

H₂ Storage Characterization and Optimization Research Effort

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National Renewable Energy Laboratory & National Institute of Standards and Technology

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Overview

Timeline*

Start: October 2015

End: TBD

% complete FY 16: ~65%

*previously a component of NREL's materials development program and supported annually since 2006

Budget

Funding FY16: \$250K*

* cost for scientists in residence at NIST.

Barriers addressed

Reversible Solid-State Material:

- M. Hydrogen Capacity and Reversibility;
- N. Understanding of Hydrogen Physi- and Chemi- sorption;
- O. Test Protocols and Evaluation Facilities.

Collaborators

PNNL - Tom Autrey, Mark Bowden,
LBL - Jeff Long, Martin Head-Gordon
HyMARC – LLNL, SNL

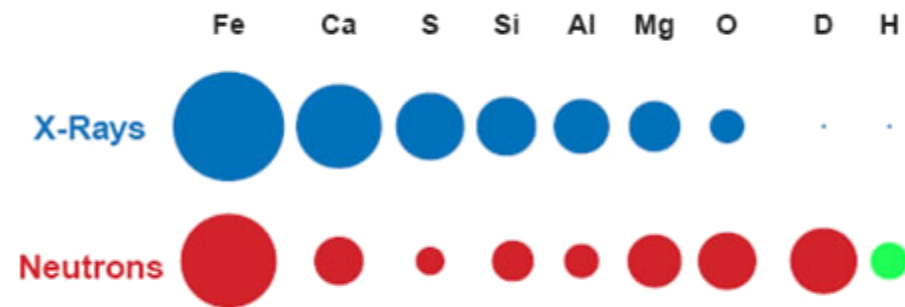
Relevance: Project Objectives

- **NREL leads a collaborative research effort involving NIST, LBNL and PNNL**
 - Seek to employ and develop characterization capabilities at each facility to understand and enhance hydrogen storage media
 - Leverage each institute's unique strengths to jointly validate hydrogen storage claims and design strategies

Relevance: Impact

- **Neutrons provide unique specificity towards determination of hydrogen properties**
 - Enables identification of isotopically-labelled hydrogen location within complex structures
 - Enables identification of hydrogen dynamics within complex structures

Relative comparison of scattering strength for x-rays and neutrons

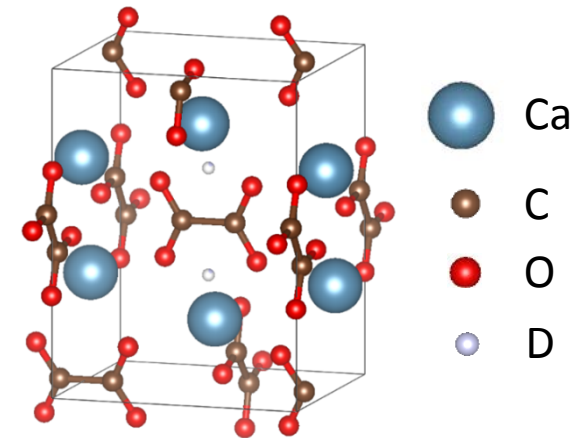
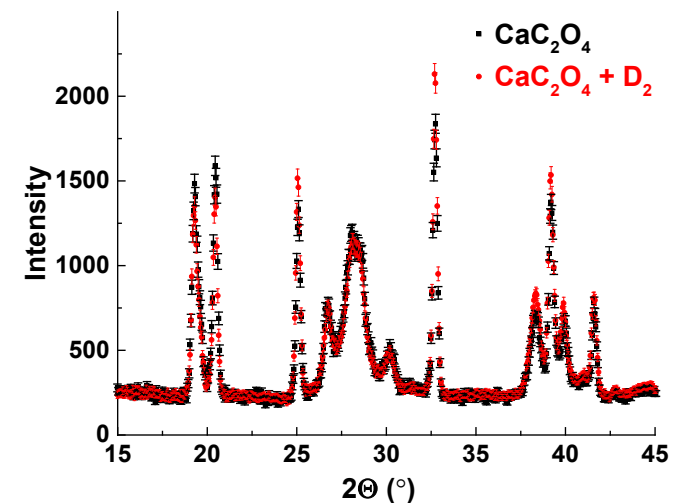


