



*Driving Innovation through Chemistry*

# High-density Hydrogen Storage in Space-filling Polyhedral Sorbents

Project ID: st150

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June 13, 2018

# Overview

## Timelines

Start: April 9, 2018

End: October 8, 2018

## Budget

Phase I SBIR

Total project budget: \$143,783

Federal share: \$143,783

## Barriers

A. System weight and volume

J. Thermal management

E. Charging/discharging rates

## Partners

NEI Corporation

# Relevance

## Objective

Increase the packing density of adsorbent materials *using a scalable process* without compromising their hydrogen storage properties or adversely affecting storage temperatures and pressures, gravimetric capacity, or reversibility



Parameter	DOE 2020 Technical Target	Project Impact
System volumetric capacity	0.030 kg H <sub>2</sub> /liter system	Increase >50% (materials dependent)
System gravimetric capacity	0.045 kg H <sub>2</sub> /kg system	No change
Min/max delivery temperature	-40 °C / +85 °C	No change
Min delivery pressure	5 bar	No change
Minimum full flow rate	0.02 (g H <sub>2</sub> /s)/kW	No change

# Approach

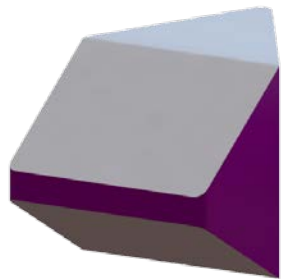
- Demonstrate the formation of space-filling polyhedral microliths from an activated carbon, a metal organic framework material, and a porous polymer
- Demonstrate the packing of space-filling polyhedral sorbents in a 1-liter container

	<b>Milestone</b>	<b>Date</b>	<b>Performance</b>
1	Formation of space-filling polyhedral shapes from at least two distinct classes of base materials with <5% change in key properties	08/09	Demonstrated for activated carbon
2	High-pressure hydrogen storage in space-filling polyhedral microliths from at least two distinct classes of base materials with <5% change in properties	09/09	Not started
3	Provide samples of space-filling polyhedral sorbents to NREL to validate adsorption measurements	10/09	Activated carbon available today
4	Demonstrate the packing of space-filling polyhedral sorbents in a 1-liter container with >95% of the theoretical packing fraction	10/09	Demonstrated at 250 ml

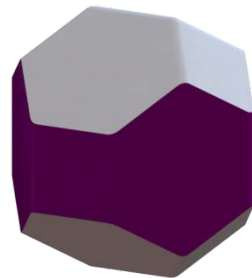
# Accomplishments and Progress: *Shapes*

- Hardware for fabrication of space-filling microliths developed
- Pressing/casting conditions established
  - Carbon: 70 – 140 MPa (10 – 20 kpsi)
  - MOF: 50 MPa (7 kpsi)
  - Polymer: cast
- 6 kg of activated carbon rhombic dodecahedra fabricated

Space-filling polyhedral dies for a high-speed rotary tablet press



Gyrobifastigium



Truncated Octahedron



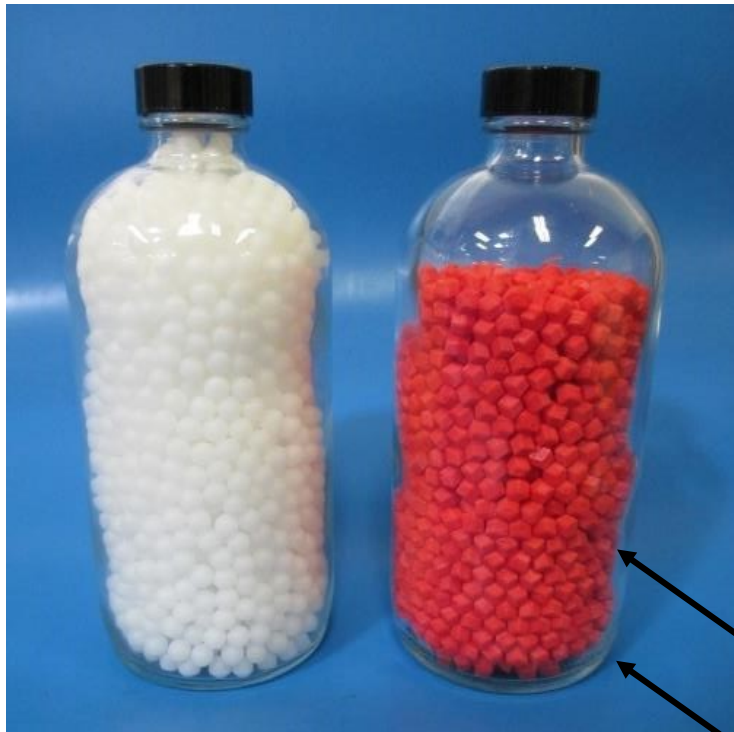
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Truncated octahedron

