

# Solid Acid Fuel Cell Stack for APU Applications

Hau H. Duong  
Superprotonic, Inc.  
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Project ID# fc\_45\_duong

# Overview

## Timeline

- Start – July 2008
- Finish – July 2010
- 40% complete

## Budget

- Total project funding
  - ▶ DOE - \$492,000
  - ▶ SPI cost share - \$123,000
- DOE funding received for FY'08
  - ▶ \$492,000

## Collaborator

- California Institute of Technology
- Richard Mistler, Inc.

## Barriers

- Barriers
  - ▶ A. Durability
    - Tolerance to reformed fuel
    - Operating hours
  - ▶ B. Cost
    - MEA fabrication
    - Catalysts
  - ▶ C. Performance

## Technical Targets

	Stack	System*
Power (W)	300	3000
Durability (h)	2000	2000
Efficiency (%LHV)	55	25
CO tolerance (%)	10	N/A
H <sub>2</sub> S tolerance (ppbv)	50	N/A
Specific power (W/kg)	55	TBD
Power density (W/L)	160	TBD
Cost (\$/kW <sub>e</sub> )	300	800

\* System projections based on conceptual design

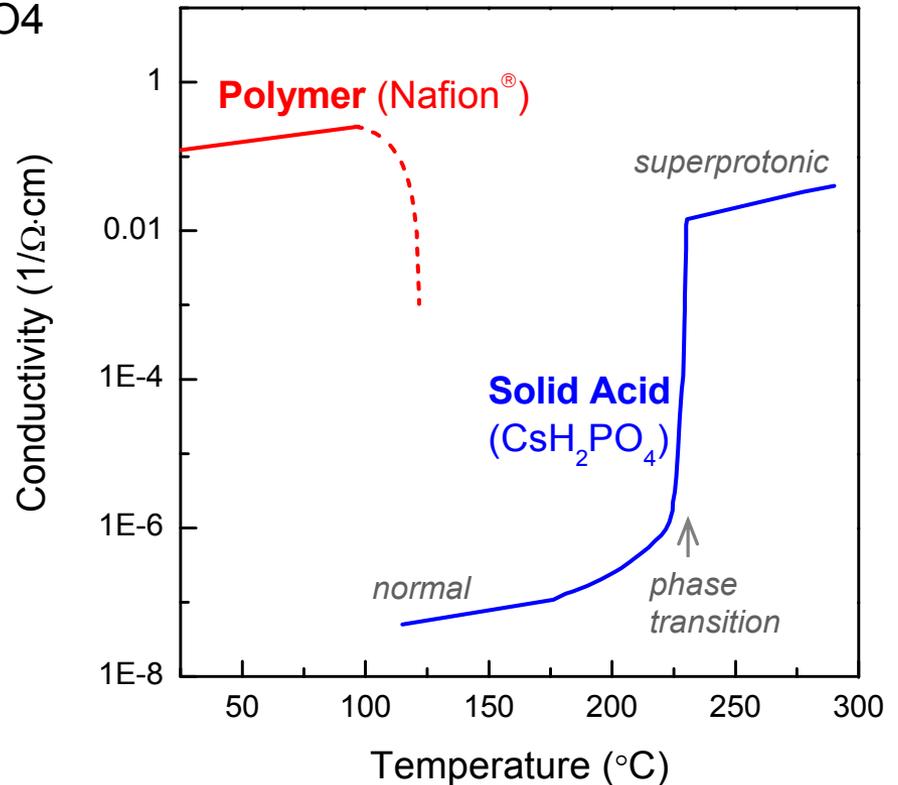
# Introduction to SAFC

# Solid Acid Electrolytes

## Fundamentals

- Intermediate salts and acids
  - ▶  $1\text{Cs}_3\text{PO}_4 + 2\text{H}_3\text{PO}_4 \rightarrow 3\text{CsH}_2\text{PO}_4$
  - ▶ General Formula:  $\text{M}_x\text{H}_y(\text{XO}_4)_z$ 
    - $\text{M} = \text{Cs, Rb, K, NH}_4, \text{TI}$
    - $\text{X} = \text{S, Se, P, As}$
- Physically similar to salts
  - ▶ Brittle
  - ▶ Water soluble
- Properties
  - ▶ Solid state proton conductivity
  - ▶ Impermeable

Solid Acid Conductivity

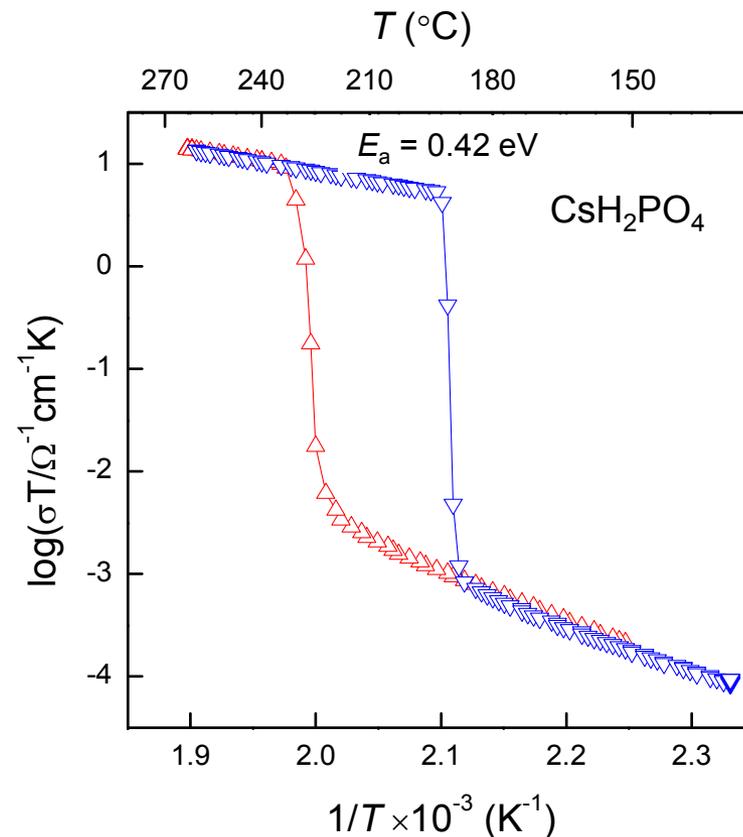


# Cesium Dihydrogen Phosphate

## Electrical properties

- Superprotonic phase transition
  - ▶  $T_{SP} = 231^{\circ}\text{C}$
  - ▶ Proton conductivity jump
    - <  $8.5 \times 10^{-6} \Omega^{-1}\text{cm}^{-1}$  at  $223^{\circ}\text{C}$
    - >  $1.8 \times 10^{-2} \Omega^{-1}\text{cm}^{-1}$  at  $233^{\circ}\text{C}$
  - ▶ Hysteresis
    - Upon cooling  $5\text{-}30^{\circ}\text{C}$
- Superprotonic conductivity
  - ▶ Typical conductivity
    - $\sim 2.5 \times 10^{-2} \Omega^{-1}\text{cm}^{-1}$  at  $250^{\circ}\text{C}$
  - ▶ Temperature range
    - $232\text{-}280^{\circ}\text{C}^*$