Approach: Neutron Scattering

- **(FY16) Utilize neutrons to characterize and validate hydrogen storage media**
 - Neutron powder diffraction with precise D₂ loading at T > 4 K and P < 100 bar
 - Elucidate crystal structure of storage materials
 - Harness isotopic sensitivity of elastically scattered neutrons to locate chemi- and physisorption sites of deuterium
 - Inelastic neutron spectroscopy with precise H₂ loading at T > 4 K and P < 100 bar
 - Harness isotopic sensitivity of inelastically scattered neutrons to identify local environment for complex hydrides and chemi- and physisorbed hydrogen

Accomplishments and Progress: Oxocarbons

- **Structure of bare and deuterium-intercalated ultramicroporous calcium oxalate (CaC_2O_4) probed using neutron powder diffraction**
 - Strong correlation identified between D_2 loading pressure and storage capacity
 - Adsorption requires dosing at $T > 175$ K and $P > 250$ torr
 - Changes in diffraction pattern upon intercalation consistent with D_2 adsorption between Ca^{2+} and O^{2-} (more work is required to fully understand the data)

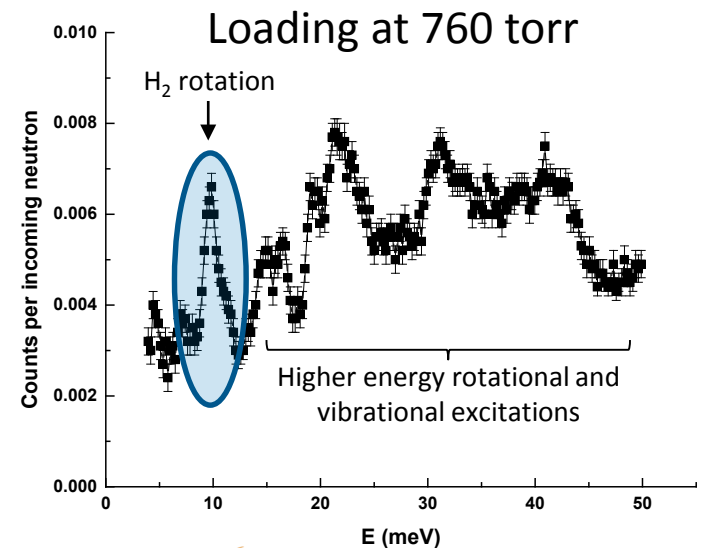
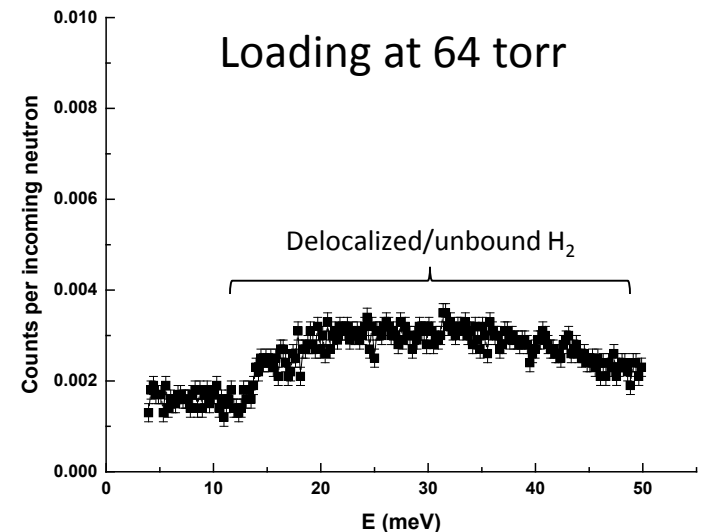


Neutrons locate adsorbed D_2 in ultra-micropores

Accomplishments and Progress: Oxocarbons

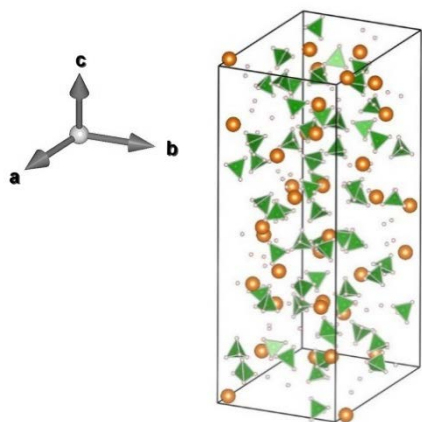
- **Local dynamics of H₂ adsorbed in CaC₂O₄ probed using inelastic neutron spectroscopy (INS)**
 - Loading at low pressure (<250 torr) leads to weakly bound H₂
 - Loading at atmospheric pressure (760 torr) leads to discrete localized adsorption sites
 - Peak near 10 meV consistent with transition from J=0 to to a sub-level of the J=1 rotational state of H₂
 - More detailed loading dependence and isotopic substitution should help assign remaining peaks to rotation or rotation+phonon coupled peaks

