



# **RU-EMN – Best-in-class Platinum Group Metal-free Catalyst Integrated Tandem Junction PEC Water Splitting Devices**

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Lawrence Livermore National Laboratory







### **Partners**

Rutgers PIs: E. Garfunkel & G. C. Dismukes, coPI: M. Greenblatt NREL: T. Deutsch, D. Friedman & A. Zakutayev

## Vision

Identify the best technical approaches to fabricate both High Performance (HP) & High Value (HV) PGM-free PECs:

-Using HP PEC NREL-proprietary tandem III-V photovoltaic

-Using HV PEC NREL nitrides & Rutgers oxynitrides in tandem

-Using Rutgers-proprietary electro-catalysts for  $\mathrm{O}_2$  - and  $\mathrm{H}_2$  - evolution

### Impact

- Reveal performance limits of two limiting PV configurations (HP vs HV) using best PGM-free Cats.
- Identify & solve interfacing problems.
- this amount does not cover support for HydroGEN resources leveraged by the project (which is provided separately by DOE)

Award #	EE0008083	
Start Date Yr 1 End Date Project End Date	09/01/17 08/31/18 TBD	
Total DOE Share Total Cost Share Year 1 DOE Funding*	\$750,000 \$83,334 \$250,000	
Bioinspired HER catalyst	Bioinspired OER catalyst	
HER Catalyst OER Catalyst $H_2$ $2H^+$ $V_2O_2 + 2H^+$ $H_2$ $H_2$ $H_2$ $V_2O_2 + 2H^+$ $H_2$ $H_$		
PV IIN M <sup>3</sup> 21	H⁺ + 2e⁻	
Inter	facial layer	

# Approach- Summary



#### **Project Motivation**

**High-performance Photovoltaics:** State of the art material competing with NREL HP III-V tandems (STH 16.7%) is: Multi-junction Silicon (STH 3.9%).

**High-value (HV)** – emerging photoabsorbers (level 3 readiness): NREL-developed  $ZnSnN_{2}$ , coupled with SrNbO<sub>2</sub>N are potential efficient and low-cost with high potential impact on global energy economy. Early development hurdles.

**TiN vs TiO<sub>2</sub> -** Rutgers developed thin-film passivation and protection layer.

**LiCoO<sub>2</sub> OER catalyst** – Rutgers developed alkali cat. w. performance on par with PGM-catalysts.

 $Ni_5P_4$  HER catalyst - Rutgers developed acid & alkali cat. w. performance on par with Pt.

### Barriers

## Replace PGMs catalysts with non-noble metal catalysts without compromising efficiency

→ Apply active thin-film of RU-eCats which are on par with PGMs catalysts.

## Reduce light losses in Cat/Absorber interface (5 nm $MoS_2/MoO_x \sim 25\%$ current loss)

→ Uses optimum thickness of catalyst with protection/antireflective layers (6.8 nm NiP<sub>2</sub>/4 nm TiN ~ no current loss).

#### Photoelectrodes corrosion in alkaline electrolyte

- → Use proper protection layer with alkaline stable RU-eCats.
- → Use compatible alkaline buffers.

### Key Impact – year 1

Metric	State of the Art	Expected Advance
HP - STH	Non-PGM STH 10%	Match or beat using non-PGM RU-eCats.
HV - Parameter 1 (BiVO₄ benchmark)	J <sub>ph</sub>   <sub>E=1.23V RHE</sub> = 1.5mA/cm <sup>2</sup> , Stable >90% for 1h, material-cost	Improve two or more metrics.

### Partnerships

HydroGEN partners:

#### **Daniel Friedman (NREL)**

III-V Semiconductor Epi-Structure Device Design and Fab MOCVD GaInP/GaAs growth & engineering capabilities

#### Todd Deutsch (NREL)

On Sun Characterization of Bulk and Interfaces *Solar testing and benchmarking capabilities* 

#### Andriy Zakutayev (NREL)

High-Throughput Thin-film Combinatorial Capabilities *Multi-source and reactive sputtering capabilities* 

# Approach- Innovation year-1 Thrust

Design of catalyst/photoabsorber interfaces using PGM-free RU-eCats with state-of-the-art performance to NREL proprietary photoabsorbers.



