



## Development of Robust Metal Membranes for Hydrogen Separation

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Reaction Chemistry & Engineering Research Group Leader  
Office of Research & Development, NETL

2009 DOE Hydrogen Program Review



# Reaction Chemistry & Engineering Group Members

## U.S. DOE - NETL

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## NETL Research Faculty

*Dr. Andrew Gellman, CMU*

*Dr. James Miller, CMU*

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## NETL Site Support Contractors

*Dr. Mike Ciocco, Parsons*

*Dr. Sonia Hammache, Parsons*

*Paul Zandhuis, Parsons*

*Nick Means, Parsons*

*Technical staff, Parsons*

# Overview

## Timeline

- Project start date: 10/1/2008
- Project end date: 9/30/2009
- Percent complete: 67%

## Budget

- FY09 Funding: \$746k
- FY08 Funding: \$1,000k
- FY07 Funding: \$1,230k

## Barriers<sup>(1)</sup>

- (G) H<sub>2</sub> Embrittlement
- (H) Thermal cycling
- (I) Poisoning of catalytic surface
- (J) Loss of structural integrity and performance

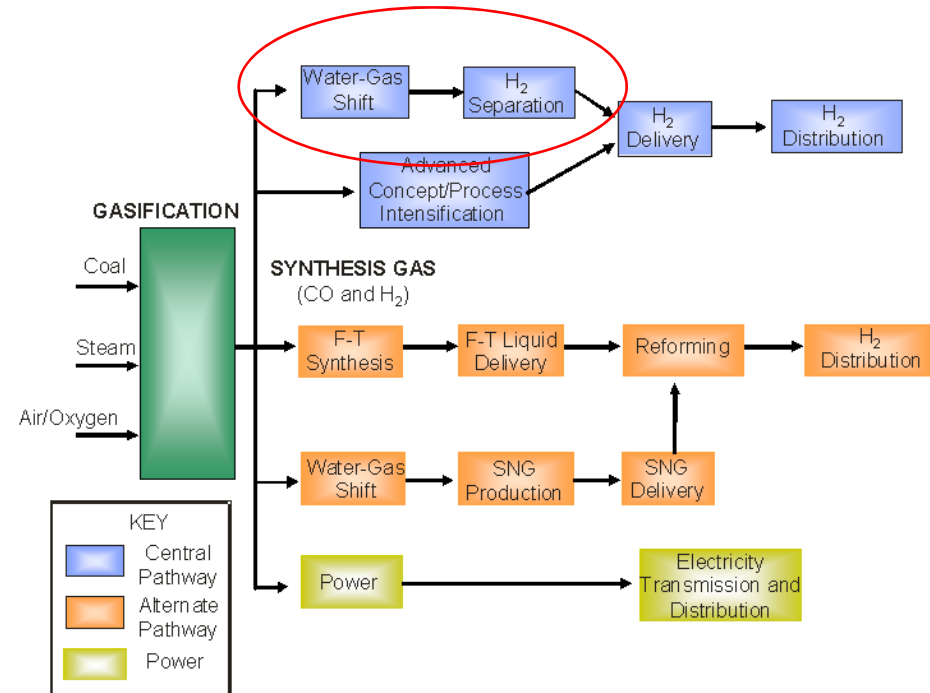
## Partners

- Carnegie Mellon University
- University of Pittsburgh
- Gas Technology Institute
- REB Research
- Los Alamos National Lab.
- NETL Computational Chemistry

# Background

*(Relevance)*

- **Overall goal**
  - Development of robust dense metal, hydrogen separation membranes for integration into coal conversion processes
- **Studies suggest that incorporating separation membranes into coal conversion processes can reduce costs by...**



