



# HydroGEN Overview: A Consortium on Advanced Water Splitting Materials

H.N. Dinh, D. Anton, R. Boardman, A. McDaniel,  
T. Ogitsu, A. Weber

Presenter: Huyen Dinh, NREL

Date: 4/30/2019

Venue: 2019 DOE Annual Merit Review

Project ID # P148

This presentation does not contain any proprietary, confidential, or otherwise restricted information.



# HydroGEN Overview

## Timeline and Budget

- Start date (launch): **June 2016**
- FY17 DOE funding: **\$3.6M**
- FY18 DOE funding: **\$9.9M**
- FY19 planned DOE funding: **\$6.5M**
- Total DOE funding received to date: **\$22M**

## Barriers

- **Cost**
- **Efficiency**
- **Durability**

## Partners





# Collaboration: HydroGEN Steering Committee



**Huyen Dinh**  
(Director)



**Adam Weber**  
(Deputy Director)



**Anthony McDaniel**  
(Deputy Director)



**Richard Boardman**



**Tadashi Ogitsu**



**Donald Anton**



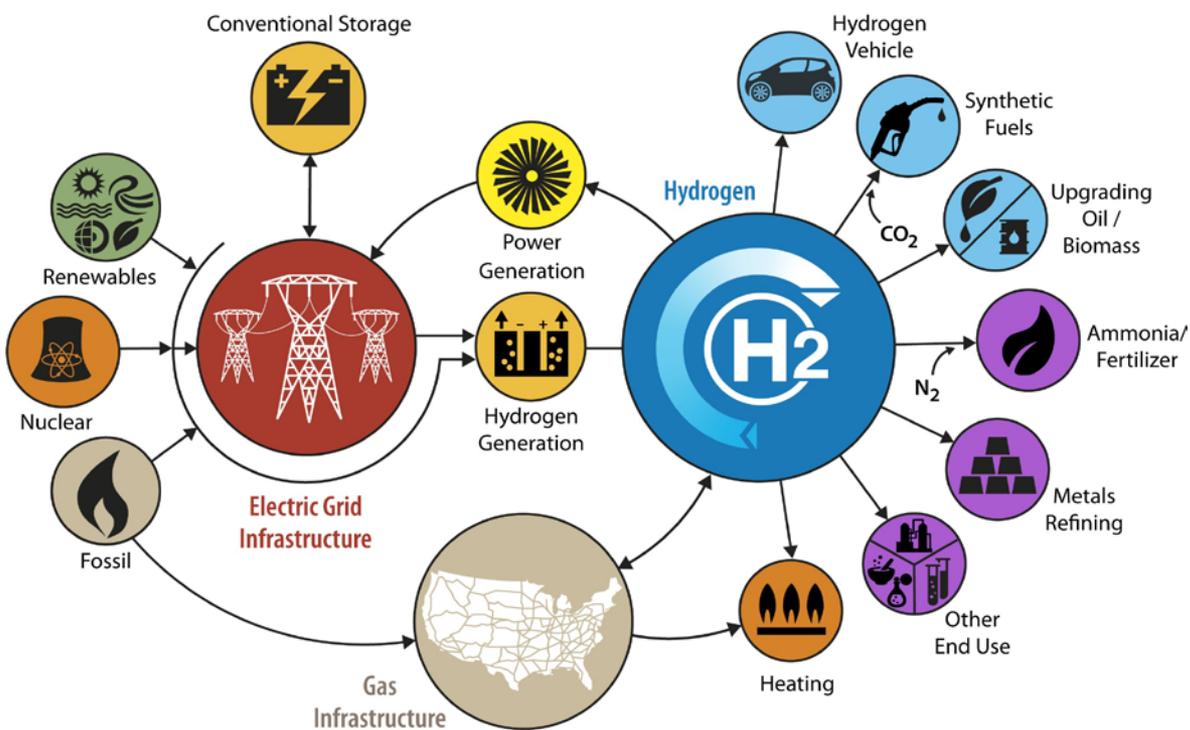
**Eric Miller and Katie Randolph, DOE-EERE-FCTO**



# H2@Scale Energy System Vision

## Relevance and Impact

### *Transportation & Beyond*



\*Illustrative example, not comprehensive

Large-scale, low-cost hydrogen from diverse domestic resources enables an economically competitive and environmentally beneficial future energy system across sectors

Materials innovations are key to enhancing performance, durability, and cost of hydrogen generation, storage, distribution, and utilization technologies key to H2@Scale

<https://energy.gov/eere/fuelcells/h2-scale>



# Energy Materials Network (EMN)

## Relevance and Impact



DOE's EMN aims to accelerate early-stage applied R&D in materials tracks aligned with some of the nation's most pressing sustainable energy challenges

Hydrogen Compatible Materials

Breakthrough Hydrogen Storage Materials

Advanced Water Splitting Materials for Hydrogen Production

Next-Generation Electro-catalysts for Fuel Cells

*example tracks*



DISCOVERY > DEVELOPMENT > OPTIMIZATION > SYSTEM INTEGRATION > CERTIFICATION

COMMERCIAL SCALE-UP AND MANUFACTURING

PROPELLING CLEAN ENERGY MATERIALS DEVELOPMENT FORWARD

*Accelerating early-stage materials R&D for energy applications*



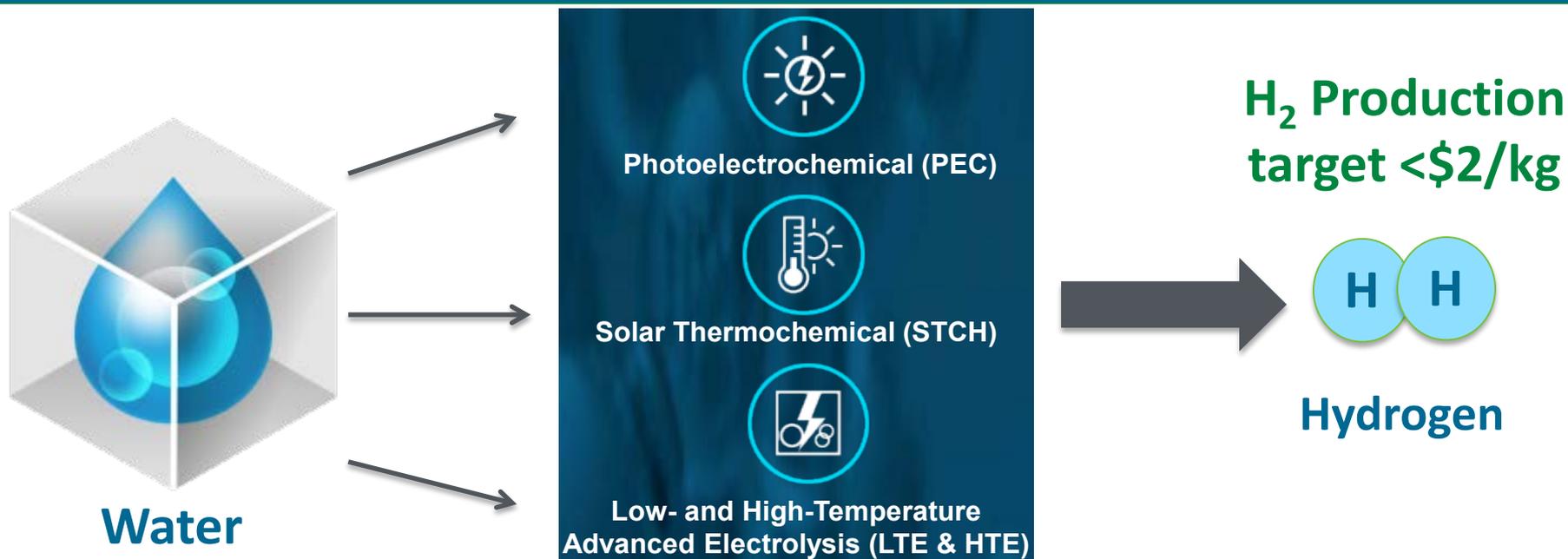
# Advanced Water-Splitting Materials (AWSM)

## Relevance, Overall Objective, Impact, and Approach

**AWSM Consortium**  
**Six Core Labs:**



**Accelerating R&D** of innovative materials critical to advanced water splitting technologies for clean, sustainable, and low cost H<sub>2</sub> production, including:



**HydroGEN consortium supports early stage R&D in H<sub>2</sub> production**

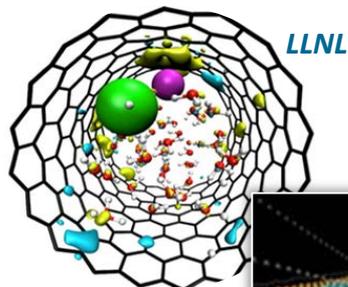


# HydroGEN-AWSM Consortium

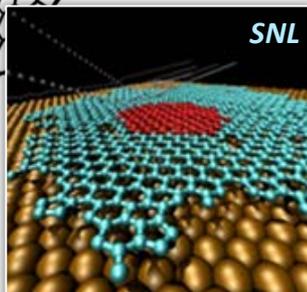
## Relevance, Overall Objective, Impact, and Approach

Comprising more than 80 unique, world-class capabilities/expertise in:

### Materials Theory/Computation

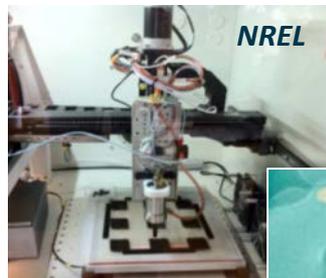


Bulk & interfacial models of aqueous electrolytes

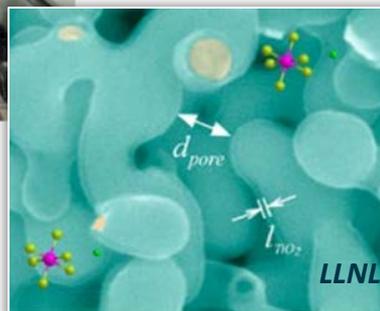


LAMMPS classic molecular dynamics modeling relevant to  $H_2O$  splitting

### Advanced Materials Synthesis



High-throughput spray system for electrode fabrication



Conformal ultrathin  $TiO_2$  ALD coating on bulk nanoporous gold

### Characterization & Analytics



Stagnation flow reactor to evaluate kinetics of redox material at high-T



TAP reactor for extracting quantitative kinetic data

**HydroGEN fosters cross-cutting innovation using theory-guided applied materials R&D to advance all emerging water-splitting pathways for hydrogen production**

Website: <https://www.h2awsm.org/>



# Accomplishments: Updated Capability Nodes on the User-Friendly Node Search Engine for Stakeholders

## Added 3 new and updated >40 current capability nodes:

- Hybrid Organic Inorganic Perovskites for Water Splitting
- Understanding catalyst inks and ionomer dispersions
- Electronic-Structure Modeling for Atomistic Understanding of Catalytic Materials with Real-World Distributions of Facets and Defects

Considered capability assessment from benchmarking team

Search

[Reset filtering](#)

**CAPABILITY CLASS**

- Analysis
- Benchmarking
- Characterization
- Computational Tools and Modeling
- Data Management
- Material Synthesis
- Process and Manufacturing Scale-Up
- System Integration

**WATER-SPLITTING TECHNOLOGY**

- High-Temperature Electrolysis
  - HTE 1  HTE 2  HTE 3
- Low-Temperature Electrolysis
  - LTE 1  LTE 2  LTE 3
- Photoelectrochemical
  - PEC 1  PEC 2  PEC 3
- Solar Thermochemical
  - STCH 1  STCH 2
  - STCH 3
- Hybrid Thermochemical
  - HT 1  HT 2  HT 3

[Node Readiness Categories](#)

**NATIONAL LABORATORY**

- Idaho National Laboratory (INL)
- Lawrence Berkeley National Laboratory (LBNL)
- Lawrence Livermore National Laboratory (LLNL)
- National Renewable Energy Laboratory (NREL)
- Sandia National Laboratories (SNL)
- Savannah River National Laboratory (SRNL)



## Node Readiness Category (NRC) Chart



Node is **fully developed** and has been used for AWSM research projects

Node requires **some development** for AWSM

Node requires **significant development** for AWSM

Showing 1 to 12 of 82 entries

1 2 3 4 Next

Ab Initio Modeling of Electrochemical Interfaces

LLNL      PEC 1, LTE 2

Advanced Electron Microscopy

SNL      HTE 1, LTE 1, PEC 1, STCH 1

Advanced Materials for Water Electrolysis at Elevated Temperatures

INL      HTE 2

Show

12

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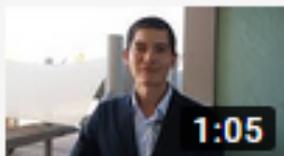
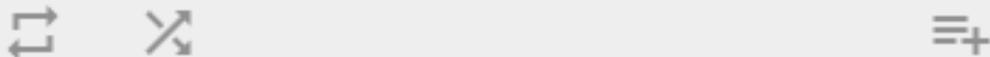
*Annual capability review is a rigorous process and keeps nodes updated and relevant*



# Accomplishments: Maintained HydroGEN Website and Developed “Working with HydroGEN” Video Testimonials

## HydroGEN Advanced Water Splitting Materials

National Renewable Energy Laboratory - NREL - 1 / 4



**Working With HydroGEN: Tom Jaramillo, Stanford University**

National Renewable Energy Laboratory - N...



**Working with HydroGEN: Chris Capuano, Proton OnSite**

National Renewable Energy Laboratory - N...



**Working with HydroGEN: Samantha Millican, University of Colorado Boulder**

National Renewable Energy Laboratory - N...



**Working with HydroGEN: Nemanja Danilovic, Lawrence Berkeley National**

National Renewable Energy Laboratory - N...

**Partner testimonial videos can be found here:**

<https://www.youtube.com/watch?v=1eK8ZnVo2Y&list=PL3GM1pirYAchur4Aq1pojTC2cMPf8xs6P>



## WORKING WITH HYDROGEN: WHAT OUR PARTNERS SAY



Tom Jaramillo, principal investigator on a HydroGEN seedling project, talks about how the capabilities and expertise of the HydroGEN Advanced Water Splitting Materials consortium can help researchers working on hydrogen technologies.

**4,373 users**  
**24,011 pageviews**  
**564 downloads**

**Visitors:**  
**85% new**  
**15% returning**

**Traffic:**  
**46% search**  
**37% direct**  
**17% referral**

**Posted**  
**10 news**  
**items**

**Added “Work**  
**with Us” page**  
**to top menu**



# Accomplishments: HydroGEN Data Hub: Making Digital Data Accessible

<https://datahub.h2awasm.org/>

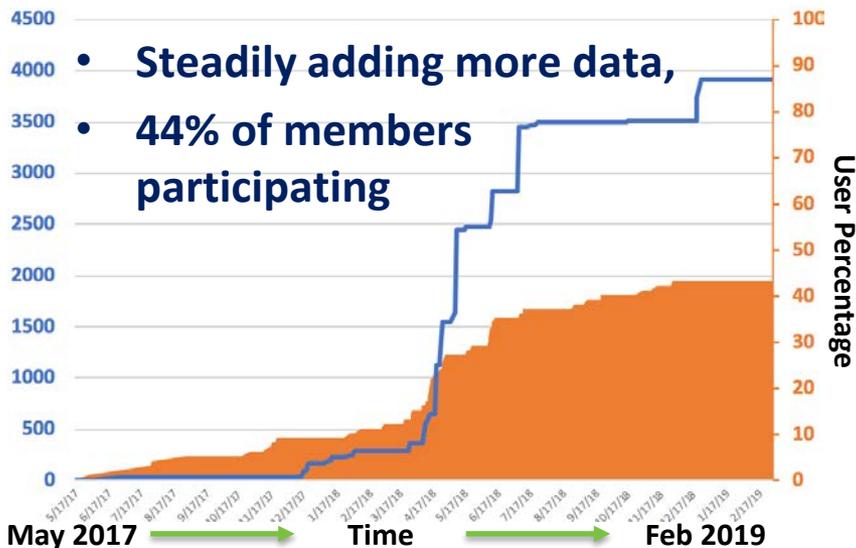
179  
Users



4055  
Files



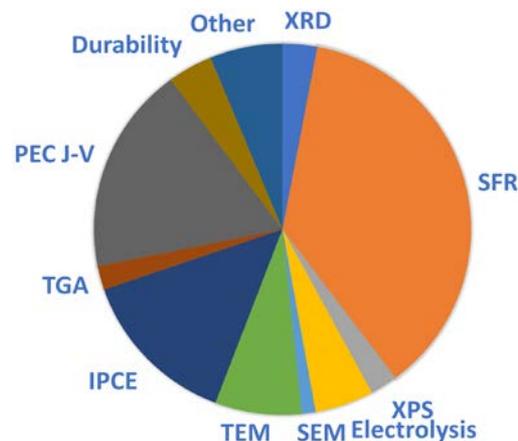
## Cumulative Data on H2AWSM Data Hub



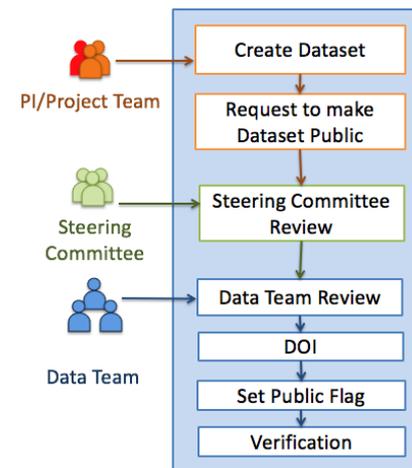
## Data Hub implemented in May 2017

- Secure project space for team members
  - Centralized authentication
- Data publication process developed (FY2018Q3 QPM-see Reviewer-Only slides)
- Metadata tools and improvements to support advanced search

## Types of Experimental Data



## Data Publication Process



## Data Team

Other = Raman spectroscopy, rheology, helium ion microscope images, conductivity, dilatometry, kinetic, XRF





# Accomplishments: Data Tools for Data Ingestion, Visualization and Analysis (Leveraging other EMNs)

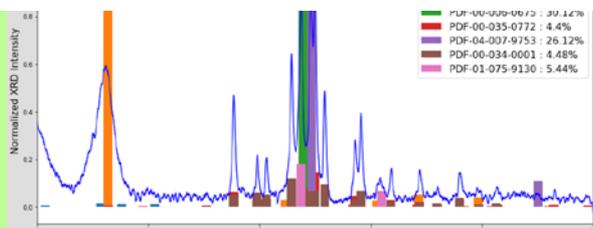
5 currently available, 5 under development

