



HydroGEN: High Temperature Electrolysis (HTE) Hydrogen Production

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Presenter: Richard Boardman

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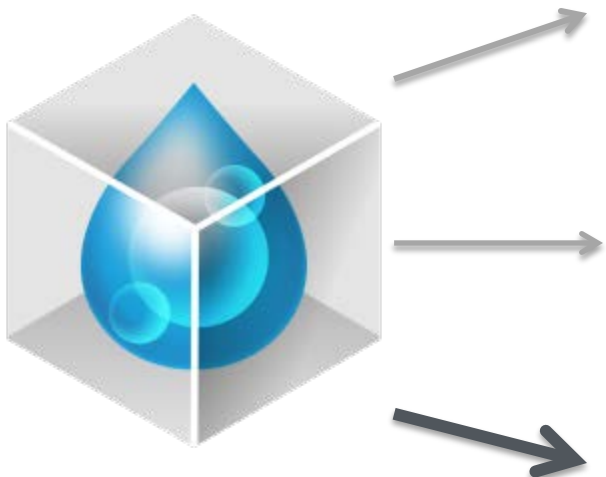


Advanced Water-Splitting Materials (AWSM) Relevance, Overall Objective, and Impact

**AWSM Consortium
6 Core Labs:**



Accelerating R&D of innovative materials critical to advanced water splitting technologies for clean, sustainable & low cost H₂ production, including:

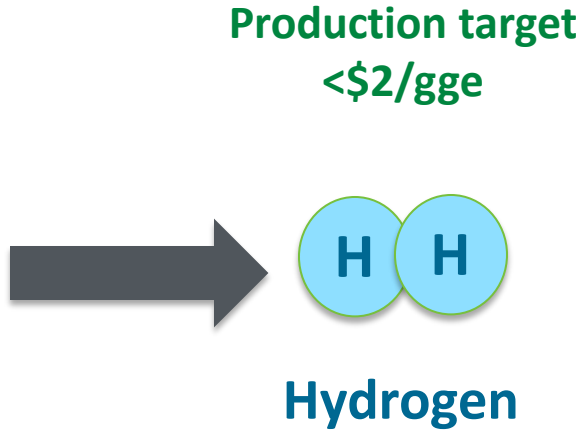


Water

Photoelectrochemical (PEC)

Solar Thermochemical (STCH)

Low- and High-Temperature Advanced Electrolysis (LTE & HTE)



- Powder Synthesis
- Electrolyte ion-conduction
- Combinatorial materials deposition

- Electrodes microstructure
- Infiltrated catalysts metal support electrodes
- Perovskite formulations

Lower TRL

HydroGen Consortium

HTE Cell Improvement

HTE electrodes & electrolytes

Thermo-electrical energy to H₂
 Catalysis & H.T. ion-conduction
 Cell/stack efficiency & stability
 Balance of plant reactor designs
 Techno-economic life cycle assessments

- Conductivity & polarization measurements
- Durability testing
- Embedded sensors

HTE Stack Improvement

Higher TRL

- Seals and interconnections
- Stack design and fabrication
- Heat management

Looking Inward: Crosscutting challenges that bind us together
 Looking Outward: Unique materials development frontiers

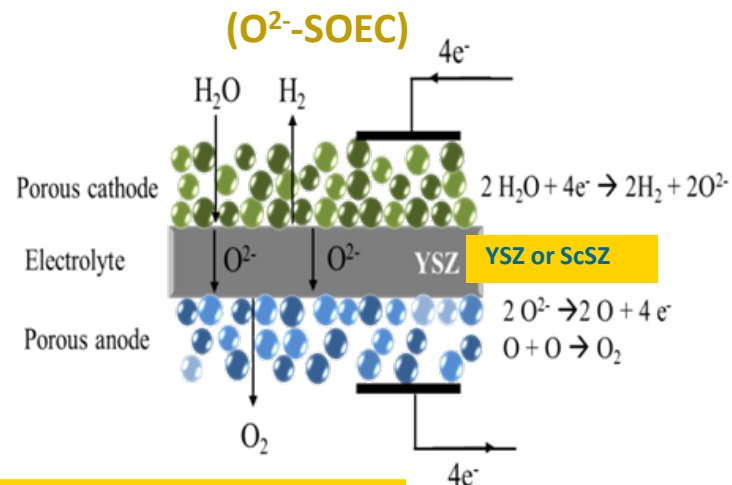
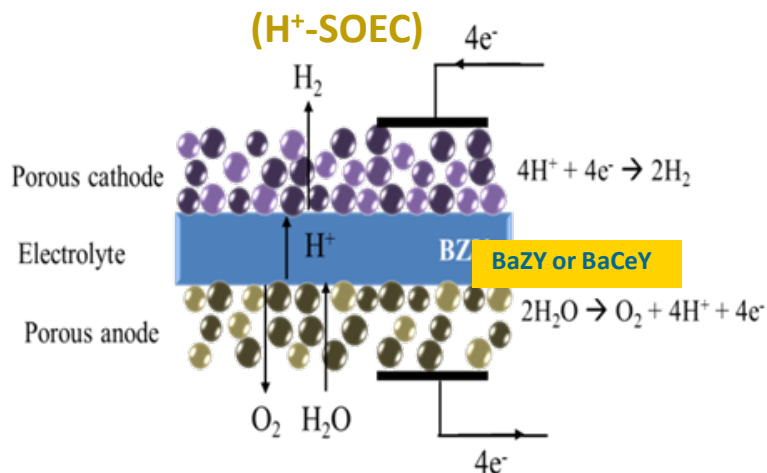


High Temperature Electrolysis (HTE) Technologies

Approach

Proton-Conducting Solid-Oxide Electrolysis

Oxygen Ion Transport Solid-Oxide Electrolysis



| Attributes | H ⁺ - SOEC | O ²⁻ - SOEC |
|--------------------------|----------------------------------|-----------------------------------|
| Operating Temperature | 550-750°C | 650-850°C |
| Electrolyte Conductivity | 0.01 S.cm ⁻¹ at 650°C | 0.015 S.cm ⁻¹ at 850°C |
| Cathode Products | Pure H ₂ | H ₂ O + H ₂ |
| Anode Products | O ₂ + sweep gas | H ₂ O + O ₂ |

