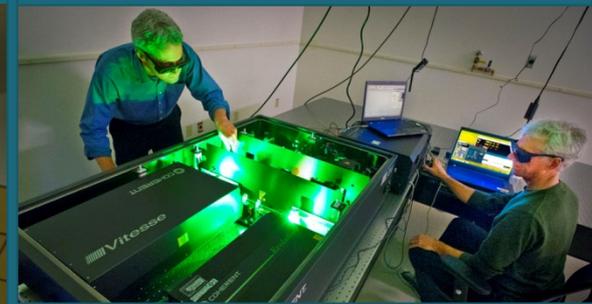
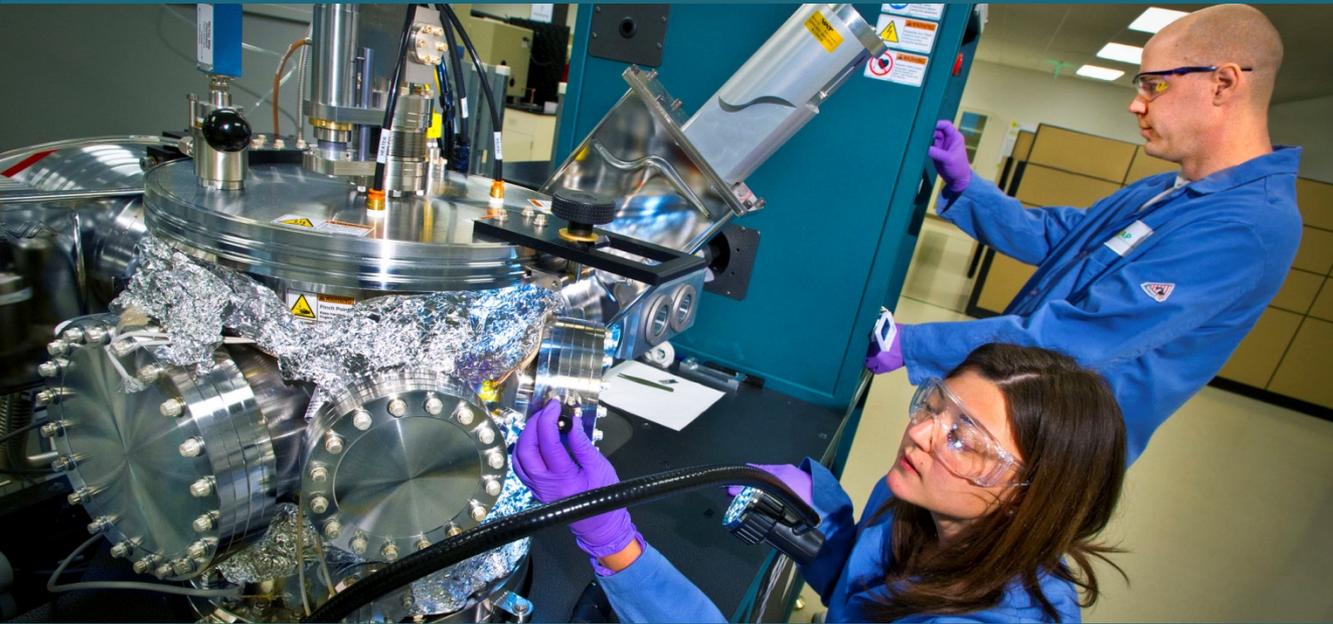




JCAP Research Overview for HTAC



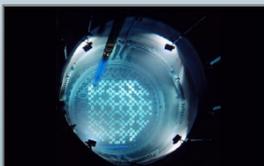
Carl Koval, Director
October 30, 2013



DOE ENERGY INNOVATION HUBS

All five DOE Energy Innovation Hubs have challenging goals

CONSORTIUM FOR ADVANCED SIMULATION OF LIGHT WATER REACTORS (CASL)



Objective: Develop computer simulation models that enhance the safety, reliability, and economics of nuclear energy.

ENERGY EFFICIENT BUILDINGS HUB (EEBHUB)

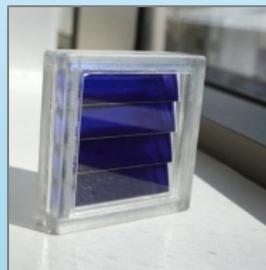


Objective: Improve energy efficiency of buildings in the Philadelphia metropolitan area by 20% by 2020.

JOINT CENTER FOR ARTIFICIAL PHOTOSYNTHESIS (JCAP)



Objective: Develop an efficient, scalable and robust prototype that generates fuel from sunlight, water, and carbon dioxide.



Objective: Increase the energy density of batteries by a factor of five within five years at one-fifth of the cost.



JOINT CENTER FOR ENERGY STORAGE RESEARCH (JCESR)



CRITICAL MATERIALS INSTITUTE (CMI)

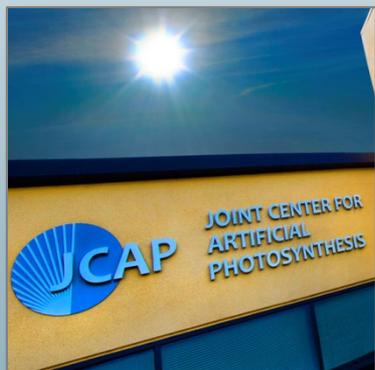


Objective: Identify approaches to reduce supply risk of rare-earth elements that are critical to clean energy technologies.



THE JOINT CENTER FOR ARTIFICIAL PHOTOSYNTHESIS – AT A GLANCE

5 LEADING RESEARCH INSTITUTIONS



TWO DEDICATED RESEARCH LABORATORIES

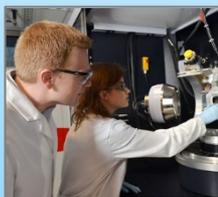


Jorgensen Laboratory

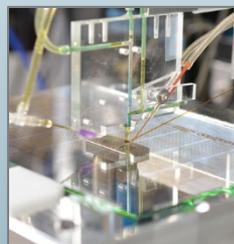
Solar-Energy Research Center



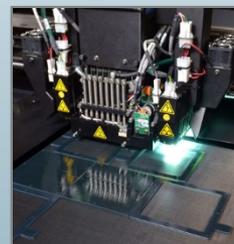
140 SCIENTISTS AND ENGINEERS



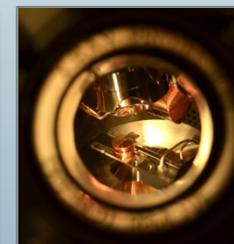
INSTRUMENTS & UNIQUE CAPABILITIES



High throughput electrochemistry



Rapid prototyping systems



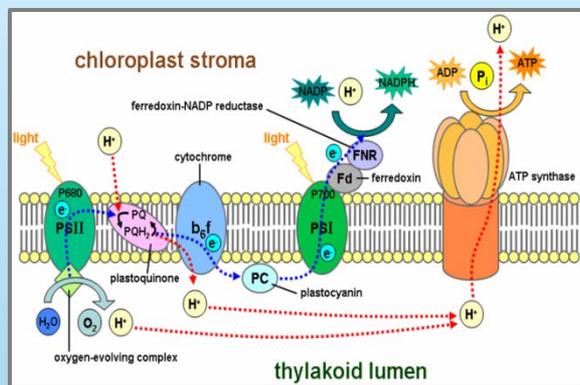
Advanced surface characterization



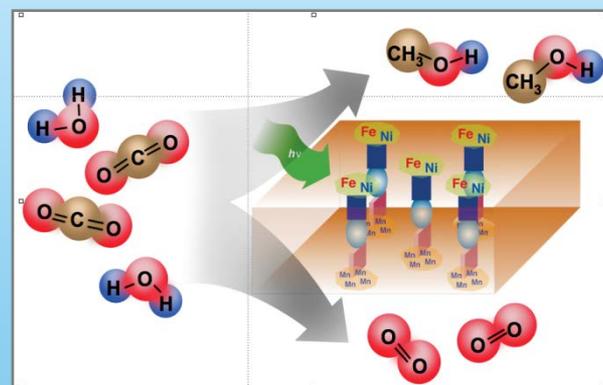
JCAP 5-YEAR GOAL

Discovery of robust, Earth-abundant light absorbers, catalysts, linkers and membranes; and scale-up science required to assemble the components into a complete photosynthetic system.

NATURAL PHOTOSYNTHESIS



ARTIFICIAL PHOTOSYNTHESIS

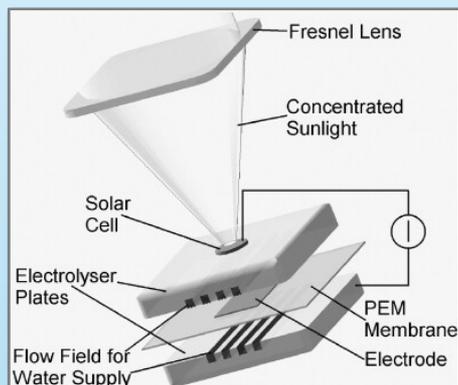


“It is time to build an actual artificial photosynthetic system, to learn what works and what does not work, and thereby set the stage for making it work better”

Melvin Calvin (1961 Nobel Laureate)

GENERAL APPROACHES TO ARTIFICIAL PHOTOSYNTHESIS

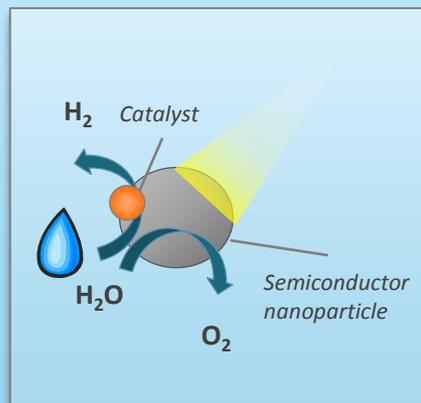
DISCRETE PHOTOVOLTAIC WIRED TO ELECTROLYZER



Advantages: Operational system has already been demonstrated with 18% efficiency.¹

Challenges: Demonstrated system demands expensive components; lack of integration further reduces cost efficiency.

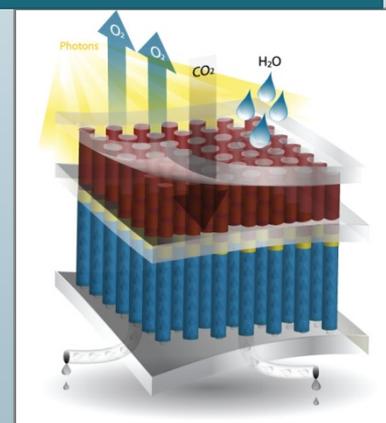
SOLAR-FUEL GENERATING PARTICLE DISPERSIONS



Advantages: Offers a simple architecture with the potential for low materials cost.

Challenges: Co-generation of fuel and oxidizer pose operational safety issues.