## Simple data interface developed

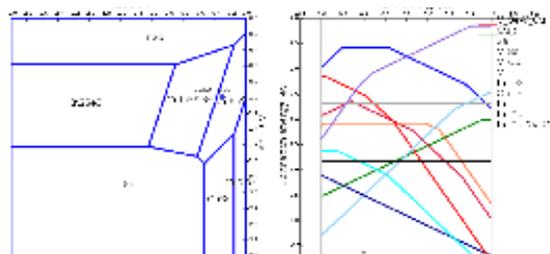


## Materials properties (A&V examples)

### Structural information: XRD interface in collaboration with ElectroCat



### Phase stability & Defect properties

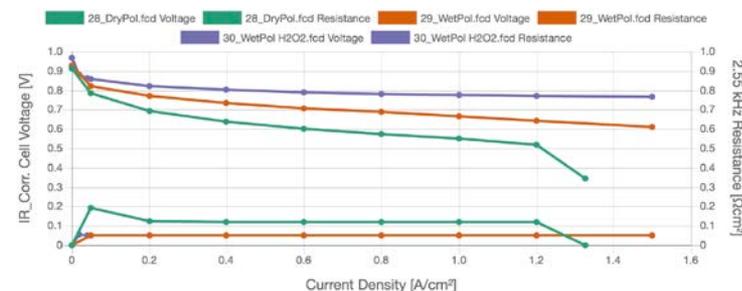


## Data Tools for

- EMN collaboration
  - Data exchange and exploration
  - Data analysis and visualization (A&V)
- External users
  - Access to comprehensive database
    - Experimental and computational
  - Materials properties
    - Spectroscopy, phase stability, etc.
  - Device performance

## Device performance (A&V example)

### Electrolysis Performance curve

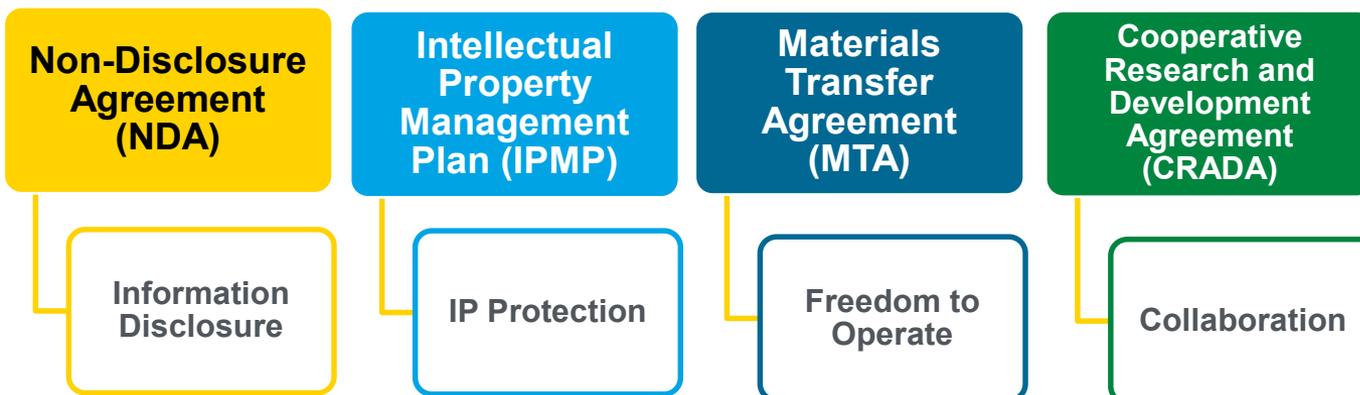




# Accomplishments: Technology Transfer Agreements (TT/A)

## ➤ Streamlined Access

- ✓ Four standard, pre-approved TT/A between all consortium partners
- ✓ Executed all 21 project NDAs and 2 MTAs
- ✓ Updated multiple, more flexible CRADAs
  - Single Lab Single Participant
  - Multi-Lab Single Participant
  - Multi-Lab Multi-Participant
- ✓ Established Work For Others Agreements outside of FOA-awarded projects:
  - DeNora : GDE development on porous transport layers (LBNL Cell Testing Node)
  - German Aerospace Center: Concentrated solar power and solar fuels research (SNL)



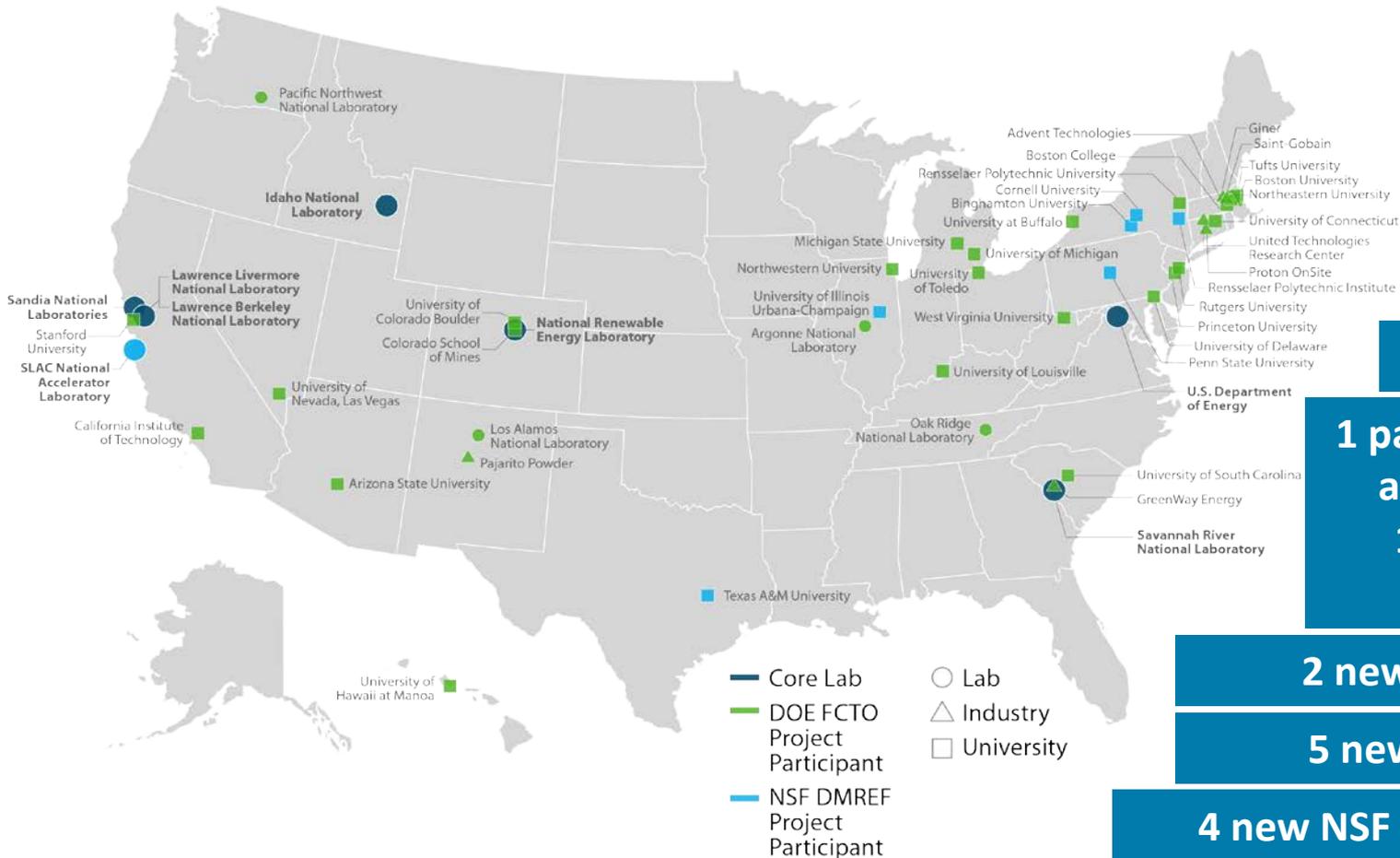
<https://www.h2aws.org/working-with-hydrogen>





# National Innovation Ecosystem Collaboration/Accomplishments

11 Labs 7 Companies 30 Universities 2 Funding Agencies



**A Community  
Benchmarking  
Workshop**

**2 MRS Symposia**

**17 papers  
published**

**88 presentations**

**1 patent, 2 provisional  
applications filed;  
1 SLAC proposal  
accepted**

**2 new HTE projects**

**5 new Supernodes**

**4 new NSF DMREF projects**

***HydroGEN is vastly collaborative, has produced many high value products, and is disseminating them to the R&D community.***



# 4 New NSF DMREF/DOE EERE HydroGEN EMN Projects Collaboration/Accomplishments

## NSF DMREF PSU LTE (IA023 Poster)



### Membrane Databases – New Schema and Dissemination

**Recipient** Penn State University (PI: Michael A. Hickner)

**Subs** National Institute of Standards and Technology/NIST (PI: Debra Audus) and Rensselaer Polytechnic Institute/RPI (PI: Chulsung Bae)

#### HydroGEN Node Experts

National Renewable Energy Laboratory:

- Shaun Alia
- Guido Bender
- Kristin Munch
- Bryan Pivovarov

## NSF DMREF CSM STCH (IA022 Poster)



### High Temperature Defects: Linking Solar Thermochemical and Thermoelectric Materials

**Recipient** Colorado School of Mines (PI: Eric Toberer and Vladan Stevanovic)

**Subs** University of Illinois Urbana-Champaign (PI: Elif Ertikin) and SLAC National Accelerator Laboratory (PI: Michael Toney)

#### HydroGEN Node Experts

National Renewable Energy Laboratory:

- Robert Bell
- David Ginley
- Stephan Lany
- Philip Parilla

## NSF DMREF PSU PEC (IA024 Poster)



### Experimental Validation of Designed Photocatalysts For Solar Water Splitting

**Recipient** Penn State University (PI: Ismaila Dabo and Raymond E. Schaak)

**Subs** Cornell University (PI: Héctor D. Abruña)

#### HydroGEN Node Experts

National Renewable Energy Laboratory:

- Todd Deutsch
- James Young

## NSF DMREF UB PEC (IA021 Poster)



### Collaborative Research: A Blueprint for Photocatalytic Water Splitting: Mapping Multidimensional Compositional Space to Simultaneously Optimize Thermodynamics and Kinetics

