

# **Compact (L)H<sub>2</sub> Storage with Extended Dormancy in Cryogenic Pressure Vessels**

**Salvador Aceves, Gene Berry,  
Francisco Espinosa, Guillaume Petitpas,  
Tim Ross, Vernon Switzer, Ray Smith**

**Lawrence Livermore National Laboratory  
June 8, 2010**

This presentation does not contain any proprietary or confidential information

**Project ID #  
ST003**



# Overview

## Timeline

- Start date: **October 2004**
- End date: **Sept. 2011**
- Percent complete: **80%**

## Budget

- Total project funding
  - DOE: **\$4.5M**
- Funding for FY09:
  - **\$2.25M**
- Funding for FY10:
  - **\$440k**

## Barriers

- **A. Volume and weight**
- **B. Cost**
- **O. Hydrogen boil-off**

## Targets

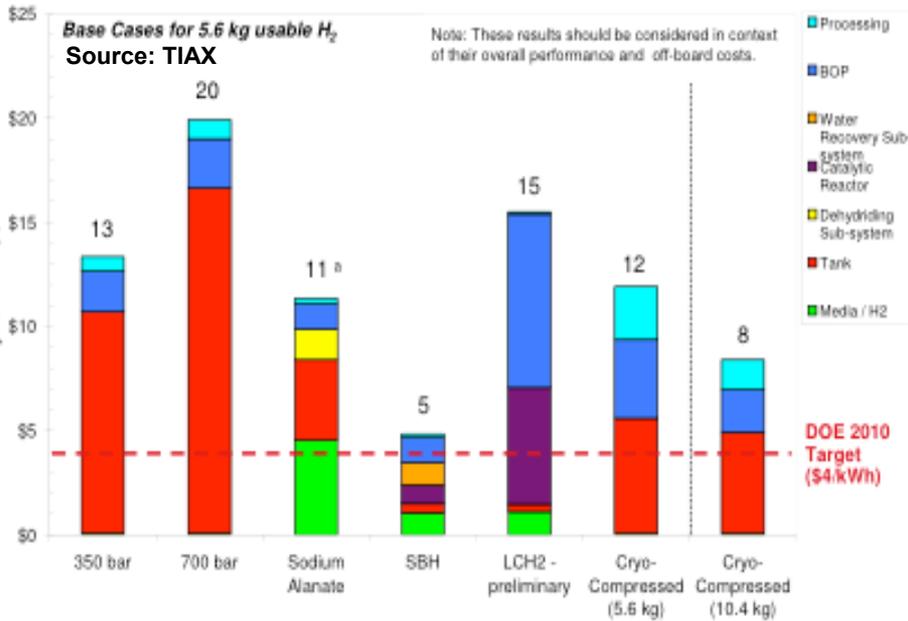
- **Ultimate volume target**
- **Ultimate weight target**

## Partners

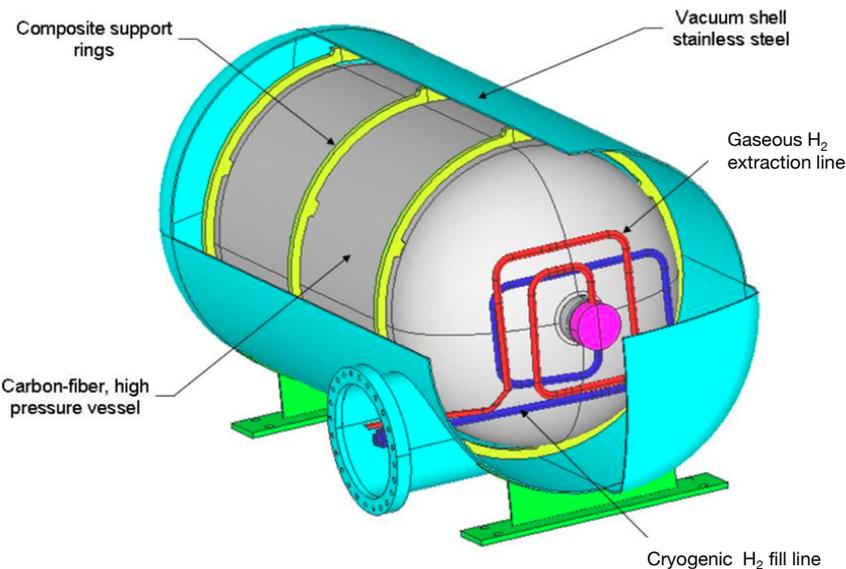
- **CRADA** with BMW
- **CRADA** with Structural Composites Industries (SCI)



# Relevance: High density cryogenic hydrogen enables compact, lightweight, and cost effective storage



- **Cost effective:** Cryogenic vessels use 2-4x less carbon fiber, reducing costs sharply at higher capacity

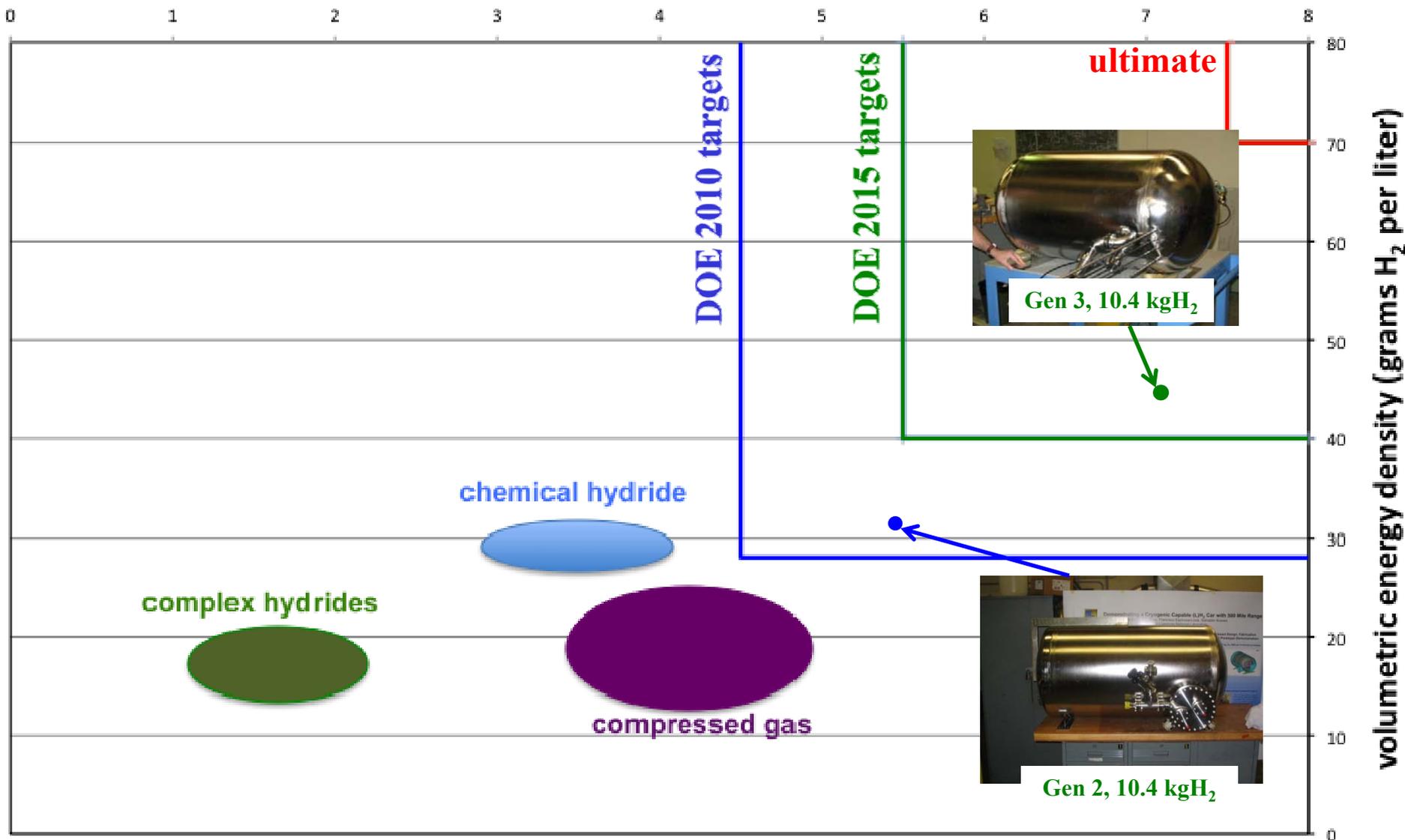


- **Compact:** 235 L system holds 151 L fuel (10.3-10.7 kg H<sub>2</sub>)

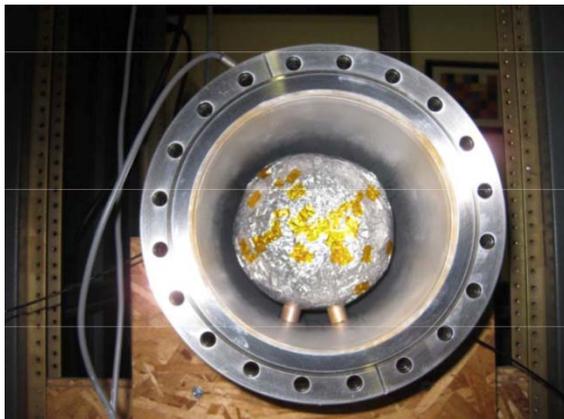
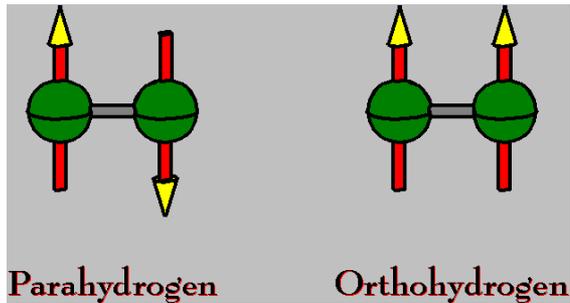


# Relevance: Cryogenic pressure vessels can exceed 2015 H<sub>2</sub> storage targets and approach *ultimate*

gravimetric energy density (H<sub>2</sub> Weight %)



# Approach: reduce/eliminate H<sub>2</sub> venting losses by researching vacuum stability, insulation, and para-ortho conversion

