

# **Development of a Novel Efficient Solid-Oxide Hybrid for Co-generation of Hydrogen and Electricity Using Nearby Resources for Local Application**

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Materials & Systems Research Inc., Salt Lake City, UT

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**Project ID#: PDP 33**

# Overview

## Timeline

- Project started: 02/10/2006
- Project ends: 07/31/2009
- Percent completed: 70%

## Budget

- Total budget funding
  - DOE \$2,480k
  - Contractor \$ 620k
- Funding received in FY07
  - \$ 1,265k
- Funding for FY08
  - \$ 755k

## Barriers

Hydrogen generation by water electrolysis

- G – Capital cost
  - Low-cost, durable high-temperature materials development
  - Lower operating temperature
- H – System Efficiency
- K – Electricity costs

## Partners

- University of Alaska Fairbanks – anode supports fracture mechanisms and modeling of residual stresses (S. Bandopadhyay; N. Thangamani)
- University of Missouri-Rolla – cathode & seal materials development (H. Anderson; R. Brow)
- University of Utah – interconnect development (A. Virkar)

# Objective

Overall Objective	<ul style="list-style-type: none"><li>• To develop a low-cost and highly efficient 5 kW SOFC-SOFEC hybrid system co-generating both electricity and hydrogen to achieve the cost target of &lt; \$3.00/gge when modeled with a 1500 gge/day hydrogen production rate.</li><li>• The project focuses on materials R&amp;D, stack design &amp; fabrication, and system design &amp; verification.</li></ul>
2007	<ul style="list-style-type: none"><li>• 5 kW SOFC-SOFEC hybrid system development<ul style="list-style-type: none"><li>– Materials development and application (electrodes &amp; seals)</li><li>– Stack design and development</li><li>– Hybrid system design</li><li>– BOP components design and development</li></ul></li></ul>
2008	<ul style="list-style-type: none"><li>• 5 kW SOFC-SOFEC hybrid system fabrication and assembly<ul style="list-style-type: none"><li>– Cell &amp; non-cell repeat units fabrication</li><li>– BOP components fabrication</li><li>– Stack assembly and integration</li><li>– Hybrid module evaluation</li><li>– Control system assembly &amp; programming</li></ul></li></ul>

# Milestones

Quarters, FY	Milestone or Go/No-Go Decision
1 <sup>st</sup> Quarter, FY07	Go/No-Go decision: assess the appropriate load and strains that SOFC-SOFEC can withstand.
2 <sup>nd</sup> Quarter, FY07	Go/No-Go decision: assess the viability of 1 kW hybrid module for cogeneration based on performance.
4 <sup>th</sup> Quarter, FY07	Go/No-Go decision: finalize the cathode system for hybrid application.
4 <sup>th</sup> Quarter, FY07	Milestone: Complete the design of the 5 kW system and major BOP components. Down select two “invert” glass compositions from 81 compositions that are thermal-mechanically compatible and thermal-chemically stable. Finalize the cathode system for SOFEC applications.
3 <sup>rd</sup> Quarter, FY08	Milestone: Complete the high temperature flexural tests and creep studies of SOFEC anode substrates.
4 <sup>th</sup> Quarter, FY08	Milestone: Complete fabrication of all cell/non-cell repeat units. Complete fabrication and pre-test of BOP components. Transfer the glass seal technology from the subcontractor to MSRI.

# Approach

## Materials Development

- A. Cathode materials Dev.
- B. Anode optimization
- C. Electrolyte optimization
- D. Catalyst studies
- E. Seals development
- F. Fabrication Q.A.

80% complete

## Cell / Stack / System Design

- A. Stack design
- B. 5kW system design
- C. BOP design/dev.
- D. Stresses analyses
- E. Seals application
- F. Economic analysis

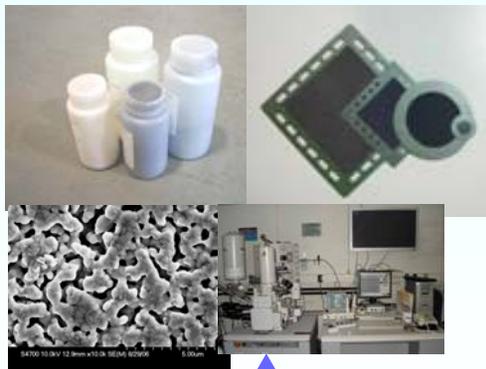
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## Experimental Verification

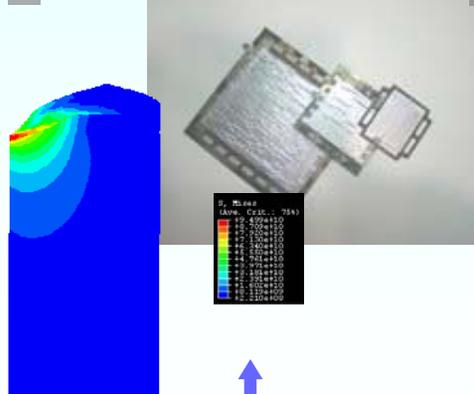
- A. Short stacks in dif. modes
- B. 1 kW hybrid stack
- C. Durability evaluation
- D. BOP design & evaluation
- E. 5 kW hybrid system development & evaluation

50% complete

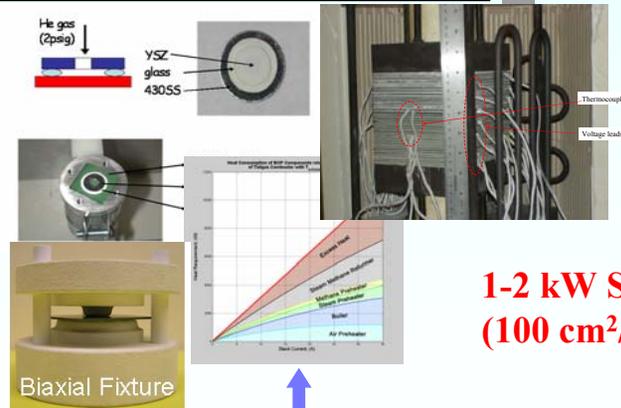
Success



MSRI, UMR



MSRI, UAF, UMR, UU

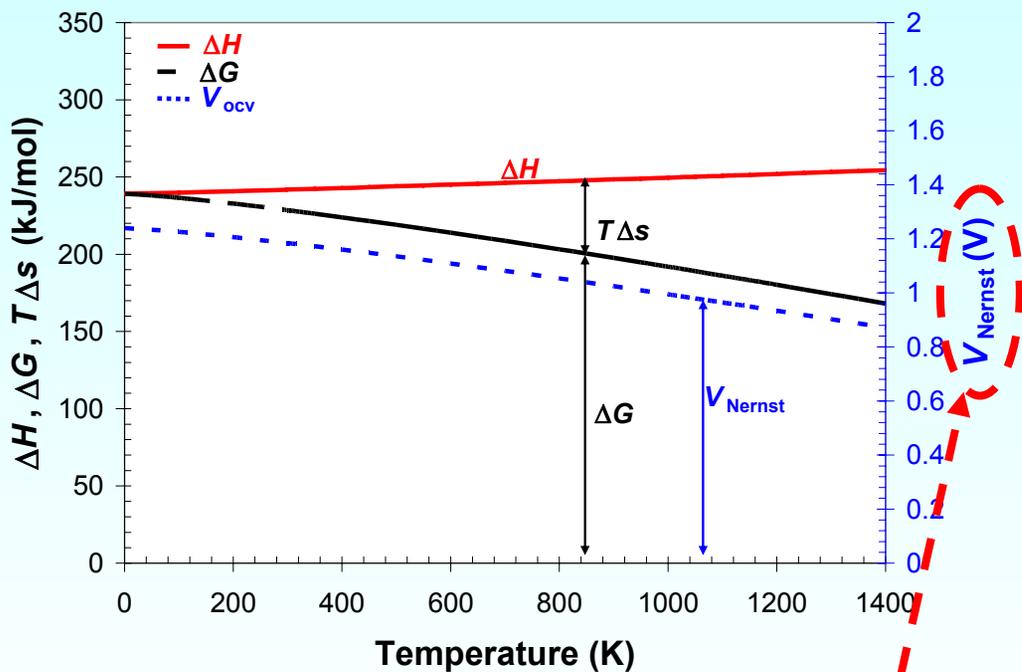


1-2 kW Stack  
(100 cm<sup>2</sup>/cell)

MSRI, UU, UMR

# Background

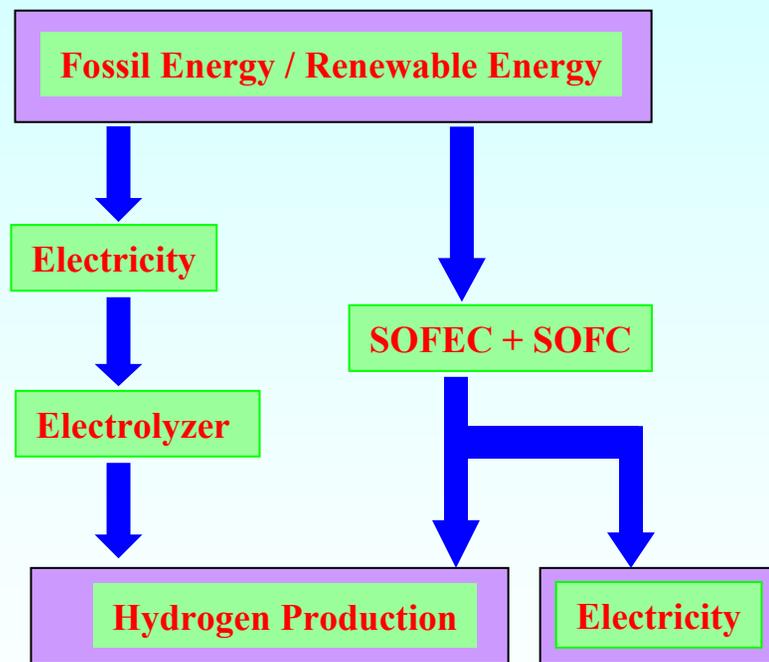
A Solid Oxide Fuel-Assisted Electrolysis Cell (SOFEC) directly applies the energy of a chemical fuel to replace the external electrical energy required to produce hydrogen from water/steam; decreasing the cost of energy relative to a traditional electrolysis process.



