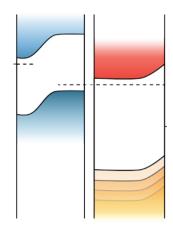


## NSF/DOE Solar Hydrogen Fuel Engineering Surfaces, Interfaces, and Bulk Materials for Unassisted Solar Photoelectrochemical (PEC) Water Splitting



Prof. Thomas Jaramillo Dept. of Chemical Engineering Stanford University June 9<sup>th</sup>, 2015 Project ID#: PD119



This presentation does not contain any proprietary, confidential, or otherwise restricted information



#### Timeline

- Project Start Date: 1/1/15
- Project End Date: 12/31/2017

### Budget

- Total Project Budget: \$750,000
  - Total Recipient Share: \$750,000
  - Total Federal Share: \$750,000
  - Total DOE Funds Spent\*: \$67,258
    \* As of 4/15/15

#### Partners

- National Renewable Energy Laboratory (NREL)
- PEC Working Group

#### **Barriers**

- Materials Efficiency Bulk and Interface (AE)
- Materials Durability Bulk and Interface (AF)
- Integrated Device Configurations (AG)

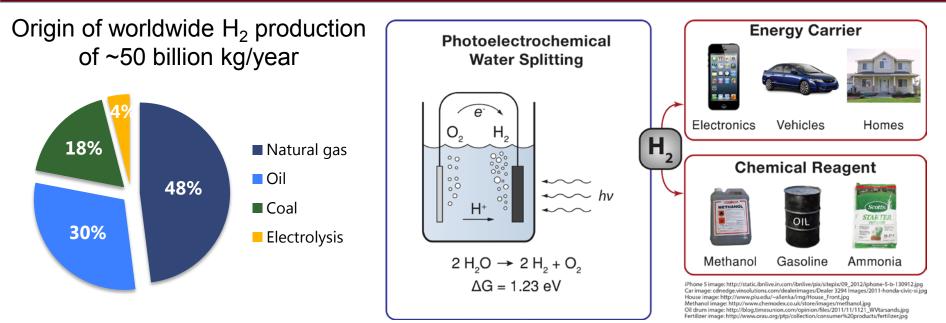
#### **Targets**

- Photoelectrochemical Hydrogen Cost
- Annual Electrode Cost per TPD H<sub>2</sub>
- Solar to Hydrogen (STH) Energy Conversion Ratio
- 1 sun Hydrogen production rate

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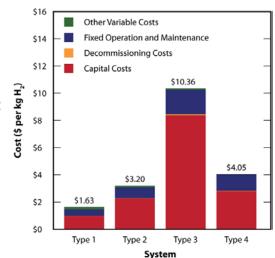
## Relevance and Impact

#### H<sub>2</sub> Production via Photoelectrochemical Water Splitting



Hydrogen is an important industrial chemical and potential future fuel. Photoelectrochemical (PEC) water splitting offers the potential for sustainable and  $H_2$  production from renewable resources, sunlight and water. Technoeconomic analysis of centralized PEC  $H_2$ production facilities shows that this process can become economically competitive with further improvements in device efficiency, durability, and cost.

U.S. Department of Energy & National Hydrogen Association. A. Midilli & I. Dincer. *International Journal of Hydrogen Energy* **2007**, 32, 511 Pinaud, B.A., T.F. Jaramillo, et al. *Energy & Environmental Science*, 2013. **6** (7): 1983-2002.



