

Metal Oxide Semiconductor Nanotubular Arrays for Photoelectrochemical Hydrogen Generation

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Overview

Timeline

- Project start date: October, 2006
- Project end date: September, 2014
(Pending DOE approval)
- Percent complete: 85

Barriers

- Barriers addressed:
 - AP. Materials efficiency
 - AQ. Materials durability
 - AR. Bulk material synthesis
 - AS. Device configuration and scale up

Budget

- Total project funding: \$ 3,650 K
 - DOE share: \$ 2,970 K
 - Contractor share: \$ 680 K
- Funding for FY06: \$ 2,970 K

Partners

- National Renewable Energy Laboratory
- University of Arkansas at Little Rock
- University of Nevada Las Vegas
(pending DOE approval)

Objectives

- Overall** **Develop high efficiency hybrid-semiconductor nanotubular materials for hydrogen generation by water splitting**
- 2006-2007**
- Develop new anodization techniques to synthesize high quality and robust titanium dioxide (TiO_2) nanotubes with wide range of nanotubular architectures
 - Understand kinetics and formation mechanism of the TiO_2 nanotubes under different synthesis conditions
- 2007-2008**
- Develop organic-inorganic hybrid photo-anodes
 - Develop multi-junction photoanodes
 - Develop cost-effective cathode materials
- 2008-2009**
- Develop mixed metal oxide nanotubular photoanodes
 - Develop multi-junction photoanodes
 - Design PEC systems for on-field testing under real solar irradiation
- 2009-2010**
- Develop semiconductors which absorb in the visible region of the solar spectrum
- 2010-2011**
- Develop visible light sensitive ferroelectric BiFeO_3 photoanodes based on DFT modeling
- 2011-2012**
- Synthesis of titania nanotubes in mixed acid electrolytes to dope transitional metals
- 2012-2013**
- Enhance visible light absorption of metal oxide nanotubes
 - Develop new synthesis techniques for enhanced photoelectrochemical performance

Approach

Task A. Synthesis and fabrication of photocatalysts

- Increase visible light utility of metal oxide nanotubes by
- Examine new synthesis techniques
 - Effect of surface treatment and effect of light irradiation during synthesis
- Characterization and fundamental understanding of the materials prepared

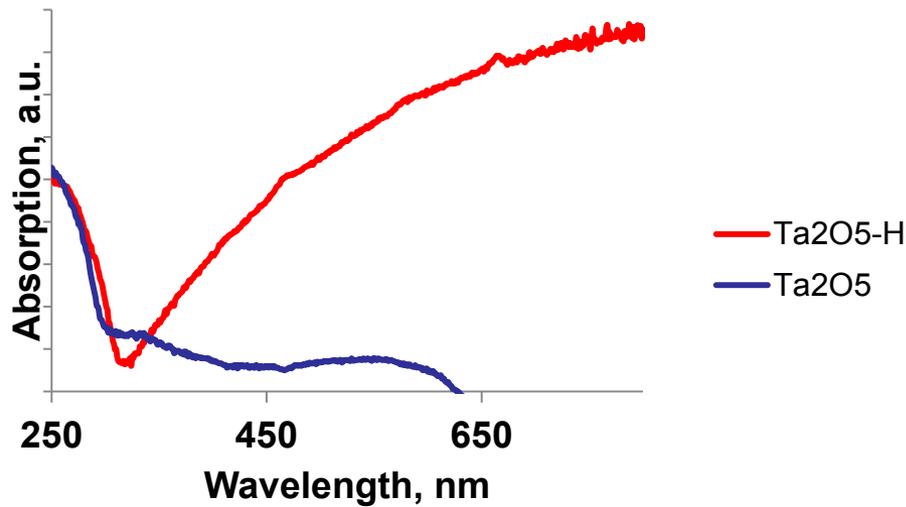
Task B. Application of the nanotubular materials for photoelectrochemical generation of H₂ from Water

- Evaluate photoelectrochemical behavior of nanotubular oxide composite photoanodes