Arrhenius Plot of Conductivity

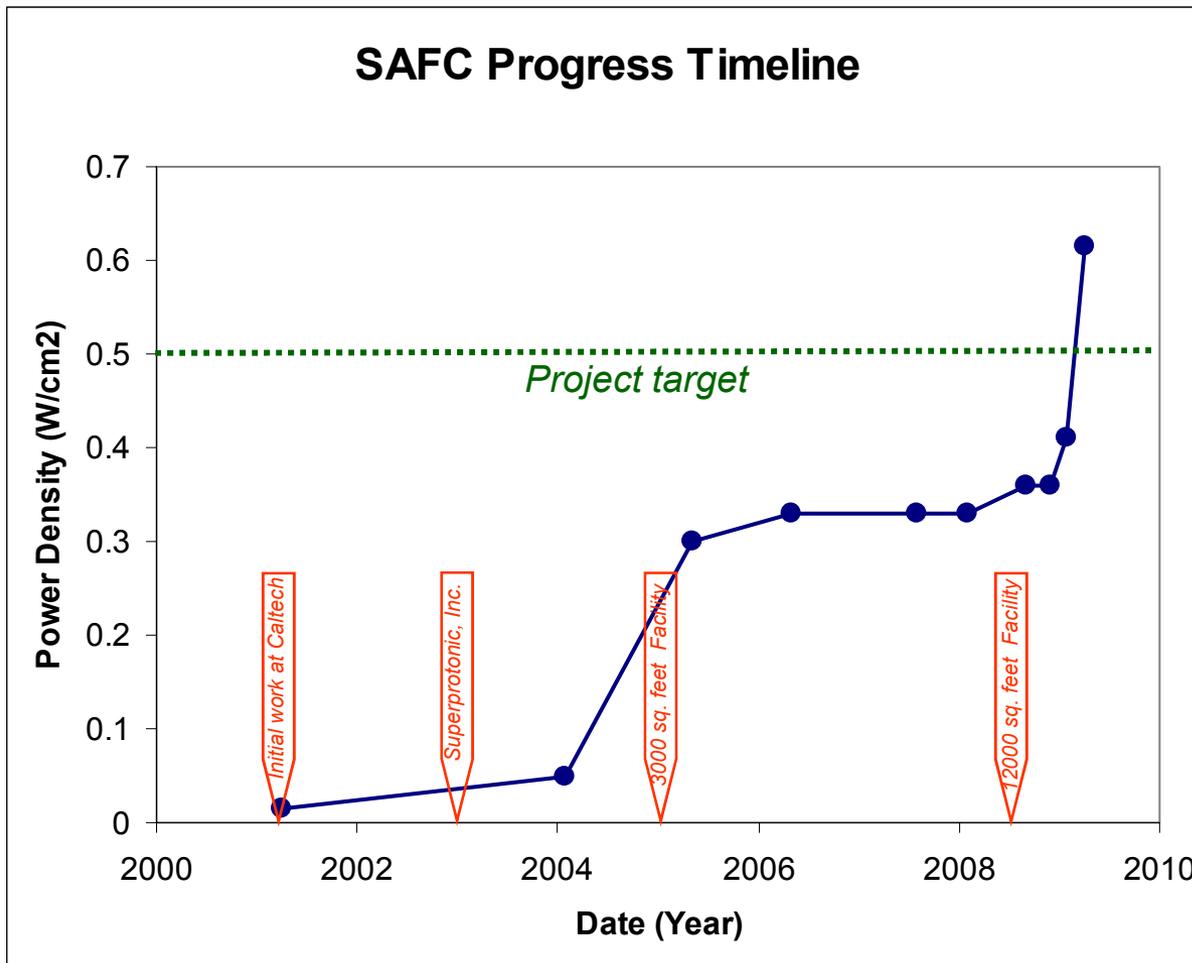


\* Dependant on  $P_{\text{H}_2\text{O}}$

# SAFC Characteristics

- Solid acid electrolytes
  - ▶ Solid state proton conducting membranes -  $\text{CsH}_2\text{PO}_4$
  - ▶ Operate at favorable temperatures - 230-280°C
- Fuel flexibility
  - ▶ Hydrogen, methanol, ethanol, reformed fossil fuels
  - ▶ Optimal temperature for low T gas shift
  - ▶ High tolerance to CO (>10%),  $\text{H}_2\text{S}$ (100ppm), &  $\text{NH}_3$  (100ppm),
- Low cost/high performance potential
  - ▶ Optimized microstructure for improved catalysis
  - ▶ Better catalysis/less platinum

# SAFC History at Superprotonic



# Project Update

# Objective - Relevance

The main focus of the project is to develop a solid acid fuel cell stack

- that operates on diesel reformat
- with performance characteristics approaching or exceeding most of the DOE's 2010 technical targets for an APU.

Performance Measure	Units	SPI Stack	SPI System*	DOE 2010 APU System Target	Technical Barrier(s) Addressed**
Power	Watts	300	3000	N/A	N/A
Durability	Hours	2000	2000	20000	A,B,C
Efficiency	% LHV	55	25	35	B,C
CO tolerance	%	10	N/A	N/A	A,B
H <sub>2</sub> S tolerance	ppbv	50	N/A	N/A	A,B
Specific power	W/kg	55	TBD	100	B,C
Power density	W/L	160	TBD	100	B,C
Cost*	\$/kW <sub>e</sub>	300	800	400	B

\*SPI System projections will be based on scaled up stack and conceptual system design

\*\*Barriers: A-Durability, B-Cost, C-Performance

\* Cost will be projected based on production of 100000 3 kW units/year

# Approach

## ■ Task 1: Development of SAFC MEA

- ▶ Fabricate and test 3 cm<sup>2</sup> and 20 cm<sup>2</sup> MEA
- ▶ Optimize MEA performance
- ▶ Develop proposal for scaling up MEA fabrication to 125 cm<sup>2</sup>

## ■ Task 2: Fabrication of 125 cm<sup>2</sup> SAFC MEA

- ▶ Fabricate using method based on 3 cm<sup>2</sup> and 15 cm<sup>2</sup> MEA
- ▶ Fabricate using scaled up method

## ■ Task 3: Characterization of SAFC MEA

- ▶ Evaluate SAFC performance on reformat with CO and H<sub>2</sub>S
- ▶ Evaluate 3 cm<sup>2</sup> and 15 cm<sup>2</sup> MEA performance
- ▶ Evaluate 125 cm<sup>2</sup> performance (durability, efficiency, specific power, power density, etc.)