**Recipient** University at Buffalo (PI: David Watson)

**Subs** Texas A&M University (PI: Sarbajit Banerjee) and Binghamton University (PI: Louis Piper)

#### HydroGEN Node Experts

Lawrence Berkeley National Laboratory

- Jinghua Guo
- David Prendergast



# Collaboration: HydroGEN FOA-Awarded Projects

16 projects passed GNGs and in phase 2 + 2 new HTE projects awarded  
49 unique capabilities being utilized across six core labs

**Advanced Electrolysis (10)**

**LTE (5)**

**HTE (5)**

**PEC (5)**

**Benchmarking &  
Protocols (1)**

**STCH (5)**

**2-Step MO<sub>x</sub> (4)  
Hybrid Cycle (1)**



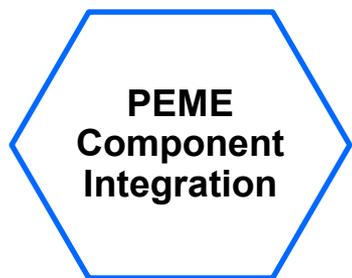


# A Balanced AWSM R&D Portfolio

## Accomplishments/Collaborations



### Low Temperature Electrolysis (LTE) (G. Bender: P148A; 5 Projects)



**PEM Electrolysis**

- PGM-free OER and HER catalyst
- Novel AEM and Ionomers
- Electrodes

**AEM Electrolysis**

### High Temperature Electrolysis (HTE) (R. Boardman: P148B; 5 Projects)

- Degradation mechanism at high current density operation
- Nickelate-based electrode and scalable, all-ceramic stack design

**O<sup>2-</sup> conducting SOEC**

- High performing and durable electrocatalysts
- Electrolyte and electrodes
- Low cost electrolyte deposition
- Metal supported cells

**H<sup>+</sup> conducting SOEC**

### Photoelectrochemical (PEC) (A. Weber: P148C; 5 Projects)

- III-V and Si-based semiconductors
- Chalcopyrites
- Thin-film/Si
- Protective catalyst system
- Tandem cell

**Semiconductors**

- PGM-free catalyst
- Earth abundant catalysts
- Layered 2D perovskites
- Tandem junction

**Perovskites**

### Solar Thermochemical (STCH) (A. McDaniel: P148D; 5 Projects)

- Computation-driven discovery and experimental demonstration of STCH materials
- Perovskites, metal oxides

**STCH**

- Solar driven sulfur-based process (HyS)
- Reactor catalyst material

**Hybrid Thermochemical**



# Accomplishments/Collaborations: HydroGEN Collaborative R&D Technical Highlights

## Low Temperature Electrolysis (LTE)

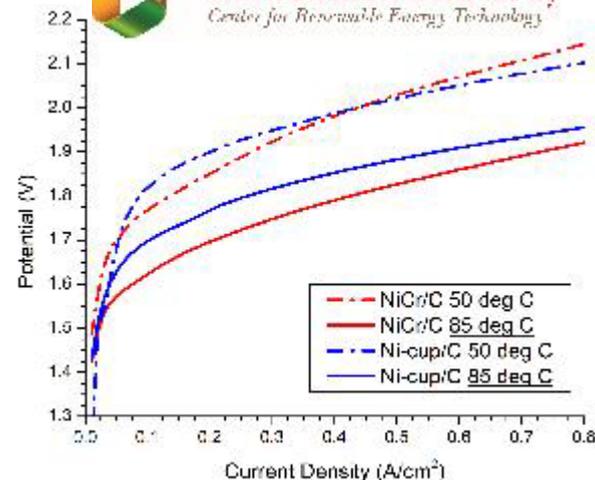
NEU and UD demonstrated **high AEM electrolyzer performance (0.46 A/cm<sup>2</sup> at 1.8 V, 85°C)**, using **PGM-free Ni-Fe/Raney Ni OER and NiCr/C HER catalysts, PAP-TP-MQN AEM, and 1% K<sub>2</sub>CO<sub>3</sub> solution**, enabling durable, high-performing materials for efficient and low cost H<sub>2</sub> production. LBNL and SNL modeling nodes helped NEU better understand the AEM/catalyst interface, NREL nodes helped NEU characterize the membrane physical properties and optimize the catalyst formulation.

## Photoelectrochemical (PEC) Water Splitting

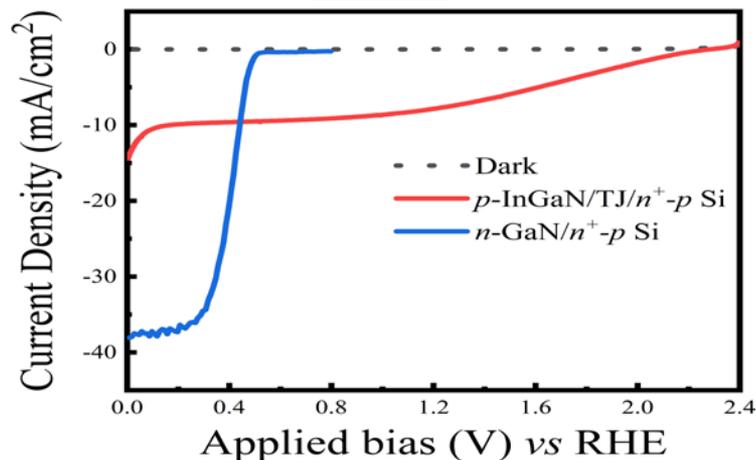
UM achieved a **first demonstration of a functional Si/InGaN tandem photoelectrode**. A GaN/Si photocathode with **stable operation (>100 h) at ~38 mA/cm<sup>2</sup>**, without using surface protection, was also demonstrated. UM is focused on developing Si-based, low cost, high efficiency (>15%), and stable (>1,000 h) PEC tandem water splitting devices, using scalable, low cost semi-conductors, nanowire tunnel junction, and N-rich GaN self-protection. The photoelectrodes were characterized and optimized by the NREL and LBNL nodes, while LLNL carried out modeling to understand the protection role of N-rich GaN surfaces.



Northeastern University  
Center for Renewable Energy Technology



Anode: Ni-Fe/Raney Ni; Membrane: polyaryl piperidiny triphenyl (PAP-TP-MQN), prepared by University of Delaware (UD)





# Accomplishments/Collaborations: HydroGEN Collaborative R&D Technical Highlights

## High Temperature Electrolysis (HTE)

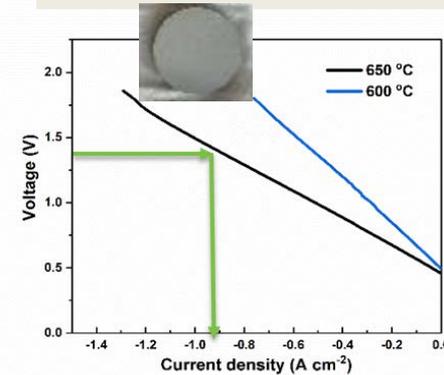
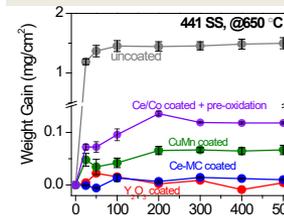
United Technologies Research Center (UTRC), with help from INL, LBNL, and NREL nodes, identified high performance proton-electrolytes and steam electrodes that resulted in **button cell performance exceeding the DOE performance target, stable protective coatings for 500 h, and a cell model for SOEC performance characterization and cell/stack operation simulation.** The HydroGEN nodes provided critical support by addressing technical barriers in metal alloy durability, electrode/electrolyte material optimization and stability, and SOEC modeling.



**United Technologies  
Research Center**

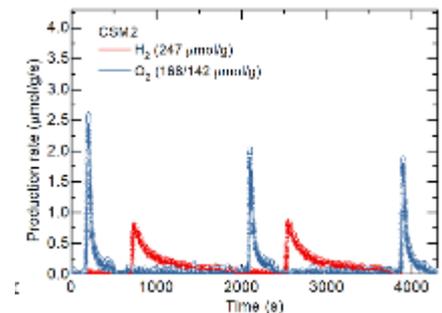
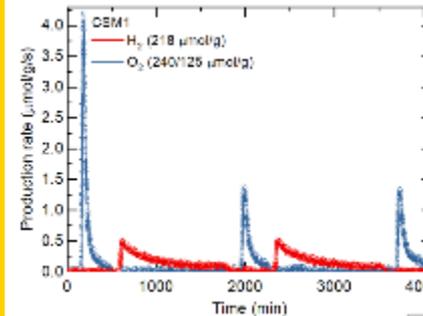
Button cell with SPS electrolyte exceeded BP1 target (>0.8 A/cm<sup>2</sup> at 1.4V & 650°C)

Protective coatings for SS alloys, 500h no oxidation



## Solar Thermochemical (STCH) Water Splitting

CSM exceeded its hydrogen production target of 59 μmol H<sub>2</sub>/g sample by producing 218 and 247 μmol H<sub>2</sub>/g using two compositions within Ce<sub>x</sub>Sr<sub>2-x</sub>MnO<sub>4</sub> family, x = 0.1 and 0.2, respectively. This promising performance further motivates the search for additional perovskite compounds with STCH water splitting properties. The objective is to discover new STCH materials that can meet steam to hydrogen efficiency of 20% and low cost H<sub>2</sub> production. The project leverages NREL and SNL modeling, material synthesis and characterization nodes.



