Composite Materials for Hazard Mitigation of Reactive Metal Hydrides Project ID# ST013

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Project Overview

Timeline

- Project start date: June 2007
- Project end date: Sept.
 2011
- Percent complete: 90%

• Budget

- Total project funding
- DOE share: \$2.6M
- Funding received in FY10: \$310
- Funding for FY11: \$475

• Barriers & targets addressed

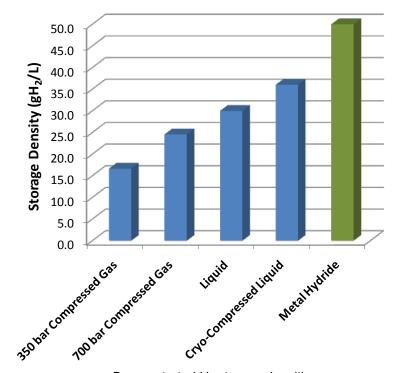
- On-Board Hydrogen
 Storage
- Durability/Operability (D)
- Codes and Standards (F)
- Reproducibility of Performance (Q)
- Partners
 - Collaboration with United Technologies Research Center



Relevance

• Metal hydrides show promise for compact hydrogen storage, enabling hydrogen usage in diverse applications.





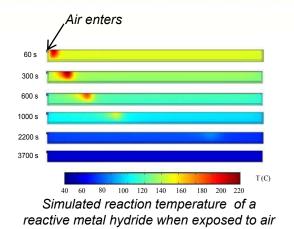
Demonstrated H₂ storage densities



Relevance (continued)

• However some show unfavorable reactivity when exposed to air or water (i.e., in an accident).



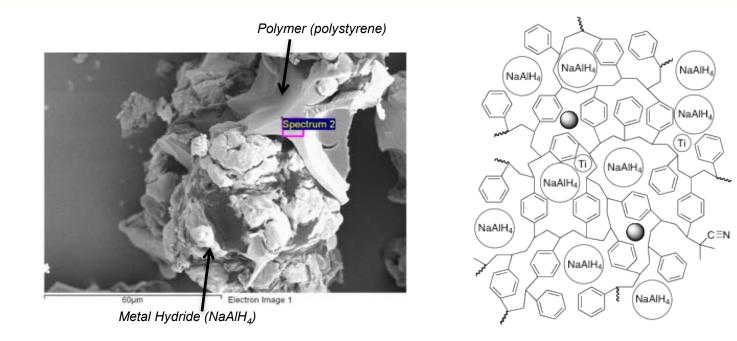


- A hazard mitigation strategy would help enable widespread use and commercialization of metal hydrides. Upon a breach in tank, something that:
 - Slows the reaction rate
 - Stops the penetration of oxygen, and/or
 - Absorbs the heat of reaction.



Approach

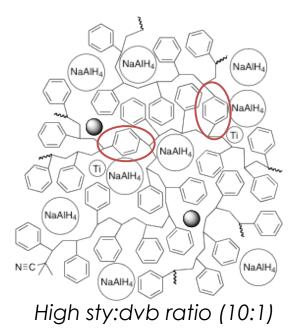
• A composite mixture of the metal hydride with a polymer may have these mitigating features.



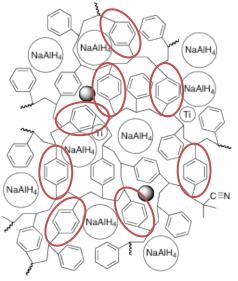
This methodology is applicable to other metal hydrides with similar reactivity issues.



 A metal hydride composite with copolymer polystyrenedivinylbenzene (dvb) was successfully synthesized, including differing cross-linking ratios.



