# **Optically Read MEMS Hydrogen Sensor**



### Barton Smith Scott Hunter

Oak Ridge National Laboratory

Annual Merit Review Washington, DC June 9, 2010

Project ID #:SCS018

This presentation does not contain any proprietary, confidential, or otherwise restricted information.



### **Overview**

#### Timeline

- Start: Late FY 2007
- Finish: Project continuation & direction determined annually by DOE

#### Budget

- Project funding profile
  - FY 2007: \$125k
  - FY 2008: \$300k
  - FY 2009: \$200k\*

\*Project moved to VT for 1 year

#### **Barriers**

- D. Liability Issues
  - E. Variation in Standard
    Practice of Safety
    Assessments for Components
    and Energy Systems
- Technical Targets on next slide

#### **Partners & Collaborators**

- University of Tennessee
- Advanced Catalyst Systems
- Agiltron, Inc.
- United Protective Technologies



### **Overview**

### Technical Targets

- H<sub>2</sub> concentration measurement range: 0.1%-10%
- Operating temperature: -30 to 80°C
- Response time: less than one second
- Accuracy: 5% of full scale
- Gas environment: ambient air, 10-98% relative humidity range
- Lifetime: 10 years
- Interference resistant (e.g., hydrocarbons)



# **Relevance - Objectives**

**Project goal:** Develop optics-based sensing technology that achieves DOE R&D targets for hydrogen safety sensors.

Month-Year	Milestones
May 2009	Milestone: Complete characterization of response time, recovery time, sensitivity and accuracy within the operating temperature range (100% complete)
Jul 2009	Milestone: Establish commercialization partnership (75% complete)
Sep 2009	Milestone: Demonstrate sensor performance and compliance with safety goals (75% complete)



# **Technical Highlights**

### FY 2009

- Functionalized Polymer-Film Microcantilever Arrays (MCA)
  - Work performed in collaboration with University of Tennessee, Department of Chemistry
  - Fabricated and tested MCA coated with composite films of select polymers
    - Cantilevers are responsive to H<sub>2</sub> but minimally responsive to chemical interferents
    - Reversible cantilever responses to all chemicals in air sample
    - MCA provided patterned response for each chemical; pattern recognition software enabled quantification of each chemical present in the air sample
    - Determined best performing phases with respect to ability to discriminate against CO<sub>2</sub>, water vapor, methane
  - Completed selectivity, sensitivity and regeneration analyses of the functionalized polymer-film cantilevers
  - We now have a prototype MCA sensing device that provides qualitative and quantitative information on H<sub>2</sub> present in air sampling



# **Technical Highlights**

# FY 2010

- Developed new highly porous nanoparticle palladium coated microcantilever sensor arrays
  - Developed a new wet chemical Pd film deposition technique on Ag/Cr coated cantilever sensors using a galvanic PdCl<sub>2</sub> exchange reaction
  - Large sensor surface area in comparison with film volume leads to very fast sensor response and recovery times
  - Completed selectivity, sensitivity, response and recovery time and regeneration analyses of the nanoporous Pd coated sensors excellent response characteristics
  - Excellent stability, repeatability and life measurements have been observed during more than a year of operation
  - First field portable instrument fabricated to demonstrate operational performance of the Pd based sensors
  - Sensor commercialization partners identified and agreements in progress



# **Technical Approach –**

# **Microcantilever Hydrogen Detection**



7 Managed by UT-Battelle for the Department of Energy National Laborator

# Testing Hydrogen Sensors – Microcantilever sensor lab bench system



Lab bench setup for the H<sub>2</sub> sensitivity and potential interferent measurements



### **Sensor Material Development -**Functionalized multi-cantilever arrays



We use commercially available microcantilever arrays which are differentially functionalized with materials that potentially interact with H<sub>2</sub> to produce a bending response

Schematic of cantilevers coated with different thin films to invoke an array of response to H<sub>2</sub> and potential interferents





### **Sensor Material Development -**Nanostructured Palladium Surface Coatings



0.5 µm

- Wet chemical galvanic PdCl<sub>2</sub> exchange reaction with thin silver/chromium sputtered film leads to very nanoporous textured Pd metal surfaces
- > Conjectured to lead to fast response and recovery times



Low threshold and wide dynamic range response



Representative sensitivity measurements using Pd/Ag coated microcantilever H<sub>2</sub> sensors.

