

# Spirocyclic Anion Exchange Membranes for Improved Performance and Durability

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Project ID: FC178

# Overview

## Timeline and Budget

- Project start date: 01/01/18
- Project end date: 09/30/19
- Total project budget: \$150k with additional \$150K possible
  - Total recipient share: \$0K
  - Total federal share: \$150K
  - Total DOE funds spent\*: \$20K

## Barriers

- Cost
- Performance
- Durability

## Partners

- NREL only project
- Multiple interactions across AEM space, leverage significant effort at NREL on related projects

\* As of 3/31/18

# Relevance/Impact

## DOE (Preliminary) Milestones for AMFCs\*

- **Q2, 2017:** Develop anion-exchange membranes with an area specific resistance  $\leq 0.1 \text{ ohm cm}^2$ , maintained for 500 hours during testing at  $600 \text{ mA/cm}^2$  at  $T > 60 \text{ }^\circ\text{C}$ .
- **Q4, 2017:** Demonstrate alkaline membrane fuel cell peak power performance  $> 600 \text{ mW/cm}^2$  on  $\text{H}_2/\text{O}_2$  (maximum pressure of 1.5 atma) in MEA with a total loading of  $\leq 0.125 \text{ mg}_{\text{PGM}}/\text{cm}^2$ .
- **Q2, 2019:** Demonstrate alkaline membrane fuel cell initial performance of 0.6 V at  $600 \text{ mA/cm}^2$  on  $\text{H}_2/\text{air}$  (maximum pressure of 1.5 atma) in MEA a total loading of  $< 0.1 \text{ mg}_{\text{PGM}}/\text{cm}^2$ , and less than 10% voltage degradation over 2,000 hour hold test at  $600 \text{ mA/cm}^2$  at  $T > 60 \text{ }^\circ\text{C}$ . Cell may be reconditioned during test to remove recoverable performance losses.
- **Q2, 2020:** Develop non-PGM catalysts demonstrating alkaline membrane fuel cell peak power performance  $> 600 \text{ mW/cm}^2$  under hydrogen/air (maximum pressure of 1.5 atma) in PGM-free MEA.

### Impact/Team Project Goals

**Novel Synthesis** - Improve novel perfluoro (PF) anion exchange membrane (AEM) properties and stability.

**Fuel Cell Optimization** - Employ high performance PF AEM materials in electrodes and as membranes in alkaline membrane fuel cells (AMFCs).

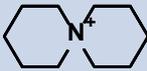
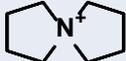
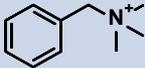
**Model Development** - Apply models to AMFCs to determine and minimize losses (water management, electrocatalysis, and carbonate related).

\*taken from D. Papageorgopoulos presentation AMFC Workshop, Phoenix, AZ, April 1, 2016

# Relevance/Objectives

## Alkaline exchange membranes continue to be challenged with cation degradation at high temperature and pH conditions

- State of the art trimethyl ammonium cations exhibit limited durability under fuel cell operating condition
- Research has indicated that cations with a spirocyclic structure have improved durability
  - Higher activation energy for both Hoffman elimination and substitution degradation mechanisms
- Incorporation of spirocyclic ammonium cations into alkaline exchange membranes to improve durability

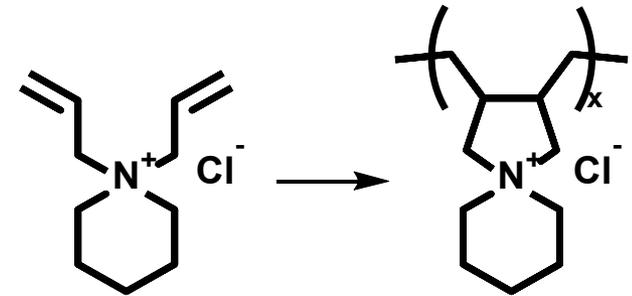
Quaternary Ammonium	Abbreviation	Half-life [hr]
	ASU	110
	DMP	87.3
	DMPy	37.1
	ASN	28.4
	BTMA	4.18

Marino, M. G.; Kreuer, K. D., Alkaline Stability of Quaternary Ammonium Cations for Alkaline Fuel Cell Membranes and Ionic Liquids. *ChemSusChem* **2015**, 8 (3), 513-523.