# Designing RU-protection layer to stabilize photoabsorber interface and RU-eCat adhesion.



#### Rutgers' proprietary HER catalyst

1766-2016



#### Budget period 1 summary:

Design, fabricate, characterize, and test first example of Rutgers protection (TiN) and eCat ( $Ni_5P_4$ , and  $LiCoO_2$ )layers on NREL HP (np-GalnP/n<sup>+</sup>-GaAs) and NREL HV (ZnSnN<sub>2</sub>).

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TEA of Type 4, tracking concentrator array PEC Type 4



 This project supports the development of non-PGM and scalable electrocatalysts for PEC water splitting. It will lower H2 costs by increasing STH efficiency, decreasing production costs using lower cost materials, and increase lifetime. These are critical to achieving the DOE target of hydrogen for <\$2/kg.

Pinaud et al. Energy Environ. Sci. 6, 1983 (2013)

#### HydroGEN consortium

Rutgers utilizes photoabsorbers and characterization facilities from three NREL nodes: 1) MOCVD GainP/GaAs growth & engineering capabilities (Friedman)[2) Solar testing and benchmarking capabilities (Deutsch); 3) Multi-source and reactive sputtering capabilities (Zakutayev).

This team develops innovative approaches to making interfaces with combined protection/antireflective layers. Bi- and tri-weekly meetings coordinating efforts have been undertaken.

Rutgers offers unique Nion STEM with ultra-high resolution EELS vibrational spectroscopy, Rutherford (and medium ion) backscattering spectroscopy and high resolution He-ion microscopy capabilities through collaboration. HydroGEN: Advanced Water Splitting Materials





Milestones	Antici pated Qtr	Milestone Description (Go/No-Go Decision Criteria)	Milestone Verification Process (What, How, Who, Where)	Status
#1	Q1	Determine which substrate has the best performance/cost while optimizing the loading of the LiCoO <sub>2</sub> catalyst	E vs OER < 500mV, stable for 24h, loadings ≥ RDE pellets, Electrochemical test in flooded cells, Rutgers	complete
#2	Q2	Device test under illumination showing: V_{OC} > 1.85 V, J_{SC} > 10 mA/cm^2, and $\eta_{eff} > 25\%$	Photoeletrochemical benchmarking, NREL	complete
Go/No-Go Crtieria for Thrust 1	Q4	Evaluation of high performance PEC device with renewable catalysts	STH> 10%, stability > 24h, and/or J <sub>ph</sub> (at 0 V vs. RHE) > 10mA/cm <sup>2</sup> , Photoelectrochemical benchmarking, Rutgers and NREL	50%
#3	Q3	Fabricate and report optimized ZnSnN <sub>2</sub> device	Optimize reflector - max reflectivity, Balancing the highest possible carrier mobility with closeness to 1.1eV bandgap, Rutgers	25%
Go/No-go Criteria for Thrust 2	Q4	LiCoO <sub>2</sub> /ZnSnN <sub>2</sub> photoanode fabricated and benchmarked against BiVO <sub>4</sub>	Must outcompete on (at least 2): J <sub>ph</sub> (1.5mA/cm <sup>2</sup> at 1.23 V vs. RHE), stability >90% after 1h, and material cost, Rutgers, NREL	2%, started

All milestones are completed or on-track for on-time completion

## **Accomplishments – Overview**





-0.8 -0.6 -0.4 -0.2 0.0 0.2 0.4 0.6 0.8 1.0 Potential (V vs. RHE)

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## Accomplishments: Electrochemical (dark) anode

 Milestone 1 (achieved Q1) realize a stable non-PGM alkali LiCoO<sub>2</sub> electroanode (on par with published cubic-LiCoO<sub>2</sub> benchmark) with at least long-term stability (>1 day) required in milestone #1.



- Technical target: Overpotential, η ≤ 430 mV at 10 mA/cm2geo in pH 14 electrolyte
- Raman spectroscopy signals indicates localized spinel structure. XPS confirm only  $Co^{3+}$  precluding unstable  $Co_3O_4$ formation (Data not shown).