# Facilities & Capabilities

## •Reactor Systems

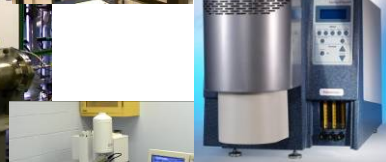
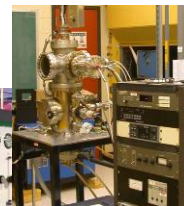
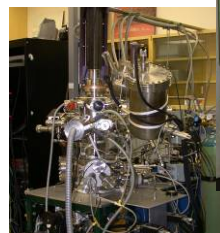
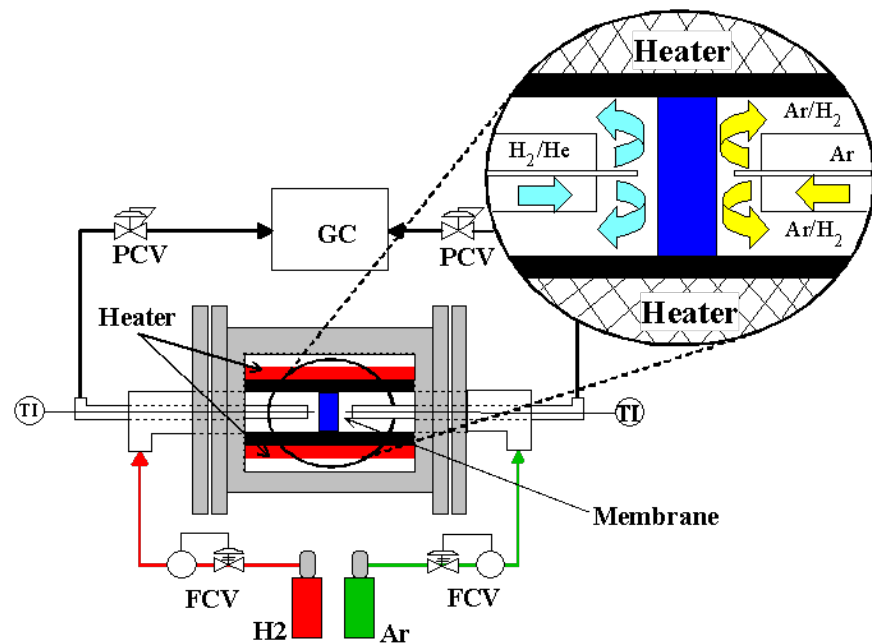
- Reactor and separation configurations
  - Continuous or batch
- Major and minor syngas constituents
- T to 1000°C, P to 1000 psi

## •Fabrication Lab

- Depositions chamber
- Vacuum arc-melter
- Micro-welder
- High-T box and annealing ovens

## •Characterization Instruments

- UHV chambers
  - Gradient doser, AES, XPS, LEIS, TPD, PVD
- XRD w/hot-stage
- SEM w/EDS
- TGA for use with H<sub>2</sub>S



# Outline

**Task 1: H<sub>2</sub> Membrane Test Protocol**

**Task 2: PdCu System**

**Task 3: Robust Metal Membrane Development**

- **Objective**
- **Approach**
- **Technical Accomplishments**
- **Collaborations**
- **Proposed Future Work**

# Task 1: H<sub>2</sub> Membrane Test Protocol

## •Objective

- Define a H<sub>2</sub>-membrane test protocol that
  - will advance the technology towards application to coal conversion processes
  - is consistent with overall FE program metrics, and
  - yields a basis for an “apples-to-apples” comparison

## •Approach

- Apply understanding of engineering principles, membrane technology and coal conversion processes to define a sequential protocol

Performance Criteria	Units	2007 Target	2010 Target	2015 Target
Flux <sup>(a)</sup>	sccm/cm <sup>2</sup>	51	102	152.4
Temperature	°C	400–700	300–600	250–500
S Tolerance	ppmv	----	20	>100
Cost	\$/ft <sup>2</sup>	150	100	<100
WGS Activity	-	Yes	Yes	Yes
? P Operating Capability <sup>(b)</sup>	psi	100	Up to 400	Up to 800 to 1,000
Carbon Monoxide Tolerance	-	Yes	Yes	Yes
Hydrogen Purity	%	95%	99.5%	99.99%
Stability/Durability	years	1	3	5

<sup>a</sup> For 100 psi ΔP (hydrogen partial pressure basis)

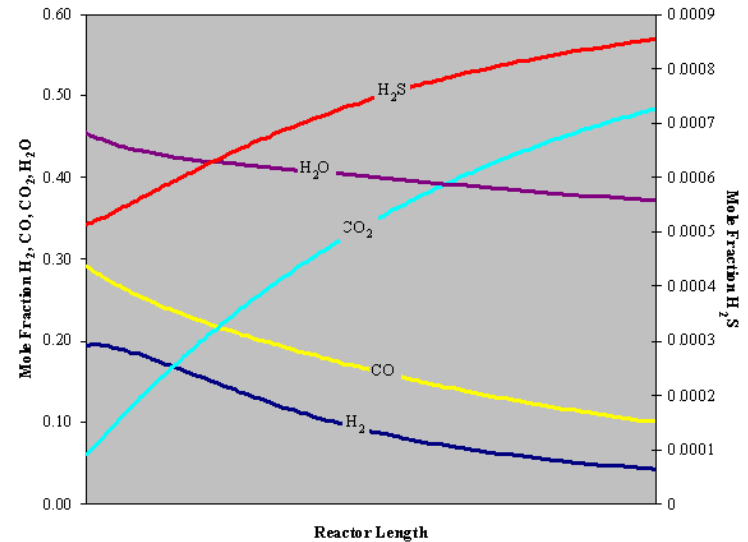
<sup>b</sup> ΔP = total pressure differential across the membrane reactor