# LTE/Hybrid Supernode: Linking LTE/Hybrid Materials to Electrode Properties to Performance (NREL, SRNL, LBNL; 8 Nodes)

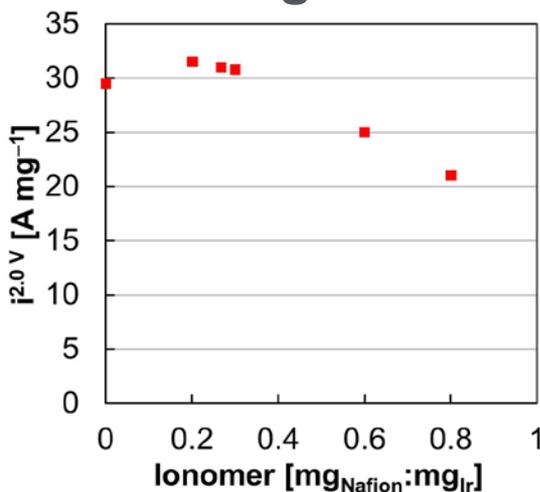


*Goal: Better integration between ex situ and in situ performance, more relevant ex situ testing, and improved material specific component development to achieve optimized electrolyzer cell performance and durability.*

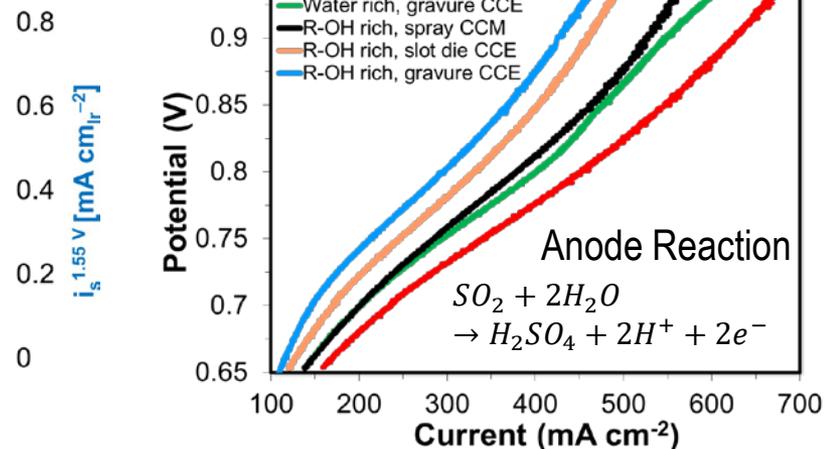
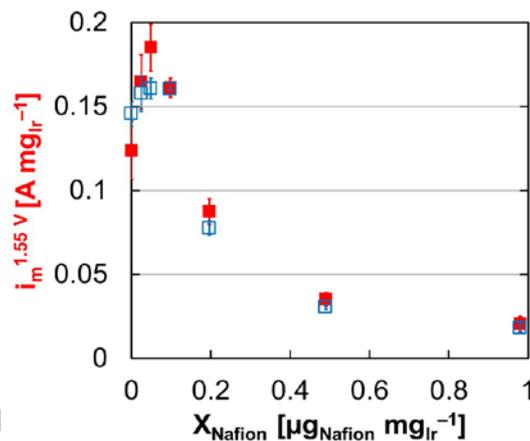
## Correlating Half- and Single- LTE Cells Similar Ionomer Content Effect

## Ink Formulation and Electrode Processing Affect $\text{SO}_2$ Electrolysis Performance

### MEA Single-Cells



### RDE Half-Cells

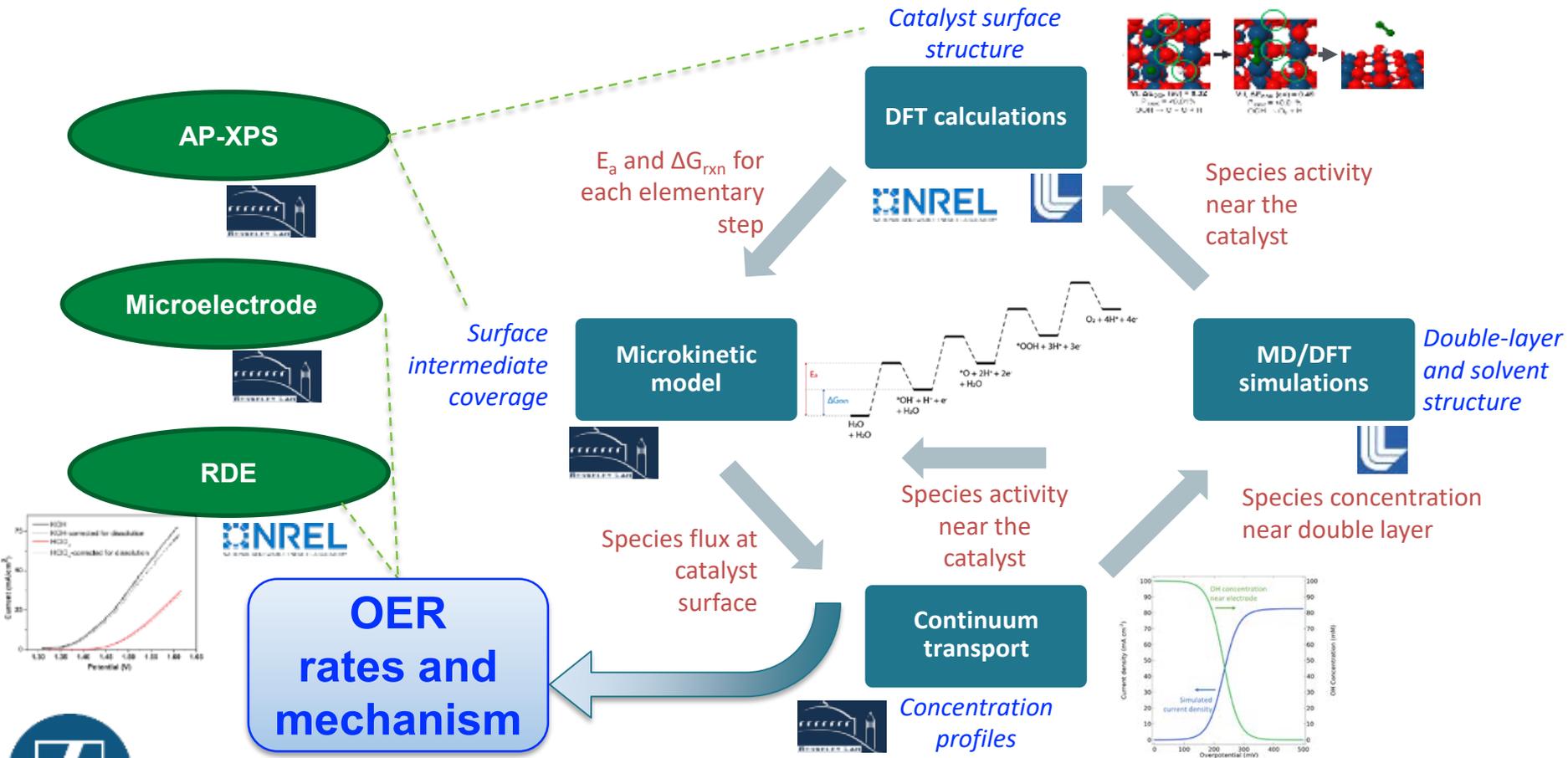


- Ionomer content has a similar effect on MEA and half-cell performance
- Ionomer balance is needed for optimal performance

- Material exchange between NREL and SRNL
- Water-rich inks and slot-die-made MEAs performed better than alcohol-rich (R-OH) inks and gravure-prepared MEAs.



# OER Supernode: Validated Multiscale Modeling To Understand OER Mechanisms across the pH Scale (NREL, LLNL; 6 Nodes)



- Developed various model handoff requirements and interactions
  - Microkinetic model for OER pathways formulated
  - Initial barriers for OER on IrO<sub>2</sub> in acid calculated
- Experimental characterization of OER on IrO<sub>2</sub> (RDE, AP-XPS, ME) initiated

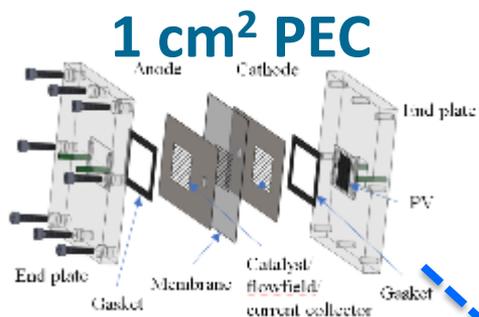




# PEC Supernode: Emergent Degradation Mechanisms with Integration and Scale Up of PEC Devices (NREL, LBNL; 7 Nodes)

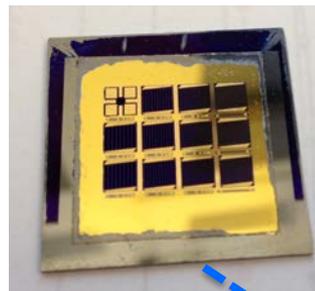


*Goal: Understand integration issues and emergent degradation mechanisms of PEC devices at relevant scale, and demonstrate an integrated and durable 50 cm<sup>2</sup> PEC panel.*



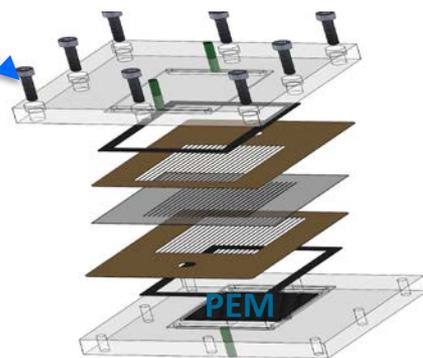
1 cm<sup>2</sup> PEC

0.1 cm<sup>2</sup> PV



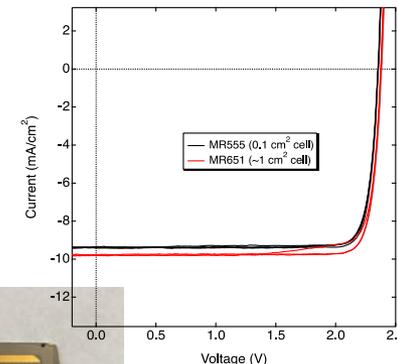
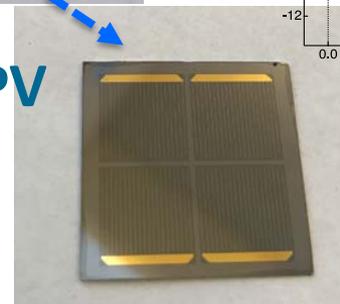
4 cm<sup>2</sup> PEC