## **Relevance and Impact**

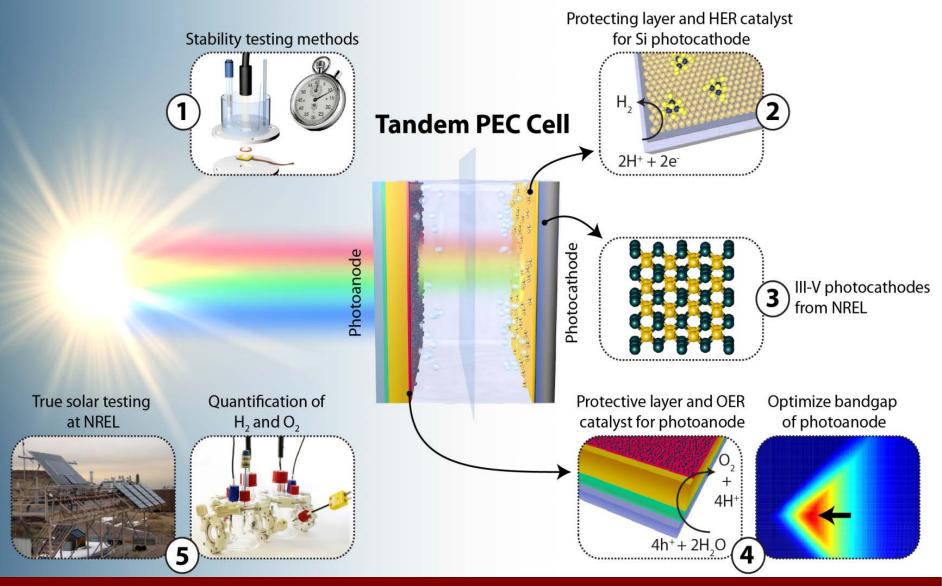
#### **Objectives**

- Method and protocol development to understand photoelectrode corrosion *in acid*.
- Interfacial engineering of the Si surface to provide enhanced catalytic activity and corrosion resistance <u>in acid</u> by means of molybdenum sulfide nanomaterials.
- Interfacial engineering of III-V photocathode surfaces with similar approaches, in collaboration with Dr. Todd Deutsch at the National Renewable Energy Laboratory (NREL).
- Interfacial engineering of the BiVO<sub>4</sub> surface to provide enhanced electronic properties, catalytic activity and corrosion resistance <u>in acid</u> with a series of ultra-thin metal / metal oxide films.
- Quantification of  $H_2$  and  $O_2$  and true solar testing at NREL.

#### **Technical Targets:**

10% STH Efficiency 100 J/s per m<sup>2</sup> of Hydrogen Production

### Approach Testing Design and Tandem Device Engineering

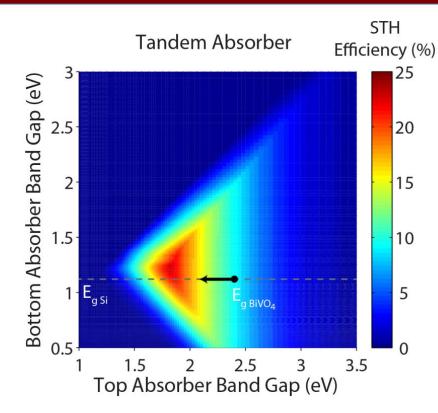


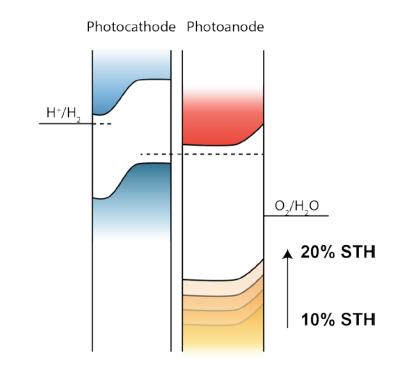
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#### Approach Theory Guided Material Selection





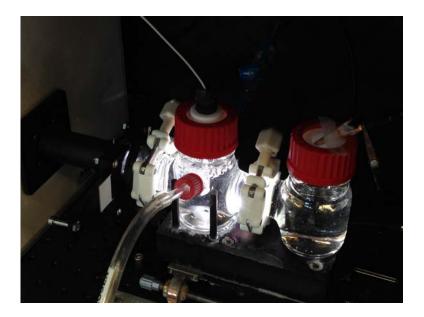
Modeling of realistic STH efficiency as a function of band gaps for a tandem absorber PEC system shows that 20% STH can be achieved with a tandem device with band gaps of 1.2 and 1.8 eV Theoretical model is used to direct material selections and provide realistic goal for STH of each tandem system

Seitz, L. C.; Chen, Z.; Forman, A. J.; Pinaud, B. A.; Benck, J. D.; Jaramillo, T. F. ChemSusChem 2014, 7 (5), 1372-1385.

