

Photoelectrochemical Material Synthesis at LANL

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Project ID #
PD097

Overview

Timeline

- Project start date: 10/01/2010
- Project end date: 9/30/2013*
- Percent complete: **75%**

Budget

- Total project funding
 - DOE share: \$ 365K
 - Contractor share: \$ 0K
- Funding received in FY12: \$ 100K
- Funding for FY13: \$ 65K

Barriers

- Barriers addressed
 - AE. Materials Efficiency – Bulk and interface
 - AF. Materials Durability – Bulk and interface
 - AG. Integrated Device Configurations

Partners

- Interactions/ collaborations
 - NREL
- Project lead
 - Todd Williamson, LANL

Relevance

- The objective of this work is to explore the use of Group III Nitrides, specifically InGaN, for Photoelectrochemical Hydrogen production.
- The focus of this year's work has been on creating p-type and metal-polar InGaN which should offer increased stability and higher conversion efficiency than n-type, N-polar InGaN that has been produced previously

Table 3.1.8 Technical Targets: Photoelectrochemical Hydrogen Production: Photoelectrode System with Solar Concentration ^a

Characteristics	Units	2011 Status	2015 Target	2020 Target	Ultimate Target
Photoelectrochemical Hydrogen Cost ^b	\$/kg	NA	17.30	5.70	2.10
Capital cost of Concentrator & PEC Receiver (non-Installed, no electrode) ^c	\$/m ²	NA	200	124	63
Annual Electrode Cost per TPD H ₂ ^d	\$/yr-TPDH ₂	NA	2.0M	255K	14K
Solar to Hydrogen (STH) Energy Conversion Ratio ^{e, f}	%	4 to 12%	15	20	25
1-Sun Hydrogen Production Rate ^g	kg/s per m ²	3.3E-7	1.2E-6	1.6E-6	2.0E-6

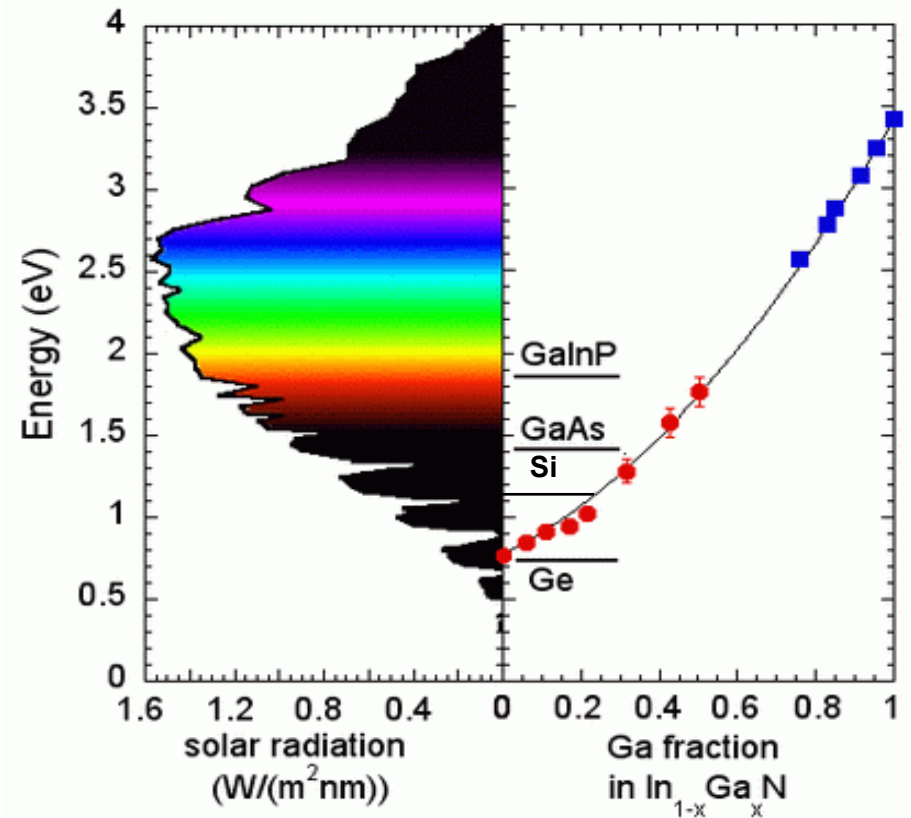
Approach

Our approach is to take utilize the InGaN ternary alloy system, a materials system whose bandgaps can span nearly the entire solar spectrum.

Pure nitride $\text{In}_x\text{Ga}_{1-x}\text{N}$ has potential to be both stable and efficient PEC electrode

InGaN materials are grown at Los Alamos National Lab (LANL) and their PEC H_2 production performance is characterized at the National Renewable Energy Lab (NREL)

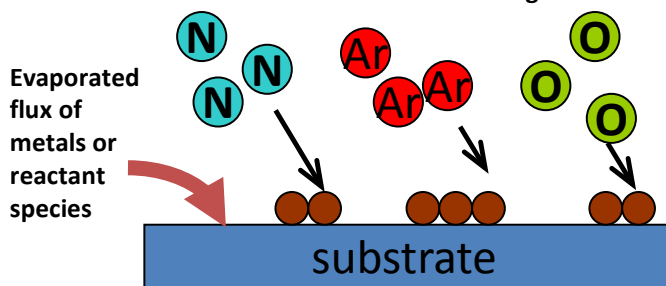
Walukiewicz et al, J. Appl. Phys 94, 6477(2003)