Task C. Materials stability of hybrid oxide nanotubular photo-anodes

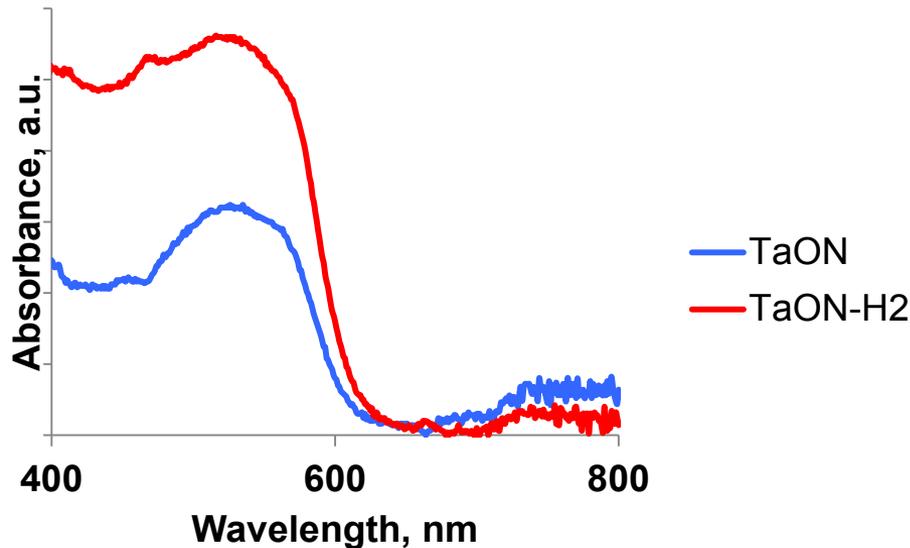
- Electrochemical methods
- Spectroscopic and Electron Microscopic analysis

