

# Tunable Photoanode-Photocathode-Catalyst Interface Systems for Efficient Solar Water Splitting

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AMR 2017

NSF-CBET/DOE-EERE joint  
Project ID# PD121

## Timeline

- Project Start Date: Sept 1, 2014
- Project End Date: Aug 31, 2017
- Percent complete: 89%

## Budget

- Total Project Budget: \$749,996
  - Total Recipient Share: NSF disallows
  - Total Federal Share: \$749,996
  - Total DOE Funds Spent: \$667,496 as of 4/10/17

## Barriers addressed

- AE. Materials Efficiency – Bulk and Interface
- AF. Materials Durability – Bulk and Interface
- AI. Auxiliary Materials
- AJ. Synthesis and Manufacturing

## Unfunded Collaborators

### Current:

- Andrew Rappe (Upenn)
  - Ni-P HER theory

### EMN (pending):

- Andriy Zakutayev (NREL)
  - $\text{ZnSnN}_2$  photoabsorber
- Todd Deutsch (NREL)
  - High flux/diurnal analysis
- Daniel Friedman (NREL)
  - $\text{GaInP}_2/\text{GaAs}$  photoabsorber

## Objectives:

**Long-term:** Effectively leverage the knowledge base of materials chemistry developed in the prior years to fabricate semiconductor/catalyst interfaces, both at the photoanode and photocathode, that utilize abundant elements and low cost fabrication methods. Combine these individual cells in tandem forming a monolithic device that attains or exceeds the DOE benchmark STH efficiency (~10%).

### Specific to Current Year (2016-2017):

- Develop a novel passivation layer (TiN) that prevents corrosion of the low bandgap photoabsorber (Si).
- Fabricate a photocathode consisting of a low cost photoabsorber (Si), HER catalyst (nickel phosphide, here NiP<sub>2</sub>), and a passivation layer (TiN). Test performance and stability under acidic and basic conditions.
- Investigate wider bandgap semiconductors as complementary tandem photoabsorbers in the photoanode, specifically perovskite oxynitride (SrNb(O/N)<sub>3</sub>) tantalum nitride (Ta<sub>3</sub>N<sub>5</sub>), and halide double perovskite (Cs<sub>2</sub>AgBiBr<sub>6</sub>)

## Relevance to DOE H<sub>2</sub> & FC Program

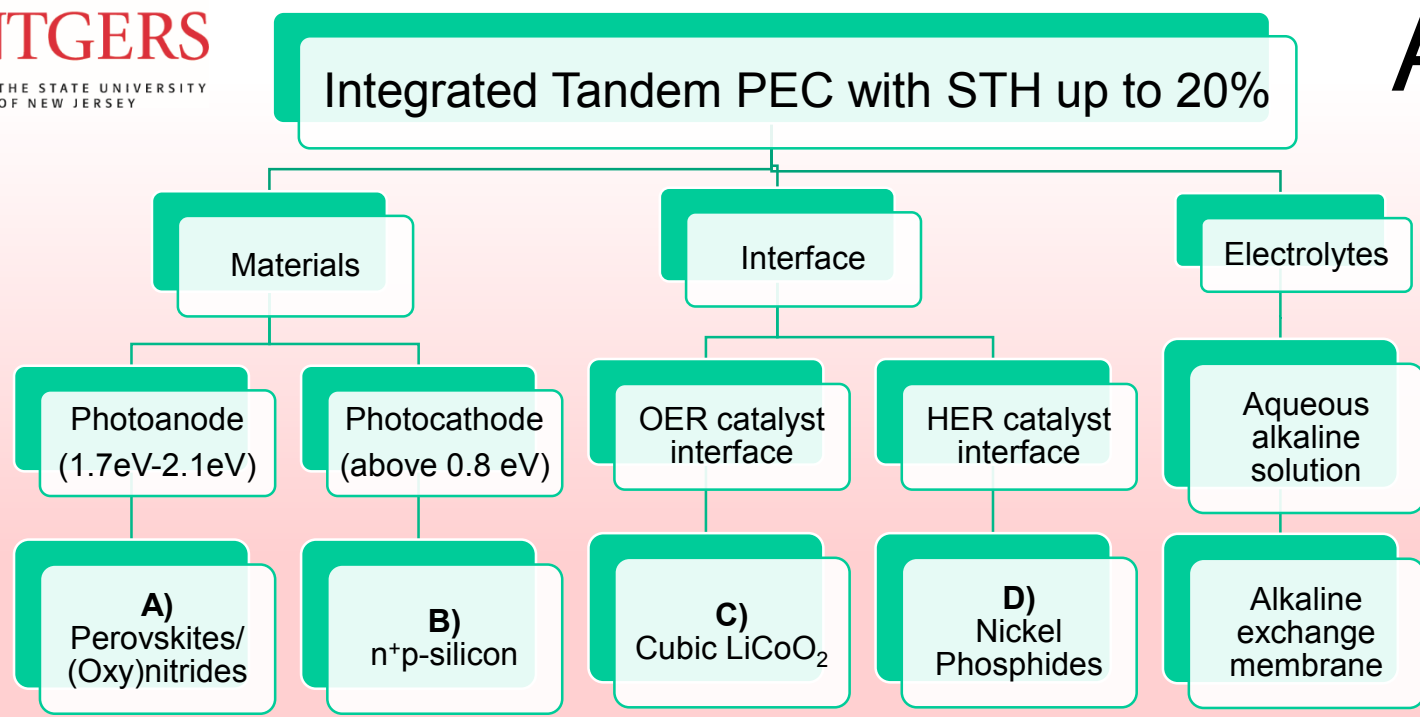
### (DOE section 3.1.5 of FCTO Multi-Year Research, Development, and Demonstration Plan)

*Materials Efficiency – Bulk and Interface (Barrier AE):* Developing methods that utilize Si adds value the high efficiencies of commercial cells. When paired with a TiN, absorption and injection efficiency is high.

*Materials Durability – Bulk and Interface (Barrier AF) –* Fabricated photocathode shows long term stability (>24 hours) while retaining high performance (145mV overpotential vs Pt for HER at 10mA/cm<sup>2</sup>)

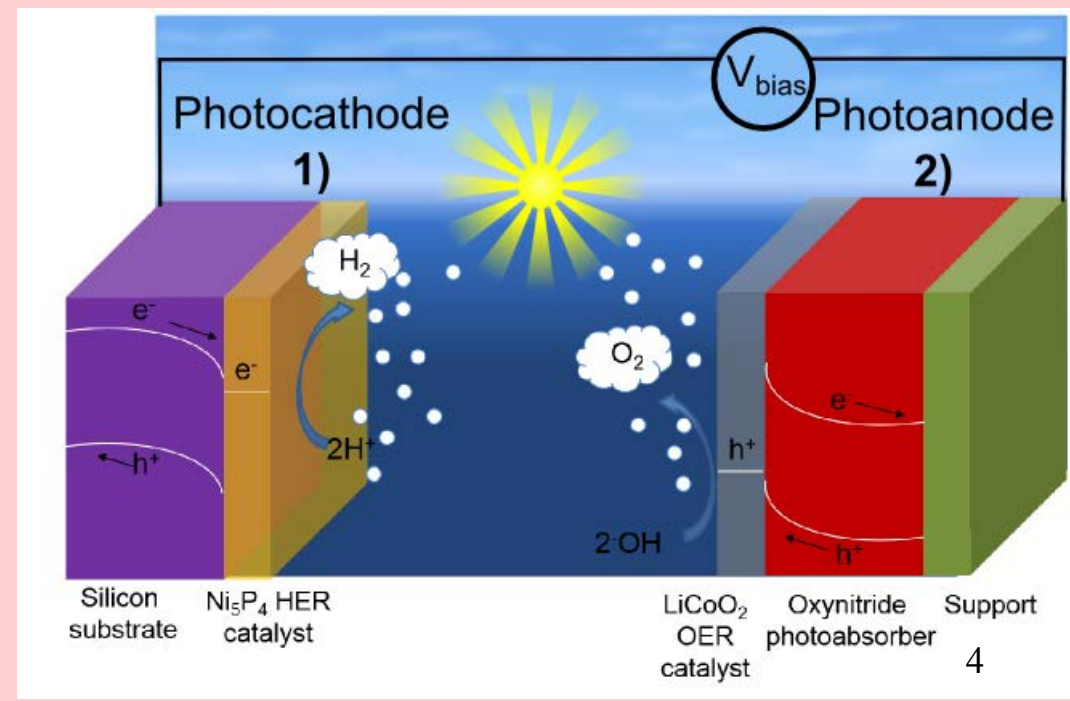
*Auxiliary Materials (Barrier AI) –* With TiN passivation, silicon can now be employed as both a substrate and a low bandgap photoabsorber in the photocathodic half cell. This reduces interfaces and materials costs.

*Synthesis and Manufacturing (Barrier AJ) –* Wide bandgap photoabsorber thin films that are fabricated by spin coating has significantly lower capital costs compared to CVD or PVD techniques.



