



Sweet Hydrogen: High-Yield Production of Hydrogen from Biomass Sugars Catalyzed by *in vitro* Synthetic Biosystems

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Overview

Timeline

- Project Start Date: 06/15/2015
- Project End Date: 06/30/2017 (Two years)
- Project renewal and continuation will be determined by DOE EERE's Go/No-Go milestones

Barriers

Hydrogen production from Biomass

A. Reformer (Bioreactor) Capital Costs and Efficiency

B. Operations and Maintenance (O&M)

C. Biomass Feedstock Issues (starch and biomass sugars)

Budget

- Total Project Budget: \$937,602 (two years)
 - Total Recipient Share: \$187,602
 - Total Federal Share: \$750,000
 - Total DOE Funds Spent*: \$320,522

* As of 3/31/2016

Collaborators

- Virginia Tech (lead)
- University of Georgia

Relevance

Fuel Cell Technologies Office Objective -- Develop advanced biological generation technologies to produce hydrogen with a projected cost of \$10.00/gge at the plant gate by 2020.

Our novel approach – use renewable sugars (e.g., biomass sugars or starch) to split water to produce H_2 catalyzed by enzyme cocktails

 $C_6H_{10}O_5 + 7 H_2O = 12 H_2 + 6 CO_2$

- Low-carbon or even nearly carbon-neutral production in terms of entire life cycle
- High-purity hydrogen generated (no CO)
- Mild reaction conditions (1 atm, 20-90°C, pH 7.0, aqueous phase)
- Local resources for distributed hydrogen production
- Highest biological hydrogen generation rates

Targets	Units	June 2016 Target	June 2017 Target (estimated)	Year 2020 Target (Plant gate)
Production cost	\$/kg H ₂	1000	10	10.00 (year 2020)
Productivity	mmole H ₂ /L/h	300	750	2,000
Reactor volume	L of reactor	0.01	1	2,777*
				*200 kg H ₂ per day

Background

Production of H₂ from biomass sugars (overview)



- 1. Logan et al. 2007. PNAS 104:18871.
- 3. Schmidt. 2007. Agnew Chem. 46: 586

- 2. Schmidt. 2004. Science. 303:993
- 4. Dumesic et al. 2002. Nature 418:964

5. Zhang et al. PLoS One 2007, 2:e456; ChemSusChem 2009; Angew Chem 2013; Metab Eng 2014; PNAS 2015.

Background

Microbes versus in vitro enzymatic biosystems



(b) Enzymatic systems



- Closed natural biosystems
- Dominant bio-manufacturing platform
- Enhanced performance by synthetic biology and systems biology

- Open man-made biosystems
- Emerging low-cost bio-manufacturing platform
- Surpass mother nature of cells, such as, bioenergetics, mass transfer rate, etc.
 - High yield,
 - Fast reaction rate,
 - Easy product separation,
 - Broad reaction conditions

Background

Increasing volumetric rate (Rxn)

New Moore's law ? - Doubling speed every two years



- Highest Rxn of i.e., 153 mmole H₂/L/h
- 750-fold increase in past 10 years

Planned Solutions

- Better pathway design
- Better enzymes (complexes)
- High temperature
- Kinetic modeling
- More addition of ratelimiting enzymes
- High substrate levels
- More enzyme loadings

This project Rxn goals

- 300 mmole H₂/L/h @Jun 2016
- 750 mmole H₂/L/h @Jun 2017

Decreased H₂ production costs

Quantitative indicator of enzymes and coenzyme: **total turn-over number (TTN)**, mole product/mole enzyme



Preconditions for low-cost hydrogen production

- Low cost enzyme production (\$10-20/kg)
- \Box High stability of enzymes (TTN = 10^{8-9} mole product/mole enzyme)
- High stability of coenzymes (TTN = 10⁶⁻⁷ mole product/mole enzyme)
- Low-cost of coenzymes (\$100/kg)

Key directions: (1) discovery of better enzymes (Task 1.3), (2) engineering of dehydrogenases on biomimetic cofactors (Task 1.2), and (3) mass production of enzymes (Tasks 3.1 & 3.2).

Zhang, Biotechnol. Adv. 33, 1467–1483 (2015).