## ■ Task 4: Design & Modeling of SAFC stack

- ▶ Design 300W stack components
  - Bipolar plates
  - Cooling plates
  - Sealing
- ▶ Model 300W stack
  - Flow field distribution
  - Pressure gradient

## ■ Task 5: Fabrication of 300W SAFC stack

## ■ Task 6: Characterization of SAFC stack

- ▶ Durability
  - Long term stability
  - Thermal cycling
- ▶ Efficiency
- ▶ Specific power
- ▶ Power density
- ▶ Test on reformat

## ■ Task 7: Design 3kW SAFC stack

- ▶ Scale up 300W design
- ▶ Modeling
- ▶ High volume cost projection

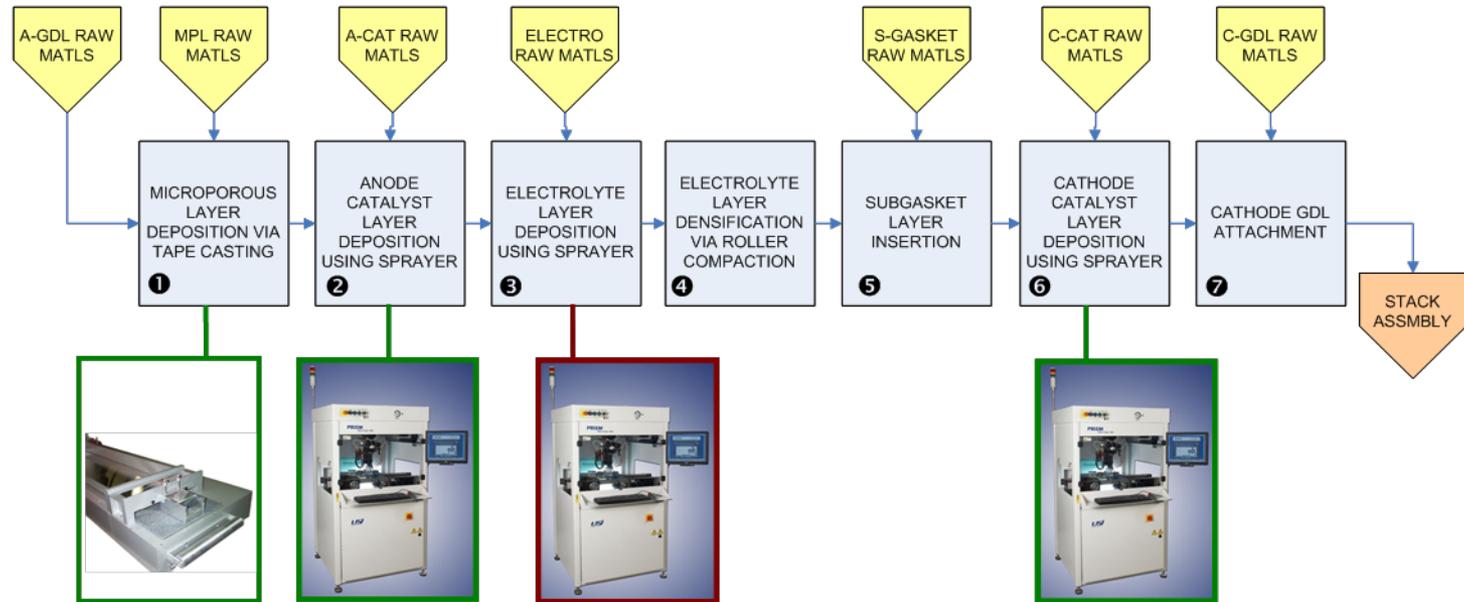
## ■ Task 8: Design 3kW SAFC system

- ▶ Control and ancillary subsystems
- ▶ Start-up and shutdown procedures and algorithm

# Milestones

Month/Year	Milestones or Go/No-Go Decision
August 2008	Go/No-Go decision: Complete evaluation of MEA fabrication and assess scalability with existing manufacturing processes. Conduct feasibility of electrolyte layer deposition using a sprayer and electrolyte densification via roller compaction. (Status – met)
September 2008	Go/No-Go decision: Complete evaluation of SAFC stack functionality and stability on synthetic reformat with high CO (10%) and H <sub>2</sub> S (100 ppm) content. (Status – met)
October 2008	Milestone: Complete 300W stack design completed based on existing 50W stack configuration. (Status – 75% complete)
November 2008	Milestone: Complete evaluation of 20 cm <sup>2</sup> SAFC stack functionality and stability on reformed methanol and liquid propane gas (Status – 100% complete)
January 2008	Milestone: Model flow field and pressure gradient on a 300W SAFC stack. (Status – 50% complete)
February 2008	Milestone: Fabricated 125 cm <sup>2</sup> SAFC MEA based on current 20 cm <sup>2</sup> MEA fabrication process (Status – 10% complete)
March 2008	Go/No-Go decision: Complete feasibility of SAFC stack functionality on reformed diesel fuel. (Status met)
October 2009	Milestone: Demonstrate micro-porous layer deposition using tape casting technique (Status – planned)
November 2009	Milestone: Demonstrate anode and cathode catalyst layer deposition using a sprayer (Status – planned)
December 2009	Go/No-Go: Fabricate 125 cm <sup>2</sup> SAFC MEA (10 minimum) using proposed fabrication process. (Status – planned)

# Development of SAFC MEA fabrication

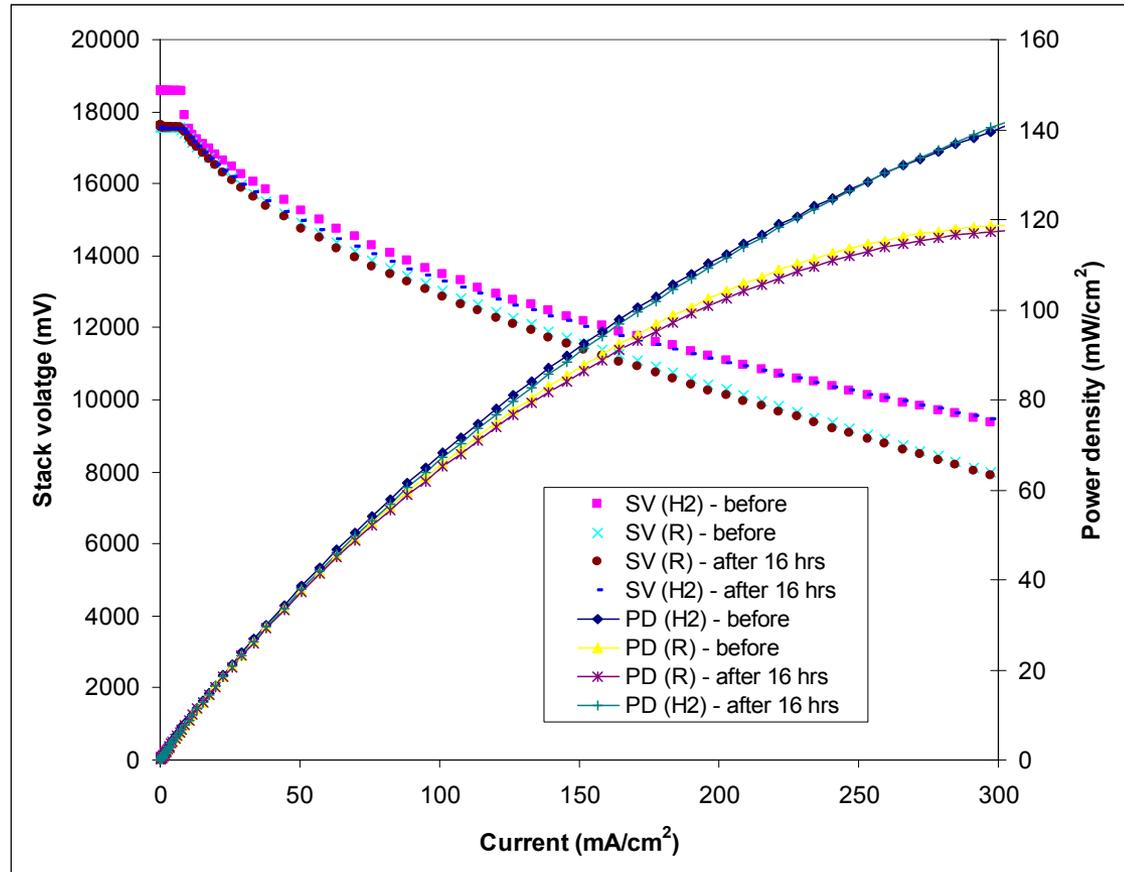


1. Micro-porous layer (MPL) deposition via the tape casting process;
2. Anode catalyst layer (ACL) deposition using a sprayer with positive displacement pumps;
3. Electrolyte layer deposition using a sprayer with positive displacement pumps;
4. Electrolyte layer densification via the roller compaction method;
5. Sub-gasket layer insertion;
6. Cathode catalyst layer deposition using a sprayer with positive displacement pumps;
7. Cathode gas diffusion layer (GDL) attachment.