Figure and Table Adapted from:
Singh and Hu, UConn

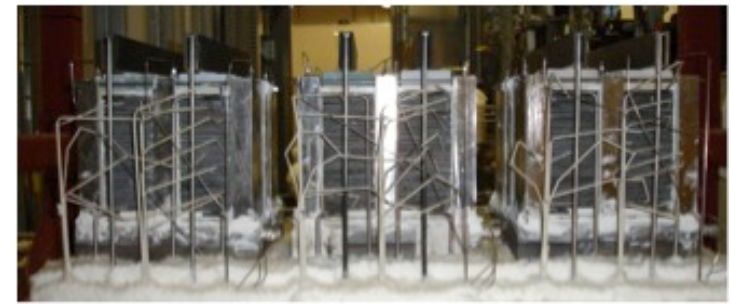
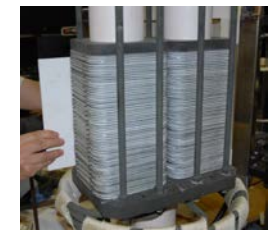
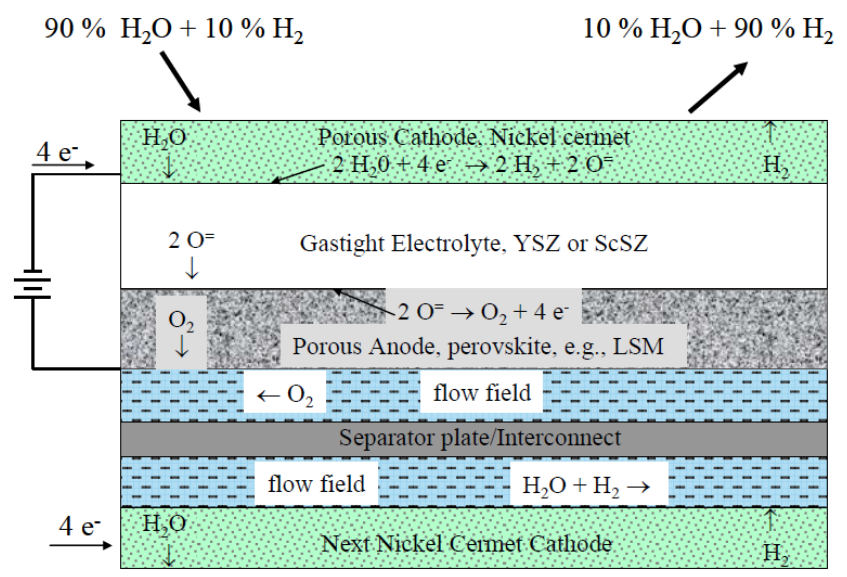
Trade-offs:

- Thermodynamic efficiency increases at higher temperature
- Solid oxide conduction and kinetics increase with temperature
- Solid oxide is purely proton conducting < 600°C
- Materials durability decreases with higher temperatures



Example: HTE Oxygen-Transporting Solid-Oxide Cell

Approach/Impact



- **Stacks must be tolerant to corrosion (Cr migration) and thermal cycling (discontinuous and dissimilar materials)**
- **Fabrication of SOECs cells require several steps and is energy intensive**



HydroGEN HTE Projects & Collaboration

Approach

High Temperature Electrolysis (HTE)

- Oxygen Conducting SOEC (o-SOEC)
- Proton Conducting SOEC (p-SOEC)

- Cost
- Efficiency
- Durability

HTE Node Labs



Support through:



Personnel
Equipment
Expertise
Capability
Materials

Data

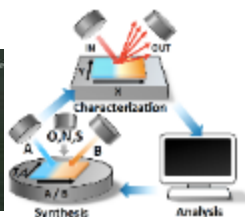
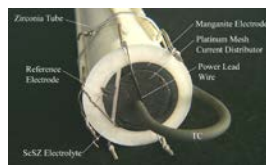
HTE Projects



United Technologies
Research Center

UConn

Northwestern



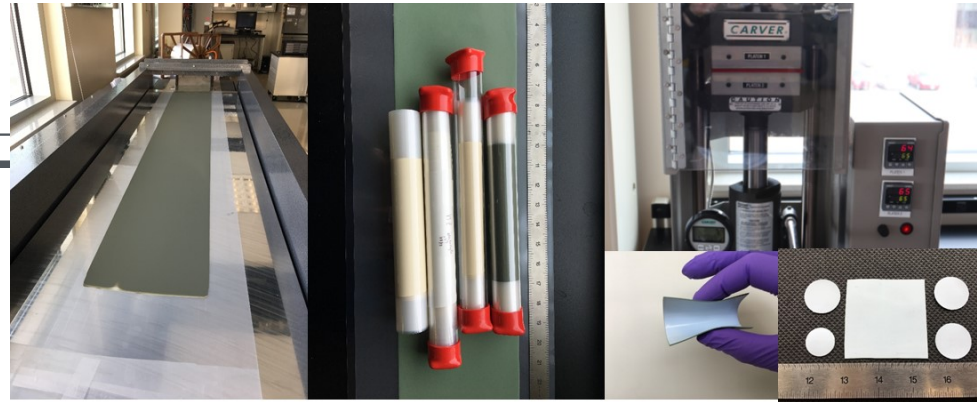


Example node: INL - HTE 1

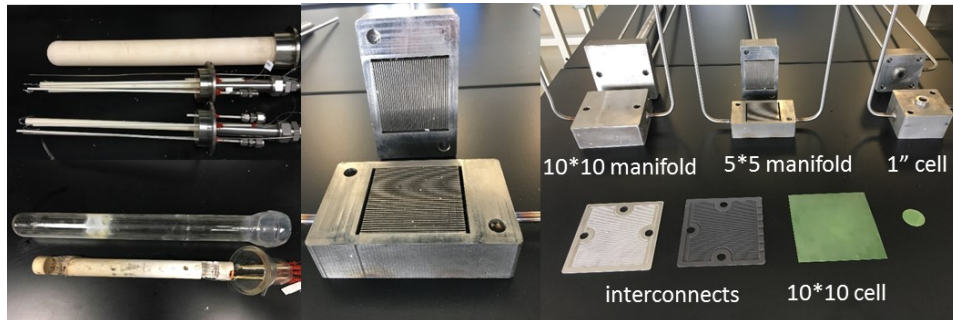
Advanced Materials for Elevated Temperature Water Electrolysis

HT Roll-to-Roll Manufacturing for button cells to large planar cells

Multi-functional sintering



High Throughput materials testing/screening



Testing fixtures for symmetrical cells, half cells, button cells, planar square cells and stacks (up to 19 channels)



Testing facilities for cells and stacks (17 channels with potentiostat and EIS)

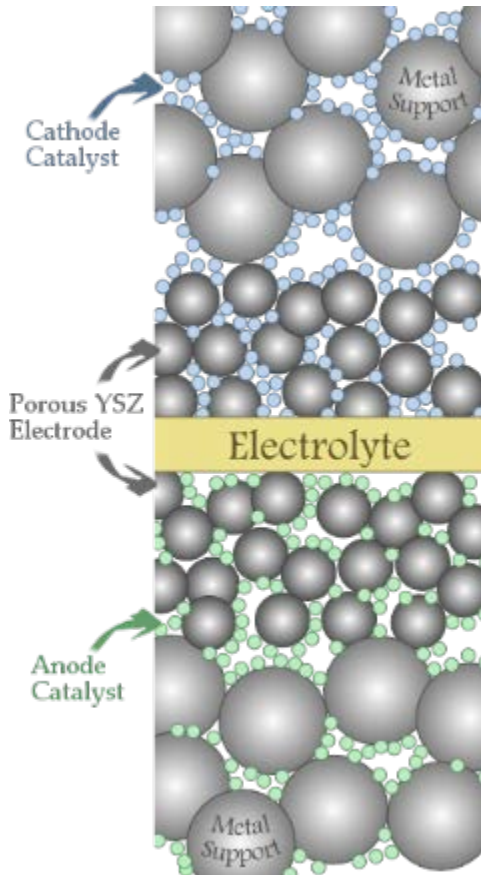
**Dong Ding,
Node Expert**



Example node: LBNL - HTE 3

Metal-Supported Solid Oxide Cells

Approach/Impact



➤ Cost

- Stainless steel order of magnitude less expensive than ceramics
- Single high-T sintering step