INS indicates adsorption of H₂ in ultra-micropores

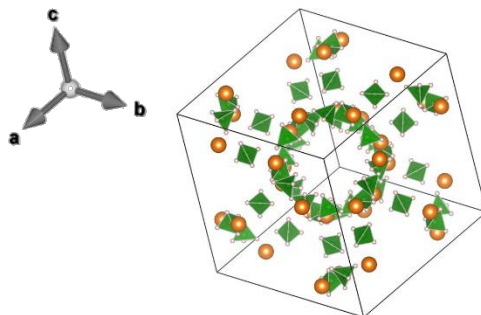


Accomplishments and Progress: Borohydrides

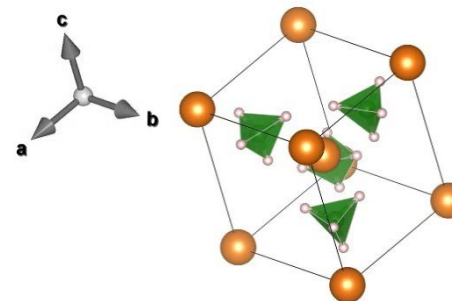
- Structural and phase dynamics in $\text{Mg}(\text{BH}_4)_2$ probed with inelastic neutron spectroscopy
 - Three potential polymorphs identified



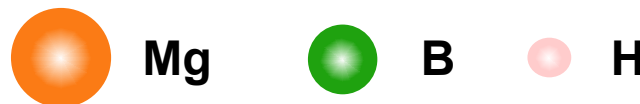
α - $\text{Mg}(\text{BH}_4)_2$
P6₁22



γ - $\text{Mg}(\text{BH}_4)_2$
Ia-3d



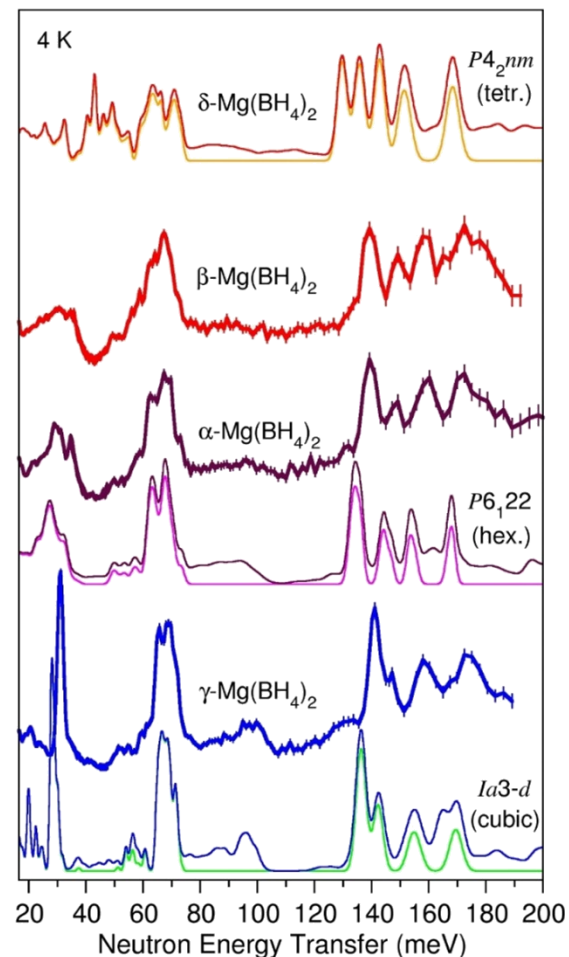
δ - $\text{Mg}(\text{BH}_4)_2$
P4₂nm



Motivation: INS can be used as a fingerprint in non-crystalline environments

- **Neutron vibrational spectra of different $\text{Mg}(\text{BH}_4)_2$ polymorphs**
 - Measured at 4 K
 - Agreement established with one-phonon and one+two-phonon densities of states simulated from first-principles phonon calculations of the DFT-optimized structures