# SAFC Performance on Synthetic Reformate

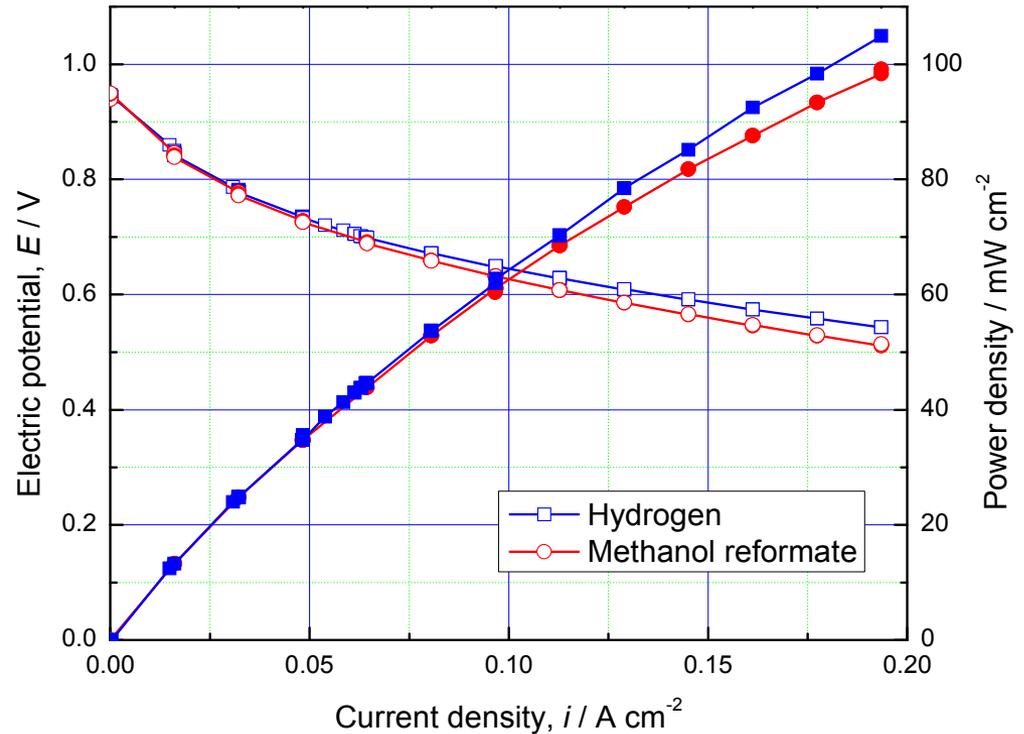


- Minimal effect of reformate on twenty 20 cm<sup>2</sup> cell stack performance
  - ▶ Mostly H<sub>2</sub> dilution effect
  - ▶ Degradation was ~0.06%/hr on reformate
- Gas flow/compositions
  - ▶ Cathode: 7 SLPM air + 0.3 bar H<sub>2</sub>O
  - ▶ Anode:
    - Hydrogen: 2.0 SLPM H<sub>2</sub> + 0.3 bar H<sub>2</sub>O
    - Reformate: 4 SLPM synthetic reformate + 0.3 bar H<sub>2</sub>O
  - ▶ **synthetic reformate:**
    - 50% H<sub>2</sub>
    - 40% N<sub>2</sub>
    - 10% CO
    - 100 ppm H<sub>2</sub>S



# SAFC Performance on Methanol Reformate

- Minimal effect of reformate on twenty 20 cm<sup>2</sup> cell stack performance
  - ▶ Mostly H<sub>2</sub> dilution effect
- Gas flow/compositions
  - ▶ Cathode: 7 SLPM air + 0.3 bar H<sub>2</sub>O
  - ▶ Anode:
    - Hydrogen: 1.0 SLPM H<sub>2</sub> + 0.3 bar H<sub>2</sub>O
    - Reformate: 2.8 SLPM methanol reformate + 0.3 bar H<sub>2</sub>O
  - ▶ **methanol reformate:**
    - 74.2% H<sub>2</sub>
    - 6.8% CO
    - 18.2% CO<sub>2</sub>



# SAFC Performance on Propane Reformate

- Minimal effect of reformate on twenty 20 cm<sup>2</sup> cell stack performance

- ▶ Mostly H<sub>2</sub> dilution effect

- Gas flow/compositions

- ▶ Cathode: 6.4 SLPM air + 0.3 bar H<sub>2</sub>O

- ▶ Anode:

- Hydrogen: 1.0 SLPM H<sub>2</sub> + 0.3 bar H<sub>2</sub>O

- Reformate: 5.8 SLPM commercial propane reformate + 0.3 bar H<sub>2</sub>O

- ▶ **commercial propane reformate:**

- 41.1% H<sub>2</sub>

- 34.6% N<sub>2</sub>

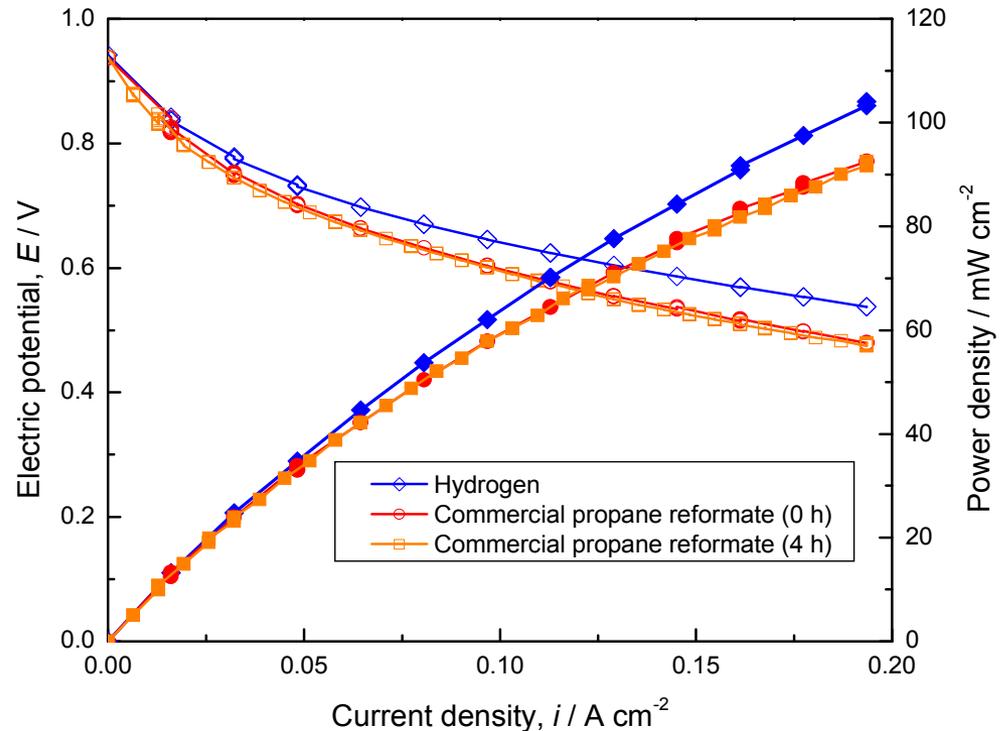
- 10.6% CO

- 9.84% CO<sub>2</sub>

- 0.24% CH<sub>4</sub>

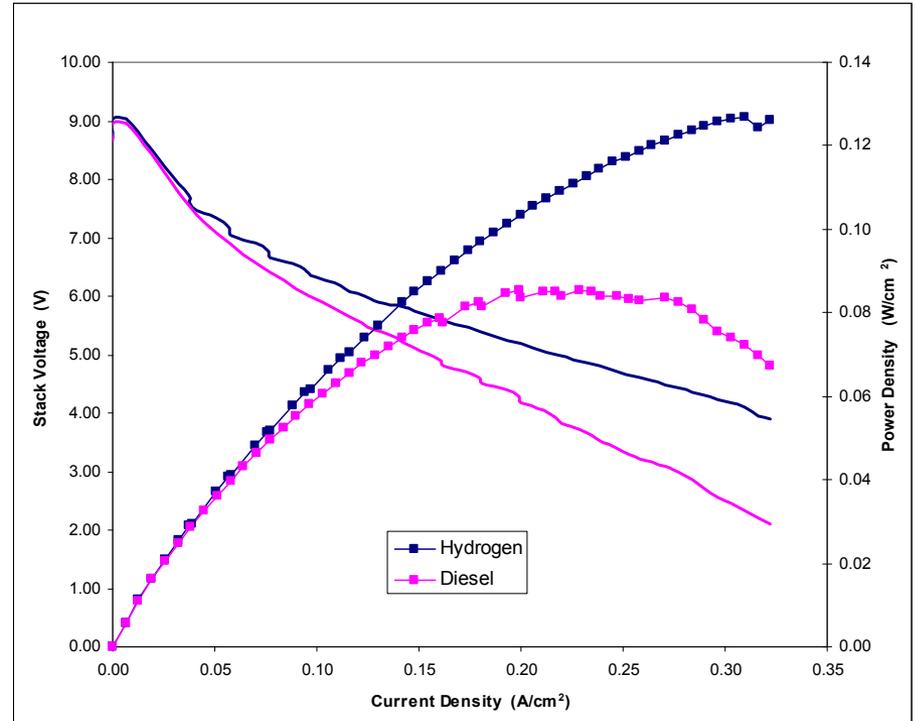
- 0.005% C<sub>3</sub>H<sub>8</sub>

- 0.5ppm H<sub>2</sub>S

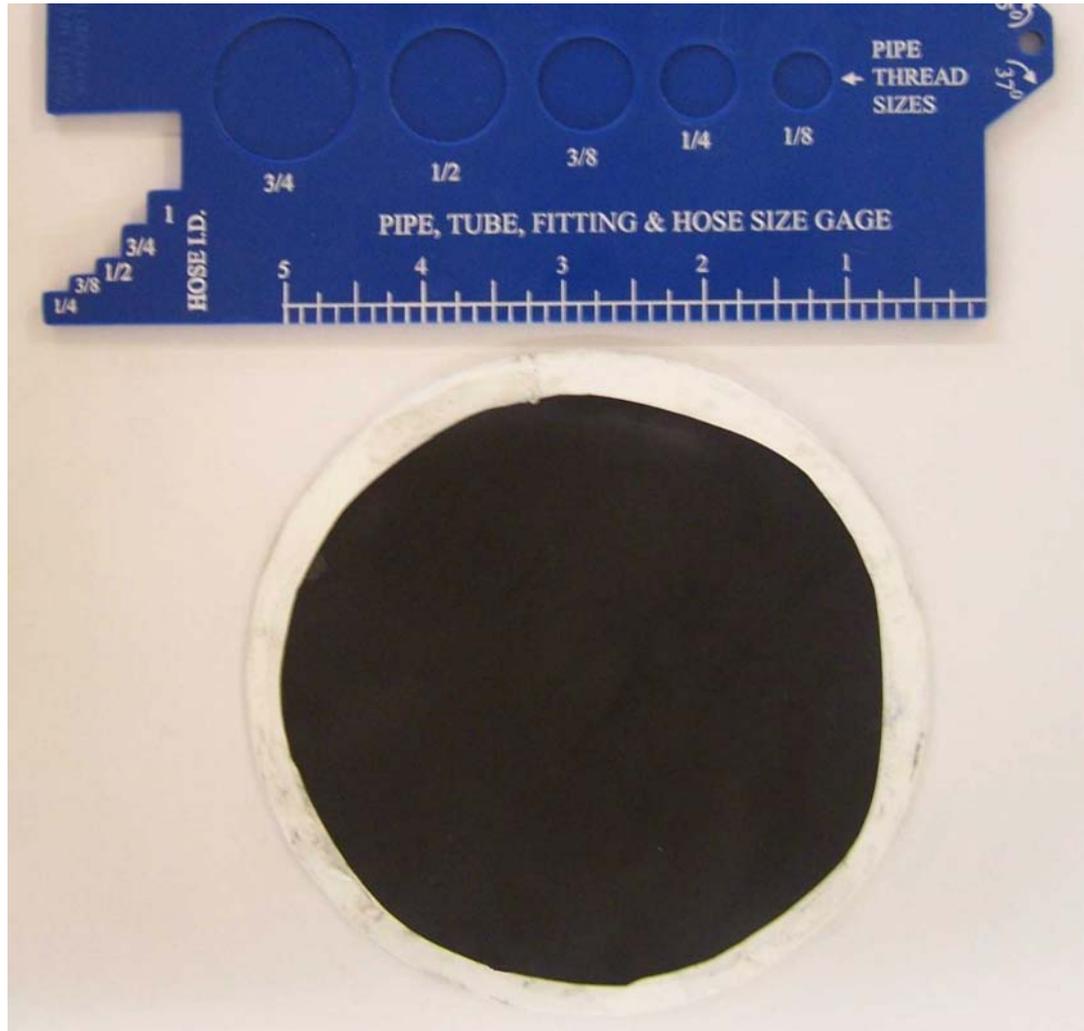


# SAFC Performance on Diesel Reformate

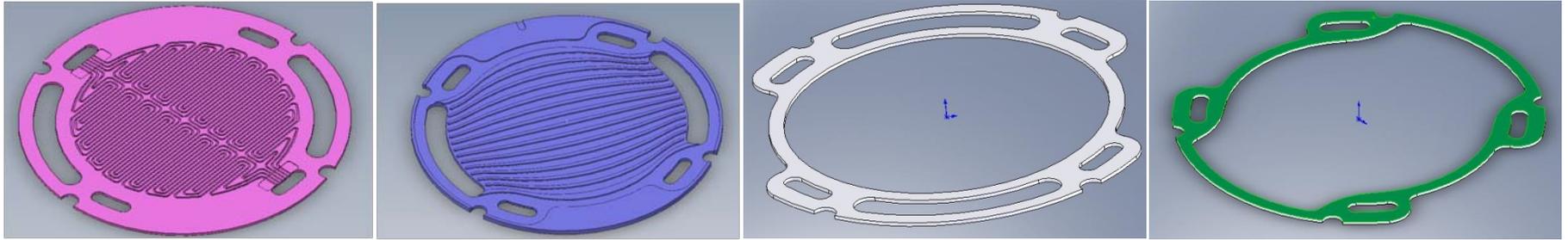
- Some effect of reformate on ten 20 cm<sup>2</sup> cell stack performance
    - ▶ Strong H<sub>2</sub> dilution effect
  - Gas flow/compositions