# Approach

## Synthesis

- Diallyl monomers undergo ring closing radical polymerization
- Polymerization of diallylpiperidinium chloride produces polymer of ASU/ASN hybrid structures



## Polymer & copolymer characterization

- Structure
- IEC
- Conductivity

## Accelerated aging

- Polymer & AEM durability
- Degradation pathways and rates

## MEA fabrication and characterization

- Fuel cell performance
- Long term durability

Leverage NRELs in-house expertise and MEA testing equipment

- Previous work generated multiblock copolymers of polydiallylpiperidinium segments in a high performance polysulfone backbone
- Current synthesis focuses on scaling synthetic procedure for production of larger (>20 g) batches
  - Provide ample material for complete MEA characterization and durability studies

# Approach - Milestones

Milestone Name/Description	End Date	Type
Produce sufficient materials (> 20 g) to accomplish degradation evaluation of both the homopolymer and multiblock copolymer membrane.	12/31/2017	Progress Measure
Quantify poly(polydiallylpiperidinium hydroxide) degradation rates using temperature as an accelerating factor up to 160 °C.	3/31/2018	Progress Measure
Demonstrate membrane ASR $\leq 0.02 \Omega$ in fuel cell tests.	6/30/2018	Progress Measure
Demonstrate AEM fuel cell initial performance of 0.6 V at 600 mA/cm <sup>2</sup> on H <sub>2</sub> /air (maximum pressure of 1.5 atma) in MEA, and less than 10% voltage degradation over 1,000 hour hold test at 600 mA/cm <sup>2</sup> at T>60 °C.	9/30/2018	Annual Milestone

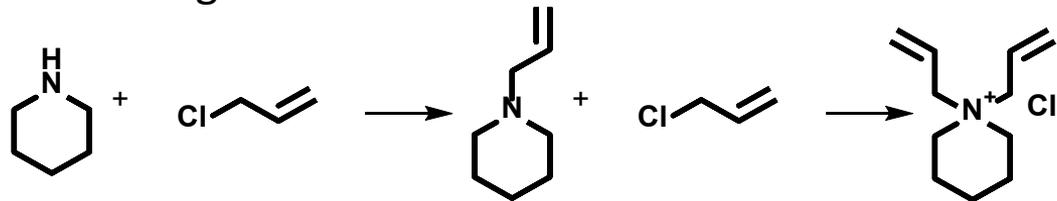
Go/No-Go Description	Criteria	End Date
AEMFC Durability	In alignment with DOE 2019, Q2 AEM target, demonstrate AEM fuel cell initial performance of 0.6 V at 600 mA/cm <sup>2</sup> on H <sub>2</sub> /air (maximum pressure of 1.5 atma) in MEA, and less than 10% voltage degradation over 1,000 hour hold test at 600 mA/cm <sup>2</sup> at T>60 °C.	9/30/2018