INTEGRATED PHOTOELECTROCHEMICAL SOLAR-FUEL GENERATOR



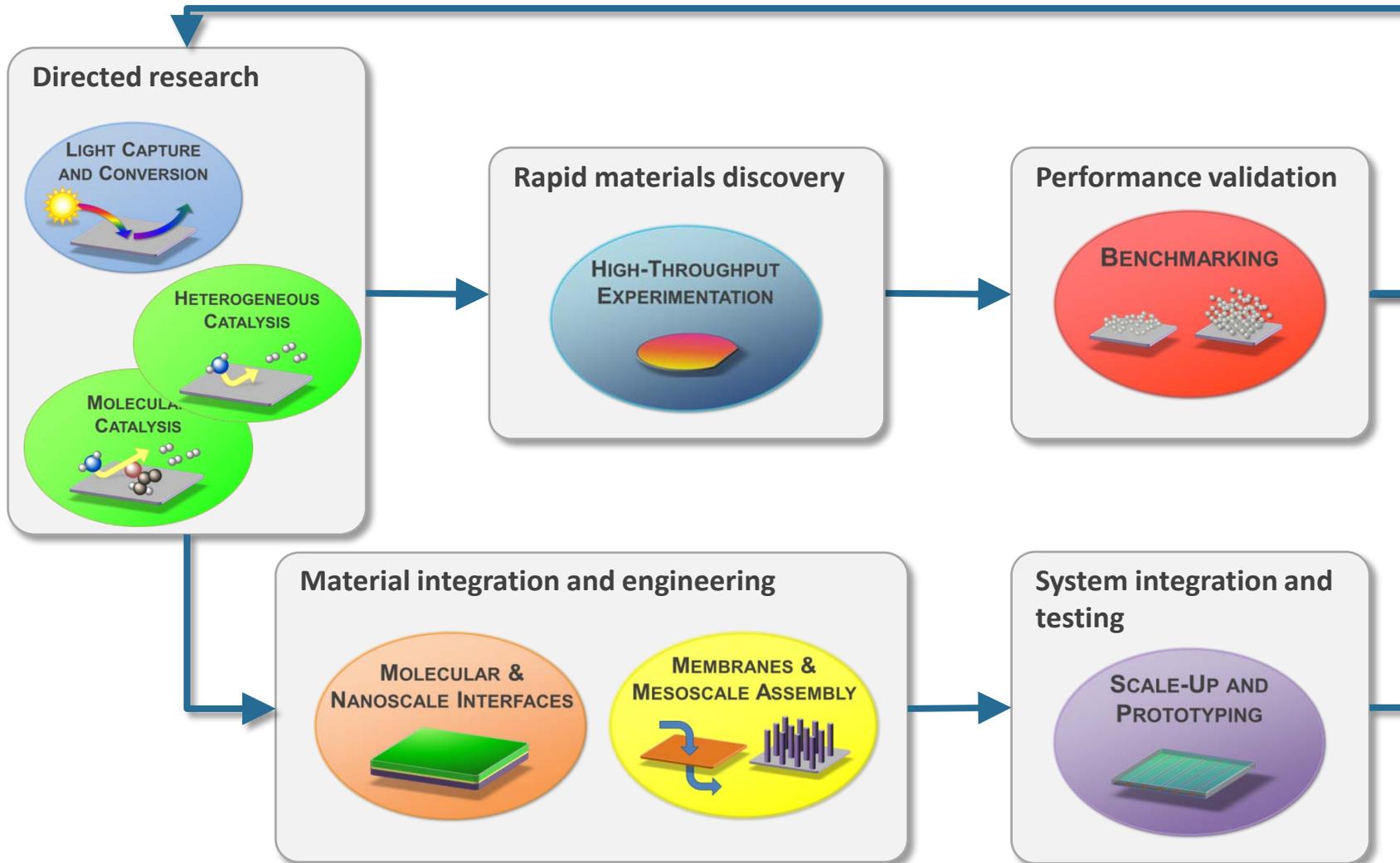
Advantages: Potentially lower component costs than a discrete system with reduced complexity.

Challenges: Requires that semiconductor, catalysts, and membranes operate efficiently under identical conditions.

¹ G. Peharz, F. Dirmouth, and U. Wittstadt *Int. J. of Hydrogen Energy* **2007**, *32*, 3248-3252 ([DOI: 10.1016/j.ijhydene.2007.04.036](https://doi.org/10.1016/j.ijhydene.2007.04.036))

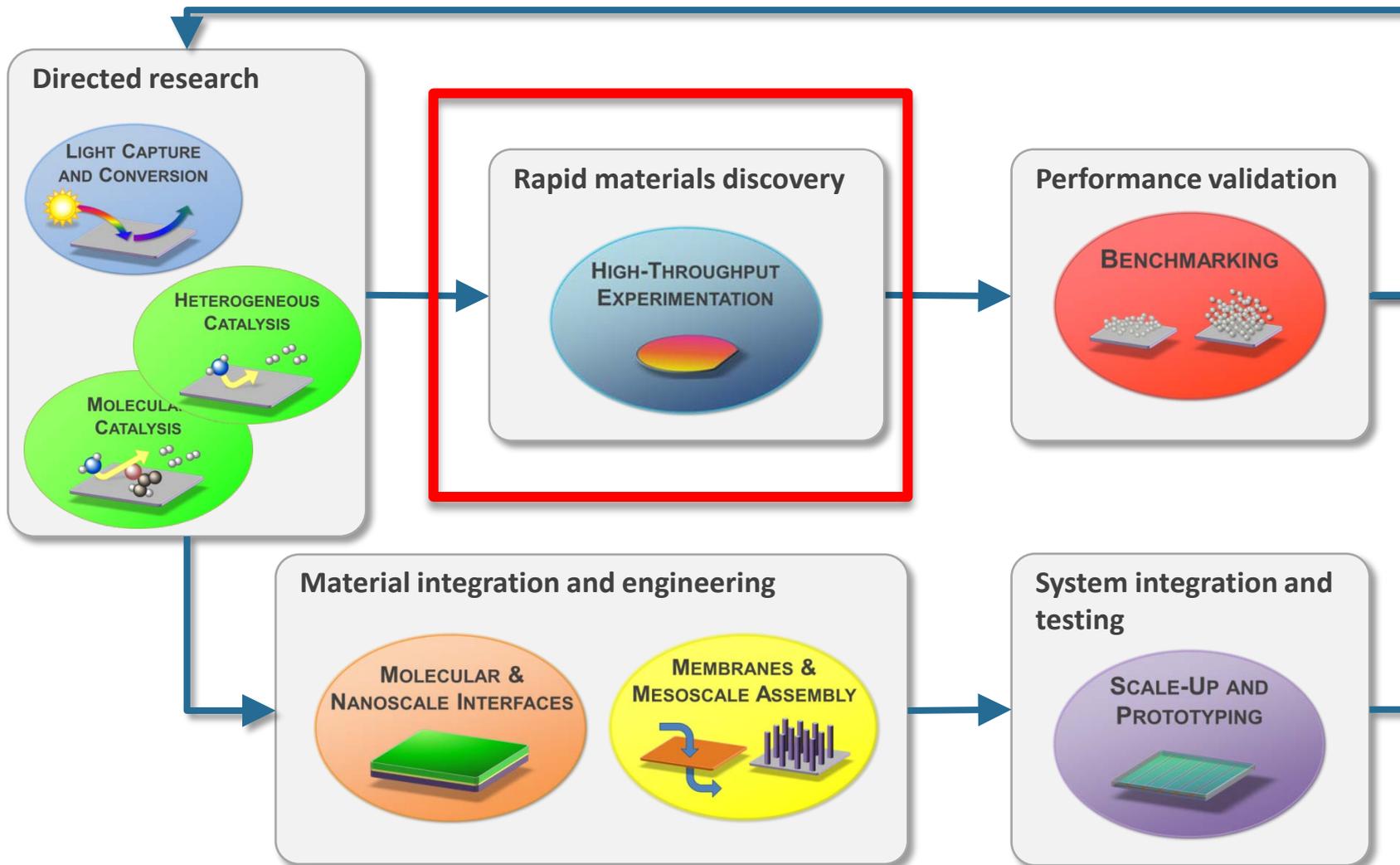
JCAP APPROACH TO ARTIFICIAL PHOTOSYNTHESIS

Projects are coordinated to accelerate materials discovery and prototype development

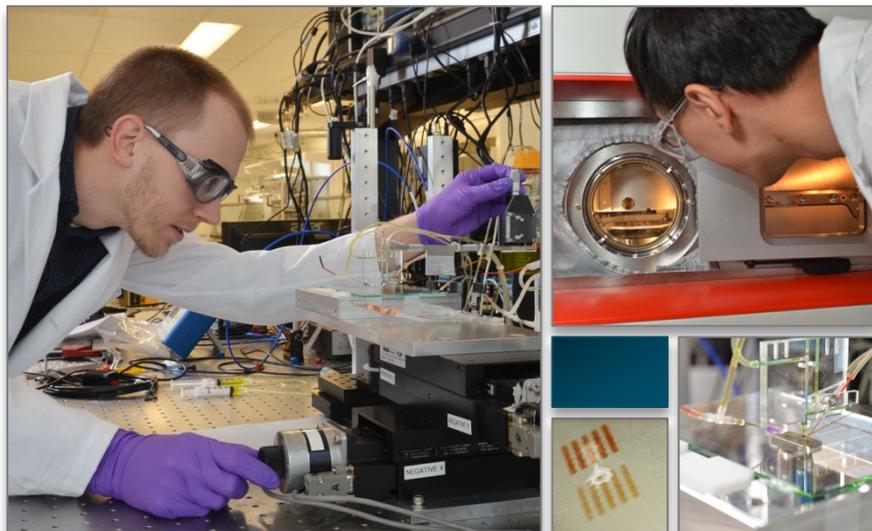


JCAP APPROACH TO ARTIFICIAL PHOTOSYNTHESIS

Projects are coordinated to accelerate materials discovery and prototype development



OVERVIEW OF THE HIGH-THROUGHPUT EXPERIMENTATION PROJECT



OBJECTIVES

JCAP's High-Throughput Experimentation (HTE) Project focuses on automated, high-throughput discovery of light absorbers and heterogeneous catalysts. The project employs combinatorial techniques that can produce new alloys from Earth abundant elements, rapid screening methods that can identify high-performing light absorbers and catalysts, and surface-science analysis tools that can characterize the structure and composition of promising materials.

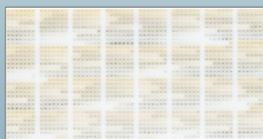
PROJECT LEADER



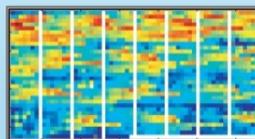
John
Gregoire

APPROACH

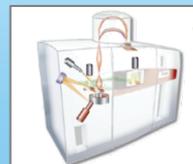
Combinatorial materials synthesis



High-throughput screening



High-throughput characterization



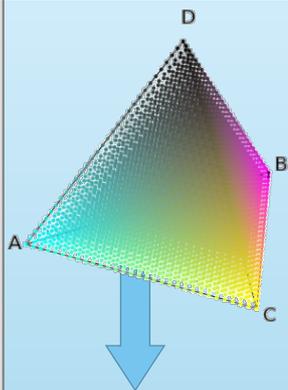
Planning and analysis



COMBINATORIAL MATERIALS SYNTHESIS

Discrete composition libraries are produced using ink jet printers

MAPPING OF COMPOSITION SPACE



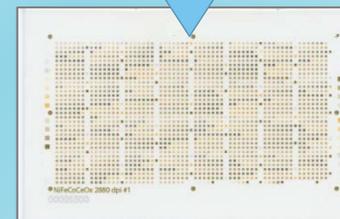
- 4-element space
- 3.33% intervals.
- 5456 samples (1mm²) on 3 plates



INK FORMULATION AND PRINTING



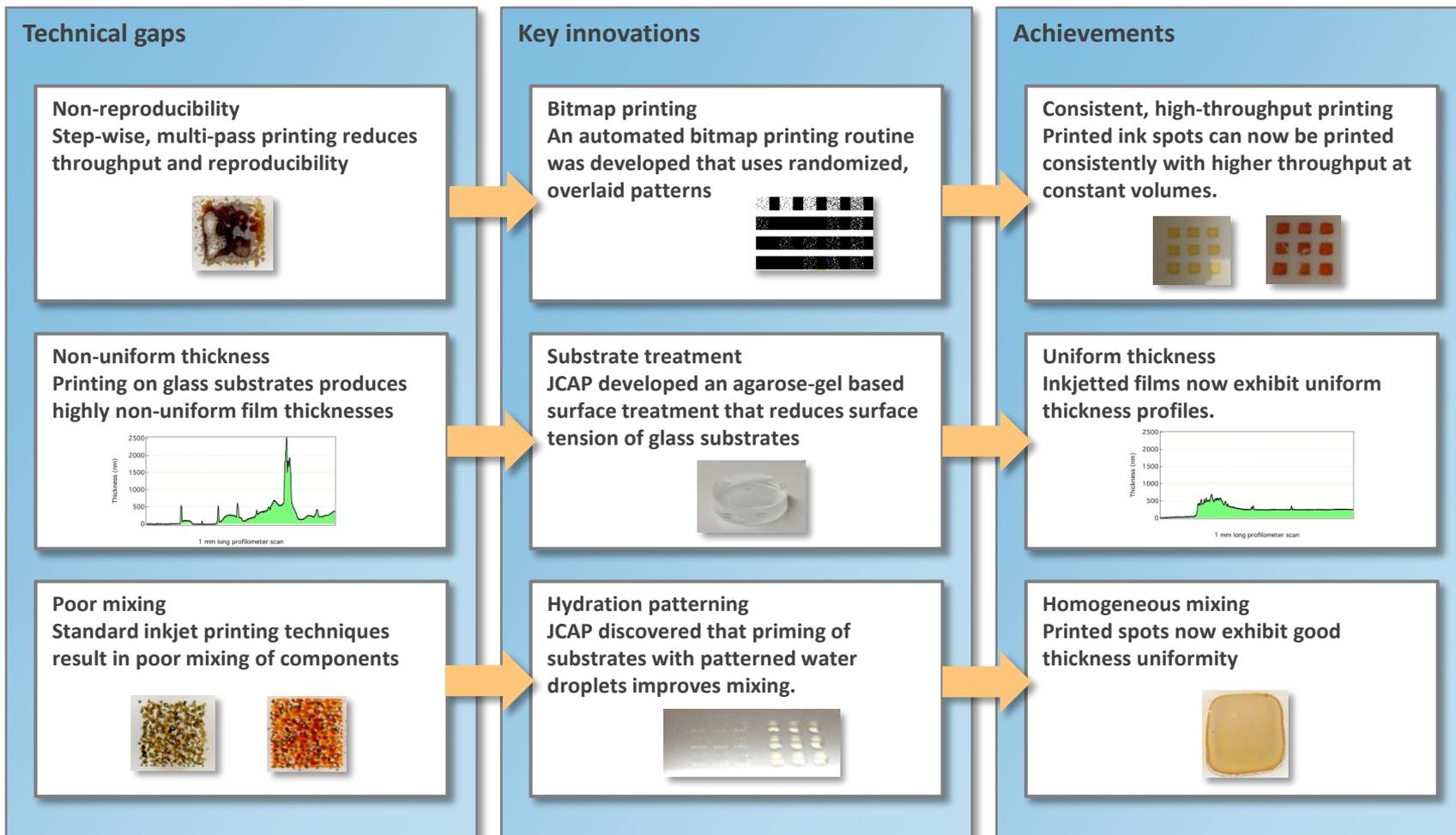
CALCINATION



Material libraries consisting of pseudo-binary, tertiary, and quaternary compositions can be produced with unprecedented throughput using inkjet printers

