

Systems Engineering of Chemical Hydrogen, Pressure Vessel, and Balance of Plant for On-Board Hydrogen Storage

J. Holladay (P.I.), K. Brooks, K. Simmons, E. Rönnebro, M. Weimar, A. Karkamkar, R. Pires, M. Westman

> DOE Fuel Cell Technology Program Annual Merit Review

Washington, DC May 14, 2013 Technology Development Managers: Ned Stetson and Jesse Adams



U.S. Department of Energy Energy Efficiency and Renewable Energy Bringing you a prosperous future where energy is clean, abundant, reliable, and affordable



This presentation does not contain any proprietary, confidential, or otherwise restricted information

Project ID: ST005

Overview

Timeline

- Start: Feb. 2009
- Project End: Jan. 2014
 - End Phase 1: 2011
 - End Phase 2: 2013
 - End Phase 3: 2014
- Percent complete: 65%

Budget

- \$5.8M Total (PNNL) Program
 - DOE direct funded
 - No cost-share required for National Lab
- FY13: \$700k
- FY12: \$895k
- FY11: \$960k



United Technologies Research Center

Barriers

- A. System Weight and Volume
- B. System Cost
- C. Efficiency
- D. Durability

Partners

GΜ

HEXAGON

- E. Charging / Discharging Rates
- G. Materials of Construction
- H. Balance of Plant (BOP) Components
- J. Thermal Management
- O. Hydrogen Boil-Off

1.1

S. By-Product/Spent Material Removal

National Renewable

HSM

Pacific No

Energy Laboratory Innovation for Our Energy Future

Relevance: Hydrogen Storage

Impact to FCT Program

- Demonstrate hydrogen storage system that meets DOE 2017 targets for light duty vehicles using chemical hydrogen storage
- Apply materials discoveries from the Materials Centers of Excellence
- Discover/develop engineering solutions to overcome material's deficiencies
- Identify minimal performance for materials to be applicable in engineered H₂ storage systems for light duty vehicles.
- Hydrogen Storage Community at Large
 - Develop and/or advance modeling and simulation tools for the optimum design and engineering of on-board storage systems
 - Provide functional prototype systems available to OEMs
 - Provide engineering methodologies, analysis tools, and designs applicable to stationary storage and portable power applications
 - Demonstrate on-board storage to advance state of the art.
 - Identify, develop and validate critical components either for performance, mass, volume, or cost.

Approach:

- PNNL's Roles Supporting Engineering Center Structure
 - Technology Area Lead (TAL) for Materials Operating Requirements
 - Coordinate activities as the Technology Team Lead (TTL)
 - Bulk Materials Handling (Transport Phenomena)
 - Pressure Vessels (Enabling Technologies)
 - Manufacturing and Cost Analysis (Performance Analysis)
 - Liaison to VT Program projects and resources
- Technical Objectives of PNNL Scope:
- Design chemical hydrogen H₂ storage system & BOP components Chemical Develop system models to predict mass, volume, performance Hydrogen Reduce system volume and mass while optimizing storage capability,
 - fueling and H₂ supply performance
 - Mitigate materials incompatibility issues associated with H₂ embrittlement, corrosion and permeability
 - Demonstrate the performance of economical, compact lightweight
- All Systems

4

- vessels for hybridized storage
- Guide design and technology down selection via cost modeling and manufacturing analysis
 - Perform value engineering of BOP to minimize cost, volume and mass
- Phased/ gated progressions aligning with HSECoE go/no-go decisions

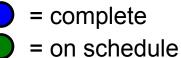
FY13 Objectives

- Chemical Hydrogen Storage Design
 - Validate models and concepts via experiments
 - Scale-up slurry production
 - Assess feasibility of liquid-slurry chemical hydrogen storage
 - Assess feasibility of volume-exchange tank
 - Assess feasibility of slurry use with heat exchanger, pump, valves.
- Pressure Vessel for Cryo Adsorbent Hydrogen Storage
 - Exercise "tankinator" model to assess materials and design options for type I, III, and IV vessels
 - Optimize vessel design in terms of cost
 - Assess vessel cost as function of pressure and temperature
- Balance of Plant
 - Maintain BOP library
 - Size components (heat exchangers, valves, pumps,...)
 - Determine material compatibility
 - Identify where improvements can be made
- Cost Modeling



Accomplishments: Milestones FY13

Q2 🔵	Task 1	Report on Feasibility of a volume exchange tank operation with AB slurry	
Q2 🔵	Task 1	Report on ability to identify a system design having mass<97kg, volume <118L, and be able to meet all the drive cycles	
Q1 🔵	Task 2	Report on settling and flocculation rates for a 45 wt% AB slurry (both fresh and spent)	
Q3 🔘	Task 2	Report on feasibility to scale-up the slurry production process	
Q1 🔵	Task 4	Complete the analysis on which tank type to move forward with in Phase III.	
Q2 🔵	Task 4	Complete the tank design mass and volume models for publication on the HSECoE website.	
Q3 О	Task 4	Complete the materials database from components in the SA designs	
4/30	Task 5	Update Cost Analysis for Chemical Hydrogen and Cryo- Sorbents	