## Project Focuses for FY2017:

- A-D** and combinations thereof
  - Prepare and characterize thin films of **A)** SrNbO<sub>2</sub>N, Ta<sub>3</sub>N<sub>5</sub>, and Cs<sub>2</sub>BiAgBr<sub>6</sub>
  - Fabricate functioning half cells, **1** and **2** (right) from the combination of **A + C** and **B + D** using above developed methodology
  - Benchmark half-cells for stability and performance



## Photocathode

- HER catalyst/passivation/photoabsorber =  $\text{NiP}_2/\text{TiN}/\text{n}^+\text{p-Si}$ 
  - Determine most optimal nickel phosphide compound ( $\text{Ni}_x\text{P}_y$ ) for integration with device
  - Identify a passivation layer that is transparent, highly conductive, has good band matching with Si, and prevents the formation of nickel silicides
  - Fabricate and characterize performance of device in both acidic and basic electrolytes
  
- Future effort (FY 2017)
  - Optimize system variables:
    - Film thickness (passivation layer and HER catalyst), different nickel phosphide polymorphs ( $\text{Ni}_5\text{P}_4$ , and  $\text{Ni}_3\text{P}$ ), fabrication steps (temperature, time)
  - Begin long term testing >100h using chronoamperometry
  - Stabilize interface, investigate corrosion mechanism
    - Time-dependent AFM, ion scattering and cross-section TEM will monitor surface and interface changes

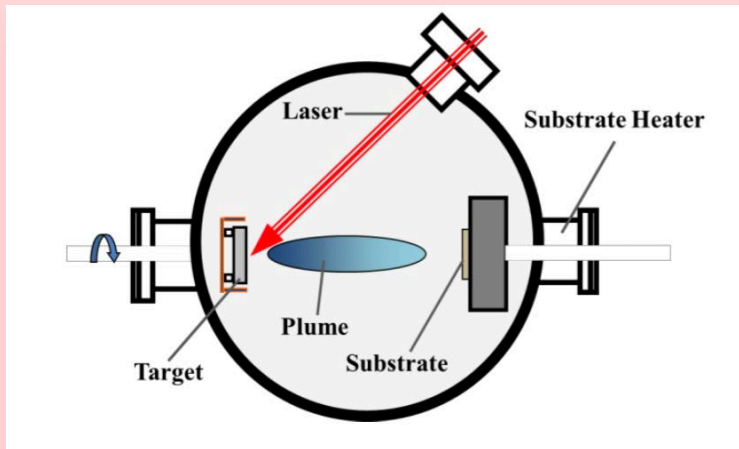
## Photoanode

- Double perovskites as a wide band gap photoabsorber material
  - Synthesized bulk powders of  $\text{Cs}_2\text{AgBiBr}_6$ . Verified the stability and optical properties of this class of compounds.
  - Developed a method to thin films with these materials
  
- $\text{Ta}_3\text{N}_5$  as a wide band gap photoabsorber material
  - Synthesized thin films of  $\text{Ta}_3\text{N}_5$
  - Confirmed viability of a tandem photoabsorber:  $\text{Ta}_3\text{N}_5$  on Si.
  - Tested  $\text{Ta}_3\text{N}_5$  as a photoanode for OER
  - Developed a passivation layer that is transparent, conductive, and stable
  
- Future effort (FY 2017)
  - Determine the band structure of  $\text{Cs}_2\text{AgBiBr}_6$  thin films experimentally
  - Develop a half device:  $\text{LiCoO}_2$ /passivation layer/photoabsorber #1 and #2
    - Achieve a clean interface between OER catalysts and photoabsorber and establish baseline performance
    - Introduce passivation layer (NiO et. al.) between OER catalyst and photoabsorber to increase stability of devices

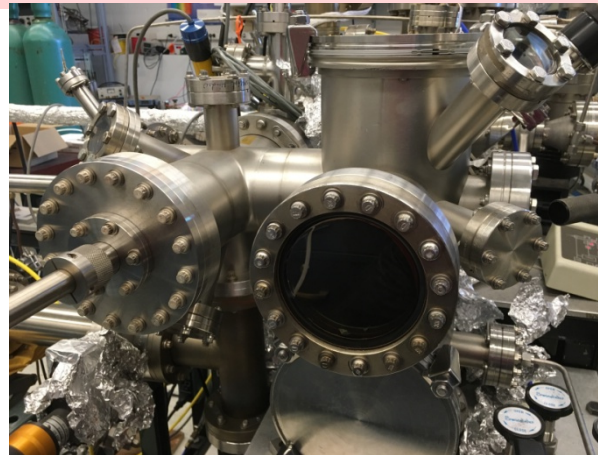
# Accomplishments: Photocathode device

Developed a thin film fabrication route to a complete device

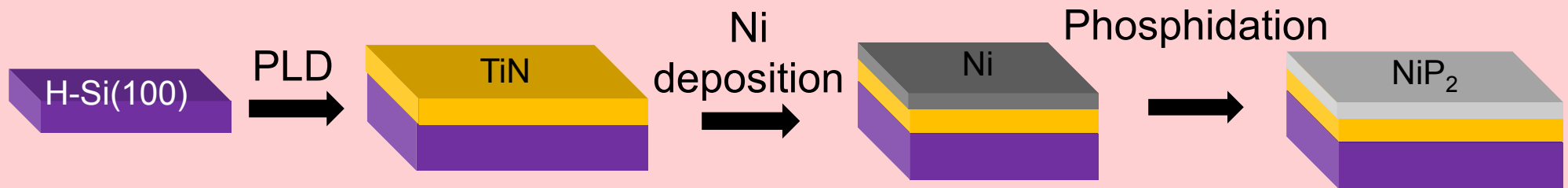
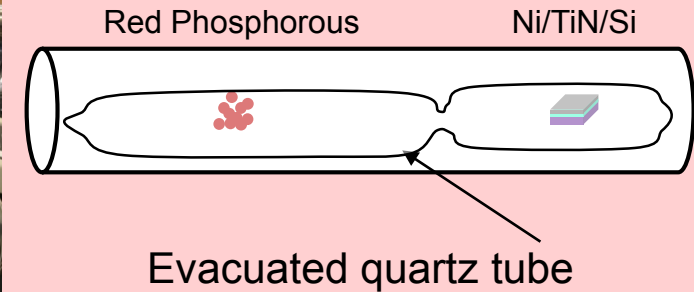
**Pulsed laser deposition**