Improved Visible Light Absorbance of TaON Nanotubes



Ta₂O₅: annealed in air at 500 °C 2h

Ta₂O₅-H: annealed in N₂/H₂
(95/5 mol) 500 °C 2h

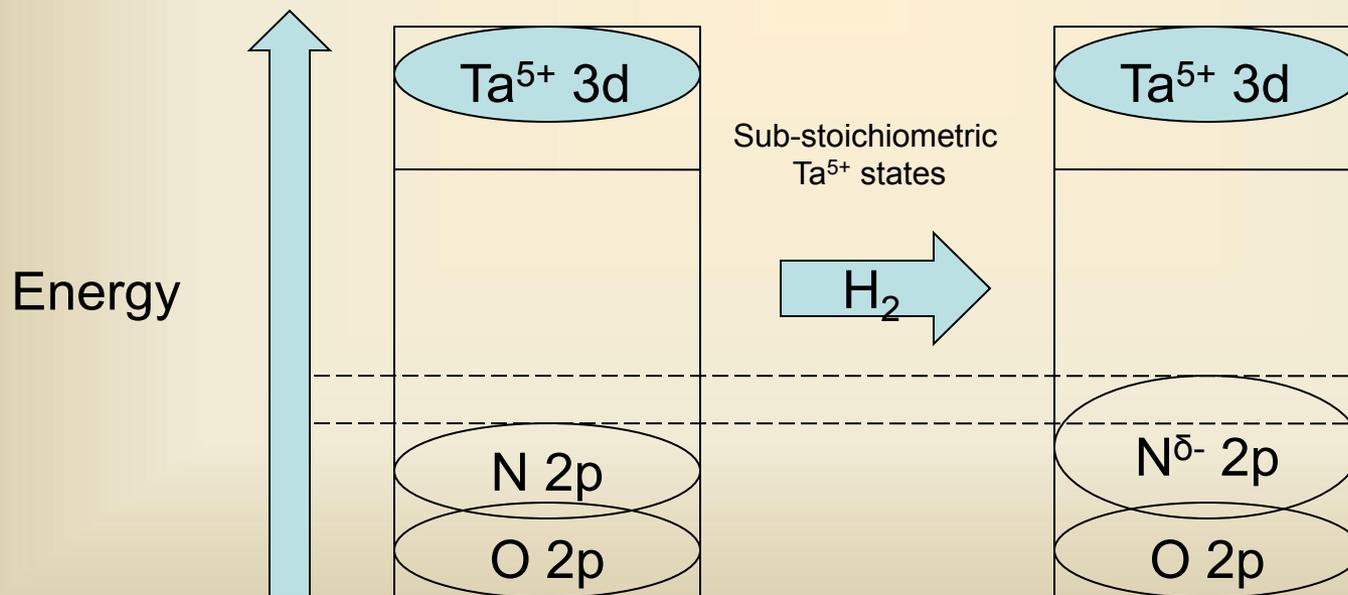


TaON: annealed in NH₃ at 700 °C 2h

TaON-H: Ta₂O₅-H annealed in NH₃
at 700 °C 2h

Improved Visible Light Absorbance of TaON Nanotubes

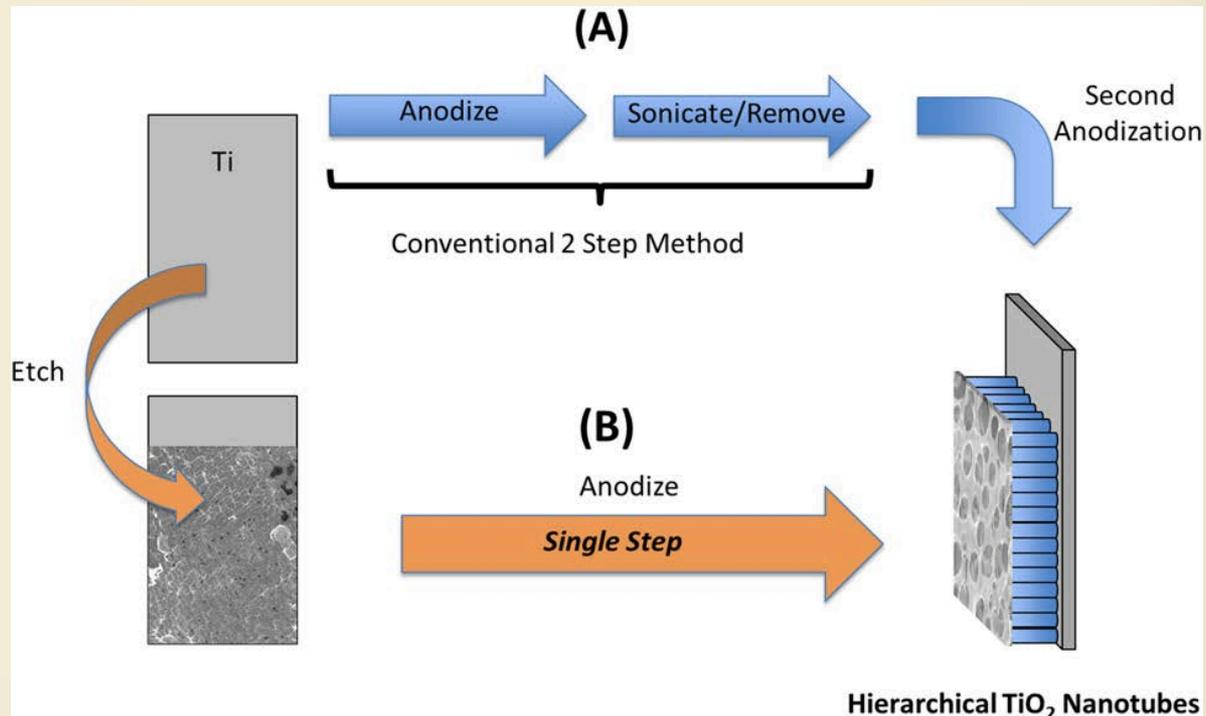
- Reduction in band gap of TaON-H due to N 2p states in valance band
- Improvement in visible light absorption due to synergistic interaction between sub-stoichiometric Ta and $N^{\delta-}$ 2p [2]



New Approach to Synthesize Hierarchical TiO₂ Nanotubes (NTs)

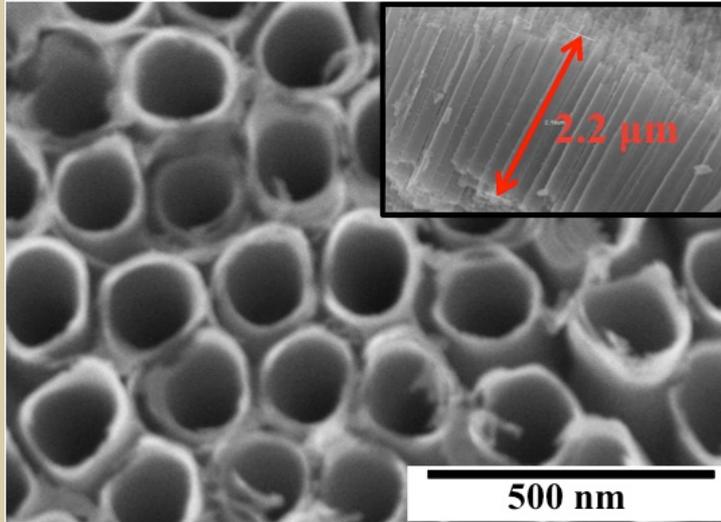
Hierarchical TiO₂ NTs demonstrate higher photoelectrochemical performance over plain TiO₂. Current synthesis of these nanostructures requires a two-step anodization process (A)

Etch solution:
HF (35 %)
HNO₃ (68-70%)
DI H₂O
1:3:50 ml, respectively

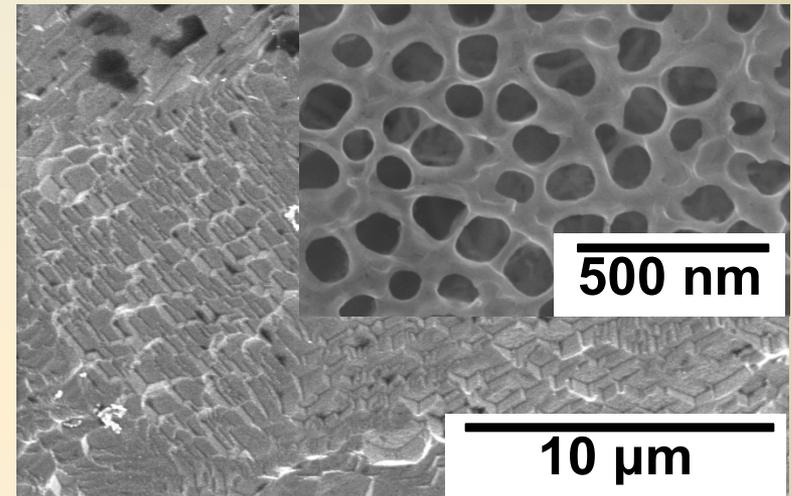


The same top-layered type nanostructure can be obtained by a single anodization process (B) utilizing a surface etching treatment prior to anodization

