Project ID: fc172

### Highly Active and Durable PGM-free ORR Electrocatalysts through the Synergy of Active Sites

Yuyan Shao

#### Pacific Northwest National Laboratory 05/20/2020

This presentation does not contain any proprietary, confidential, or otherwise restricted information

# Overview

#### Timeline

- Project Start Date: 10/01/17
- Project End Date: 03/31/21\*
  - \* Project continuation and direction determined annually by DOE
  - \* No-cost extension for 6 months (Year 1: 10/01/17-03/31/19)

#### Budget

- Total Project Budget: \$2,223,776
  - Total Recipient Share: \$223,776
  - Total Federal Share: \$2,000,000
  - Total DOE Funds Spent\*: \$1,190,297
    - \* As of 05/20/19

#### **Barriers**

- Barriers addressed
  - Cost (catalyst)
  - Activity (catalyst; MEA)
  - Durability (catalyst; MEA)

#### Partners

- Washington Univ. in St. Louis
- Univ. of Maryland, College Park
- Ballard Power Systems Inc.
- Oregon State Univ.
- ElectroCat Consortium
- Project lead: PNNL

# Relevance

PGM-free ORR catalysts could significantly decrease fuel cell cost, but their durability (and activity) needs improvement.



Ref.: 1) Adv. Mater. 2019, adma.201807615

## Relevance

H<sub>2</sub>O<sub>2</sub>(radicals) play a significant role in PGM-free ORR catalysts degradation.



This project will decrease  $H_2O_2$  (radical) formation through dual active site catalysts, doubling stability than baseline, while maintaining the high activity.

Ref.: 1) Energy Environ. Sci. 2018, DOI: 10.1039/C8EE01855C; 2) Adv. Mater. 2019, adma.201807615



Fundamental understanding using ElectroCat capabilities includes HR-STEM, synchrotron X-ray, *In situ* degradation measurement/detection, and MEA diagnosis.

# Milestones

Milestone	Milestone Description (Go/No-Go Decision)	Due date	Status
GNG2 (Year 2)	Demonstrate a PGM-free catalyst $\geq 25 \text{ mA/cm}^2$ at 0.90 V (iR-corrected) in an H <sub>2</sub> -O <sub>2</sub> fuel cell, maintain partial pressure of O <sub>2</sub> + N <sub>2</sub> at 1.0 bar (cell temperature 80 °C), demonstrate MEA 1.5X stability over baseline.	04/30/20	100%
M2.1	Identify dual-site catalysts with $\Delta E_{1/2}$ < 45 mV (vs. Pt/C) under RRDE test	07/31/19	100%
M2.2	Demonstrate ALD deposition of radical scavenger on PGM-free catalysts with < 5 nm particle size, and >50% decrease of $H_2O_2$ formation	10/31/19	100%
M2.3	Demonstrate AHD deposition of radical scavenger into micropores of PGM- free catalysts, >50% decrease of $H_2O_2$ formation	12/31/19	100%
M2.4	Identify dual-site catalysts with $H_2O_2$ generation less than half of Pt/C (2%) under RRDE test	01/31/20	100%
GNG3 (Year 3)	Demonstrate a PGM-free catalyst $\geq$ 30 mA/cm <sup>2</sup> at 0.90 V (iR-corrected) in an H <sub>2</sub> -O <sub>2</sub> fuel cell and 150 mA/cm <sup>2</sup> at 0.80 V in an H <sub>2</sub> -air fuel cell (measured); maintain partial pressure of O <sub>2</sub> + N <sub>2</sub> at 1.0 bar (cell temperature 80 °C). Provide six 50cm <sup>2</sup> MEAs to DOE while showing a reasonable pathway to achieve DOE performance and durability targets.	04/30/21	On- going
M3.1	Demonstrate integrated ALD and AHD deposition of radical scavengers with >75% decrease of $H_2O_2$ formation and >50% decrease in carbon oxidation	07/31/20	On- going
M3.2	Demonstrate MEA 2X durability of dual-site catalysts over baseline	10/31/20	On- going
Any propos	ed future work is subject to change based on funding levels.		