- = on sched
- = at risk
- = behind

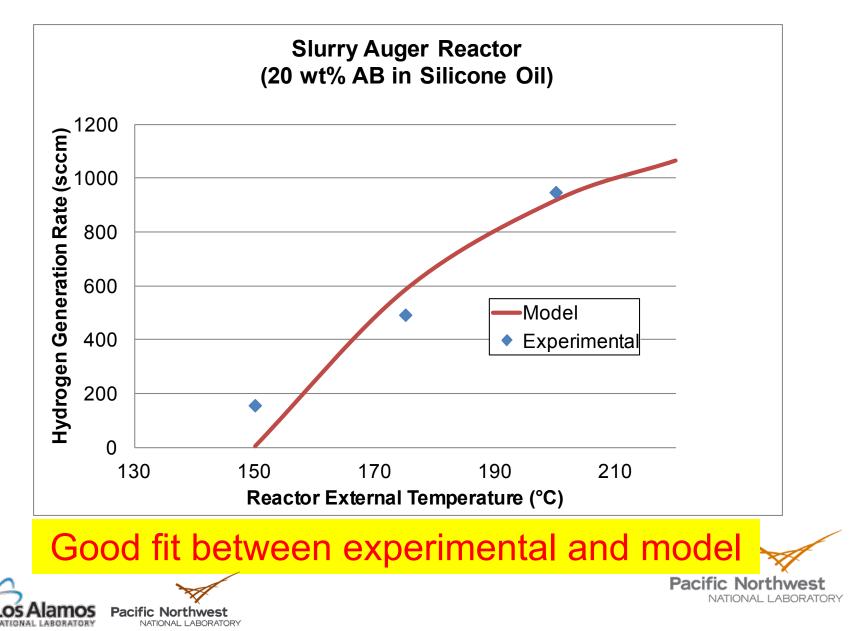


Chemical Hydrogen Storage Development

- Modeling and Validation
 - FY12: System Modeling and Operational Envelope
 - FY13: System improvements (mass, volume, cost)
 - FY13 Component Validation
 - Reactor: LANL Model validation: PNNL
 - Gas-Liquid- Separator: UTRC
 - Radiator/Heat Exchanger: PNNL
 - Pump : PNNL
 - Displacement volume tank: PNNL
- Liquid-Slurry development (reviewer section)
 - Endothermic liquid-slurry: Alane surrogate leverage BNL's work
 - Exothermic liquid-slurry: Ammonia borane surrogate
 - 45-55 wt%
 - Flow reactor
 - Scale-up
 - Settling/flocculation rates

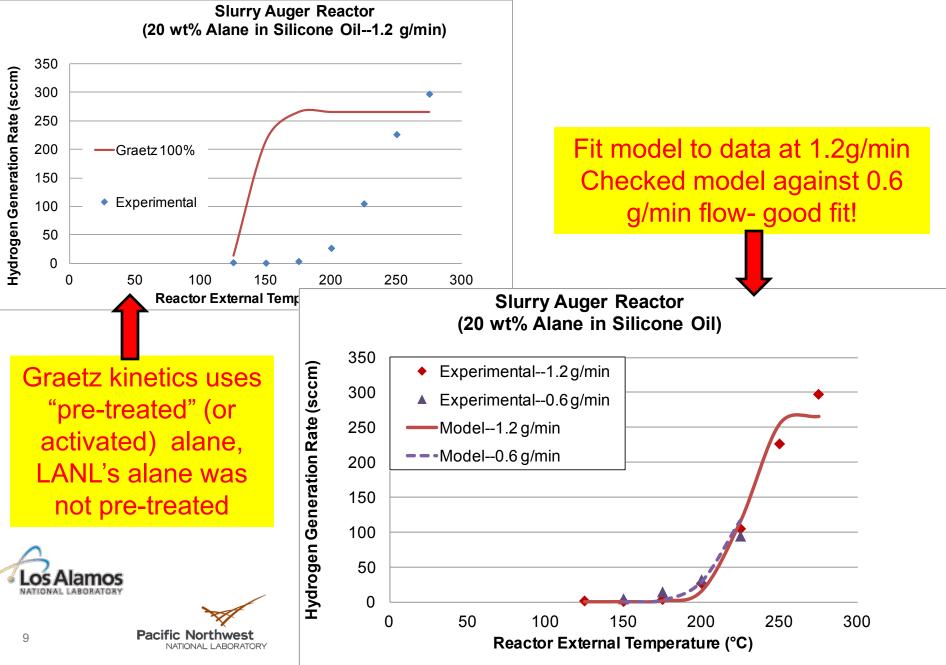


Accomplishments: Reactor Model Validation



8

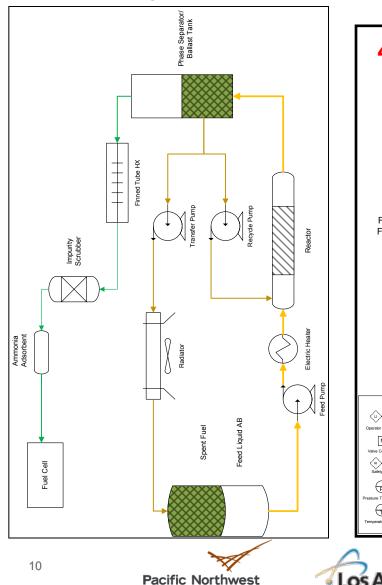
Accomplishments: Reactor Model Validation