# Accomplishments and Progress

## Synthesis

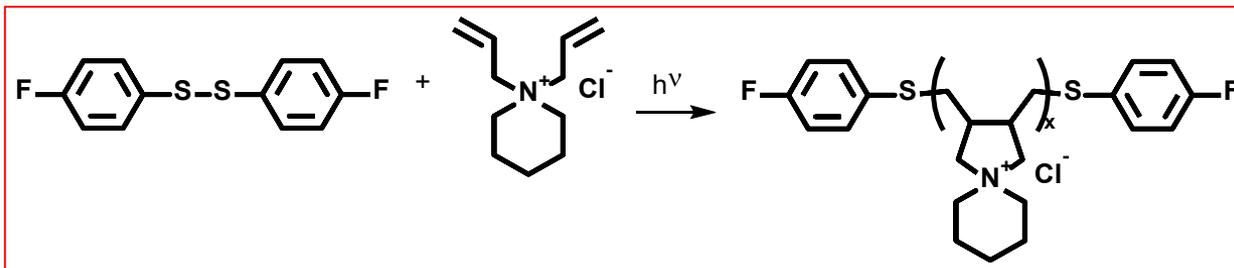
### Monomer

- Two step process completed in 4-5 days
- Easily produce > 100 g batches



### Poly(diallylpiperidinium chloride) (PDApip)

- 24 hr. photopolymerization
- Small batch ( $\approx$  5 g) 50 % recovery
- Larger batch (30-40 g) 15 % recovery
- **bottleneck for larger scale copolymerization**