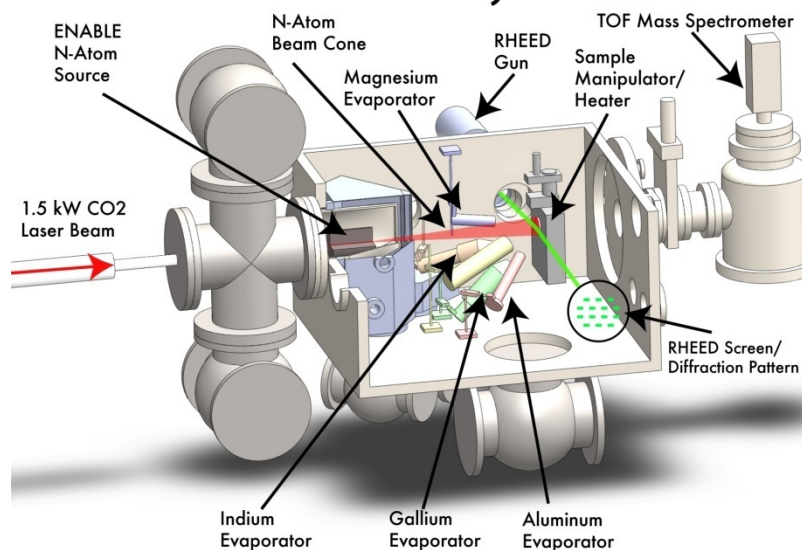
Approach – Film Synthesis

Energetic neutral atom beam assisted surface chemistry

1 to 5 eV atomic beam energies



The ENABLE System



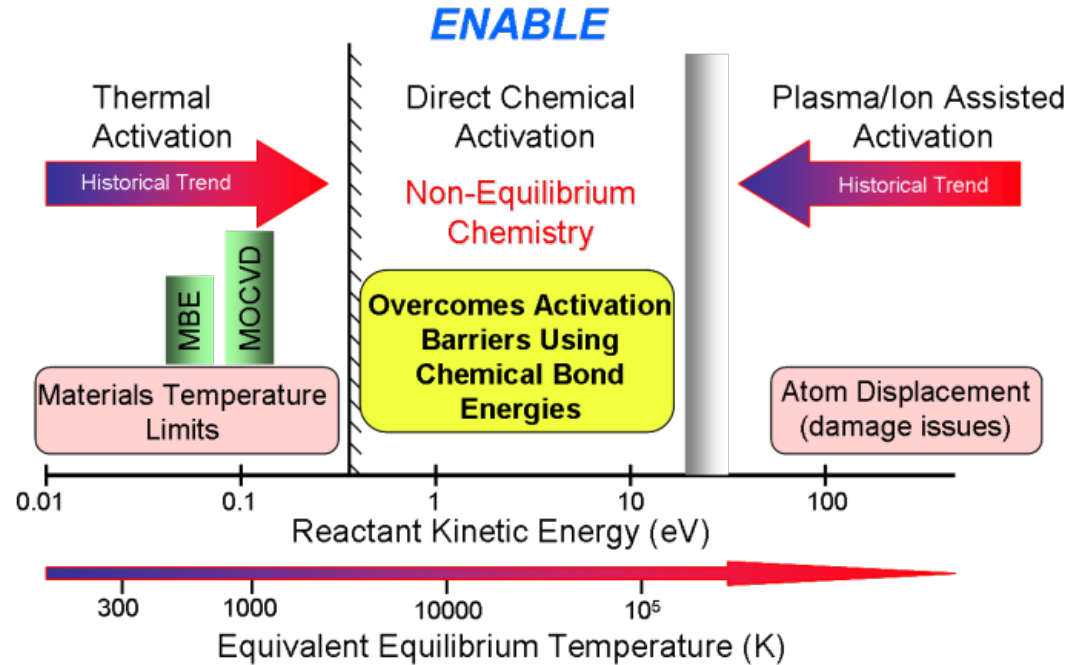
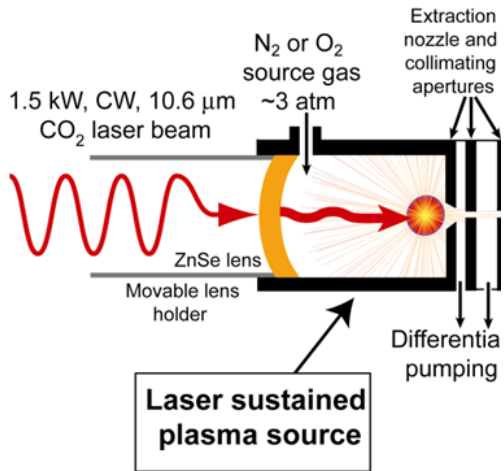
What is ENABLE:

ENABLE is a unique MBE-type film growth technology that uses a *high-flux* beam of *energetic* neutral nitrogen (or oxygen) atoms with kinetic energies of 1 to 5 eV for overcoming reaction barriers. Conventional thermal sources are used for metals.