TGERS

## Accomplishments: Electrochemical (dark) anode





- Chronoamperometric analysis for OER confirms >12 days of stability.
- Surface area normalized CVs shows same intrinsic activity. Minor decreased activity (during durability test) is due to loss of surface area attributed to particle detachment.
- Achieved Task 3 "Optimize LiCoO<sub>2</sub> electrode loading" and milestone 1.

## Accomplishments: Successful integration of Ni<sub>5</sub>P<sub>4</sub>/TiN on p-type GaInP

Transferring optimized synthesis condition for proprietary NREL PV - Synthesis development of catalyst/protection layers, Tech slide 1, 2



Helium Ion Microscope (HIM) image shows uniform catalyst coverage



Top layer XPS confirms successful formation of nickel phosphide catalyst without atomic diffusion of In and Ga

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# Accomplishments: Successful integration of Ni<sub>5</sub>P<sub>4</sub>/TiN on p-type GaInP



- Ni<sub>5</sub>P<sub>4</sub>/TiN layers on p-GalnP are on par with PtRu/p-GalnP. ( $V_{onset}$ , and  $J_{sat}$ )
- Durability test demonstrates 89 h stability of Ni<sub>5</sub>P<sub>4</sub>/TiN/p-GaInP device in acid
- Achieved Task 2 "Optimize catalyst/photoabsorber (Nickel Phosphide/GalnP) interface"

## Accomplishments: Benchmarking Ni<sub>5</sub>P<sub>4</sub>/TiN/np-GaInP at NREL\_Deutsch



- PEC measurement were verified at the **NREL node**.
- Introducing buried junction np-GaInP increases  $V_{onset}$  (from 0.4 V to 1.0 V vs. RHE).
- Test proves the performance of Ni<sub>5</sub>P<sub>4</sub>/TiN/np-GaInP is on par with benchmark PtRu/TiN/np-GaInP.

## Accomplishments: NREL fabrication of GaInP/GaAs high-efficiency PV

Milestone #2 (**achieved Q2**) fabricate a high-efficiency PV cell for photoabsorbers:  $V_{OC} > 1.85$  V,  $J_{sc} > 10$  mA/cm<sup>2</sup>, and  $\eta_{eff} > 25\%$ 

- NREL\_Friedman node provided an upright tandem photoabsorber (GaInP/GaAs) that exceeds required properties. This photoabsorber will be coupled with RU-eCat and protection layer for unassisted water splitting.
- Achieved Milestone #2



## Accomplishments: NREL fabrication of thin-film ZnSnN<sub>2</sub>



### Milestone #3: Fabricate ZnSnN<sub>2</sub>thin film:



- NREL\_Zakutayev node provided synthesis (confirmed by XRD, XRF) and investigated post-annealing conditions for ZnSnN<sub>2</sub> that can generate proper conductivity for photoanode applications.
- Task9 On-track to achieve milestone #3, "Fabricate ZnSnN<sub>2</sub> photoabsorber "







## **Collaboration: Multiple interactions** with HydroGEN nodes



NREL: III-V semiconductor & semiconductor NREL: I characterization nodes experim

NREL: High throughput experimental thin-film node

Date	Meetings w/	Presenter	Outcomes
11/13/17	NREL NREL	N/A	HydroGEN kick-off meeting – project initiation
1/26/18	NREL	Rutgers	Understanding native oxides on GaInP for improving TiN adhesion <b>(Task 2)</b>
2/23/18	NREL	Rutgers	Optimize fabrication conditions for Ni <sub>5</sub> P <sub>4</sub> thin-film Successful fabrication of Ni <sub>5</sub> P <sub>4</sub> /TiN on p-GaInP <b>(Task 2)</b>
3/16/18	NREL	Rutgers	Characterization of Ni <sub>5</sub> P <sub>4</sub> /TiN/p-GaInP and npGaInP Discussion for benchmarking test (IPCE) <b>(Task 2,4, and Go-No/Go1)</b>
3/22/18	NREL	NREL	Optimization of fabrication condition for ZnSnN <sub>2</sub> (Task 9, Milestone 3)
4/13/18	NREL	Rutgers NREL	Prove Ni <sub>5</sub> P <sub>4</sub> is on par with PtRu on GaInP (Task 2, Go-No/Go1) Fabrication of np-GaInP/np-GaAs tandem PV <b>(Task 1, Milestone 2)</b>