- Detection threshold 0.01% H<sub>2</sub>
- Dynamic range > 3 orders of magnitude



Fast sensor response and recovery

Fast Response (< 3 sec) and recovery (< 10sec) in simulated air



Sensor response to a 4% H<sub>2</sub>/Ar gas mixture after the microcantilever had been passivated by flowing the various carrier gases through the flow cell



### Excellent specificity to common impurities and carrier gases

#### Nanostructured Pd coated cantilever shows negligible response to most potential interferents



Time 30 seconds/division



Differences in temporal response and signal magnitude allow discrimination between more significant interfering species



Sensor accuracy, repeatability and lifetime



High sample-to-sample repeatability of the sensor bending response in a simulated  $H_2/air$  environment

Single microcantilever response to a 4%  $H_2$  /Ar-O<sub>2</sub> mixture over an 8 month period



### First Field portable instrument



First field portable prototype sensor hardware incorporating the same features as the benchtop apparatus but on a much reduced scale. This unit is operated by a standard laptop computer. Imaging camera used for optical alignment



The unit is fabricated on a 4" optical breadboard



### Advanced field portable instrument



Coated arrays of cantilever sensors have been used in arrays and combined with pattern recognition algorithms to obtain quantitative estimates for H<sub>2</sub> and various potential interferents for enhanced selectivity



# **Results** – Cantilever array measurements and neural network give accurate estimates for $H_2$ and $CO_2$



# Experiment to demonstrate the ability of an artificial neural network (ANN) algorithm to predict component mixtures of multiple species

- > Two component mixture comprised of varying concentrations of  $H_2$  and  $CO_2$
- Experiments were performed using arrays of silicon microcantilevers coated successively with nanostructured metal coatings.
- The graph shows the ability of the ANN to accurately predict the relative concentration of the two components



# **Collaborations**

#### **Partners:**

#### University of Tennessee - Improved H<sub>2</sub> sensor response

- Faster response/recovery times now around 3-10 sec reduce to  $\sim$  1 sec
- Improved specificity reduced water vapor response
- Further studies of potential interferents CO, EtOH, CH<sub>4</sub>, other hydrocarbons
- Multi-component mixture studies
- IP and patent filings techniques, methods, coatings, applications
- Agiltron Commercialization partnership agreements finalized

### **Technology Transfer:**

- Agiltron collaboration
  - Tech transfer to commercialization partner for preproduction prototype development
  - Manage preproduction instrument fabrication at commercialization partner facilities

Agiltron's existing low cost microcantilever based infrared camera optical readout circuitry can easily be modified to interrogate an array of microcantilever chemical sensors



# **Future Work**

#### FY 2010

#### Improved H<sub>2</sub> sensor response

- Faster response/recovery times now around 3-10 sec reduce to  $\sim$  1 sec
- Improved specificity reduced water vapor response
- Further studies of potential interferents CO, EtOH, CH<sub>4</sub>, other hydrocarbons
- Multi-component mixture studies
- IP and patent filings techniques, methods, coatings, applications
- Commercialization partnership agreements finalized

### FY 2011

- Finish fabrication and preliminary testing of field portable instrument
- Implement commercialization plan with commercial partner
  - Tech transfer to commercialization partner for preproduction prototype development
  - Manage preproduction instrument fabrication at commercialization partner facilities
- Test pre-commercialization prototype testing at NREL Hydrogen Sensor Laboratory
  - Short-term repeatability test
  - H<sub>2</sub> mixtures in air
  - Temperature and Relative Humidity
  - Response and recovery kinetics

Linearity/dynamic range test Atmospheric pressure sensitivity Long term stability Interferent testing





# **Summary Table** – Measured Performance

# Present microcantilever based H<sub>2</sub> sensors meet all but the most stringent requirements for automotive applications

	Performance Requirement*	Measured Performance
Sensitivity Range	< 0.1% to > 4%	0.01 – 4%
Survivability Limit	100%	Linear response to 100% H <sub>2</sub>
Response Time	Automotive: < 3 sec Stationary: < 30 sec	~ 3-5 sec
Recovery Time	Automotive: < 3 sec Stationary: < 30 sec	~ 10-20 sec
Temperature Range	Automotive: -40°C to +125°C Stationary: -20°C to +50°C	Yes
Pressure Range	Automotive: 62-107 kPa Stationary: 80-110 kPa	Yes
Ambient Relative Humidity Range	Automotive: 0 – 95% Stationary: 20 – 80%	0-100%
Interferent Resistance	No false positive responses	Excellent
Power Consumption	< 1 Watt	0.5-1 Watt
Lifetime	Automotive: 6,000 hr Stationary: > 5 years	Demonstrated > 9 months
Accuracy and Repeatability	Automotive: 5-10% Stationary: 10%	> +/- 5%

\* L. Boon-Brett et al., "Identifying performance gaps in hydrogen safety sensor technology for automotive and stationary

applications," Int. J. Hydrogen Energy, 35, 373-384 (2010).



# **Project Summary**

#### **Relevance:**

Commercialize optical technology that achieves DOE R&D targets for hydrogen safety sensors **Approach:** 

Develop and characterize an optically based, cost effective hydrogen sensor that can be used in distributed hydrogen leak sensing applications

#### **Technical Accomplishments and Progress:**

Demonstrated that optically read microcantilever based sensors meet all the H<sub>2</sub> sensor performance and life operating requirements

#### **Technology Transfer/Collaboration:**

Active partnership with the University of Tennessee. New partnerships with Agiltron, Advanced Catalyst Systems and UPT – publications, patents, preproduction prototype development.

#### **Proposed Future Research and Development:**

Improved H<sub>2</sub> sensor response and interferent selectivity. Portable preproduction instrumentation development for competitive benchmarking at NREL.