Anode  
Flowfield/GDL



Cathode  
Flowfield/GDL

1 cm<sup>2</sup> PV



*Accomplishments: PV fabrication scale up from 0.1 to 1 cm<sup>2</sup>  
PEC vapor cell scale up from 1 to 4 cm<sup>2</sup>  
Benchmarking PEC cell performance between NREL and LBNL*

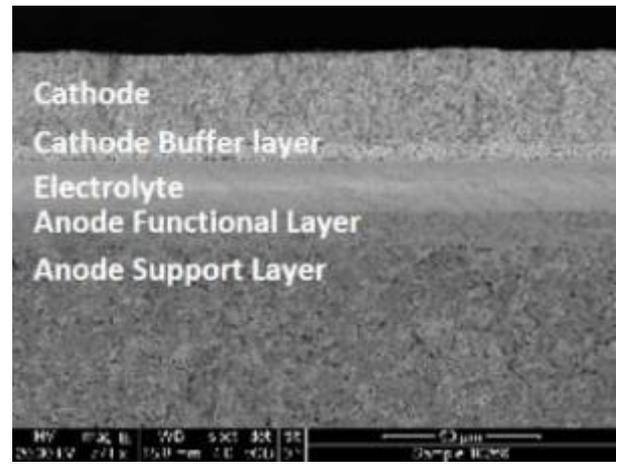


# HTE Supernode: Characterization of Solid Oxide Electrode Microstructure Evolution (INL, NREL, LBNL, LLNL, Sandia; 7 Nodes)



*Goal: Deeper understanding of high-temperature electrolysis (HTE) electrode microstructure evolution as a function of local solid oxide composition and operating conditions.*

- Accomplishments:**
- ✓ First batch of YSZ-based SOEC fabricated using high-purity precursors.
  - ✓ Sintering aid used in cathode buffer layers.
  - ✓ Electrolyte thickness: 10-15  $\mu\text{m}$ .



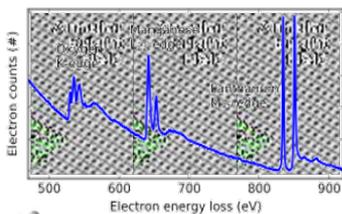


# STCH Supernode: Develop Fundamental Understanding of the Mn-O Ligand Field's Influence on Water Splitting (LLNL, NREL, SNL; 6 Nodes)

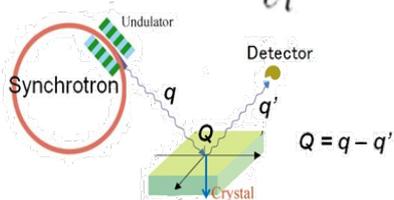


*Goal: Develop a fundamental understanding of how unique electronic structures, induced by Mn-O ligand bond arrangements, influence favorable water-splitting material behavior.*

## operando probes of Mn-O electronic structure

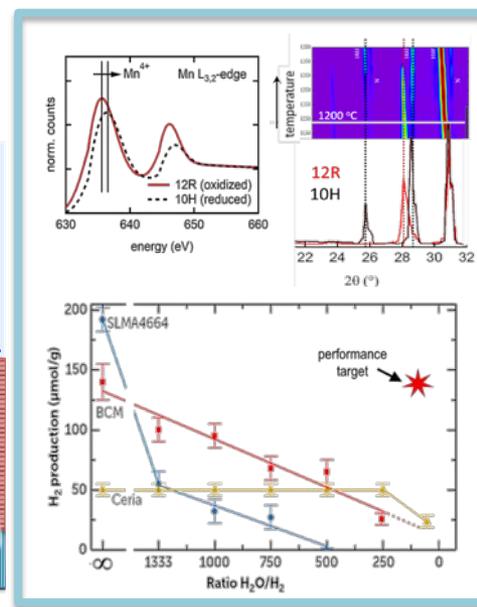
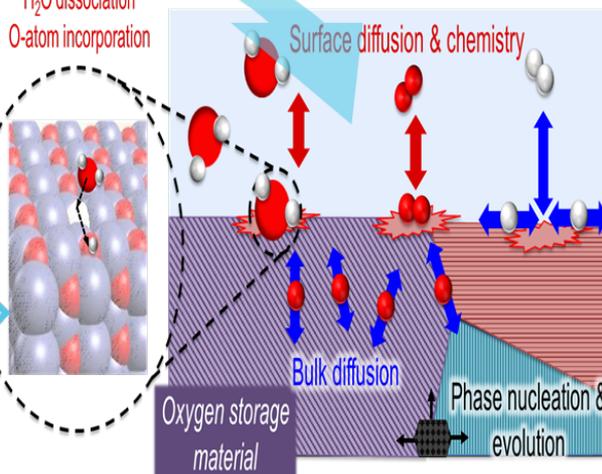


$$\left[ \frac{-\hbar^2}{2m} \nabla^2 + V \right] \psi = i\hbar \frac{\partial}{\partial t} \psi$$



Key governing processes in STCH materials

- O-atom vacancy formation
- H<sub>2</sub>O dissociation
- O-atom incorporation



## Accomplishments:

- Develop synthetic routes to “ideal” Mn-O compounds
- Develop DFT methods to model core-hole spectroscopies
- Deploy operando test cell at SLAC (X-ray scattering)
- Conduct HR/STEM EELS studies on “ideal” compounds



# Accomplishments: HydroGEN Benchmarking Advanced Water Splitting Technologies Project (P170)

## Best Practices in Materials Characterization

PI: Kathy Ayers, Proton OnSite (LTE)  
Co-PIs: Ellen B. Stechel, ASU (STCH);  
Olga Marina, PNNL (HTE);  
CX Xiang, Caltech (PEC)  
Consultant: Karl Gross



### Accomplishments:

- 1<sup>st</sup> Annual AWS community-wide benchmarking meeting (ASU, Oct. 24–25, 2018)
- 4 test frameworks developed
- 4 AWS technology surveys and result summaries published on the Data Hub
- 4 preliminary AWS roadmaps developed
- 1<sup>st</sup> round of test protocols defined and written
- Quarterly newsletters disseminated to AWS community
- >80 EMN capability nodes assessed; node gaps identified and communicated to HydroGEN

### Goals:

- *Develop standardized best practices for characterizing and benchmarking AWSMs*
- *Foundation for accelerated materials RD&D for broader AWS community*
- *Extensive collaboration and engagement with HydroGEN steering committee, node subject matter experts, and broad water splitting community*

***Development of Best Practices in Materials Characterization and Benchmarking:  
Critical to accelerate materials discovery and development***



## Responses to Previous Year Reviewers' Comments

- The fact that about half of the 80+ nodes (capabilities) are being utilized suggests that the interaction with the HydroGEN-supported R&D projects/community is a benefit toward helping DOE realize its goals. It is unclear how to put into perspective the number of users, page views, downloads, etc., as well as the publications and presentations, and whether these are helping DOE achieve its goals/targets. As HydroGEN “matures,” a better metric would be clear evidence of how the nodes had an impact in making measurable/quantifiable benefits toward advancing R&D to meet DOE goals.

**Response:** We agree that testimonials and specific points of collaboration as witnessed by joint publications and presentations will be key in evaluating whether the nodes are providing critical or ancillary support for the FOA projects. Furthermore, cost reduction and technology maturity remain key metrics for success of the EMN.

- Probably one of the most pressing issues under Proposed Future Work is the development of an effective data management program, not so much in managing the data but in presenting to the R&D community in a format that is of value to the community. Developing benchmarking standard protocols and metrics is another pressing issue to make sure that the protocols for evaluating the various technologies provide an apples-to-apples comparison.

**Response:** We agree and are currently working on the data management in terms of dissemination with full metadata. In terms of protocols and metrics, this is an active area of work with the Benchmarking team lead by NEL/Proton OnSite, where HydroGEN core labs are actively engaged in helping to evaluate and establish the protocols.



# Responses to Previous Year Reviewers' Comments

- Some of the secondary, visible metrics, such as the use of the data hub (~250 data files in a year), should get more emphasis either on boosting participation or in communicating the complexity of the data contained within the hub. There is ambiguity as to what a single file contains: whether it is a single resistance measurement or a summary from an entire collection of measurements from a unique tool on the beamline. In short, if the scale of the databank were conveyed in person-hours per data file that makes up the ~250 total, that could strike an audience as more impressive/appropriate than leaving the number of files to remain as an abstract concept, which risks sounding underwhelming.

**Response:** There has been lots of activities in the data hub this past year, hence there are 2 slides on data, summarizing the publication process developed, the various tools that have been developed, and the type of data that are on the data hub (e.g., microscopy, pol curves, XRD, XPS). Each data file represents one of these data. We hope these two slides better communicate the level of participation and the complexity of the data in the Data Hub.

- Although individual projects have milestones and go/no-go points, the AWSM Consortium lacks clear success metrics at a higher level to guide its pathways and projects.