FY12

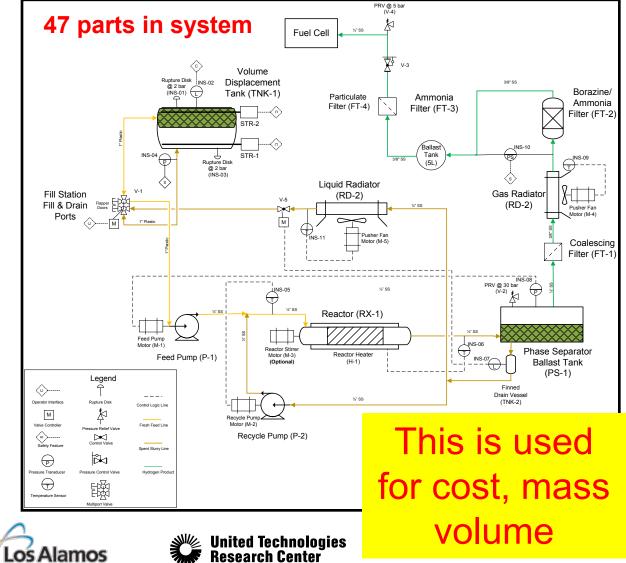
FY13

12 parts in system



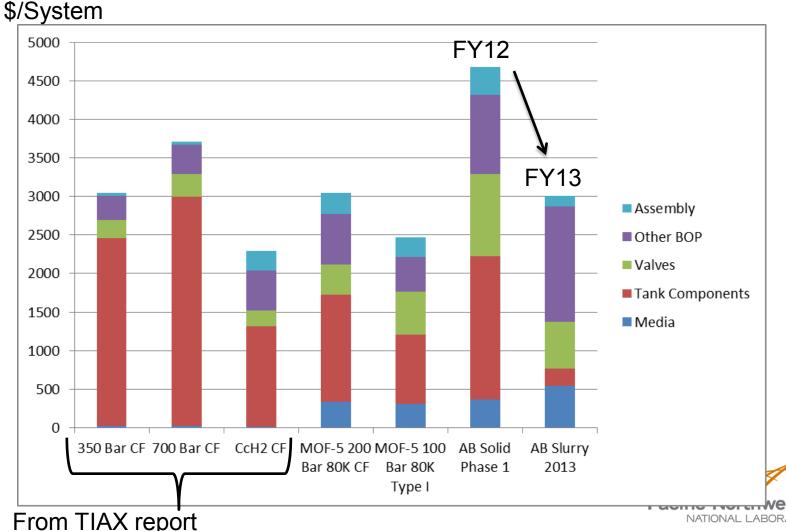
NATIONAL LABORATORY

Added Instrumentation, Controls, Control Logic, Engineering Improvements



Accomplishment: Chemical Hydrogen Storage Cost

Tank significantly reduced costs by moving from high pressure carbon fiber tank to low pressure fiberglass tank

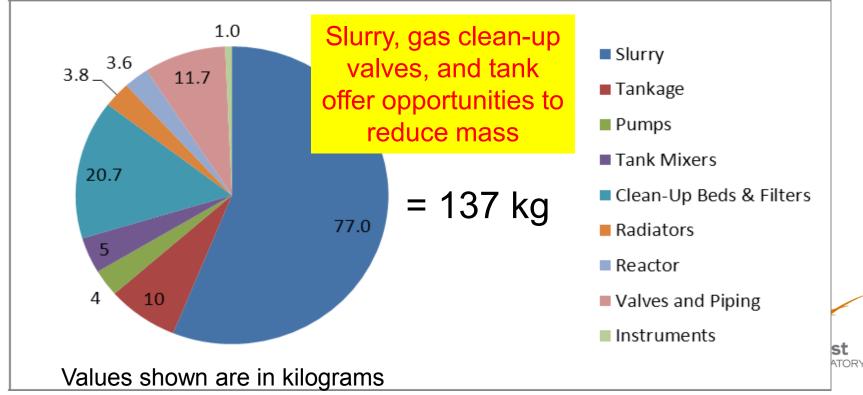


NATIONAL LABORATORY

Accomplishment: Milestone: System Mass and Volume

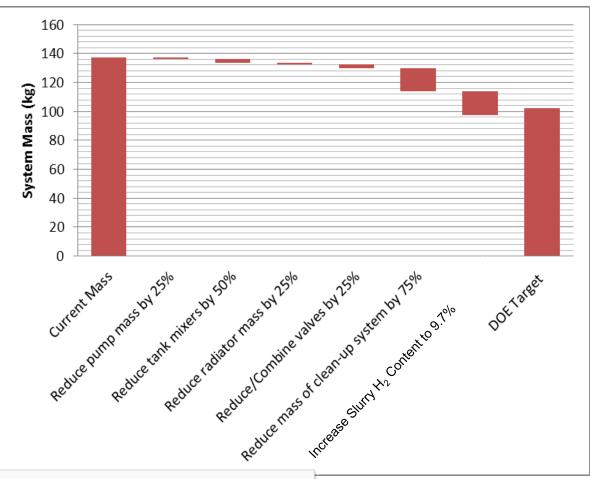
Purpose

- Demonstrate SMART Milestone: "Report on ability to identify a system design having ..."
 - mass less than 97 kg
 - volume less than 118 liters
 - meeting the all of the HSECoE drive cycles