New Approach to Synthesize Hierarchical TiO₂ Nanotubes (NTs)

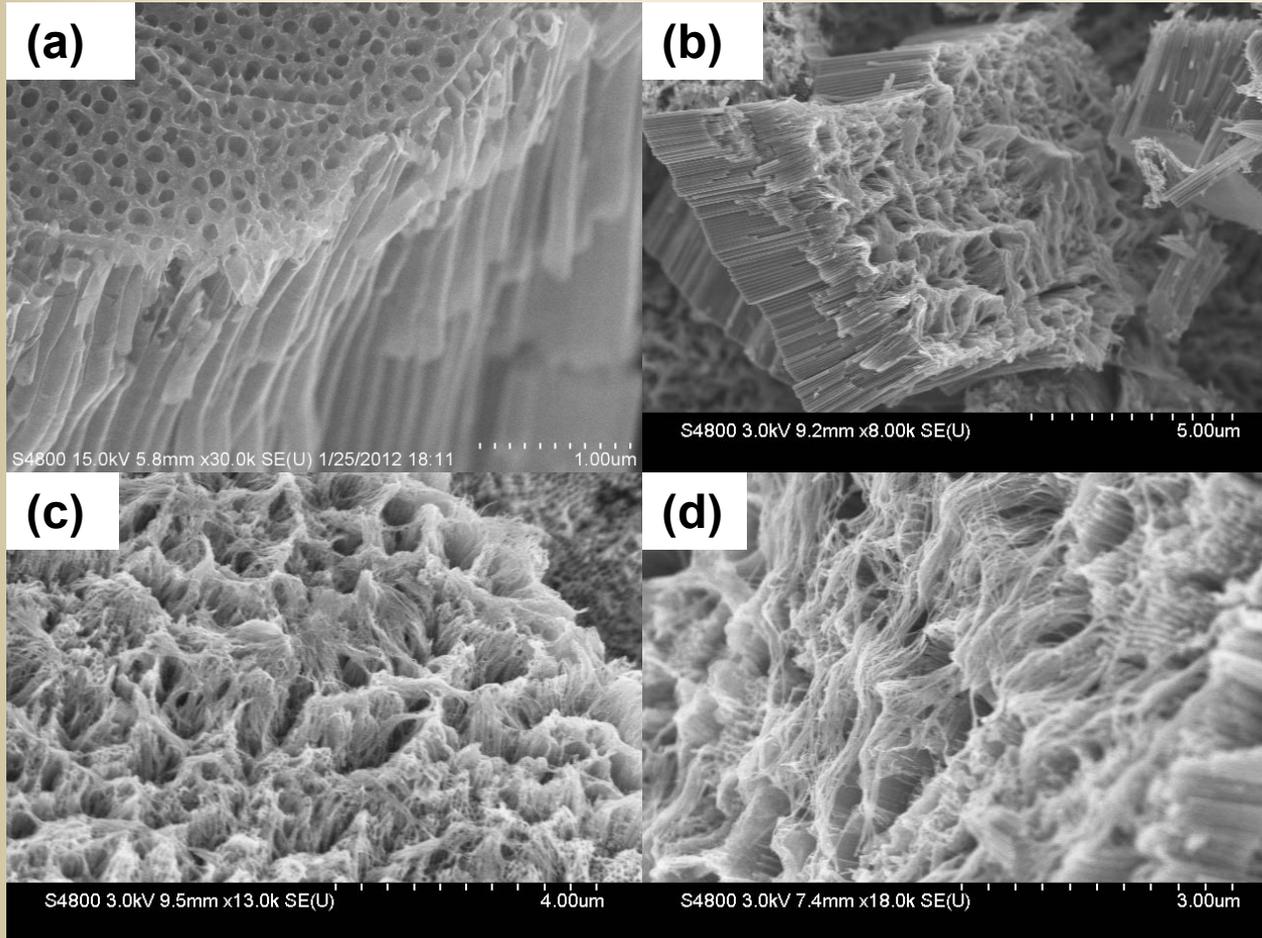


SEM TiO₂ NTs formed at 40 V for 1 h in an ethylene glycol solution (10 wt% H₂O + 0.5 wt% NH₄F without any surface treatment prior to anodization



SEM of Ti surface after etching treatment. The inset shows the surface morphology after anodization at 40 V for 1h. The etching treatment results in the formation of a thin nano-porous layer on top of the TiO₂ nanotubes