Electricity from Grid

Electrochemical Process  
at cathode  
at anode

Unique process



Co-generation

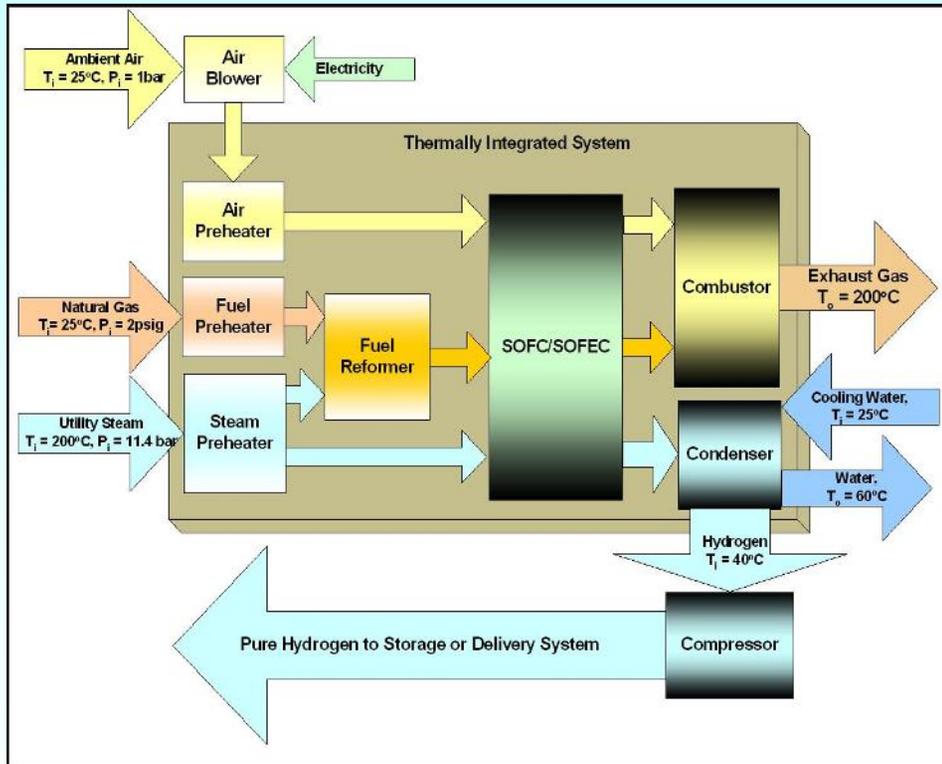
CH<sub>4</sub>-assisted SOFEC Reaction



Pure H<sub>2</sub> formed. No need for H<sub>2</sub> separation membranes. Lower electricity requirement.

# Concept of Hybrid SOFC-SOFEC Integral System

## Technical Challenges and Solutions



### Cost of Hydrogen

#### SOFC-SOFEC performance

- ❖ cathode materials: composition, microstructure, catalytic characteristics;
- ❖ anode material: porosity, tortuosity, composition;
- ❖ low operating temperature: inexpensive materials, non-precious metal catalysts;
- ❖ manufacturability.

#### System efficiency

- ❖ SOFC-SOFEC hybrid architecture;
- ❖ thermal integration;
- ❖ co-generation concept.

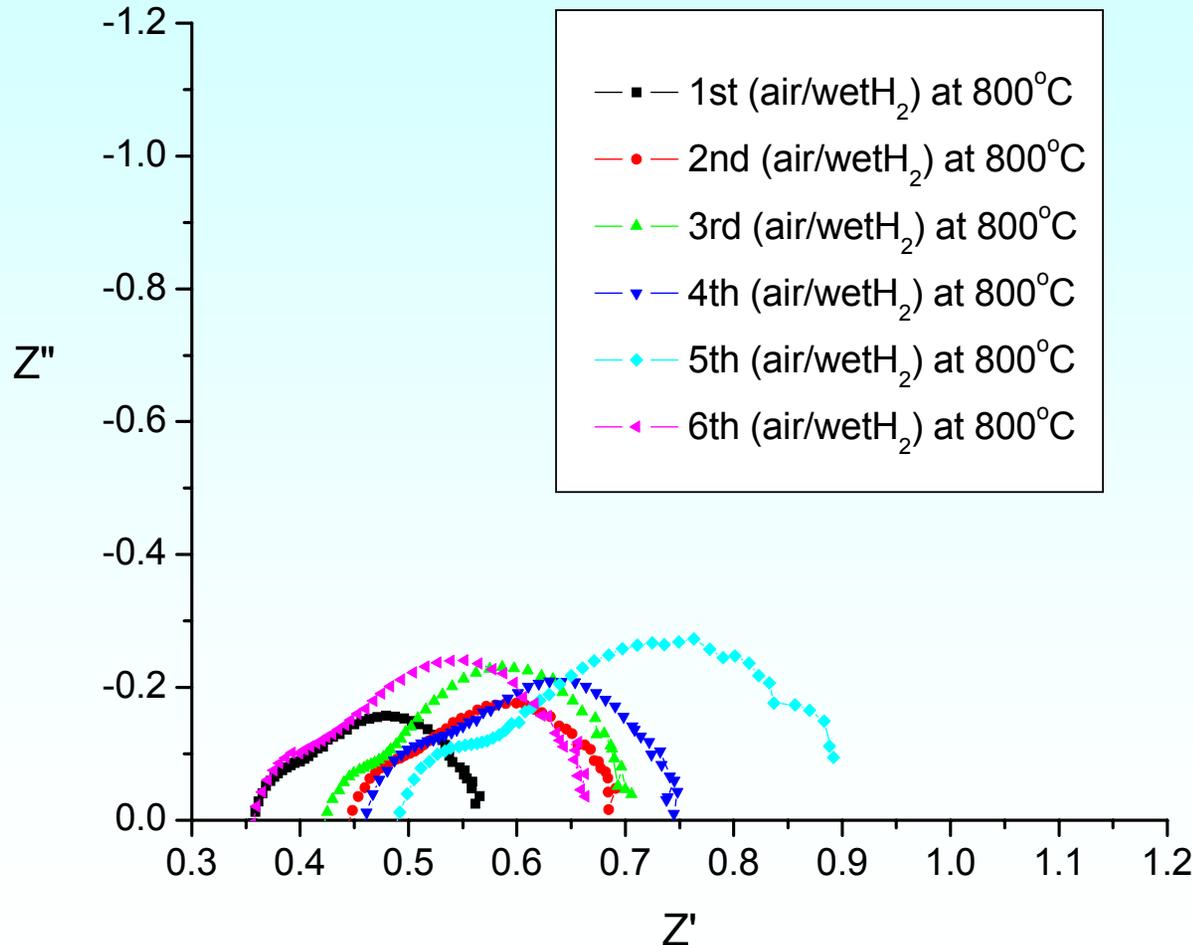
#### Long-term durability

- ❖ "invert" seal material thermalmechanical compatibility and thermalchemical stability;
- ❖ anode mechanical strength.