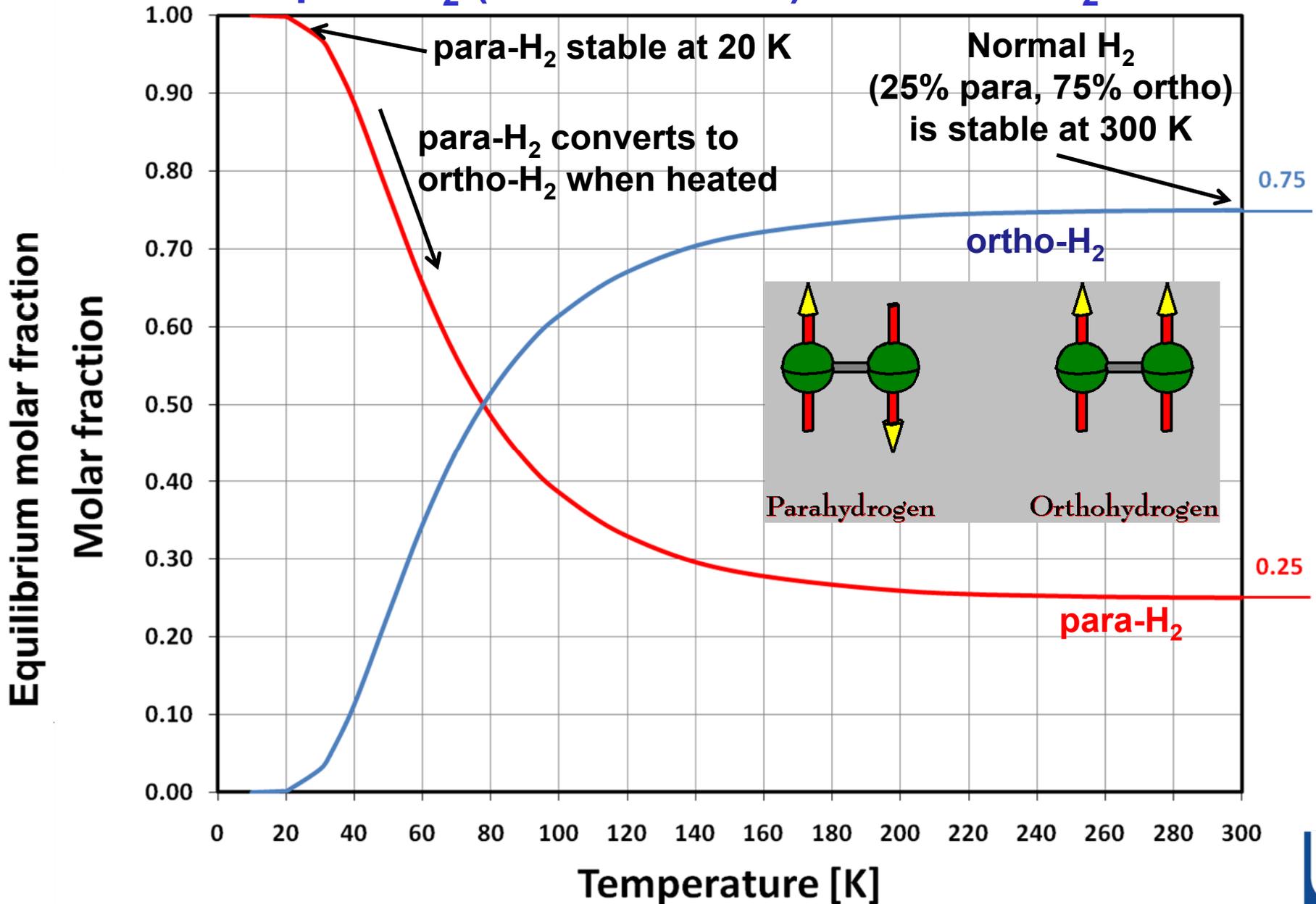


Hydrogen annual merit review, LLNL, June 8, 2010, p. 3

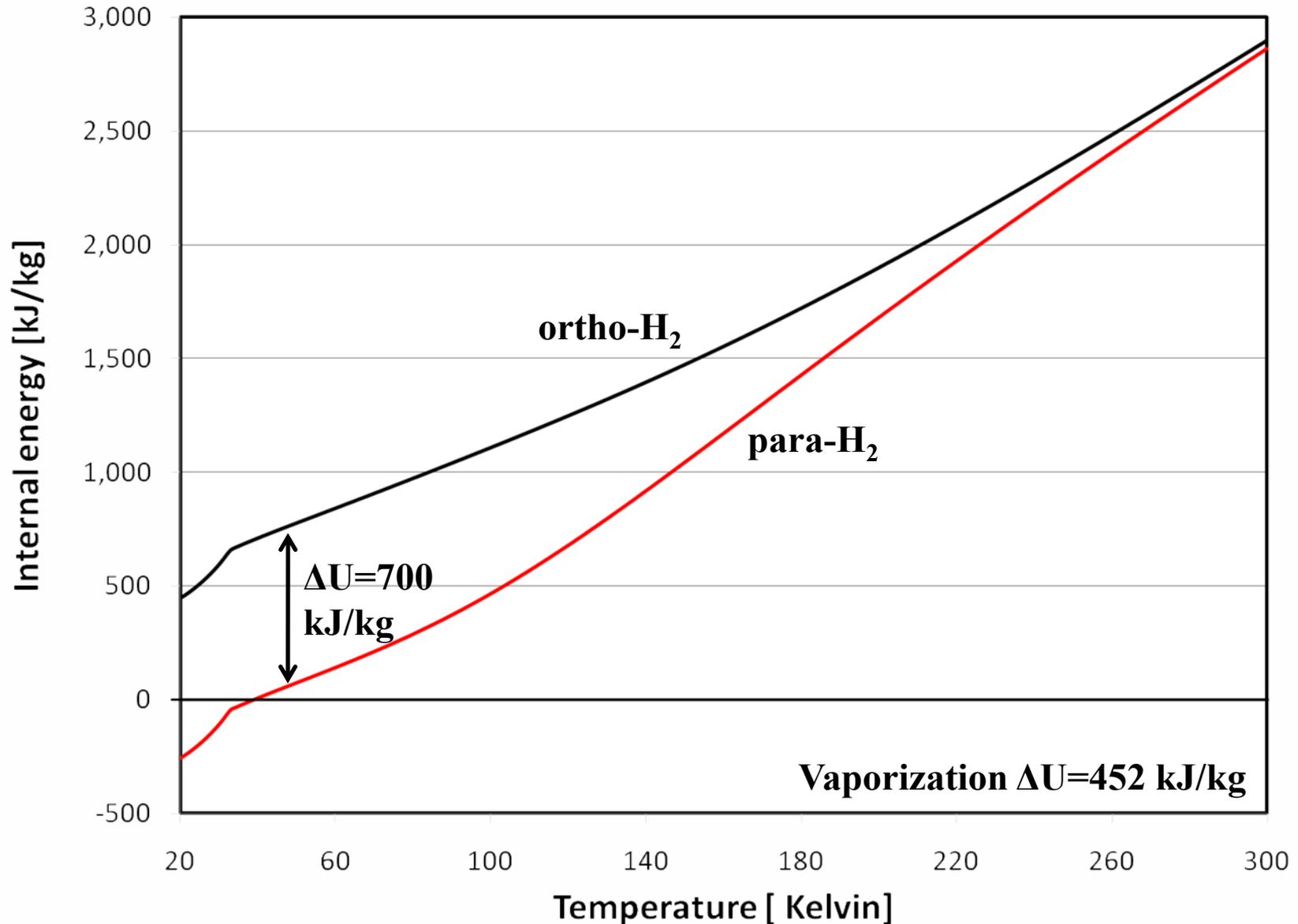
- *Determine para-ortho effect on pressurization and venting losses*
- *Directly measure para-ortho populations*
- *Determine vessel heat transfer mechanism (radiation vs. conduction)*
- *Evaluate vacuum stability by measuring pressure vessel outgassing*
- *Test ultra thin insulation for improved vessel volume performance*
- *Improve vessel design based on experimental results*



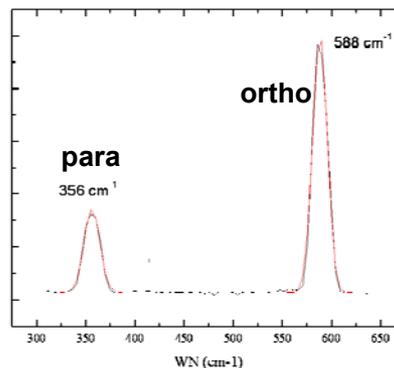
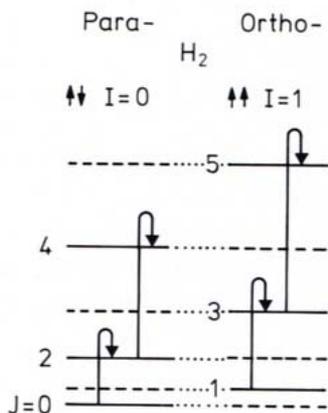
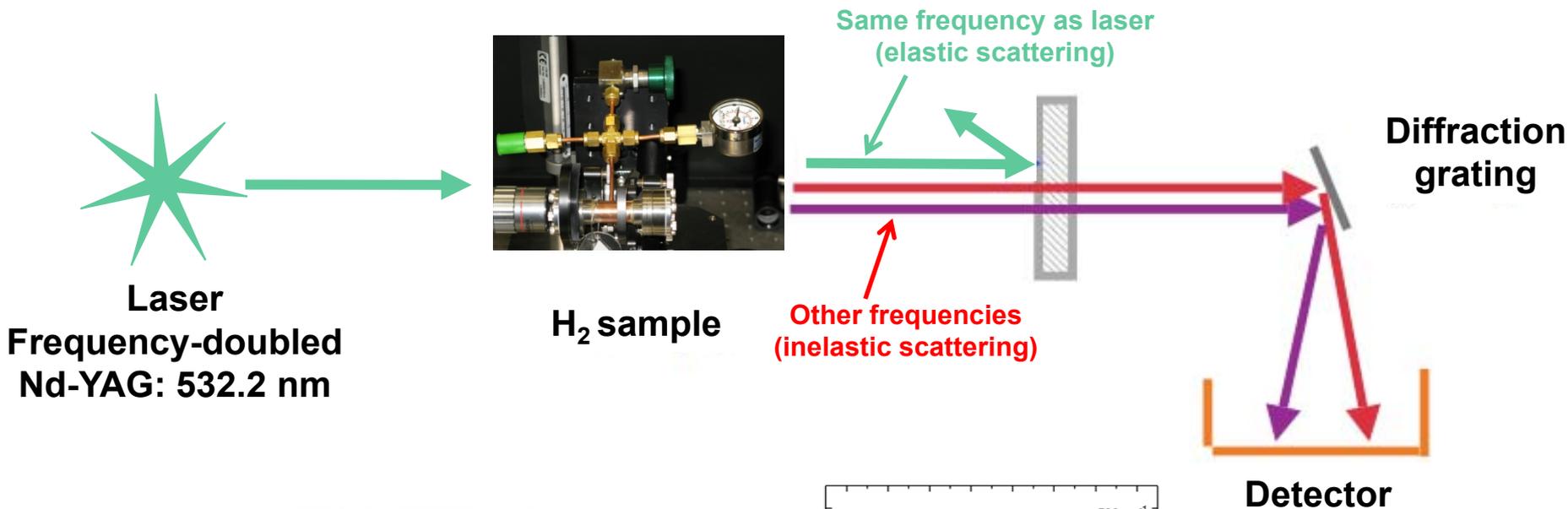
# Hydrogen has two nuclear spin states: para-H<sub>2</sub> (stable at 20 K) and ortho-H<sub>2</sub>



# Para-ortho conversion absorbs energy & increases dormancy (equivalent to a second evaporation)



# Rotational Raman spectroscopy quantifies the population of para-H<sub>2</sub> and ortho-H<sub>2</sub> energy levels



$$J=0 \rightarrow J=2 \quad J=1 \rightarrow J=3$$

Rotational transition lines for normal H<sub>2</sub> (75% ortho)



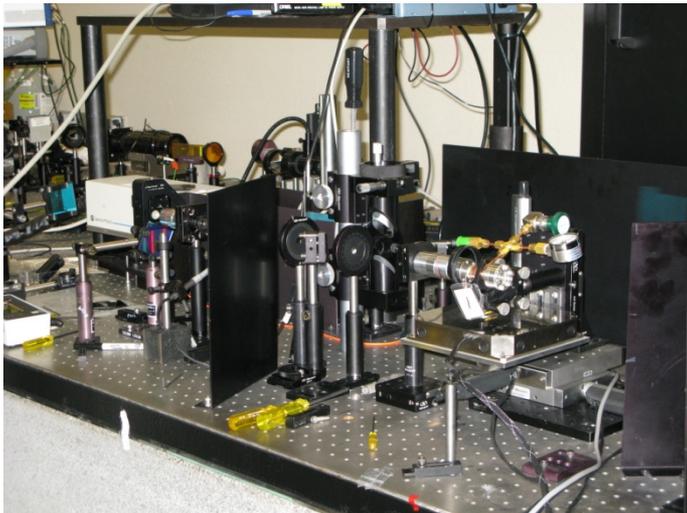
# Gas samples from full-scale vessel were analyzed for para-H<sub>2</sub> concentration within 10 minutes of collection