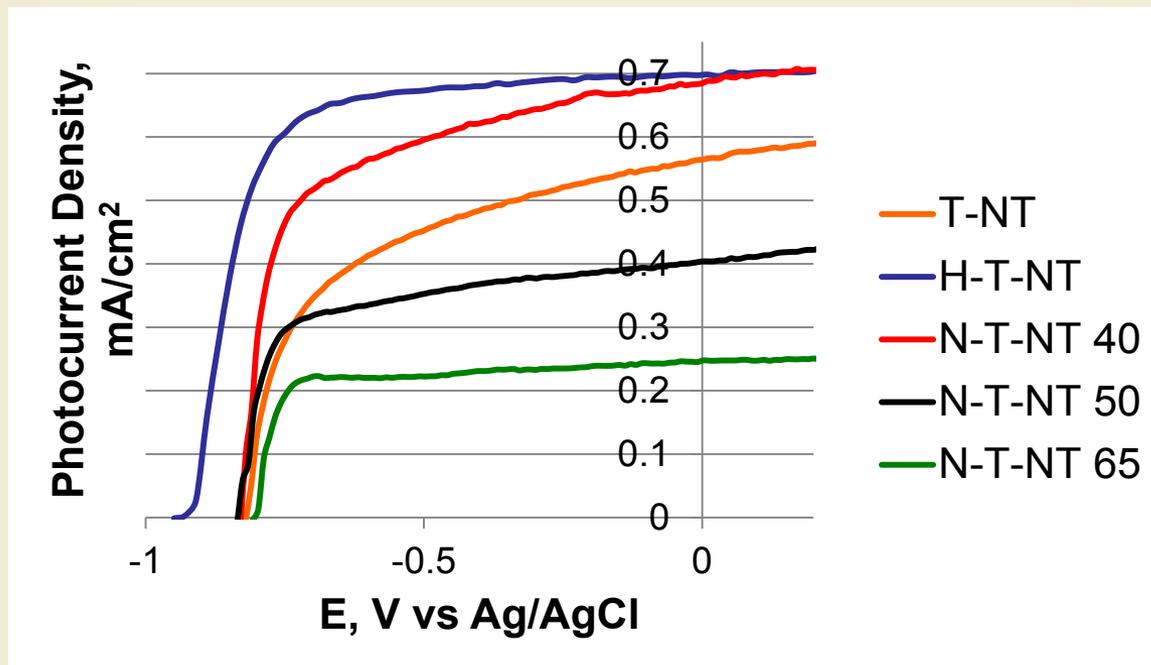
New Approach to Synthesize Hierarchical TiO₂ Nanotubes (NTs)



(a) Hierarchical TiO₂ NT (b-d) TiO₂ NTs with a nanogras top layer. This type of architecture is obtained by increasing the anodization bath temperature. Temperatures of (b) 40 °C, (c) 50 °C, and (d) 65 °C were examined.

Hierarchical, Nangrass and Plain TiO₂ Nanotubes Photoelectrochemical Performance

AM 1.5 irradiation (100 mW cm⁻²) in 1 M KOH with a Pt mesh cathode.

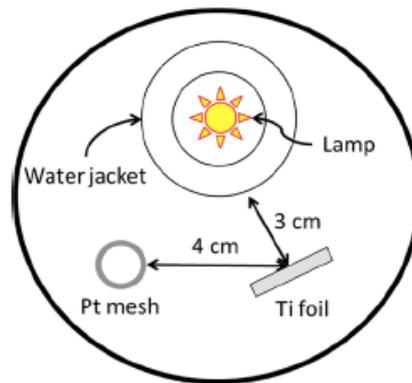


The increased performance of H-T-NT and N-T-NT 40 can be attributed to an increase in surface area for light harvesting and electrolyte interaction sites. The addition of a thin top layer can increase the workfunction over plain TiO₂ NT [3]. Too thick of a layer (eg N-T-NT 50) results in increases electron-hole recombination

Light-Assisted Anodized TiO₂ Nanotubes (NTs) for Enhanced Photoelectrochemical Performance

- Photoelectrochemical etching (PEC-E) has shown to enhance the photoelectrochemical performance of rutile titania [4]
- Can potentially affect nanotube formation morphology