#### Accomplishments Stability measurement setup

New photoelectrochemical setup improves the reliability of photoelectrochemical measurements

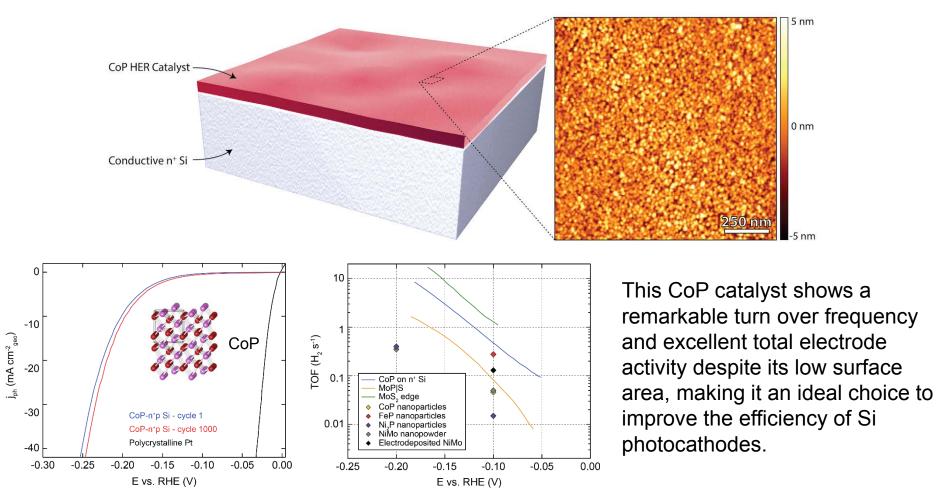
- Precise control over the position of the electrode
- High uniformity in the light spot





#### Accomplishments Cobalt Phosphide HER Catalyst

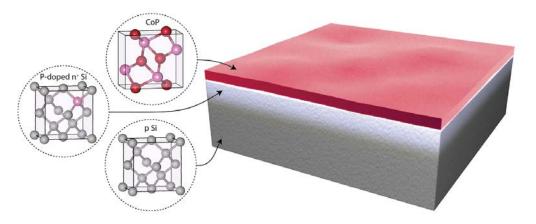
CoP, a nonprecious metal phosphide, shows very high activity for the HER. We synthesized a flat, thin film CoP catalyst on conductive silicon.



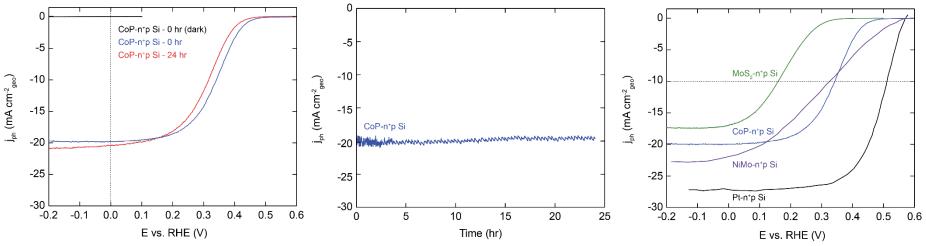
T.R. Hellstern, T.F. Jaramillo, et al. submitted (2015)

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### Accomplishments CoP-n<sup>+</sup>p Si Photocathode



The CoP catalyst incorporated onto photoactive n<sup>+</sup>p Si shows excellent activity and stability. This performance puts it among the best non-precious metal Si photocathodes ever tested in acidic or basic electrolyte.



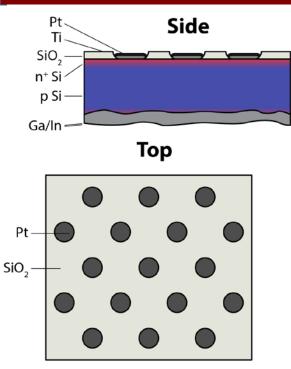
CoP-n<sup>+</sup>p Si data: T.R. Hellstern, T.F. Jaramillo, et al. Submitted (2015)

MoS<sub>2</sub>-n<sup>+</sup>p Si data: J.D. Benck, S.C. Lee, K.D. Fong, J. Kibsgaard, R. Sinclair, & T.F. Jaramillo. *Advanced Energy Materials* 4 (18) 2014, 1400739 Pt-n<sup>+</sup>p Si data: S.W. Boettcher, et al. *JACS* (2011) 133, 1216. NiMo-n<sup>+</sup>p Si data: E.L. Warren, et al. *E&ES* (2012) 9653-9661.