Lower Cross-linking Density Flexible matrix, tough material → holds together



Low sty:dvb ratio (3:4)

Higher Cross-linking Density Stiff matrix, brittle material → easily crushed

Red circles indicate styrene cross link between polystyrene chains

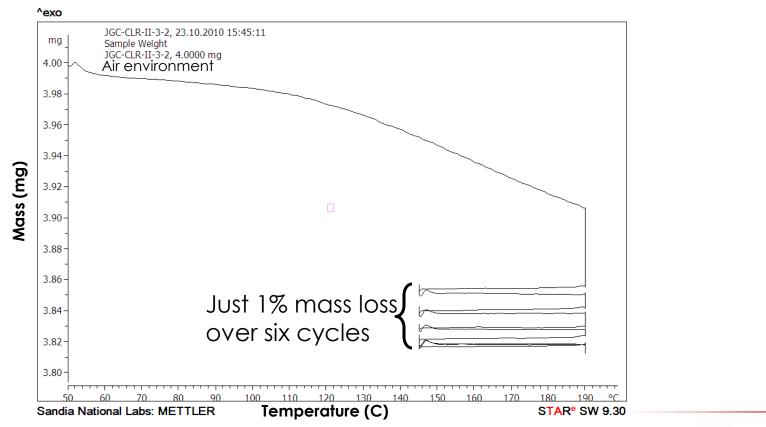


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- Large-scale synthesis was attempted at the 30 g, 50 g, and 300 g scales (compared to 2 g typical for other tests):
 - NaAlH4 with 20 wt% copolymer polystyrene-divinylbenzene (psdvb) composite.
 - Initial attempts in a steel-reaction vessel gave heterogeneous mixtures.
 - Further attempts to make composite material with mechanical agitation of the mixture were unsuccessful due to incomplete polymerization.
 - Further attempts at large scale synthesis of ps-dvb composite deferred due to subsequent findings showing it is not a suitable material.
 - If future large scale syntheses with a successfully performing composite material (in development) are attempted, a more suitable reaction vessel must be acquired.

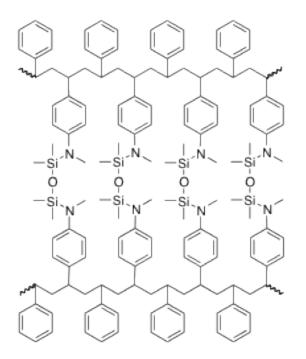


3. The polystyrene backbone was verified to be <u>thermally stable</u> (in air at the same thermal conditions as the H_2 charge/ discharge cycling: 30 min. at 145 C and 60 min. at 190 C).

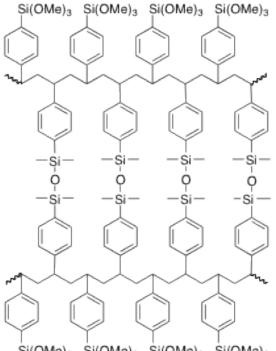




4. Synthesis routes for other polymer formulations (polystyrene with siloxane crosslinker, with and without siloxane moieties) have been devised.