Approaches/Milestones

(*As of 3/31/2016)

FY 15 Q4	FY 16 Q1	FY 16 Q2	FY 16 Q3	FY 16 Q4	FY 17 Q1	FY 17 Q3	FY 17 Q3
Task 1.1. Co-expression of multiple enzymes in one host(FY16 Q1)100%Task 1.2. Two redox enzymes on biomimics at 0.1 U/mg(FY16 Q3)80%Task 1.3. Discovery of four novel enzymes(FY16 Q1)100%Task 1.3. Discovery of eight novel enzymes(FY16 Q3)100%							
Task 2.1. Optimization of enzyme ratios by modeling(FY16 Q1)100%Task 2.2. Data fitting and validating of rate-limiting steps(FY16 Q4)100%Task 2.3. Construction of artificial electron transport chains(FY16 Q2)100%Task 2.4. Construction of synthetic enzyme complexes(FY16 Q3)60%							
Task 3.1. Task 3.2. Task 3.3.	High-densi Mass produ 10-mL leve	ty of prote uction of h I demonsti	in express ydrogenas ration	sion in <i>E. c</i> a se (SH1) (L	oli JGa)	(FY16 Q3) (FY16 Q3) (FY16 Q3)	90% 80% 80%

This reporting period (June 2016, Go/No-Go milestones)
□ 10-fold volume scale-up (10 mL reactor) with 2-fold increase in H₂ peak production rate (i.e., 300 mmol H₂/L/h)

Accomplishments and Progress: Responses to Previous Year Reviewers' Comments

• This project started on June 15, 2015. No previous year reviewers' comments are available.

Accomplishments: Multi-enzyme expression in one host





We can control multiple (two-four) enzyme expression levels by novel strategies. High-level expression of more than four heterologous enzymes is technically difficult.

Accomplishments: Discovery of better enzymes

Desired enzyme characteristics: expected reaction, thermostability, high activity, easy expression and purification in *E. coli*

Technical approaches:

- Data mining thermostable enzyme candidates from megagenomic DNA databases (KEGG, NCBI, JGI)
- Compare available enzyme characteristics from Brenda database and literature
- Clone, express and characterize enzyme candidates

Enzyme	Abb.	EC	Original enzyme Source	New Enzymes Discovered
a-glucan phosphorylase	aGP	2.4.1.1	C. thermocellum	T. maritima (Q1)
4-glucanotransferase	4GT	2.4.1.25		Thermococcus litoralis (Q2)
Phosphoglucomutase	PGM	5.4.2.2	C. thermocellum	Thermococcus kodakarensis (Q2)
Glucose-6-phosphate dehydrogenase	G6PDH	1.1.1.49	G. stearothermophilus	T. maritima (Q1)
6-phosphoglucolactonase	6PGL	3.1.1.31		T. maritima (Q2)
6-phosphoglucanate dehydrogenase	6PGDH	1.1.1.44	M. thermoacetica	T. maritima Q1)
Aldolase	ALD	4.1.2.13	T. thermophilus	Sulfolobus tokodaii STK_03180 (Q2)
Fructose-1,6-bisphosphatase	FBP	3.1.3.11	T. maritima	Sulfolobus tokodaii STK_03180 (Q2)
Phosphoglucose isomerase	PGI	5.3.1.9	C. thermocellum	Thermus thermophilus (Q1)
NAD(P)H: rubredoxin oxidoreductase	NROR	1.18.1.4	P. furiosus	<i>P. furiosus</i> (PF1197) used for electron transfer from NADPH to EM (Q2)

We have obtained nine new thermostable enzymes, all of whose activities are higher than previous ones. They can be purified by easy heat precipitation.

Progress: Replace NAD(P) of dehydrogenases with NR

Wild-type dehydrogenases (i.e. glucose 6-phosphate dehydrogenase, **G6PDH**, and 6-phosphogluconate dehydrogenase, **6PGDH**, cannot work on biomimetic coenzymes.



Comparison of coenzymes

Coenzyme engineering Rossmann Catalytic module A B C MAD-binding MAD-binding Rational design Directed evolution Module swap I Output of the system I Output of the syste

\$4,500/kg

Nicotinamide riboside (NR) is better than NAD(P)

- Less costly
- More stable
- Small size better mass transfer

- We have three strategies to change coenzyme preference of dehydrogenases
- New area very high risk.
- Directed evolution of enzymes is preferred.

Progress: Novel high-throughput screening for coenzyme engineering



- We established high-throughput screening for dehydrogenases working on biomimetic coenzyme with minimum influences from inherent coenzymes (NAD(P)).
- Engineered glucose 6-phosphate dehydrogenase (G6PDH) and 6-phosphogluconate dehydrogenase (6PGDH) can work on NR for the first time.