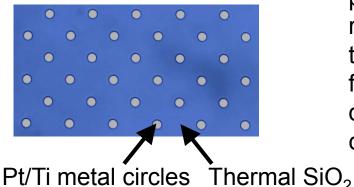
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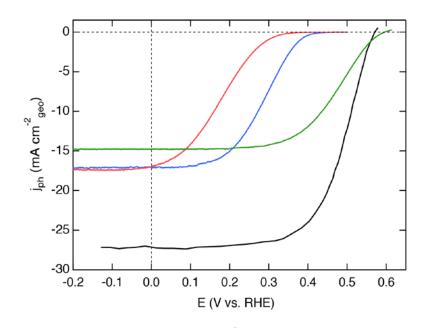
Stanford University

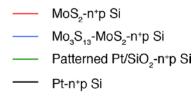
### Accomplishments Lithographically Patterned Si Photocathode



**Optical Microscope Image** 





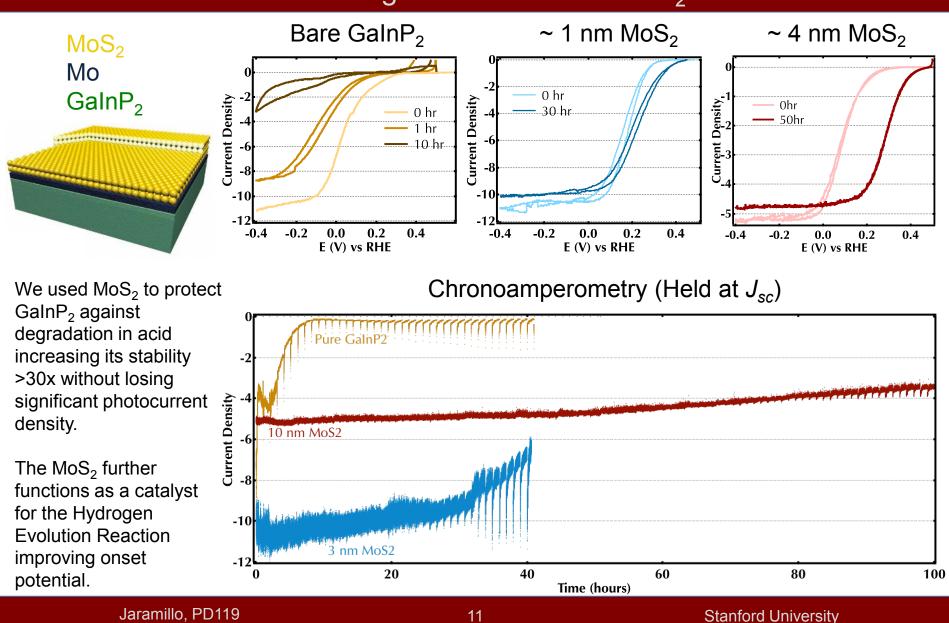


 $Mo_3S_{13}/MoS_2$ -n<sup>+</sup>p Si data: J.D. Benck, S.C. Lee, K.D. Fong, J. Kibsgaard, R. Sinclair, & T.F. Jaramillo. *Advanced Energy Materials* 4 (18) 2014, 1400739

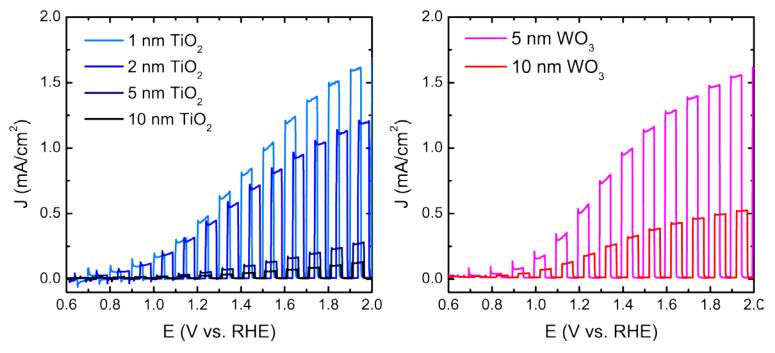
Pt-n<sup>+</sup>p Si data: S.W. Boettcher, et al. *JACS* (2011) 133, 1216.

