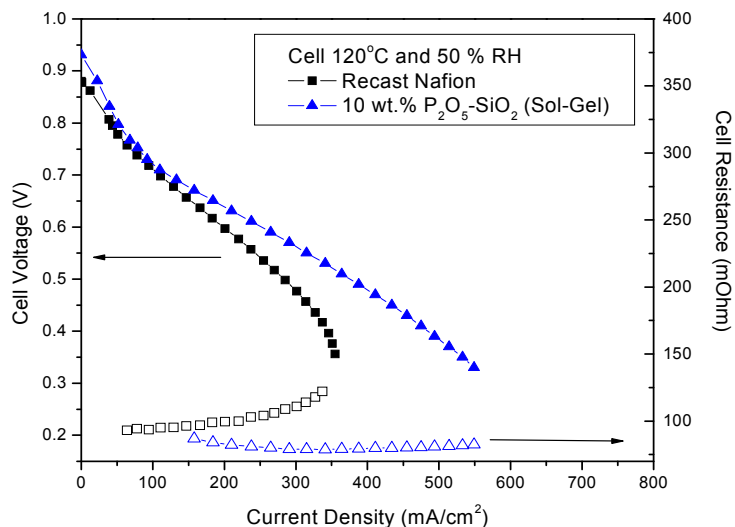


## Task 3

## Polymer/Phosphosilicate Gel Composite Membranes

### Fuel Cell Performance

At 120 °C and 50 and 20% RH, resistance of Nafion®/ 10% P-Si gel membrane in a fuel cell 40-50% less and limiting current 40-50% higher than that of Nafion® membrane.



Performance and FC Resistance of Nafion®/10%P-Si Gel Membranes in H<sub>2</sub>/Air Fuel Cell at 120 °C and Different RH

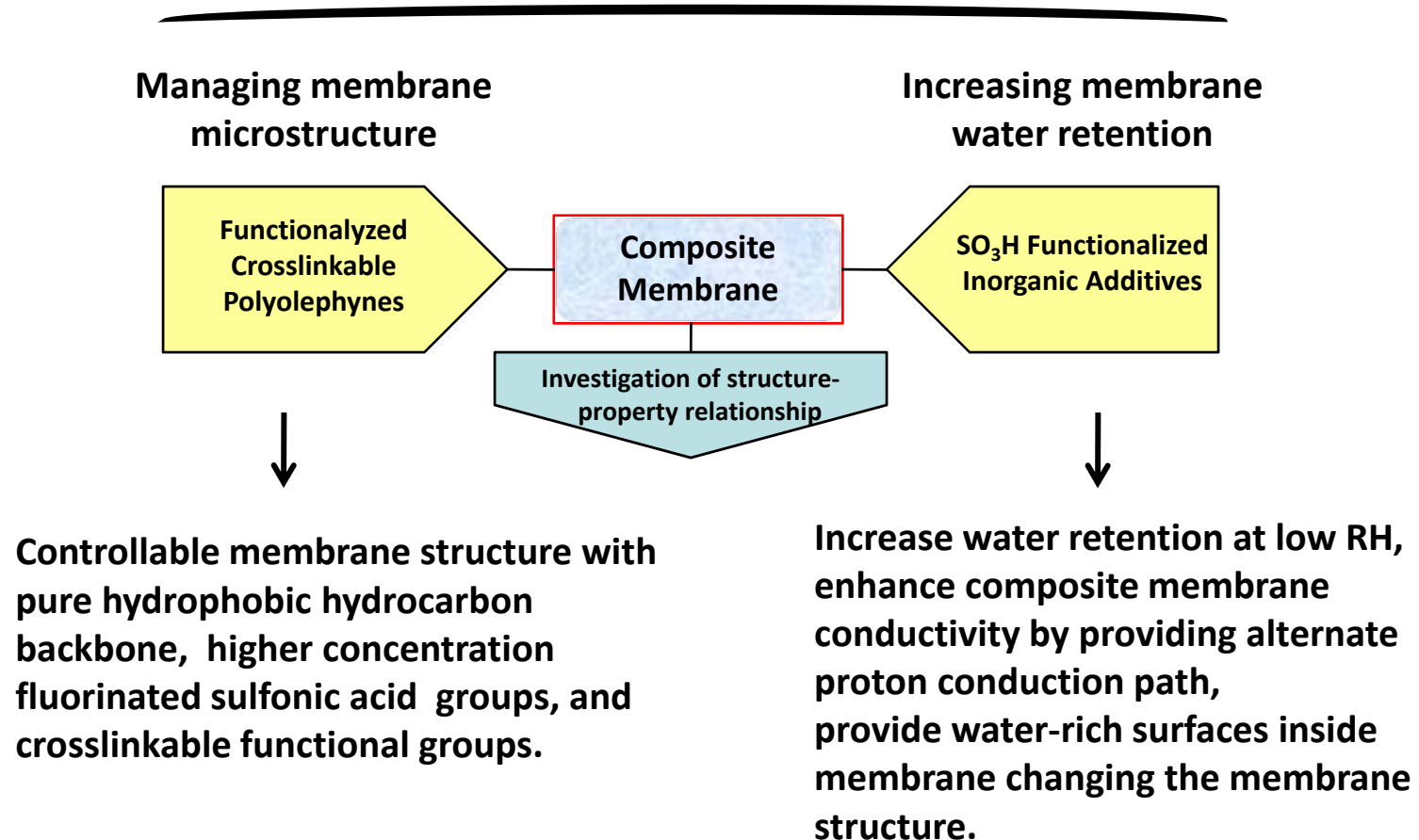
**Partner:** *BekkTech LLC*

- Fuel Cell Testing & Diagnostic Services
- Within of the DOE H<sub>2</sub> Program
- Control Measurements of New Samples Conductivity

# Possible Future Work

## Our approach to achieve DOE goals on membrane conductivity

- 100 mS/cm at 120°C and 50% RH
- 100 mS/cm at  $\leq 120^\circ\text{C}$  and inlet water vapor partial pressure  $< 1.5$  kPa





# Project Summary

- Relevance:** Development of high temperature membrane necessary for energy efficient PEM fuel cell implementation
- Approach:** Composite membrane based on functionalized polymeric material and hydrophilic proton conductive inorganic capable to hold water more tightly than the ionomer and to increase membrane conductivity at higher temperature and low RH.
- Technical Accomplishment and Progress:** Synthesis of new chemically and mechanically stable cross-linkable polyolefin based proton conductors; synthesis and characterization of highly conductive inorganics; selecting the technique for composite membrane fabrication; fabrication of Nafion®/phosphosilica gel composite membrane with enhanced conductivity and substantially improved fuel cell performance
- Collaborations:** *BekkTech LLC*, fuel cell testing and diagnostic service
- Possible Future Research:** Development of composite membrane based on new chemically and mechanically stable cross-linkable polyolefin proton conductors and highly conductive inorganic to increase membrane water retention and conductivity up to the level necessary to meet the DOE milestones.