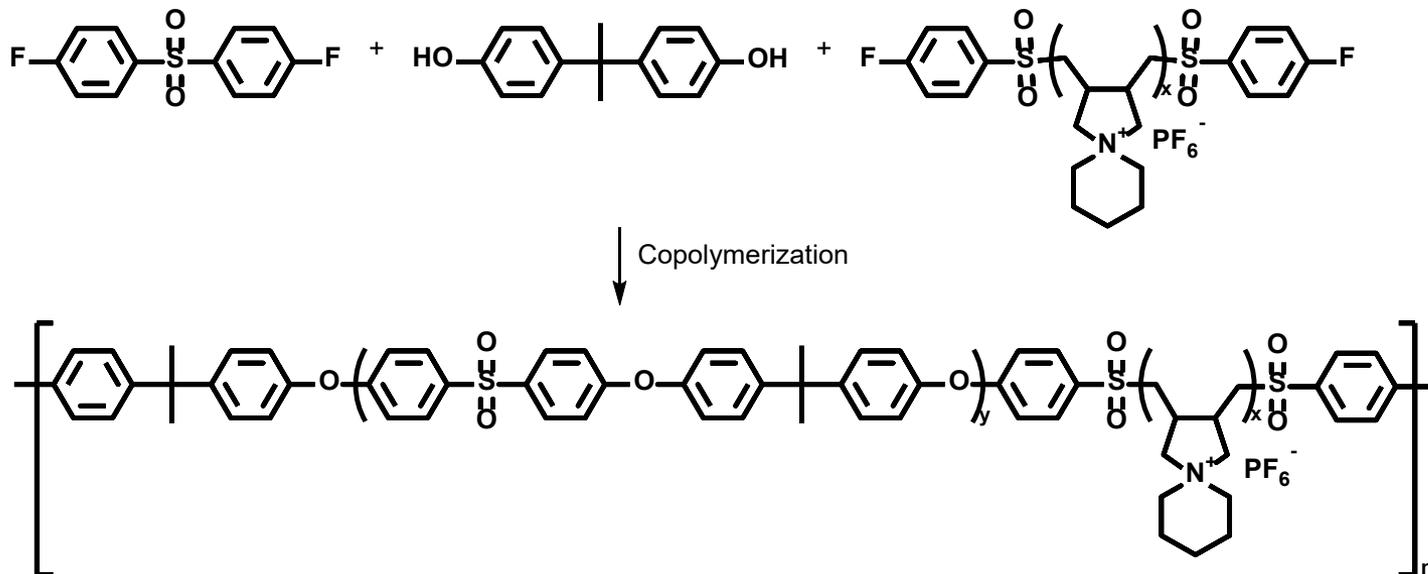


# Accomplishments and Progress

## Synthesis

### Polysulfone-PDApip multiblock copolymers

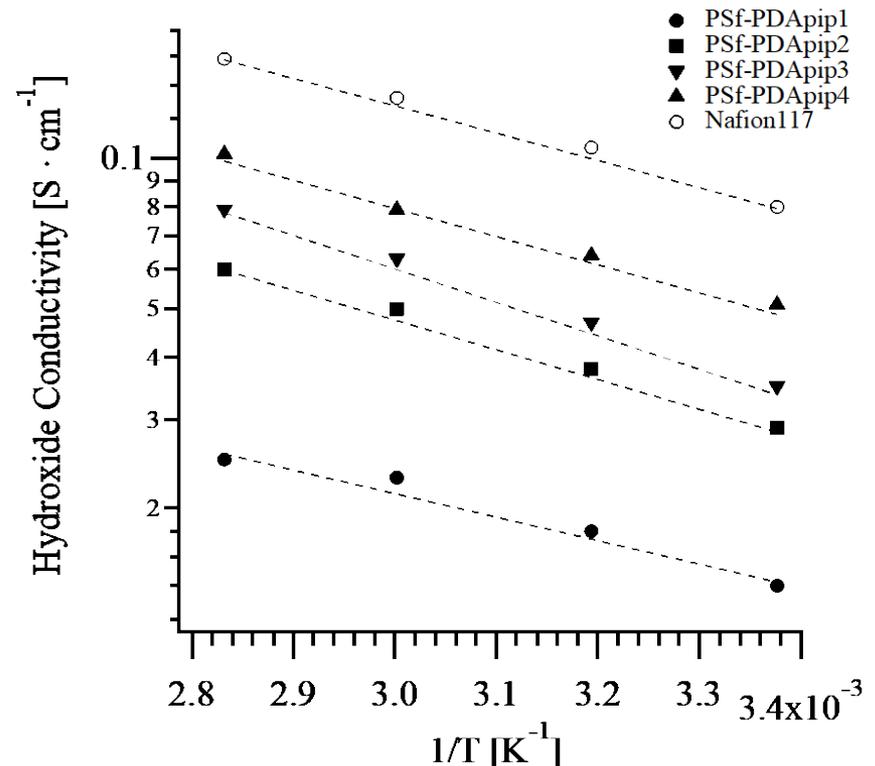
- A single 16 g batch has been copolymerization completed
- Copolymerization had low viscosity – indicating low molecular weight
- $^1\text{H}$  NMR estimation of PDApip end group concentration was insufficient
  - Indicates the presence of non-functional end groups



# Accomplishments and Progress

## Polymer & Copolymer characterization

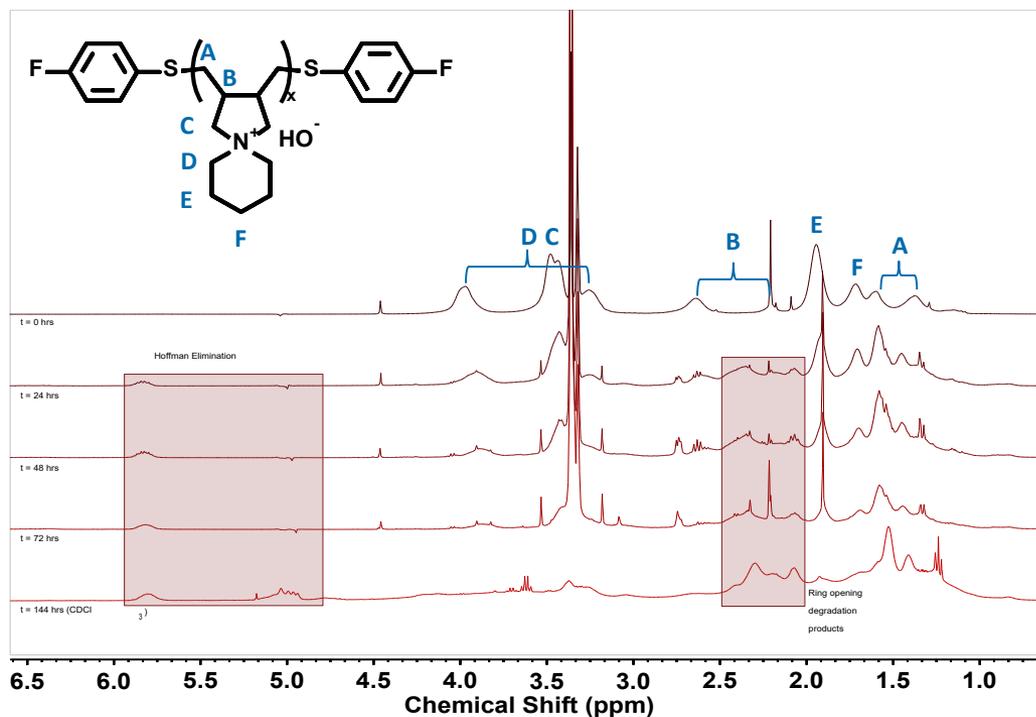
- Previous work has shown the Polysulfone-PDApip multiblock copolymer membranes to efficiently conduct hydroxide
- The activation energy for hydroxide conduction was very similar to Nafion117
- Hydroxide conductivity was able to reach 102 mS/cm<sup>2</sup> at 80 °C