We created a Si photocathode with lithographically patterned surface catalyst to reduce charge carrier recombination inside the semiconductor, improving the photocurrent onset potential. Future work will focus on increasing the saturation photocurrent density and incorporating nonprecious metal HER catalysts into this architecture.

#### Accomplishments Protecting the surface of GalnP<sub>2</sub>



# $\begin{array}{c} \textbf{Accomplishments} \\ \textbf{TiO}_2 \text{ and WO}_3 \text{ protection layers on a BiVO}_4 \text{ Photoanode} \end{array}$



Chopped-illuminated I-V curves of W-doped BiVO<sub>4</sub> photoanodes coated by wide band gap oxides with various thickness tested in H<sub>2</sub>SO<sub>4</sub> (pH 1)

- Atomic layer deposition (ALD) of Ultrathin TiO<sub>2</sub> and WO<sub>3</sub> protection layers were developed
- Comparing to ALD TiO2, WO<sub>3</sub> layer has lower resistivity. Hence, thicker protection layer to be utilized

## Collaborations

#### National Renewable Energy Laboratory

Todd Deutsch James Young

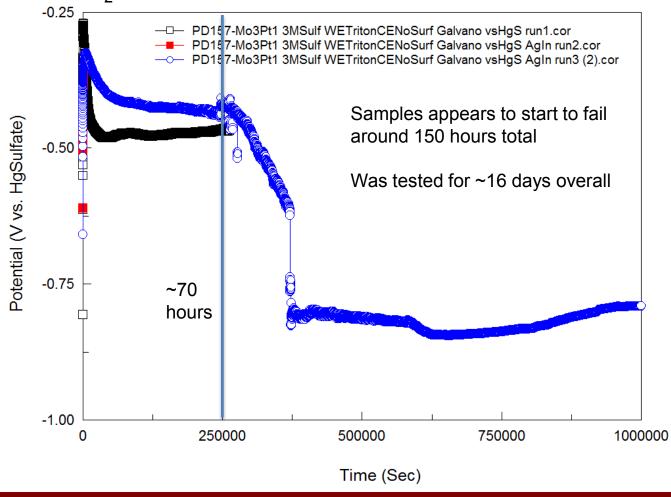
We collaborate together with Todd Deutsch and James Young on stabilizing GaInP<sub>2</sub> photocathodes. These III-V semiconductors are fabricated at NREL, while at Stanford we engineer catalysts and surface protection layers.

#### **PEC Working Group**

Collaboration among a community of PEC researchers to set standards for device testing and to discuss and address technical challenges in the field. Key organizers include Dr. Eric Miller (DOE), Dr. Heli Wang (NREL), and Prof. Thomas F. Jaramillo (Stanford).