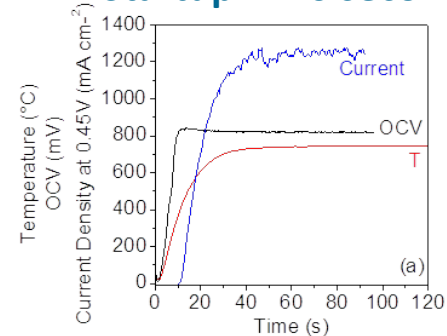
➤ Mechanical

- Rugged, strong cell
- Welded electrical connections
- Tolerates rapid T-ramp
- Tolerates large T-gradient

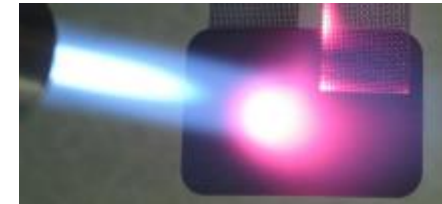
➤ Operational

- Very high performance
- T can vary quickly during operation (increase/decrease performance on the fly)
- Tolerates intermittent fuel/hydrogen (redox cycling)
- Imbalanced pressure acceptable

Startup in 20 seconds



>300° C/cm



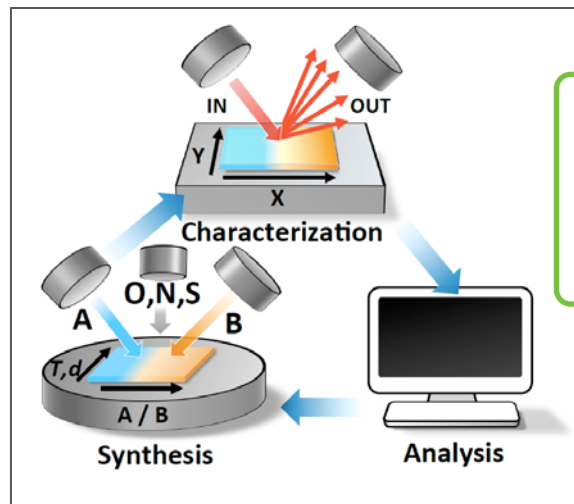


Example nodes: NREL - HTE 2 (also STCH and PEC)

High-Throughput Experiments Thin Film Combinatorial Capabilities

Approach/Impact

Existing: High-Throughput Experiments

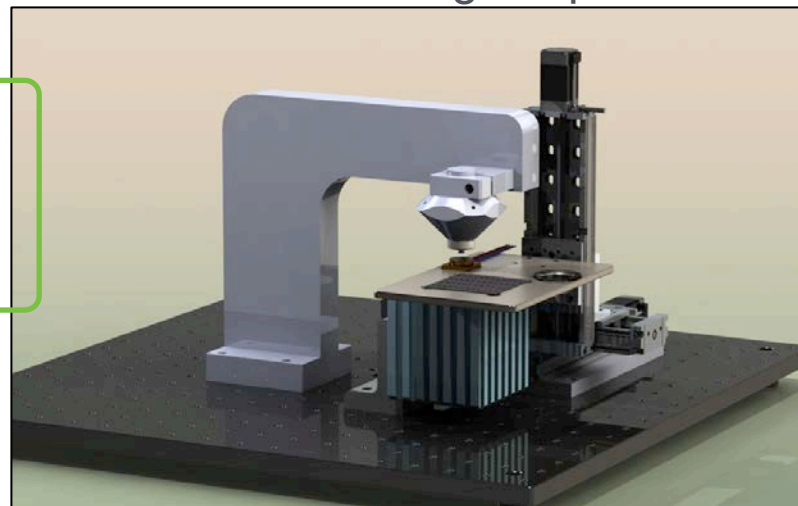


Andriy Zakutayev,
Node
Expert

- 6 combinatorial synthesis chambers
- 14 spatially-resolved measurement tools
- Big data management and analysis

Capabilities developed using BES and SETO

New: Scanning Droplet Cell



- Electrochemical measurement under light
- 44 spots tested on 1 sample library
- 1-2 mm spatial sample resolution

Capabilities developed using HydroGen EMM

<https://www.h2aws.org/capabilities/high-throughput-experimental-thin-film-combinatorial-capabilities>

HydroGEN EMM projects:

- **CSM**: Accelerated Discovery of Solar Thermochemical Hydrogen Production Materials
- **Rutgers**: Best-in-class PGM-free Catalyst Integrated Tandem Junction (PEC) Devices
- **UConn**: Proton-Conducting Solid Oxide Electrolysis Cells at Intermediate Temperatures
- **HNEI**: Novel Chalcopyrites For Advanced Photoelectrochemical Water Splitting



Overview: FOA Seedling Projects

5 Nodes utilized
7 Lab PIs engaged
6 samples exchanged
Files being compiled for Data Hub

Approach

Northwestern University (NWU, Scott Barney PD153):

Characterization and Accelerated Life Testing of a New Solid Oxide Electrolysis Cell

United Technologies Research Center (UTRC, Tianli Zhu; PD154):

Thin-Film, Metal-Supported High-Performance and Durable Proton-Solid Oxide Electrolyzer Cell

University of Connecticut (Uconn, Prabhakar Singh; PD15X):

Proton-Conducting Solid Oxide Electrolysis Cells for Large-scale Hydrogen Production at Intermediate Temperatures

- **Interact with the HTE Working Group and 2B (PNNL, Olga Marina)**



Collaboration: HydroGEN HTE Node Utilization

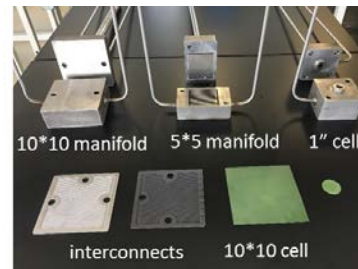
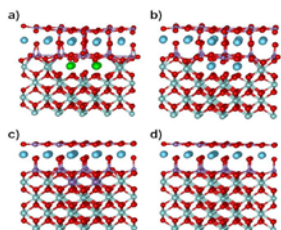
Approach

| Lab | Node | Node PI | NWU | UConn | UTRC |
|------|--|---------------------------------|-----|-------|------|
| NREL | Engineering of Balance of Plant (BOP) for High-Temperature Systems | Ma, Zhiwen Martinek, Janna | | | ✓ |
| LBNL | Metal-Supported SOEC | Tucker, Michael Wang, Ruofan | ✓ | | ✓ |
| INL | Advanced Materials for Elevated Temperature Water Electrolysis | Ding, Dong | ✓ | ✓ | ✓ |
| INL | SOEC Characterization | O'Brien, James Stoots, Carl | ✓ | | |
| NREL | High-Throughput Experimental Thin Film Combinatorial Capabilities | Zakutayev, Andriy | | ✓ | |

Computation

Material Synthesis

Characterization



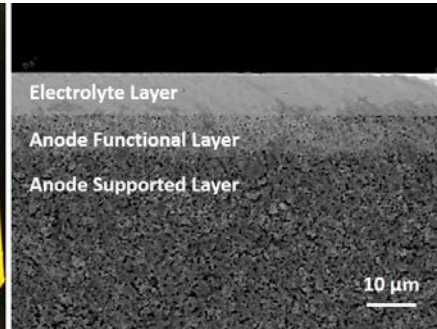
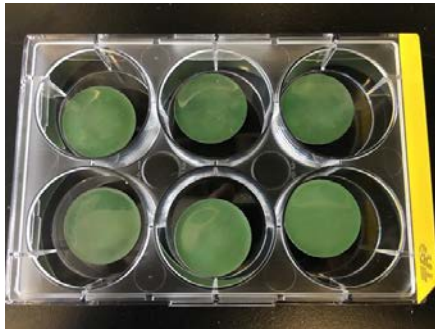


Project Vision

Study degradation mechanisms in solid oxide electrolysis cells operated at high current density using accelerated testing closely coupled with theory and modeling.