Advantages:

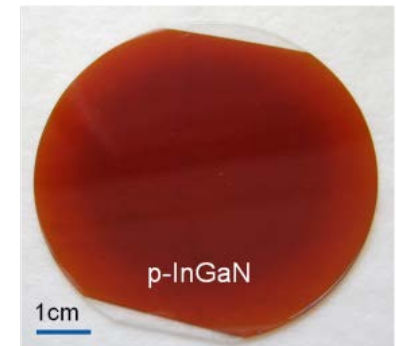
- With ENABLE, the *kinetic energy* of *neutral atoms* provides the reactant energy required to grow thin films via *direct chemical activation* (not high T or charge)
- The atom's kinetic energy also *promotes surface diffusion* and byproduct removal
- ENABLE provides a *high flux* of reactive species facilitating *high film growth rates* (>5 microns/hr) and *non-equilibrium* growth conditions allowing thermodynamic limits to be overcome
- No toxic chemical precursors or hazardous waste stream

Approach – Film Synthesis



ENABLE advantages for Group-III Nitride films:

- InGaN films can be grown over the full alloy composition range with good epitaxy & crystallinity
- Excellent compositional uniformity with no phase segregation
- Permits compositional grading for complex device architectures



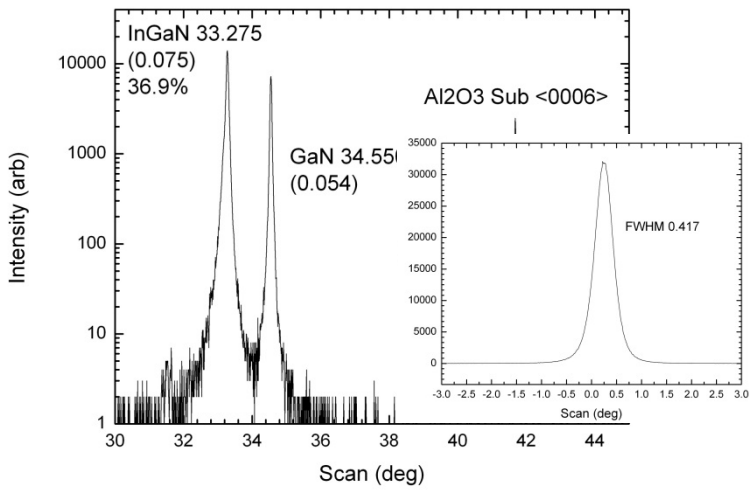
“red” In-rich InGaN film with 2.0 eV band gap grown by ENABLE

Approach – Milestones

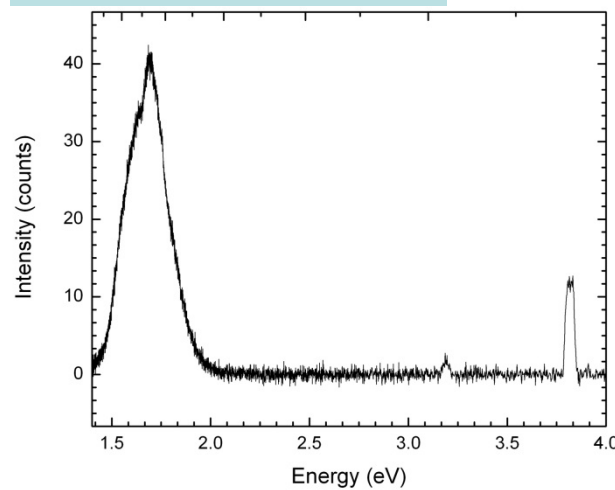
Milestone	Completion Date	Status
Demonstrate growth of single phase n-type InGaN with In compositions 15% - 30% on double-side polished sapphire and p-type Si with carrier concentrations $\sim 10^{18}/\text{cm}^3$ or less and carrier mobilities $\sim 100 \text{ cm}^2/\text{V}\cdot\text{s}$ or greater. Films will be sent to NREL for PEC performance characterizations.	January, 2013	50%
Demonstrate growth of single phase p-type InGaN with In compositions 15% - 30% on sapphire and p-type Si with carrier concentrations between $5 \times 10^{17}/\text{cm}^3 - 1 \times 10^{19}/\text{cm}^3$ and mobilities greater than $10 \text{ cm}^2/\text{V}\cdot\text{s}$. Films will be sent to NREL for PEC performance characterizations.	May, 2013	25%
With NREL, complete evaluation of InGaN material for direct water splitting or as part of a mechanically stacked tandem system and based on achievable efficiency and stability make go/no-go decision on additional studies.	September, 2013	10%

Accomplishments and Progress - Characterization results for InGaN on sapphire ~37% In and ~39% In contents

XRD measurements



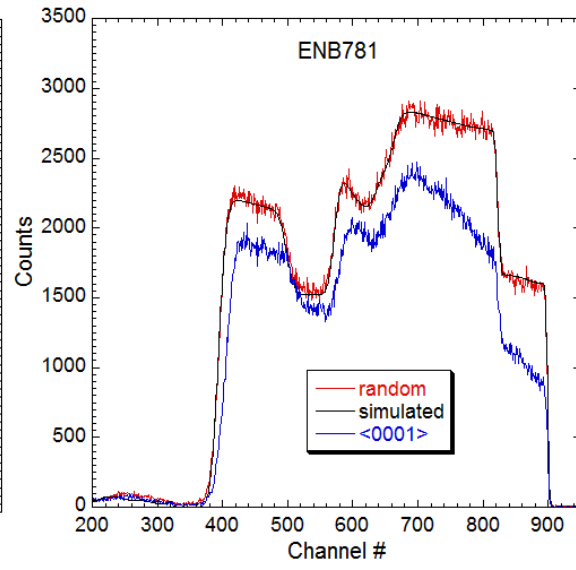
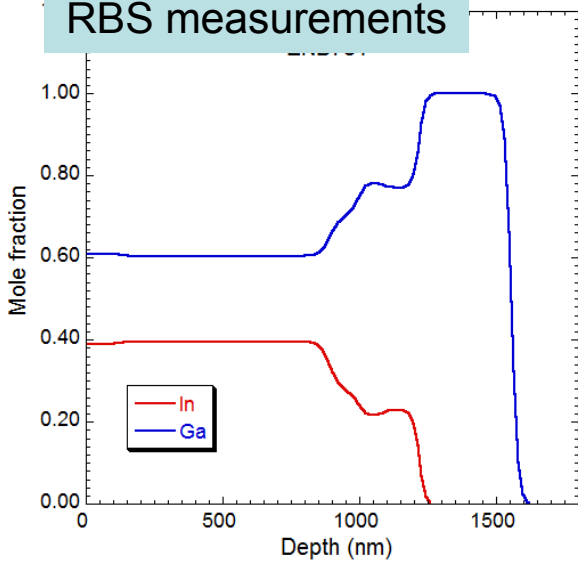
Photoluminescence