Accomplishments: Identification of key rate-limiting steps by modeling





#1. Hydrogenase (SH1) is No. 1 ratelimiting step.

#2. NADPH regeneration from NADP⁺ is No. 2 ratelimiting step catalyzed by G6PDH.

Green line, experimental data Red line, model data after data fitting by using a genetic algorithm Blue line, model robustness (leave one out) cross-validation.



#3. ALD for FDP synthesis from GA3P and DHAP is the no. 3 ratelimiting step.

Planned solutions

#1, to be addressed by artificial electron transport chains (Task 2.3), biomimetic coenzyme (Task 1.2); enzyme complexes (Task 2.4) #2, to be addressed by redesigned pathway, better enzymes & biomimic (Task 1.2) #3, to be addressed by better enzymes (Task 1.3)

Accomplishments: Increasing Rxn by using artificial electron transport chains

Previous pathway design



6-Phosphogluconolactone

Big concerns

to H₂

NADPH (-320 mV)

New artificial electron transport chain



 Thermodynamics unfavorable (from -320 to -420 mV)

Very slow reaction rate from NADPH

• Very high activation energy barrier

Key idea: the use of an electron mediator bridging NAD(P)H and H2, decreasing activation energy and increase Rxn.

Accomplishments: Increasing Rxn by using the artificial electron transport chain



- We achieve the highest biohydrogen generation rates (i.e., 310 mmole H₂/L/h, 0.62 g H₂/L/h or 6.6 L H₂/L/h) by using BV and NROR, far higher than anaerobic microbial fermentation, electrochemically-assisted microbial fuel cells, algal hydrogen production.
- Meet the rate target of the Go/No-Go milestones of Phase I project -- Increase H₂ production rate by 2-fold to <u>300 mmole H₂/L/h</u>.
- Use of another new enzyme 6-phoghoglucolactonase (6PGL) accelerates the reaction rates of oxidative phase of pentose phosphate pathway.

Progress: Complete starch utilization for H₂ generation

Η,

Hydrogen

H₂ase

H₂ase

New pathway for better substrate (starch) utilization

- Use isoamylase to debranch branched starch (amylopectin, dominant corn starch component) (increasing H_2 yield by 30%)
- Use 4-glucanotransfer (4GT) to utilize maltose, hydrolytic product of starch (increase H₂ yield about 4%)
- Polyphosphate glucokinase (PPGK) to utilize glucose (increasing yield about 4%)

2016.



Progress: Complete starch utilization for H₂ generation

Amylopectin
$$\rightarrow$$
 amylodextrin $(C_6H_{10}O_5)_n$ isoamylase $(C_6H_{10}O_5)_n + P_i \rightarrow G-1-P + (C_6H_{10}O_5)_{n-1}$ alpha-glucan phosphorylase $G-1-P \rightarrow G-6-P$ phosphoglucomutase $G-6-P + 12 \text{ NADP}^+ \rightarrow 12 \text{ NADPH} + 12 \text{ H}^+$
 $+ 7 \text{ H}_2\text{ O}$ Pentose phosphate pathway,
glycolysis & glycogenesis $12 \text{ NADPH} + 12 \text{ BV}_{ox} \rightarrow 12 \text{ NADP}^+ + 12 \text{ BV}_{red}$ NROR $12 \text{ BV}_{red} + 12 \text{ H}^+ \rightarrow 12 \text{ Bv}_{ox} + 12 \text{ H}_2$ Hydrogenase (SH1) $_6H_{10}O_5 + 7H_2O = 12 \text{ H}_2 + 6 \text{ CO}_2$ (n-1) anhydroglucose units of starch

 $C_6H_{12}O_6 + 6H_2O + (P_i)_n = 12 H_2 + 6 CO_2 + (P_i)_{n-1}$ 1 glucose of starch

We will utilize all energy stored in starch to generate theoretical yield H₂. (Note 1: We have converted all glucose and xylose from biomass to H₂ before this project.) (Note 2: We have utilized all glucose units of starch for the production of another product.)

Progress: Scale-up of reactor volume

VT glass shop-made bioreactors



0.5-4 mL 5-20 mL

200-

200-1000 mL

Now we conduct most experiments in the small bioreactor (1-2 mL working volume) for optimization of reaction condition, new pathway design, etc.

10X scale-up. We will scale up enzymatic hydrogen production in 10-mL working volume bioreactor by the end of June 2016 (the end of Phase I). (Now we are scaling up enzyme production capacity by 10 (large flasks) -1000 fold (e.g., 6-L fermenter.))