# Accomplishments and Progress

## Polymer Durability

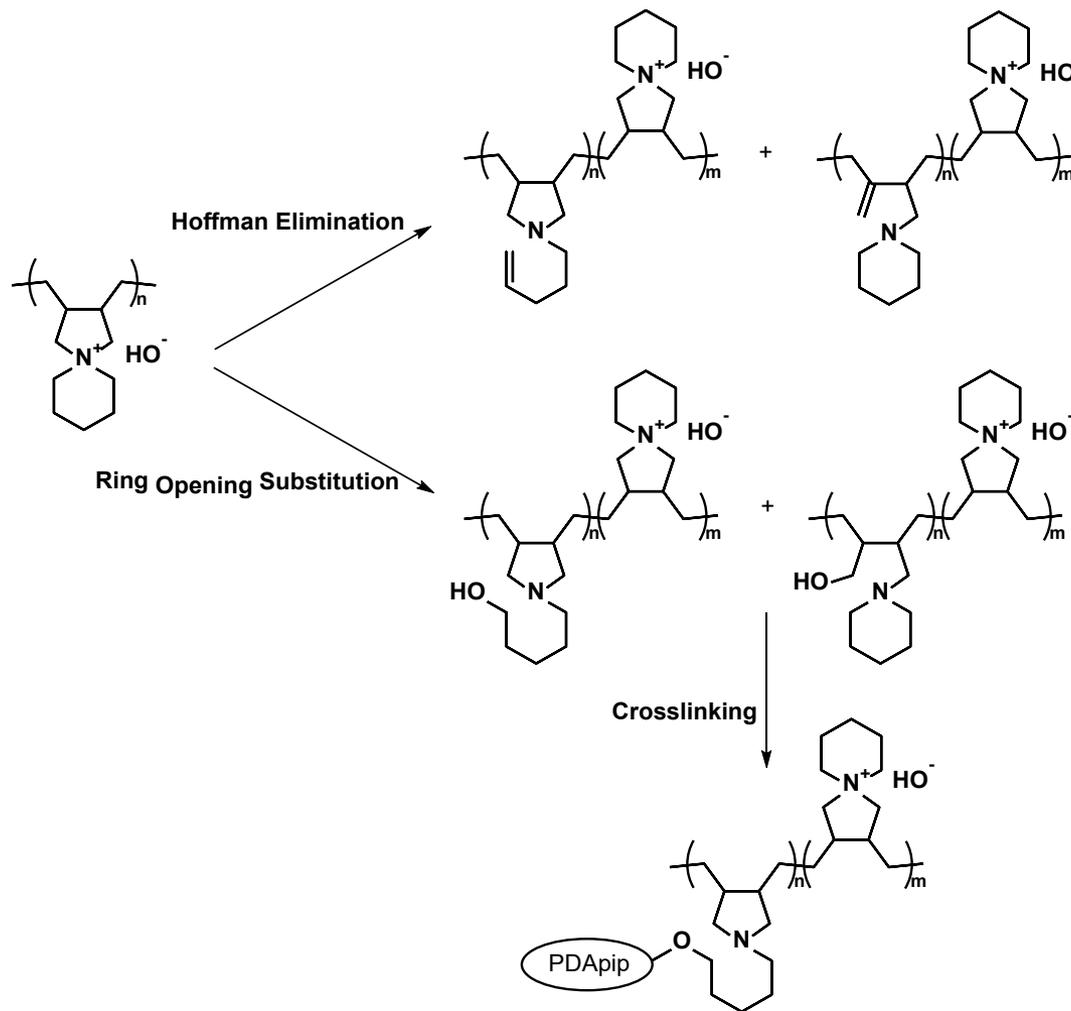
- Previous investigation of the PDApip durability indicated negligible degradation, by  $^1\text{H}$  NMR, over 1000 hours at  $80\text{ }^\circ\text{C}$  in a 1M KOH/Methanol- $\text{d}_4$  solution
- Current durability studies adapt NRELs established accelerated aging conditions
- PDApip aged at  $140\text{ }^\circ\text{C}$  for 144 hours in 2:1 MeOH/2M KOH
- Hoffman elimination and ring opening degradation was observed over 144 hours