Smart Milestone: Chemical Hydrogen BoP

Report on ability to identify BoP components (does not include AB slurry) suitable for Chemical Hydrogen system having a mass no more than 41kg and a system volume no more than 57 liters.



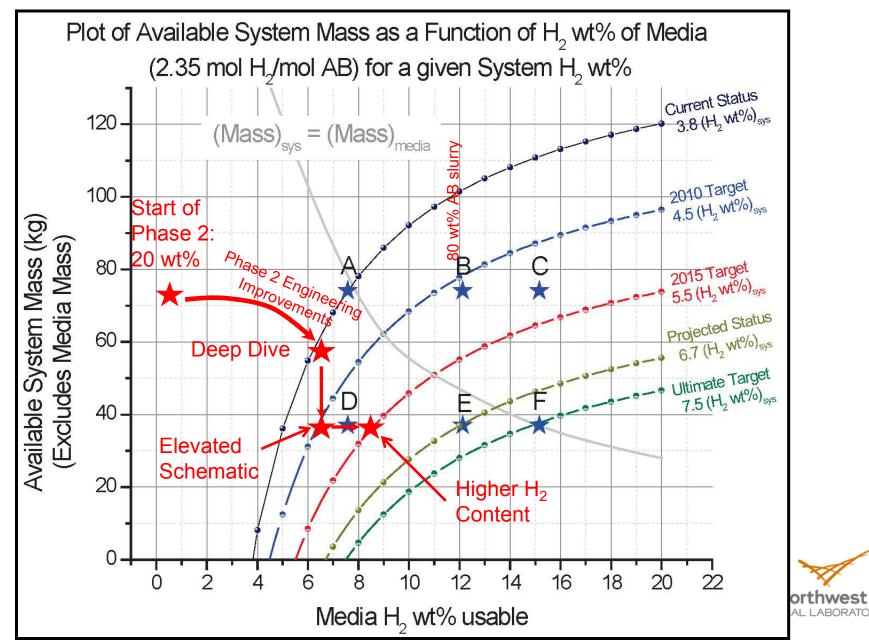
Chemical Hydrogen System BoP

Pathway to meet Milestone Targets Proposed

Current BOP Mass 53.8 kg BOP Volume 61.2 Proposed pathway BOP Mass 38kg BOP Volume 41L



Accomplishments: Progress in Phase 2



Accomplishment: What about Alane?

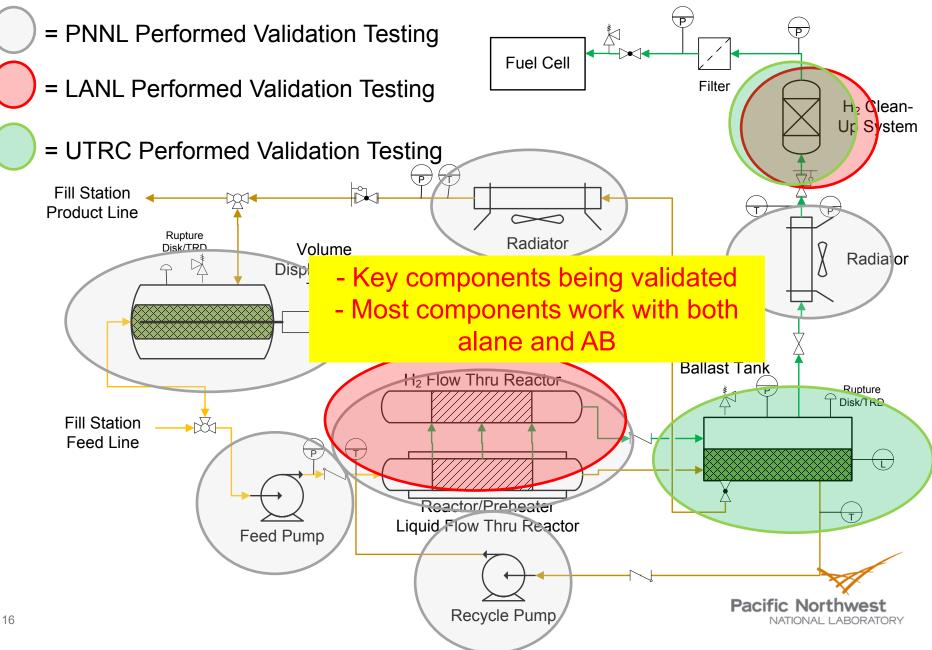
Subtractions

- No clean-up adsorbent beds (-19 kg)
- No recycle pump (-2 kg)
- Additions
 - Larger volume displacement tank
 - Lower H₂ loading (assuming 50 wt%) (+39 kg)
 - Larger fraction needed for parasitic losses (+10 kg)
 - Larger reactor (+1 kg)
 - Recuperator (+1 kg)
- Net ~ 30 kg heavier for alane
- Other considerations: Easier pumping, easier gas cleanup