COMBINATORIAL MATERIALS SYNTHESIS

High-throughput inkjet printing required extensive development



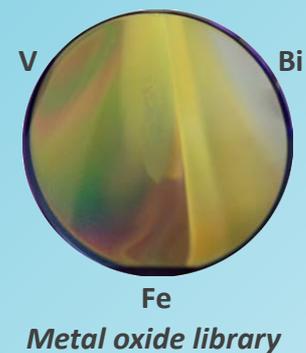
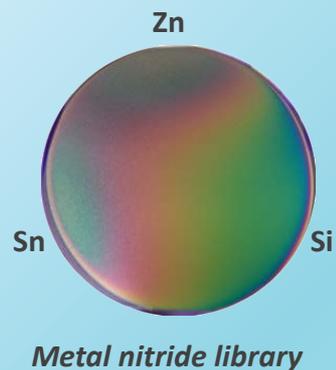
COMBINATORIAL MATERIALS SYNTHESIS

Continuous composition libraries can be produced using physical vapor deposition

OFF-AXIS SPUTTERING SYSTEM



TERNARY LIBRARY EXAMPLES



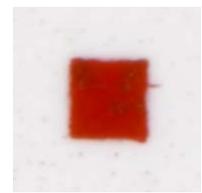
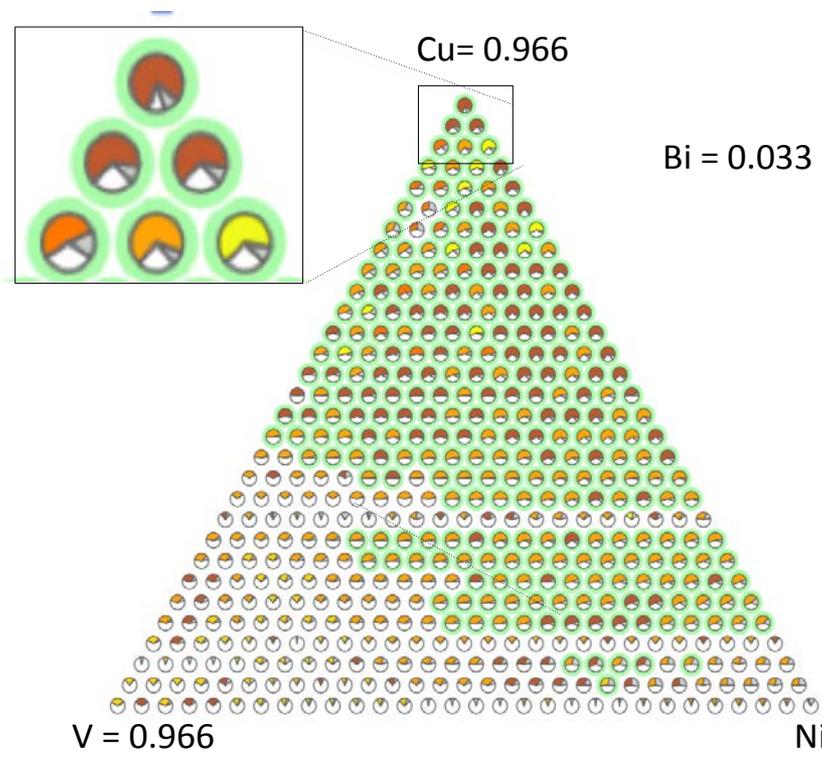
Material libraries consisting of pseudo-binary, tertiary, and quaternary compositions can be produced with physical vapor deposition methods

Color mapping: Highly parallel photoabsorber screening



- Optical absorption
- Band gap
- Quantum efficiency
- Aqueous echem.
- Band energy

In Hue – Saturation – Lightness color space, histograms reveal continuity, homogeneity



continuity **GOOD**
homogeneity **GOOD**

screen bandgap
screen absorptivity



continuity **BAD**
homogeneity **GOOD**

screen bandgap
semiquantitative absorptivity



continuity **BAD**
homogeneity **BAD**

do not scan bandgap
do not screen absorptivity

Color image processing



Smart Sampling & Database Integration

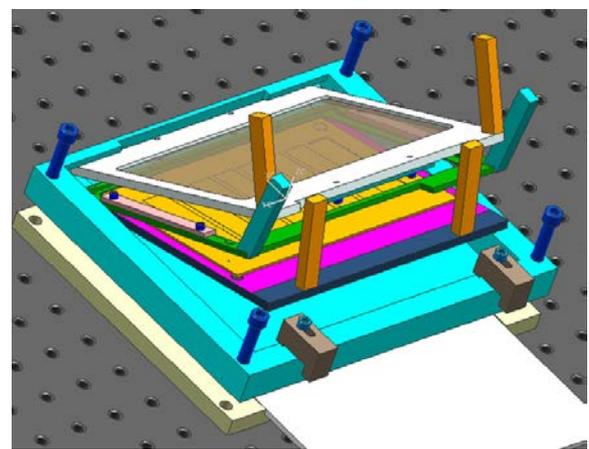
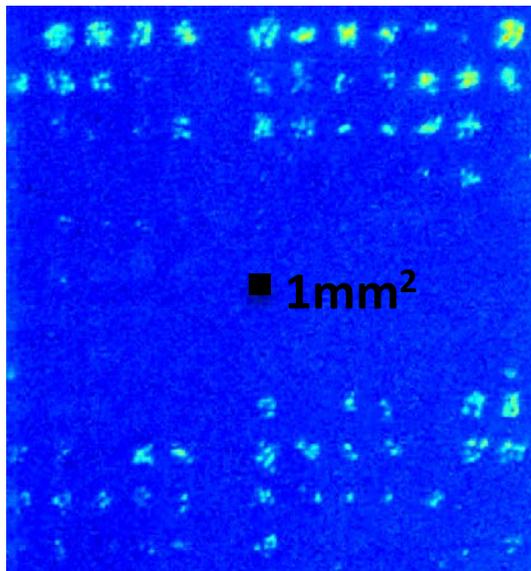
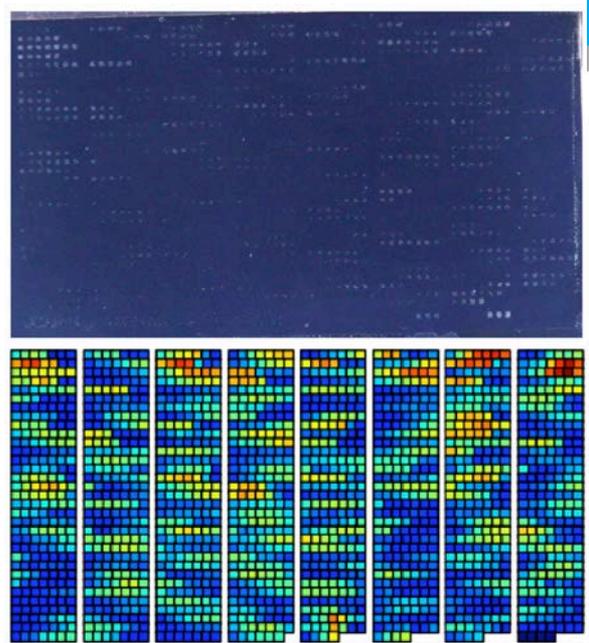
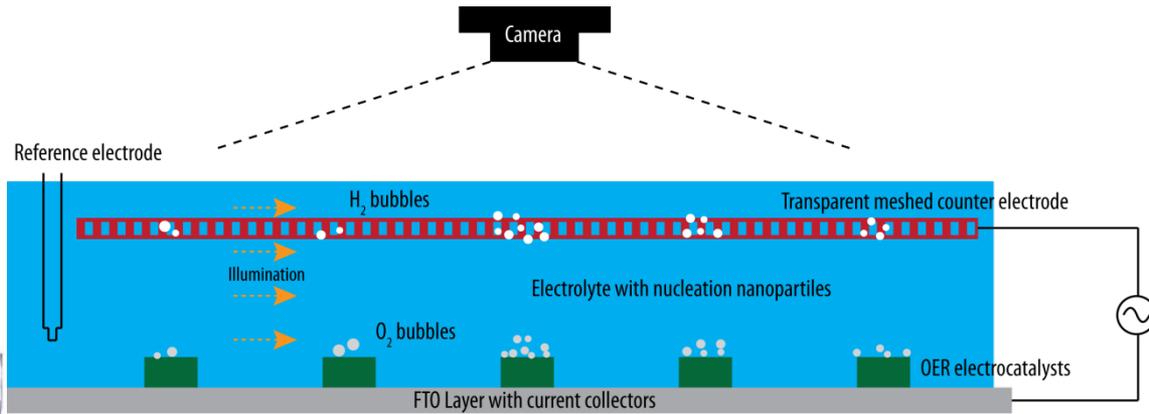


Increased UV-Vis Throughput

Bubble imaging: Highly parallel electrocatalyst screening



- Chemical Stability
- Gas production**
- Current vs. Overp.
- Voltage Stability
- Faradaic Efficiency



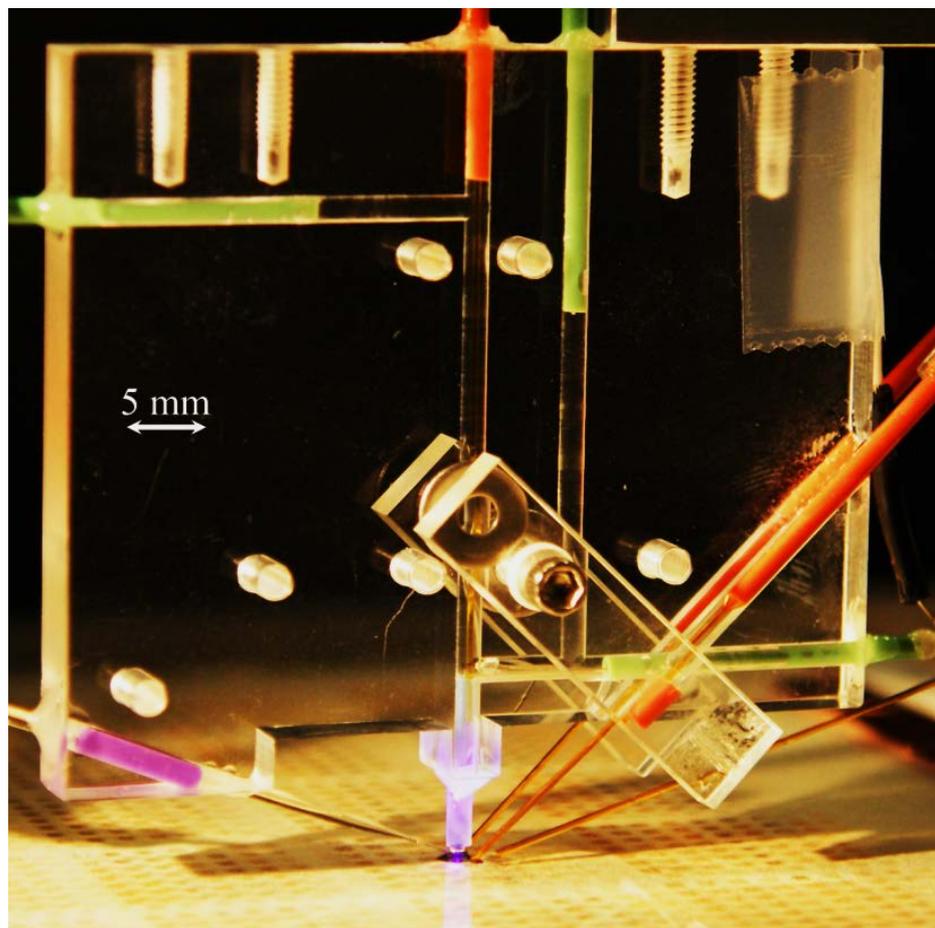
Scanning Drop Cell



Optical absorption
Band gap
Quantum efficiency
Aqueous echem.
Band energy

Chemical Stability
Gas production
Current vs. Overp.
Voltage Stability
Faradaic Efficiency