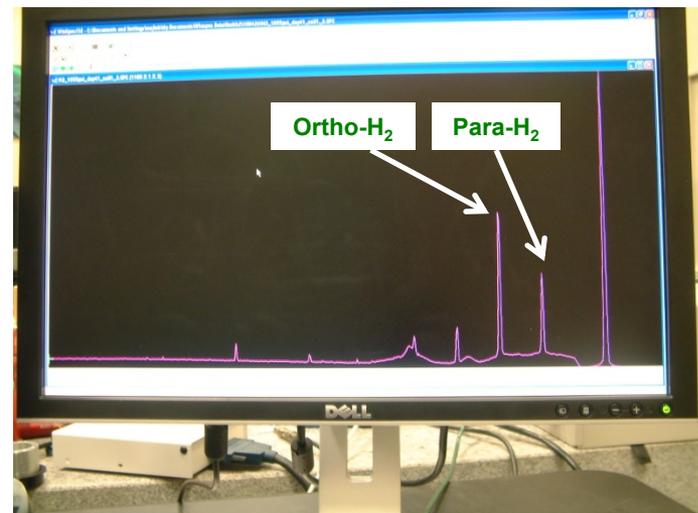
**Full-scale vessel**



**Sample delivery (~10 min) to laser lab**



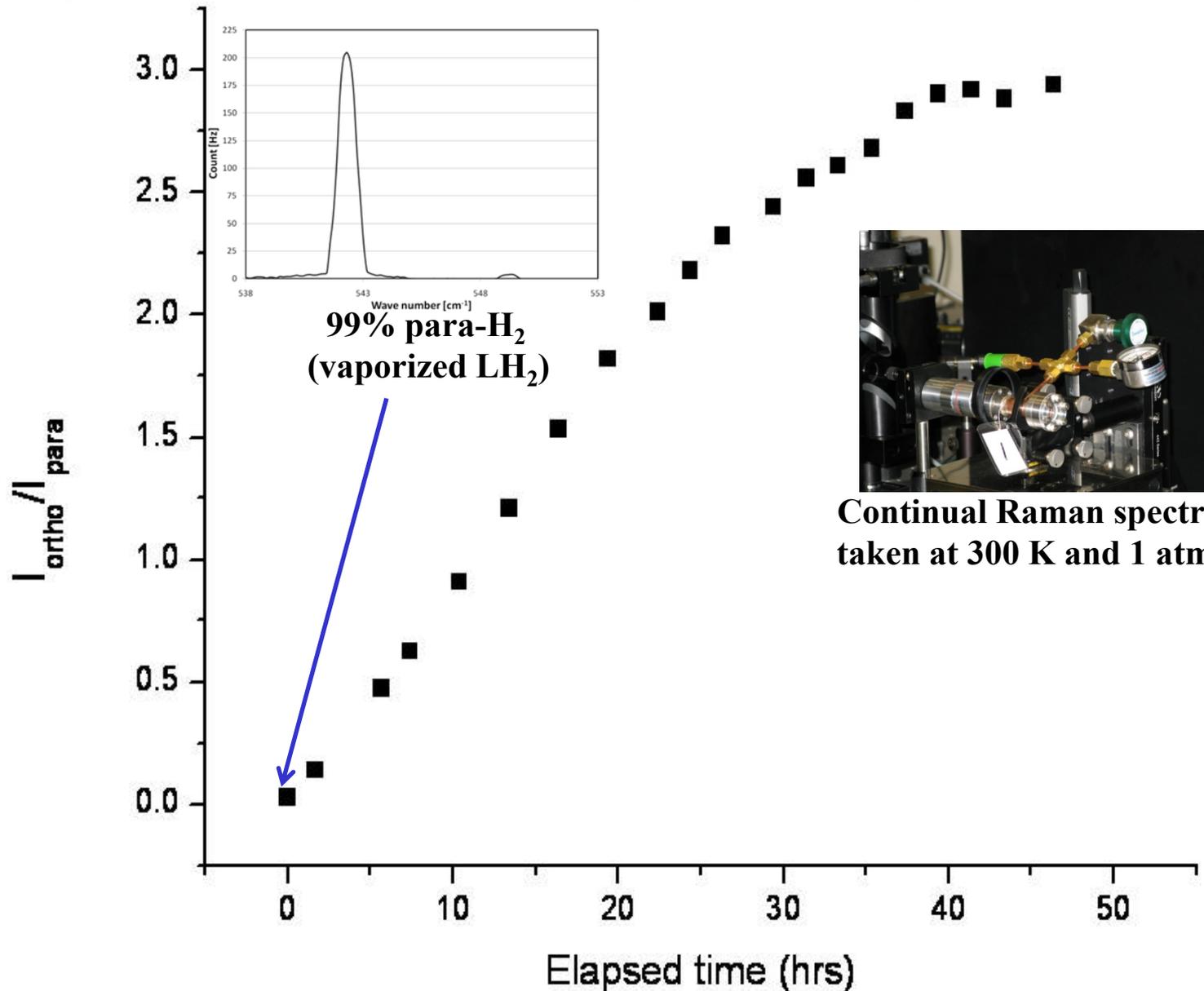
**Confocal laser Raman system**



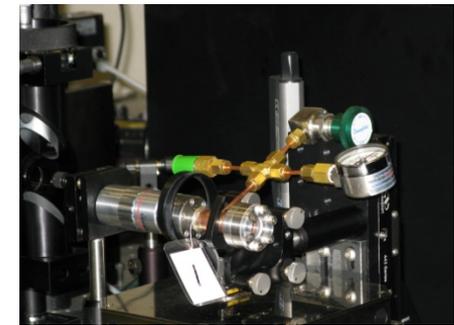
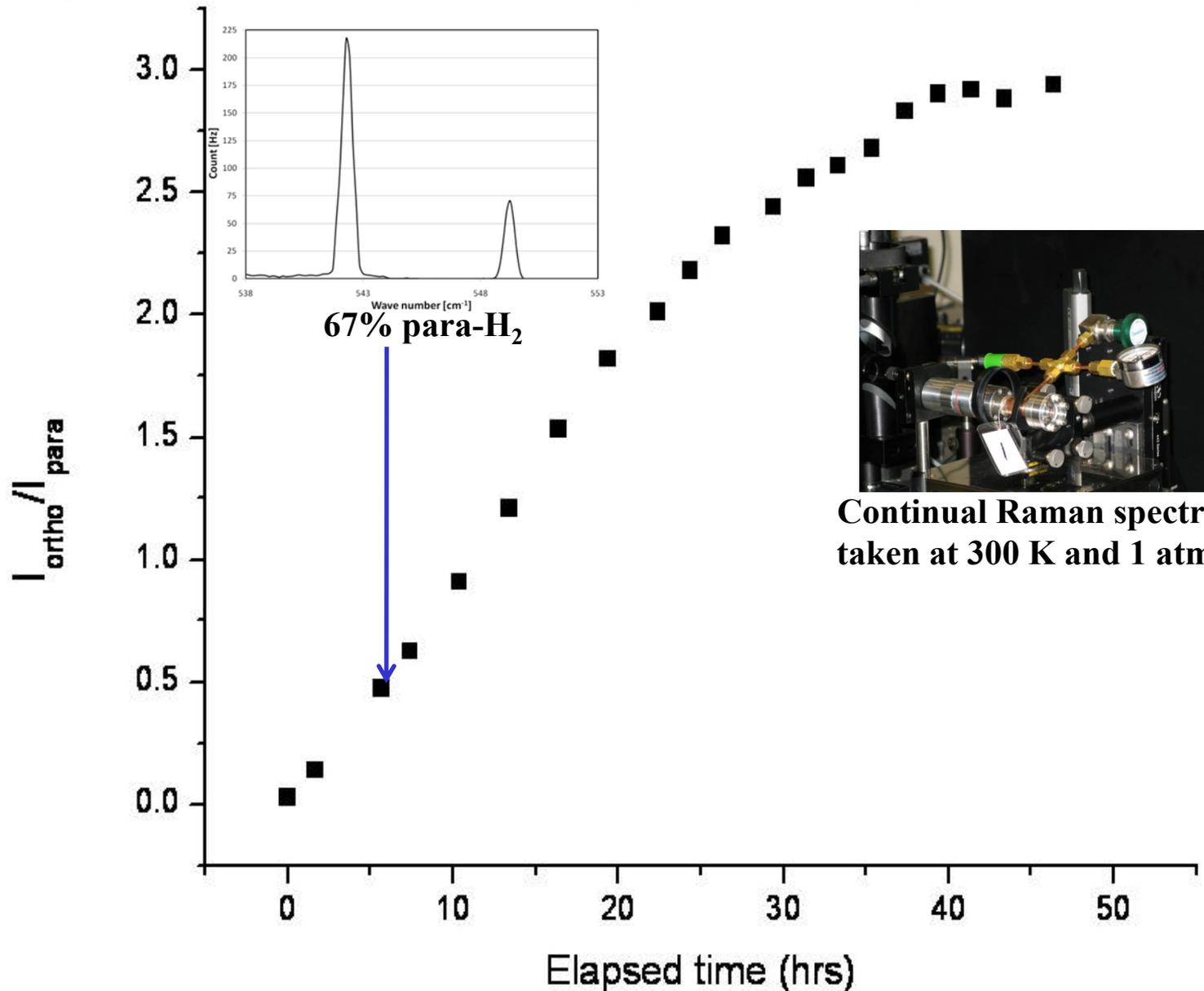
**Spectral results**