**Thermal evaporation**



**Chemical vapor phosphidation**

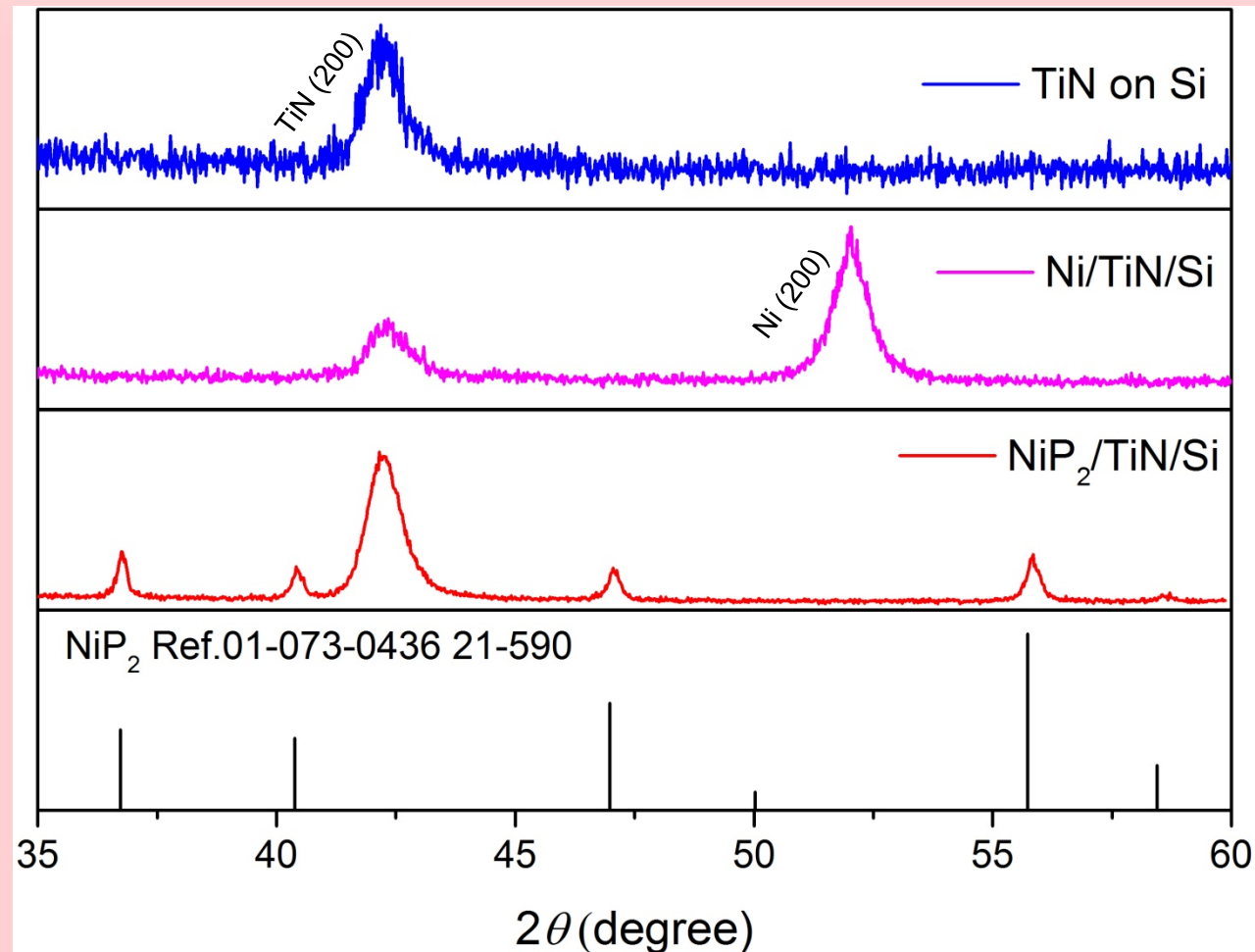




# Accomplishments: Photocathode device

## Atomic Structures Characterized

### XRD for each fabrication step



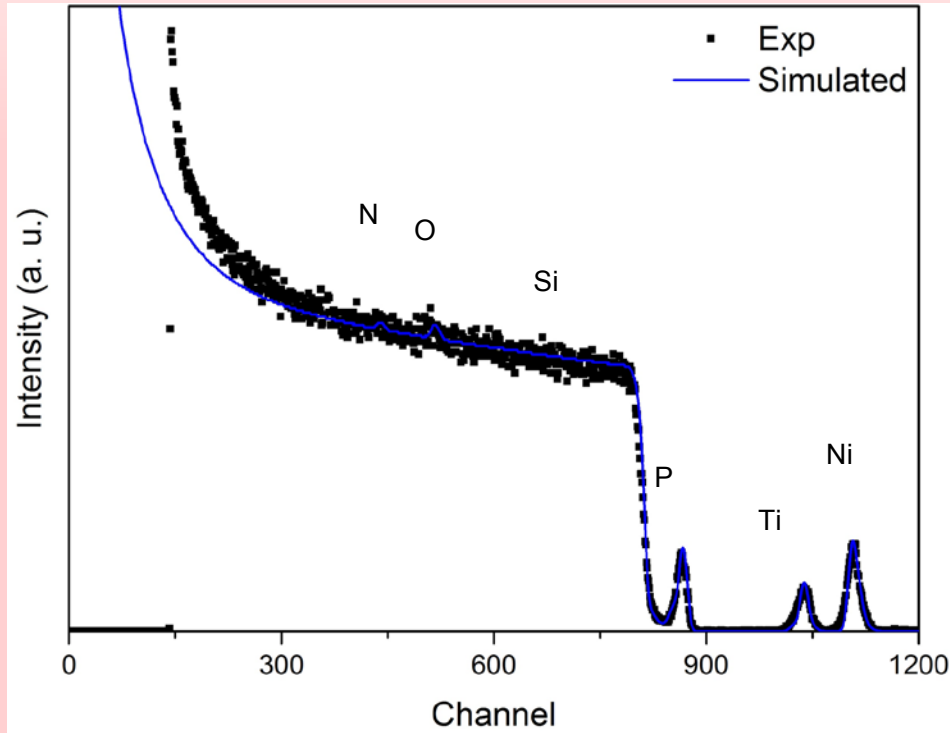
- To determine crystalline phase of each layer, thicker layers ( $\sim 50$  nm) of TiN and Ni were deposited.
- TiN shows an oriented (200) peak, which is the same orientation as the cubic Si substrate (100).
- Oriented, thermally evaporated Ni is observed.
- Polycrystalline  $\text{NiP}_2$  was successfully prepared by conversion from Ni by chemical vapor phosphidation. Other  $\text{Ni}_x\text{P}_y$  compounds can be prepared similarly.



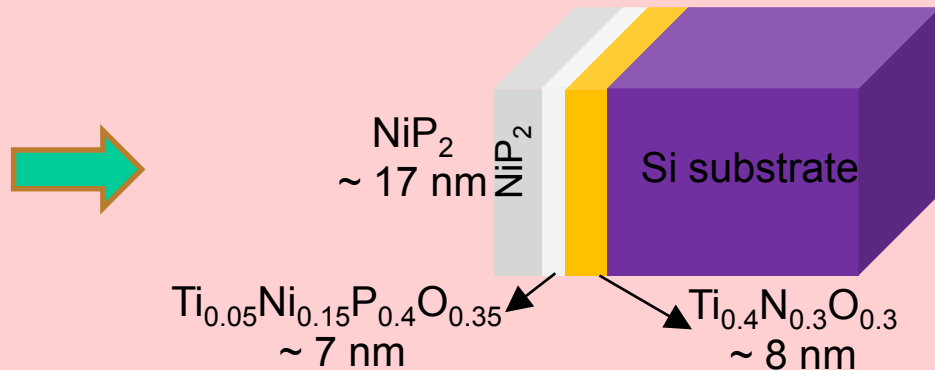
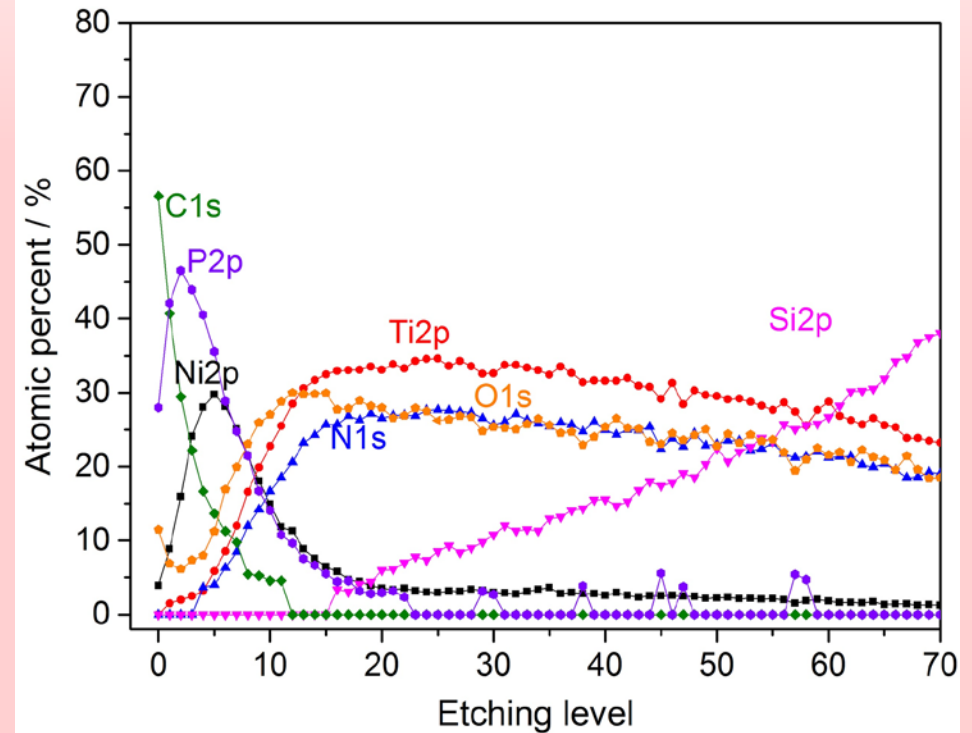
# Accomplishments: Photocathode device

