2020 DOE Hydrogen and Fuel Cells Program Annual Merit Review



Synthetic Fuels Technoeconomic Analysis and Life Cycle Analysis



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May 30, 2020

SA174

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Overview

Timeline

- Start: October 2019
- End: Determined by DOE
- % complete (FY20): 80%

Barriers to Address

- A: Future Market Behavior
 - Potential market for low value energy and potential hydrogen markets beyond transportation
- D: Insufficient Suite of Models & Tools
- E: Unplanned Studies and Analysis ≻H2@Scale is a new concept and requires analysis of its potential impacts

Budget

• Funding for FY20: \$150K

Partners/Collaborators

- NREL and INL
- DOE NE Office



H2@SCALE EXPANDS HYDROGEN UTILIZATION BEYOND CURRENT MARKETS – Relevance/Impact



- Synthetic fuels production can contribute to large scale hydrogen demand
- Need to evaluate potential hydrogen demand and associated cost

EVALUATE ECONOMIC AND ENVIRONMENTAL IMPACTS OF PRODUCING SYNTHETIC FUELS FROM ZERO-CARBON HYDROGEN – Approach



$H_2+CO_2 \rightarrow LIQUID HYDROCARBON FUELS AND CHEMICALS - Relevance$

 Electrofuels or "e-fuels" encompass energy carriers and their intermediates synthesized primarily using a carbon source and electricity (for hydrogen)





MAJOR CO₂ AND ZERO–CARBON ELECTRICITY SOURCES TO CONSIDER – Approach

Fossil power plants



LARGE NUMBER OF SYNTHETIC CHEMICALS AND FUELS PATHWAYS ARE POSSIBLE – Approach





LCA of CO₂–BASED FUELS: SYSTEM BOUNDARY – Approach





HIGH-CONCENTRATION CO₂ SOURCES – Accomplishment



OTHER LOWER-CONCENTRATION CO₂ SOURCES – Accomplishment



• Cement plant CO₂: 14-33% purity, 65 MMT /year



POTENTIAL H₂ DEMAND FOR SYNTHETIC HYDROCARBON PRODUCTION FROM CONCENTRATED CO₂ SOURCES – Accomplishment

- ~100 MMT of concentrated CO₂ sources are considered (out of total~ 5 GT CO₂)
 - ➢ 44 million MT from ethanol plants
 - ✓ Current CO_2 supply capacity of 14 MMT, and market demand of 11 MMT
 - Remainder from hydrogen SMR (refineries) and ammonia plants



Source: Supekar and Skerlos, ES&T (2014)

14 MMT POTENTIAL H₂ DEMAND WITH **100MMT CONCENTRATED CO₂ ANNULAY** - Accomplishment

Installed nuclear plants U.S. Operating Commercial Nuclear Power Reactors



Wind electricity potential



Solar electricity potential





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MODELED H₂ DEMAND FOR SYNTHETIC FISCHER-TROPSCH (FT) HYDROCARBON PRODUCTION – Accomplishment





Well to wheel (WTW) boundary of FT fuel production from electrolysis hydrogen and ethanol plant by-product CO₂

Life cycle emissions



FT FUEL PROCESS SIMULATION USING ASPEN PLUS – Accomplishment

- Six process areas were simulated
- Two systems were evaluated:
 - with H₂ recycle
 - without H_2 recycle (with electricity export)

Preliminary



INTEGRATED ETHANOL AND FT FUEL MASS CONVERSION – Accomplishment

- FT fuel production is integrated with the dry milling corn ethanol production via 154 kg CO₂
- 479 kg corn and 17 kg H₂ are converted into 161 kg ethanol, 138 kg DGS, and 23 kg FT fuel





CARBON-NEUTRAL HYDROGEN PRODUCES NET ZERO **CARBON FT FUELS – Accomplishment**



(Forthcoming by Zang et al.)

HYDROGEN COST < \$1/kg FOR FT FUEL TO BREAKEVEN WITH \$3.6/gal DIESEL – Accomplishment





SUMMARY – Accomplishment

- Modeled hydrogen demand and product yield for synthetic FT production
- Simulated two FT plant designs, with and without recycling of hydrogen, using Aspen Plus
- Conducted life cycle analysis with GREET[®] model
 - Evaluated two co-product allocation methods
 - More than 90% CO_{2e} emission reduction compared to petroleum fuels
- Conducted techno-economic analysis using H2A framework to determine cost of hydrogen for FT fuels to breakeven with petroleum fuels
 - Hydrogen cost < \$1/kg for FT to breakeven with \$3.6/gal Diesel
- Documented data sources, modeling approach and analysis in two papers
 - Currently under review
 - Publication pending



Collaborations and Acknowledgments

H2@Scale is a multi-national laboratory effort with collaboration across DOE national lab complex

- Mark Ruth and Paige Jadun: NREL
- Richard Boardman: INL



Future Work

- Consider additional CO₂ sources for synthetic hydrocarbon production
 - e.g., natural gas processing plants, cement plants, power plants and direct air capture
- Conduct economic and environmental analysis of other synthetic fuels and chemicals
 - e.g., methanol and methanol-to-gasoline, etc.
 - Determine potential CO₂ reduction for each fuel and chemical pathway compared to baseline current technologies
 - Determine H₂ cost for synthetic hydrocarbon to breakeven with baseline technology pathway
- Conduct regional analysis considering proximity of CO₂ and H₂ supplies
 - Evaluate economics of delivered H_2 vs. onsite production
 - Evaluate economics of CO₂ capture and transportation



Project Summary

- Relevance: Hydrogen from clean energy sources can be used with available CO₂ sources to produce near zero-carbon synthetic hydrocarbon chemicals and fuels
- Approach: Evaluate economic and environmental impacts of synthetic fuel production using Aspen, H2A framework and GREET models
- Collaborations: H2@Scale is a multi-national laboratory effort with collaboration across DOE national lab complex
- Technical accomplishments and progress:
 - Modeled hydrogen demand and product yield for synthetic FT production
 - Evaluated two FT plant designs, with and without recycling of hydrogen
 - Conducted life cycle analysis with GREET[®] model
 - Evaluated two co-product allocation methods
 - > More than 90% CO_{2e} emission reduction compared to petroleum fuels
 - Conducted techno-economic analysis to determine cost of hydrogen for FT fuels to breakeven with petroleum fuels
 - Hhydrogen cost < \$1/kg for FT to breakeven with \$3.6/gal Diesel</p>
 - Documented data sources, modeling approach and analysis in two papers

• Future Research:

- Consider additional CO_2 sources for synthetic hydrocarbon production
- Conduct economic and environmental analysis of other synthetic fuels and chemicals
- Conduct regional analysis considering proximity of CO_2 and H_2 supplies
- Evaluate economics of delivered H_2 vs. onsite production
- Evaluate economics of CO_2 capture and transportation