# Task 1: H<sub>2</sub> Membrane Test Protocol

## (Technical Accomplishments)

- Completed a survey to determine the effluent composition of a WGS unit
- Developed COMSOL model to predict the influence of WGS reaction and/or H<sub>2</sub> removal on overall gas composition
- Identified the test conditions and gas compositions that are relevant to syngas conversion flowsheet options:

- Test 1: Shifted syngas, with no sulfur
- Test 2a: Shifted syngas with 20ppm H<sub>2</sub>S
- Test 2b: Shifted syngas with ~50% H<sub>2</sub> removal
- Test 2c: Shifted syngas with ~90% H<sub>2</sub> removal



	Test 1	Test 2a	Test 2b	Test 2c
H <sub>2</sub>	50%	50%	33%	5%
CO	1%	1%	1%	2%
CO <sub>2</sub>	30%	30%	40%	57%
H <sub>2</sub> O	19%	19%	25%	36%
H <sub>2</sub> S	0.0%	0.2%	0.3%	0.4%
Temp	300-600oC			
P <sub>Ret</sub>	200 psi			
P <sub>Per</sub>	atm			



# Task 1: H<sub>2</sub> Membrane Test Protocol

*(Collaborations)*

- **NETL Technology Manager and Technology Team**
  - The development of the test protocol was a team effort consisting of several participants of the Technology Team
- **NETL funded H<sub>2</sub> Separation Projects**
  - Provide unbiased performance verification testing
    - REB Research
    - ORNL
    - Eltron Research
    - WRI

# Task 1: H<sub>2</sub> Membrane Test Protocol

*(Proposed Future Work)*

- **Continue to support the development of test protocols to include more “commercially relevant” conditions**
  - Higher transmembrane pressure differentials
  - Contaminants other than H<sub>2</sub>S
    - For example, Cl and N for biomass co-feed
  - Integration of WGS reactor and Membrane separator
    - (WGSMR)

# Task 2: PdCu System

## •Objective

- Complete a comprehensive performance evaluation of the PdCu system at conditions consistent with coal conversion processes

## •The intent of the study is to

- gain a fundamental understanding of the PdCu system
- address discrepancies observed in literature
- develop property-performance relationships
- provide design guidance for fabrication of membranes at commercial scales and thicknesses

Performance Criteria	Units	2007 Target	2010 Target	2015 Target
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Stability/Durability	years	1	3	5

<sup>a</sup> For 100 psi  $\Delta P$  (hydrogen partial pressure basis)

<sup>b</sup>  $\Delta P$  = total pressure differential across the membrane reactor

# Task 2: PdCu System

## (Approach)

- Utilize several PdCu compositions

- 100, 80, 60, 53wt%Pd-Cu

- Fabricated by “cold rolling” techniques

- Membrane thickness was generally  $\sim 100\mu\text{m}$

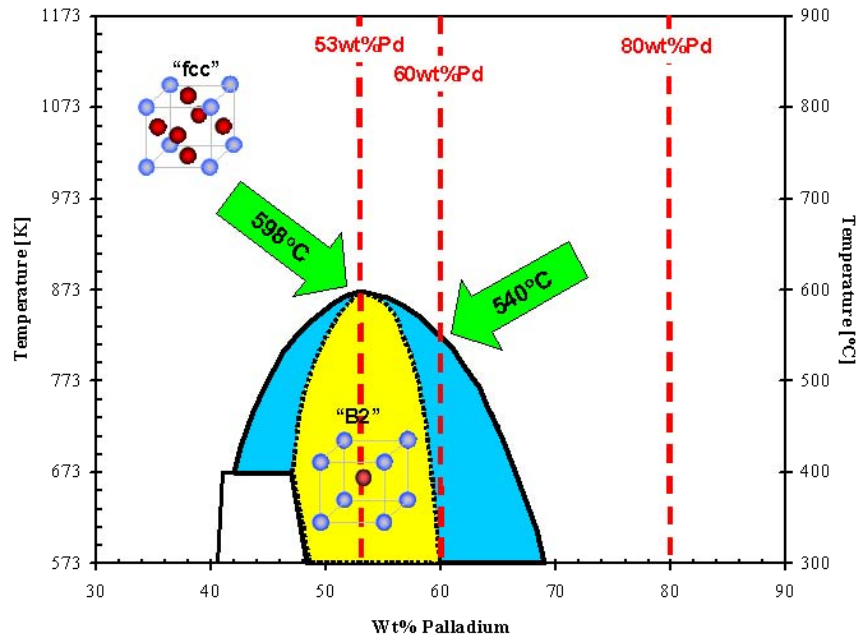
- Ease of operation and minimize failures

- Temperatures consistent with post gasification and allowed variation in crystalline structure

- 350, 450 and 635°C

- Sour gas studies

- 8hrs in clean 10%He-H<sub>2</sub> (baseline)
- 120hrs in 0.1%H<sub>2</sub>S-10%He-H<sub>2</sub>