Key update #1: MEA performance in both  $O_2$  and air has been improved continuously in the past three years.



Anode: 0.1-0.2 mg/cm<sup>2</sup> <sub>Pt</sub>; Cathode: 2.5-4.5mg/cm<sup>2</sup><sub>PGM-free</sub>; Membrane: Gore (15 $\mu$ m)/NR211; Active area: 5 cm<sup>2</sup>; 80-100% RH, 80°C, O<sub>2</sub>+N<sub>2</sub>=1 bar. Performance improvement due to increased catalyst activity (increased active site density and accessibility (through porosity optimization)) and electrode fabrication.



7

**Key update #2**: Dual active site catalyst FeNC\_NCeOx exceeds Year 2 activity and durability milestones (i.e.,  $\geq 25 \text{ mA/cm}^2$  at 0.90 V (iR-corrected) in an H<sub>2</sub>-O<sub>2</sub> fuel cell, 1.5X stability over baseline).



Anode: 0.1 mg/cm<sup>2</sup> <sub>Pt</sub>; Cathode: 2.6 mg/cm<sup>2</sup><sub>PGM</sub>. <sub>free</sub> (FeNC\_NCeOx), 40 wt% Nafion ionomer; Membrane: NR211; Active area: 5 cm<sup>2</sup>; 100% RH, 80°C,  $O_2$ + $N_2$ =1 bar. DOE ElectroCat durability test protocol.

- 1. Initial MEA performance and activity (> $30mA/cm^2$  at 0.9V) in H<sub>2</sub>/O<sub>2</sub> of dual-active site catalyst FeNC\_NCeOx is better than baseline FeNC.
- 2. The retention of current density at 0.9V for FeNC\_NCeOx is more than 100% of baseline.



8

Atomically dispersed, highly porous FeNC catalysts



STEM/EELS confirm high-density, atomically dispersed Fe and most likely FeNx moieties. The catalyst shows highly porous structure.



Atomically dispersed FeNC catalysts



XANES and EXAFS confirm atomic dispersion of Fe (no Fe-Fe and Fe-C).



FeNC catalyst exceeds Year 3 activity and performance milestones (i.e.,>30mA/cm<sup>2</sup> at 0.9V in O<sub>2</sub> and >150mA/cm2 at 0.8V in air)



chosen for comparison.

Dual active sites: FeNC\_NCeOx – improved durability



Test condition: 0.5M H<sub>2</sub>SO<sub>4</sub>, ORR staircase potential step of 0.025 V at intervals of 25 s from 1.0 to 0.0 V vs RHE. Catalyst loading: 0.6mg/cm<sup>2</sup>. 25°C.

- 1. Improved ORR stability with NCeOx
- H<sub>2</sub>O<sub>2</sub>% increases after durability testing much smaller increase with NCeOx – indicating a self-catalyzed degradation for FeNC.
- 3. MEA performance and durability in Slide #8.

Any proposed future work is subject to change based on funding levels.



12

Dual active site catalyst: FeNC(N600)\_NCeOx – improved durability





Any proposed future work is subject to change based on funding levels.

80°C.

New radical scavenger NTaTiOx improves catalyst durability



Northwest

Successful loading of NTaTiOx using ambient hydrolysis deposition method



Any proposed future work is subject to change based on funding levels.

 Improved durability for CoNC catalyst – indicating applicable to non-Fe catalyst

2. Decreased  $H_2O_2\%$ 

Test condition:  $0.5M H_2SO_4$ , ORR staircase potential step of 0.025 V at intervals of 25 s from 1.0 to 0.0 V vs RHE. Catalyst loading:  $0.6mg/cm^2$ . Stability:  $0.6 V \_ 3 s$ ; 0.95 V $\_ 3 s$ , 10,000 cycles,  $O_2$ saturation. 25°C.