## Thin Films Characterized

### Rutherford backscattering spectrometry (RBS)



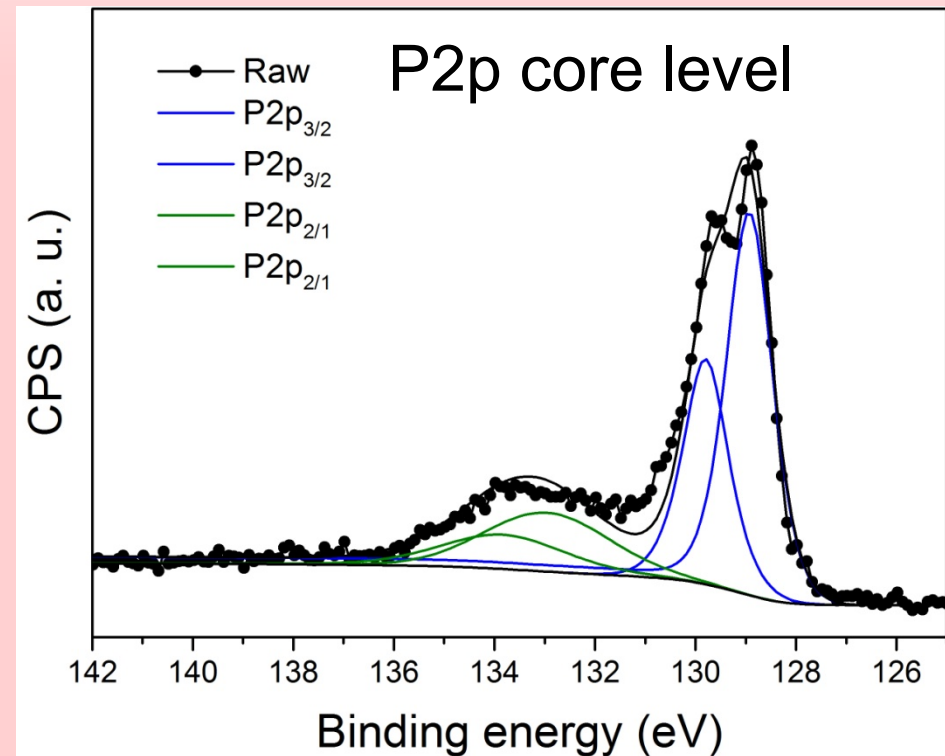
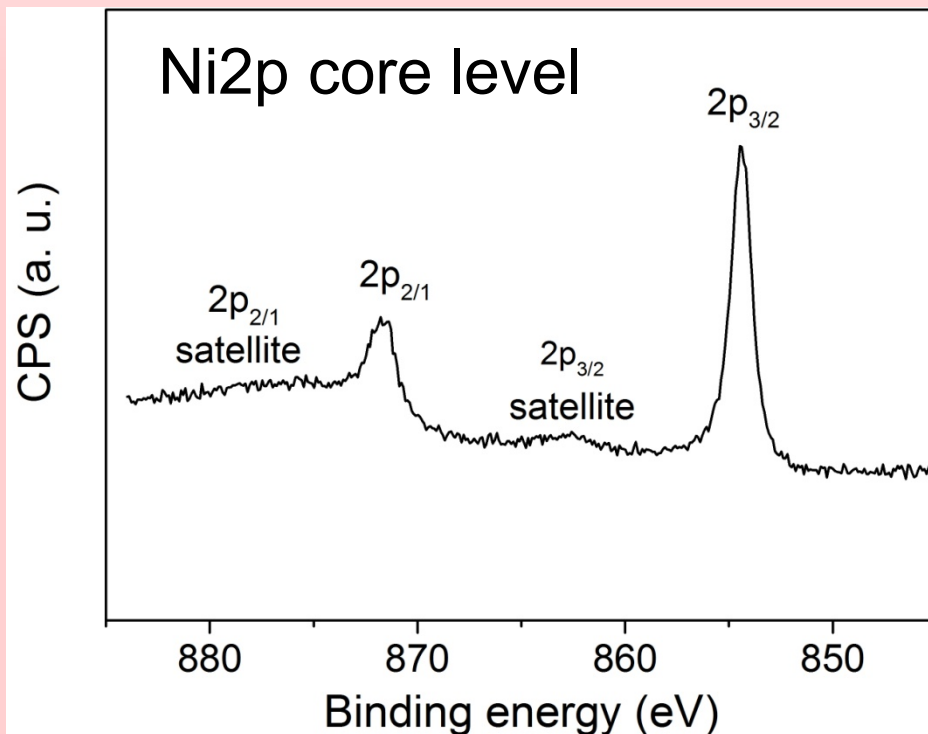
### Depth resolved XPS



The cations are well resolved by RBS and depth resolved XPS identifies the anions. The two data sets can be made into a composite representation of the fabricated film.

# Accomplishments: Photocathode device

## X-ray Photoelectron Spectroscopy



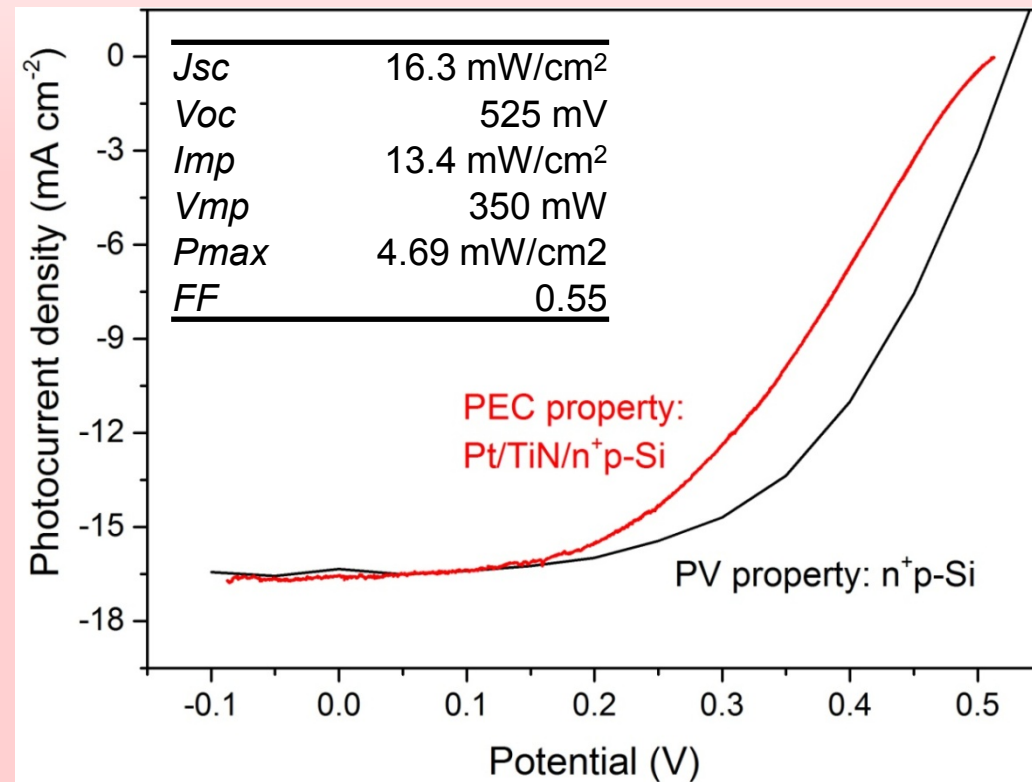
- Ni 2p core level spectrum shows metallic Ni feature without surface oxidized Ni.

- P 2p core level spectrum shows that main phosphorous has metallic character.

# Accomplishments: Photocathode device

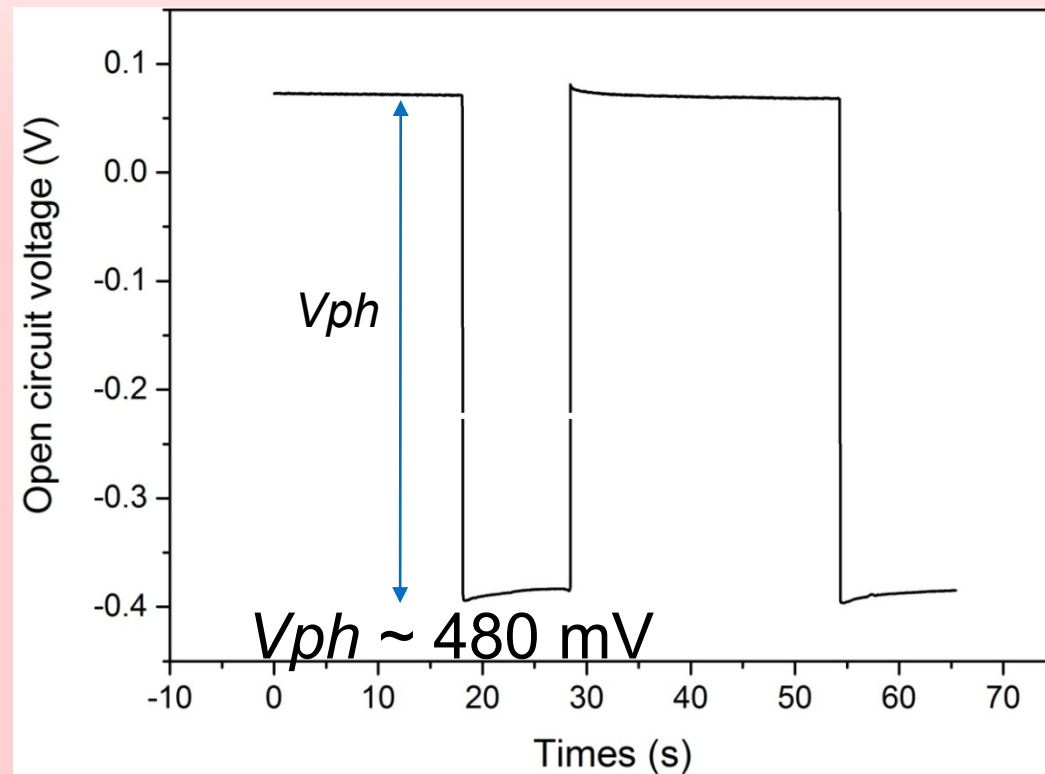
## Photo-properties of NiP<sub>2</sub>/TiN/n<sup>+</sup>p-Si

### PV



- The *J*<sub>sc</sub> of the PEC measurement from Pt/TiN/n<sup>+</sup>p-Si (reference) sample shows similar *J*<sub>sc</sub> to a solar simulated PV (n<sup>+</sup>p-Si). This indicates the TiN passivation layer has negligible light absorption.