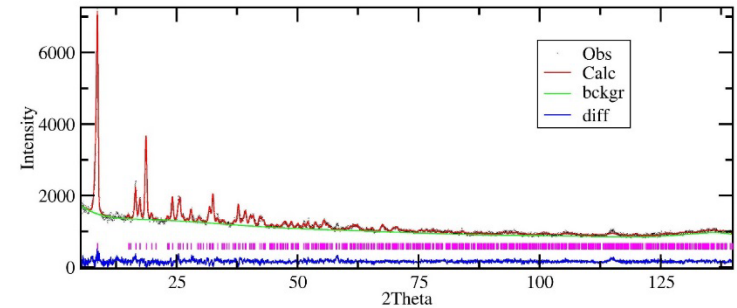
Most intense peaks are due to H-motion: clear signature of different polymorphs



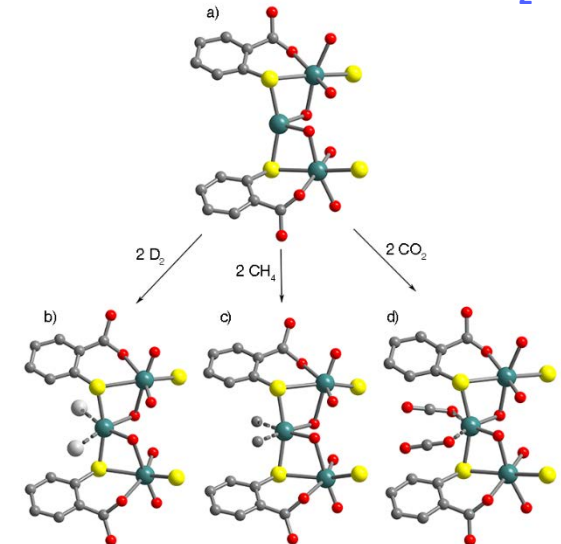
Accomplishments : MOFs (Metal Organic Framework)

- Identify multiple H₂ binding at a metal site in a MOF
 - Mn₂(dsbdc)
 - Successful desolvation and activation
 - X-ray and Neutron Diffraction
 - D₂ adsorption sites determined.

D₂ and other gases have similar binding characteristics. Clear but weak interactions.



Rietveld refinement of neutron diffraction data for 0.75 D₂ : Mn

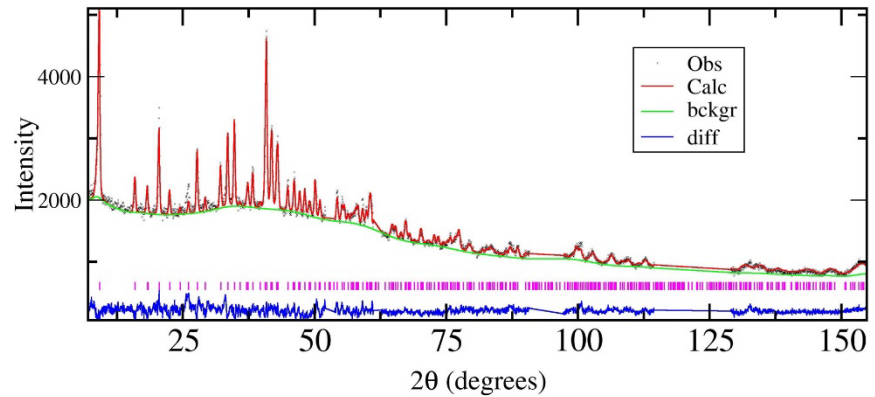


Mn can bind multiple gas molecules on one metal site

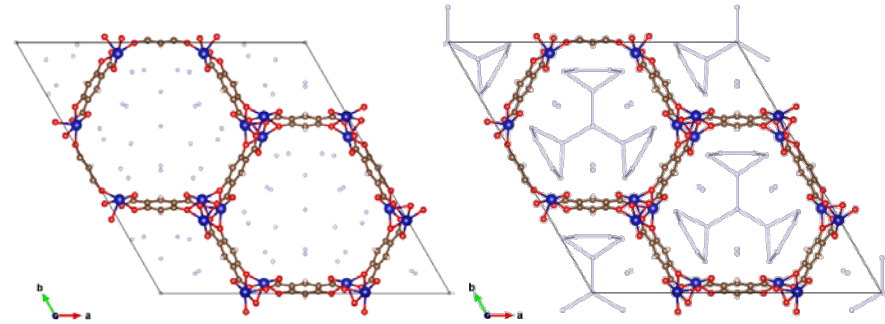
Accomplishments and Progress: MOFs

- **Volumetric capacities at pressure in MOFs**
 - Moderately high pressure neutron diffraction
 - Determine loadings and adsorption distributions in $\text{Co}_2(\text{m-dobdc})$ and $\text{Ni}_2(\text{m-dobdc})$
 - Refines to 3.5 $\text{D}_2:\text{Co}$