37% In InGaN has bandgap luminescence ~ 1.8 eV, suitable for PEC electrode

XRD shows good crystal structure

RBS measurements

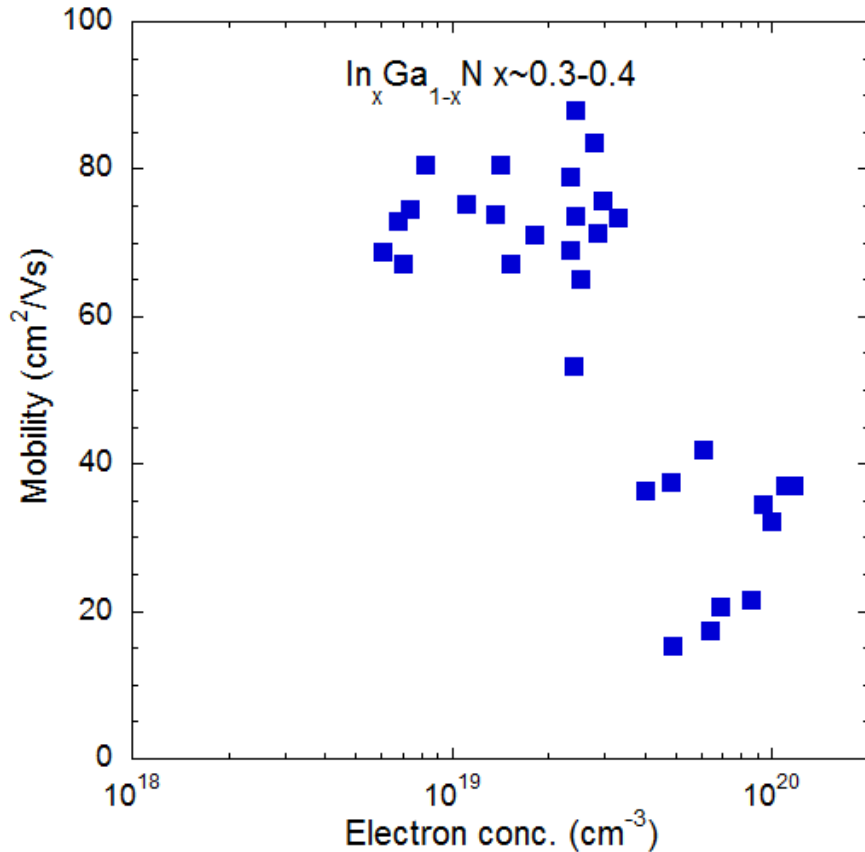


39% In content graded from ~20% to 39%.

Hall: $n=1.32 \times 10^{19}/\text{cm}^3$
 $\mu=55.6 \text{ cm}^2/\text{Vs}$

Grading of film from low In content to high produces higher quality films

Accomplishments and Progress- Mobility of InGaN alloy films with In content in the 30 to 40% range



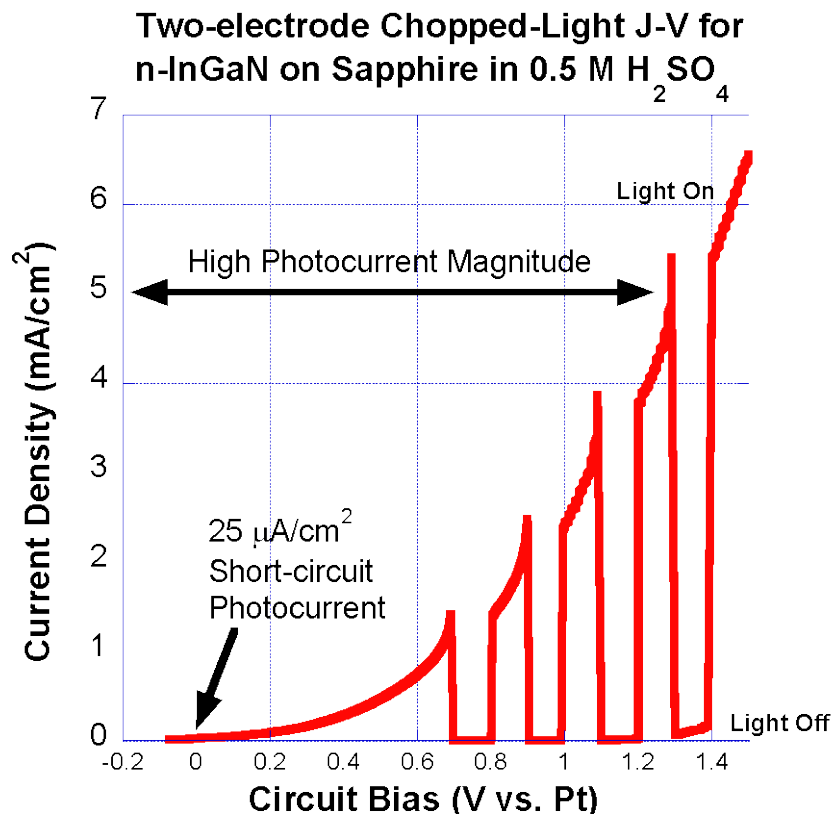
Noteworthy achievement for InGaN films:

- In-rich (30 to 40% In) InGaN films
- residual electron concentration in the mid-10¹⁸ range
- mobility ~80 cm²/Vs
- thicknesses suitable for use as PEC electrodes

Still actively working to improve:

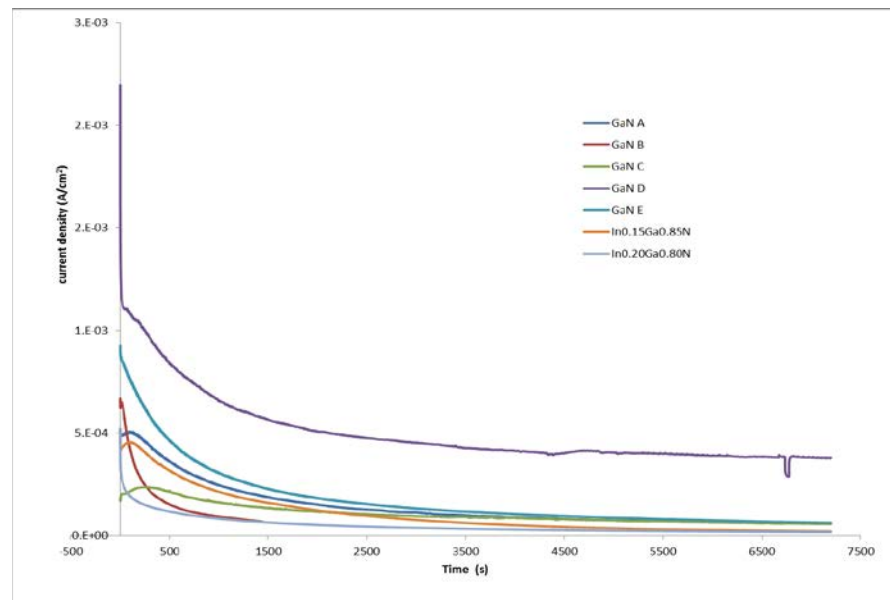
- film quality
- further reduce the electron concentrations to lower levels to improve performance and stability

Accomplishments and Progress – PEC Performance of InGaN Under Illumination



5mA/cm² is about 50% of theoretical AM1.5G max for this ~2.2 eV band gap material

Performance may be increased by using conductive substrates and p-type films



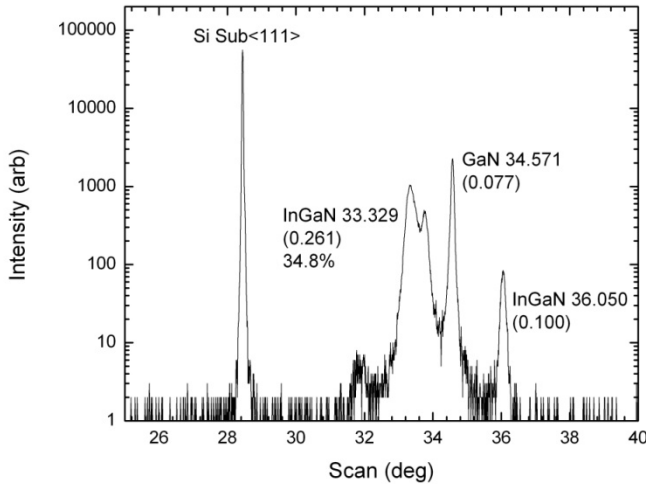
InGaN films show photocurrent at 0 V applied bias, suggesting the interfacial energetics are favorable for spontaneous photoelectrolysis

Both InGaN and GaN films show degradation with time.

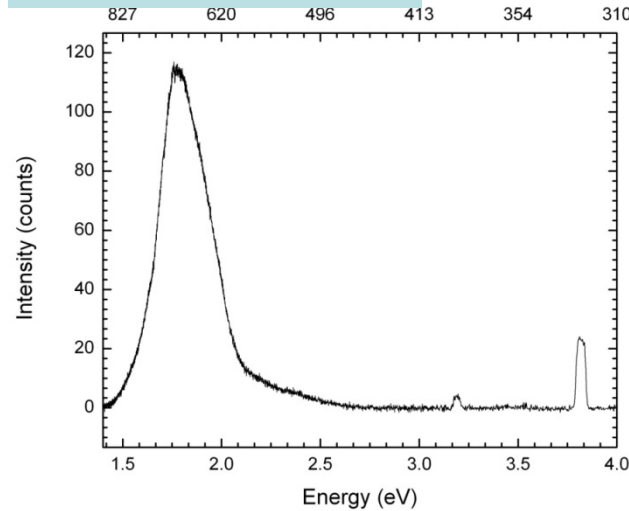
Films are N-polar, and n-type, which may be reason for stability problem.