### PEC

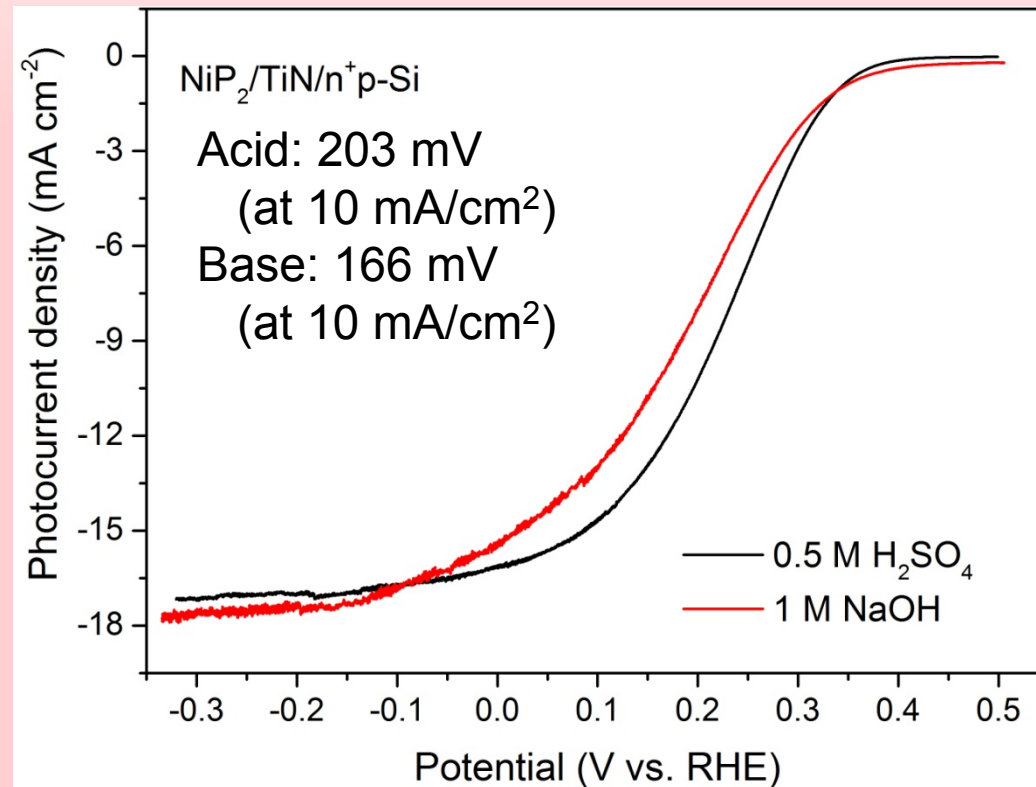


- The *V*<sub>ph</sub> of the PEC, measured by the difference between light and dark OCP demonstrates that the NiP<sub>2</sub> and TiN layer did not significantly change the *V*<sub>oc</sub> of Si photoabsorber.

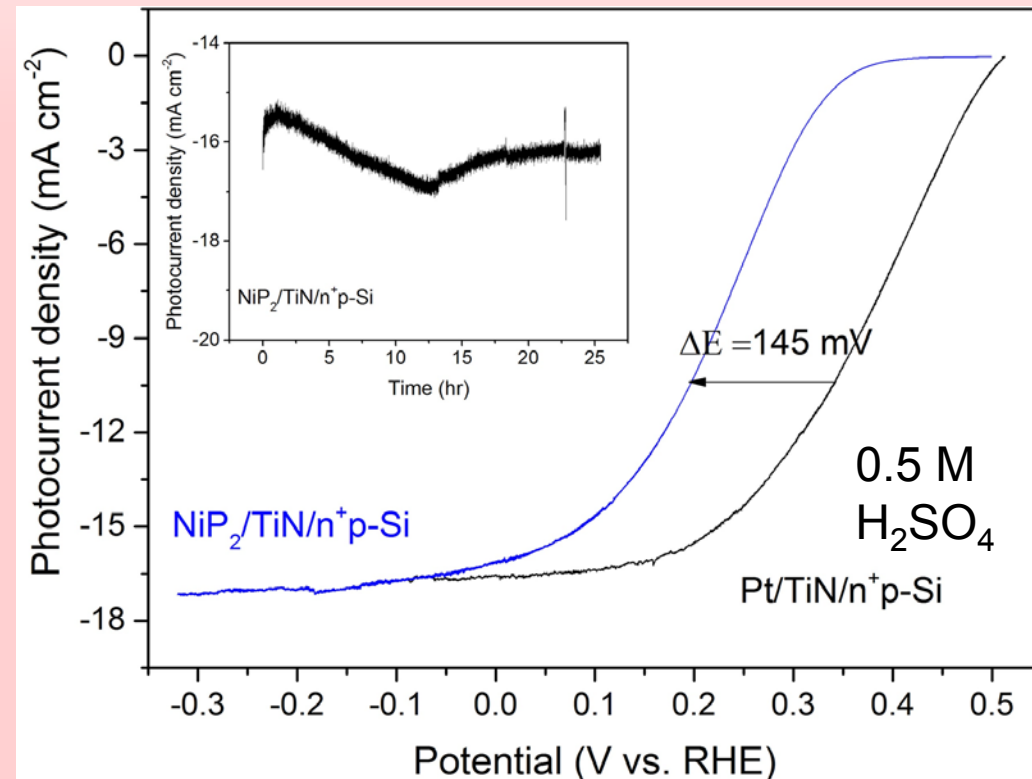
# Accomplishments: Photocathode device

## Photoelectrochemistry

### Activity



### Stability

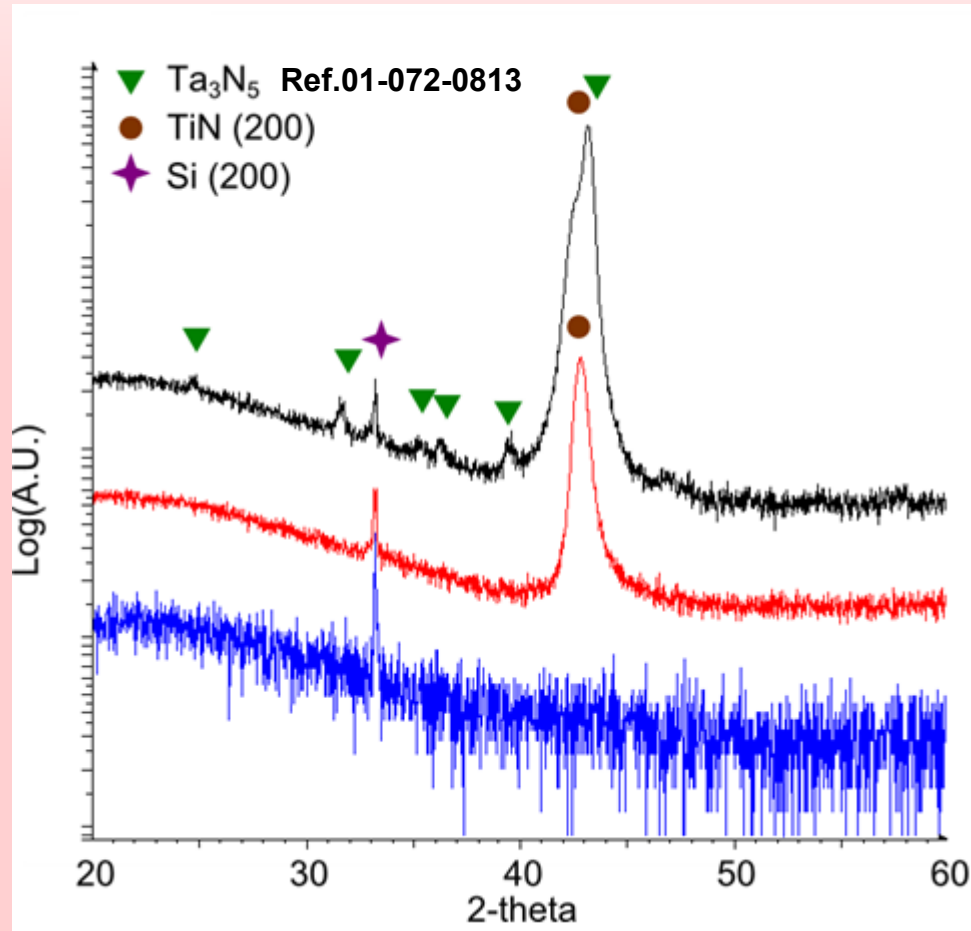


-  $\text{NiP}_2/\text{TiN}/\text{n}^+\text{p-Si}$  shows high activity in both acidic and basic media.  $J_{\text{sc}}$  of both averaged cyclic voltammograms suggest that light absorption interference is negligible from the  $\text{NiP}_2$  and TiN layer (compared to  $J_{\text{sc}}$  of  $\text{n}^+\text{p-Si}$ ). In acidic media,  $\text{NiP}_2$  requires only a 145 mV overpotential against a  $\text{Pt}^0$  catalyst, and it shows photostability for more than 25 hours without failure.