# Continual monitoring of para-ortho H<sub>2</sub> conversion In sample vessel validated experimental procedures

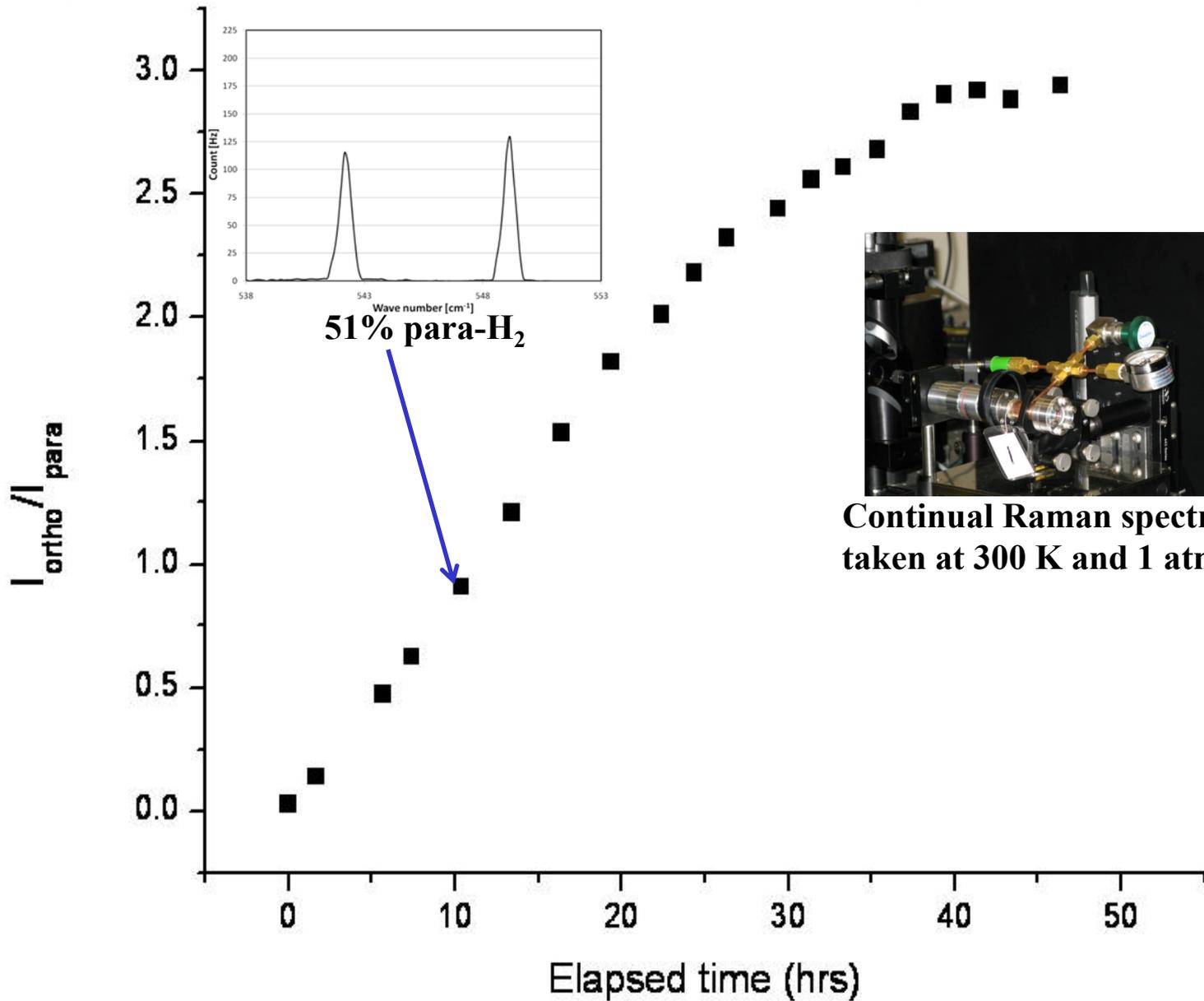


# Continual monitoring of para-ortho H<sub>2</sub> conversion In sample vessel validated experimental procedures

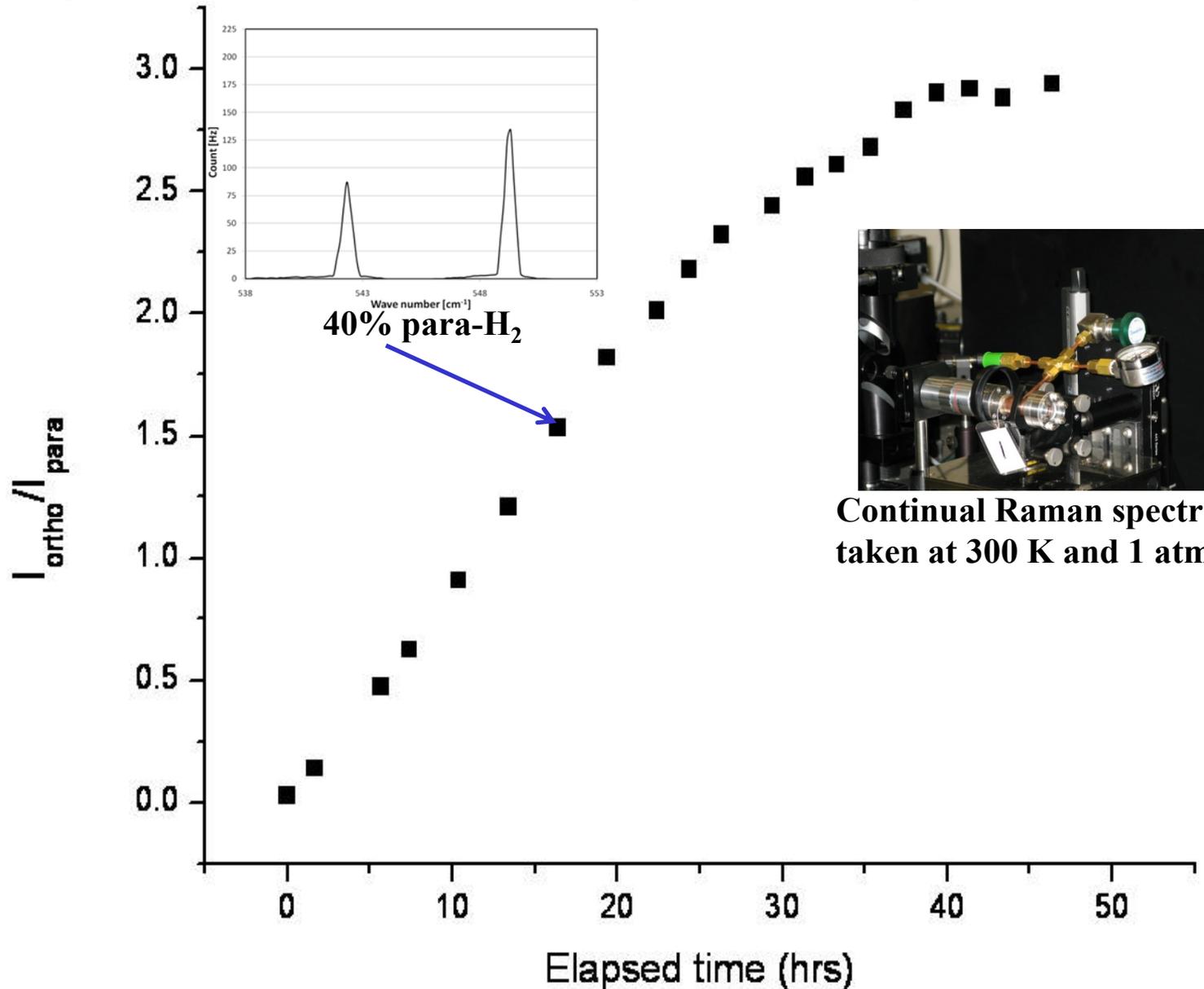


Continual Raman spectra taken at 300 K and 1 atm.

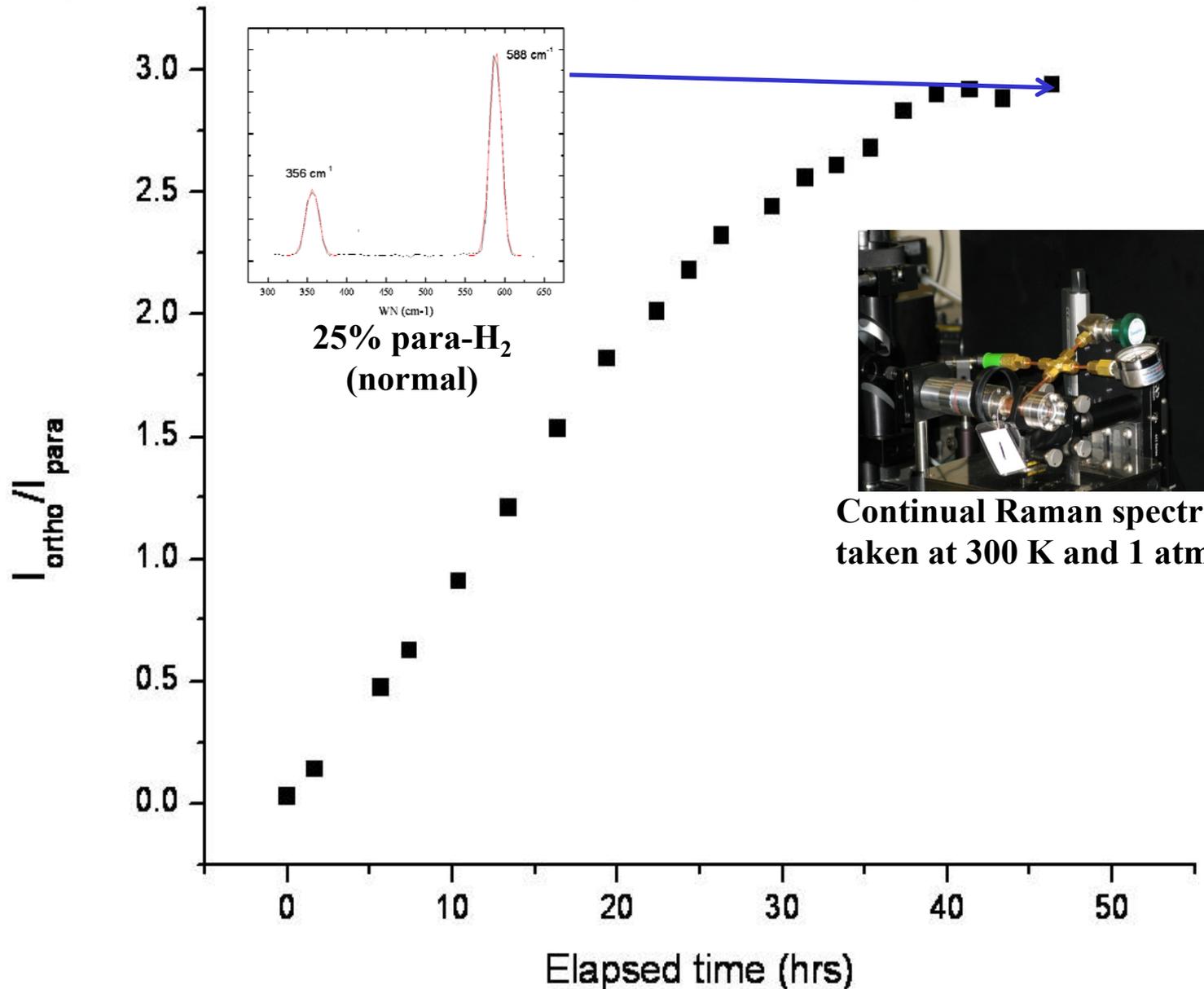
# Continual monitoring of para-ortho H<sub>2</sub> conversion In sample vessel validated experimental procedures



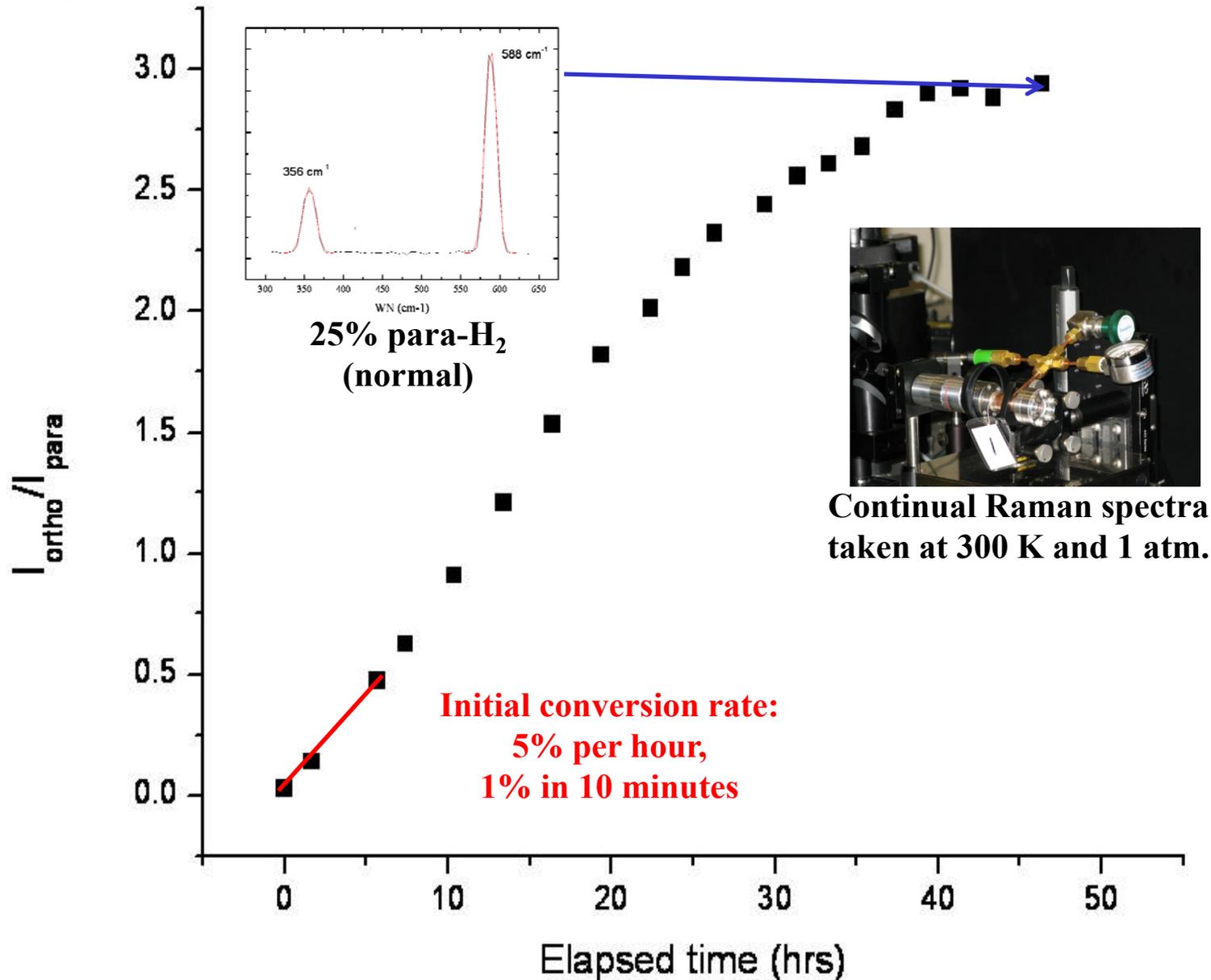
# Continual monitoring of para-ortho H<sub>2</sub> conversion In sample vessel validated experimental procedures



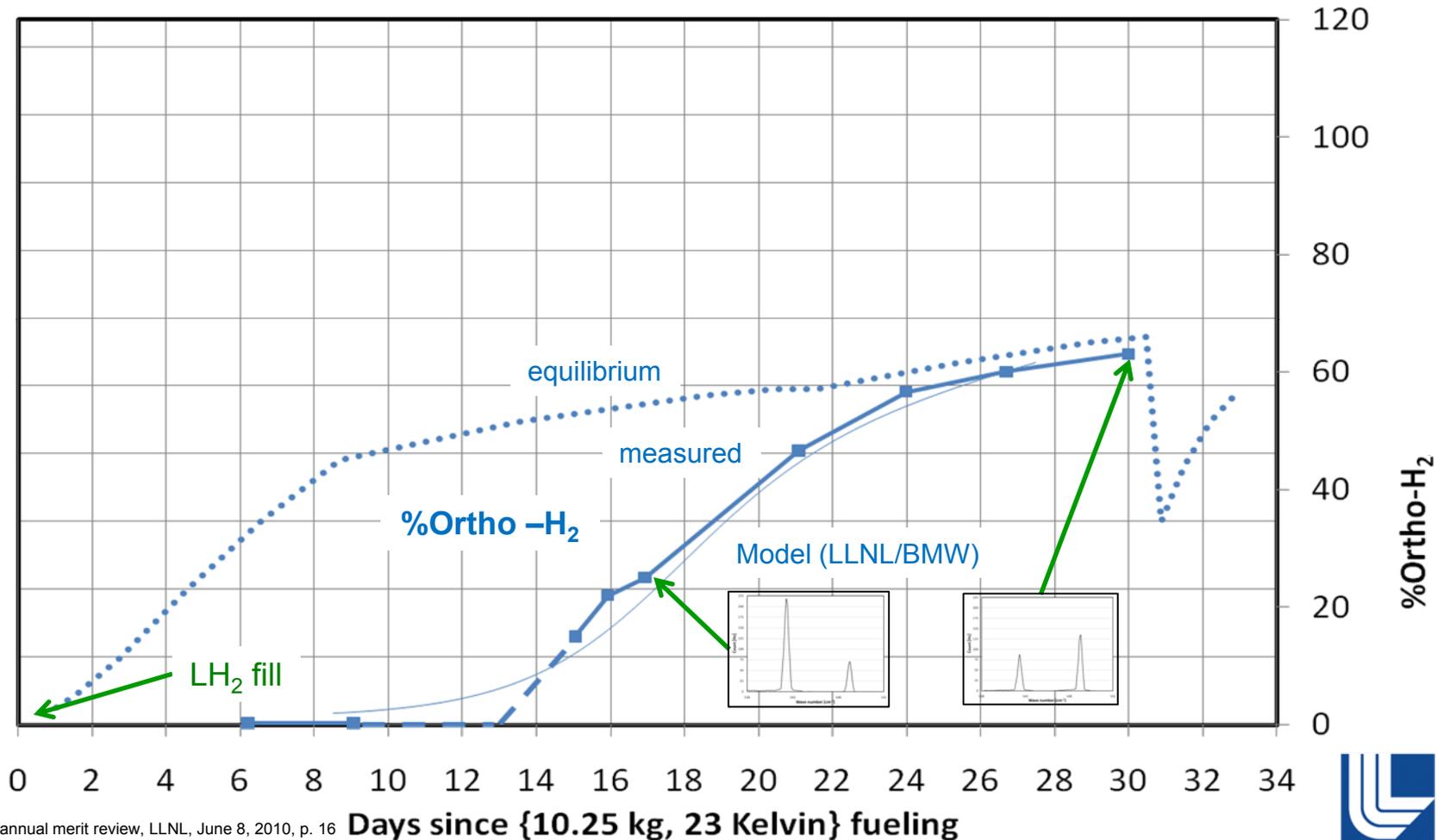
# Continual monitoring of para-ortho H<sub>2</sub> conversion In sample vessel validated experimental procedures