1000X scale-up. We will scale up enzymatic hydrogen production in 1000mL working volume bioreactor by the end of June 2017 (the end of Phase II).

Accomplishments: Scale-up of high density E. coli fermentation

1000 X scale-up



200 mL/flask OD = 3.0





4000 mL/reactor OD = 150

Recombinant enzyme production cost analysis

- Cell density (OD = 150, 50 g DCW/L)
- Protein expression level = 20-30% of cellular protein
- Fermentation costs = \$200-300/m³ in 20-50 m³ fermenters (including low-cost industrial medium and OpEx) from a few Chinese <u>contract research</u> <u>outsourcing (CRO)</u> enzyme companies and Korean companies.

Enzyme production cost = \$40-50/kg enzyme (enzyme order scale of 10 tonnes/yr)

Enzyme production cost = \$400-500/kg (enzyme order scale of 10-100 kg/yr)



Accomplishments: Over-expression of Soluble Hydrogenase I (SHI)

NADPH + $2H^+ \rightarrow H_2$ + NADP





(155 kDa, 1 Ni, [6 FeS] clusters, FAD)

P _{SLP} promoter: constitutive, high level expression; Affinity tagged Strep- or His ₉ -PF0891					
	Host	Affinity tag	Purification Steps	SHI yield (mg/10 g cells)	
Native	P. furiosus	-	4	0.25	
Overexpressed	P. furiosus	Strep tag II	1	1.7	
Overexpressed	P. furiosus	His tag	1	13.5	

We increased SH1 expression level by 54 fold, making 340 mg SH1 per batch. Problem: >30% of the SHI protein lacks the PF0894 catalytic subunit (E3).



Progress: Identification of limiting factors for SH1 expression



Progress: Over-expression of SHI accessory proteins



- Over-expression of *frxA* increased almost 200-fold in strain MW519 & three-fold higher H₂ evolution activity.
- Over-production of HypC/D in the FrxA over-expression strain did not further increase SHI production

Over expression of HypF in OE-FrxA and OE-FrxA/HydC/D strains is underway



Collaboration



- Focus on low-cost mass production of hyperthermophilic Fe-Ni soluble hydrogenase I (*P. furiosus* SH1) without a discount of specific activity of SH1
- Related to Objective 1 (decrease hydrogen production costs for enzymes) and Objective 3 (scale-up of enzymatic hydrogen production)

The University of Georgia

Remaining Challenges and Barriers

- Increase catalytic efficiencies (k_{cat}/k_M) dehydrogenases on biomimic (NR) to comparative levels of to those on their natural coenzyme (NADP).
- Decrease SH1 production costs better expression levels without a decrease in specific activity (i.e., coordinated co-expression of enzyme components).
- Construct G6PDH-6PGDH-SH1 enzyme complexes featuring electron transport channeling.
- Scale up recombinant enzyme production from gram scales at labs to kg or ton scale in pilot plants and demonstration plants

Proposed future work

- Decrease H₂ production costs by 1000 fold to \$10/kg, where production cost will be estimated by using H2A model
 Task 1.2. Replace costly NADP by biomimic -- NR
 Task 1.4. Detailed economic analysis of H2 production
- Increase H₂ production rate from 300 to 750 mmole H₂/L/h Task 2.3. Construction of artificial electron transport chains Task 2.4. Construction of synthetic metabolons (enzyme complexes)
- Scale up H₂ reaction volume from 10 mL to 1000 mL Task 3.1. High-density of recombinant protein expression in *E. coli* Task 3.2. Mass production of hydrogenase (SH1) (UGa) Task 3.3. Liter level demonstration

Technology Transfer Activities

- Before this EERE award, Virginia Tech and Oak Ridge National laboratory received issued patent US 8,211,681. Biohydrogen production by an artificial enzymatic pathway. (2012).
- PI. Prof. Dr. Percival Zhang, President, Cofounder, and CSO of a startup company, Cell Free Bioinnovations (CFB). CFB optioned to license this hydrogen technology from Virginia Tech but this company is focusing on high-end applications – sugar-powered biobatteries (direct sugar fuel cells). Both hydrogen and bio-batteies projects share most enzymes used.
- CFB is seeking for capital investment and SBIR/STTR grants.
- Several provisional patent disclosures will be filed about engineered dehydrogenases, biomimetic coenzymes, artificial electron transport chains, artificial enzyme complexes.