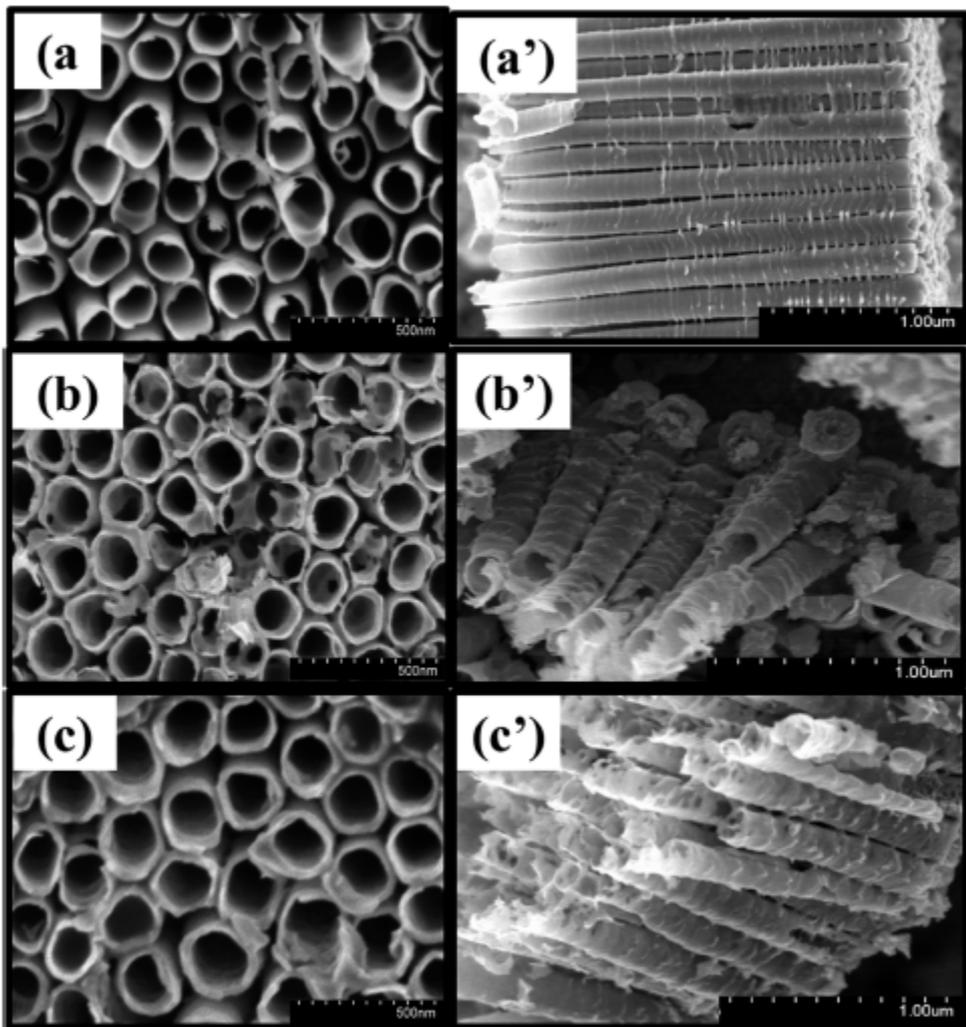
Scheme 1. Depiction of the Top View of the Experimental Setup Used for Anodization in the Presence of UV-Vis Irradiation^a



Light-Assisted Anodized TiO₂ Nanotubes (NTs) for Enhanced Photoelectrochemical Performance

Top View

Side View



Plain TiO₂ NTs (T-NTA)

60 V, 1h

Ethylene glycol electrolyte
(10 wt% H₂O + 0.5 wt% NH₄F)
annealed in N₂/H₂ (95/5 mol) at
500 °C for 2h.