Accomplishments and Progress - Characterization Results for InGaN on Si(111) ~35% In Content w/ AlN Buffer

XRD measurements

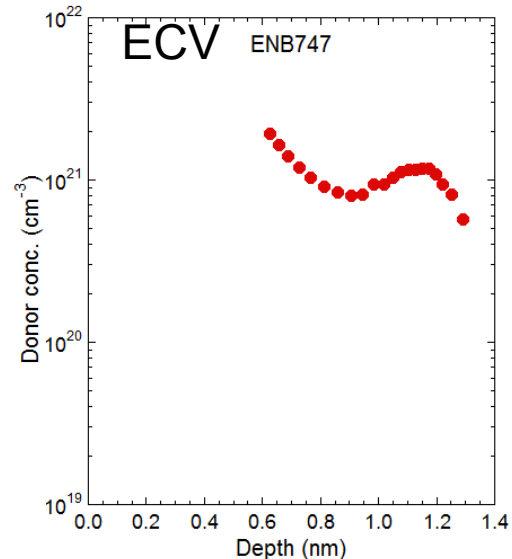
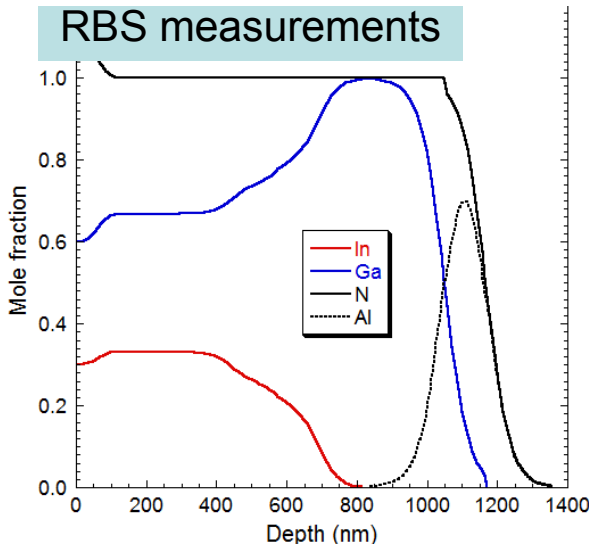


Photoluminescence



Growth of conductive substrate (Si<111>) may lead to improved electrode performance

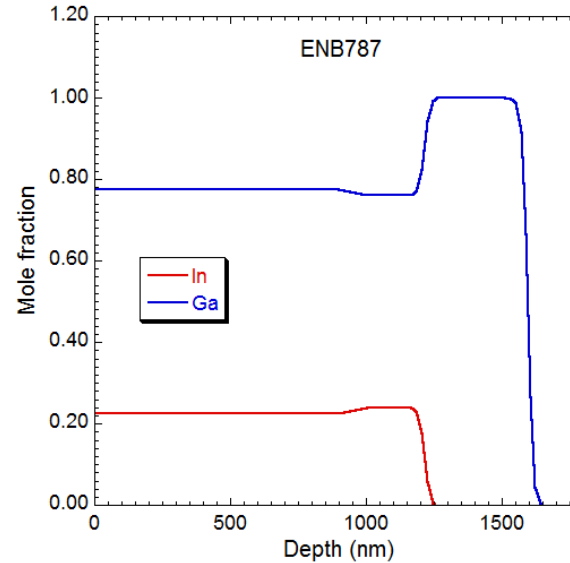
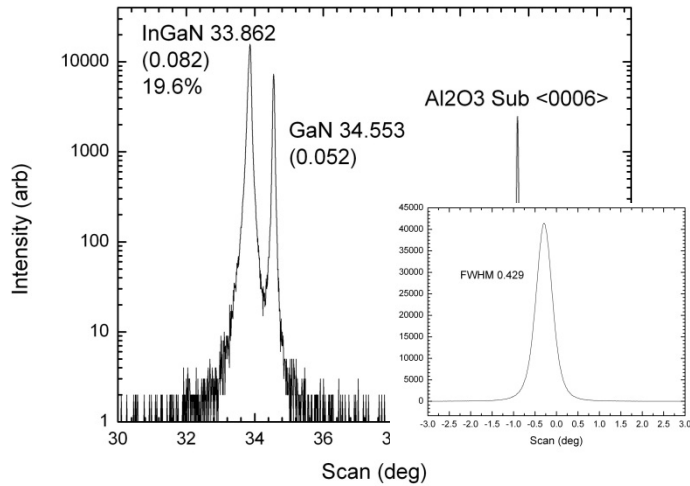
RBS measurements



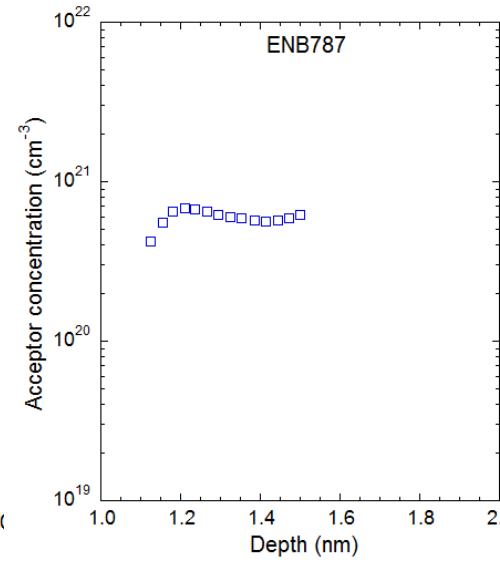
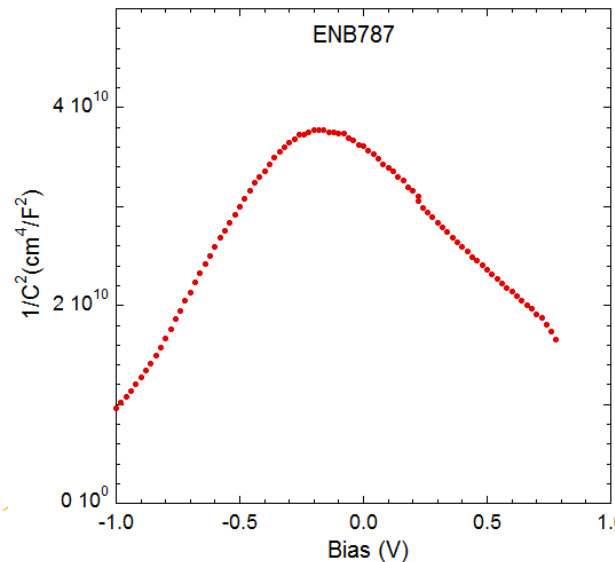
35% In content grown with composition grading
ECV: n-type $\sim 1 \times 10^{21} / \text{cm}^3$
AFM surface roughness: ~ 30 nm RMS