**Response:** The success metrics for the consortium are varied. The overall metric is cost reduction (through performance and durability gains) for the various water-splitting materials, remaining cognizant that they are at very different levels of technological maturity. Interactions and enabling the FOA projects is key to this and thus serves as metrics for the consortium. In addition, the existence of supernodes (with their more traditional research metrics, shown on FY19 Supernode Annual Milestones [AM] slide in "Reviewers Only slides"), the data analysis and data hub, and the research progress of individual nodes all provide further metrics for measuring success.



# Proposed Future Work

- Core labs will execute HydroGEN nodes to enable successful phase 2 project activities and work with new phase 1 projects
  - Core labs' interaction with a specific project will end if that project does not achieve its go/no-go decision metric
- Collaborate and perform integrated research in the 5 supernodes
- Integrate whole system (capability nodes, FOA awardees, data infrastructure, TT/A) to accelerate the R&D of HydroGEN critical materials development to deployment
- Continue to review, maintain, and develop current and identify new relevant HydroGEN capability nodes
- Continue to develop a user-friendly, secure, and dynamic HydroGEN data hub that accelerates learning and information exchange within the HydroGEN EMN labs, their partners, and other EMN, AE, PEC, and STCH communities
- Work closely with the Benchmarking Team to establish benchmarking, standard protocols, and metrics for the different water-splitting technologies
- Outreach

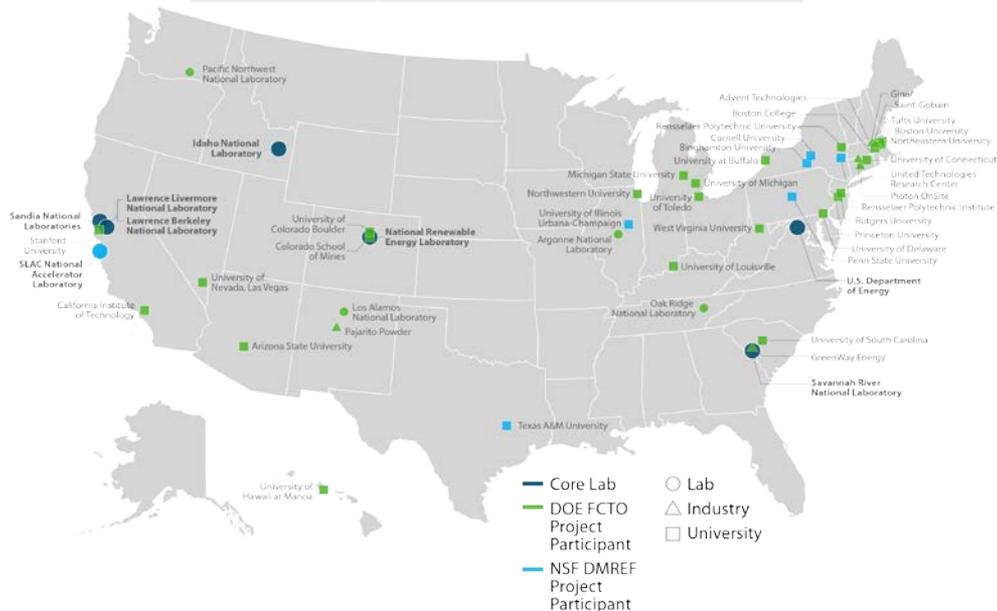


# Summary – HydroGEN Consortium: Advanced Water-Splitting Materials (AWSM)



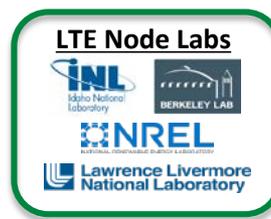
## A Nationwide, Inter-Agency, Collaborative Consortium in Early-Stage Materials R&D

11 Labs 7 Companies 30 Universities 2 Funding Agencies



>80 unique, world-class capabilities/expertise:

- **Materials theory/computation**
  - **Synthesis**
  - **Characterization and analysis**
- **16 projects successfully passed GNG**
  - **5 new Supernodes**
  - **4 new NSF DMREF projects**
  - **2 new HTE projects**
  - **2 Work for Others agreements**
  - **5 new data tools; >4,000 files**
  - **4 partner testimonial videos**
  - **1 annual benchmarking workshop**
  - **Multiple AWS standard protocols**



Support through:

- Personnel
- Equipment
- Expertise
- Capability
- Materials
- Data



*HydroGEN fosters cross-cutting innovation using theory-guided applied materials R&D to advance all emerging water-splitting pathways for hydrogen production*

# Acknowledgements



Energy Materials Network  
U.S. Department of Energy



**HydroGEN**  
Advanced Water Splitting Materials

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**Katie Randolph**



**David Peterson**



**James Vickers**



**Eric Miller**



# Acknowledgements



Energy Materials Network  
U.S. Department of Energy



**HydroGEN**  
Advanced Water Splitting Materials

## NREL Team

**Huyen Dinh, Lead  
Principal Investigators:**

Shaun Alia	Zhiwen Ma
Mowafak Al-Jassim	Scott Mauger
Guido Bender	Kristin Munch
Joe Berry	Judy Netter
Jeff Blackburn	John Perkins
Todd Deutsch	Bryan Pivovar
Daniel Friedman	Matthew Reese
David Ginley	Genevieve Saur
Mai-Anh Ha	Glenn Teeter
Kevin Harrison	Michael Ulsh
Steven Harvey	Judith Vidal
Stephan Lany	Andriy Zakutayev
Ross Larsen	Kai Zhu

## LBLN Team

**Adam Weber, Lead  
Principal Investigators:**

Nemanja Danilovic	Francesca Toma
Ian Sharp	Miquel Salmeron
Peter Agbo	Ethan Crumlin
David Larson	Jeffrey Greenblat
Lin-Wang Wang	Ahmet Kusoglu
Walter Drisdell	Frances Houle
Mike Tucker	David Prendergast

## SRNL Team

**Hector Colón-Mercado, Lead  
Principal Investigators:**

Maximilian Gorensek    Brenda Garcia-Diaz

# Acknowledgements



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Advanced Water Splitting Materials

## SNL Team

**Anthony McDaniel, Lead  
Principal Investigators:**

Mark Allendorf	Bryan Kaehr
Eric Coker	David Littlewood
Bert Debusschere	John Mitchell
Farid El Gabaly	Jeff Nelson
Lindsay Erickson	Peter Schultz
Ivan Ermanoski	Randy Schunk
James Foulk	Subhash Shinde
Cy Fujimoto	Josh Sugar
Fernando Garzon	Alec Talin
Ethan Hecht	Alan Wright
Reese Jones	

## LLNL Team

**Tadashi Ogitsu, Lead  
Principal Investigators:**

Sarah Baker	Tuan Anh Pham
Monika Biener	Christopher Spadaccini
Alfredo Correa Tedesco	Tony Van Buuren
Thomas Yong-Jin Han	Joel Varley
Tae Wook Heo	Trevor Willey
Jonathan Lee	Brandon Wood
Miguel Morales-Silva	Marcus Worsley
Christine Orme	

## INL Team

**Richard Boardman, Lead  
Principal Investigators:**

James O'Brien	Josh Mermelstein
Dong Ding	Jeremy Hartvigsen
Rebecca Fushimi	Hanping Ding
Dan Ginosar	Michael Glazoff

# Acknowledgements



Energy Materials Network  
U.S. Department of Energy



**HydroGEN**  
Advanced Water Splitting Materials

## LTE Supernode Team



Shaun Alia  
Grace Anderson  
Guido Bender  
Huyen Dinh  
Allen Kang  
Scott Mauger  
Bryan Pivovar  
Michael Ulsh  
James Young



Donald Anton  
Hector Colón-Mercado



Nemanja Danilovic  
Ahmet Kusoglu  
Adam Weber



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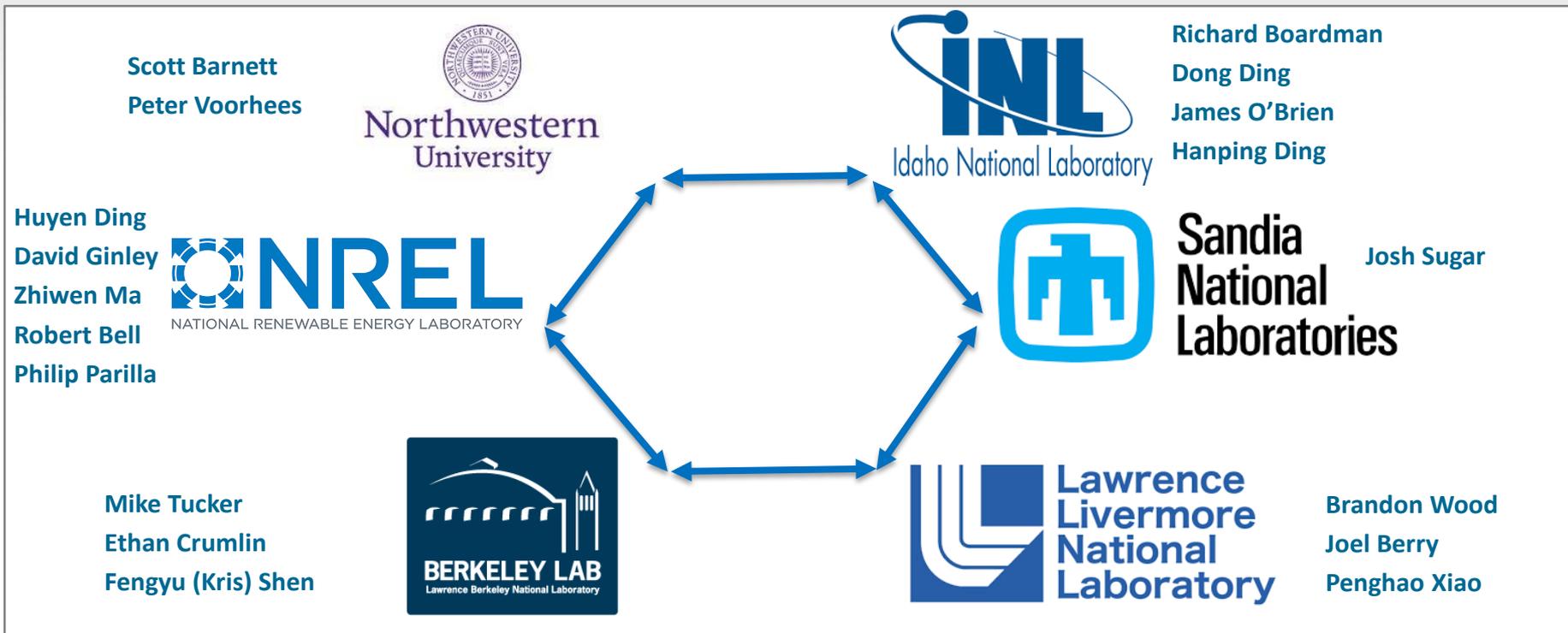