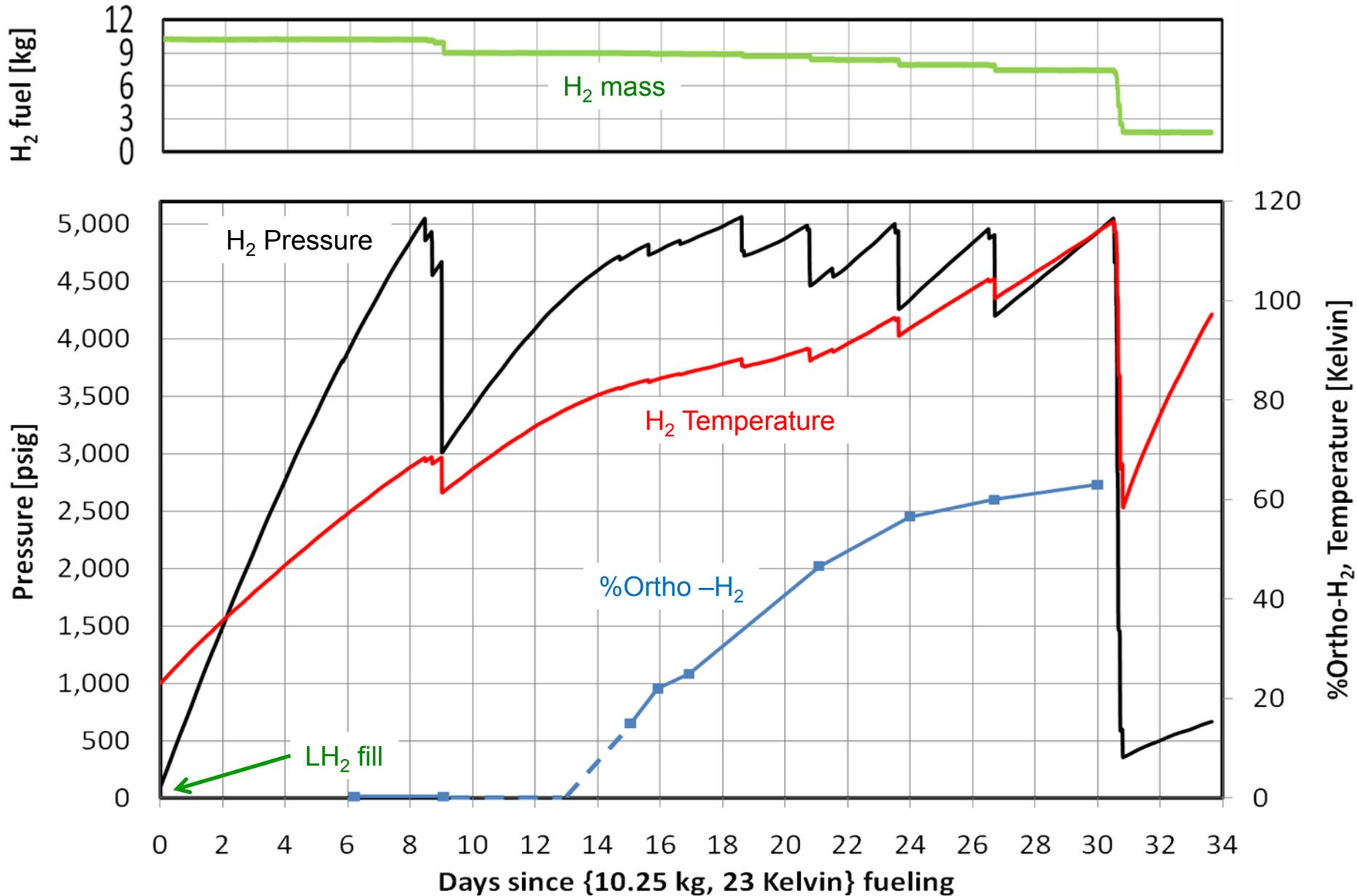
# Continual monitoring of para-ortho H<sub>2</sub> conversion In sample vessel validated experimental procedures



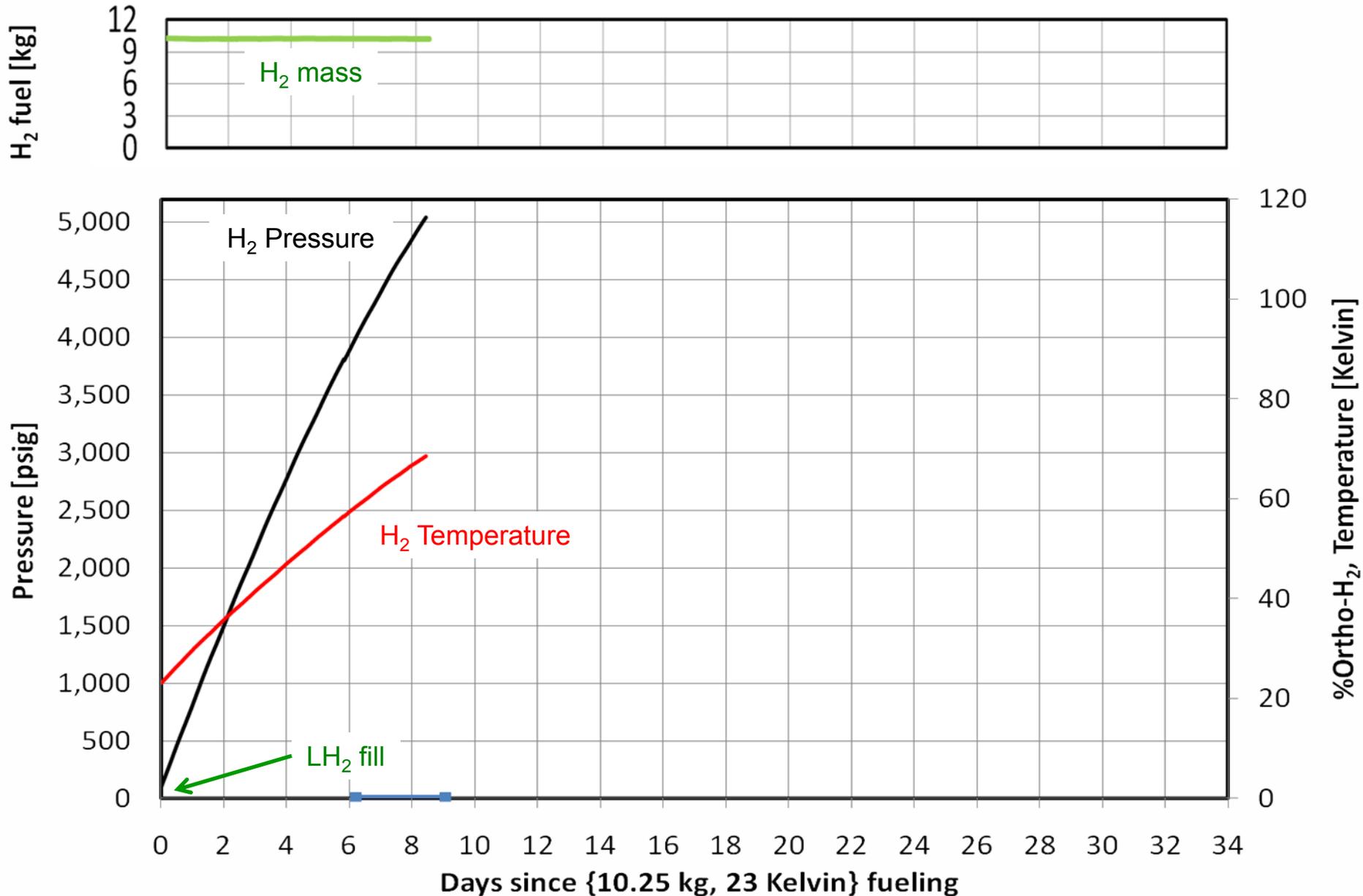
# Raman measurements of samples from Prius show a 12 day lag of ortho conversion then S-curve approach to equilibrium



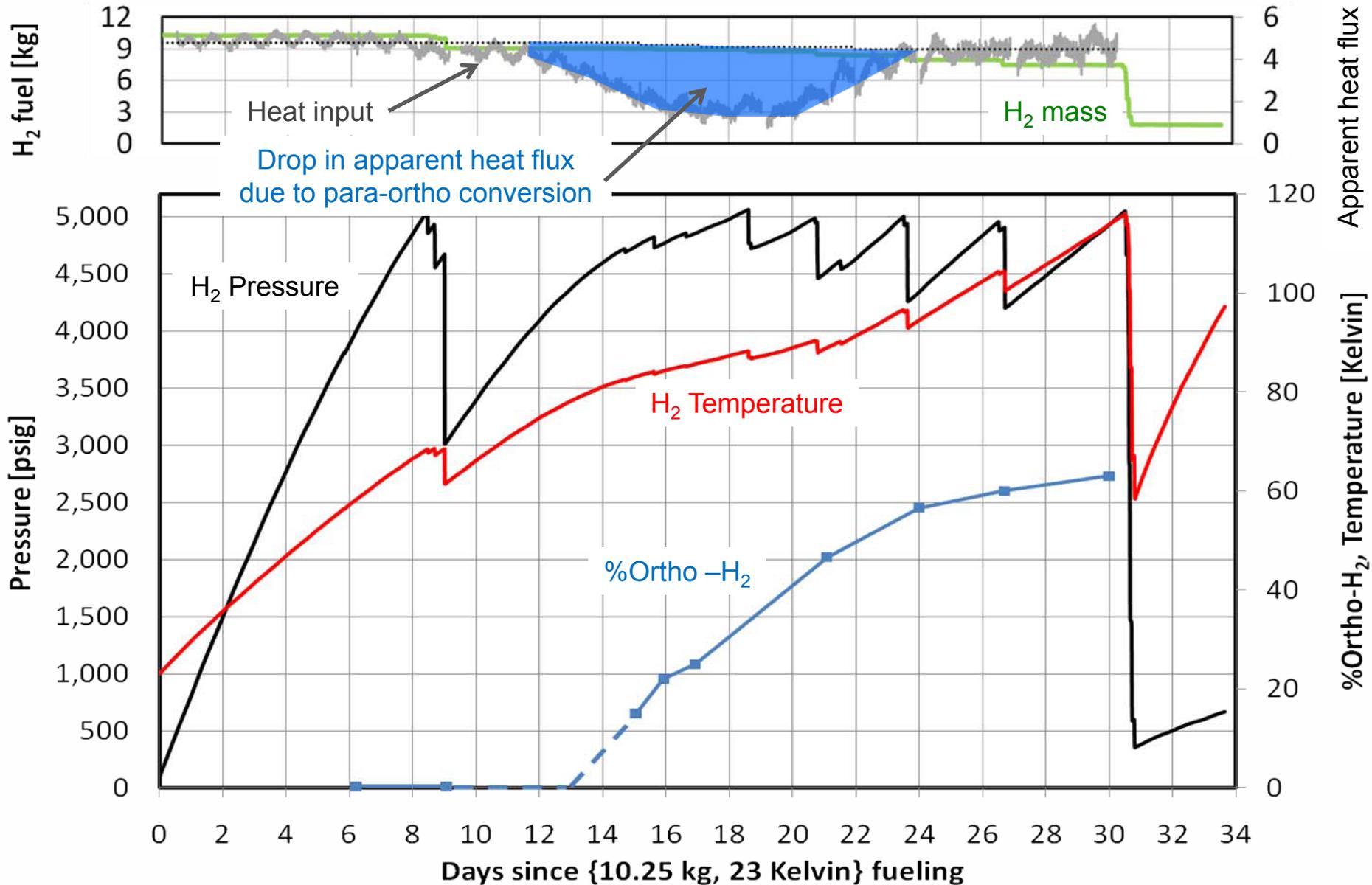
# LLNL's 5,000 psi cryogenic pressure vessel fueled with LH<sub>2</sub> retained 75% of its fuel after a 1 month experiment