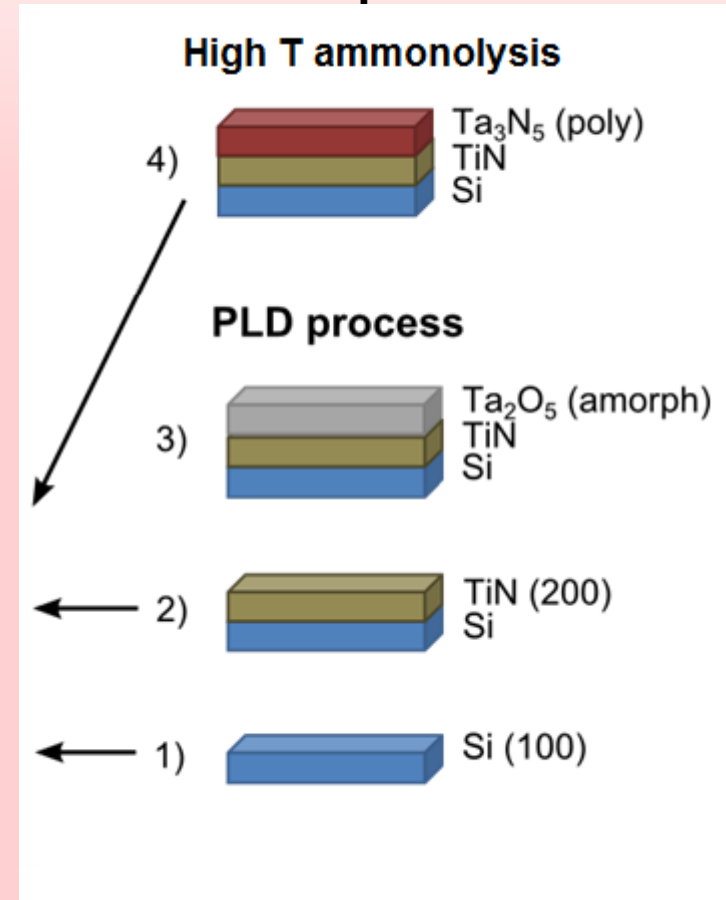
# Accomplishments: Photoanode device

Ta<sub>3</sub>N<sub>5</sub> photoabsorber

## XRD



## Fabrication Steps



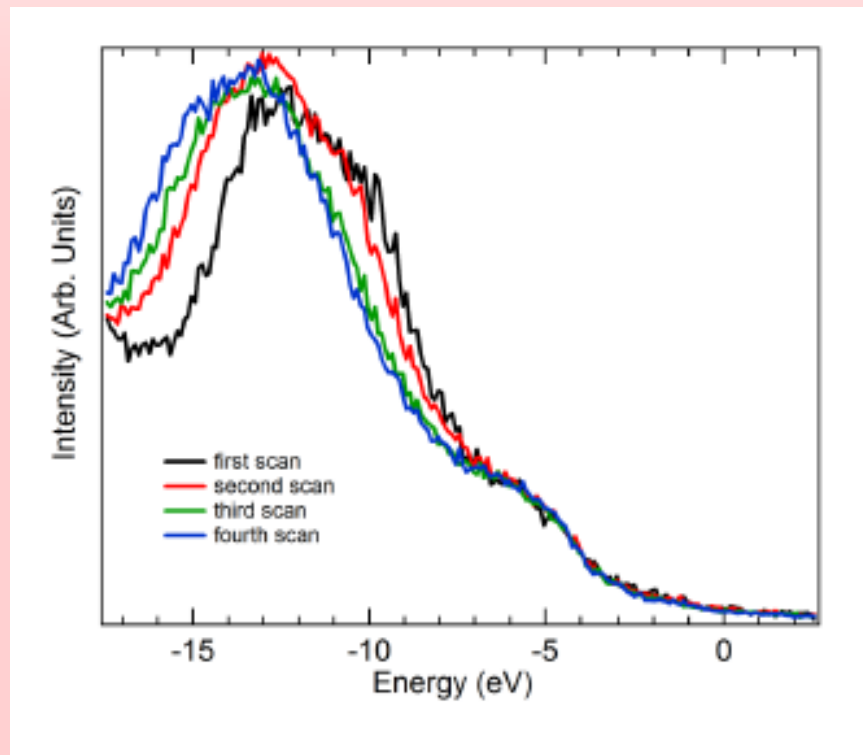
- TiN is (200) oriented while Ta<sub>2</sub>O<sub>5</sub> is deposited amorphous. During ammonolysis, the Ta<sub>2</sub>O<sub>5</sub> is converted to a polycrystalline Ta<sub>3</sub>N<sub>5</sub>.

# Accomplishments:

## Photoanode device

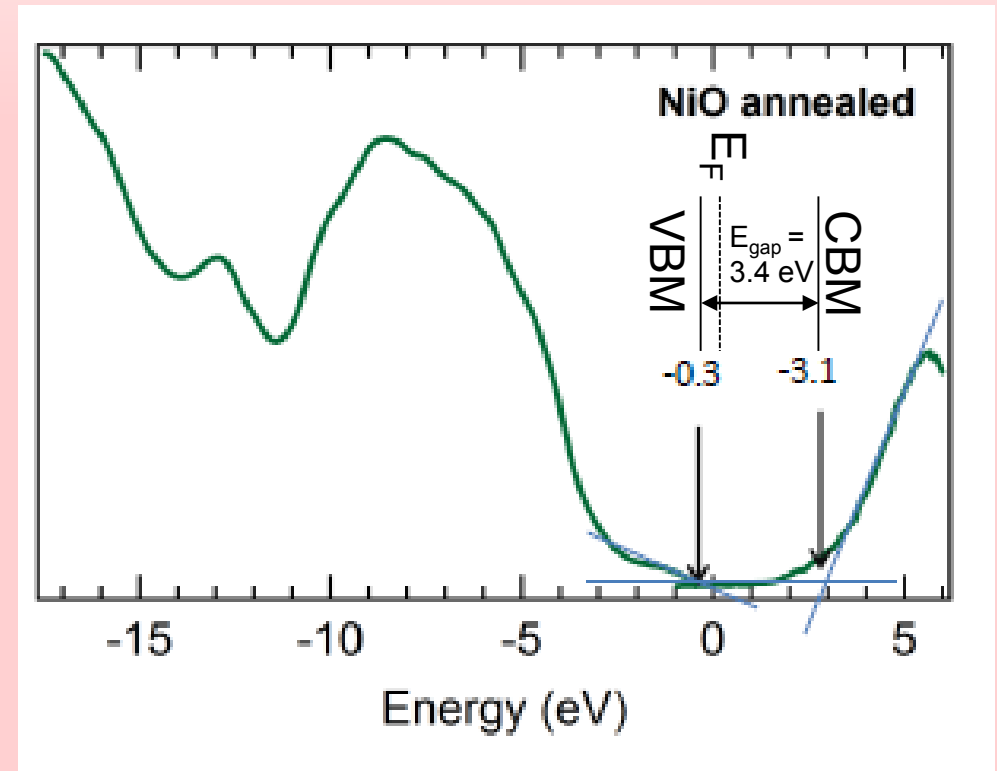
Photanode NiO passivation layer developed

### UPS



- Freshly sputtered NiO films charge due to insulating behavior. A post-sputter anneal (data not shown) allows higher hole conductivity. Conductivity changes from insulating to metallic ( $\rho < 50 \Omega \cdot \text{cm}$ )

### Inverse/Photoemission Spectroscopy



- The experimentally determined band gap of the annealed NiO is wide ( $E_{\text{gap}} = 3.4 \text{ eV}$ ), which further promotes its candidacy as a passivation layer in the photoanode.