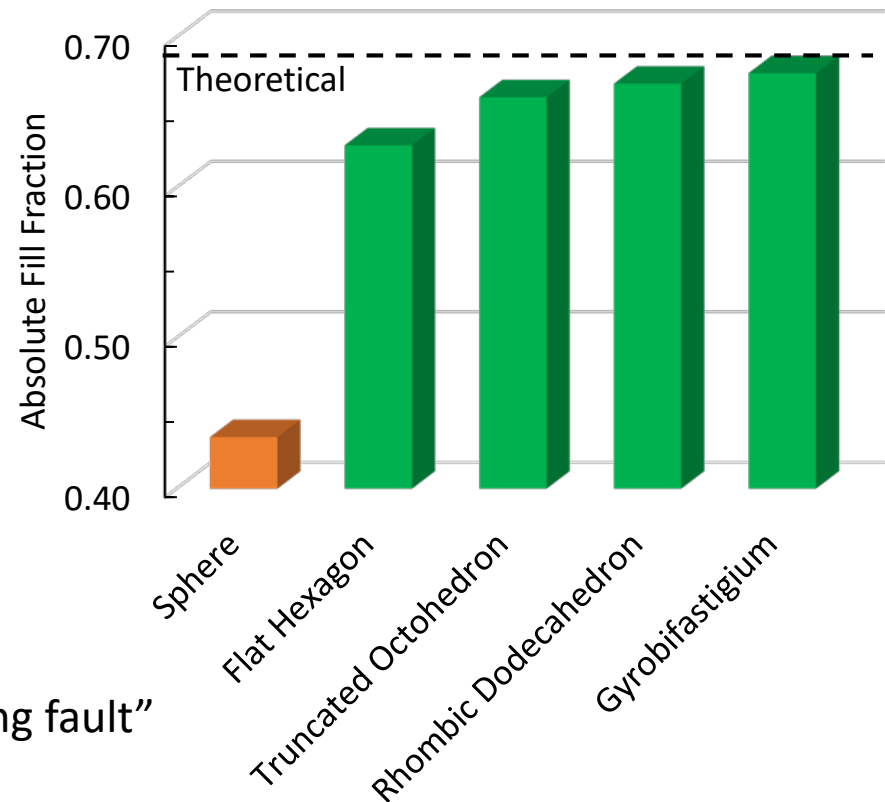
# Accomplishments and Progress: *Packing*

- Demonstrated packing of space-filling microliths in a 250 ml container with >96% of the theoretical packing fraction

Comparable volume of spheres (white, 43% packing) and rhombic dodecahedra (red, 66.9% packing) in 250 ml jars



Packing of space-filling polyhedral microliths in a 250 ml jar



"Stacking fault"

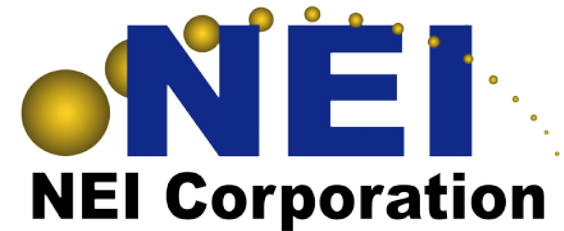
Seed layer

Accomplishments and Progress:  
*Responses to Previous Year Reviewers' Comments*

This project was not reviewed last year

# Collaboration and Coordination

- Sub-awardee: NEI Corporation
- Small business outside the DOE Hydrogen and Fuel Cell Program
- Assist with materials synthesis, shaping, characterization (surface area, density, low-pressure hydrogen uptake), tank filling
- *Critical to the success of the project*





# Remaining Challenges and Barriers

- Within the project
  - Procurement and/or synthesis of MOF-5
  - Synthesis of hyper-crosslinked polymer
  - Additional materials to test?
- To reach DOE hydrogen storage goals
  - Higher hydrogen density storage materials