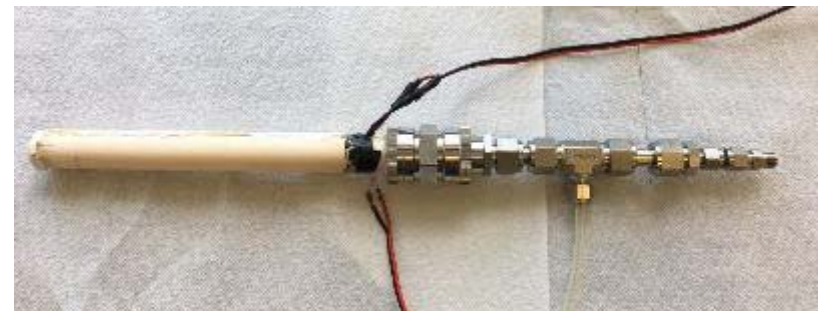
- INL prepared materials for NWU testing
- Testing at NWU and INL
- Student from NMU supporting testing at INL (Summer, 2018)

Fabricated YSZ-based SOECs as requested (ongoing)



Anode-supported cell with YSZ electrolyte

Metal-supported cell mounted on ceramic tube with current collector



Button cell test fixture prepared



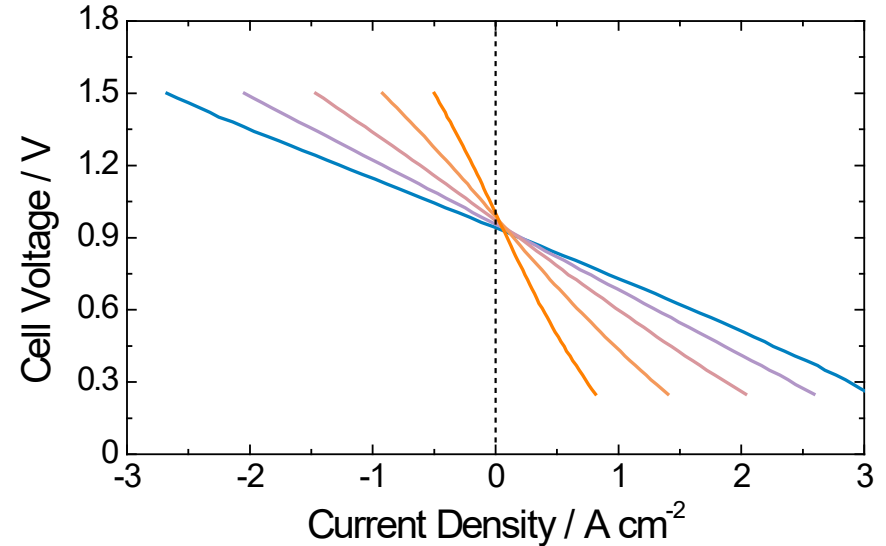
NWU Accomplishments Test Results at NWU

See Poster PD153

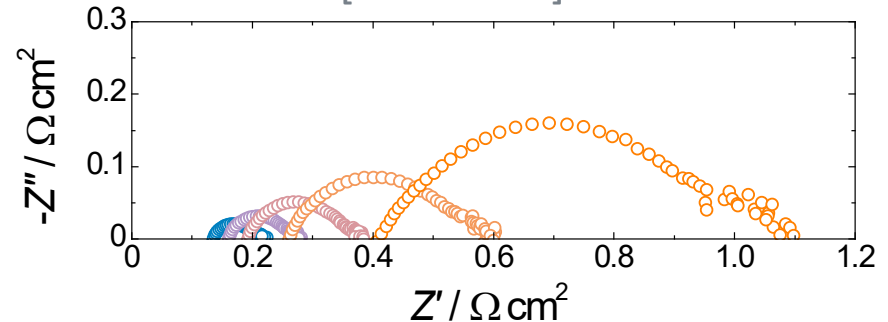
Collaboration

- Cell performance has been fully characterized
- Cell current density at expected electrolysis voltage, and cell resistance, are as desired
- The performance results meet the second part of
 - Milestone 1.1: < 1.3 V at 1.0 A/cm² (800°C, in air and humidified hydrogen)

[I-V curves with different temperatures]



[EIS results]



- The SOEC indicates the cell voltage of ~ 1.15 V at 1 A cm⁻² (800 °C).



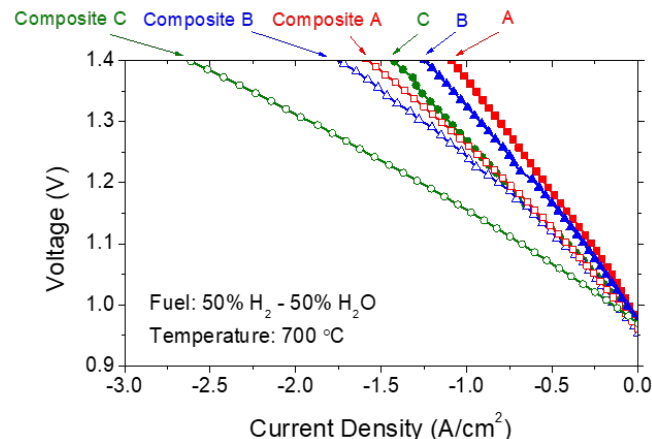
Technology Innovation & Accelerated Testing

- M-SOEC cell testing underway at LBNL
 - ✓ Information exchange is on-going between Node and PI
 - ✓ Consistent test matrix has been established
 - ✓ First time testing of well-developed metal supported solid-oxide cells in electrolysis mode
- Optimizing cell catalysts for SOEC
- Technical manuscript is being prepared by Tucker, et al.