- Establishes 3 electrode cell for each sample
- Gasket-free for rapid, on-demand rastering
- Low uncompensated resistance for rapid scanning and data interpretation
- Fiber-coupled for photoelectrochemistry
- Flow cell eliminates cross contamination
- Demonstrated 1V CV at 4s per sample, **~50x faster than any previous scanning instrument**
- Complete software automation and real-time analysis

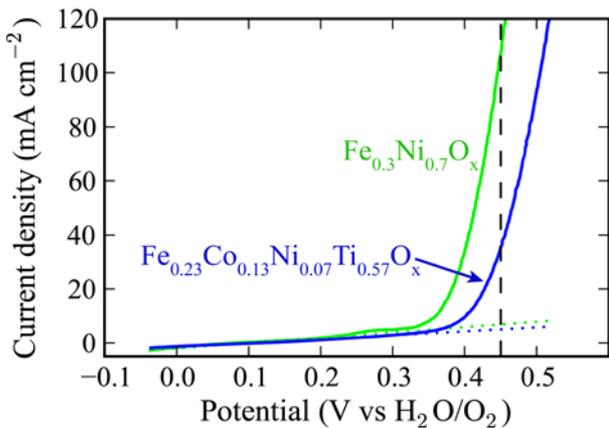


J. M. Gregoire, C. Xiang, X. Liu, M. Marcin, J. Jin, Rev. Sci. Instrum. 84, 024102 (2013)

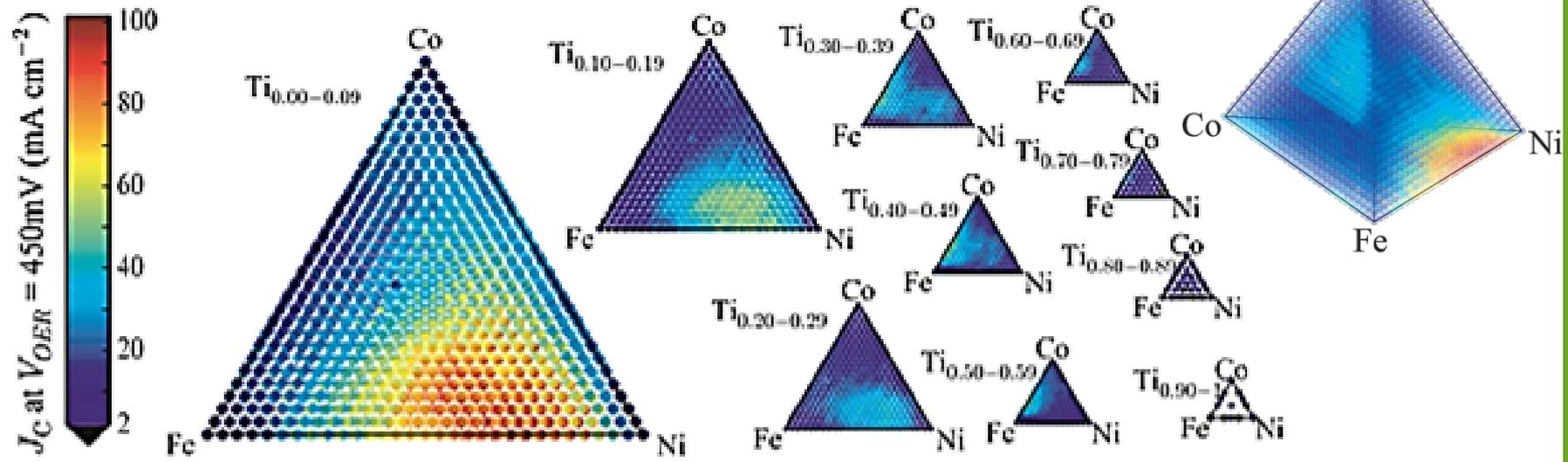
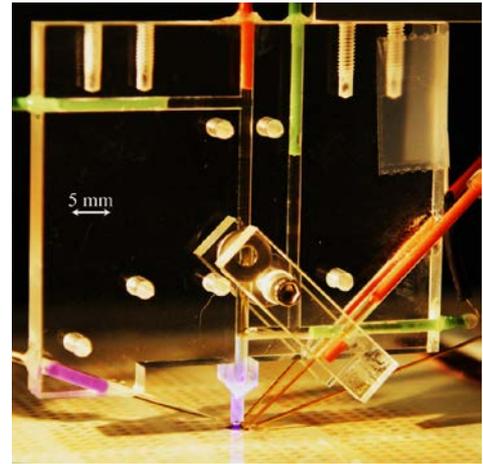
Scanning Droplet Cell: OER Electrocatalysis



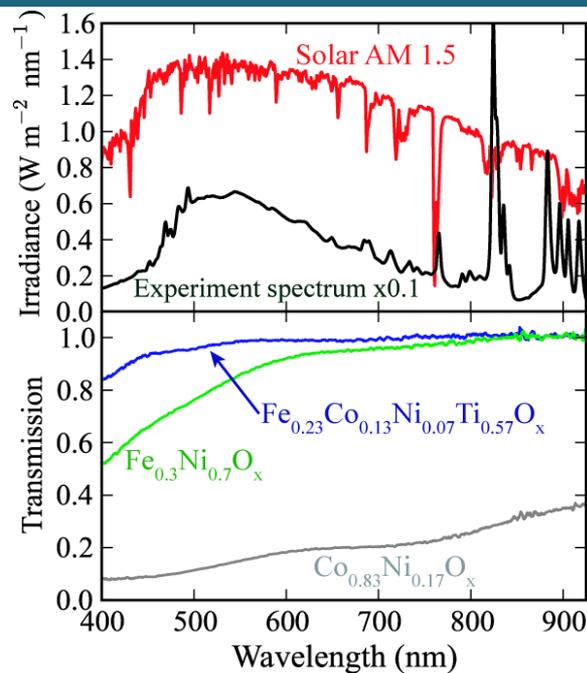
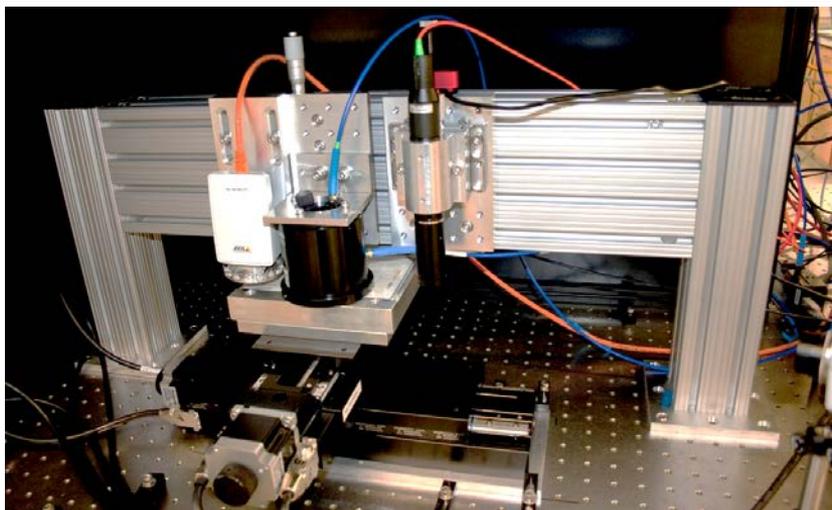
- Chemical Stability
- Gas production
- Current vs. Overp.**
- Voltage Stability
- Faradaic Efficiency



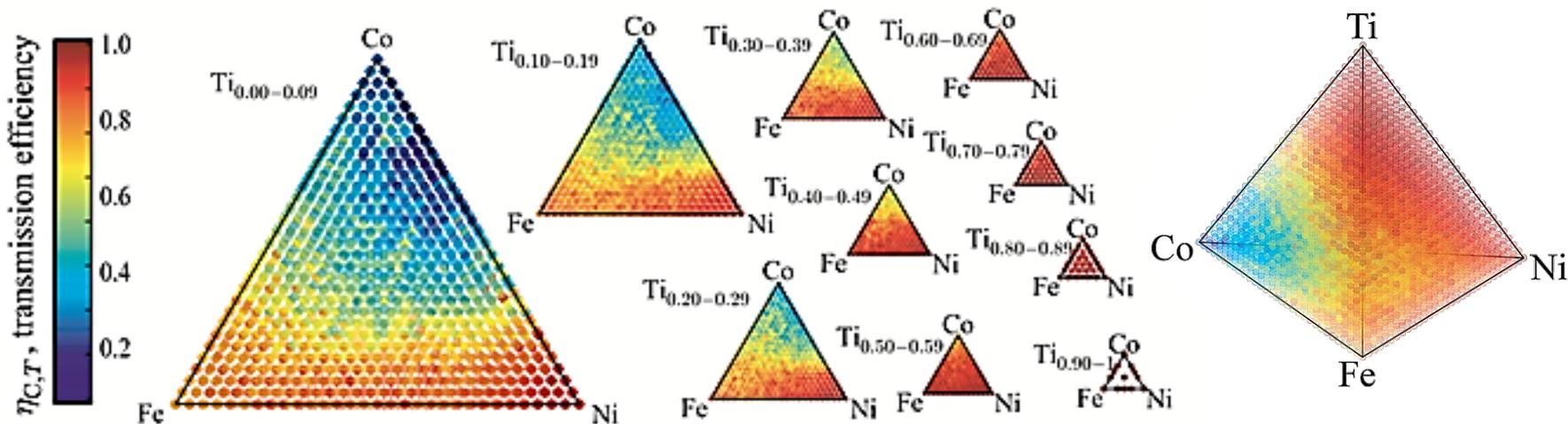
Perform CV on every sample, map current at selected overpotential



Optical Screening, In This Case For OER Transparency



What fraction of the solar energy is transmitted through the catalyst?

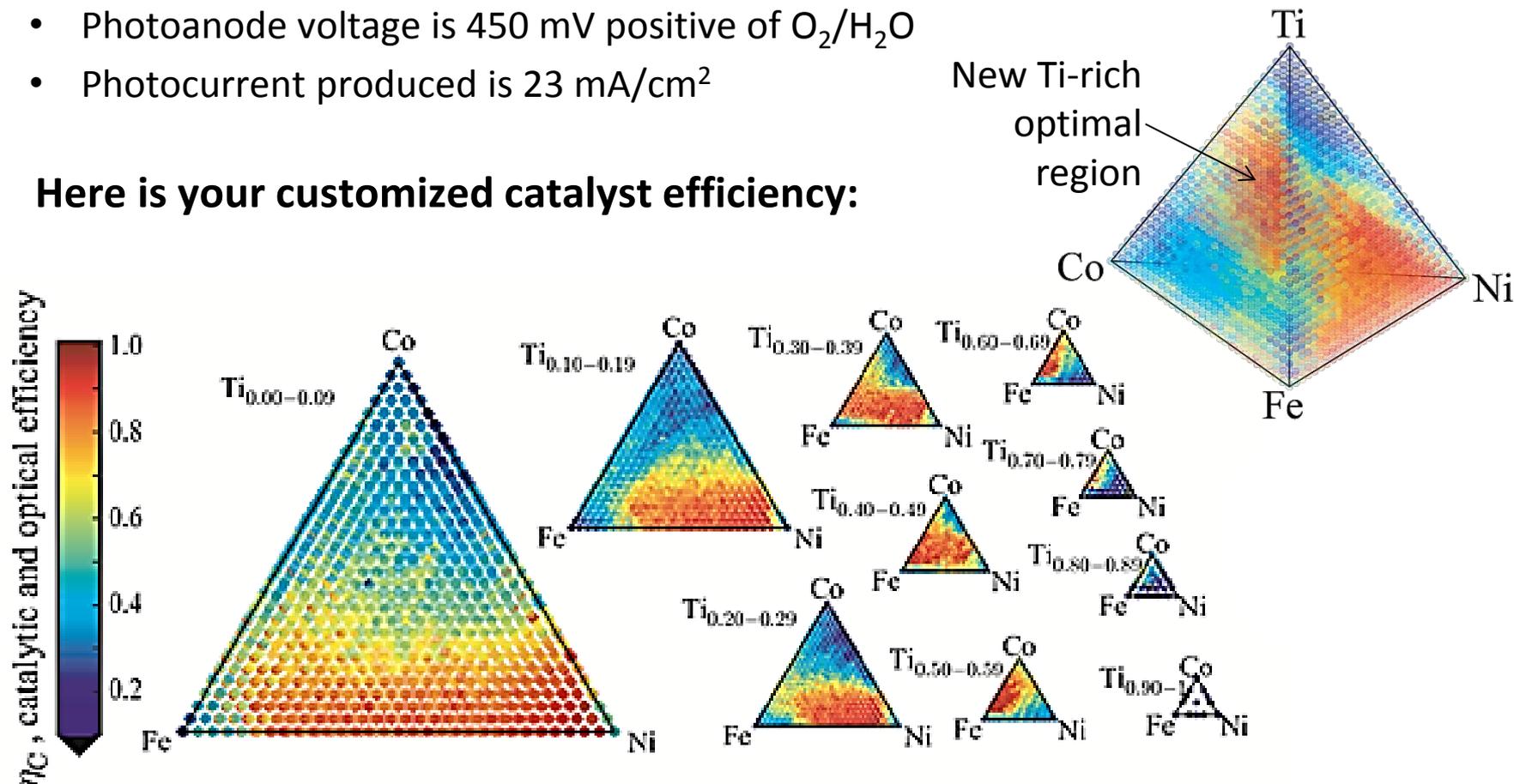


User-Defined Efficiency Model -> Database Generated Figure of Merit

Describe your light absorber device:

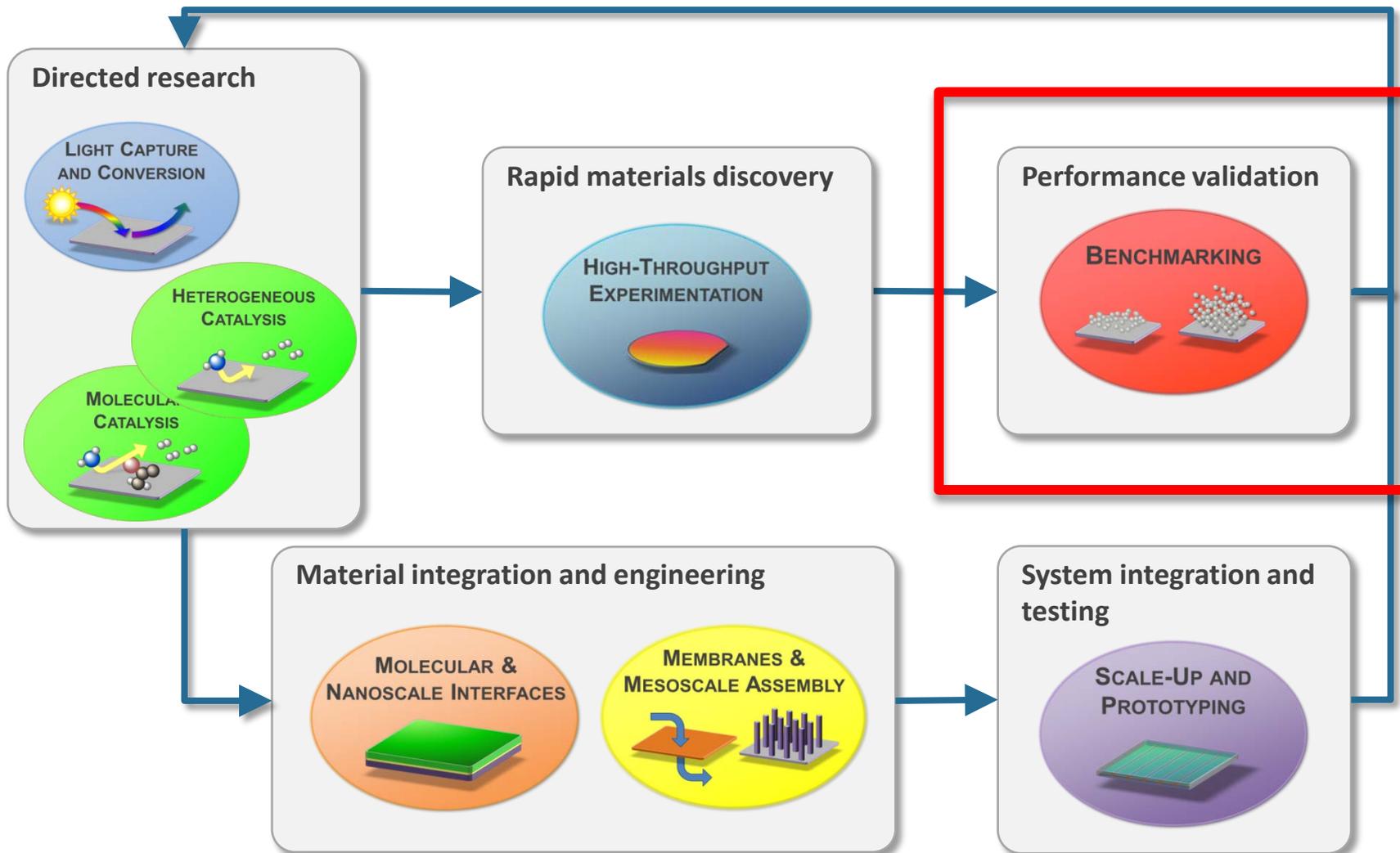
- Device operating at pH 14
- Photoabsorber stack using all solar radiation between 400 nm and 900 nm
- Photoanode voltage is 450 mV positive of O₂/H₂O
- Photocurrent produced is 23 mA/cm²

Here is your customized catalyst efficiency:



JCAP APPROACH TO ARTIFICIAL PHOTOSYNTHESIS

Projects are coordinated to accelerate materials discovery and prototype development



OVERVIEW OF BENCHMARKING PROJECT



OBJECTIVES

The Benchmarking Project serves as a community resource for the performance validation of electrocatalysts and photocatalysts. By employing a standard set of measurement protocols, unbiased evaluation by the Benchmarking Project provides comparisons that are as accurate as possible between materials/devices coming from different laboratories.

PROJECT LEADERS



Jonas
Peters



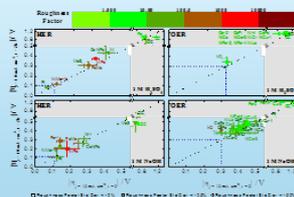
Tom
Jaramillo

APPROACH

Design of
benchmarking
standards and
protocols



Benchmarking
of
electrocatalysts



Benchmarking of light absorbers



BENCHMARKING STANDARDS AND PROTOCOLS FOR



Electrolyte Solutions:

Acidic: 1 M H₂SO₄, Alkaline: 1 M NaOH
HER preelectrolysis at -0.8 V vs SCE with sacrificial carbon

Working Electrode:

Glassy Carbon Disk (0.195 cm²)
Polished with 9 μm, 6 μm, 1 μm, and 0.1 μm diamond suspension

Auxiliary Electrode:

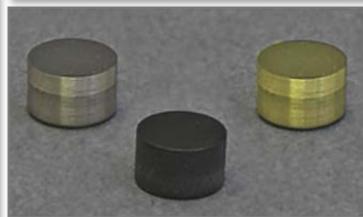
Carbon Rod or B-doped Diamond Plate
Separated from main solution with fine-porosity glass frit

Reference Electrode:

Saturated Calomel Electrode (SCE)
Externally referenced to ferrocenecarboxylic acid in 0.2 M Phosphate Buffer at pH 7 (0.28 vs SCE)

Primary Figure of Merit:

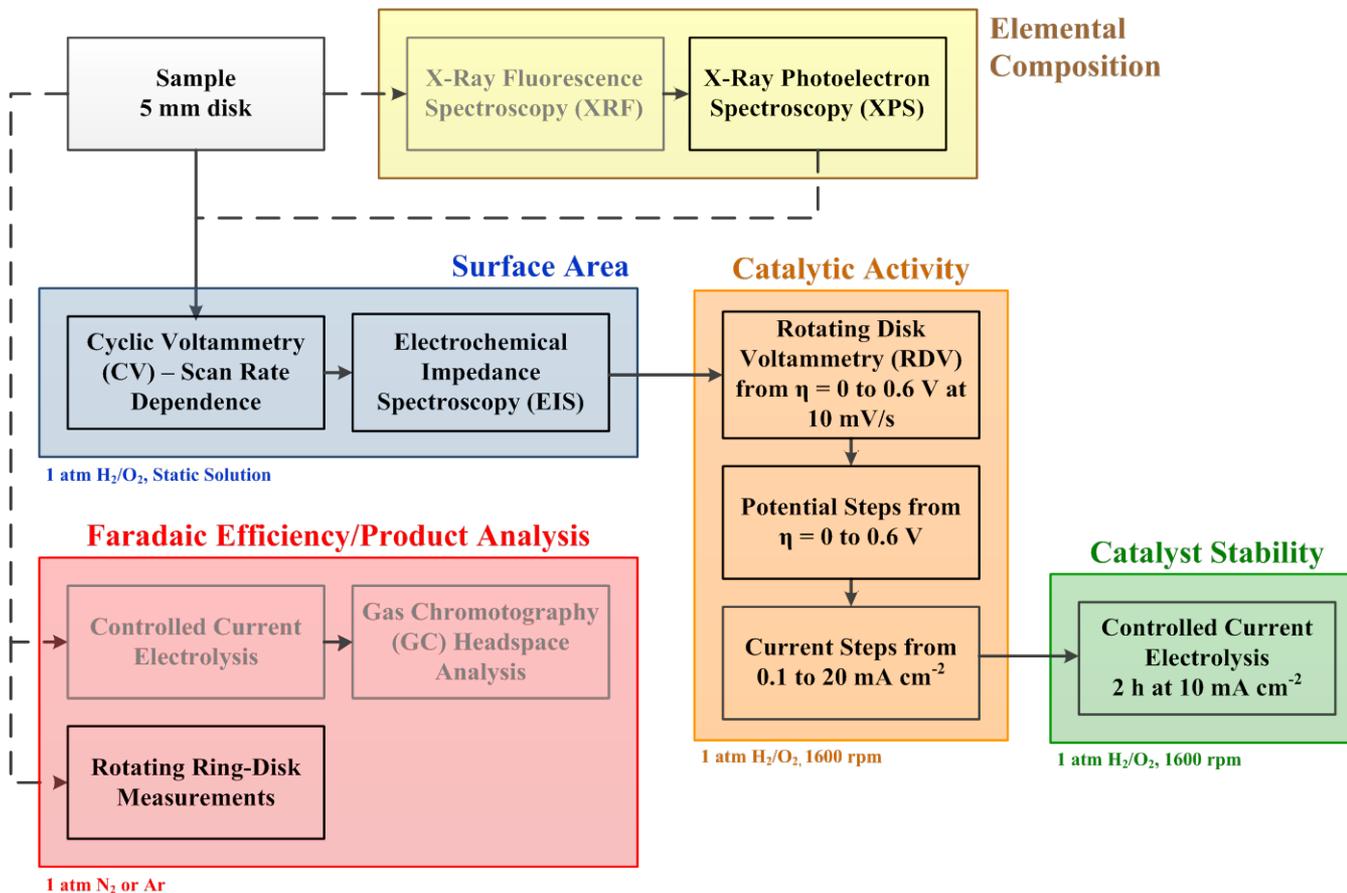
Overpotential required to achieve 10 mA cm⁻² per geometric area



BENCHMARKING METHODOLOGY

Benchmarking protocols and standards have been established for electrodeposited catalysts