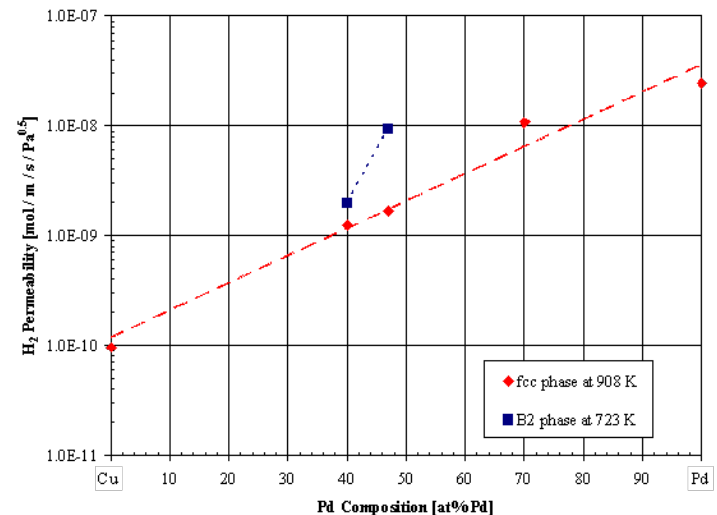
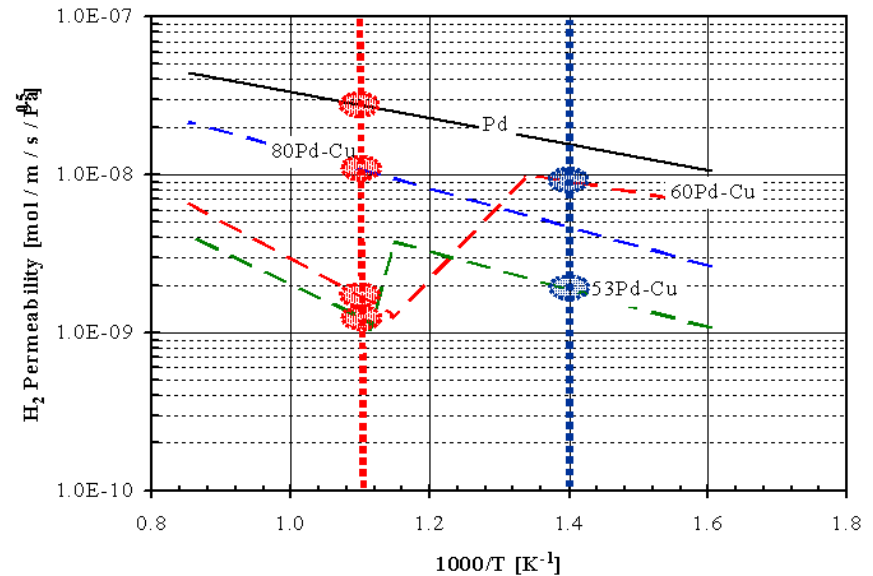
Subramanian, Laughlin, Binary Alloy Phase Diagrams, Second Ed., ASM International, (1990) 1454-1456.

# Task 2: PdCu System

## (Technical Accomplishments)

### •Completed evaluation of hydrogen permeability of Pd and PdCu

- 60Pd-Cu exhibits the highest permeability at temperatures below  $\sim 500^{\circ}\text{C}$ , corresponding to a B2 crystalline structure.
- 80Pd-Cu exhibits the highest permeability at temperatures above  $\sim 500^{\circ}\text{C}$ , corresponding to a fcc crystalline structure.
- In general, Pd-Cu permeability increases with increasing Pd content.