Accomplishments and Progress - InGaN (~20%In) p-type film growth

ENB-787 2 Theta



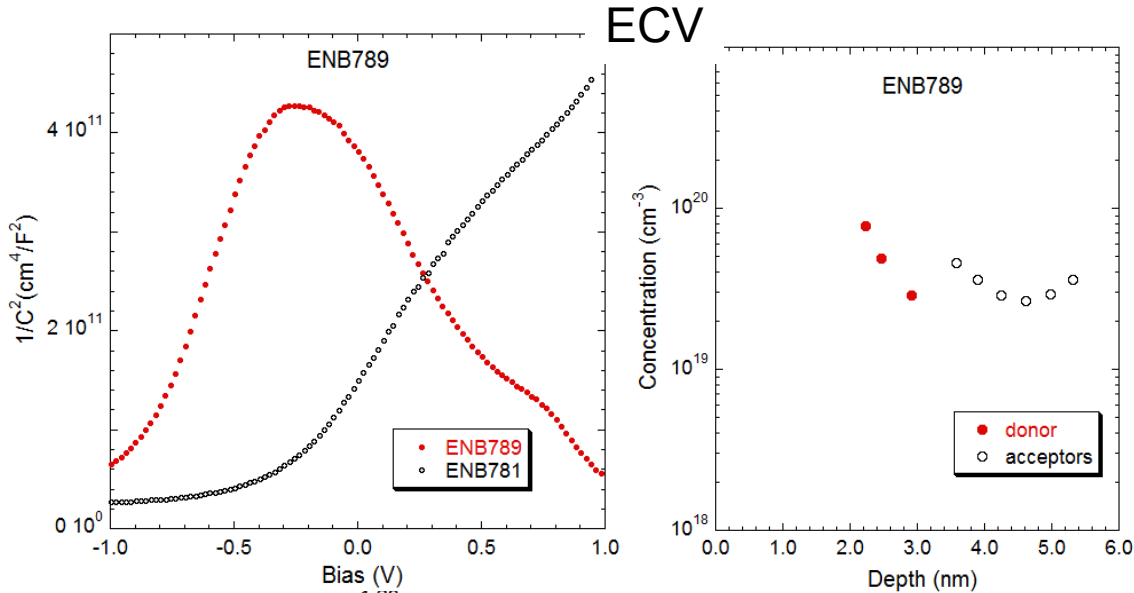
P-type films, with majority hole carriers, should be more stable under illumination



p-InGaN ~20% In:
Mg doped film, PL quenched, as expected

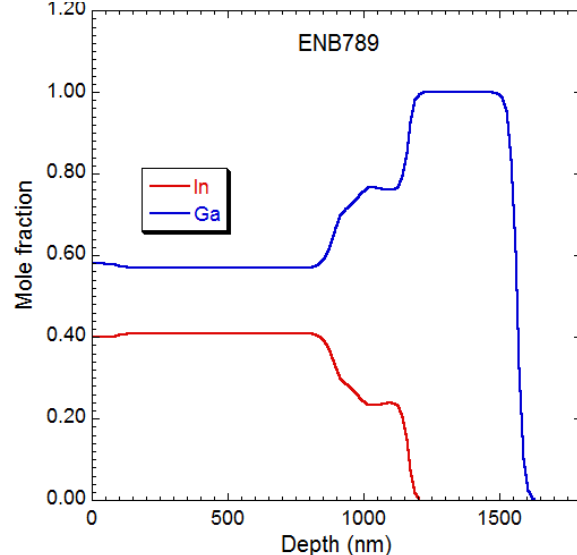
Hall: (p-type)
 $p = 2.2 \times 10^{19} / \text{cm}^3$ (assuming $0.3 \mu\text{m}$)
 $\mu = 2.75 \text{ cm}^2/\text{Vs}$
~3.5% active acceptors

Accomplishments and Progress - InGaN (~40%In) p-type film growth



P-type InGaN films with In content above ~33% have surface electron accumulation, which may offset some of stability improvement gained from p-type doping.