Determine maximum crystallographic volumetric capacities independently from manometry



Rietveld refinement of neutron diffraction data for $\text{Co}_2(\text{m-dobdc})$ at 78 bar D_2 pressure

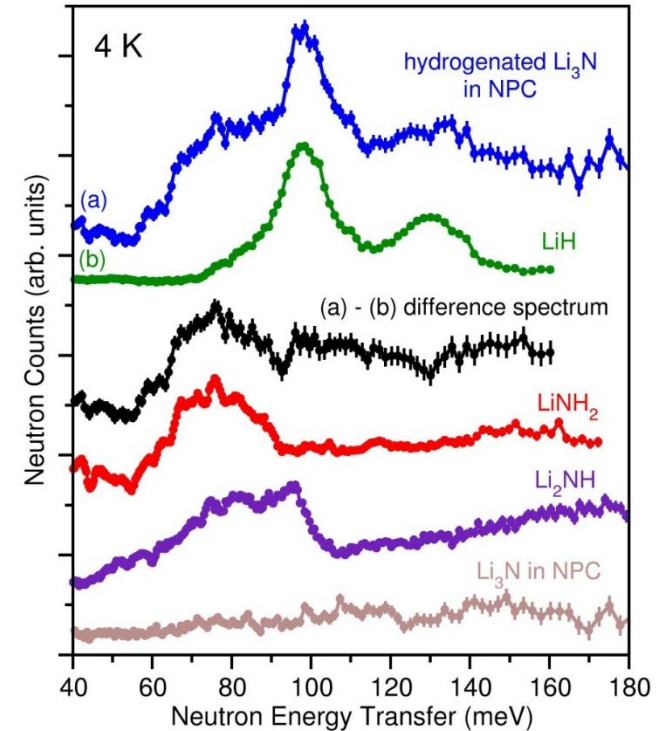


Shorter than bulk $\text{D}_2\text{-D}_2$ distances

Accomplishments and Progress: Confined phases

- **Determining products of hydrogenated Li_3N in nanoporous carbon**
 - Neutron vibrational spectra associated with hydrogenated Li_3N confined in nanoporous carbon (NPC) after 5 desorption/absorption cycles compared to reference spectra indicate that:
 - Relatively little H exists in the dehydrogenated spectrum (Li_3N in NPC)
 - Both LiNH_2 and LiH are present in the hydrogenated spectrum
 - Nanoconfinement fundamentally alters both the hydrogenation and dehydrogenation reaction pathways as a direct consequence of solid-solid nanointerfaces

INS used to identify hydrogen-containing phases that cannot be probed easily by other means



Neutron vibrational spectroscopy confirms that both LiNH_2 and LiH are hydrogenation products from carbon-nanoconfined Li_3N , with no obvious presence of Li_2NH .

Collaborations

- **NREL/NIST collaboration**
 - Characterizing ultra-microporous materials using neutron diffraction and neutron spectroscopy
- **NREL/NIST collaboration with LBNL**
 - Characterizing hydrogen adsorption in metal organic framework materials using neutron diffraction and neutron spectroscopy
 - Characterizing various hydrogen storage materials at the Advanced Photon Source
- **NREL/NIST collaboration with SNL**
 - Developing spectroscopic signatures for $\text{Mg}(\text{BH}_4)_2$
- **NREL/NIST collaboration with LLNL and SNL**
 - Identifying phases for complexes in pores
 - Determining products of hydrogenated Li_3N in nanoporous carbon

Proposed Future Work

- **Finish work on current projects**
 - Identify precise binding sites for hydrogen adsorbed in ultra-microporous oxocarbons. Establish influence of metal cation on binding distance and hydrogen capacity
 - Complete pressure dependence of volumetric capacities for MOFs
- **Continue to advance the use of neutron scattering to validate materials and concepts**
 - As determined through collaborations and discussion with EERE and HyMarc

Critical Assumptions and Issues

- **Refinement of powder diffraction patterns for oxocarbons complicated by presence of additional phases**
 - Need to determine whether additional phases stem from contaminants or represent polyphases of oxocarbon