- Pure H<sub>2</sub> & electricity co-production from feedstock: hydrocarbon fuel, steam, and air
- Hybrid comprised of SOFCs and SOFECs
- SOFECs produce pure H<sub>2</sub> and SOFCs generate electricity for a high H<sub>2</sub> production rate
- Thermal integration improves system efficiency

# SOFC Cathode Materials Development

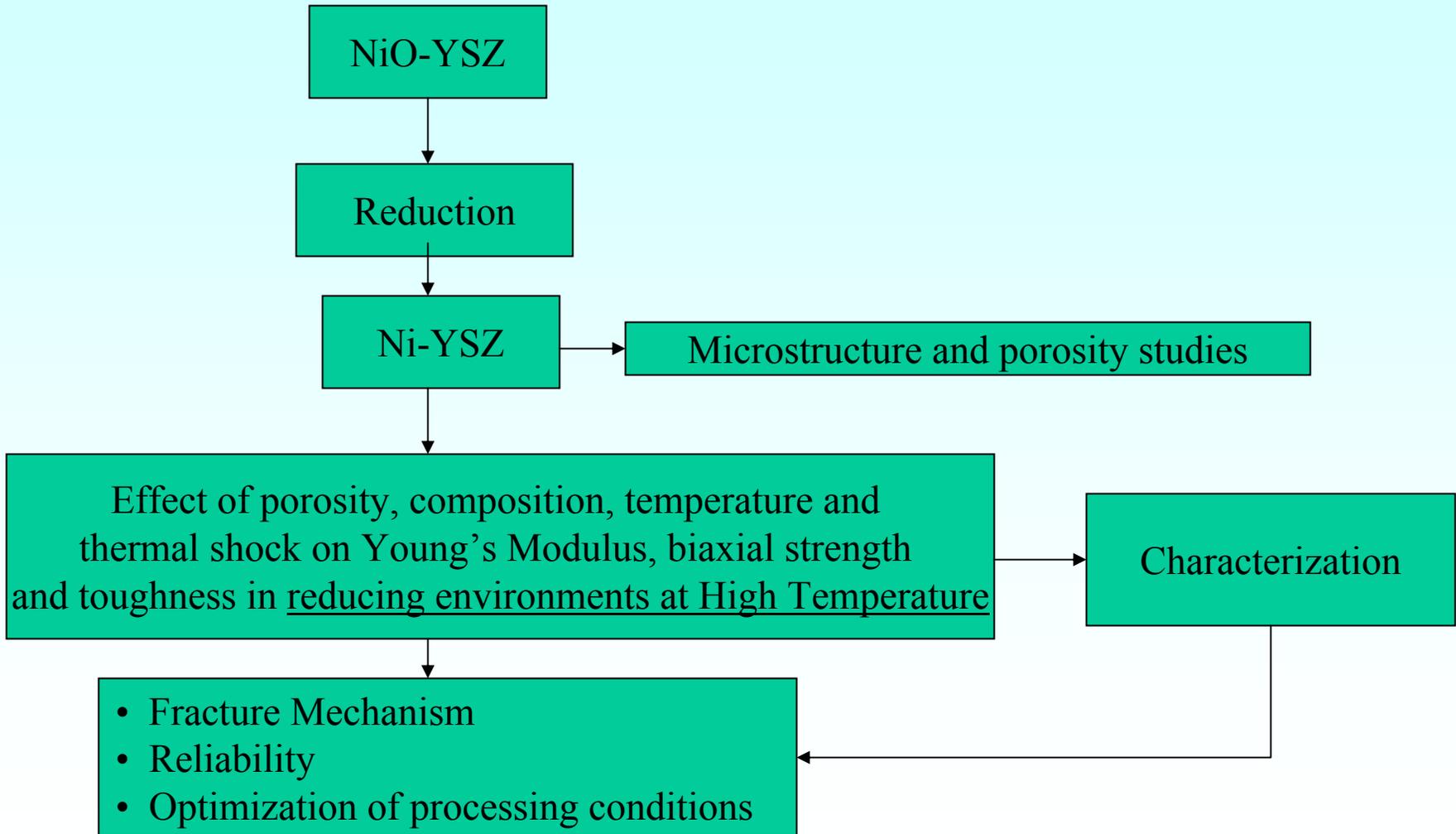
## LSCM Redox Stability Study



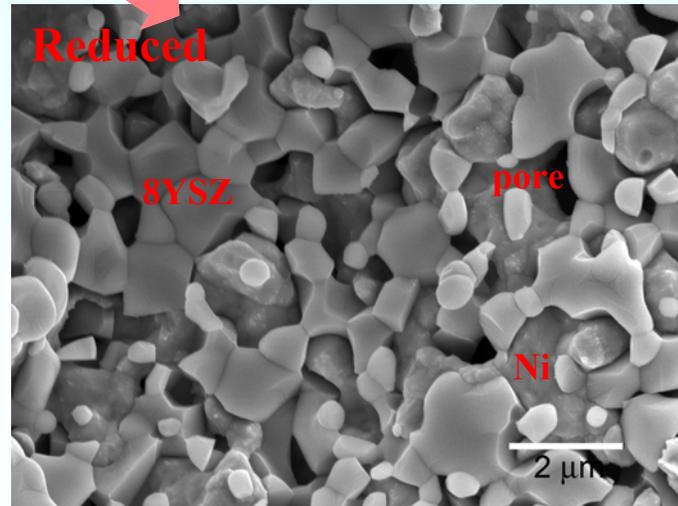
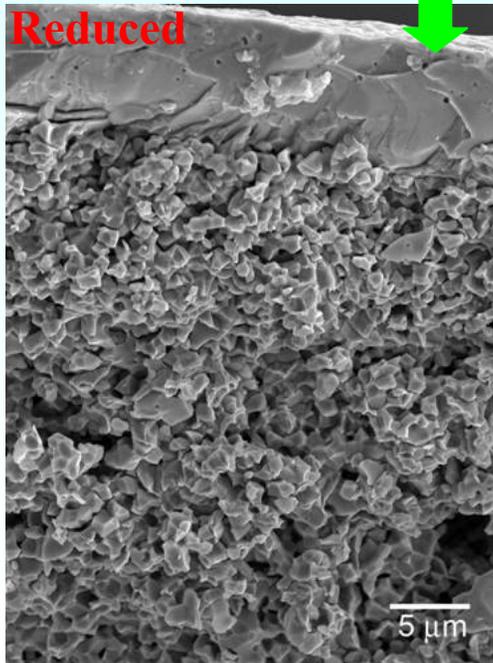
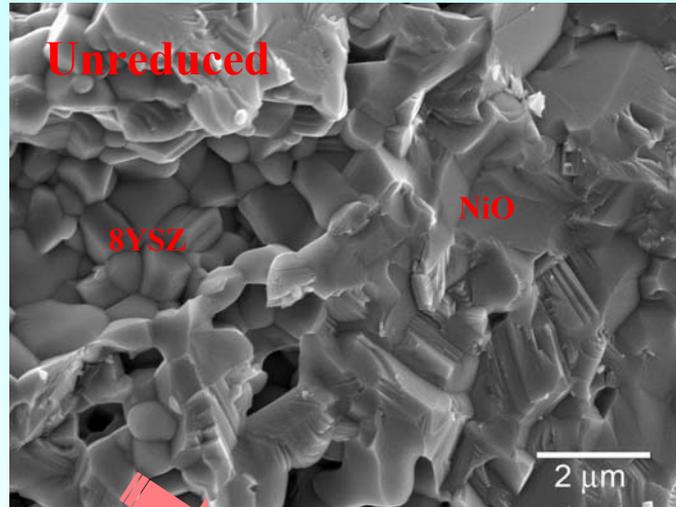
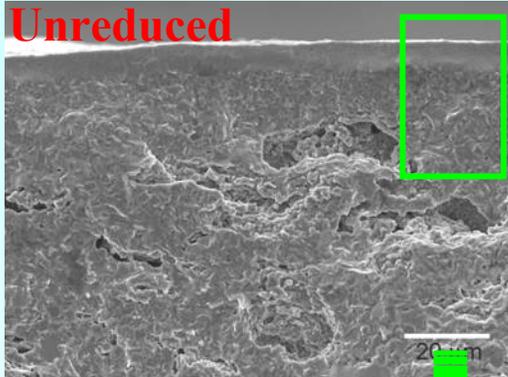
- Previous studies show that (La,Sr)(Cr,Mn)O<sub>3</sub>-based cathode material is electrocatalytically and chemically stable in both reducing and oxidizing atmospheres
- Previous long-term tests show degradation rate < 1% per 1000hrs over a 4500 hrs continuous test in the SOFC mode.
- Redox stability is desired for reversible applications