    - ▶ Cathode  
3 SLPM air + 0.3 bar H<sub>2</sub>O
    - ▶ Anode:  
Hydrogen: 1.0 SLPM H<sub>2</sub> + 0.3 bar H<sub>2</sub>O  
Reformate: 3.0 SLPM ultra low sulfide diesel reformate + 0.3 bar H<sub>2</sub>O
- Ultra Low Sulfide Diesel :**
- 35% H<sub>2</sub>
  - 41% N<sub>2</sub>
  - 14% CO
  - 10% CO<sub>2</sub>
  - 0.3% CH<sub>4</sub>
  - 1ppm H<sub>2</sub>S (by Volume)



# Fabrication of 125 cm<sup>2</sup> SAFC MEA



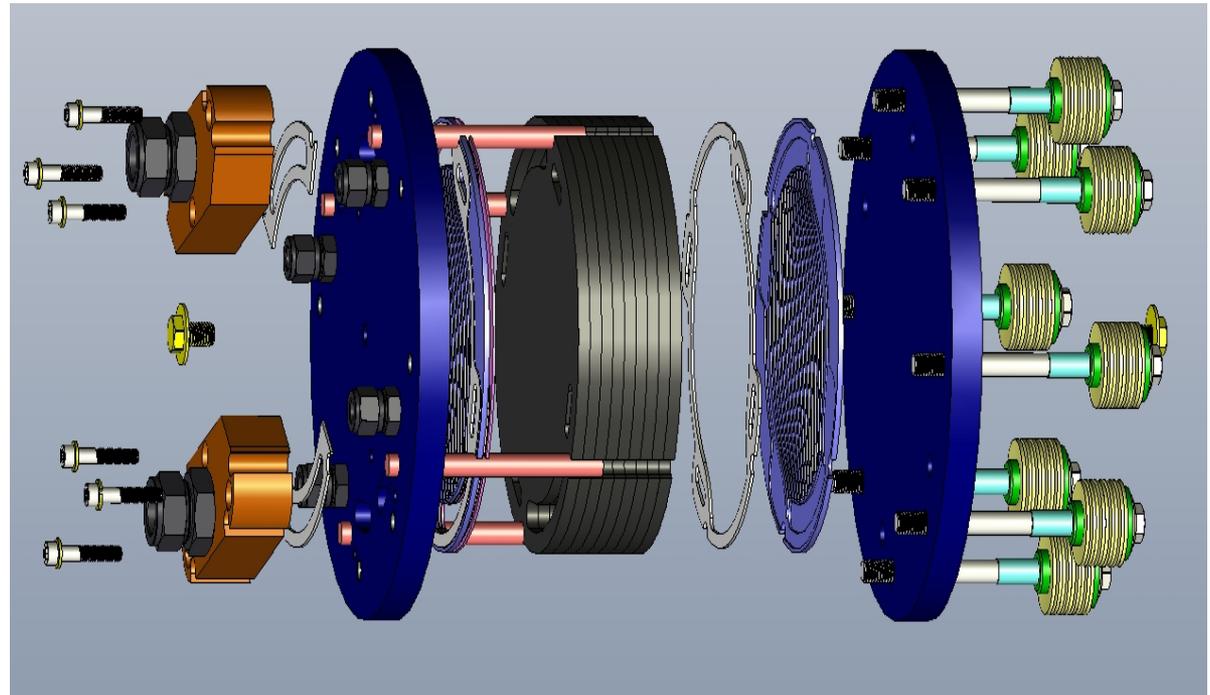
# 300 W Stack Design



- *Stack parameters\**

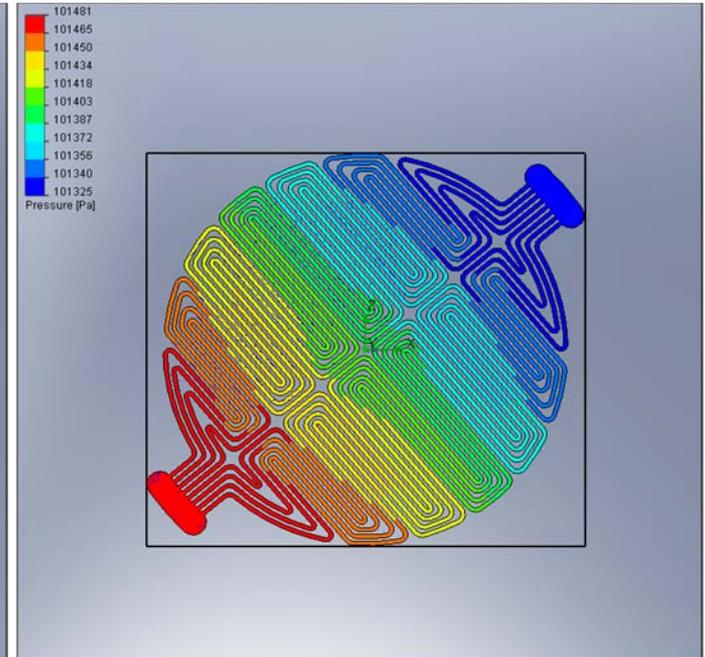
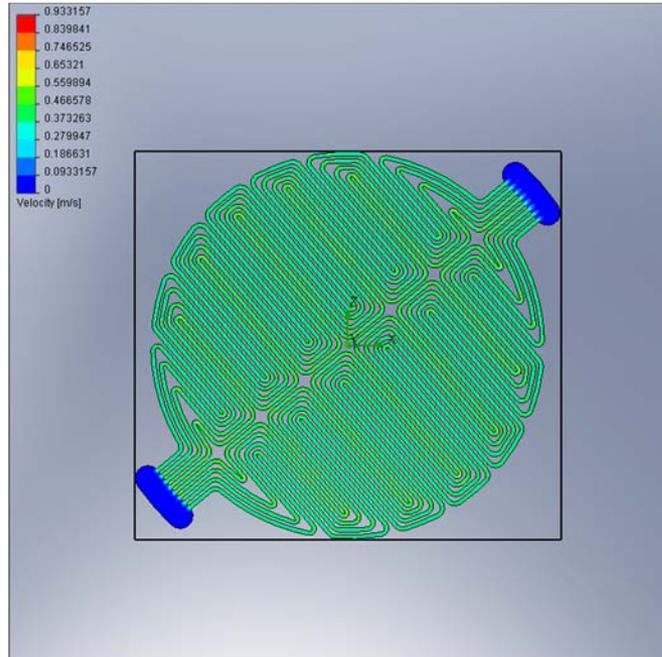
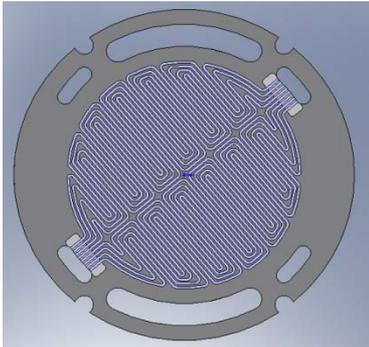
- *Diameter: 18 cm*
- *Height: 11 cm*
- *Weight: ~10 kg*