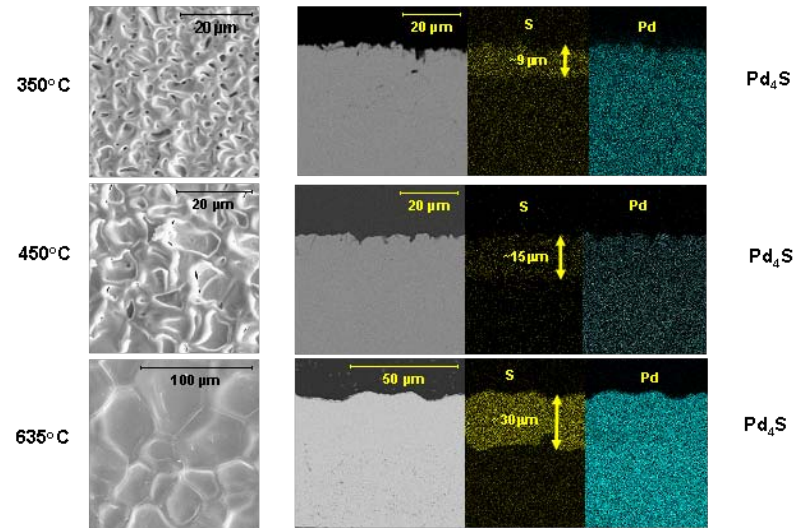


# Task 2: PdCu System

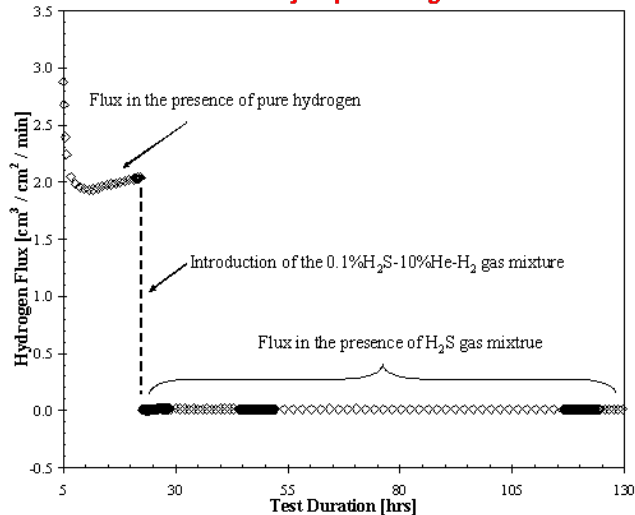
## (Technical Accomplishments)

### Completed evaluation of Pd and PdCu alloys in presence of H<sub>2</sub>S

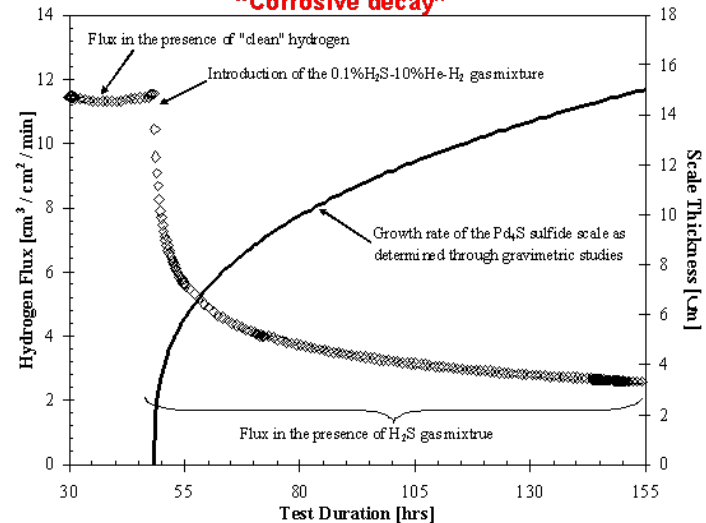
- Catalytic poisoning: Immediate decrease: no significant surface scale
  - 60Pd-Cu, 53Pd-Cu
- Corrosive decay: Gradual decrease: significant surface scale
  - Pd, 80Pd-Cu
- No change in performance upon the introduction of H<sub>2</sub>S: no surface scale.
  - 80Pd-Cu, 60Pd-Cu, 53Pd-Cu at T>450°C



“Catalytic poisoning”



“Corrosive decay”