# SOFC-SOFEC Anode Substrate Development

## Effects of Residual/Chemical/Applied Stresses on Mechanical Integrity



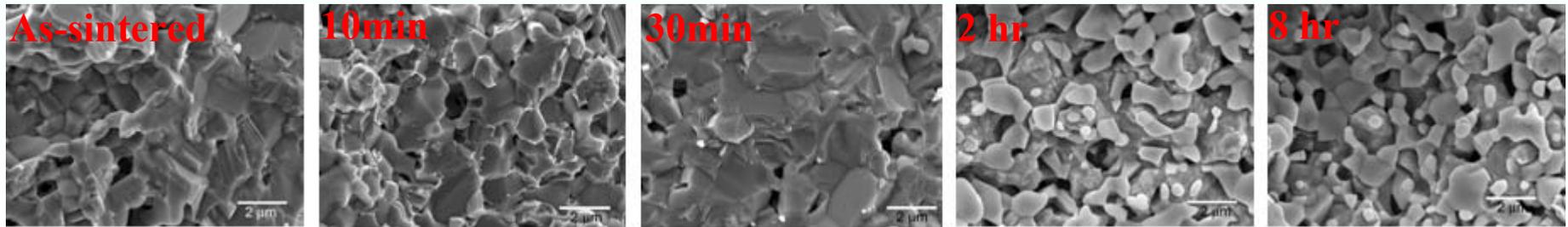
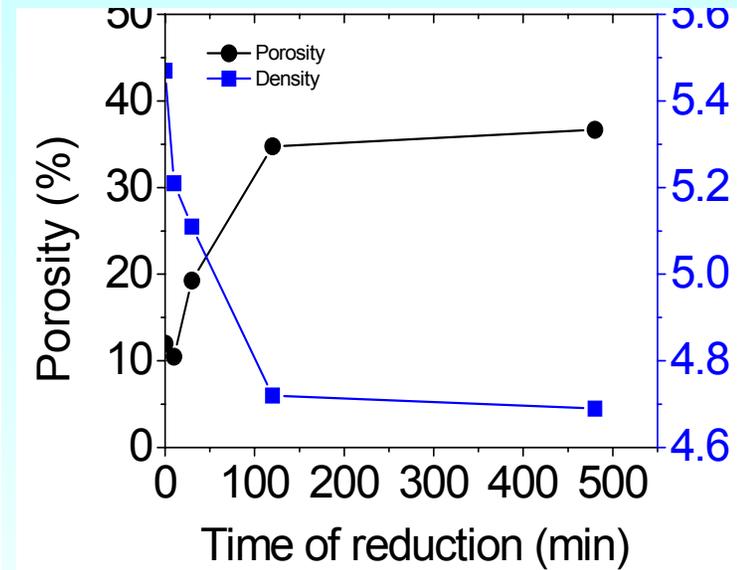
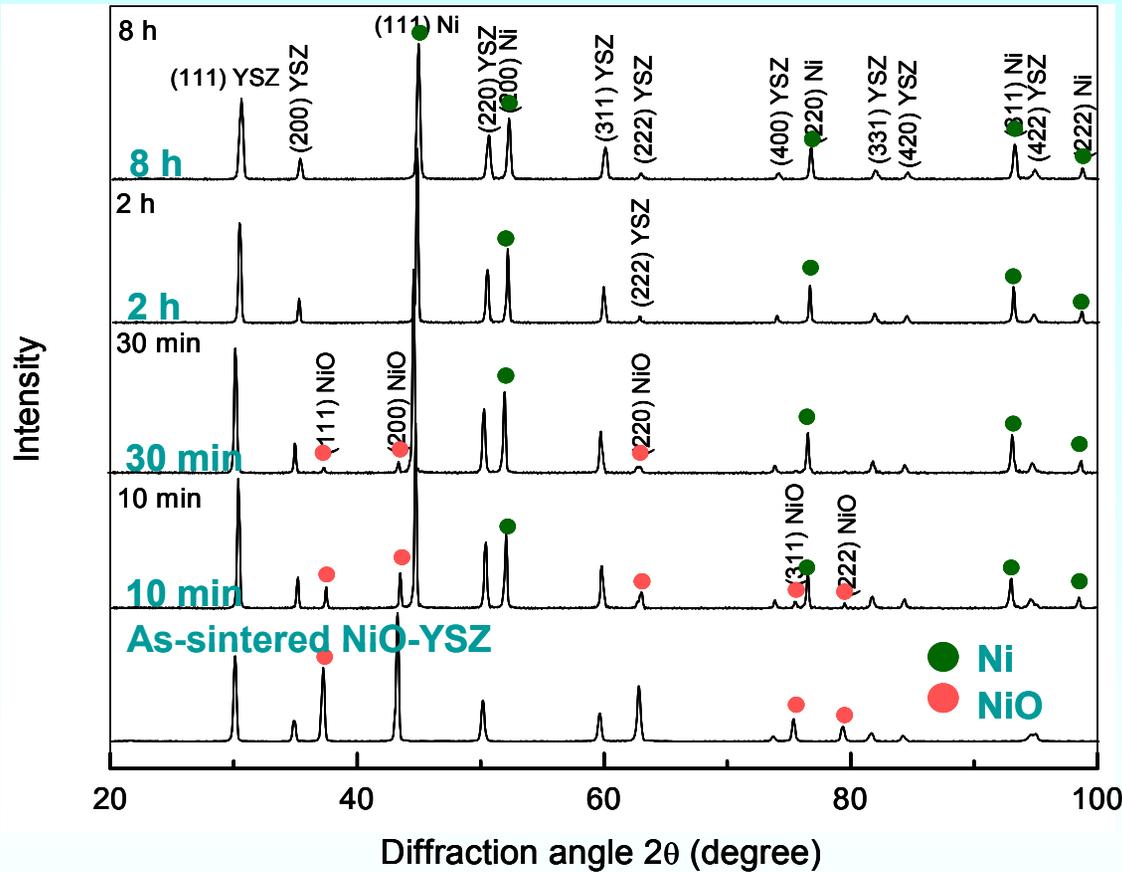
# Reduction of Half-cells



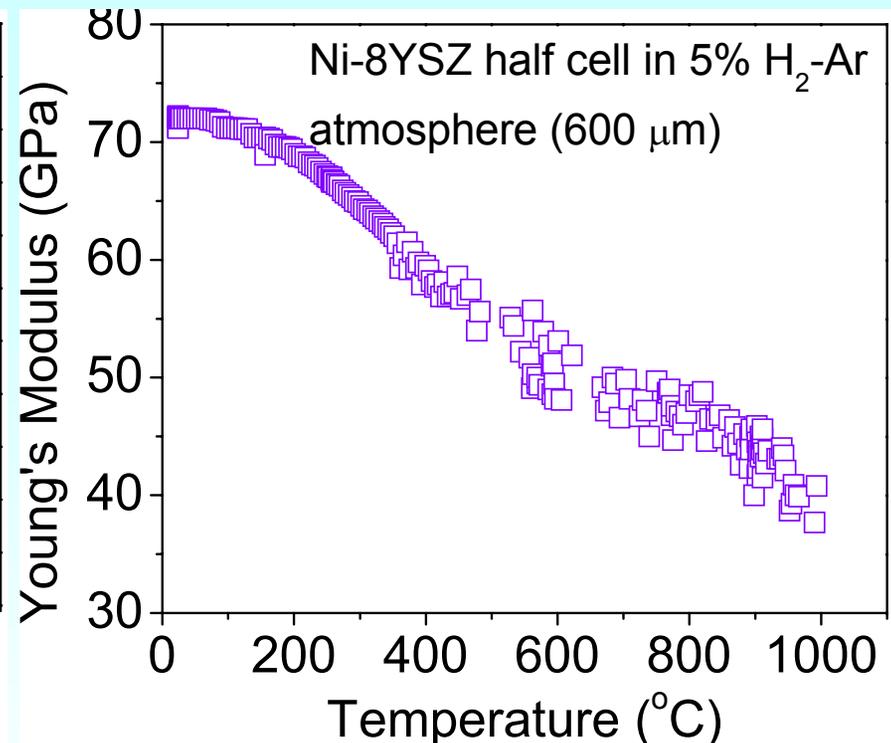
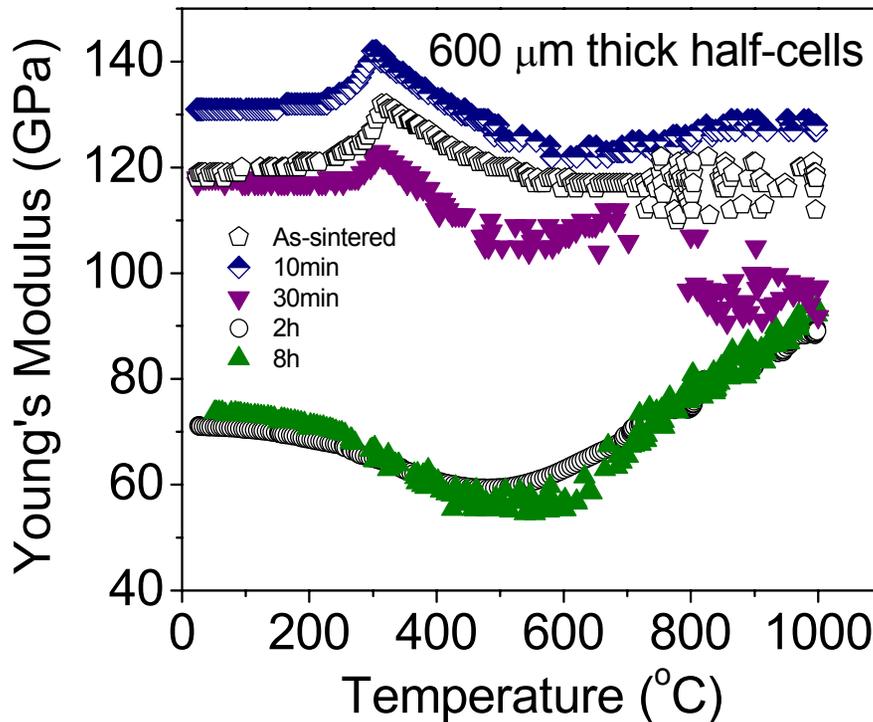
- Porosity
- Triple phase boundary (TPB)

