

Spirocyclic Anion Exchange Membranes for improved Performance and Durability

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Project ID: FC178

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Overview

Timeline and Budget

- Project start date: 01/01/18
- Project end date: 09/30/19
- Total project budget: \$300k
 - Total recipient share: \$0K
 - Total federal share: \$300K
 - Total DOE funds spent*: \$170K

Barriers

- Cost
- Performance
- Durability

Partners

- NREL only project
- Multiple interactions across AEM space, leverage significant effort at NREL on related projects

* As of 3/31/19

Relevance/Impact DOE (Preliminary) Milestones for AMFCs*

- Q2, 2017: Develop anion-exchange membranes with an area specific resistance ≤ 0.1 ohm cm², maintained for 500 hours during testing at 600 mA/cm² at T >60 °C.
- **Q4, 2017:** Demonstrate alkaline membrane fuel cell peak power performance > 600 mW/cm² on H_2/O_2 (maximum pressure of 1.5 atma) in MEA with a total loading of $\leq 0.125 \text{ mg}_{PGM}/\text{cm}^2$.
- Q2, 2019: Demonstrate alkaline membrane fuel cell initial performance of 0.6 V at 600 mA/cm² on H₂/air (maximum pressure of 1.5 atma) in MEA a total loading of < 0.1 mg_{PGM}/cm², and less than 10% voltage degradation over 2,000 hour hold test at 600 mA/cm² at T>60 °C. Cell may be reconditioned during test to remove recoverable performance losses.
- Q2, 2020: Develop non-PGM catalysts demonstrating alkaline membrane fuel cell peak power performance > 600 mW/cm² under hydrogen/air (maximum pressure of 1.5 atma) in PGM-free MEA.

Impact/Team Project Goals

Synthesize novel spirocyclic ionomers to membranes and ionomers in Alkaline Membrane Fuel Cells (AMFCs).

Optimize performance to meet DOE AMFC targets

Relevance/Objectives

Alkaline exchange membranes continue to be challenged with cation degradation at high temperature and pH conditions

- State of the art trimethyl ammonium cations exhibit limited durability under fuel cell operating condition
- Research has indicated that cations with a spirocyclic structure have improved durability
 - Higher activation energy for both Hoffman elimination and substitution degradation mechanisms
- Incorporation of spirocyclic ammonium cations into alkaline exchange membranes to improve durability

Quaternary Ammonium	Abbreviation	Half-life [hr]
	ASU	110
N ⁺	DMP	87.3
$\langle \mathbf{N}^{\mathbf{N}} \rangle$	DMPy	37.1
	ASN	28.4
N ⁺⁻	BTMA	4.18

Marino, M. G.; Kreuer, K. D., Alkaline Stability of Quaternary Ammonium Cations for Alkaline Fuel Cell Membranes and Ionic Liquids. *ChemSusChem* **2015**, *8* (3), 513-523.

Approach



D. Strasser, PhD Thesis, Colorado School of Mines

Provide ample material for complete MEA characterization and durability studies

Approach - Milestones

Milestone Name/Description			End Date	Туре
Q1	In alignment w AEM fuel cell i H2/air (maxim than 10% volta 600 mA/cm2 a	with DOE 2019, Q2 AEM target, Demonstrate nitial performance of 0.6 V at 600 mA/cm2 on um pressure of 1.5 atma) in MEA, and less age degradation over 1,000 hour hold test at at T>60 oC.	12/31/2018	Milestone - Met at 500 hours
Q2	² Develop ionomer solutions based on spirocyclic AEM ^{3/} polymers.			Progress Measure - Met
Q3	Quantify fuel of using spirocycl performance.	6/30/2019	Progress Measure – On track	
Q4	4 Optimize fuel cell performance based on a fully spirocyclic system where cathode and anode ionomer and membrane are spirocyclic based.			Progress Measure – On track
AEM FC durability In alignment with DOE 2019, Q2 AEM target, Demonstrate AEM fuel 2 cell initial performance of 0.6 V at 600 mA/cm ² on H ₂ /air (maximum pressure of 1.5 atma) in MEA, and less than 10% voltage degradation over 1.000 hour hold test at 600 mA/cm ² at T>60 °C.				12/31/2018 – conditionally met

Synthesis/Characterization





Polymer	IEC (measured) [mmol/g]	Cl ⁻ Conductivity @RT in Water [mS/cm]	Water Uptake (%)	Peak Power Density (W/cm²)
SpiroCyclic AEM	1.3	14.0	22	0.85
SpiroCyclic AEM	1.5	13.8	42	1.22
SpiroCyclic AEM	1.7	16.1	80	1.48
Gen 2 PF AEM	0.9	13.4	18	1.10