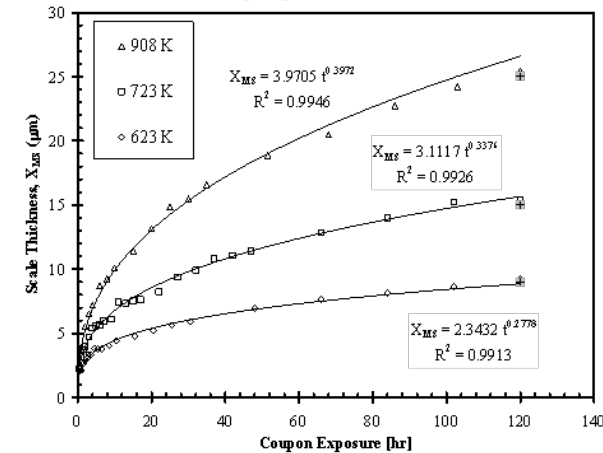
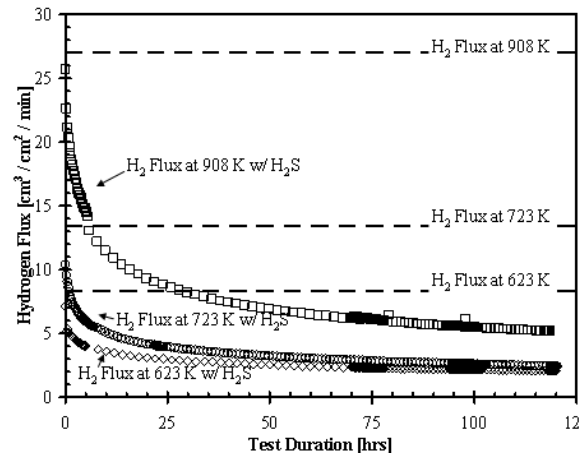
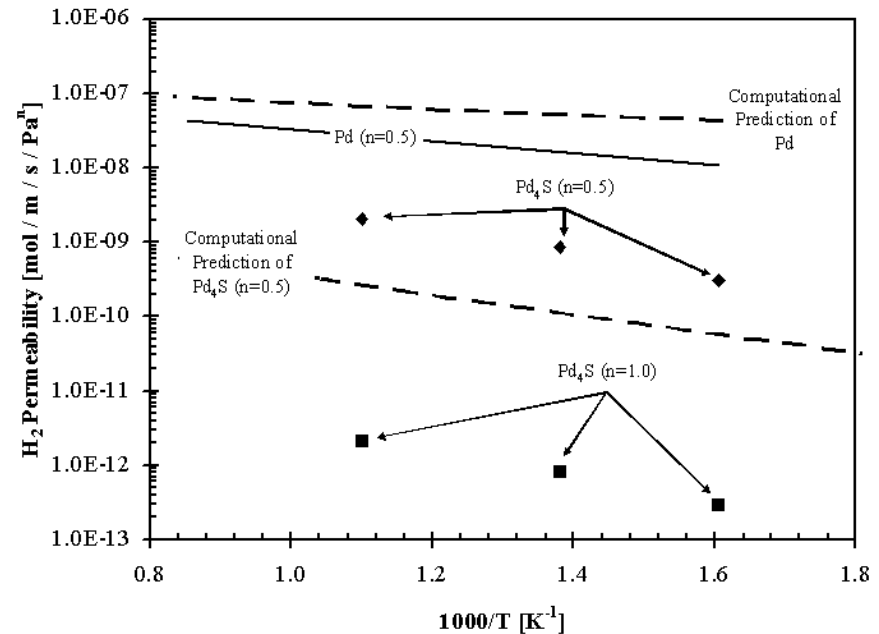
# Task 2: PdCu System

## (Technical Accomplishments)

- Reported the first ever permeability of Pd<sub>4</sub>S using both experimental and computational techniques

- Approach can be used for developing new membrane systems

- Pd<sub>4</sub>S permeability is ~1/10 Pd permeability



# Task 2: PdCu System

## *(Collaborations)*

- **The research team conducting the work on the PdCu system consisted of participants from local universities**
  - University of Pittsburgh
    - Provided technical support
    - Performance testing of the membranes
    - Membrane characterization
  - Georgia Institute of Technology
    - Utilized computational method to predict the first reported permeability of palladium-sulfide



# Task 2: PdCu System

*(Proposed Future Work)*

- **Characterization of the PdCu system in clean and H<sub>2</sub>S-contaminated environments has been successfully completed**
- **No additional work specific to PdCu is planned**
- **The results of our work with PdCu will help set the direction of future Robust Metal Membrane Development (Task 3):**
  - how to think about and characterize interaction of sulfur with multicomponent materials, like alloys
  - alloys' contribution to corrosion resistance
  - the role of minor components in imparting sulfur tolerance to metal membrane systems

# Task 3: Robust Metal Membrane Development

- Identify membrane compositions and configurations that meet the criteria outlined in FE H<sub>2</sub> from Coal RD&D plan per the NETL Membrane Test Protocol
- Provide design guidance to collaborators who will fabricate membranes at commercial scales and thicknesses

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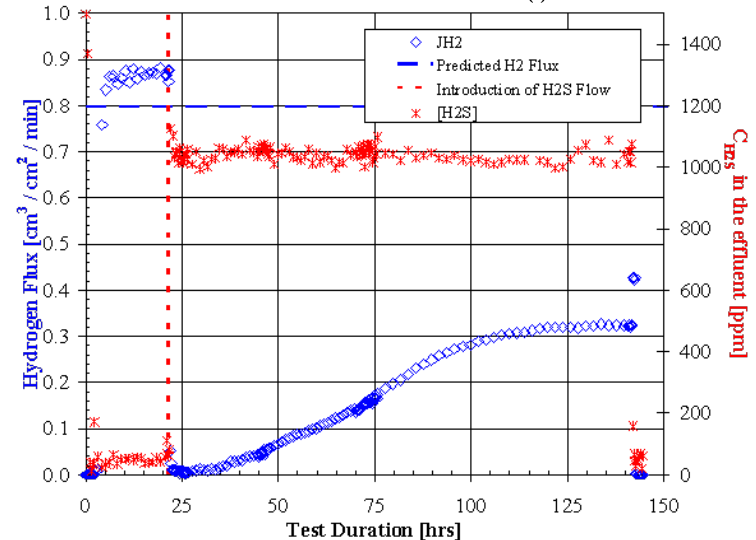
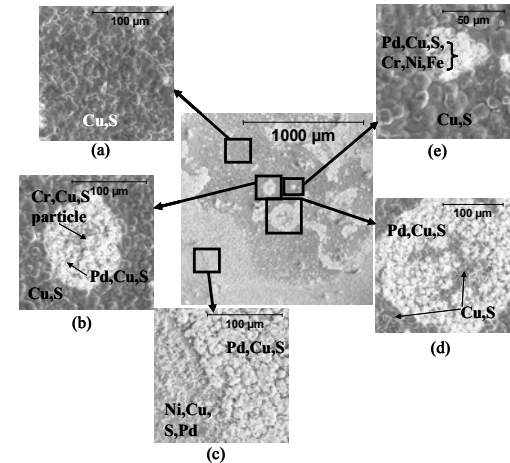
<sup>a</sup> For 100 psi  $\Delta P$  (hydrogen partial pressure basis)