# Proposed Future Work

- Complete formation of space-filling microliths from MOF-5 (or other MOF), hyper-crosslinked polymer, other? – Milestone due 08/09
- Measure hydrogen uptake at low (1 atm) and high (100 atm) pressures for all samples – Milestone due 09/09
- Optimize filling of 1 liter vessel with space-filling microliths – Milestone due 10/09
- Outreach to industry for feedback – ongoing
- Write successful SBIR Phase II proposal (FY19) – TBD

Any proposed future work is subject to change based on funding levels

# Technology Transfer Activities

- US patent 9,370,765: *Space-filling Polyhedral Sorbents* (issued June 21, 2016)
- Collaboration with InnerProduct Partners (venture capital)
- Industrial outreach begun

# Summary

## Goal

- Demonstrate the formation of space-filling polyhedral microliths from an activated carbon, a metal organic framework material, and a porous polymer and show that their hydrogen storage capacity has not been diminished from that of the base material by more than 5%

## Accomplishments

- Hardware for fabrication of space-filling microliths developed
- Pressing/casting conditions established
- 6 kg of activated carbon rhombic dodecahedra fabricated
- Demonstrated packing of space-filling microliths in a 250 ml container with >96% of the theoretical packing fraction

# Technical Back-up Slides

# Shape Packing

	<b>Advantages</b>	<b>Disadvantages</b>
Spheres	<ul style="list-style-type: none"> <li>• Added through the neck of a standard gas cylinder</li> </ul>	<ul style="list-style-type: none"> <li>• Low packing density</li> <li>• Poor thermal conductivity</li> </ul>
Monoliths	<ul style="list-style-type: none"> <li>• Near 100% packing fraction</li> <li>• Improved thermal conductivity</li> </ul>	<ul style="list-style-type: none"> <li>• Expensive and difficult to form</li> <li>• Low gas flow rate</li> <li>• Custom cylinder</li> </ul>
Space-filling polyhedra	<ul style="list-style-type: none"> <li>• Near 100% packing fraction</li> <li>• Added through the neck of a cylinder</li> <li>• Improved thermal conductivity</li> <li>• High gas flow rate</li> </ul>	

	<b>Monodispersed Spheres</b>	<b>Bimodal (10X) Size Distribution</b>	<b>Monolith</b>	<b>Space-filling Polyhedra</b>
Random packing	0.534	0.726	N/A	0.97
Ideal	0.740	0.933	1.0	1.0

Spheres: S. Yamada, et al., IPSJ Trans. Math. Model. and Appl. **4** (2011) 23

Monoliths: ATMI/Entegris

Space-filling polyhedra: US patent 9,370,765

# Sorbent Agnostic: Literature Observations

	Measurement	Powder	Monolith
<b>Pvdc-based activated carbon (BrightBlack) (1)</b>	Size	200 $\mu\text{m}$	10+ cm
	Surface Area	1030 $\text{m}^2/\text{g}$	1030 $\text{m}^2/\text{g}$
	Density	0.68 $\text{g}/\text{cm}^3$	1.10 $\text{g}/\text{cm}^3$
	Pore Volume	0.4 $\text{cm}^3/\text{g}$	0.4 $\text{cm}^3/\text{g}$
	CO <sub>2</sub> , CH <sub>4</sub> Uptake	Identical	
<b>MOF 5 (2)</b>	Size	0.25 $\mu\text{m}$	0.5 cm
	Surface Area	2762 $\text{m}^2/\text{g}$	2265 $\text{m}^2/\text{g}$
	Density	0.13 $\text{g}/\text{cm}^3$	0.5 $\text{g}/\text{cm}^3$
	Pore Volume	1.37 $\text{cm}^3/\text{g}$	1.12 $\text{cm}^3/\text{g}$
	H <sub>2</sub> Uptake	7.4 g/l	26 g/l
<b>Hyper-crosslinked polymer (3)</b>	Surface Area	>2000 $\text{m}^2/\text{g}$	>2000 $\text{m}^2/\text{g}$
	Pore Volume	0.70 – 0.78 $\text{cm}^3/\text{g}$	0.70 – 0.78 $\text{cm}^3/\text{g}$
	CH <sub>4</sub> Uptake	0.156 g/g and 82.7 $\text{cm}^3/\text{cm}^3$	0.156 g/g and 164 $\text{cm}^3/\text{cm}^3$

(1) ATMI/Entegris

(2) J. J. Purewal, et al., Int. J. H. Energ. **37** (2012) 2723

(3) W. T. Yongqin and F. Svec, Appl. Energ. **183** (2016) 1520