In content may need to be limited to less ideal values

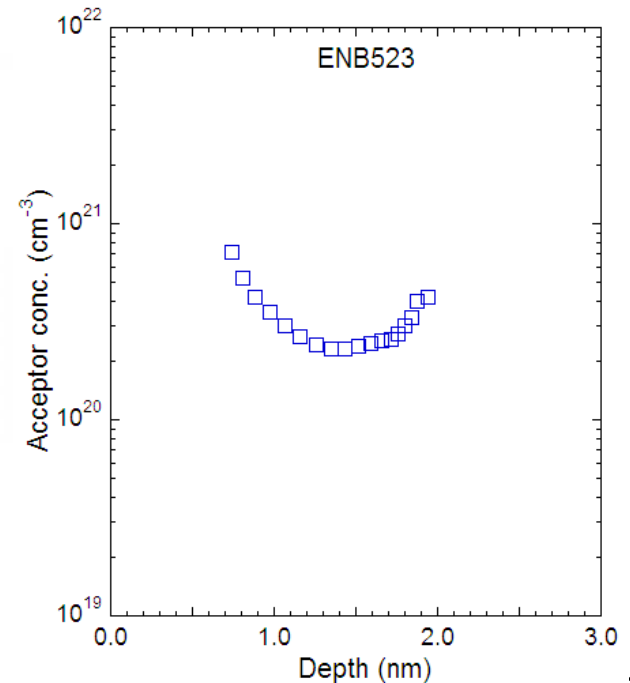
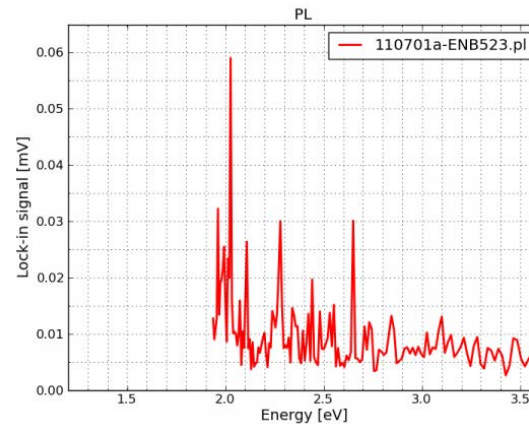
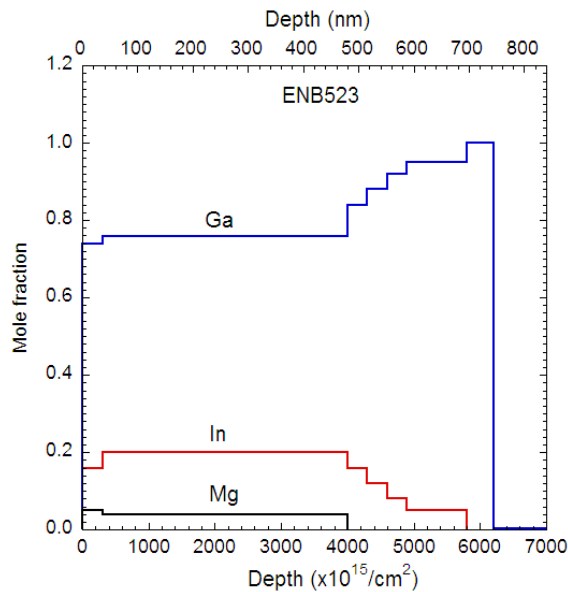


p-type InGaN graded from 20%In to 40%In
p-InGaN ~40% ~1200 nm

Hall: (n-type)
 $n=2.25 \times 10^{17}/\text{cm}^3$
(assuming 1 μm)
 $\mu=4.6 \text{ cm}^2/\text{Vs}$

Accomplishments and Progress - 20% InGaN on <111> Si, Mg-doped

- InGaN on <111> Si with AlN and graded InGaN buffer, with graded layer starting at 5% InGaN
- Graded buffer was graded over ~200 nm from 5% InGaN to 20% InGaN
- InGaN film is 500 nm thick
- Film was Mg-doped, shows active acceptors by ECV, PL quenched



Collaborations

Within DOE Hydrogen and Fuel Cells Program

- National Renewable Energy Laboratory (NREL)
 - InGaN films' PEC electrode performance evaluated by John Turner and Todd Deutsch. Team is prime POC for this work, NREL analysis is key for ongoing work on InGaN for PEC H₂ Production

Outside of DOE Hydrogen and Fuel Cells Program

- Lawrence Berkeley National Laboratory
 - Materials characterization on InGaN films including RBS, electrical characterization, and PL performed by team of W. Walukiewicz, K.M. Yu, and L. Reichertz
- Arizona State University
 - C. Honsberg and F. Ponce groups assist with InGaN film growth strategies and provides advanced XRD and PL characterization of films.

Proposed Future Work

Address issue of low photocurrent

- Low photocurrent is most likely related to material quality and use of n-type films. To improve, we will seek to improve material quality and improve p-type doping.

Address issue of electrode stability

- N-polar InGaN material, that is currently produced, is known to corrode in conditions similar to what is used during PEC H₂ production.

Investigate tandem cell configurations for improved efficiency



$\text{In}_{.29}\text{Ga}_{.71}\text{N}$

$\text{In}_{.33}\text{Ga}_{.67}\text{N}$

$\text{In}_{.40}\text{Ga}_{.60}\text{N}$

$\text{In}_{.41}\text{Ga}_{.59}\text{N}$

$\text{In}_{.48}\text{Ga}_{.52}\text{N}$

Mandatory Summary Slide

Relevance: The objective of this work is to explore the use of Group III Nitrides, specifically InGaN, for Photoelectrochemical Hydrogen Production

Approach: Utilize unique MBE capability at LANL to produce InGaN with quality, thickness, and In-content not available by conventional film synthesis techniques

Technical Accomplishments: InGaN films show photocurrent at 0 V applied bias, suggesting the interfacial energetics are favorable for spontaneous photoelectrolysis. However, N-polar InGaN appears to corrode under operating conditions. To address this issue, we have begun to produce p-type material and are investigating strategies for producing more stable, Metal-polar InGaN

Collaborations: Collaborations with NREL for evaluation of InGaN PEC performance; Collaborations with LBNL and ASU for film characterizations and film growth strategies

Proposed Future Work: Address issue of low photocurrent; Address issue of electrode stability; Investigate tandem cell configurations for improved efficiency