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# Task 3: Robust Metal Membrane Development

## (Approach)

- Building on the PdCu foundation, apply fundamental and applied science to engineer membrane alloys and composites suitable for coal conversion processes
  - Corrosion resistance
    - fundamental thermodynamics,
    - gravimetric analysis
  - Surface activity
    - H<sub>2</sub>/D<sub>2</sub> exchange
    - computational studies
  - H<sub>2</sub>-transport



# Task 3: Robust Metal Membrane Development

## *(Technical Accomplishments)*

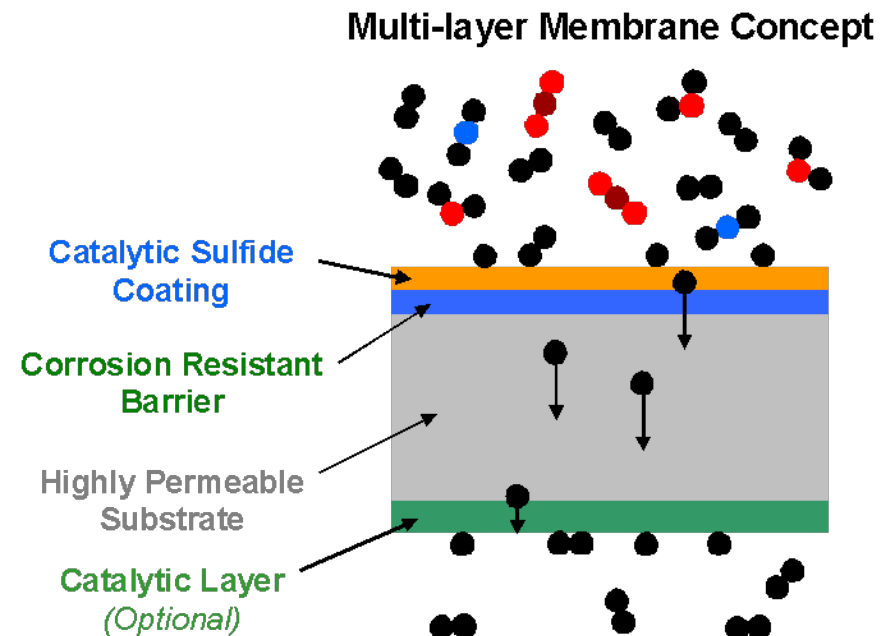
### •Provisional Patent Filed:

#### “Sulfur Induced H<sub>2</sub>-Membrane”

- Concept: Use B2-structured Pd-Cu alloy as sulfide corrosion barrier in multilayered membrane structure
- Utilize “S-based surface catalyst” to provide atomic hydrogen for transport