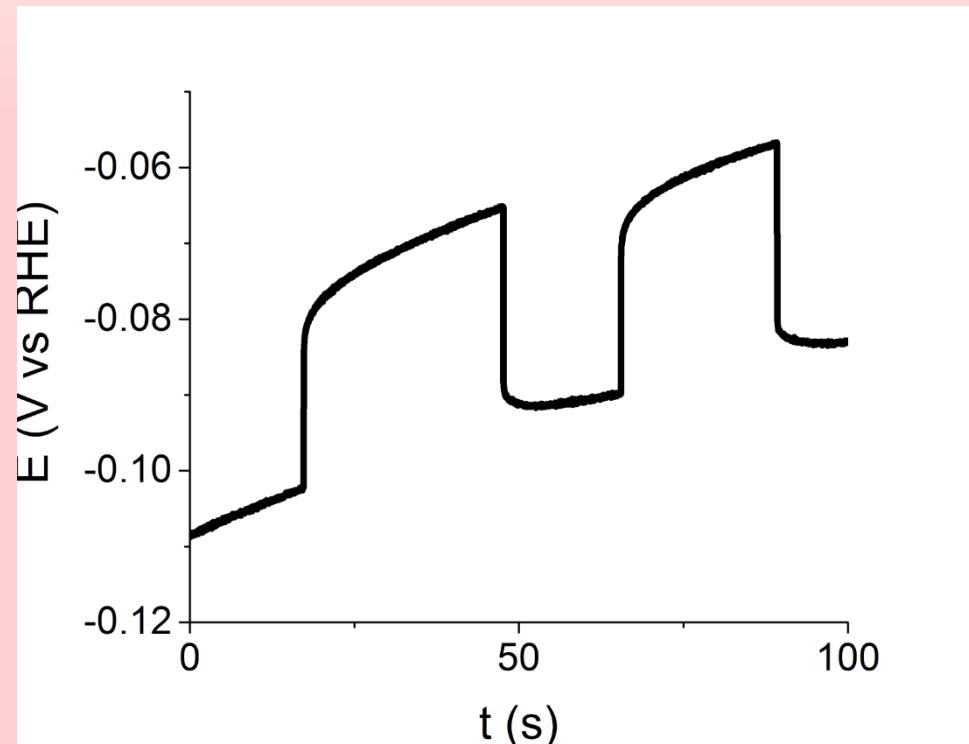


# Accomplishments:

## Photoanode device

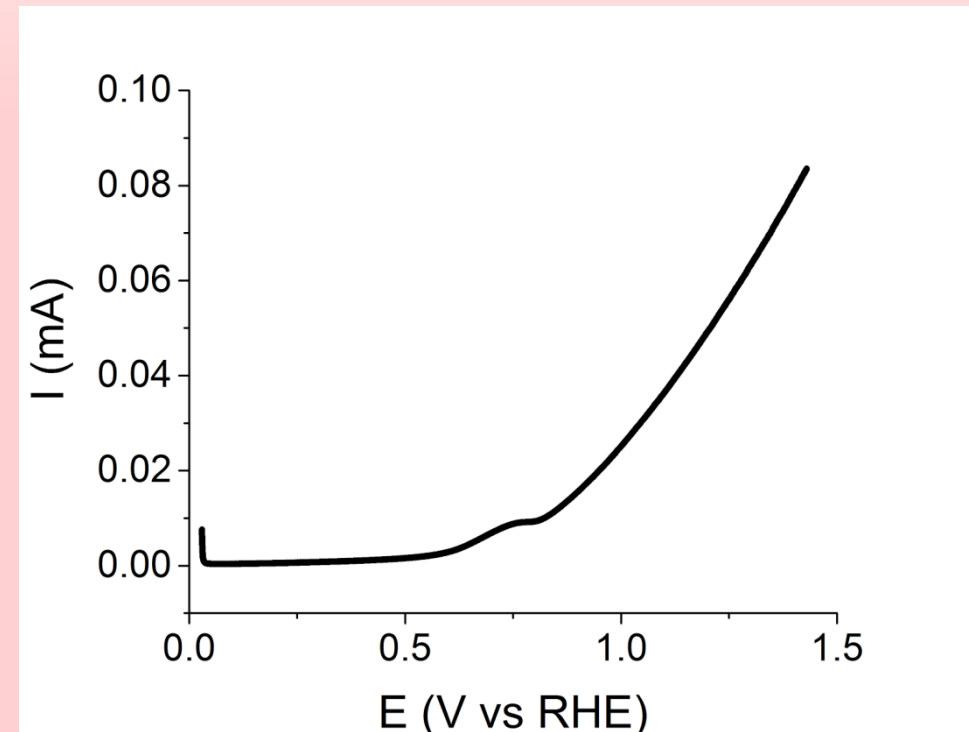
### Preliminary PEC of NiO/Ta<sub>3</sub>N<sub>5</sub>/TiN/n-Si

**OCV**



-  $V_{ph}$  is observed but is very low (20mV).

**Illuminated LSV**



- At increasing potentials an anodic current is observed.

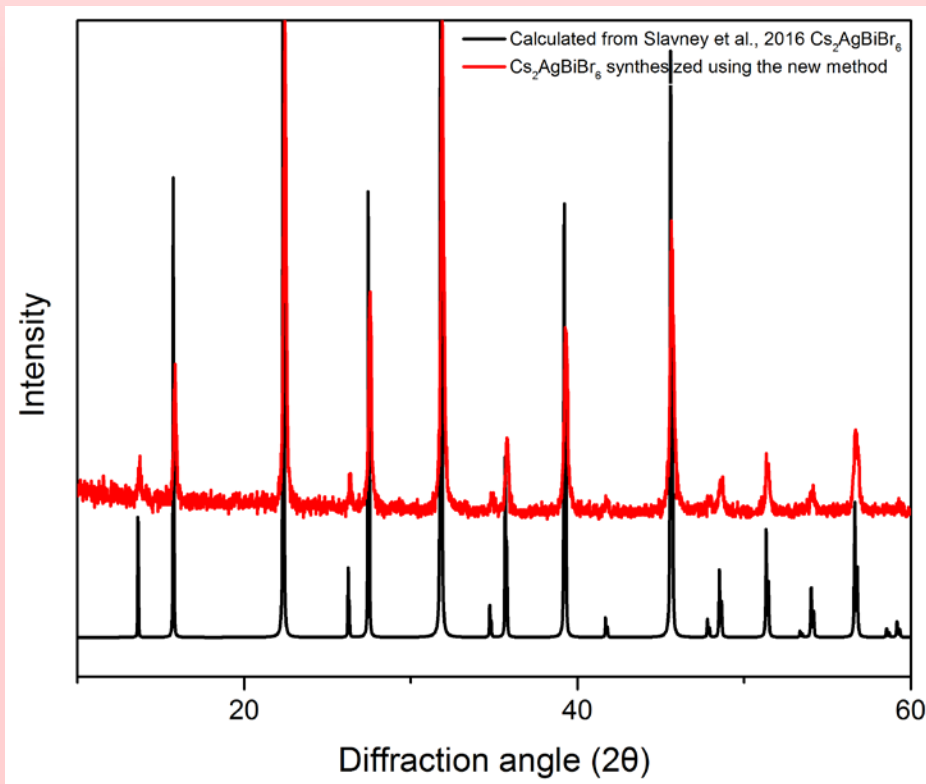
Half-cells are unstable due to photocorrosion. Optimization of the passivation layer (thickness and post-anneal) and experiments to thoroughly evaluate each layer's contribution to the PEC properties are in progress.

# Accomplishments:

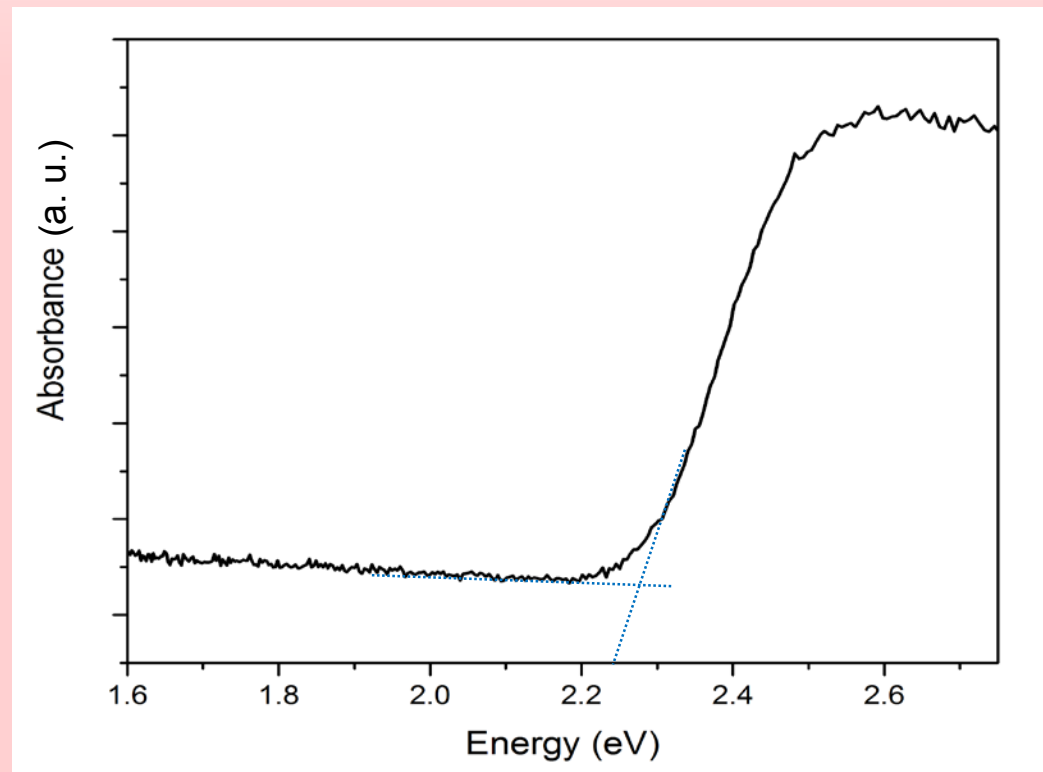
## Photoanode device

Double perovskite halide ( $\text{Cs}_2\text{AgBiBr}_6$ ) synthesized as photoabsorber

### PXRD



### Optical properties



- Bulk powders match those reported previously. Large crystals are formed during solution synthesis.