Summary

- We achieved the highest biological hydrogen generation rate (310 mmole H₂/L/h, 0.62 g H₂/L/h) by using sugar energy to split water, reaching the rate target of the Go/No-Go milestone (June 2016).
- We demonstrated the feasibility of changing coenzyme preference of engineered dehydrogenases from NADP to NAD and biomimic.
- We can produce 12 H_2 each sugar unit of starch regardless of linear or branched starch.
- We scaled up recombinant *E. coli* enzyme production by 1000-fold and recombinant *P. furiosus* SH1 (hydrogenase) production by 50-fold.

Targets	Units	June 2016 Target	June 2017 Target (estimated)	Year 2020 Target (Plant gate)
Production cost	\$/kg H ₂	1000	10	10 (year 2020)
Productivity	mmole H ₂ /L/h	310 (achieved)	750	2,000
Reactor volume	L of reactor	0.01 (to be finished)	1	2,777*
				*200 kg H ₂ per day

Technical Back-Up Slides

Appraisal of enzymes as biocatalysts

Basic facts

- Most enzymes are proteins
- Biocatalysis catalyzed by enzymes has highly chemical selectivity (no side product)
- Most enzymes work at mild reaction conditions (low temperature, 1 atm, neutral pH, and aqueous phase)
- Enzymes do not require costly precise metals

Conflicting concepts (academic researchers versus industrial enzyme experts)

- × For academic researchers, enzymes are VERY costly (e.g., **billion dollars per kg**)
- $\sqrt{}$ For industrial enzyme experts, bulk enzymes are less costly (e.g., **10 dollars per kg**)
- × For academic researchers, enzymes are very labile, losing activities within hours or days
- ✓ For industrial enzyme experts, some enzymes (e.g., immobilized or engineered) are very stable, lasting months and years (e.g., glucose isomerase for HFCS production, protease in detergent, glucose dehydrogenase in blood sugar test strips).

 \times For academic researchers, enzymes have narrow reaction conditions in terms of pH, temperature and solutions.

 $\sqrt{100}$ For industrial enzyme experts, some enzymes (engineered or discovered from extremophiles) can work on a large temperature range from 0 – 100°C, pH from 1 to 14, from aqueous solution to 100% organic solvent

Review of enzyme production costs & approaches



Enzyme costs are NOT showstopper for economically viable production of high-volume products (> 1 million tons per year) via in vitro synthetic enzymatic biosystems.

Enzyme stability

EC	Enzyme name	yme name Source	Form	Cond.	TTN
1.1.1.44	6-phosphogluconate hydrogenase	hosphogluconate <i>T. maritima</i> Irogenase	Free	80°C	2.4 x10 ⁸
2.2.1.2	Transaldolase	nsaldolase <i>T. maritima</i>	Free	60°C	1.7 x 10 ⁷
3.1.3.11	Fructose 1,6- Bisphosphatase	ctose 1,6- <i>T. maritima</i> ohosphatase	Free	60°C	2 x 10 ⁷
5.4.2.2	Phosphogluomutase	sphogluomutase C. thermocellum	Free	60°C	7.1 x 10 ⁷
5.3.1.5	Xylose (glucose) isomerase	ose (glucose) nerase	Immobil ized	50-60°C	5.0 x 10 ⁸
5.3.1.6	Ribose-5-phosphate isomerase	ose-5-phosphate <i>T. maritima</i> nerase	Free	60-70°C	2.2 x 10 ⁸
5.3.1.9	Phosphoglucose isomerase	nerase <i>C. thermocellum</i>	Free Immob.	60°C	3.2 x 10 ⁷ 1.1 x 10 ⁹
	Our goal:				
	1 kg of enzyme produces	H ₂ 32			

Enzymatic H₂ costs

Cost goal = $\frac{2.00}{\text{kg H}_2}$



Expected enzyme expenditure = \$0.10/kg H₂ Expected coenzyme = \$0.025/kg H₂ 1 kg enzyme produces 300 kg of H₂.

All estimates were conducted based on key inputs of Slide 7 – enzyme costs & TTN by the PI.

Progress: H₂ production cost analysis

Distributed H₂ Production



All estimates were conducted by Cell Free Bioinnovations and VT business school but will be redone by using H2A model.

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Key inputs:
CapEX= $6 MM
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Other inputs \$1.35/kg of hydrogen for sugar

(\$0.18/kg sugar) \$0.10/Kg of hydrogen for enzyme \$0.03/Kg of hydrogen for coenzyme

Ultimate cost

= \$1.77/kg H2
@8000 ton H2 per year
= \$2.00/kg
@200 kg H2 per day