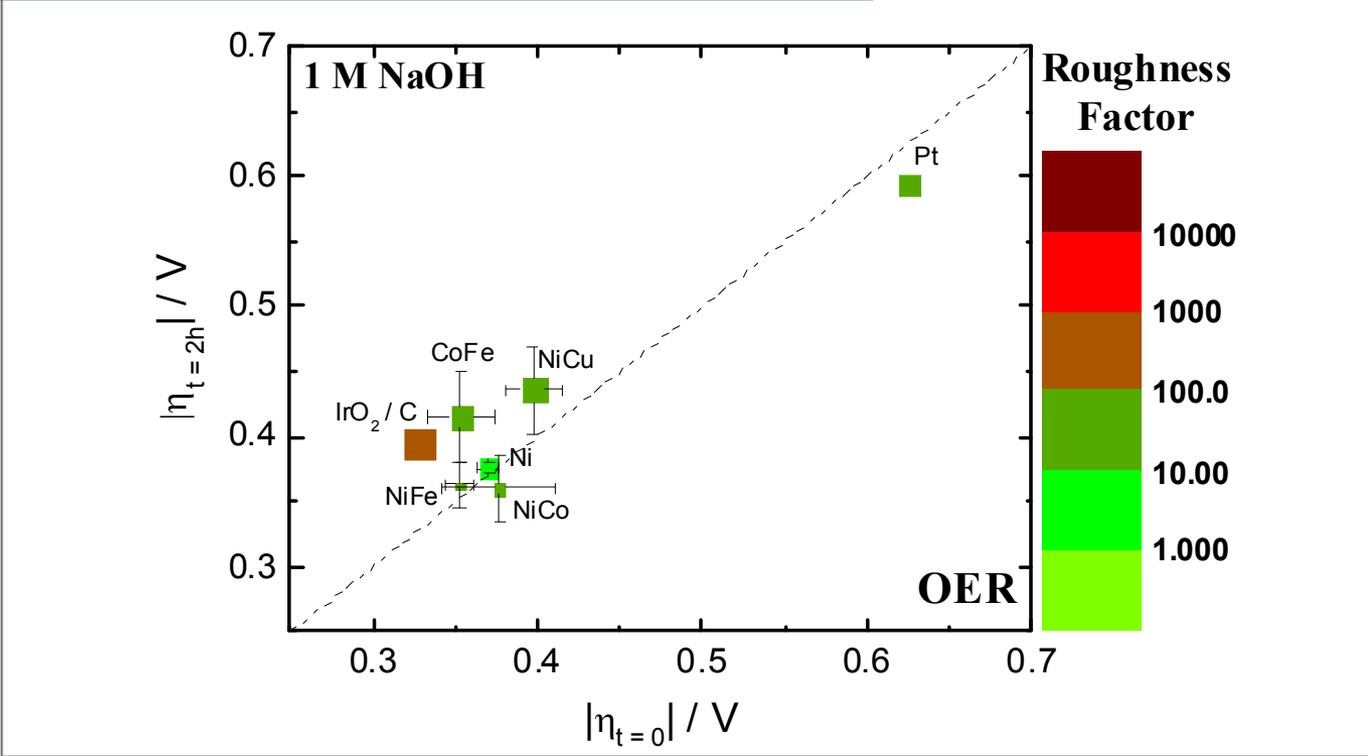
BENCHMARKING FLOWCHART



BENCHMARKED OXYGEN EVOLUTION CATALYSTS

Catalysts are now readily comparable on standardized plots

CATALYSTS FOR OXYGEN EVOLUTION UNDER ALKALINE CONDITIONS

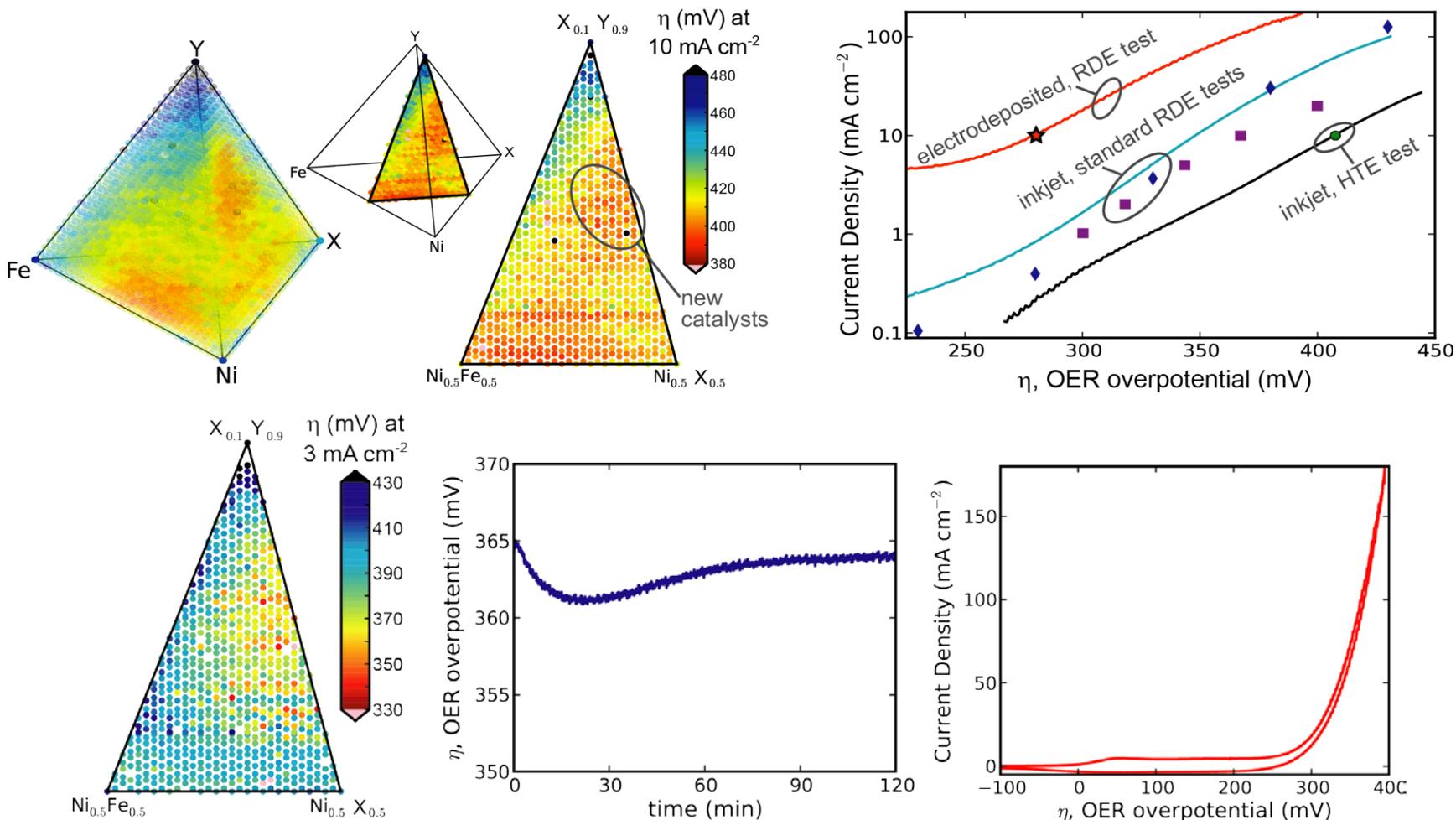


Catalytic activity and stability can now be assessed by measuring electrochemical behavior under standardized conditions

Discovery and Rapid Technological Development of a new Oxygen Evolution Catalyst

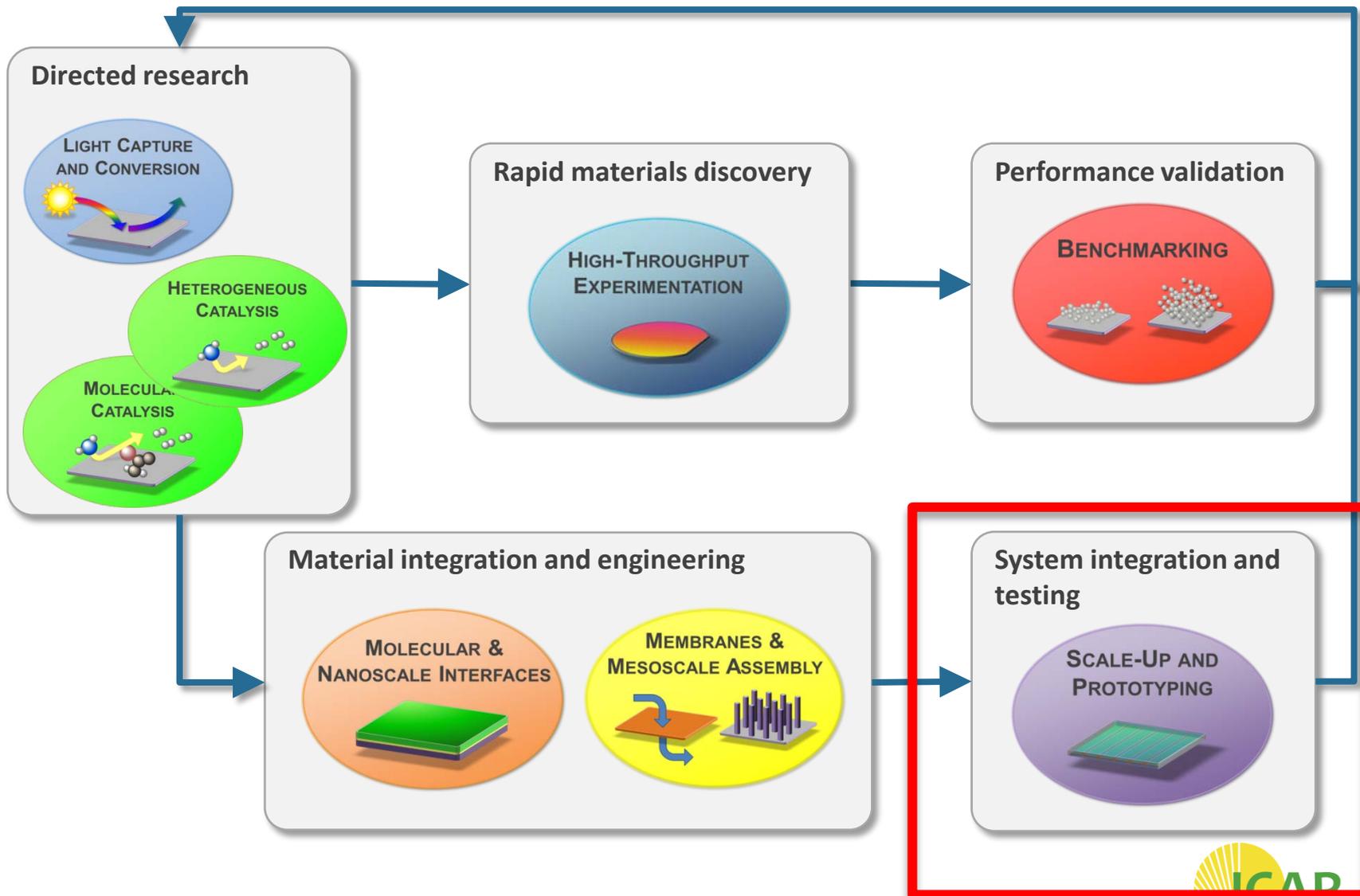
Scientific Achievement

Using high throughput experimentation (HTE), JCAP discovered a new family of electrocatalysts for the oxygen evolution reaction (OER). The catalytic activity observed via high throughput synthesis and screening was reproduced using traditional electrodeposition synthesis and traditional electrochemical experiments. Using a testbed system, solar-powered water splitting with the new OER catalyst was demonstrated for 100 hours.



JCAP APPROACH TO ARTIFICIAL PHOTOSYNTHESIS

Projects are coordinated to accelerate materials discovery and prototype development



Project Overview – Scale-up and Prototyping



The goal of the Scale-up and Prototyping project is to develop robust, high-performance, scalable solar-fuels generators by implementing systems engineering methodologies and actively integrating newly discovered robust and Earth-abundant materials and components.