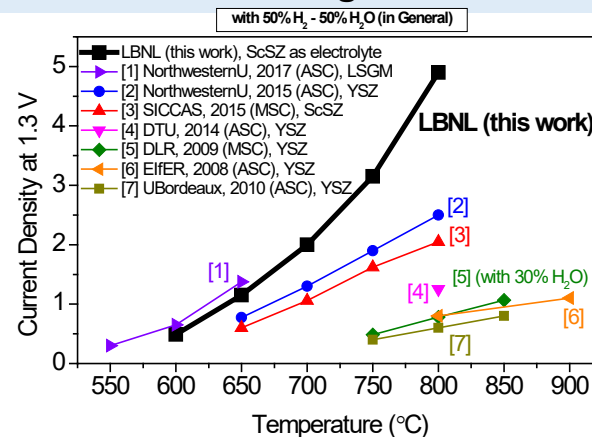
Collaboration

Zirconia oxide-conducting

High Performance of OER catalysts



Comparison with other oxygen-conducting SOECs





UTRC Project Accomplishments

See Poster PD154

Collaboration

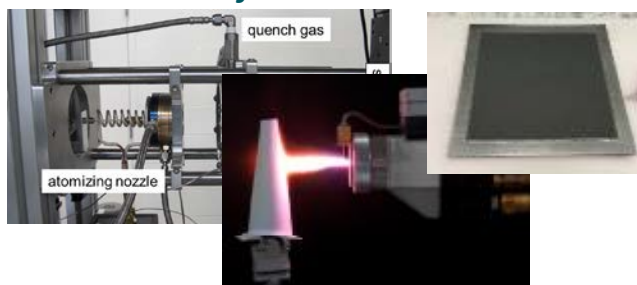
PI: Tianli Zhu

Partner: University of Connecticut

Project Vision

Develop low cost, scalable fabrication of metal-supported cell, and further material optimization for an efficient & durable proton conducting SOEC (p-SOEC) at 550-650°C.

UTRC low cost cell fabrication

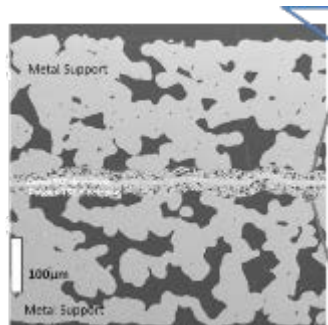


Metal supported p-SOEC

UTRC stack & system testing (up to 10kWe)



*Reactive Spray Deposition (<math><10\mu\text{m}</math>)
Suspension plasma spray (<math><15\mu\text{m}</math>)*



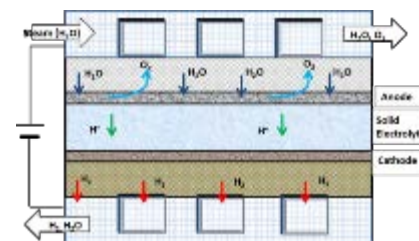
*Co-sintered metal cell
LBNL*



*Metal alloy selection:
oxidation rig at LBNL*



*Material optimization:
high throughput testing
stands at INL*



*Electrochemical modeling
at NREL*



**United Technologies
Research Center**

This material is based upon work supported by the U.S. Department of Energy's Office of Energy Efficiency and Renewable Energy (EERE) under the Fuel Cells Technology Office (FCTO) Award Number DE-EE00080808

HydroGEN: Advanced Water Splitting Materials

Partnerships: UCONN and Electrochem



UTRC Project Accomplishments

See Poster PD154

Collaboration

PI: Tianli Zhu

Partner: University of Connecticut

- **UTRC:** Hydrogen electrode deposition on porous metal support is complete
- **UTRC:** Electrolyte deposition by Reactive Spray Deposition Technique (RSDT) and Suspension Plasma Spray (RSD) in progress, focus on optimization
- **UTRC:** Open circuit voltage (OCV) results of the electrolyte indicate density of electrolyte needs improvement
- **LBL:** Co-Sintering doped BZY compatible stainless steel support material
- **LBL:** Selection of interconnect alloy and coating through oxidation study
 - Severe oxidation of uncoated SS metal alloy in 50% steam-air (650 °C)
 - Minimal oxidation observed with coated metals in 50% steam-air (650 °C)
- **INL:** Electrolyte and electrode material optimization
 - Tested BaZrCeY proton-conducting based ceramics in high steam
 - Evaluated the conductivities of different electrolytes and their stability in high water content at operating temperatures
- **NREL:** Thermal/Electrochemical modeling of Solid Oxide Electrolysis Cell (SOEC)
 - Extended the fundamental mathematical description of the cell electrochemical process for the proton-conducting SOEC.



Project Vision

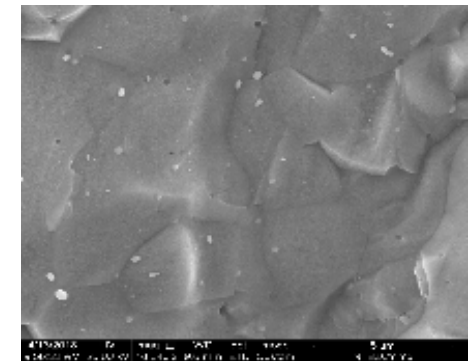
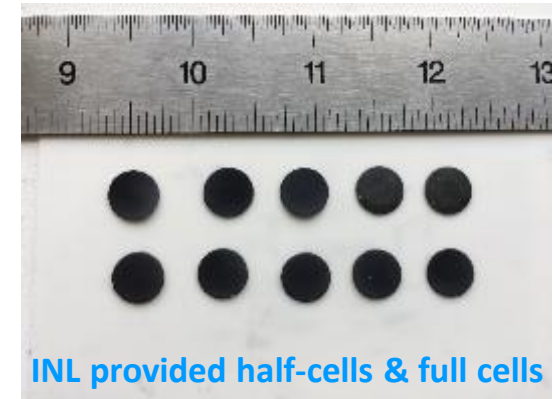
Develop innovative, cost effective and efficient proton-conducting solid oxide electrolysis cells (H-SOECs) for large-scale hydrogen production at intermediate temperatures (600-800C).

Development of electrolyte densification technique and determination of corresponding ionic conductivities.

- INL provided proton-conducting half-cells and full cells (1 cm diameter) for SOEC testing and characterization.
 - The measured conductivity and the thickness (~20 μm) of dense electrolyte at elevated temperature
 - ✓ Electrical conductivity >0.01 S/cm below 700°C
 - ⇒ Meets Program Milestone
 - INL will provide large size full cells (1.3 cm or 2.5 cm diameter) and electrode materials for SOEC testing

Anode microstructural modification for performance improvement and develop mechanisms for polarization losses.

- Uconn made electrode and electrolyte materials
- Tape cast and sintered multi-layer laminated
- ✓ Full densification below 1400°C
 - ⇒ Meets Program Milestone
- TEM analysis at PNNL showed no grain boundaries or secondary phases



Dense electrolyte at 1350°C



Engagement with 2B Benchmarking Team

Olga Marina, PNNL

Collaboration

- Completed preliminary assessment of the EMN Node capabilities
- Developed HTE Test Framework and HTE Benchmark Questionnaire.
 - Drafts sent to node experts for critical review.
 - The framework is categorized into Materials Property Testing, Cell Testing and Stack Testing.
 - Looking at INL to support SOEC stack testing
 - Assembled a list of experts who should be participating in this survey.
 - Both test matrix and questionnaire have been sent out to community for inputs.