- Density
- Mechanical properties

# Effect of Reduction on Phases & Microstructure



# Effect of Reduction on Mechanical Properties



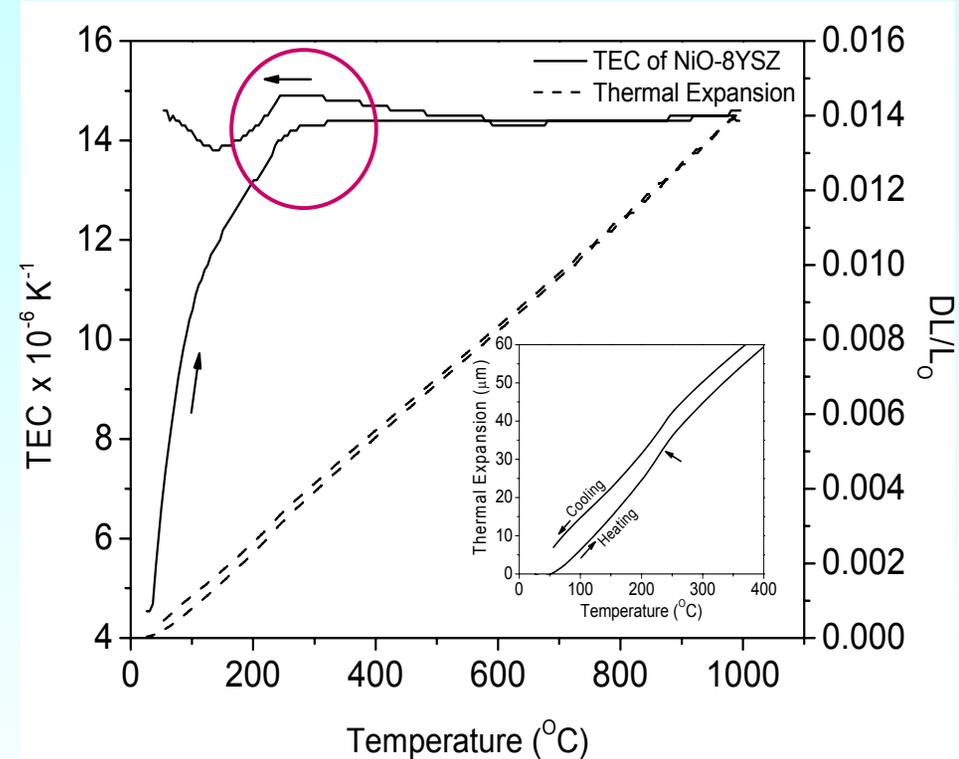
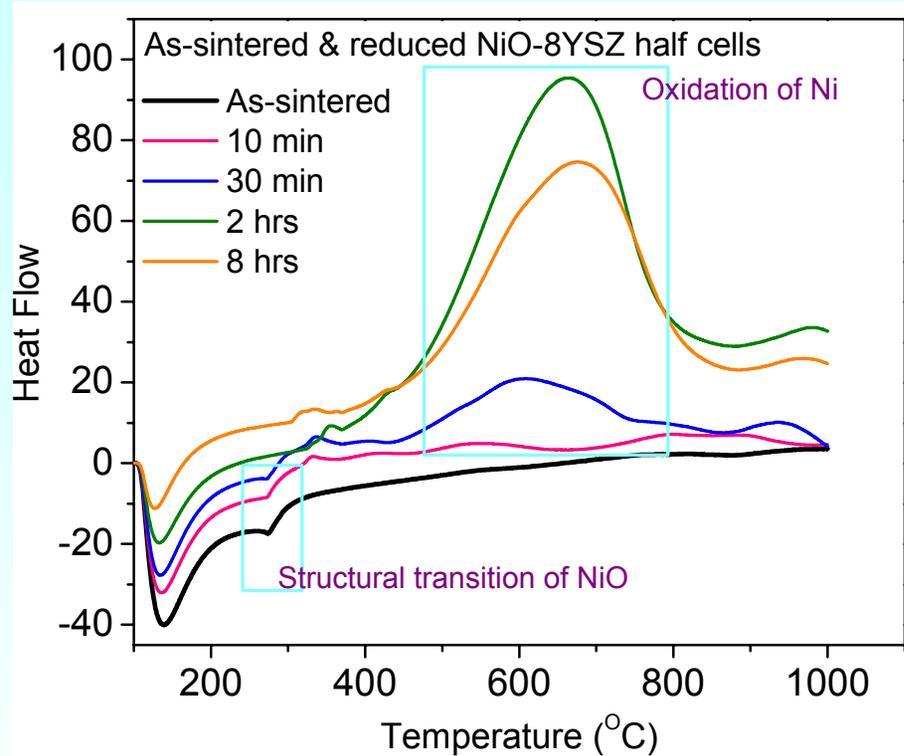
## Half-cells that have NiO:

- Higher initial E and a peak in E at 300 $^{\circ}\text{C}$  due to rhombohedral to cubic transition

## Half-cells that have no or negligible NiO:

- No structural transition assisted change in E; Oxidation of Ni increases E
- Scattering of E at high temperatures due to thermodynamic instability of c-ZrO<sub>2</sub>

# Events Affecting 'E' at Elevated Temperatures



- Rhombohedral to cubic transition of NiO that affects the thermal expansion and Young's modulus @  $\sim 300^{\circ}\text{C}$
- Oxidation of Ni @  $\sim 500^{\circ}\text{C}$

# Hermetic Seals Development

**Evaluated More Than 81 'Invert' Glass Compositions**

"Invert" silicate:

Glasses with

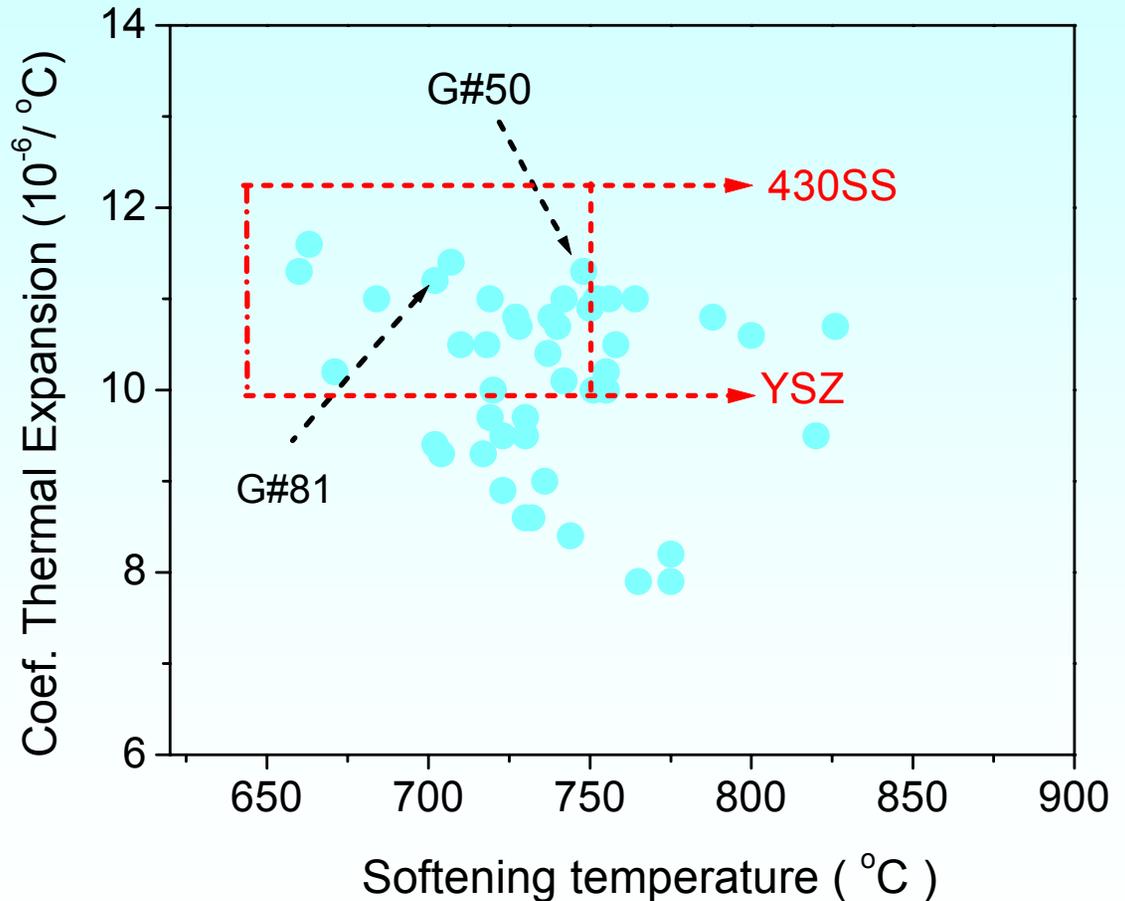
$\text{SiO}_2 < 45 \text{ mole\%}$

Compositions based on:

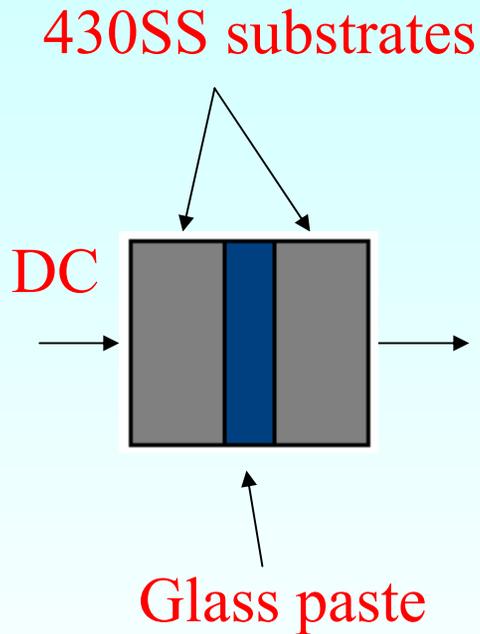
Pyrosilicate

and

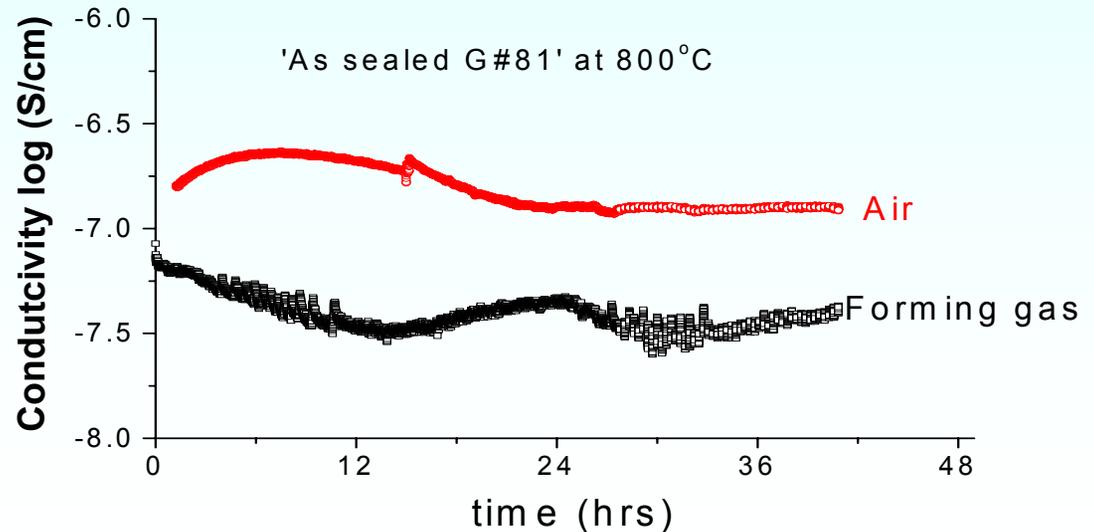
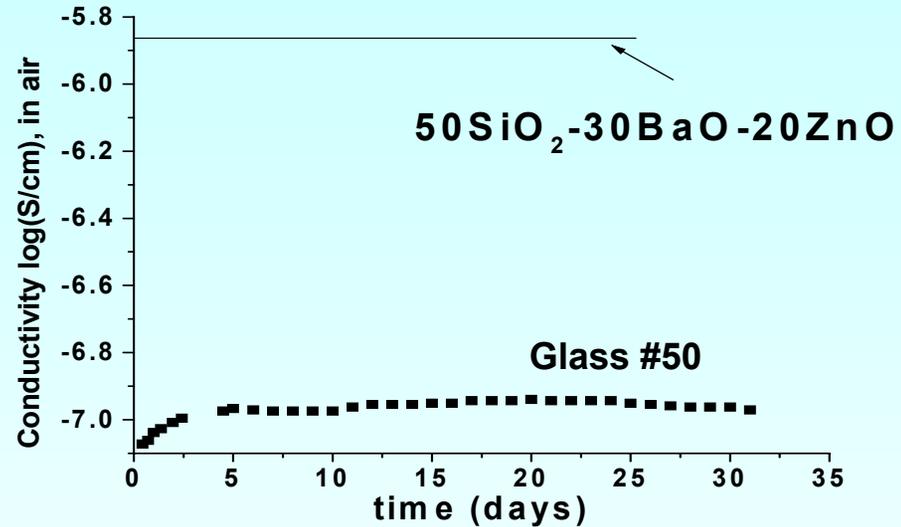
Orthosilicate



# Electrical Conductivity in Air and Forming Gas



Test conditions:  
800°C in air and  
forming gas



# Thermal Cycle Effects on Seals

## Hermetic Seals After Initial Multiple Thermal Cycles

Sealing materials	Test conditions	Notes
430SS/G50 <sup>1</sup> (45μm)/Ni-YSZ	wet forming gas	Failed after 20 cycles; Ni-YSZ fracture
430SS/G50 <sup>1</sup> (45μm)/YSZ	air	Failed after 60 cycles; YSZ fracture
430SS/G81 <sup>2</sup> (20μm)/Ni-YSZ	forming gas	10 cycles; Ni-YSZ fracture
430SS/G81 <sup>2</sup> (20μm)/Ni-YSZ	forming gas	Failed after 30 cycles; Ni-YSZ fracture
430SS/G81 <sup>2</sup> (25μm)/Ni-YSZ	forming gas	32 cycles without failure

(Samples held at pressure for 24 hours at 800°C; tested for leaks using a 4 psig differential in the testing gas)

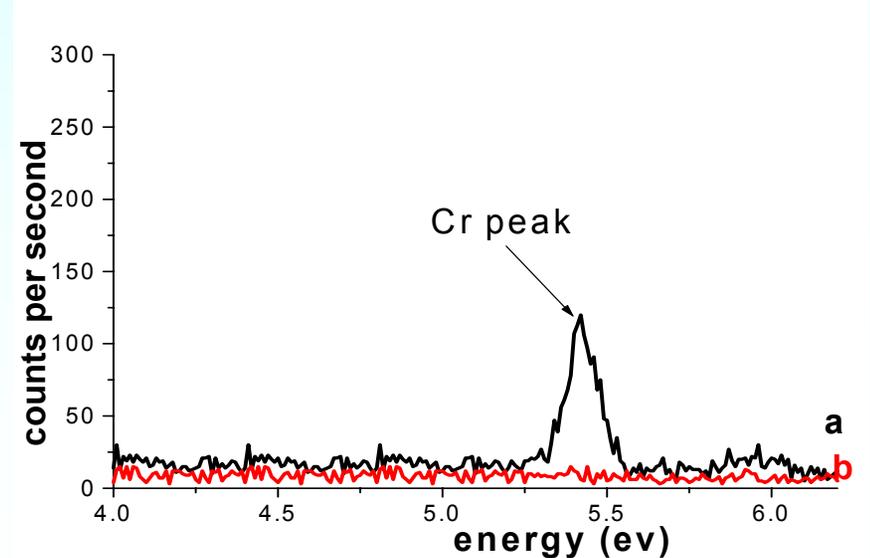
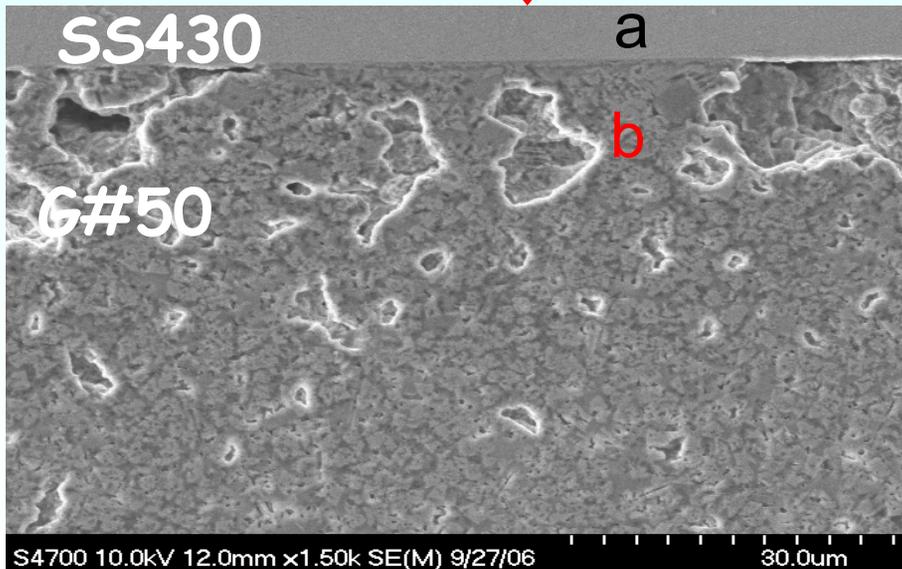
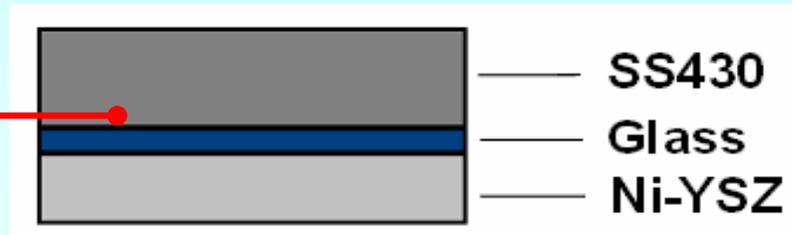
<sup>1</sup> Glass prepared by a commercial vendor (MO-Sci)