Meeting mass target more difficult with alane



FY13: Component Validation



Component Photographs

not shown: 3- way valves and PRV







Results of System Flowability Tests

- FY12 FMEA identified as major concern
- Settling Tests
 - Pump had no difficulty pumping any of the simulants after settling
 - Polyimide powder—100 hours settling: Pressure fluctuations associated with slugs of slurry moving through flow restrictions
 - Spent AB—70 hours settling: No increase in start-up load, no pressure fluctuations
 - System disassembly revealed no major slurry issues
 - Some slurry solids deposited in radiator bends with polyethylene slurry after ~14 hours of settling and running: large radius bends recommended
 - Additional tests need to be done
- Cold Start-up Tests
 - Polyethylene simulant flows at -24°C
 - Polyimide flows at -29°C and spent AB at -20°C

Tests with spent AB and simulants show no immediate show stoppers



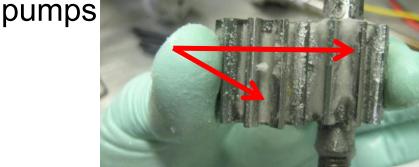
Flowability Tests with Fresh AB

Approach

- Tested 50 mL of 20 wt% AB
- Pumped in a simple loop with 3/8" tubing, pressure gauge and valve
- Smaller loop required because limited material available

Results

- Spent AB- no issues
- AB formed clogged gears
- Peristaltic pump and syringe
- R&D needed to scale down positive displacement type



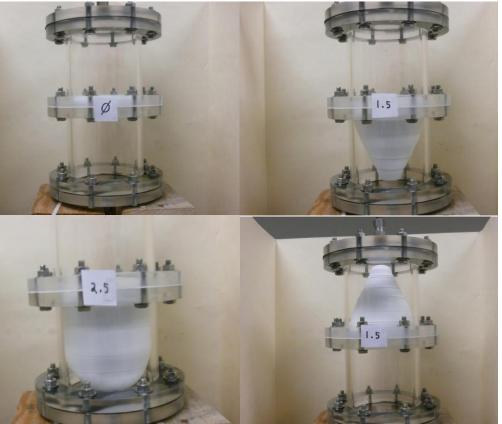
- Peristaltic/ syringe pump okay for Phase III
- Gear pumps clogged
- R&D needed



Accomplishments: Volume Exchange Tank-"Pleated" Membrane

- Purpose
 - Minimize the amount of stretching of the membrane to prolong its life
 - Allow the use of a wider range of elastomers
 - Pleated membranes enable conformable tanks
- Approach
 - Fabricated a small scale (4" diameter) membrane with silicone
 - Test with water and air to understand the issues and demonstrate the concept

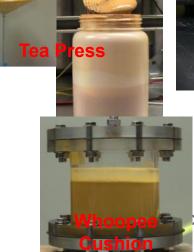




Accomplishments: Mixing in Volume Displacement Tank

- Scoping Tests Performed
 - Fresh AB/Fresh AB simulant (polyethylene in water/glycerin)
 - Poor Mixing: Ultrasound Testing, vibration
 - Good Mixing: Mixing with impeller
 - Good Mixing: Tea Press Mixing
 - Spent AB/Spent AB simulant (polyimide in silicone oil)
 - Poor mixing: Ultrasound Testing, vibration, whoopee cushion test
 - Mixed, but had other issues: mixing with impeller, extrusion
 - Good Mixing: Tea Press Mixing
 - Good Mixing: Jet Mixing

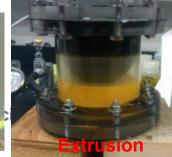
Jet Mixing and Tea Press moved forward to larger scale testing





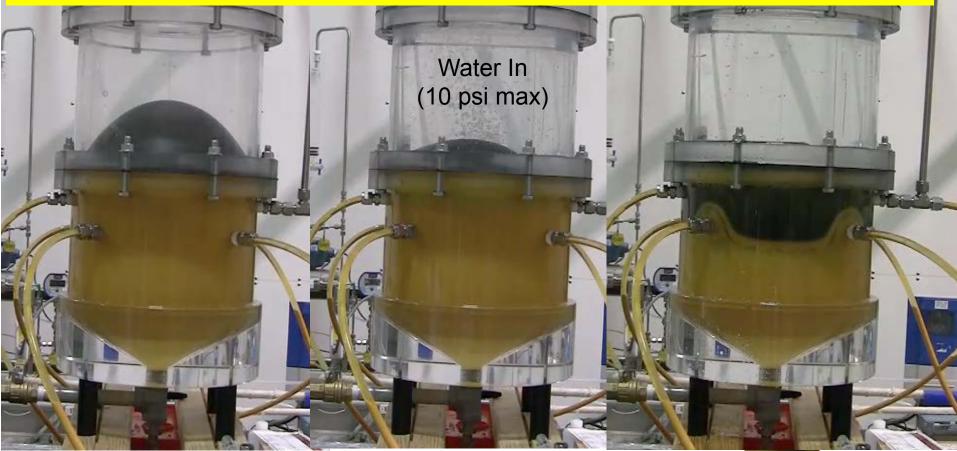






Accomplishment: Volume Exchange Tank Spent Chemical Hydride Tea Press Test

1.7 minutes to fluidize settled solids and remove 10.9 kg of simulant from tank meeting exchange tank milestone



t = 0 Tea Press 8 strokes 22 7 kg max force t = 30 sec Drain 10.9 kg simulant 35 wt%