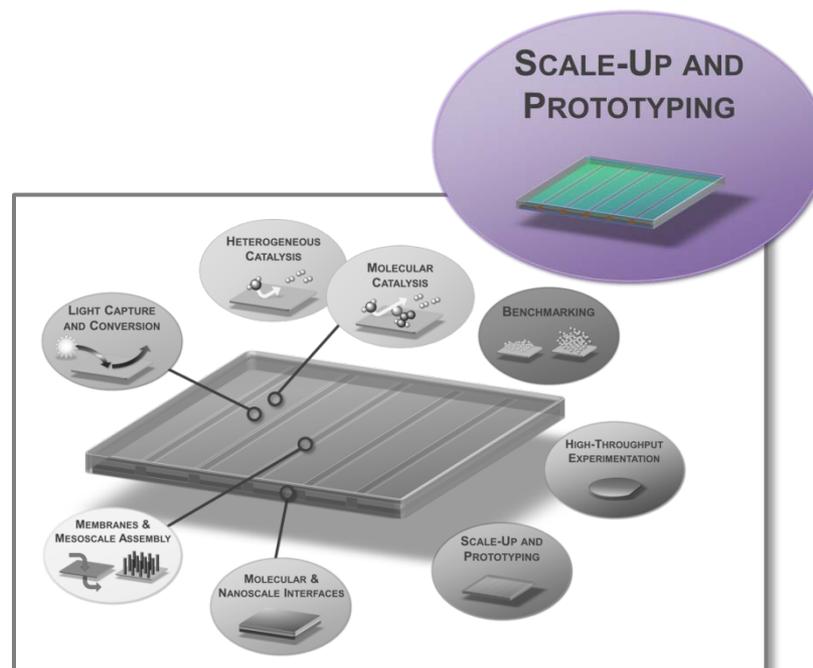
Project Leaders



Jian Jin

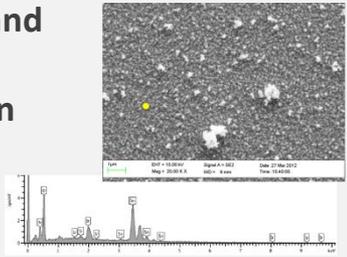


Will West



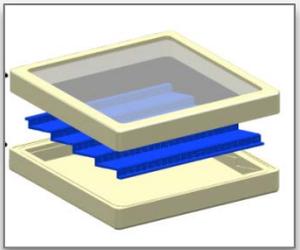
JCAP APPROACH TO SYSTEM INTEGRATION OF SOLAR-FUELS GENERATORS

Material and process integration

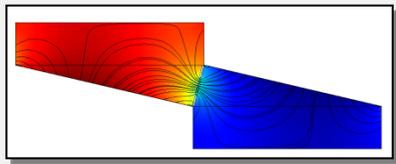


Element	Atomic %
Na	2.51
O	66.13
Si	6.35
Sn	20.93
H	3.08
Total	100.00

Early design test beds and feasibility studies



Multi-dimensional, multi-physics modeling

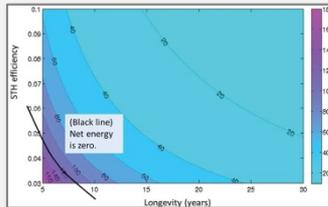


0.12
0.11
0.1
0.09
0.08
0.07
0.06
0.05

Fabrication and characterization of prototypes



Investigations of manufacturing, scale-up, and sustainability



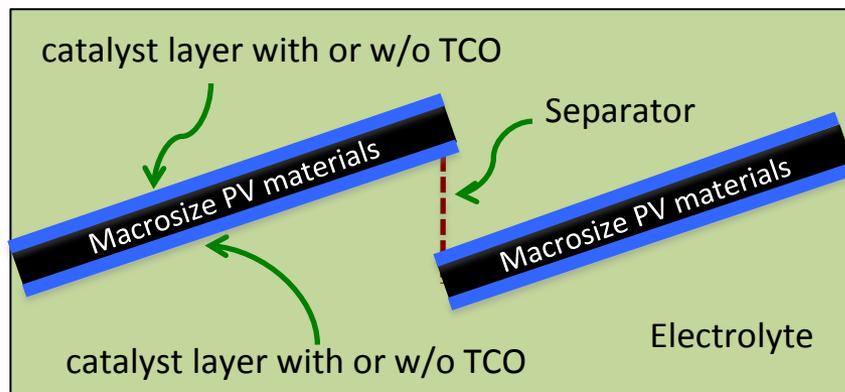
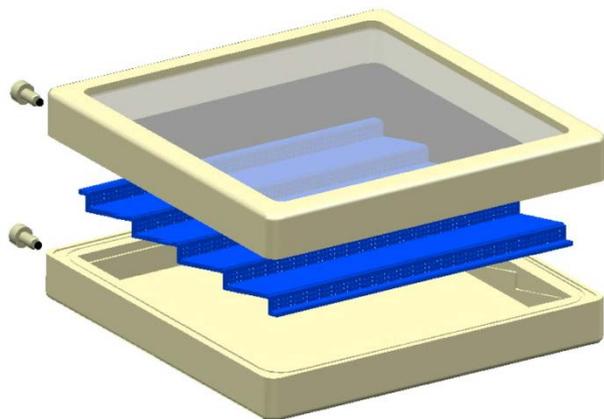
STH efficiency

Longevity (years)

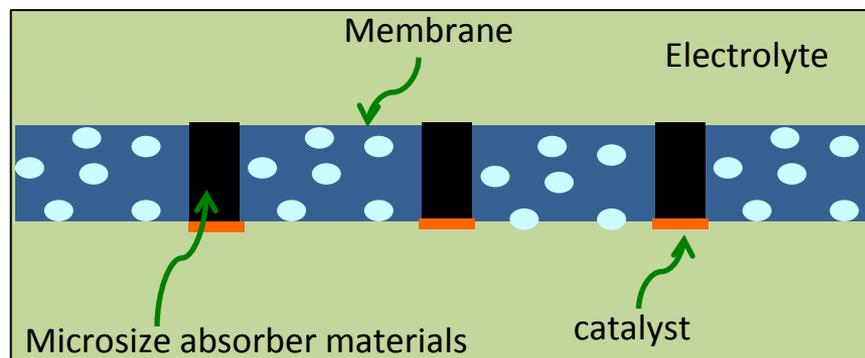
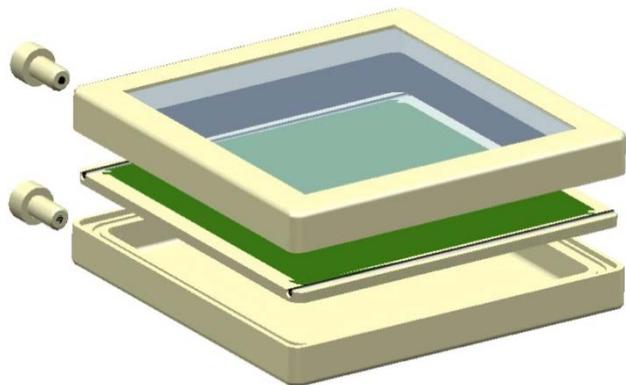
(Black line)
Net energy is zero.

Prototype Designs for a Solar-Driven Water-Splitting Device

Louvered prototype

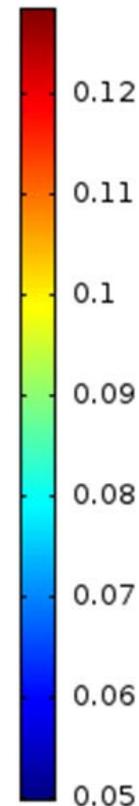
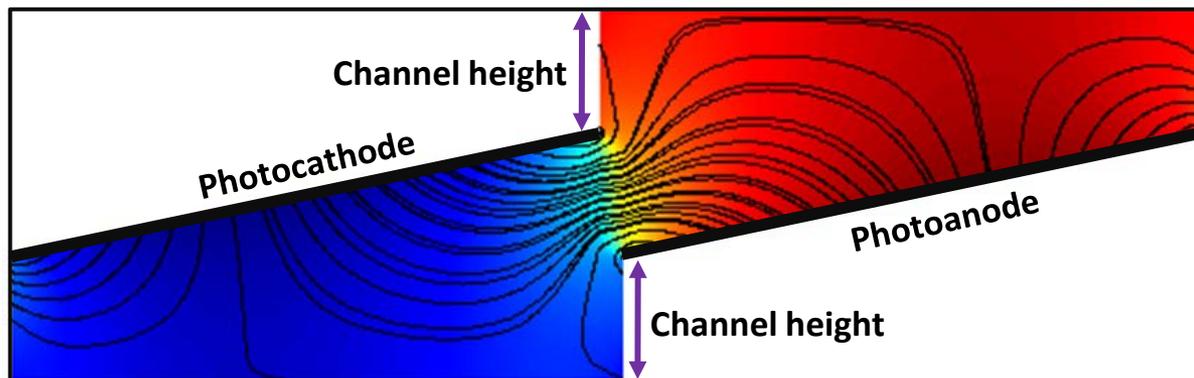


Absorber-in-membrane prototype



Research Highlight: Modeling of Resistive Losses

Electrolyte potential (V) and current density (stream lines) for the louvered prototype design

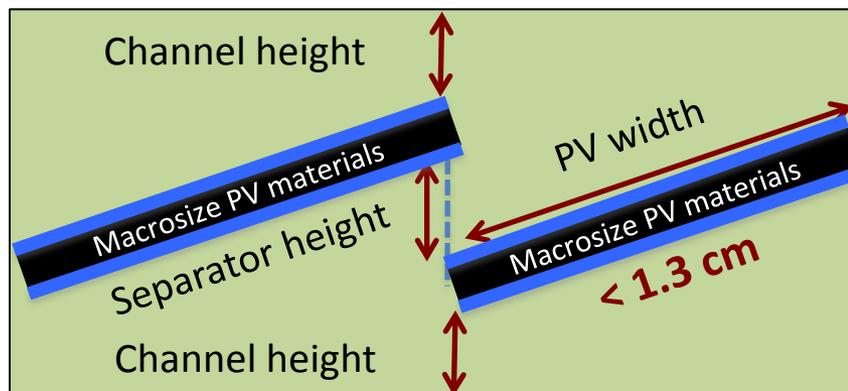
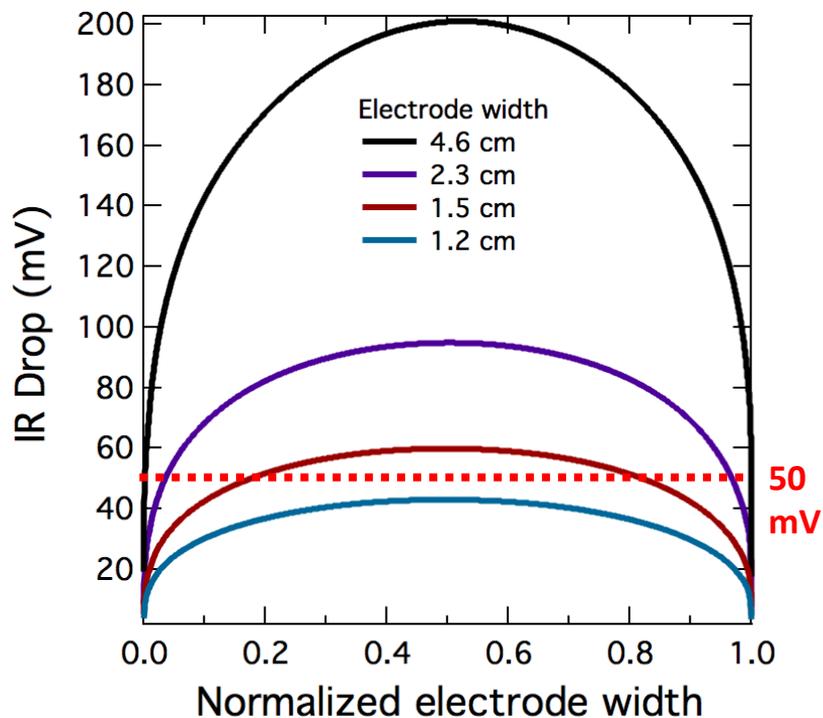


Semiconductor width = 2 cm
Nafion height = 4 mm
Channel height = 4 mm



Computational modeling using Poisson's equation solver in COMSOL Multiphysics provides electrical potential and current distributions

Research Highlight: Optimization of Electrode Geometries

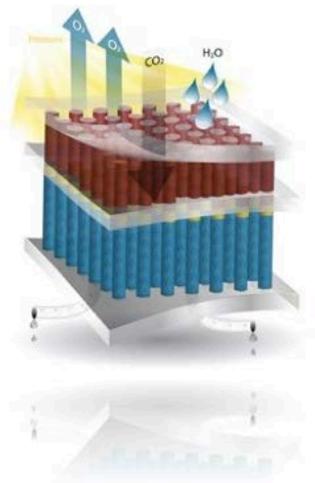


Separator height = 4 mm
Channel height = 4 mm

not to scale

For the louvered prototype design, the electrode width cannot exceed 1.3 cm for a 50 mV potential drop