Durability

Before and (where possible) after Degradation

- Teflon-lined Parr reactors (20 mL)
- Dry samples of ~100 mg in Cl⁻ form
- 1 M KOH (10 mL)
- 80 °C in oven for 1000 h
- Room Temperature, Liquid H₂O, Cl⁻ Conductivity and IEC tested before/after





- More opaque color
- Still soft/flexible
- Broken/tears easily
- IEC ↓ 2.8%
- Conductivity $\downarrow 21\%$

PSF-1.7



- Darker orange color
- Hard/inflexible
- Breaks easily
- IEC $\downarrow 50\%$
- Conductivity $\downarrow 4\%$





PSF-1.5





- Orange/brown color
- Hard/inflexible
- Broken/tears easily
- IEC ↓ 66%
- Conductivity ↓ 60%

Spirocyclic AEM in-situ Fuel Cell Performance



- Cell Temp: 60°C
- Active Area: 5 cm²
- Pressure: 121 kPa_{abs}
- Ionomer: Varcoe ETFE AEI ionomer
- GDL: Toray H-060
- Gases: H₂ var°C, O₂ var°C, 1.0 slpm
- Anode: 0.8 mg/cm² PtRu/Vu
- Cathode: 0.5 mg/cm² Pt/Vu





~ 1.48 W/cm² in peak power density

- ASR ~ 0.007 Ω (0.036 Ω-cm²)
- Close to state-of-the-art performance

In-situ Fuel Cell Durability: Spirocyclic VS PF AEM

Durability 600mA/cm² hold, H₂/Air, 122 kPa_{abs}



Spirocyclic AEM in-situ Fuel Cell Durability



PGM-free Cathode, Low loading Spirocyclic AEM MEA

<u>Go/No-Go Decisions:</u> Completed. Demonstrate AEM fuel cell initial performance of 0.6 V at 600 mA/cm² on H₂/air (max. 1.5 atma) in MEA at T > 60°C with loading < 0.1 mg PGM/cm². Membrane Electrode Assembly

- Cell Temp: 70°C
- Active Area: 5 cm²
- Pressure: 132 kPa_{abs}
- GDL: Toray H-060
- Anode: 0.1 mg/cm² PtRu/Vu
- Cathode: 2.2 mg/cm² Ag/Vu
- Membrane: Spirocyclic, 1.7mmol/g, (42μm)



Spirocyclic as Ionomer Incorporated into Electrodes

PF AEM Gen2 AEM, Spirocyclic ionomer



- Spirocyclic serves as a good ionomer with high performance and durability.
- At 37 hours of durability test, cell shut down occurred due to H₂ outage.

- Cell Temp: 60°C
 Active Area: 5 cm²
- Pressure: 121 kPa_{abs}
- GDL: Toray H-060
- Anode: 0.5 mg/cm² Pt/Vu
- Cathode: 0.5 mg/cm² Pt/Vu
- Membrane: Gen 2 PF AEM (40μm)

Membrane Electrode Assembly

• Dew points: 60/60/60

Collaborations and Coordinations

- NREL only (limited-funds) project that highly leverages significant effort at NREL on related projects including
 - PF AEM project (FC 142)
 - Membrane Working Group (FC 301)
 - ARPA-E efforts
 - FCTO HydroGEN AWSM (EMN) efforts

Summary

- We synthesized spirocyclic polymers of multiple ion exchange capacity (IEC)
- Lower IEC polymers were synthesized due to trends witnessed for increased durability with decreasing IEC.
- Thinner membranes were employed to limit the impact of lower IEC membranes in fuel cell tests.
- With optimized membranes greater than 500 hours of durability above 0.6V at 600 mA/cm2 was demonstrated.
- High performance with low total PGM loading was shown using spirocyclic membranes.
- Initial performance and durability of spirocyclic ionomer based electrodes showed promise.

Response to Reviewers comments: Project presented last year but not reviewed.

Remaining Challenges and Barriers

- Polymer stability remains an issue as IEC and conductivity losses in ex-situ testing were high, but fuel cell durability was reasonable when used as a membrane and in initial studies as an ionomer
- Further optimization as an electrode ionomer
 - Ink composition and processing
 - Specific IEC of ionomer material

Future Work

- Focus on ionomer implementation of spirocyclic polymer with a focus on probing cation-catalyst interactions
- Further investigation of low PGM-loaded MEAs

Technical Backup Slides

Degradation Pathways – Accelerated testing

- After 144 hours some insoluble material was left behind in the reactor
- Insolubility indicates crosslinking resulting from ring opening attack of hydroxyl degradation products



Characterization - conductivity

0 3 6 9 12 15 18 21 24

Time (h)







Characterization – water uptake





Durability

Post 1000 h, 80°C Degradation Testing







- PSF-1.3 could not be measured
- Some correlation between IEC and conductivity loss, however morphology changes also seem significant