<sup>2</sup> Glass prepared at UMR

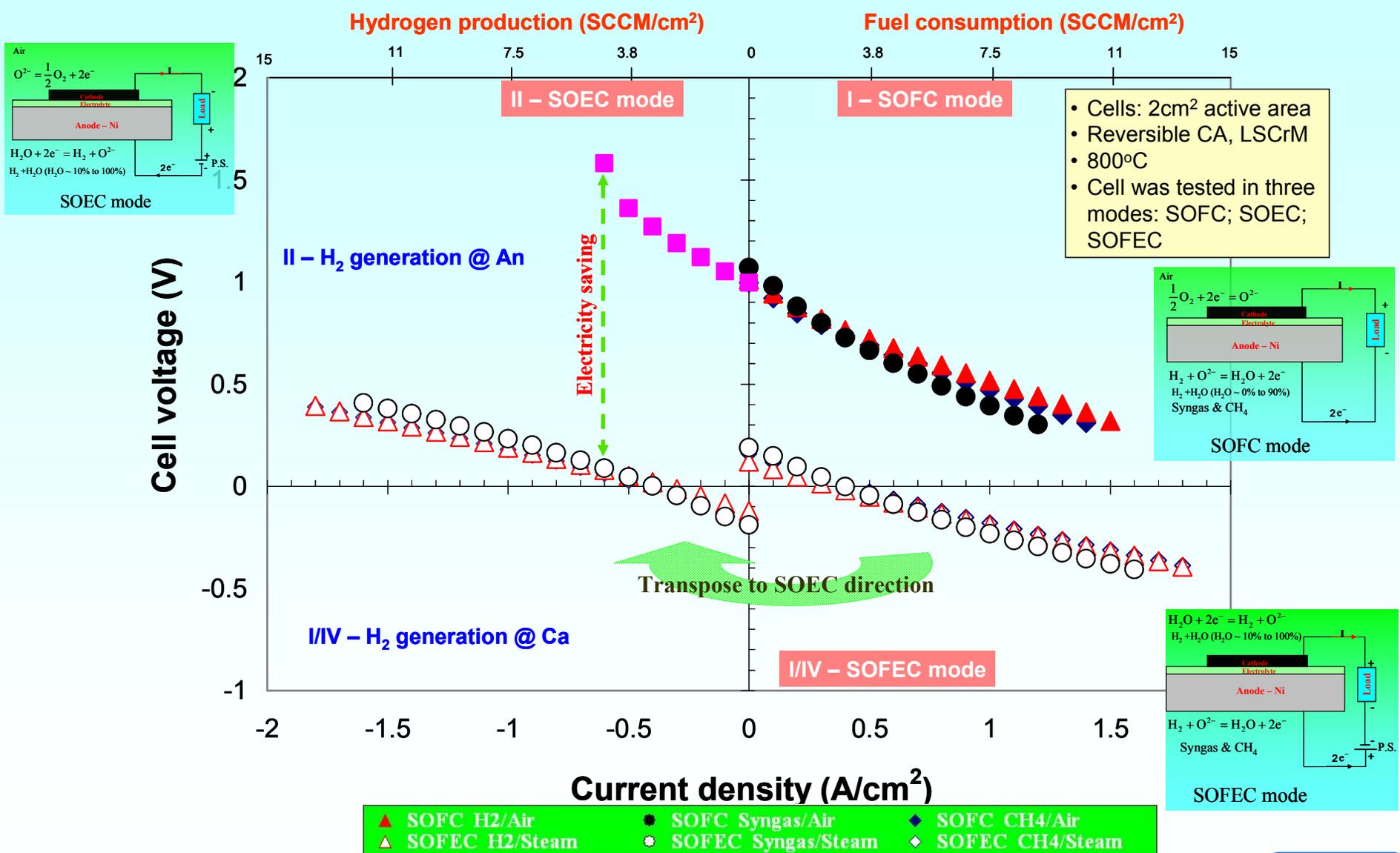
# Hermeticity and Materials Compatibility