Welcome to JCAP Modeler

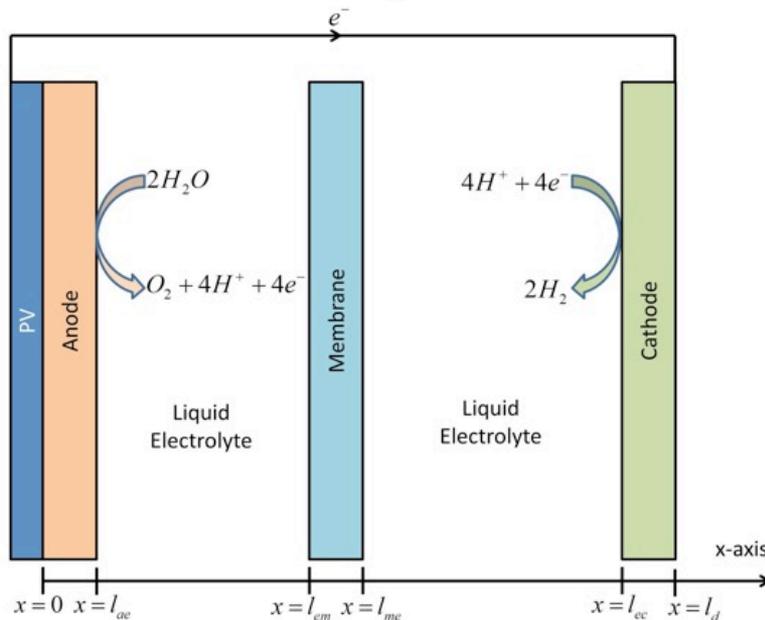


Modeling using JCAP Modeler

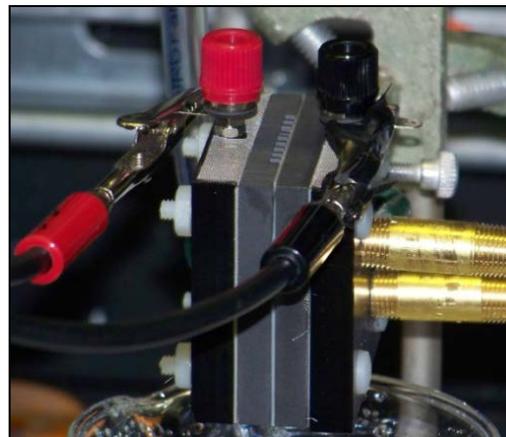
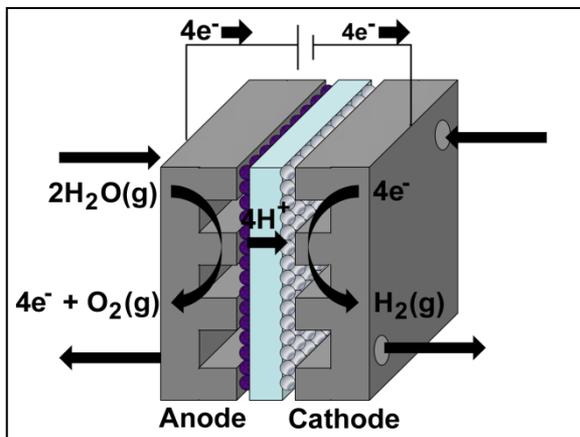
JCAP Modeler is a web application to study the behavior of various photoelectrochemical water-splitting devices. Description of each model and associated parameters are given on a designated webpage. Currently, the JCAP Modeler is capable to simulate:

Description of 1D Model

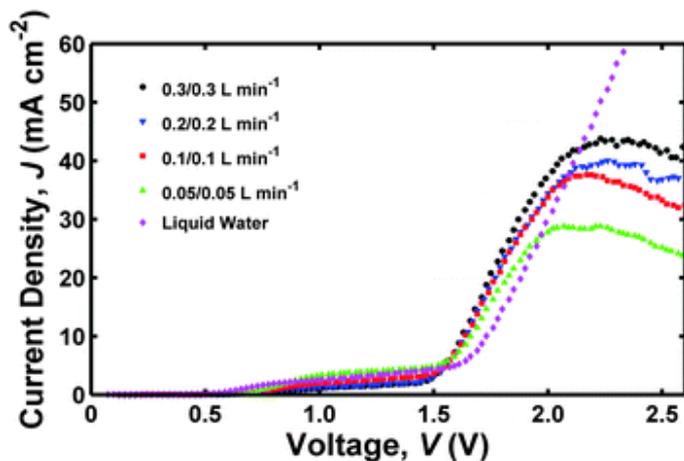
Schematic Diagram



Research Highlight: Solar-Fuels Generation Using Water Vapor



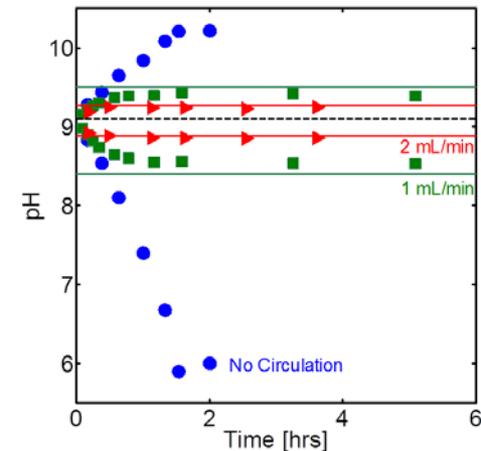
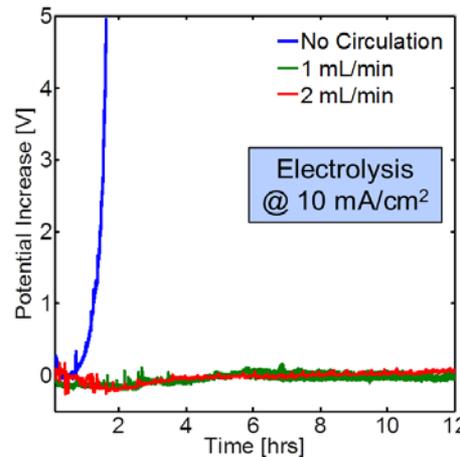
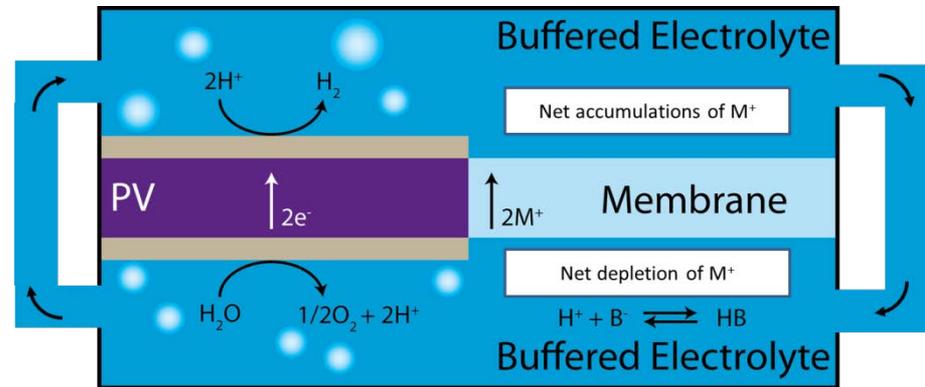
Test set-up for proton exchange membrane electrolysis by water vapor.



An efficient solar photoelectrolyzer can be operated using water vapor in place of liquid water

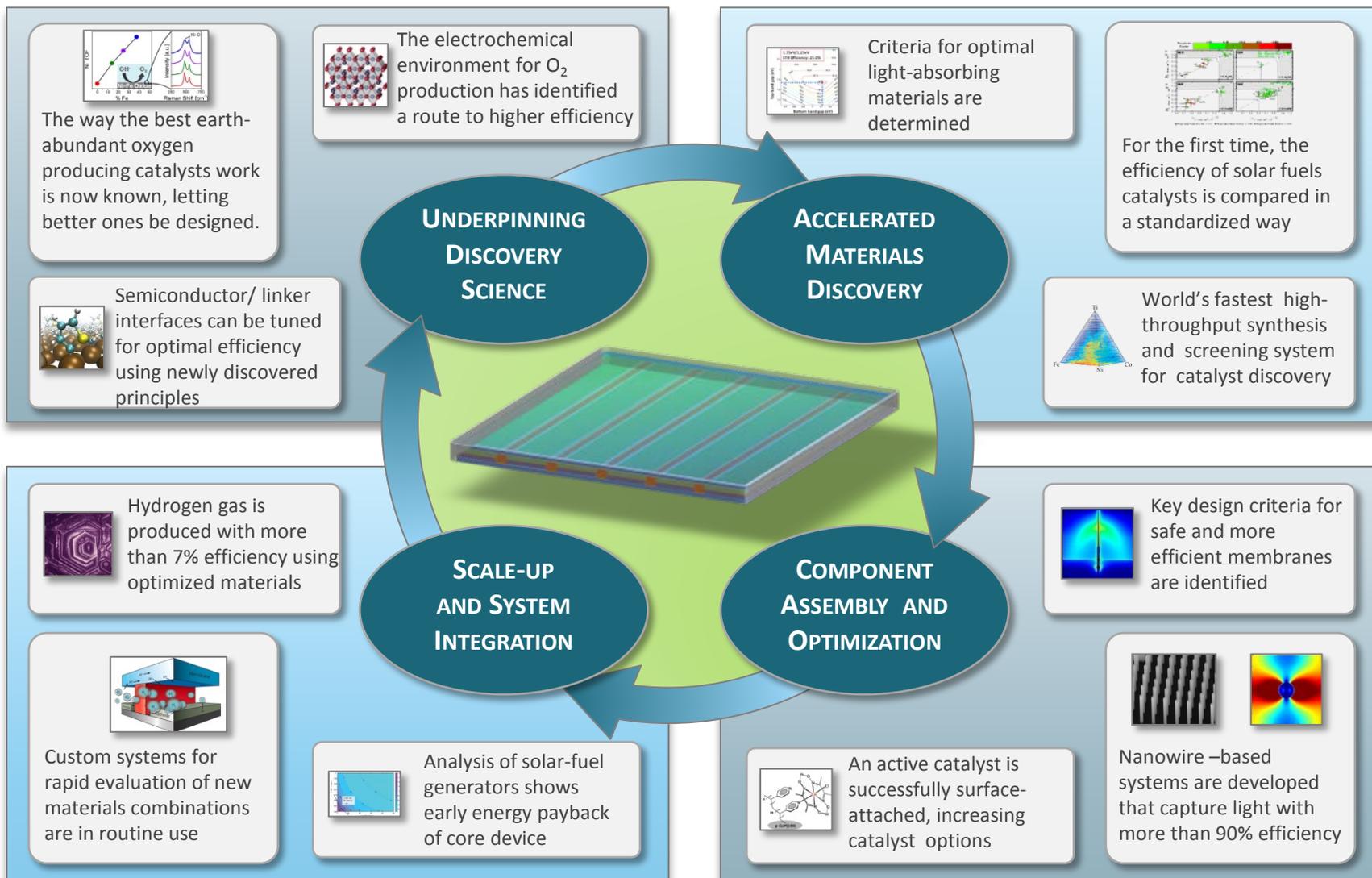
DEVELOPMENT AND TESTING OF A RECIRCULATING PROTOTYPE

- (1) Membrane separation of H_2 and O_2 compartments necessitates either high or low pH solutions
- (2) Corrosion has been identified as a key life-limiting mechanism
- (3) Buffered pH systems yield vast improvements in semiconductor/catalyst lifespan, but shut down due to membrane transport limitations
- (4) A design was explored to allow for buffered mid-pH solutions
 - (1) Leveraged direct technical input from JCAP Membrane team members (Led by Miguel Modestino)
 - (2) Includes a membrane with a bypass recirculation system
 - (3) Explosive gas mixtures are avoided by engineering recirculation physically distant from electrodes
- (5) Demonstrated that with minimal recirculation, can avoid concentration gradients over diurnal cycles at relatively low recirculation rates



Recirculation Model design (above) and electrolysis data as a function of recirculation rate (below).

OVERVIEW OF RECENT DISCOVERIES IN THE JOINT CENTER FOR ARTIFICIAL PHOTOSYNTHESIS



The Joint Center for Artificial Photosynthesis (JCAP) is the nation's largest research program dedicated to the development of an artificial solar-fuel generation technology. Established in 2010 as a U.S. Department of Energy (DOE) Energy Innovation Hub, JCAP aims to find a cost-effective method to produce fuels using only sunlight, water, and carbon-dioxide as inputs. JCAP is led by a team from the California Institute of Technology (Caltech) and brings together more than 140 world-class scientists and engineers from Caltech and its lead partner, Lawrence Berkeley National Laboratory. JCAP also draws on the expertise and capabilities of key partners from Stanford University, the University of California campuses at Berkeley (UCB), Irvine (UCI), and San Diego (UCSD), and the Stanford Linear Accelerator (SLAC). In addition, JCAP serves as a central hub for other solar fuels research teams across the United States, including 20 DOE Energy Frontier Research Center.

For more information, visit <http://www.solarfuelshub.org>.



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