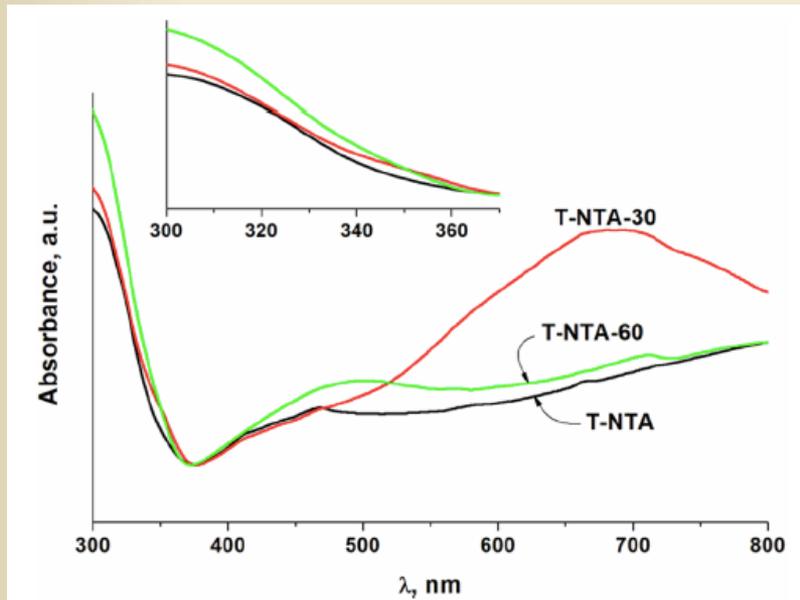
T-NTA-30

Anodized 30 min. without light
Anodized 30 min. with light

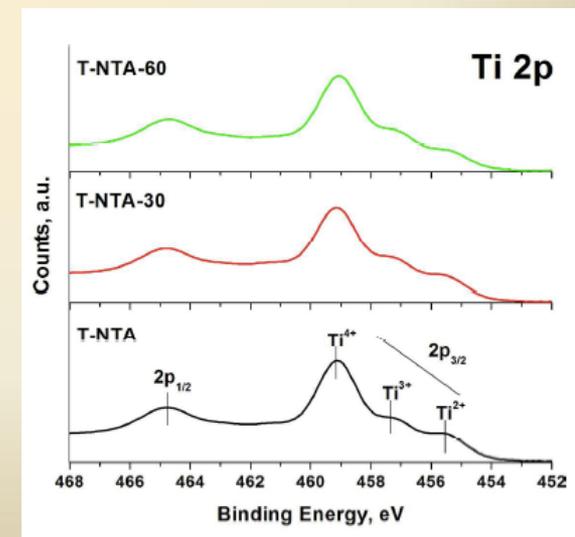
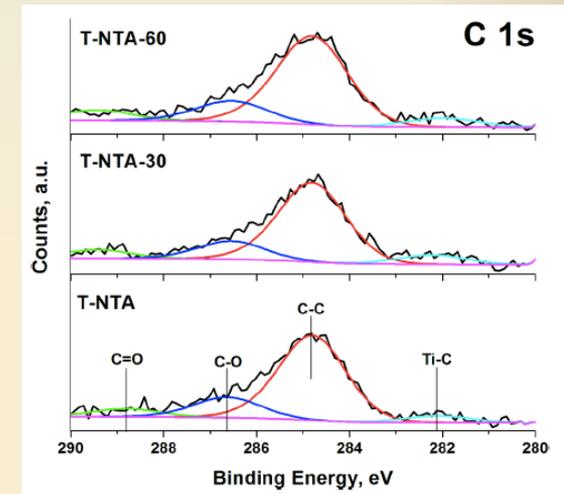
T-NTA-60

Anodized 60 min. with light

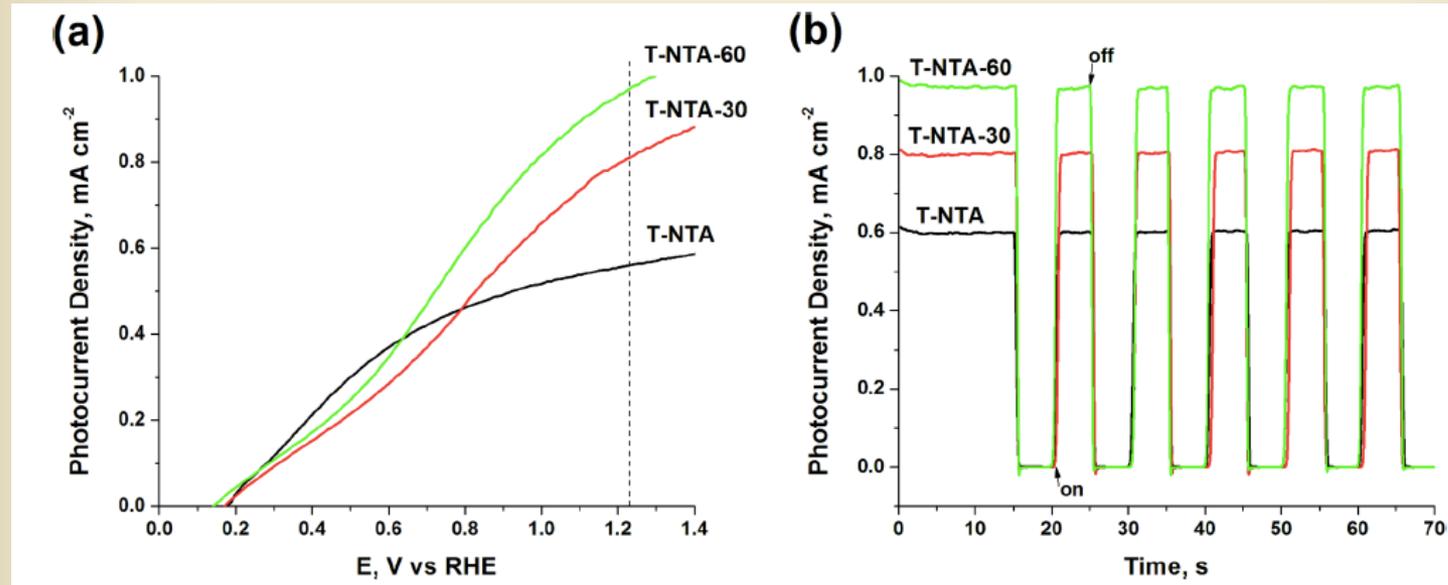
Light-Assisted Anodized TiO₂ Nanotubes (NTs) for Enhanced Photoelectrochemical Performance