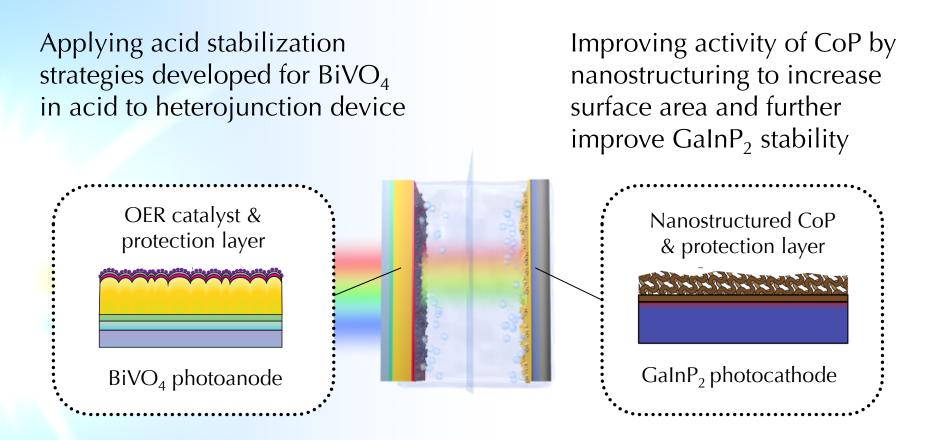
## **Collaborations - NREL**

Sister samples were sent to NREL for chronopentiometry experiments to assess stability, measured at NREL by Todd Deutsch and James Conway to further study the stability of GalnP<sub>2</sub>



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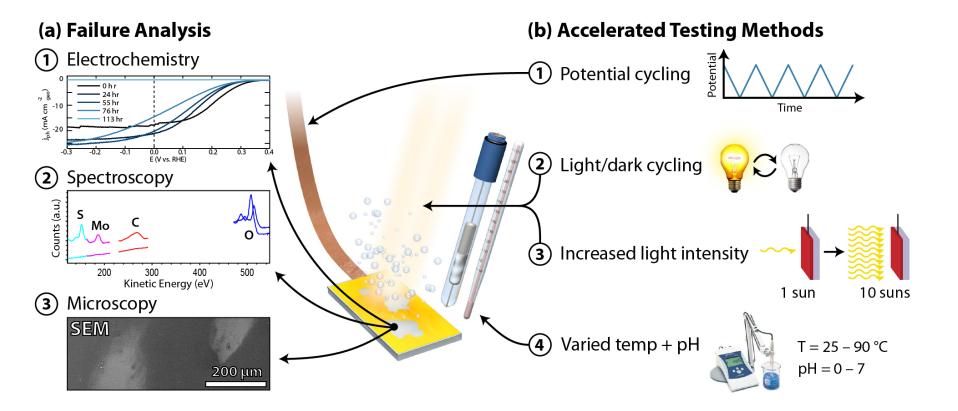
### Remaining Challenges & Future Work Immediate goals



Goal: To combine current and future innovations to make a non-precious metal unassisted water splitting device that is stable in acid.

### Remaining Challenges & Future Work Long term goals

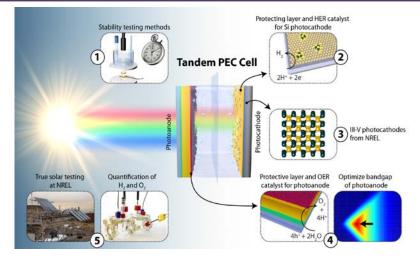
Developing methods to identify photoelectrode failure mechanisms and predict long-term performance using short-term tests



## Summary

#### Approach

• We developed a work plan for addressing stability and activity of both the photoanode and photocathode in acid.



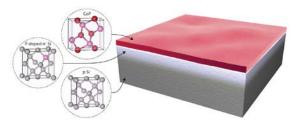
#### Accomplishments

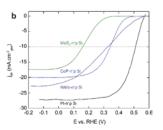
• Engineered lamp setup for stability testing.





• Developed a highly active and stable CoP HER catalyst + Si photocathode

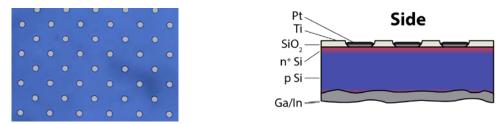




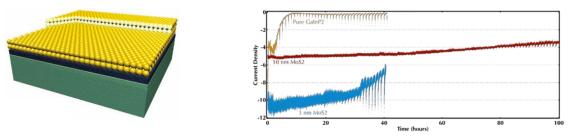
## Summary

#### Accomplishments cont.

• Interfacial engineering of the Si photocathode



• Improving stability of GaInP<sub>2</sub> over 30x using MoS<sub>2</sub> protection layers



• Using ALD of TiO<sub>2</sub> and WO<sub>3</sub> to protect BiVO<sub>4</sub> photoanodes