- Leverage HydroGEN Nodes at the labs to enable successful Go/No-Go of Phase 1 projects
 - New SOEC electrodes, new $O^=$ and H^+ conduction electrolytes, new robust metal-support materials and coatings
 - Increased durability and lifetime at optimum cell temperatures
 - Increased production rates
- Enable research in Phase 2 work for some projects and enable new seedling projects
- Work with the 2B team and HTE working group to establish testing protocols and benchmarks
- Utilize data hub for increased communication, collaboration, generalized learnings, and making digital data public
- Leverage related projects, particularly projects of similar objectives and scope (e.g., Strategic Partnerships Projects)



Summary

- Supporting 3 FOA projects with 5 nodes and 7 PIs
 - Synthesis, benchmarking, modeling, characterization
- Working closely with the project participants to advance knowledge and utilize capabilities and the data hub
- Projects demonstrate improvements in durable, less expensive materials with high performance
- Future work will include continuing to enable the projects technical progress and develop & utilize lab core capabilities

Acknowledgements



Energy Materials Network
U.S. Department of Energy



HydroGEN
Advanced Water Splitting Materials

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Andriy Zakutayev

Node Experts

Dong Ding
Michael Tucker
Zhiwen Ma
James O'Brien
Carl Stoots
Andriy Zakutayev

HTE Project Leads

Scott Barnet
Tianli Zhu
Prabhakar Singh

Research Teams

UConn



**United Technologies
Research Center**

Northwestern



Pacific Northwest
NATIONAL LABORATORY

**ElectroChem
Ventures**





Technical Backup Slides

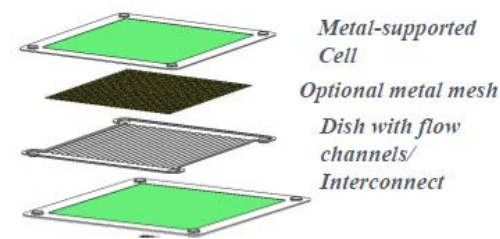


UTRC Project Accomplishments

Testing of materials (LBNL Node)

See Poster PD154

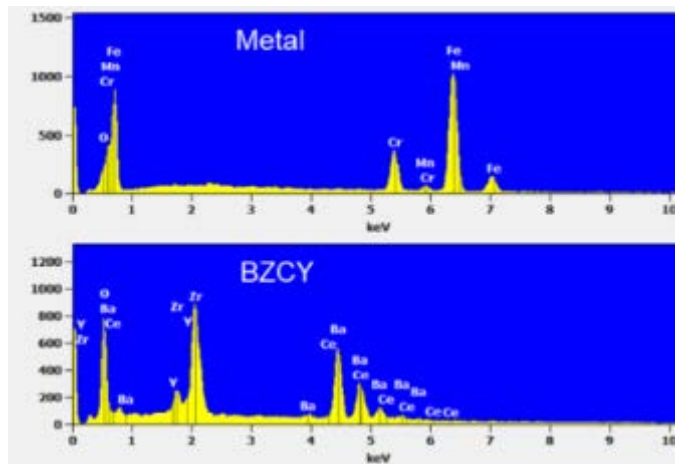
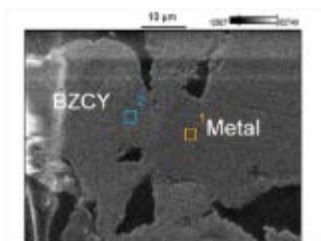
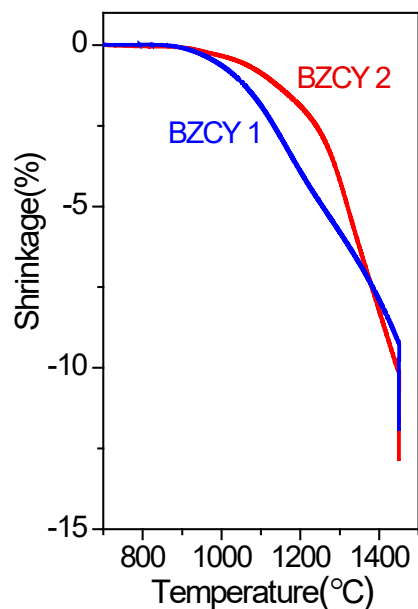
- Testing oxidation rates of various metals and metals coating <math>< 650^{\circ}\text{C}</math>
- BaZrCeY proton-conducting based ceramics in high steam



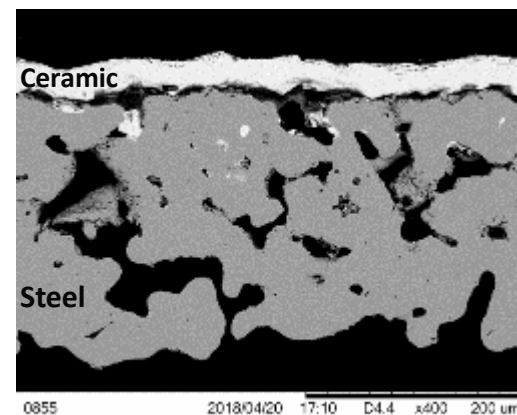
Assessing compatibility with stainless steel and processing conditions

Inter-diffusion avoided

Sintering in H_2 atmosphere



Stainless steel and BZCY-based co-sintered in H_2 atmosphere

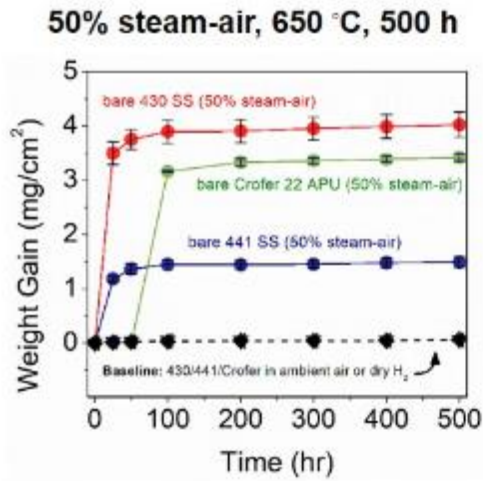


See poster for current project test outcomes

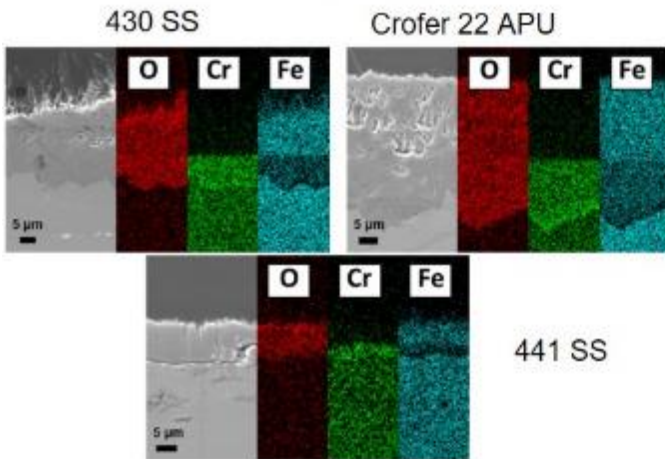
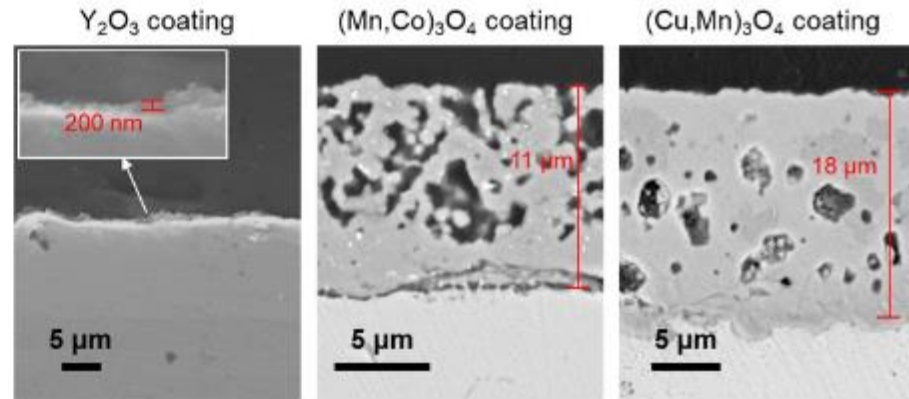