Critical Assumptions and Issues

- **Oxocarbons (specifically, CaC_2O_4) exhibit unusual behavior upon thermal cycling**
 - In-situ x-ray diffraction at APS planned to measure structural changes upon activation, thermal cycling, and hydrogen dosing of oxocarbons

Publications

- Outlook and challenges from hydrogen storage in nanoporous materials. D. P Broom, C. J. Webb, **K. E. Hurst**, **P. A. Parilla**, **T. Gennett**, **C. M. Brown**, R. Zacharia, E. Tylanakis, E. Klontzas, G.E. Froudakis, Th. A. Steriotis, P. N. Trikalitis, D. L. Anton, B. Hardy, D. Tamburello, C. Corngale, B. A. van Hassel, D. Cossement, R. Chahine, M. Hirscher. *Appl. Phys. A*. 122; 151, 2016.
- Hydrogen Storage in the Expanded Pore Metal-Organic Frameworks $M_2(\text{dobpdc})$ (M = Mg, Mn, Fe, Co, Ni, Zn). Gygi, D.; Bloch, E. D.; Mason, J. A.; Hudson, M. R.; Gonzalez, M. I.; Siegelman, R. L.; Darwish, T. A.; Queen, W. L.; **Brown, C. M.**; **Long, J. R.** *Chem. Mater.* 2016, 28, 1128-1138.
- Dynamics of Pyramidal SiH_3^- Ions in ASiH_3 (A = K and Rb) Investigated with Quasielastic Neutron Scattering. C. Österberg, H. Fahlquist, U. Häussermann, **C. M. Brown**, **T. J. Udovic**, M. Karlsson, *J. Phys. Chem. C* 120, 6369, 2016.
- Hydrogen Storage and Selective, Reversible O_2 Adsorption in a Metal-Organic Framework with Open Chromium(II) Sites. Bloch, E. D.; Queen, W. L.; Hudson, M. R.; Mason, J. A.; Xiao, D. J.; Murray, L. J.; Flacau, R.; **Brown, C. M.**; **Long, J. R.**, *Angew. Chem.* Accepted.
- Adsorption of Two Gas Molecules at a Single Metal Site in a Metal-Organic Framework. Runčevski, T.; Kapelewski, M. T.; Torres-Gavosto, R. M.; **Tarver, J. D.**; **Brown, C. M.**; **Long, J. R.**, submitted.
- Nanointerface-driven Reversible Hydrogen Storage in the Nanoconfined Li-N-H System, **B.C. Wood**, V. Stavila, N. Poonyayant, T. W. Heo, K. G. Ray, L. E. Klebanoff, **T. J. Udovic**, J.R. I. Lee, N. Angboonpong, P. Pakawatpanurut submitted.

Key: **NREL**, **NIST**, **LBL**, **SNL**



Technical Slides: Sample preparation

- Samples activated by slowly heating in tube furnace under dynamic vacuum ($\sim 10^{-7}$ torr)
- Samples transferred to He-filled dry box with < 0.1 ppm H_2O and < 5 ppm O_2 and sealed to copper blocks using an indium o-ring
- Stainless steel tubing connected to copper blocks permits delivery of D_2/H_2 to sealed sample
- Samples mounted onto bottom-loading closed cycle refrigerator and connected to gas delivery manifold of known volume
- Residual He removed using turbo molecular pump

Technical Slides: Neutron Powder Diffraction

- **High resolution neutron powder diffraction data collected at 7 K using a Ge(311) monochromator with an in-pile 60' collimator**
 - Instrument: BT1
 - Corresponds to $\lambda = 2.078 \text{ \AA}$
- **Initial measurements collected on evacuated bare sample**
- **Volumetric dosing of D₂ performed above 20 K with incrementally increasing amounts**
 - Full adsorption ensured by letting pressure fall to zero
- **Sample + adsorbed D₂ cooled to 7 K for measurement**

Technical Slides: Inelastic Neutron Spectroscopy

- **Inelastic neutron spectra collected using the Filter Analyzer Neutron Spectrometer (FANS)**
- **Residual He removed using turbo molecular pump**
- **Spectra collected at 7 K using a pyrolytic graphite (002) monochromator with 20'-20' collimation**
 - Corresponds to energy range of 4.1 – 49.6 meV, resolution of 1.1 meV
- **Volumetric dosing of H₂ performed above 20 K with incrementally increasing amounts**
 - Full adsorption ensured by letting pressure fall to zero
- **Sample + adsorbed H₂ cooled to 7 K for measurement**