Cryo-Adsorbent Hydrogen Storage

▶ FY12:

- Exercise "tankinator" model to assess materials and designs
- Optimize vessel design in terms of cost
- Assess chemical compatibility of polymer liners for type IV
- ▶ FY13
 - Sent tankinator model to SRNL for posting
 - Integrate tankinator model into SRNL model (SRNL complete)
 - Examined tank pressure and types to understand costs
 - Costed135 vessels as function of pressure and temperature (with Hexagon Lincoln and SRNL)
 - Identified welding techniques for use with adsorbents
 - Out selected type IV
 - BoP assessment
 - Thermos bottle design
 - Test in Phase III



Smart Milestone: Cryo-Adsorbent Type IV Vessel Liner

Report on ability to identify Type IV tank liner materials suitable for 40K operation having a mass less than 8 kg and a volume less than 3 liters (2.55 mm thickness). Type IV Cryogenic Pressure Vessel Manufacturing







Metric not feasible

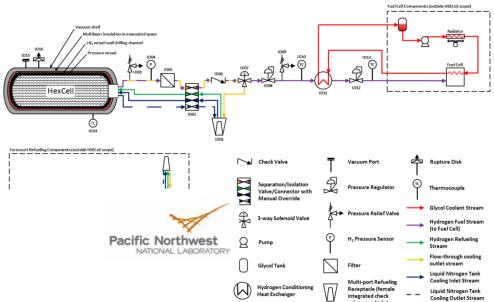
Evaluated 8 different materials. 2.55 mm liner separates from shell at pressure<35bar <u>Recommended type I or III vessel</u>



NATIONAL LABORATORY

Smart Milestone: Cryo-Adsorbent Balance of Plant

Report on ability to identify BoP materials (excluding internal HX, external HX, and combustor) suitable for 60 bar cryogenic adsorbent system having mass less than 17 kg and a volume less than 18.5 liters.

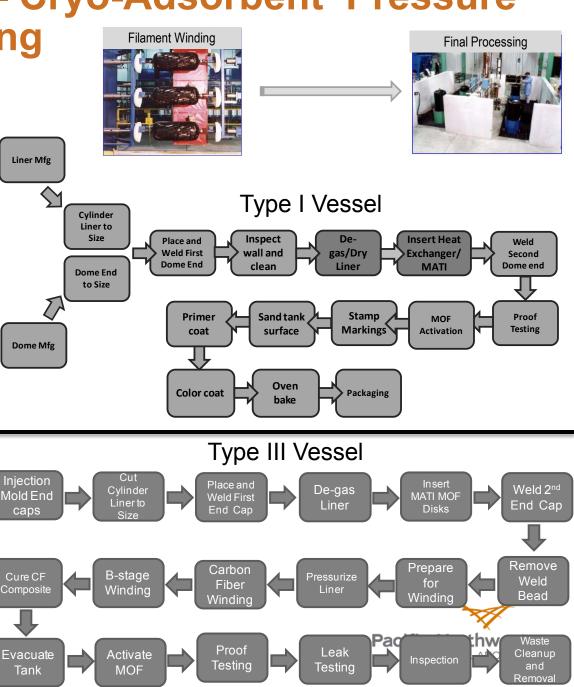




Pacific Northwest NATIONAL LABORATORY

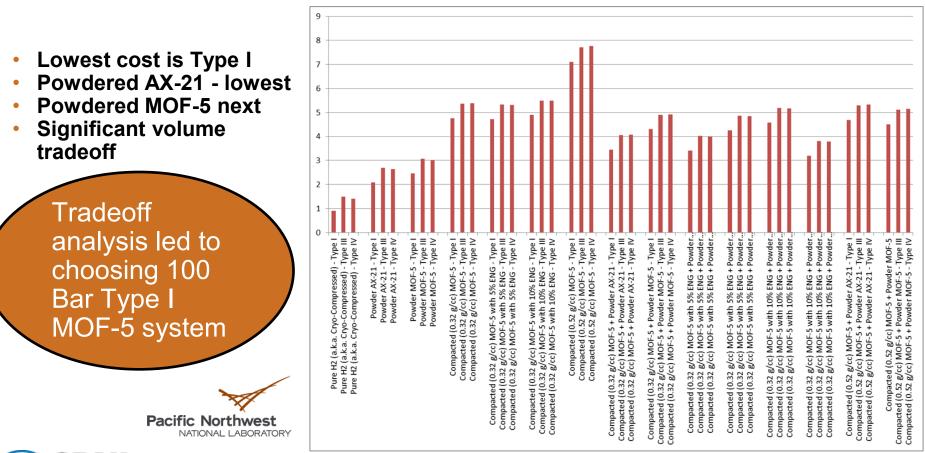
Accomplishments – Cryo-Adsorbent Pressure Vessel Manufacturing