UTRC Accomplishments: Selection of Interconnector Alloy and Coating through Oxidation Study in SOEC Conditions (LBNL)

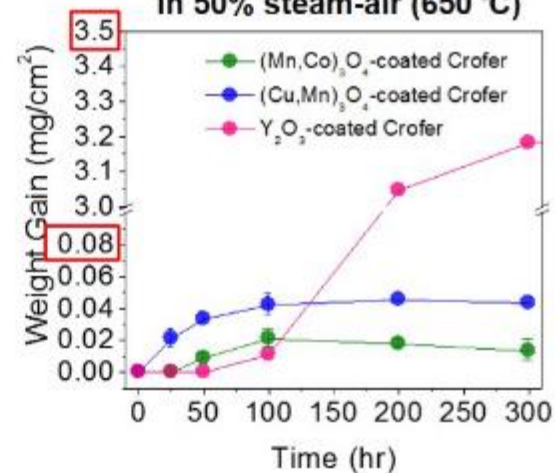
- Severe oxidation of uncoated SS metal alloy in 50% steam-air (650 °C)
- Minimal oxidation observed with coated metals in 50% steam-air (650 °C)



Potential protective coating



Minimal oxidation for MnCo- and CuMn-coated metals in 50% steam-air (650 °C)





UTRC Accomplishments

Materials synthesis and testing (INL Node)

See Poster PD154

- Evaluated the conductivities of different electrolytes and their stability in high water content at operating temperatures
- Helped test 1" metal-supported electrolysis cells

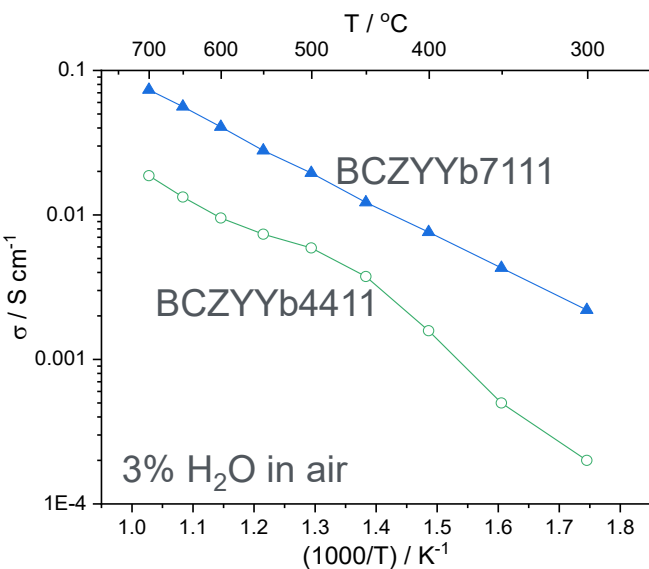


Figure 1a: The temperature dependence of electrical conductivities of BCZYYb7111 and BCZYYb4411 pellets measured by four-probe method in wet air.

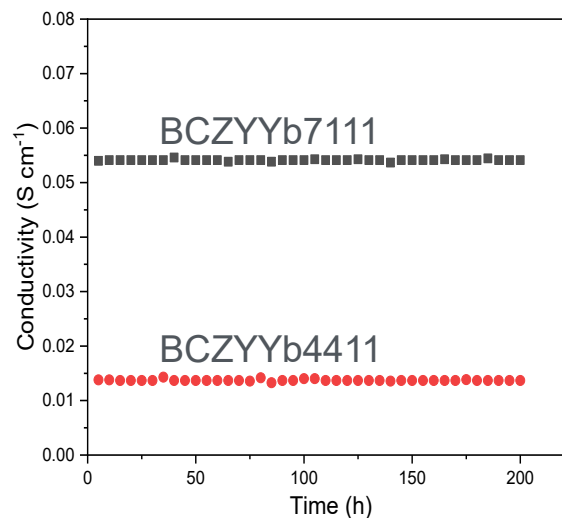
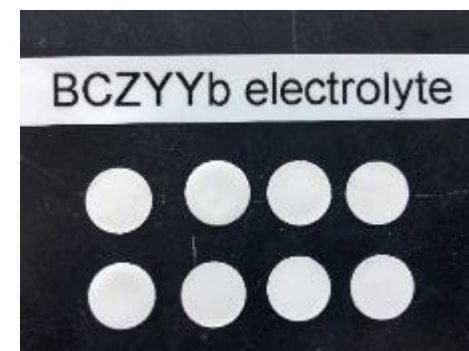


Figure 1b: The stability test of two proton conducting electrolytes in 50% H₂O (carried by air) at 650 °C.



1" cell on testing fixture



UConn Project Accomplishments

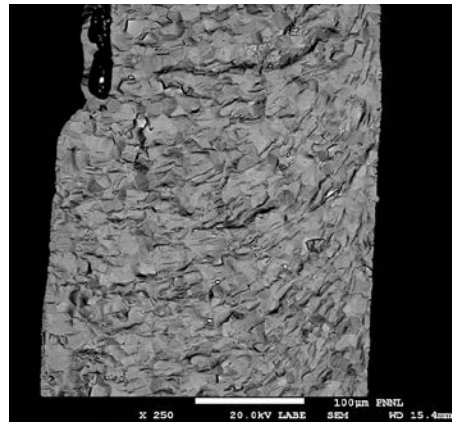
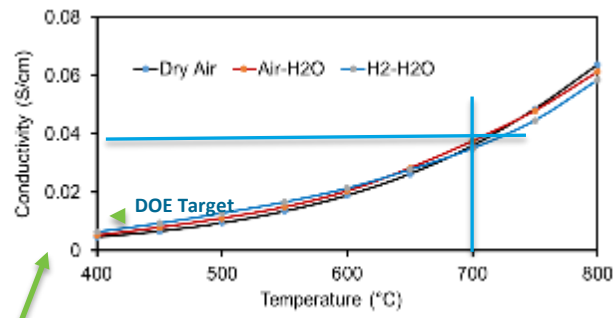
Conductivity Measurements and Densification Studies