\* Based on power density of  $0.2\text{W}/\text{cm}^2$



# Flow field and pressure gradient modeling

- Active area ~ 125 cm<sup>2</sup>
- Flow field
  - Channel depth: 0.9 mm
  - Channel width: 1.1 mm
  - Channel length: 68~76 cm
- Manifold Area: 2 cm<sup>2</sup>



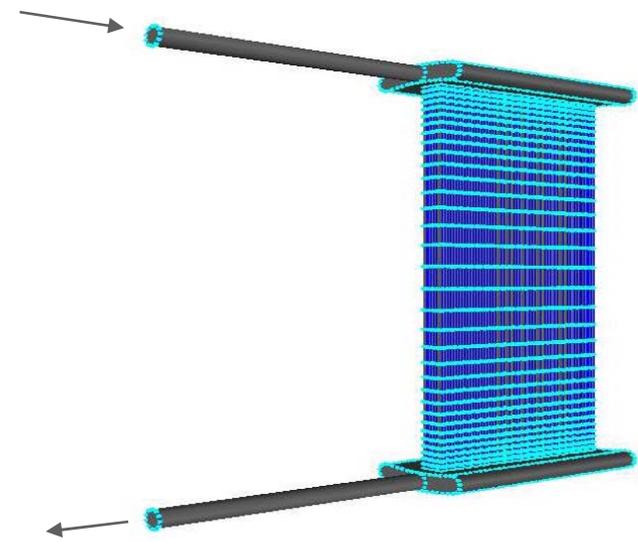
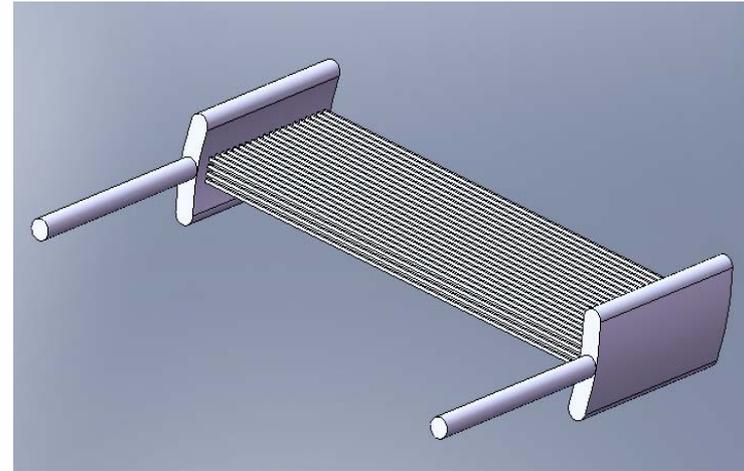
*FEA modeling suggests*

- Uniform flow distribution across electrode surfaces
- Uniform pressure drop between flow field channels

# 300 W Stack Modeling

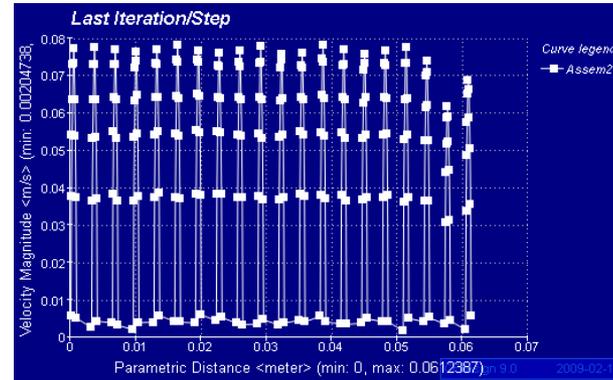
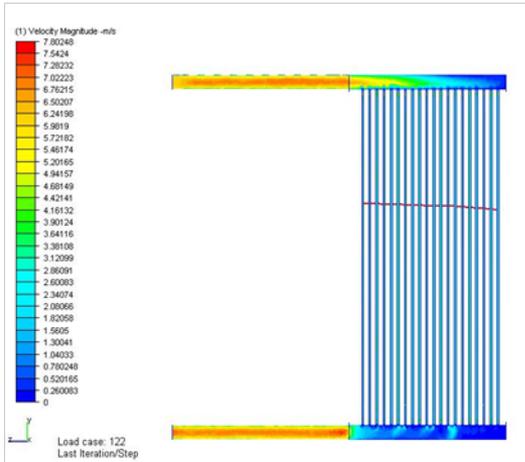
*Full modeling requires too much memory, it was necessary to simplify the model*

- 1. Straight channels replaced serpentine channels*
- 2. Shorter channels replaced full-length channels to save memory*
- 3. 20-cell stack was considered and each cell has four straight channels*
- 4. Calculated velocity distribution between cells*
- 5. Modeling geometry*
  - Channel depth: 0.9 mm*
  - Channel width: 1.1 mm*
  - Channel length: 14~15 cm*
  - Tube length: 7.6 cm*
  - Manifold Area: 2 cm<sup>2</sup>*

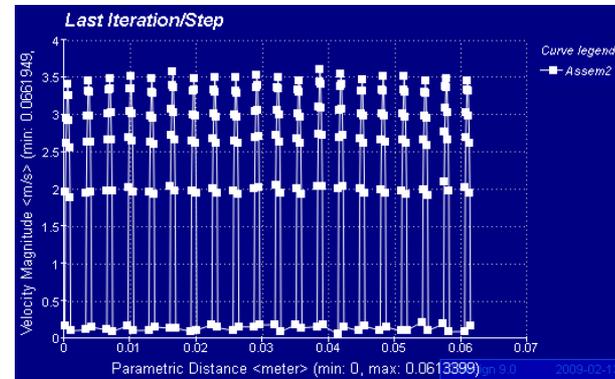
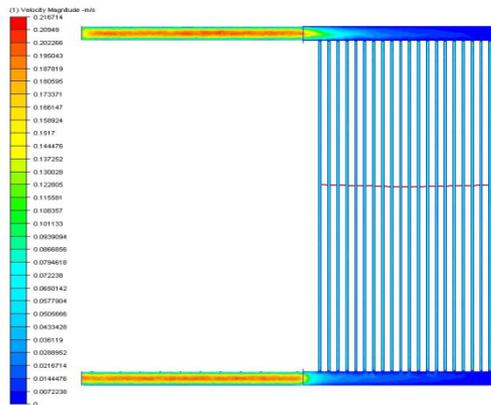


# Flow Distribution Modeling

4 l/min



10 l/min



## ■ Academic

- ▶ Collaboration with the California Institute of Technology
  - Resource for analytical instrumentation;
  - Synthesis of  $\text{CsH}_2\text{PO}_4$  electrolyte nanoparticles to increase the power density.