- Developed a Type I & III manufacturing approach
 - Processing conditions
 - Developed manufacturing approach for MATI or heat exchanger insertion
 - Tradeoff study in manufacturing and joining techniques
 - Identified friction stir welding as best technique to minimize temperature effects on MOF and heat treated material



Accomplishments: Costed 135 Cryo-Adsorbent Tank Configurations

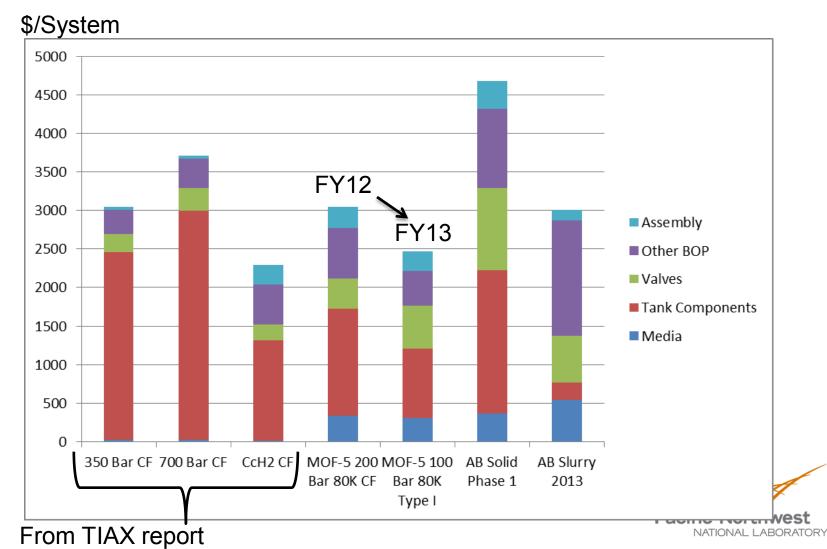
- Series of tank evaluations at 60, 80, 100 Bar with appropriate BOP
 - Type I, Type III & Type IV with different media
 - Type III weighs 61% of Type I, but 14% more costly
 - Cost for MOF-5 systems still decreasing at 100 Bar





Accomplishment: Cyro-Adsorbent Hydrogen Storage Cost

- More expensive than CcH2 CF
- Less than previous estimates at 200 Bar



Collaborative Activities

Hydrogen Storage Engineering Center of Excellence	 Hexagon Lincoln - study of CF cost and pressure vessel design modeling GM - design of structured media bed for MH Ford - characterization of absorbent materials UQTR - design and materials characterization of carbon absorbent OSU - microarchetecture device concept development and thermodynamic analysis UTRC - develop solutions for H₂ impurities filtering LANL - AB system design and measure H₂ impurities NREL - input for tank to wheels analysis and system cost models SRNL - study AB reactivity and kinetics model development
SSAWG	Participate in group discussions and analysis
Materials 'Reactivity' Program	 Khalil (UTRC) and Anton (SRNL) - understand reactivity properties of AB Van Hassel (UTRC) - study impurities in H₂
Independent Analysis	 SA - provide design details for AB refueling cost and feasibility assessment, plus share cost parameters for system cost modeling

Future Work

- Design sub-scale prototype (joint with partners)
 - Exothermic slurry (AB), Exothermic liquid, Endothermic slurry (Alane) to be tested
 - Cryo-adsorbent system design
- Chemical hydrogen
 - AB Slurry production
 - Alane slurry characterization
 - Model verification
 - Compare model to prototype system results
 - Refine model if necessary
 - Materials requirements
- Demonstrate advanced pressure vessel system for dormancy etc. (w/Hexagon Lincoln)
- Elevated design concepts
 - Sorbent: reduce part count by 30%
 - Chemical: reduce part count by 25%
- Refine cost model
 - Update BOP

30

Improve manufacturing estimates



Summary

> Relevance:

Addressing most of the engineering challenges for materials based hydrogen storage for endothermic and exothermic chemical hydrogen and cryo-adsorbents. Providing feedback and recommendations on materials requirements

> Approach:

Developed system models, experimental validation of the models, component validation, sub-scale prototype demonstration, refine models, cost estimates to guide selections.

Fechnical Accomplishments:

Developed 45+ wt% slurry, tested major components, demonstrated flow reactor, developed pathway to achieve mass/volume targets for chemical hydrogen, achieved mass/volume targets for adsorbent system, down selected vessel type for adsorbent, cost models

> Collaborations:

Extensive collaboration with all of our partners

Proposed future work:

Demonstrate sub-scale prototype, reduce part count of proposed full scale systems, validate models, cost analysis.