- Increase in carbon doping and sub-stoichiometric Ti⁴⁺ when anodized with light
- Increased visible light absorbance



Light-Assisted Anodized TiO₂ Nanotubes (NTs) for Enhanced Photoelectrochemical Performance

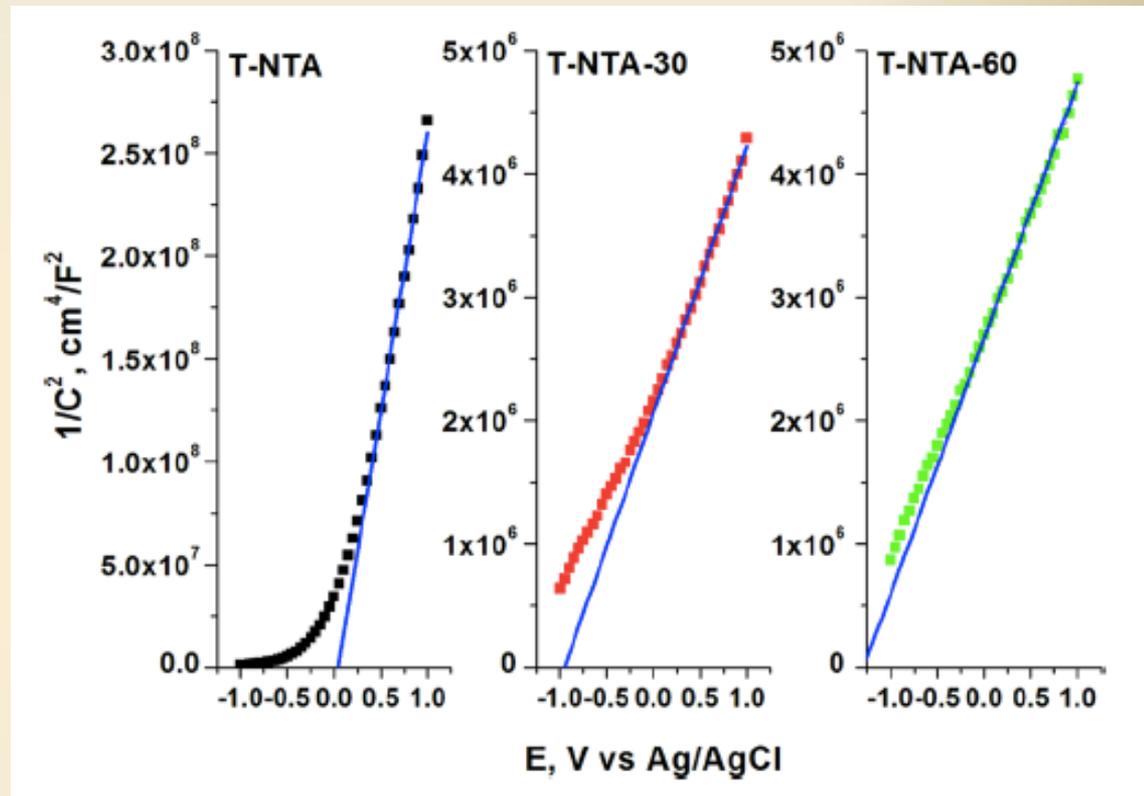


Photoelectrochemical testing: 1.5 AM irradiation (100 mW cm⁻²) in 1 M KOH with a Pt coil cathode and Ag/AgCl (3 M) reference electrode. (A) Potentiodynamic plot and (B) Potentiostatic plot with intermittent irradiation at 1.5 V RHE (0.5 V vs Ag/AgCl)

Light-Assisted Anodized TiO₂ Nanotubes (NTs) for Enhanced Photoelectrochemical Performance

Increase in flat-band potential as well as an increase in charge carrier density from 10^{16} to 10^{18} cm^{-3} account for enhanced photoelectrochemical performance

Greater band bending from thicker NT geometry allows for increased charge separation



Mott-Schottky Analysis: AM 1.5 irradiation at a frequency of 1 kHz with an AC imposed bias of 10 mV in 1 M KOH.

Summary

- **Relevance:** Develop a stable and efficient photoelectrochemical cell for solar hydrogen generation by water splitting
- **Approach:** Synthesize visible light sensitive hybrid nanotube arrays as photoanode material through combinatorial approach
- **Technical accomplishments and process:** Developed a method to extend the visible light absorbance of TaON nanotubes. Developed new synthesis techniques for higher order 1D titania nanotube architectures which demonstrate enhanced photoelectrochemical water splitting.
- **Proposed future research:** (a) Synthesize photoanodes that can harvest the full spectrum of sunlight, (b) theoretical investigation on the materials synthesized (c) scale-up the PEC system, and (d) on-field testing under real solar irradiation.