## ■ Industry

- ▶ Richard E. Mistler, Inc.
  - Determine whether the current formulation used at Superprotonics to produce micro porous layer can be scaled to production on a full length tape casting machine;
  - Develop a technique for the production with one step casting operation.

- FY2009
  - ▶ Continue to optimize SAFC MEA performance to levels of commercial viability
  - ▶ Build and characterize a 300W SAFC stack operating on hydrogen fuel
    - Employ micro-porous layer deposition using tape casting
    - Employ anode and cathode catalyst layer deposition using sprayer
    - Employ CDP electrolyte densification using compaction
    - Fabricate 125 cm<sup>2</sup> SAFC MEA using above processes
- FY2010
  - ▶ Build and characterize a 300W SAFC stack operating on diesel reformat
  - ▶ Complete conceptual design of 3 kW SAFC stack
  - ▶ Complete conceptual design of 3 kW SAFC system

# Project Summary



- **Relevance:** Demonstrate the feasibility of SAFC for energy efficient APU applications.
- **Approach:** Build and characterize 300 W SAFC stack on hydrogen and reformed diesel fuels; design 3 kW SAFC stack and system.
- **Technical Accomplishment and Progress:** Demonstrated SAFC functionality and stability on methanol, propane, and diesel fuels.
- **Technology Transfer/Collaborations:** Active partnership with the California Institute of Technology and Mistler, Inc..
- **Proposed Future Work:** Scale up SAFC MEA size and quantity; build 300 W SAFC stack; design 3 kW SAFC system.

# Acknowledgements



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This support does not constitute an endorsement by DOE of the material expressed in this presentation.

# Further Information



**Hau H. Duong, Ph.D., M.B.A.**

Vice President of Product Development

(626) 793-9314x112

[hau.duong@superprotonic.com](mailto:hau.duong@superprotonic.com)

**Calum Chisholm, Ph.D.**

Founder & Vice President

(626) 793-9314 x103

[calum.chisholm@superprotonic.com](mailto:calum.chisholm@superprotonic.com)

**Superprotonic, Inc.**

81 W Bellevue Drive

Pasadena, CA 91105

[www.superprotonic.com](http://www.superprotonic.com)

# Supplemental Slides

# Cesium Dihydrogen Phosphate

## *Thermal properties*



- Long debated in literature
- Stability proven under water partial pressure ( $P_{\text{H}_2\text{O}}$ )
- How much  $P_{\text{H}_2\text{O}}$  to stabilize superprotonic  $\text{CsH}_2\text{PO}_4$ ?

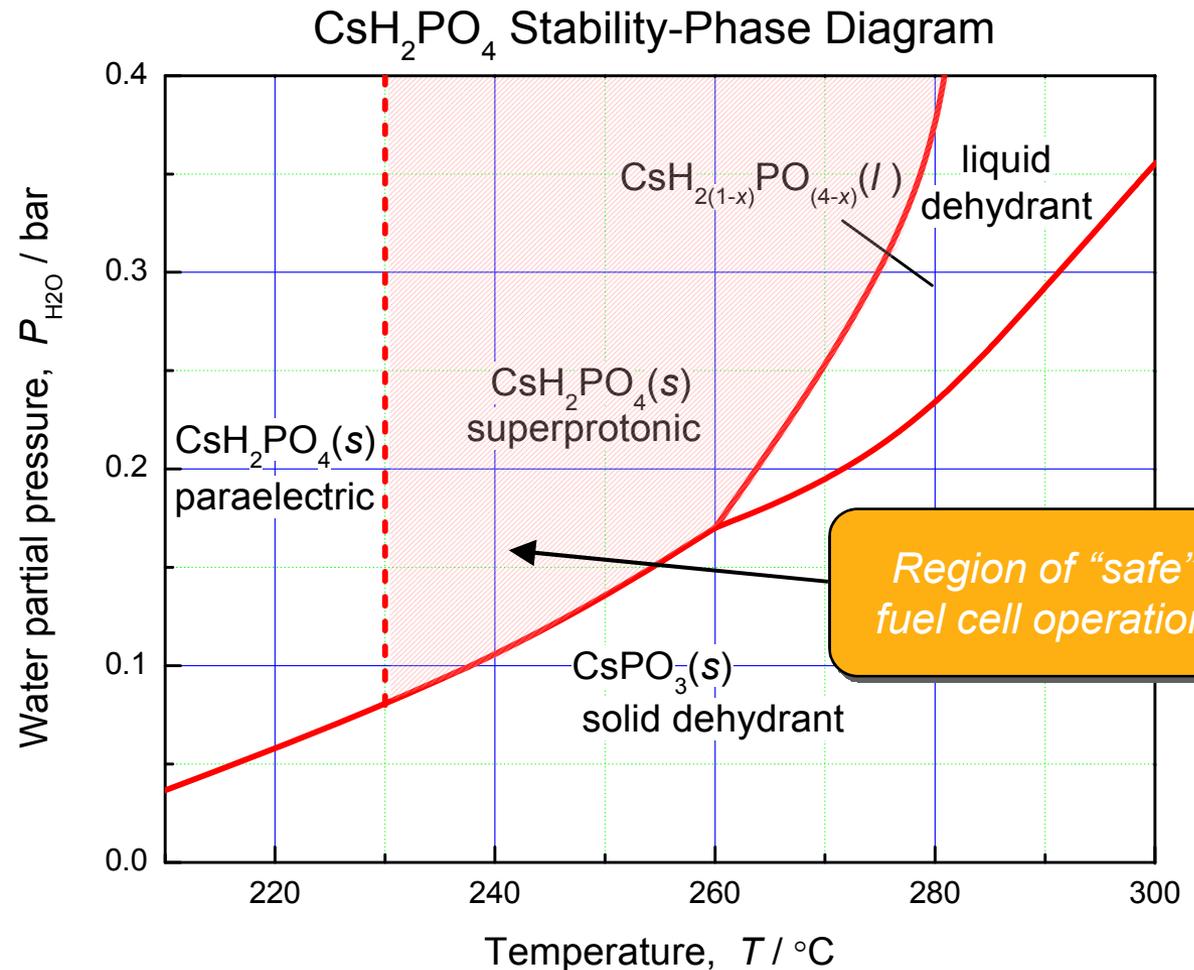
*Dehydration temperature ( $T_d$ ) is almost commensurate with superprotonic phase transition temperature ( $T_{\text{SP}}$ ) at ambient humidity levels*

$$T_d \approx T_{\text{SP}} \quad (P_{\text{H}_2\text{O}} < 0.1 \text{ bar})$$

# Cesium Dihydrogen Phosphate

## Humidity-temperature stability-phase diagram

Can easily avoid electrolyte dehydration with slight humidification ( $T_{dew} > 65\text{ }^{\circ}\text{C}$ )



\*Taninouchi, *J. Mater. Chem.*, 2007; Taninouchi, *Solid State Ionics*, 2007

# Solid Acid Fuel Cells

## Ammonia tolerance (100 ppm $\text{NH}_3$ )

SAFC stable under  
100 ppm  $\text{NH}_3$