NATIONAL LABORATORY

Technical Back-up Slides



Accomplishments: Chemical Hydrogen Settling/flocculation Observations

- 45wt% Fresh AB and Spent AB
 - Settling during 3 hours; need to stir to reform slurry

At time zero

Time	Fresh AB Flocculation	Spent AB Settling
0 min	-	-
5 min	10.4%	21.1%
10 min	21.1.%	26.3%
1h	21.1%	26.3%
2 h	26.3%	36.8%
3 h	26.3%	36.8%
24 h	26.3%	36.8%



Note: % is based on volume change/total volume



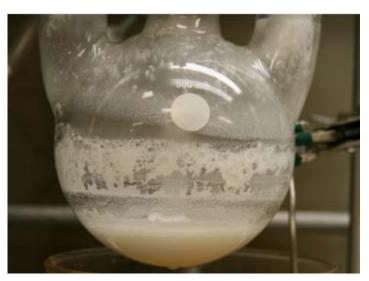
After 3 hours

Accomplishments: Chemical Hydrogen Slurry: Additive Screening Dow Triton[®] Products

- 1-3wt% loading in 35wt% Weylchem AB slurries
- X-15 significantly reduced foaming
 - Kept slurry stable throughout rxn, i.e. no solidification
- X-35 reduced foaming
 - Not as significant as X-15
- X-45 no effect

R = octyl (C8) x = 1.5 (avg)









SMART Milestone

- Report on ability to develop/identify a radiator/HX capable of cooling the effluent from 525K to 360K (252°C to 87°C) having a mass less than 1.15 kg and a volume less than 10.9 liters.
- Heat Exchanger—1.44 kg and 1.5 liters
- Meet Milestone:
 - Reduce fins from 14/inch to 11/inch—no significant decrease in overall heat transfer (mostly interior limited)

Milestone Can Be Met



FY13: Validation of the Volume Exchange Tank

- Milestone: Report on the feasibility of using a volume exchange tank that provides 5.6 kg of equivalent hydrogen to the process as a homogeneous feed slurry and can be refueled within the DOE target time.
 - Feed shall be on-loaded and product off-loaded in 2.9 minutes (2017 goal for 5.6 kg H₂ equivalent)
 - Volume exchange tank is required to meet DOE gravimetric and volumetric density
 - Develop and demonstrate a robust membrane for the volume exchange tank
 - Demonstrate mixing

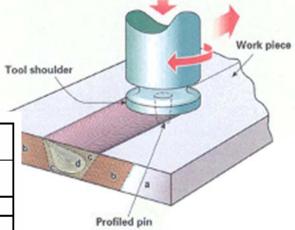


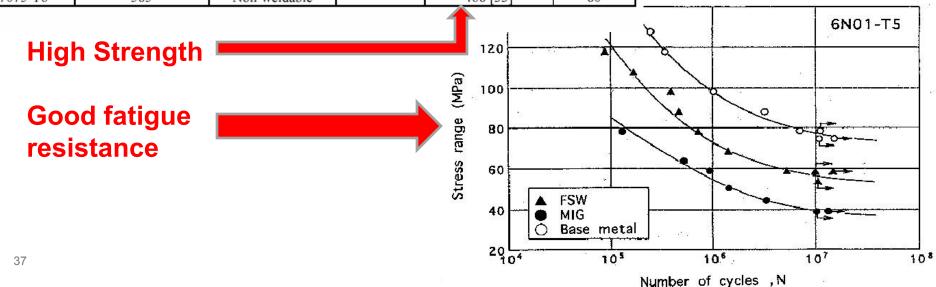


Cryo-Adsorbent Vessel Manufacturing

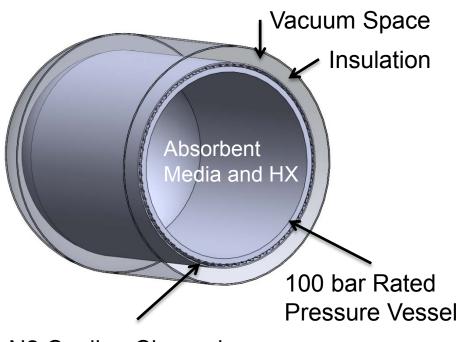
- Challenge: Traditional welding requires heat treatment to release stress. The heat treatment would damage the adsorbent
- Solution: Friction Stir Welding
 - Low temperature

Base Alloy	Parent Material [30]	Gas-Shielded Arc Welded Butt		Friction Stir Welding	
and Temper		Joint			
	Tensile strength	Tensile strength	% of Parent	Tensile strength	% of
	(MPa)	(MPa)		(MPa)	Parent
2024-T3	485	Non-weldable	-	432 [32]	89
6061-T6	415	207 [35]	50	252 [34]	61
6N01-T5	260	165 [31]	63	200 [31]	77
7075-T6	585	Non-weldable	-	468 [33]	80





Future Plans – Balance of Plant / Cost



LN2 Cooling Channels

- PNNL and Hexagon Lincoln to design and fabricate thermos bottle concept
 - Improved cooling over current designs
 - Uses liquid nitrogen for cooling not cryohydrogen
 - Decreased heat loss for improved dormancy
- Experimental validate pressure vessel cooling
- Experimental validate heat loss with tank design