See Poster PD 152

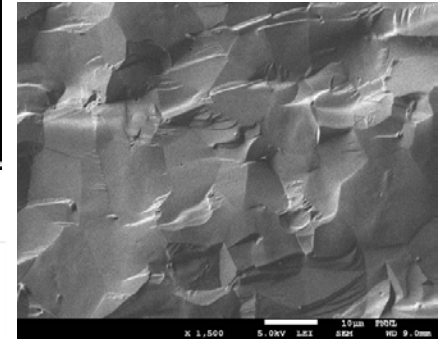
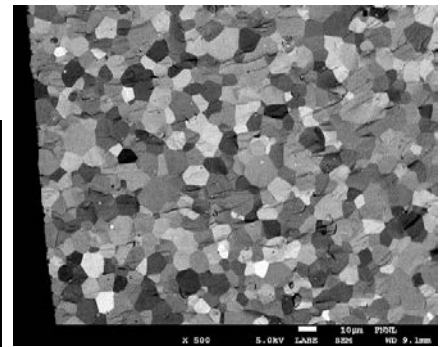


Four-probe Conductivity Test setup Assembled at UConn

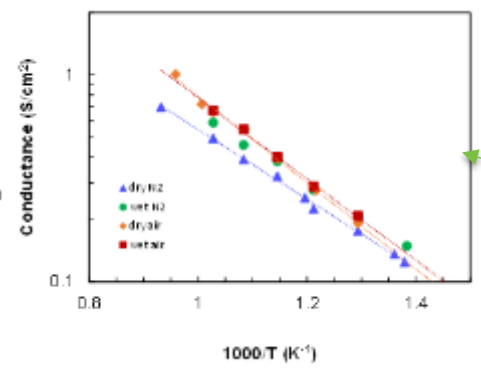
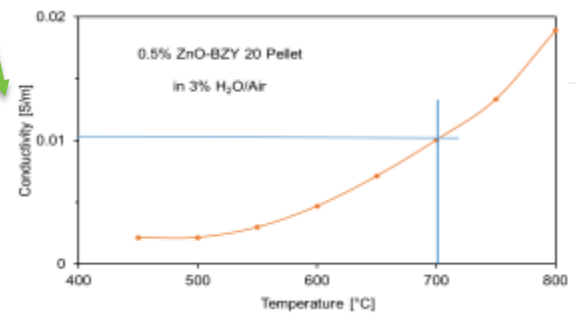
1%ZnO BZCY-Yb sintered at 1350°C for 6 h



Fractured dense sample



0.5%ZnO BZY sintered at 1450°C for 6 h



Analyzed by PNNL

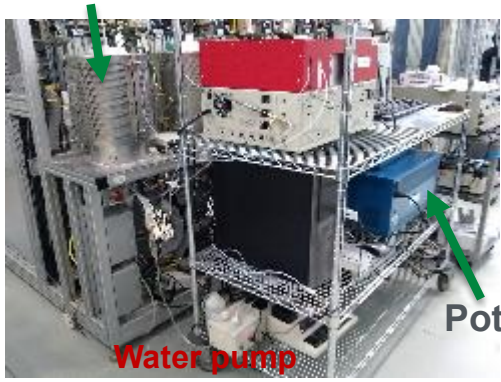
- Measurements obtained at UConn and PNNL show that the conductivity of dense BZCY-Yb (0.04 s/cm) and BZY (0.01 s/cm) electrolyte meets with the DOE Target of Milestone 3-1 (0.01 s/cm at 700°C).
- Experiments and characterization performed at UConn and PNNL show full densification at 1350°C after sintering in air. The bulk phase remains free of GB precipitates



UConn Project Accomplishments H-SOEC Cell Preparation & Testing

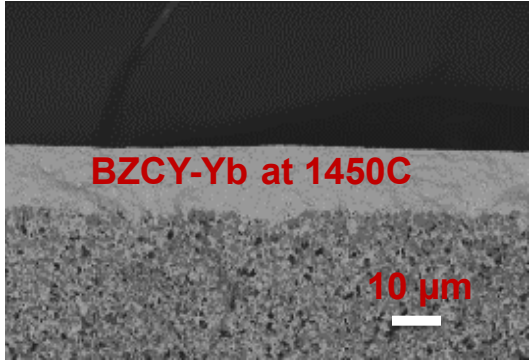
See Poster PD 152

Vertical furnaces

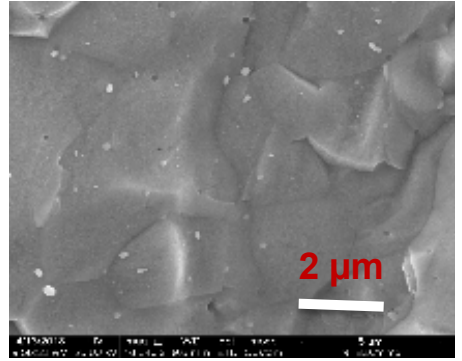


Gas chromatograph

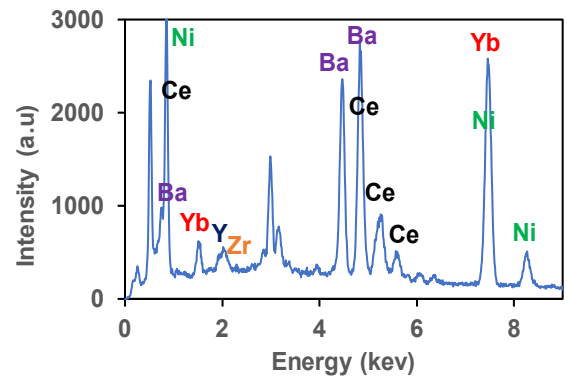
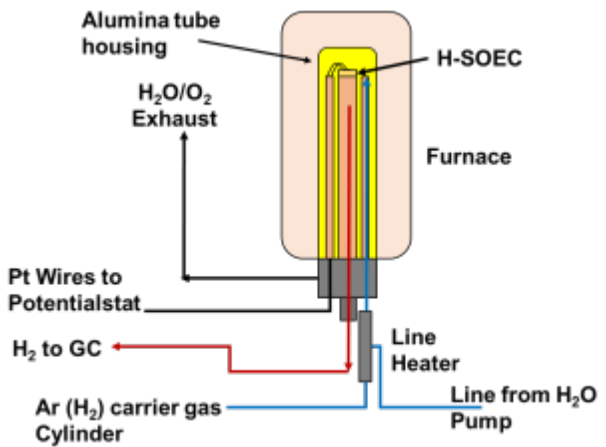
Prepared at INL



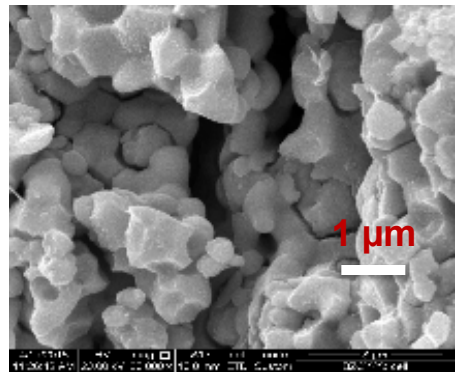
~15 μm thickness of electrolyte



Dense electrolyte at 1350°C



EDS spectra of electrode



Porous electrode

UConn and INL have fabricated and characterized H-SOEC cells with thin dense electrolyte (15-40 μm). Above SEM images shows a typical Ni-BZCY-Yb||BZCY-Yb half cell. Ni-BZY||BZY||LSCF-BZY cell are being fabricated. Both H-SOEC cells will be tested in Q3 and Q4.