# **Collaboration: Data sharing within HydroGEN**

### Data sharing:

Verified data uploaded to HydroGEN site and Datahub. Quarterly reports with input from both RU and nodes shared in HydroGEN site and Datahub. This has allowed the continues sharing of progress, samples needed, problems encountered, verification of experimental test procedures. The continuous sharing of results have been critical to the fast paced progress on this project. Sharing to the water splitting R&D community will be realized through publication as well as opening access to the verified raw data in the datahub after the project period. Data shared in publications will be opened for sharing at the time of publication.

### HydroGEN community interaction:

Pre-proposal	Means of communication	Node/point of Contact(s)	
submission	In-person (ECS conference)	James Young NREL	
	Telecon	Todd Deutsch NREL	
Post-award	Means of communication	Node/point of Contact(s)	
	Telecon	Todd Deutsch NREL	
	Telecon	Michael Ulsch NREL	
	In-person (Kick-off meeting)	Todd Deutsch NREL, Michael Ulsch NREL, Daniel Friedman NREL, Aditya Mohite LANL, PEC workgroup	
	Biweekly Telecon for progress reports	Todd Deutsch NREL, Daniel Friedman NREL	
	Telecon for progress report	Andriy Zakutayev NREL	
	In-person (Seattle ECS meeting)	PEC working group	





#### Thrust #1, High Performance PEC

**Scope:** Build complete PEC device based on NREL's incrementally advanced III/V-photoabsorbers & next generation RU-eCats.

#### - State-of-the-art PECs operate in acid electrolyte, where PGM-free anodes are not stable. Extending operation to alkali presents the first attempt to achieve this keyproperty for fully PGM-free operation.

Y2: Fabricate protection/catalyst layers for alkali operation. Improve performance under diurnal cycle (low current regime) where recombination & photocorrosion may increase. *Outcome: stable operation for > 24h in alkali, stable performance under diurnal cycle.* 

- Evaluating the long-term stability of the optimized device is crucial for an accurate H2A and LCA analysis. Costbenefit analysis will quantify the impact of the achieved improvements over state-of-the-art PEC devices to achieve target of < 2/kg H<sub>2</sub>.

**Y3:** Extended stability tests under solar concentration. Perform H2A and LCA modelling employing HydroGEN node. *Outcome: Long-term stability, effect of electrolyte, effects of PGM-free catalysts on H*<sub>2</sub> *costs.* 

#### Thrust #2, High Value PEC

- State-of-the-art HV PECs show limited performance and stability. Thin-film fabrication of oxynitride and double inorganic perovskite salt (DIPS) based electrode represent level 3 (early stage development) photoabsorbers with low-cost potential.

Y2-Y3: Development of level 3 photoabsorbers with HydroGEN nodes. Build complete PEC device based on ZnSnN<sub>2</sub> photoabsorber/RU-eCat electrodes. Development of oxynitride and DIPS photoanodes collaboration with NREL High-Throughput Experimental Thin-film Combinatorial Capabilities (Zakutayev) and High-Throughput Processing node (Ulsh).

Y2: Develop buried junction-type oxynitride thin-film. Optimize protection/catalyst layers for oxy-nitride photocathode and DIPS photoanodes. *Outcome: novel buried junction-oxynitride photoabsorber, stable operation in alkali, DIPS thin-film photoabsorber/photoelectrode.* 

**Y3:** Stability tests under solar concentration, plus H2A and LCA modelling employing HydroGEN node.

- Direct comparison of cost-benefit analysis for HV and HP devices from same team for ultimately achieving DOE targets.

# Project Summary



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# Thank you for your attention













Martha



Anders



Shinjae

Hwang

Garfunkel

Dismukes

Greenblatt

Laursen



# **Technical Back-Up Slides**



### Using Si-PV model for catalyst synthesis optimization



A thick layer is on purpose for getting X-ray diffraction

Phase-pure crystalline Ni<sub>5</sub>P<sub>4</sub> can be obtained with choice of temperature and time.







XPS depth profile shows no significant atomic diffusion was detected