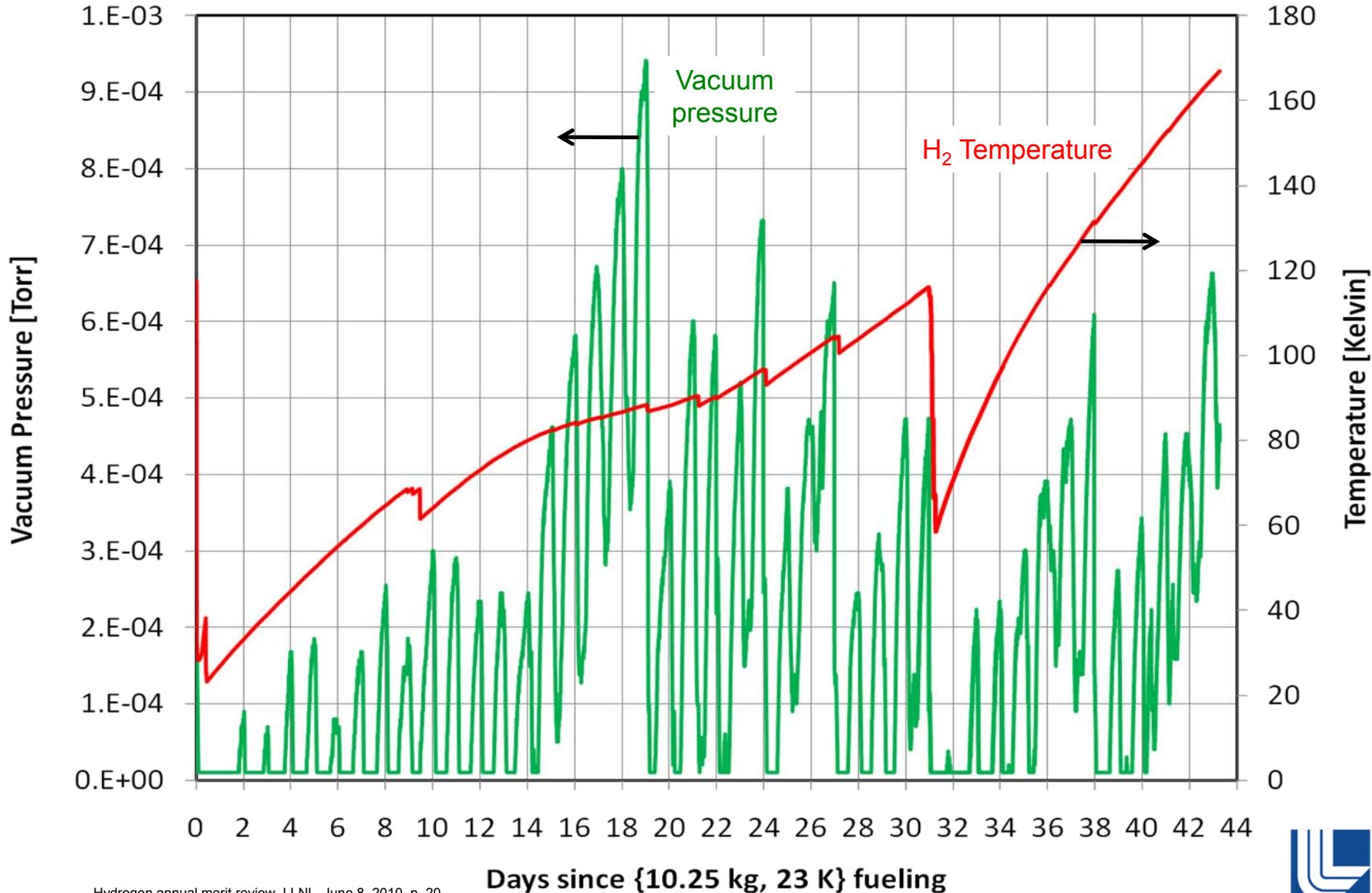
# A 95% full vessel warms from 23 K to 69 K in 8.3 days, pressurizing to 5,000 psi with no evaporative loss



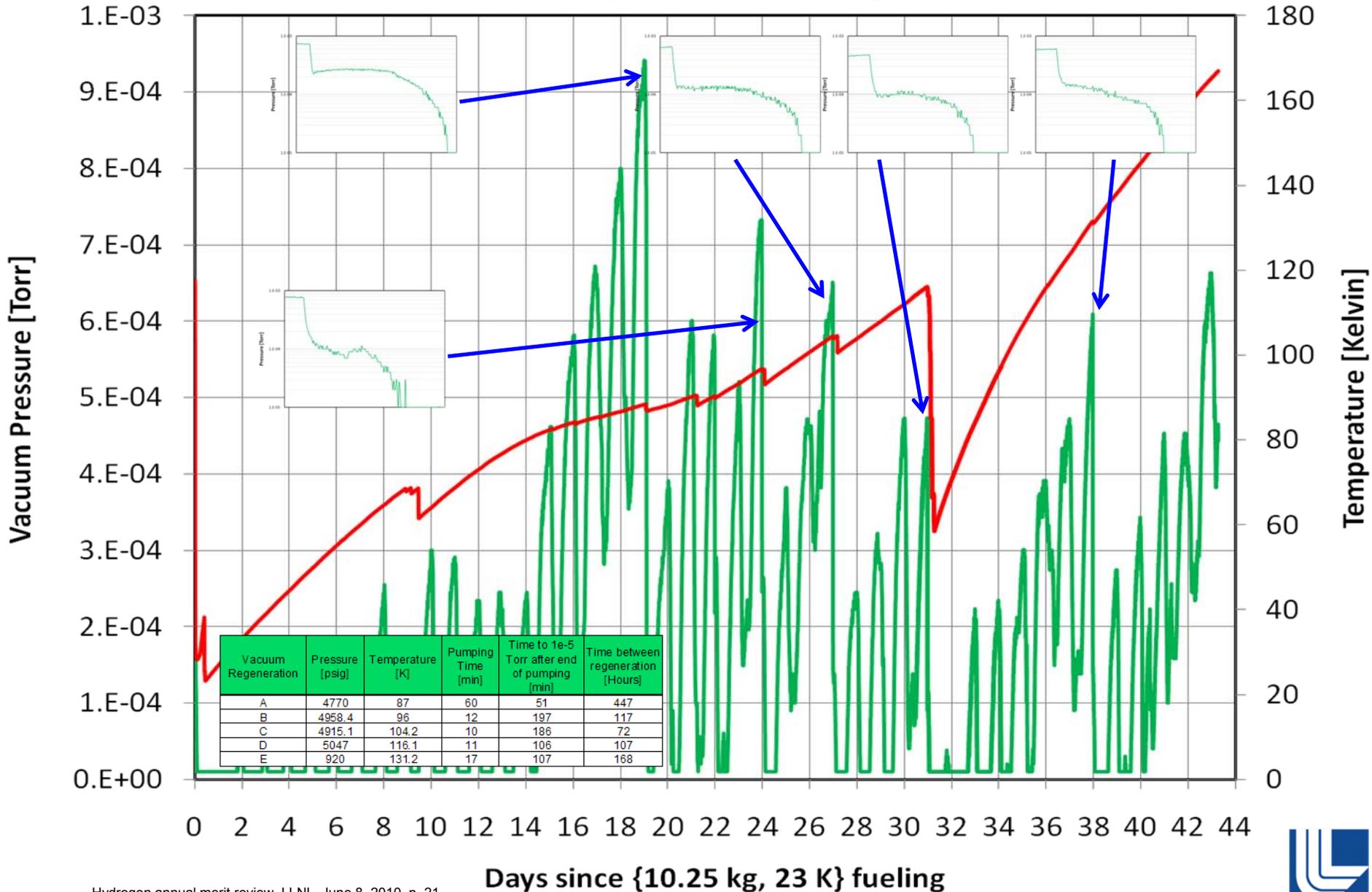
# Conversion to ortho-H<sub>2</sub> was observed between 75 and 110 K increasing dormancy by ~1 week



# Without a getter, vacuum quality degraded measurably after two weeks, at temperatures above 80 K



# Limited vacuum pumping (~ 10 minutes every 4 days) restored vacuum quality even at higher temperatures



# Composite outgassing research necessary for establishing suitable getters for long-term vacuum stability

- *Pre-bake vessels to 80°C:* Determine if H<sub>2</sub>O can be essentially eliminated
- *Run outgassing tests at 20, 60 and 80°C:* Establish effect of temperature on outgassing rate and composition
- *Cycle vessels 10 & 100 times with cooled gas:* separate mechanical and thermal effects by cycling vessels without compression heating
- *Outgassing from vacuum cured vessels with/without UV coating:* Investigate processing effects on outgassing, and potential cycling effects on coatings



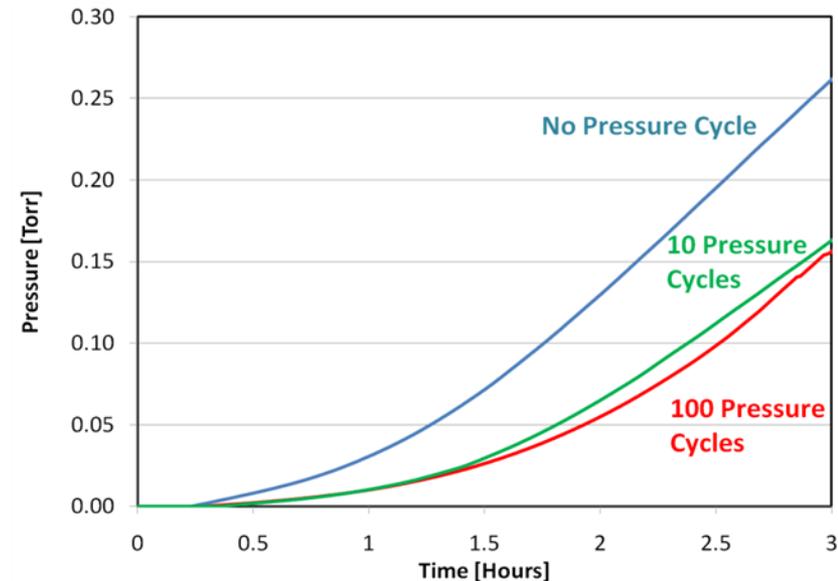
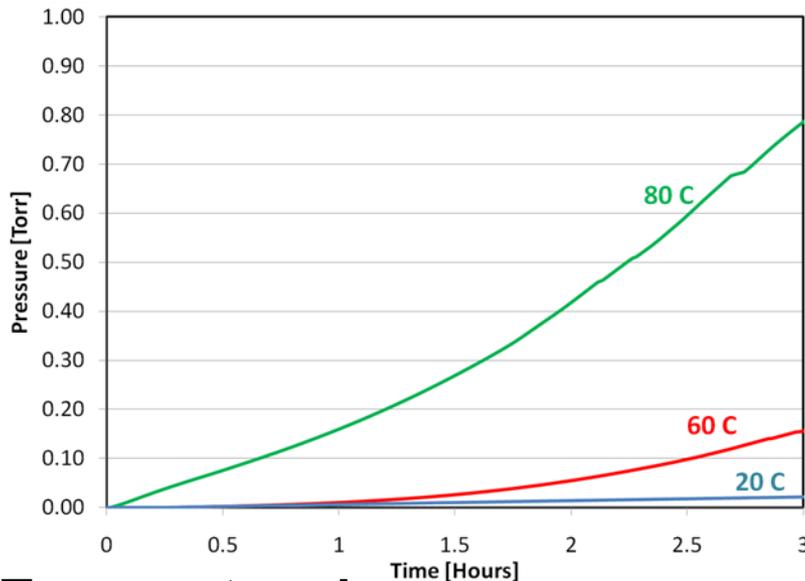
Oven in pressure cell