### •New capabilities developed

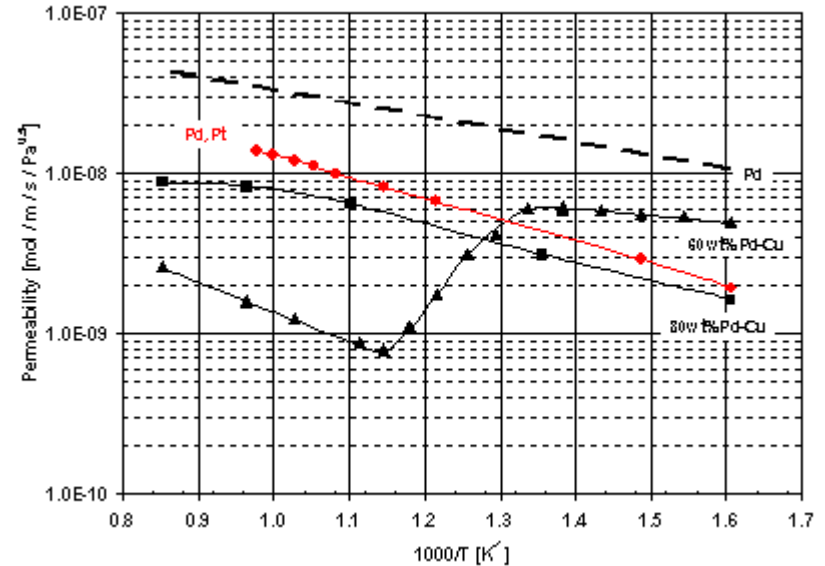
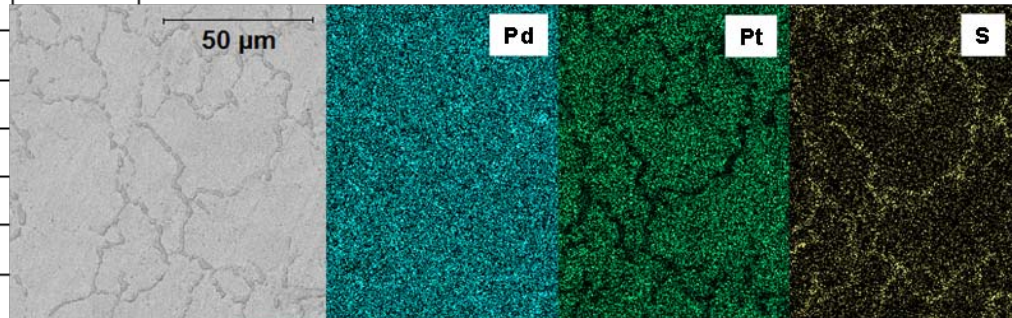
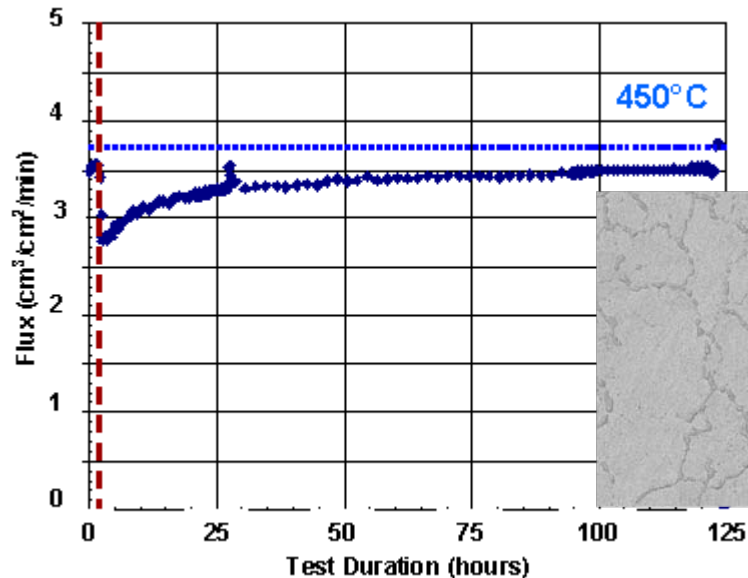
- preparation and characterization of H-atom transport through multi-layer structures in UHV
- preparation and characterization of catalytic-sulfide top-layers for multi-layer structures



# Task 3: Robust Metal Membrane Development

## (Technical Accomplishments)

- Several binary and tertiary metallic systems have been fabricated and screened
  - Pd-Ag, Au, Co, Cu, Ni, Pt
- PdPt alloy has shown significant promise for S-tolerance



# Task 3: Robust Metal Membrane Development

## *(Collaborations)*

- **The research team conducting the work on the task consisted of participants from institutions**
  - Carnegie Mellon University
    - Provided technical support
    - Assisted performance testing of the membranes
    - Utilize UHV techniques to evaluate the energetics associated with H<sub>2</sub> activation on metal and sulfide surfaces.
  - NETL Computational Research Group
    - Provide fundamental computational studies evaluating the energetics associated with H<sub>2</sub> activation on metal and sulfide surfaces.

# Task 3: Robust Metal Membrane Development

*(Proposed Future Work)*

- **Characterize H-atom transport across interfaces buried within multi-layer structures**
- **Develop options for sulfur-resistant top-layers and corrosion-resistant intermediate layers for layered structures**
- **Continue evaluation of binary and higher alloys for use alone or as functional layers in multi-component structures**
- **Provide design input to partners who fabricate membranes for practical implementation**

# Summary

- **A test protocol has been developed that allows technological progression and comparisons for application to coal conversion processes**
- **A comprehensive study of the PdCu system has been completed**
  - Conditions of complete S-tolerance have been identified.
  - Corrosion/catalytic phenomena has been identified and will be used for further membrane development
- **Several alloy compositions have been fabricated and screened for performance**
  - Some alloys have shown potential for S-tolerance