Polystyrene with siloxane crosslinker

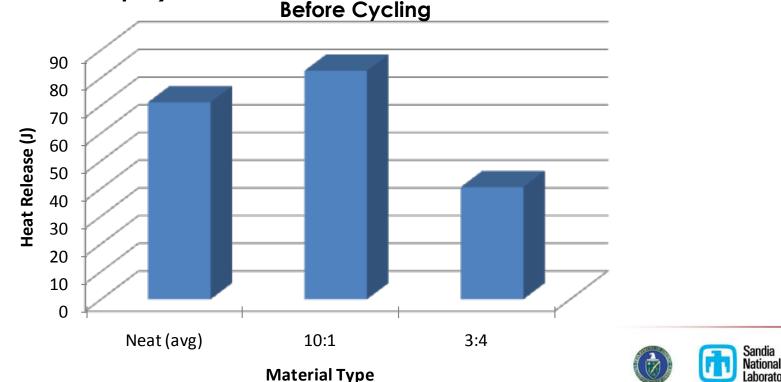


Si(OMe)₃ Si(OMe)₃ Si(OMe)₃ Si(OMe)₃

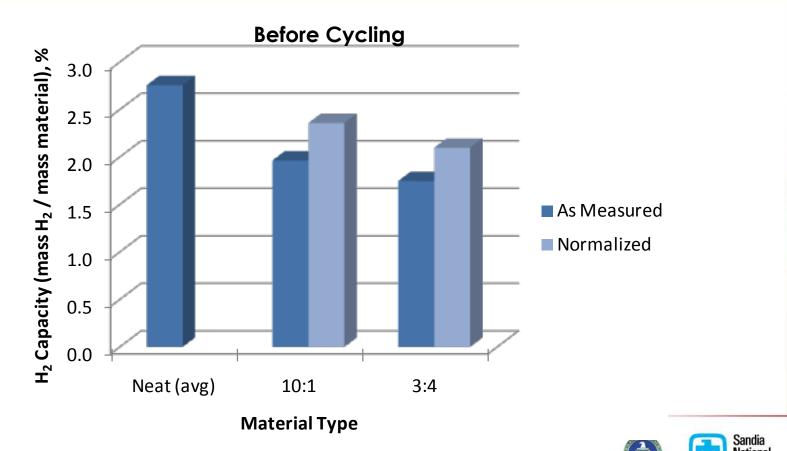
Polystyrene with siloxane crosslinker and siloxane moieties



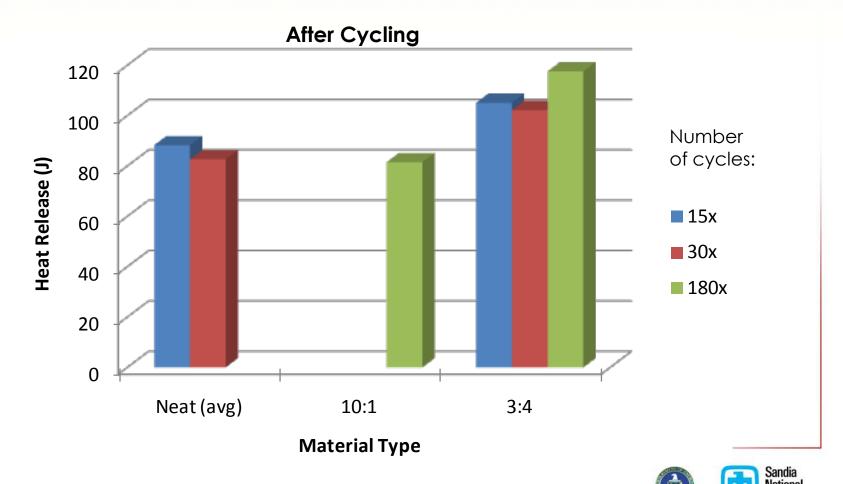
- 1. Reactivity was measured by the heat released during oxidation flow-through calorimetry. Before cycling:
 - 10:1 (ps:dvb ratio) did not reduce the reactivity (measured increase in heat release attributed to experimental error).
 - 3:4 (ps:dvb) was effective (approx. 60% of heat released by the neat sample)



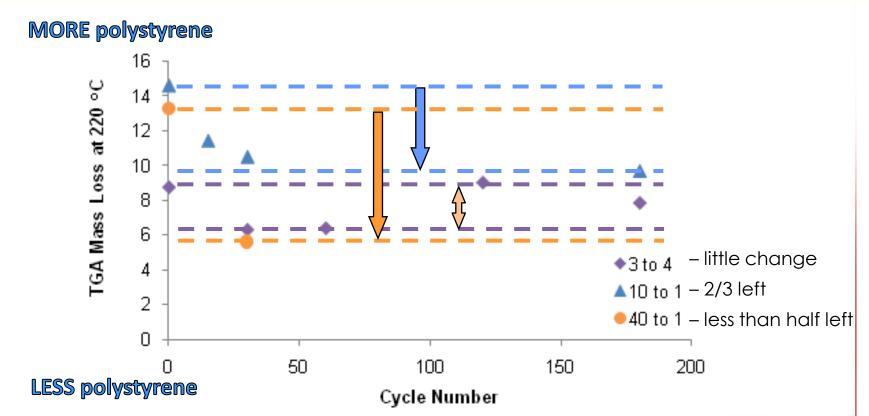
- 2. The initial hydrogen capacity (before cycling) is decreased by the addition of polystyrene.
 - Decrease in capacity directly related to amount of crosslinking.



- 3. After cycling, no mitigating effect of any polymer is observed.
 - Polymer is not stable under actual cycling conditions.