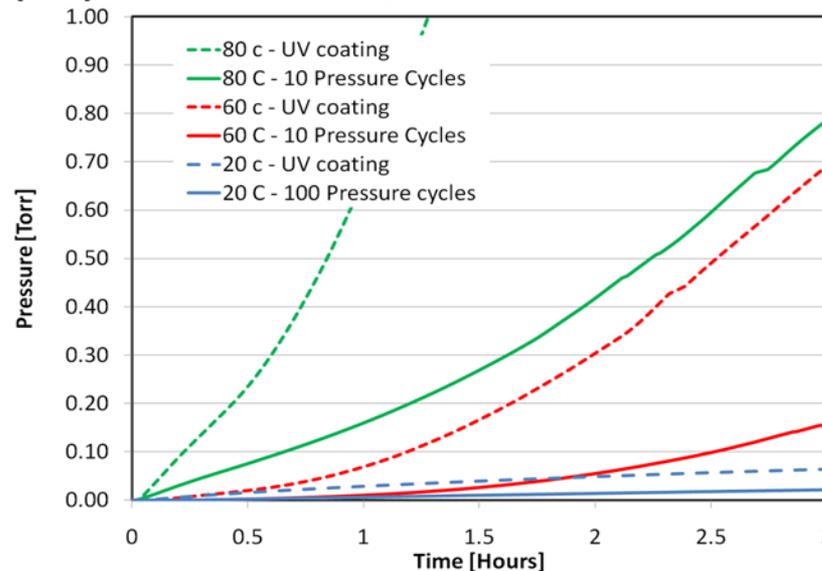
1 liter vessel under vacuum in oven



# We have quantified composite vessel outgassing as a function of temperature, pressure cycling, and surface treatment



**Temperature is most important factor in determining outgassing**



**Number of pressure cycles at fixed temperature plays a relatively minor role**

**Significant increase in residual gas pressure is mainly due to small vacuum volume and not representative of automotive-sized vessels**

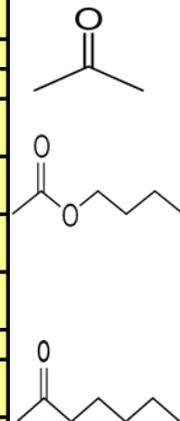
**UV coating considerably increases vessel outgassing**



# Water is eliminated by baking but hydrocarbons are not.

## Detailed outgassing composition necessary for finding appropriate getters for long-term vacuum stability

Compound (and boiling point)	Experiment 1, no cycling			Experiment 2: 10 cycles			Experiment 3: 100 cycles			Experiment 4: 100 cycles	
	20C	60C	80C	Cycle test	60C	80C	Cycle test	20C	60C	20C	60C
Water (100°C)	0	0	0	0	0	0	0	0	0	0	0
Acetaldehyde (20.2°C)					56	140	12	<1	14	<1	36
Acetone, (56.5°C)	<1	180	760	140	140	580	170	11	150	<1	390
Ethanol, (78.4°C)			96			97			20		98
Isopropyl alcohol (83.6°C)	<1	60	80	48	110	220	38	<1	80	<1	160
Acetic acid butyl ester (126°C)	<1	690	1700	270	260	1200	370	10	380	55	740
Ethyl benzene (136°C)	<1	20	50	9	10	36	16	<1	18	1.1	46
Xylenes, total (140°C)	<1	76	240	33	39	160	53	3	72	4.1	160
Styrene (145°C)	<1	21	64	9.2	11	47	14	<1	18	1.1	37
2-heptanone (151°C)	<1	1300	3000	350	570	2400	460	23	770	23	1300
1, 3, 5 trimethylbenzene (164°C)	<1	2	<1	1.2	1.7	4.4	1.6	<1	2.2	<1	<1
1, 2, 4 trimethylbenzene (169°C)	<1	2.8	10	1.6	2.1	7.1	2	<1	3.6	<1	5.4
Carbon disulfide	<1						39	48	10	<1	16
<b>Total hydrocarbons</b>	<b>0</b>	<b>2351.8</b>	<b>6000</b>	<b>862</b>	<b>1199.8</b>	<b>4891.5</b>	<b>1175.6</b>	<b>95</b>	<b>1537.8</b>	<b>84.3</b>	<b>2988.4</b>



# We are acquiring a pressurized cryogenic H<sub>2</sub> fueling capability



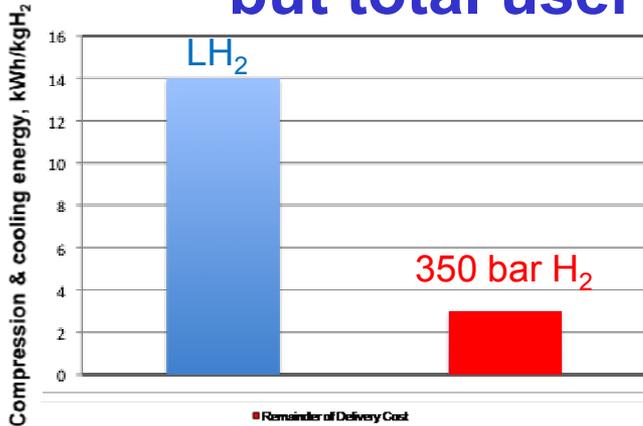
**BMW cryogenic high-pressure pump**

Hydrogen annual merit review, LLNL, June 8, 2010, p. 25

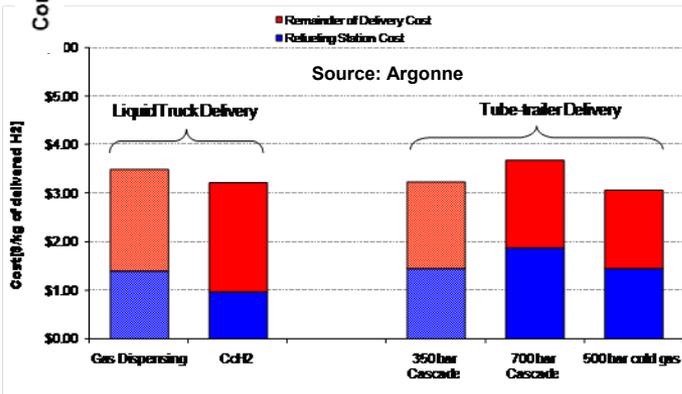
- *We currently fill at low pressure* from a conventional LH<sub>2</sub> storage vessel
- *A high pressure LH<sub>2</sub> pump* offers rapid single phase refueling without boil-off
- *Single flow refueling* can be reliable and cost effective
- *Site Permission and Utilities granted.* Will also serve for high pressure cryogenic H<sub>2</sub> testing
- *For full details,* attend PD074 tomorrow (Wed) at 1:45 pm



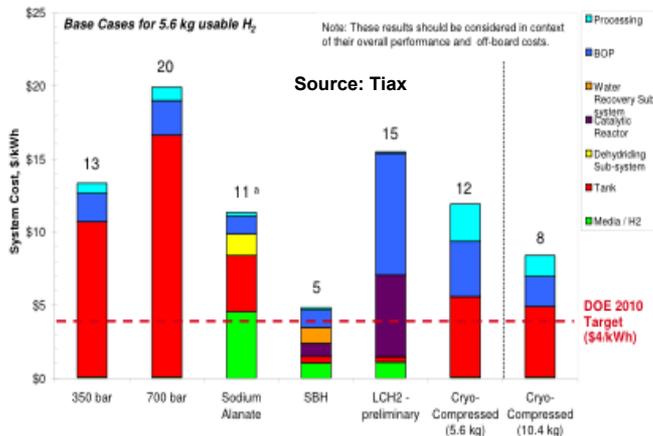
# H<sub>2</sub> liquefaction is energy and capital intensive, but total user energy and cost is what matters



- Liquefaction energy 4x greater than compression energy. Direct energy cost advantage for compression



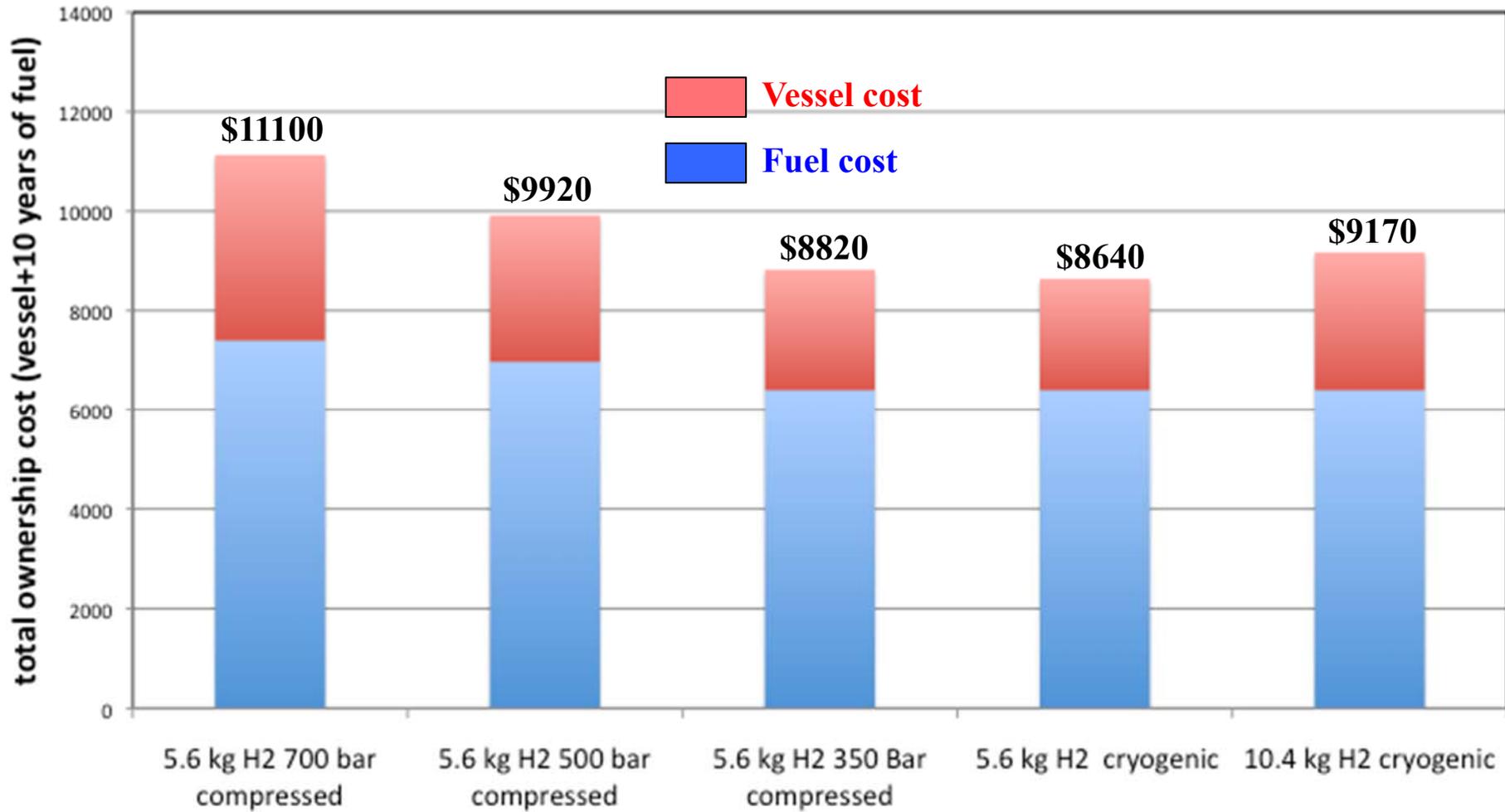
- LH<sub>2</sub> delivery cost comparable to 350 bar Liquefaction costs balanced by lower station costs.