- Bandgap is  $\sim 2.3\text{eV}$  from UV-visible diffuse reflectance, in line with other reports.

# Accomplishments:

## Photoanode device

### Spin-coated thin films of $\text{Cs}_2\text{AgBiBr}_6$

#### Development of a solution processing route



**Precursor solution**

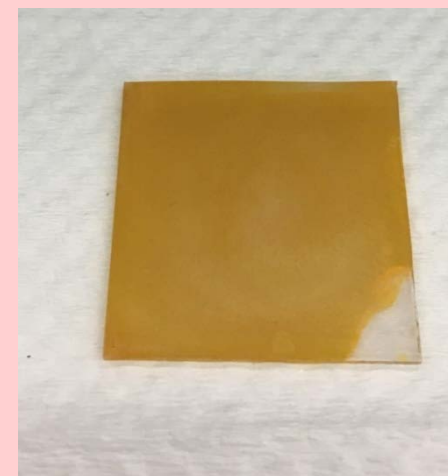
**Spin-coating solution**

**Precipitated crystals**

**Yield 72%**

- Excess solution decanted & the crystals are washed with ethanol

#### Fabricated film



- Triple layered spin-coated  $\text{Cs}_2\text{AgBiBr}_6$  films on top of ZnO/FTO.

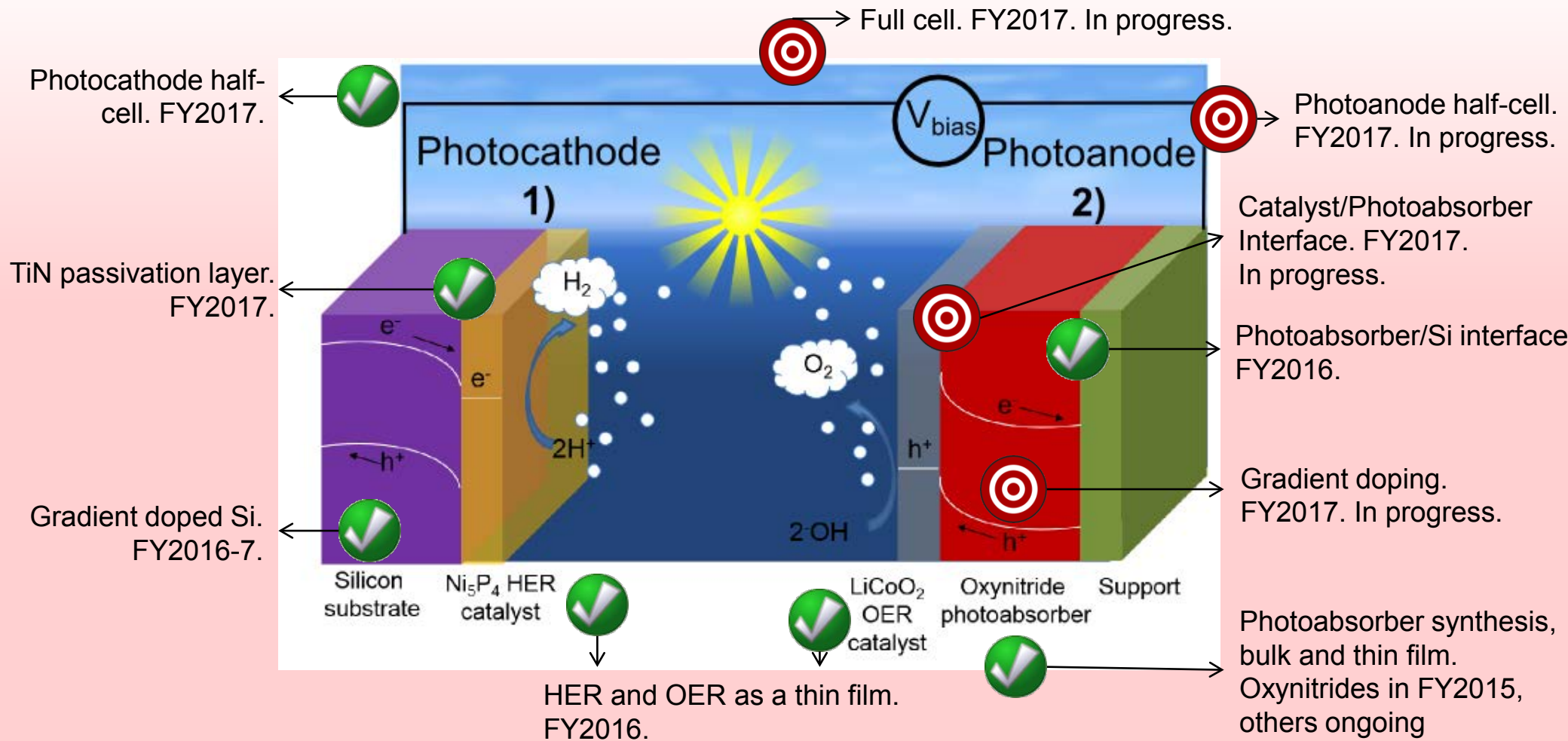
## Photocathode

- **Successful formation of the polycrystalline NiP<sub>2</sub> thin film (~ 17 nm) with TiN passivation interlayer on a silicon photoabsorber. Other Ni<sub>x</sub>P<sub>y</sub> options open.**
- **Developed TiN passivation interlayer to prevent thermal diffusion of Si during phosphidation reaction that forms NiP<sub>2</sub>, yielding a non-diffusing interface.**
- **Metallic NiP<sub>2</sub> and TiN layers did not significantly interfere with light absorption of silicon photoabsorber thereby obtaining similar *J*<sub>sc</sub> to bare Si (PV measurement).**
- **The half cell device demonstrated high activity in both acid (pH 0) and base (pH 14). Low 145 mV overpotential was required vs. ref. Pt catalyst in acid.**
- **The developed device was stable for over 25 h without failure.**

## Photoanode:

- **A route to thin films of Ta<sub>3</sub>N<sub>5</sub> on Si for tandem photoabsorber applications has been identified.**
- **NiO passivation layer is conductive and transparent promoting its use in the photoanode, though preliminary PEC data needs more understanding.**
- **Spin-coated Cs<sub>2</sub>AgBiBr<sub>6</sub> has been demonstrated as a low-cost, scalable candidate for the wide bandgap photoabsorber in a tandem PEC.**

# Full project summary



## • Key FY2017 technical results:

- A stable (>25h in acid with no significant change to operating current 10% gain in the first 10 hours due to activation; base has same stability but instead a 10% loss in operating current in the first 10 hours), highly active (145mV overpotential for HER at 10mA/cm<sup>2</sup> vs Pt in 0.5M H<sub>2</sub>SO<sub>4</sub>) photocathode utilizing a TiN passivation layer was fabricated and characterized.
- Scalable low-temperature routes to wide bandgap absorbers (SrNbO<sub>2</sub>N, Ta<sub>3</sub>N<sub>5</sub>, Cs<sub>2</sub>AgBiBr<sub>6</sub>) that can be fabricated upon silicon (low bandgap photoabsorber) were identified and investigated with the intent to optimize the interfacial chemistry on route to a full tandem device

# Relating to EMN HydroGEN

- Looking for collaborators that have these capabilities:

- Wide and/or narrow band gap semiconductors
- Diurnal testing
- In-situ e-chem analysis techniques

- Things our team has and can collaborate with:

- Thin film HER and/or OER catalyst integration with your device

- Characterization

- “ultra”-Scanning Tunneling Electron Microscope (STEM)
  - Sub-nm scale
  - Vibration, plasmon, interband-transition and core-shell transition spectra capable
- Medium Energy Ion Scattering (MEIS)
  - High-resolution-surface-“Rutherford backscatter technique”
- Helium Ion Microscopy (HeIM)
  - High resolution even for an untreated semiconductor