# Accomplishments and Progress

## Degradation Pathways

- After 144 hours some insoluble material was left behind in the reactor
- Insolubility indicates crosslinking resulting from ring opening attack of hydroxyl degradation products



# Remaining Challenges and Barriers

## Synthesis

- Large scale PDApip synthesis has poor yield
  - Limits the copolymerization scale

## Accelerated aging characterization

- $^1\text{H}$  NMR unable to quantitatively assess the amount of degradation in PDApip polymers
  - Developing titration analytical method

# Proposed Future Work

## **Synthesis**

- Continue to work on the scale up of the PDApip polymerization
  - improve conversion
- Generate ~ 10 – 15 g batches of PDApip
- Produce ~ 20 g batches of Polysulfone-PDApip copolymer

## **Polymer and copolymer characterization**

- Molecular weight analysis of produced PDApip polymers
- UV characterization of end groups – improve copolymerization
- Confirm IEC and measure ionic conductivity of Polysulfone-PDApip membranes

## **Accelerated aging**

- Continue accelerated aging experiments to further elucidate the rate of degradation and major degradation pathways

## **MEA fabrication and characterization**

- Fabricate MEAs and optimize fuel cell test conditions
- Conduct long term durability study with Polysulfone-PDApip materials

*Any proposed future work is subject to change based on funding levels*

# Summary

Milestone Name/Description	End Date	Type	Progress
Produce sufficient materials (> 20 g) to accomplish degradation evaluation of both the homopolymer and multiblock copolymer membrane.	12/31/2017	Progress Measure	Can produce > 15 g batches of multiblock membrane material
Quantify poly(polydiallylpiperidinium hydroxide) degradation rates using temperature as an accelerating factor up to 160 °C.	3/31/2018	Progress Measure	Shown accelerated degradation and pathways <ul style="list-style-type: none"> <li>• Need rates of degradation</li> <li>• Further elucidation of pathways</li> </ul>
Demonstrate membrane ASR $\leq 0.02 \Omega$ in fuel cell tests.	6/30/2018	Progress Measure	Not started
Demonstrate AEM fuel cell initial performance of 0.6 V at 600 mA/cm <sup>2</sup> on H <sub>2</sub> /air (maximum pressure of 1.5 atma) in MEA, and less than 10% voltage degradation over 1,000 hour hold test at 600 mA/cm <sup>2</sup> at T>60 °C.	9/30/2018	Annual Milestone	Not started

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