- LH<sub>2</sub> lowers onboard costs: 2-4x less carbon fiber reduces overall cost. *Substantial* cost reduction with higher capacity



# The energy cost of H<sub>2</sub> liquefaction is outweighed by onboard storage & refueling savings, with superior range/volume



**System vol. 224 L**

**262 L**

**329 L**

**160 L**

**231 L**

**Drive range 300 mi**

**300 mi**

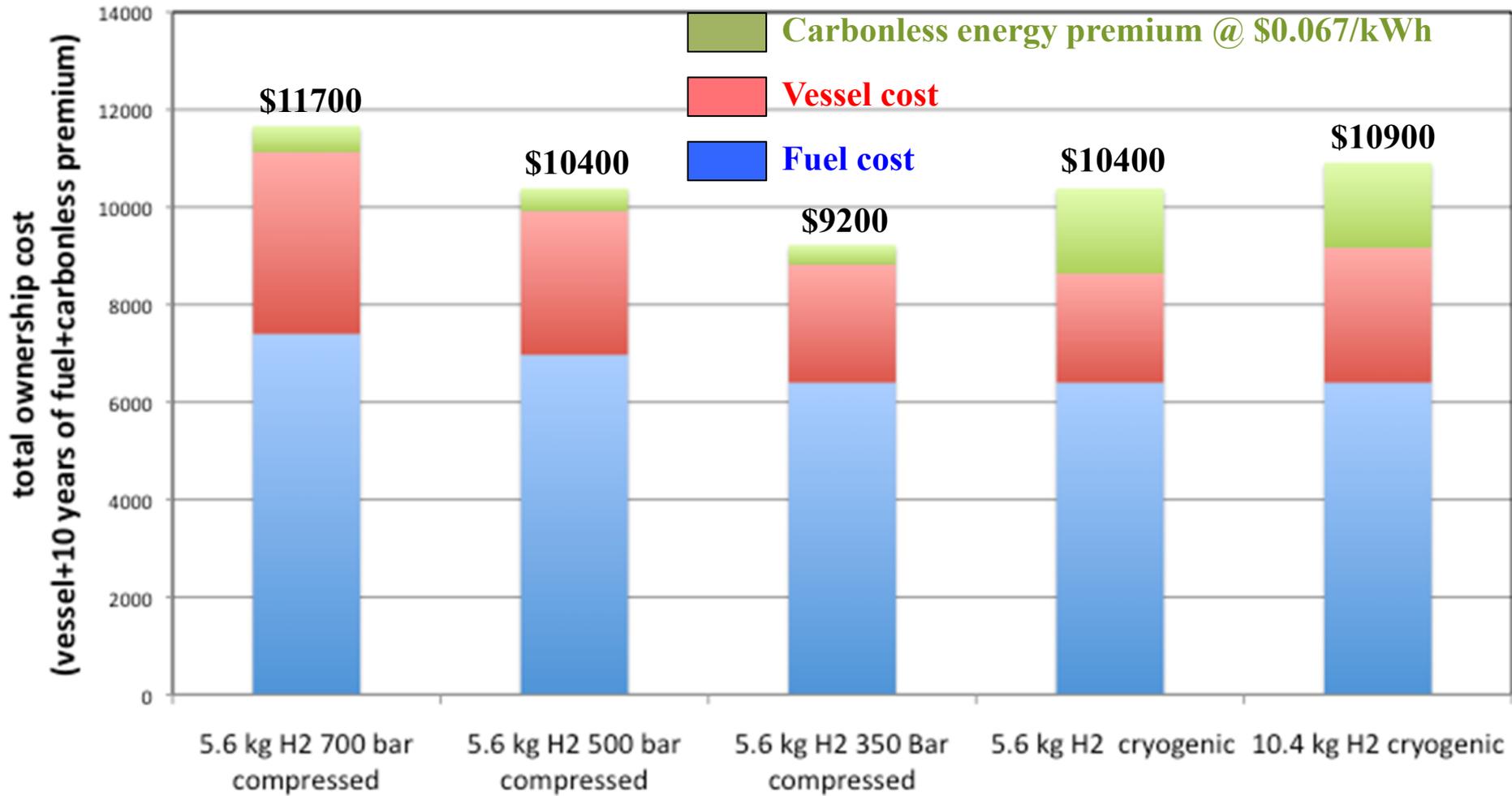
**300 mi**

**300 mi**

**560 mi**



# When considering CO<sub>2</sub> emissions from liquefaction, total energy cost savings offset carbonless energy premium

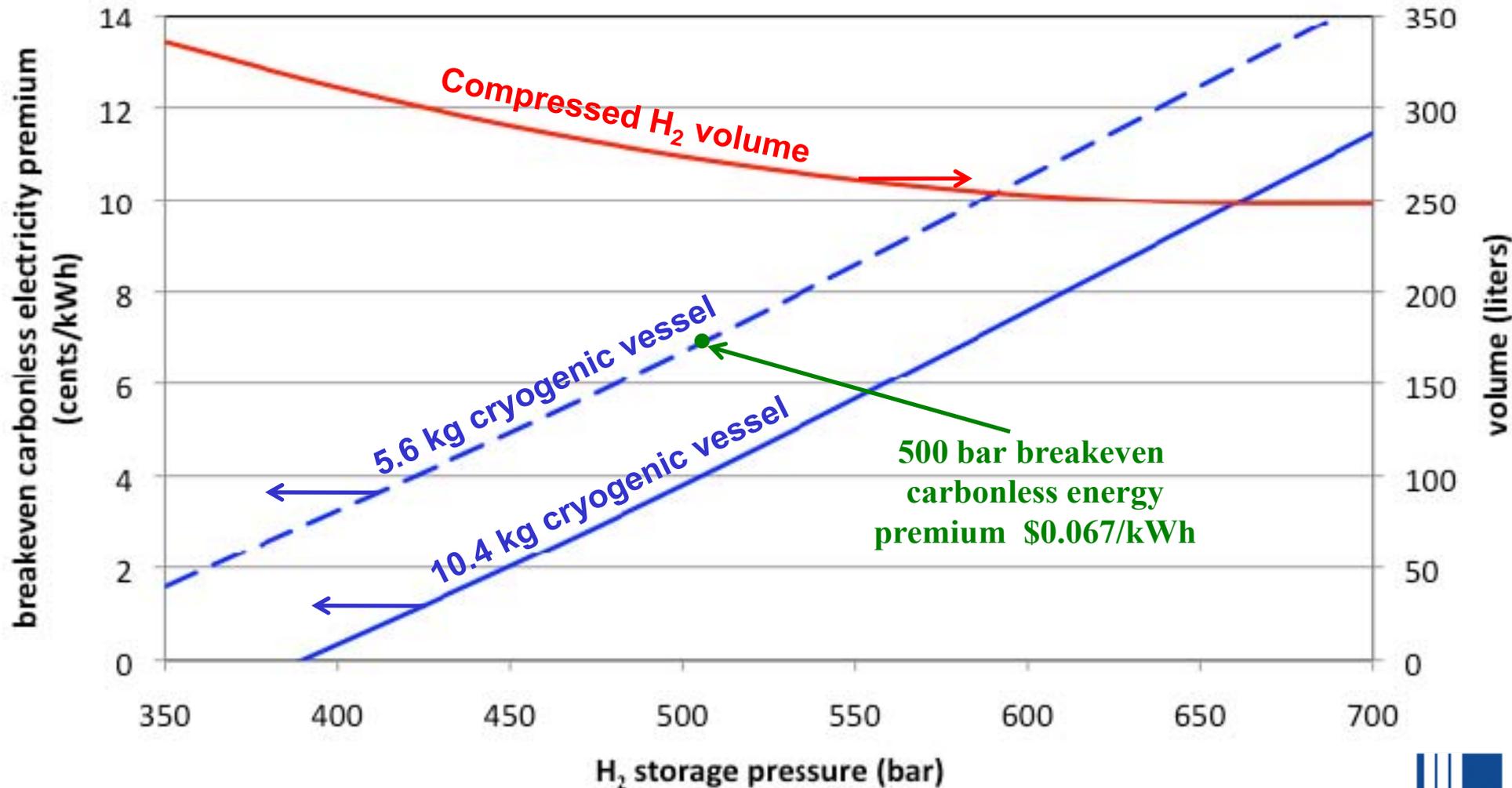


<b>System vol. 224 L</b>	<b>262 L</b>	<b>329 L</b>	<b>160 L</b>	<b>231 L</b>
<b>Drive range 300 mi</b>	<b>300 mi</b>	<b>300 mi</b>	<b>300 mi</b>	<b>560 mi</b>



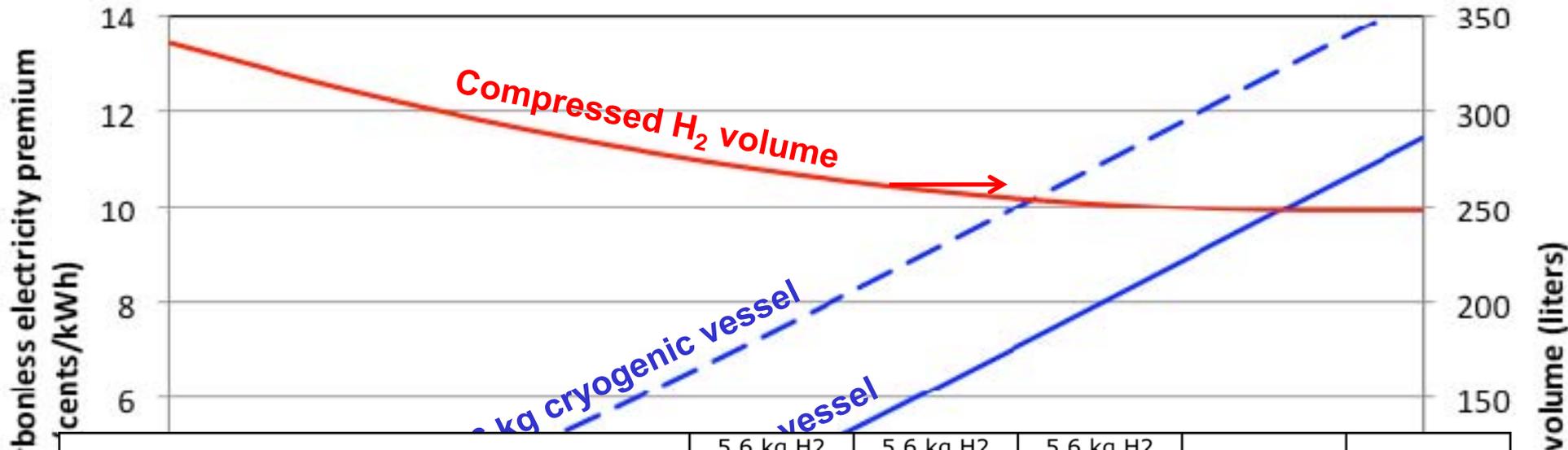
For every storage pressure there is a *reasonable breakeven* carbonless energy premium.

Cryogenic pressure vessels maintain volume/range advantage



For every storage pressure there is a *reasonable breakeven carbonless energy premium.*

Cryogenic pressure vessels maintain volume/range advantage



<b>cost analysis compressed vs. cryogenic</b>	5.6 kg H2 350 Bar compressed	5.6 kg H2 500 bar compressed	5.6 kg H2 700 bar compressed	5.6 kg H2 cryogenic	10.4 kg H2 cryogenic
system weight, kg	95	106	119	102	149
system volume, L	329	262	224	160	231
hydrogen volume, L	258	192	149	81	151
volumetric efficiency	0.78	0.73	0.67	0.51	0.65
hydrogen density, kg/m3	23.2	30.7	39.1	70.7	70.7
H2 mass, kg	6.0	5.9	5.8	5.7	10.7
Usable H2 mass, kg	5.6	5.6	5.6	5.6	10.4
Usable kWh hydrogen	186	186	186	186	346
onboard system cost, \$	2420	2950	3730	2240	2770
10 year fuel cost, \$	6400	6970	7400	6400	6400
total cost, \$	8820	9920	11100	8640	9170
10 year storage energy, kWh	6000	7000	8000	26000	26000
breakeven carbonless electricity premium, cents/kWh	-	0.000	-	0.067	0.039

# Future work: explore performance limits of vessel and cryogenic H<sub>2</sub> behavior: shape, scale, refueling speed, and energy efficiency

- ***Pressurized LH<sub>2</sub>*** offers rapid, efficient refueling and is likely necessary to achieve ultimate DOE H<sub>2</sub> storage goals
- ***Full-scale para-ortho conversion kinetics*** experiments will enable us to determine optimal vessel design parameters as well as whether liquefaction energy and cost can be reduced
- ***Generation 4 vessel*** design to maximize dormancy across the full spectrum of onboard H<sub>2</sub> capacities
- ***Multiple Volume Vessels*** offer flexible blend of capacity, weight, cost, shape, and dormancy over a single state H<sub>2</sub> storage vessel. Multiple H<sub>2</sub> storage states do add complexity



## Collaborations:

We established cooperative research & development (CRADA) agreement with automaker and pressure vessel manufacturers

- ***CRADA with BMW*** started June 2008 to investigate vacuum stability, conduct cryogenic pressure cycling, and study conversion to *ortho*-H<sub>2</sub>. BMW provides great automotive focus to our experimental and demonstration efforts.
- ***CRADA with Structural Composites Industries (SCI)*** uses LLNL's thermal/mechanical analysis capability and H<sub>2</sub> experience as well as SCI's composite cylinder design & manufacturing expertise to develop efficient and lower cost pressure vessels designed specifically for cryogenic H<sub>2</sub> storage.



## Summary: Cryogenic pressure vessel dormancy and volume advantages have been demonstrated. We are studying vacuum, as well as cryogenic H<sub>2</sub> behavior and refueling

- **8+ day dormancy** (zero evaporative losses) was demonstrated. Our 10.4 kg capacity vessel retained 7.5 kg H<sub>2</sub> after a *month*.
- **Conversion of para-H<sub>2</sub> to ortho-H<sub>2</sub>** doubled dormancy for a nearly full vessel
- **Composite vessel outgassing experiments** produced information necessary to select appropriate getters
- **Pressurized LH<sub>2</sub> refueling capability** will enable rapid refueling with minimum evaporative losses and potentially higher capacity fills