#### Accomplishments and Progress: Responses to Previous Year Reviewers' Comments

- Comments: "...The radical scavenger approach is good, and the results are promising. The team needs to further improve membrane electrode assembly (MEA) performance...." Response: we have improved MEA performance continuously, especially in FY20. (Slide #7).
- Comments: "... whether the dual-site catalyst is really more stable or only appears more stable because it starts at lower performance. ...." Response: We have demonstrated that, with the initial activity and performance of dual-site catalyst being higher than baseline, the durability is better than baseline (Slide #8).
- 3. Comments: "...What is missing in the logic chain is direct evidence showing that the improved durability in PEMFCs is indeed related to the reduced H2O2 yield." Response: in the past years (FY18, FY19), we showed the correlation between catalyst degradation H<sub>2</sub>O<sub>2</sub> yield CO<sub>2</sub> formation; we are designing experiment to confirm the direct relationship between improved durability and reduced H<sub>2</sub>O<sub>2</sub> yield.
- 4. Comments: "… It is strongly recommended that the project team design experiments or perform modeling to determine the extent to which this approach will work.." Response: This is critical for overall durability study. We have seen different degradation in liquid vs. MEA and we are working to quantify the degradation and dual-site effect.

# **Collaboration & Coordination**

Partner		Project roles			
PNNL – Lead (Y. Shao, X. Xie, L. Yan, V. Prabhakaran)		Project lead, management and coordination; catalysts design and synthesis.			
Ballard (L. Yang)		MEA design, test and analysis			
Univ. Maryland(L. Hu)		New radical scavenger, atomic layer deposition			
WashU (V. Ramani)		Electrode design and MEA assembly, MEA test and analysis			
Oregon State U. (X. Ji)		New radical scavenger, ambient hydrolysis deposition			
ElectroCat	Capabilities				
ANL	<i>In situ</i> and Operando Atomic, Nano-, and Micro-structure Characterization (X-ray adsorption, including <i>ex-situ, in-situ</i> in liquid/MEA) Electrode Microstructure Characterization and Simulation (X-ray Nano CT)				
LANL	<i>In situ</i> fluoride and carbon dioxide emission measurements (including F/metal/CO <sub>2</sub> detection simultaneously)				
NREL	Kinetics and Transport (Operando differential cell measurements of electrochemical kinetics and transport)				
	clocaconomia				

### Remaining Challenges and Barriers

- Understand degradation mechanisms of PGM-free catalysts under various conditions;
- Improve synergy of dual active sites for further enhanced durability;
- MEA performance and durability.

### **Proposed Future Work**

- Quantify each degradation pathway (e.g., radical oxidation, demetallation,...) and identify mitigation strategy for each of them;
- Develop and optimize new radical scavenger (e.g., NTaTiOx) and new synthesis method (e.g., ALD and AHD) for further enhanced durability;
- Optimize PGM-free catalysts including active sites and active site density, porosity, particle size.
- Optimize MEA fabrication (e.g., interaction between catalyst radical scavenger – ionomers).

# Summary Slide

- Dual active site catalyst FeNC(N600)\_NCeOx exceeds Year 2 activity milestones (>25mA/cm<sup>2</sup>, 0.9V in  $H_2/O_2$ ) and durability milestone (1.5X durability improvement).
- FeNC catalyst exceeds Year 3 activity milestone (>30mA/cm<sup>2</sup>, 0.9V in H<sub>2</sub>/O<sub>2</sub>) and performance milestone (>150mA/cm<sup>2</sup>, 0.8V in H<sub>2</sub>/Air) – high catalyst loading.
- Dual active site catalyst improves catalyst durability while achieving higher activity than baseline.
- Collaboration with ElectroCat helped understand the catalysts (chemistry, active sites, functional mechanisms,...).
- Need improvement on catalyst degradation understanding.

# **Technical Back-Up Slides**

XPS shows the increase of Fe loading with optimal synthesis.



#### Element content in catalysts

Catalyst	C 1s (wt%)	N 1s (wt%)	O 1s (wt%)	Fe 2p3 (wt%)
FeNC	81.31	12.76	4.17	1.76
FeNC(N400)	81.12	12.84	4.23	1.81
FeNC(N500)	79.68	12.39	4.69	3.24
FeNC(N600)	79.05	12.47	4.28	4.20
FeNC(N700)	79.42	12.36	4.95	3.27