## Little Evidence for Significant Reactions @ Seal Interfaces

Ten days in wet  
forming gas at 800°C

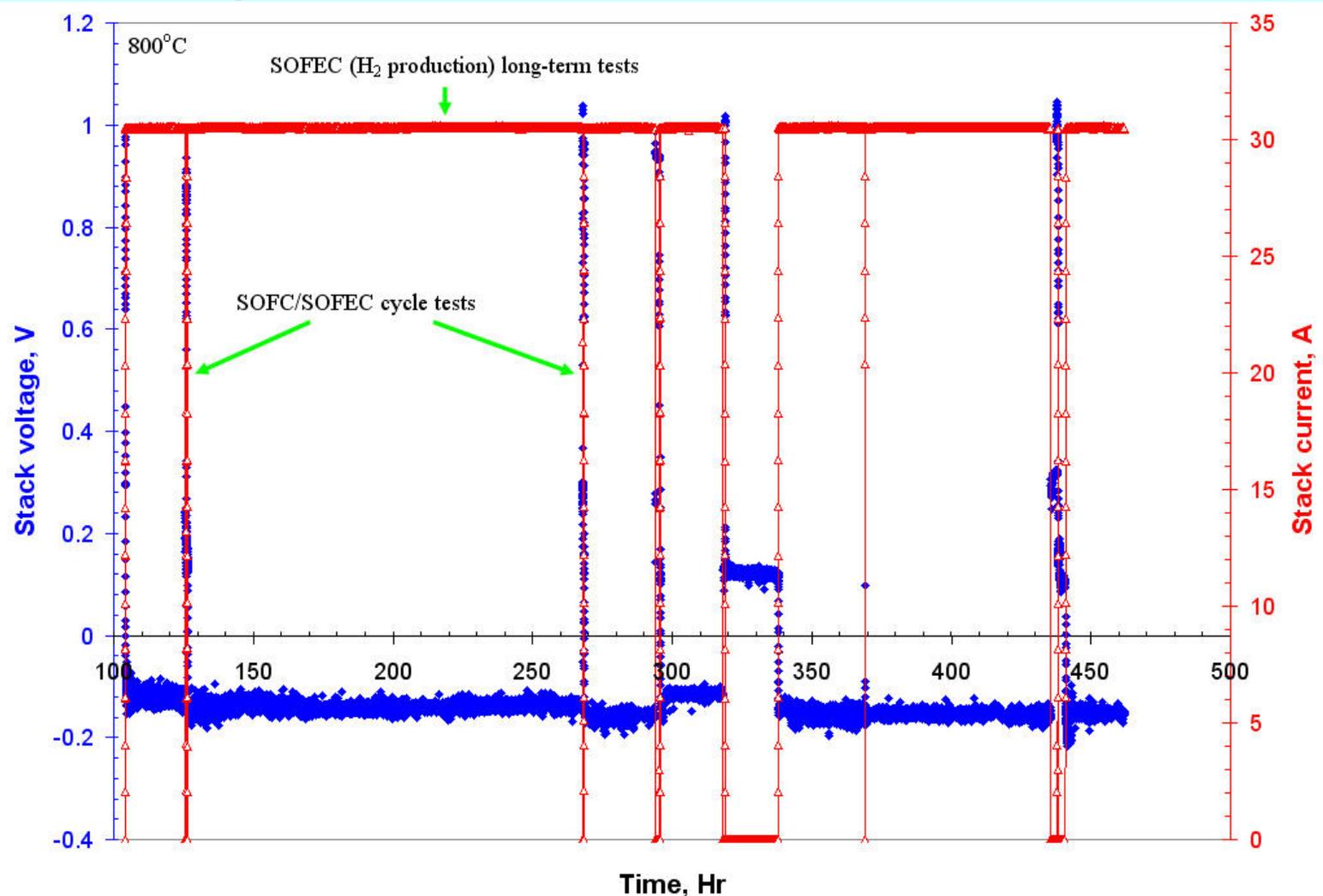


# Cathode Characteristics in SOFC/SOEC/SOFEC Modes



# SOFC Stability Test

## Single-cell Stack with 100 cm<sup>2</sup> Per-cell Active Areas



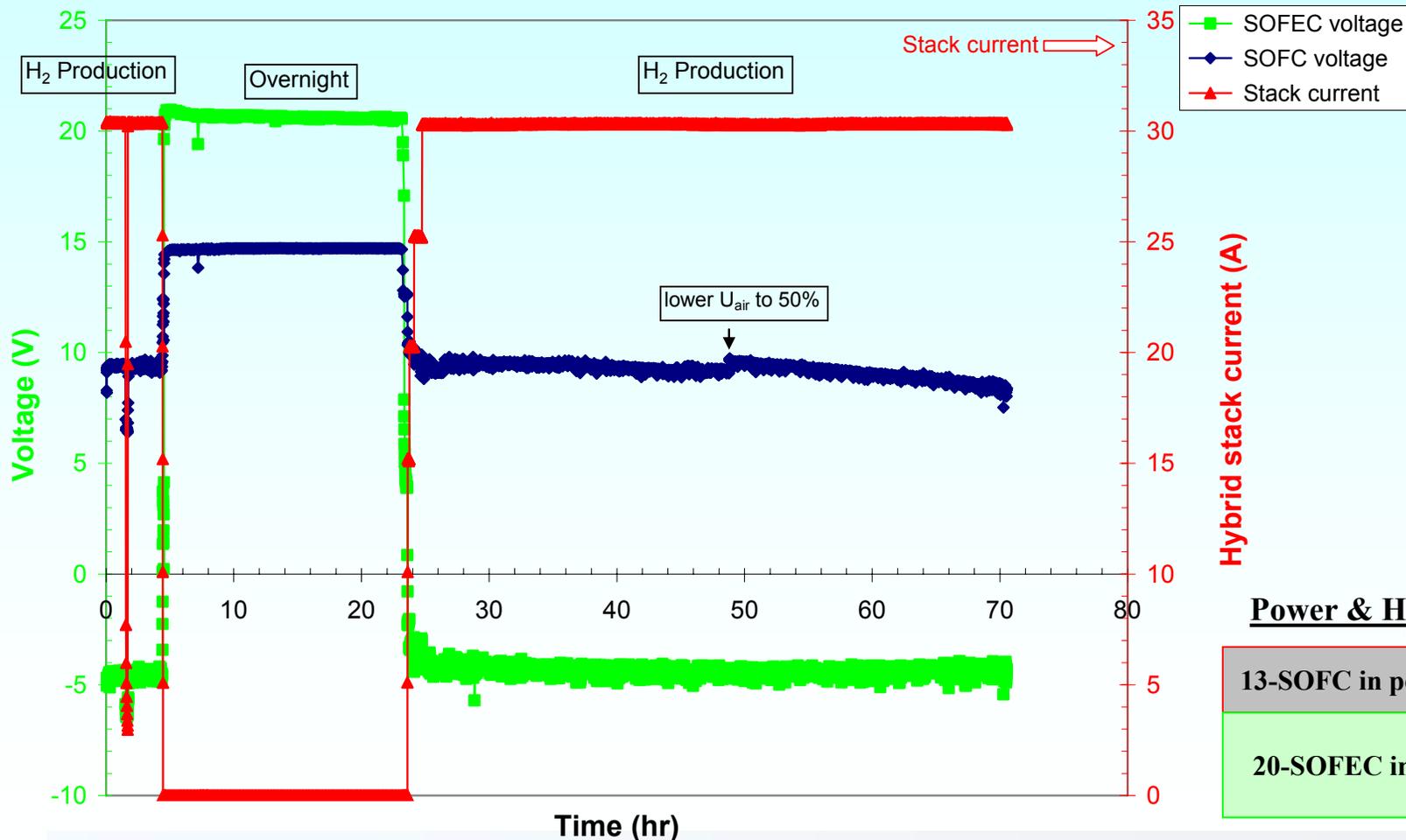
Per cell production rate: 13.92 standard liters of H<sub>2</sub> per hour, or 27.54 grams per day

# SOFC-SOFEC Hybrid Power & H<sub>2</sub> Cogeneration

Co-Production rate: Net power output @ 130 Watts and 270 standard liters of H<sub>2</sub> per hour (or 0.534 kg/day)

## SOFC (13-cell) + SOFEC (20-cell) Hybrid Stack for a Continuous H<sub>2</sub> Production

Temperature @ 780°C, AN: Syngas; CA1: air; CA2: H<sub>2</sub>O; U<sub>f</sub>/U<sub>air</sub>/U<sub>steam</sub>=50/60/40 --> 50/50/40



### Power & H<sub>2</sub> cogeneration

13-SOFC in power generation

20-SOFEC in H<sub>2</sub> production

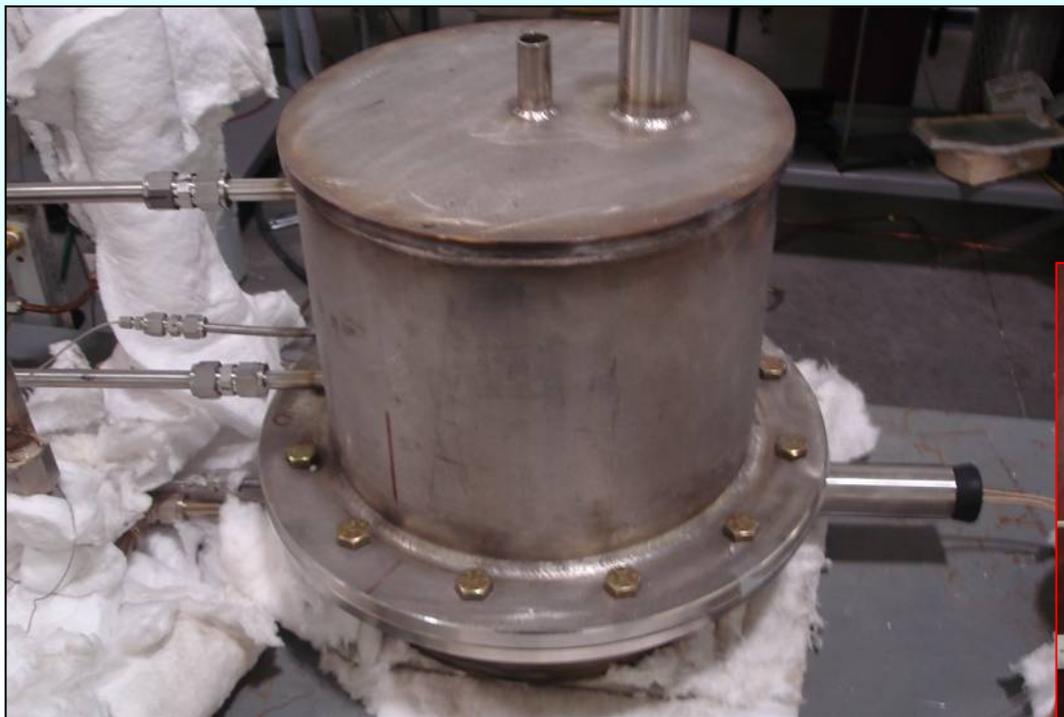
# Next Generation Stack Design for the 5 kW System

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- Larger stacks (60+ cells per stack)
- Thermal and flow management
  - Novel flow geometry for in-stack temperature and flow optimization
- Improved sealing
- Design for assembly
  - Stack components have been minimized to reduce stacking time and error
  - Improved reliability
- Reduced pressure drop over previous designs

# Balance of Plant Hardware

- Combustors, reformers and steam generators are being fabricated and tested prior to system integration.
- Catalytic combustors ensure minimal noxious byproducts.



**Catalytic combustor**



# Control System

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- Design of electronic control algorithm for the SOFC-SOFEC hybrid system.
- Autonomous operation enables to perform all stacks at maximum fuel efficiency and producing hydrogen from excess capacity.
- Multiple stacks in parallel operation to increase reliability and disrupt failure cascades.
- Acquisition of electronic control hardware and integration with system software.
- Pre-test of power electronics and load leveling systems before being integrated into the overall system.

# Future Work (FY 08 – FY 09)

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## **FY 08**

- Materials Development
  - Exploring an advanced cathode material
  - Continuous investigation of anode substrates: equi-biaxial test and flexural strength under both air and H<sub>2</sub> at working temperatures; effect of thermal cycles on strength
- 5 kW Hybrid System Fabrication
  - BOP components fabrication
  - Stack assembly, integration and burn-in
  - Implementation and optimization of system controls

## **FY 09**

- 5 kW Hybrid System Assembly and Evaluation
  - 5 kW hybrid system assembly
  - System testing and evaluation
  - Hydrogen production cost analysis using H2A model

# Project Summary

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- Relevance:** Investigate an alternative approach to provide low-cost and highly efficient distributed electricity and hydrogen
- Approach:** Develop a 5 kW SOFC-SOFEC hybrid system based on innovative materials development and system design research to co-generate hydrogen and electricity
- Technologies Accomplishments and Progresses:** Materials development: - Evaluated redox stability and long-term stability of the promising cathode material for SOFEC applications. - Finalized two promising “invert” glass compositions from 81 candidates. Seals survived after 30+ thermal cycles in reducing/oxidizing atmospheres. - Investigated effects of reduction on microstructure development and phase formation, elastic properties as a function of temperature, and effects of porosity, composition & microstructure on strength. 5 kW hybrid system development: - Evaluated long-term stability tests of hydrogen production to reduce cost. – Finalized the design of hybrid modules with improved thermal management and flow optimization. – Designed and fabricated major BOP components. – Designed hybrid system control algorithm and acquired control hardware.
- Proposed Future Research:** Continue implementing mechanical/thermal analyses of anode supports; fabricate and evaluate BOP components and optimize hybrid system controls; implement 5 kW system experimental evaluation and perform cost analyses using DOE H2A model.