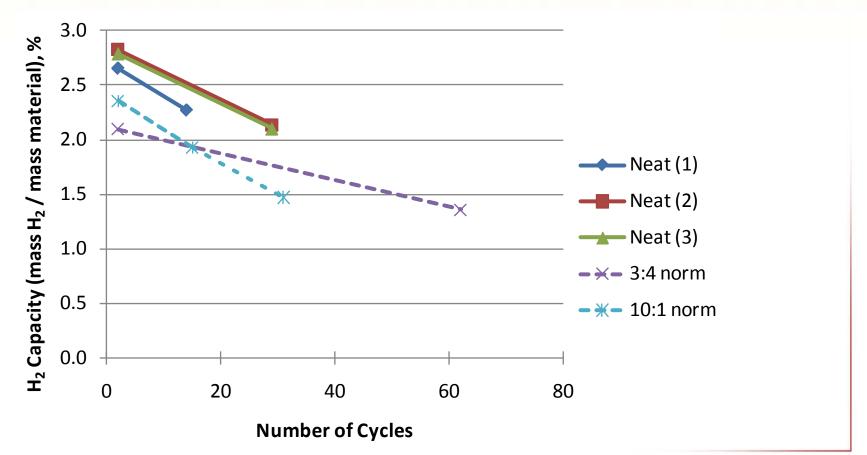


4. Mass analysis shows polystrene was lost during actual cycling, although the more crosslinked material experienced a smaller loss.



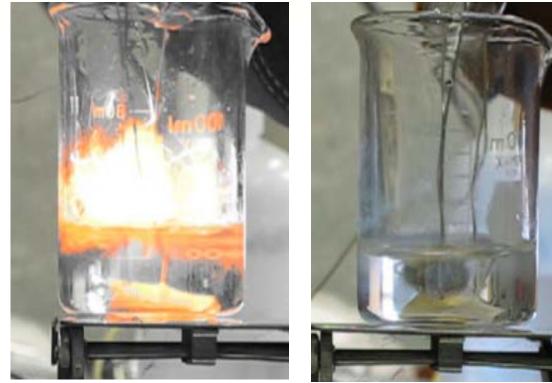


5. Less cross-linked (10:1) polystyrene has less impact on capacity, but is less stable with cycling.





6. Qualitative reactivity demonstration: pellet of material dropped into water at room temperature.



Neat material (no polystyrene)

Composite material (with polystyrene)



Collaborative Efforts

- Collaboration continues with United Technologies Research Center (Y. F. Khalil, PI). One example of tests done at UTRC:
 - While UTRC's wafer (unmitigated) ignited with flame on a hot surface at about 81 C, the Sandia (mitigated) sample did not ignite even at temperature of about 172 C.
 - → Indicates the added polymer (polystyrene) was beneficial in preventing the ignition of the heated sample.



Findings and Conclusions

- 1. Polystyrene backbone was found to be thermally stable in the same temperature conditions as cycling.
- 2. However, it broke down under actual cycling conditions (exhibited by loss of mitigating properties and verified by mass analysis).
- 3. The degree of breakdown is inversely related to the amount of crosslinking.
- 4. The degree of mitigation is directly related to the amount of crosslinking.
- The crosslinking properties seem to be the key to finding a better mitigating and more robust composite.



Future Work

- The siloxane formulations are expected to have beter mitigating properties and show more durability under charge/discharge cycling. The plan forward is to:
 - Demonstrate a reliable method to synthesize the polystyrenesiloxane polymer composites.
 - Evaluate their effectiveness in reducing heat release.
 - Evaluate their stability under cycling conditions.
 - Evaluate performance of systems during fire impingement.



Summary

- Highlights of the two project tasks over the last year:
 - Materials Development
 - Composite mixtures of metal hydride and polymers with different cross-linking ratios have been successfully synthesized.
 - Lessons learned from large scale synthesis attempt will be applied to future attempts.
 - Polystyrene backbone verified to be thermally stable.
 - Synthesis routes for more promising materials have been identified.
 - Characterization
 - Crosslinking ratios affect hydrogen capacity and mitigating ability.
 - Polystyrene composite initially mitigated heat release, but was not robust enough to withstand charge/discharge cycling.
- Collaboration is proving useful in characterization efforts
- Current and future work focuses on developing composite materials with better mitigating and stability properties.



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