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Advanced Water Splitting Materials

## HTE Supernode Team



# Acknowledgements



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## OER Supernode Team



Adam Weber  
Nemanja Danilovic  
Lien-Chung Weng

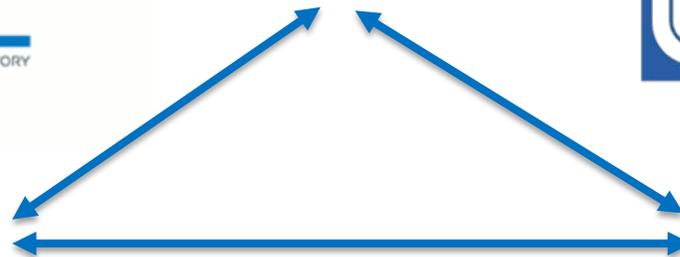
Ethan Crumlin  
David Prendergast



Ross Larsen  
Mai-Anh Ha  
Shaun Alia



Tadashi Ogitsu  
Brandon Wood  
Tuan Anh Pham



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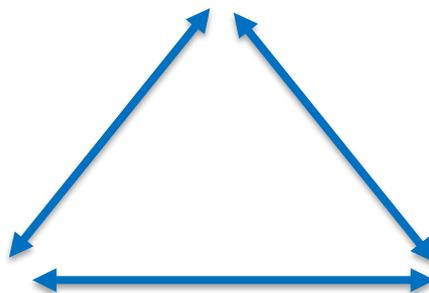
## PEC Supernode Team

### Best Practices in Materials Characterization

PI: Kathy Ayers, Proton OnSite (LTE)  
Co-PIs: Ellen B. Stechel, ASU (STCH);  
Olga Marina, PNNL (HTE);  
**CX Xiang, Caltech (PEC)**  
Consultant: Karl Gross



Todd Deutsch  
Huyen Dinh  
James Young  
Myles Steiner  
Dan Friedman



Adam Weber  
Frances Houle  
Nemanja Danilovic  
Francesca Toma  
Tobias Kistler  
Guosong Zeng

Lien-Chung Weng  
David Larson  
Jefferey Beeman



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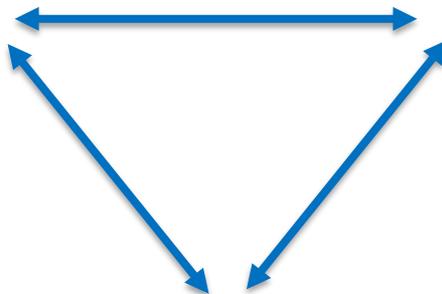
**HydroGEN**  
Advanced Water Splitting Materials

## STCH Supernode Team



**Sandia  
National  
Laboratories**

Andrea Ambrosini  
Eric Coker  
Anthony McDaniel  
Josh Sugar  
Josh Whaley



**Lawrence Livermore  
National Laboratory**

Tadashi Ogitsu  
Sabrina Wan  
Brandon Wood



NATIONAL RENEWABLE ENERGY LABORATORY

Robert Bell  
David Ginley  
Stephan Lany  
Jie Pan  
Philip Parilla





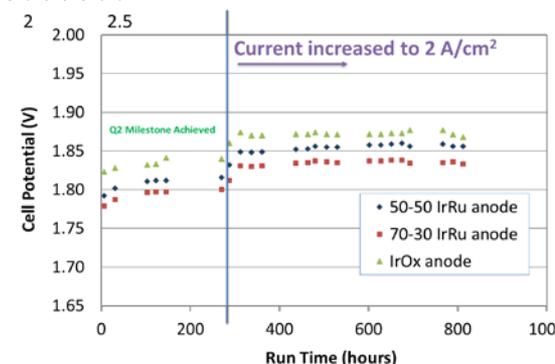
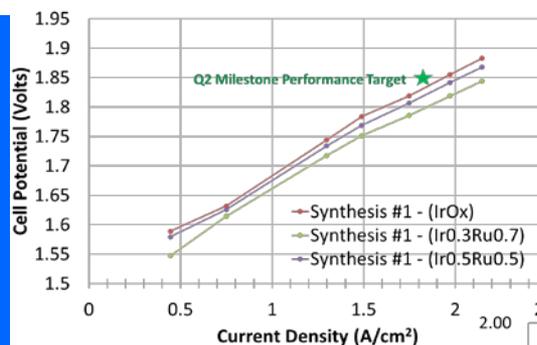
# Technical Backup Slides



# Accomplishments/Collaborations: HydroGEN Collaborative R&D Technical Highlights

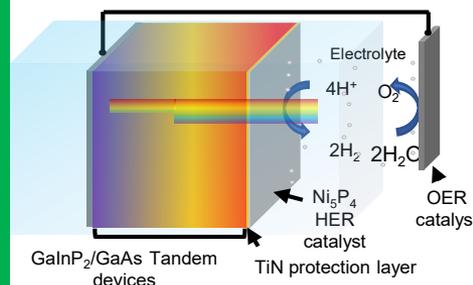
## Low Temperature Electrolysis (LTE)

NREL's contributed towards Proton Onsite achievement of **1.8 V at 2.0 A/cm<sup>2</sup>**, and **800 h PEM electrolysis durability at 2 A/cm<sup>2</sup>**, operating at 80°C and 30 bar. Proton's improved cell efficiency is a step towards achieving its PEM water electrolysis cell efficiency goal of **43 kWh/kg** (1.7 V at 90°C) and at a cost of \$2/kg H<sub>2</sub>, a significant improvement over the state-of-the-art cell efficiency of 53 kWh/kg.

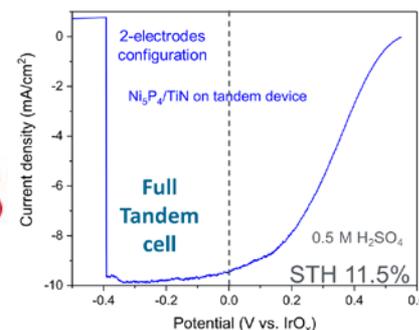


## Photoelectrochemical (PEC) Water Splitting

NREL's high performance photoabsorber (GaInP<sub>2</sub>/GaAs), integrated with Rutgers' PGM-free electrocatalysts (LiCoO<sub>2</sub> and Ni<sub>5</sub>P<sub>4</sub>) and protection layer (TiN), achieved a solar-to-hydrogen efficiency of **11.5%** for unassisted water splitting, on par performance with conventional PGM electrocatalysts (PtRu).



# RUTGERS



(NREL FY2018 Q4 AM – see Reviewer-Only slides)



# Accomplishments/Collaborations: HydroGEN Collaborative R&D Technical Highlights

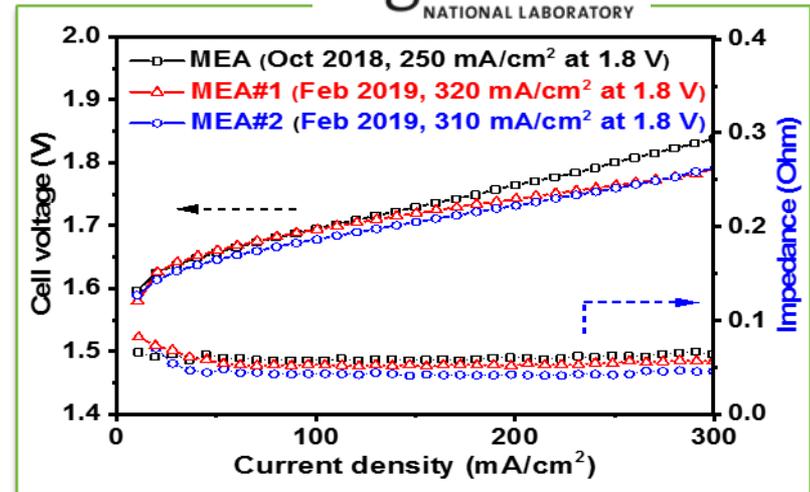
## Low Temperature Electrolysis (LTE)

ANL demonstrated a **25 % improvement in PEM electrolysis cell performance** ( $310 \text{ mA/cm}^2$  at  $1.8 \text{ V}$ ), using **PGM-free Co-ZIF derived OER** and Pt/C HER catalysts, N115 membrane,  $60^\circ\text{C}$  &  $10 \text{ psi}$  DI water, enabling the goal to reduce the anode catalyst cost by  $> 20$  folds for the widespread implementation of PEME  $\text{H}_2$  production. LBNL, LLNL, NREL, and SNL modeling, XPS, and microscopy nodes, respectively, contributed towards this achievement.

## Photoelectrochemical (PEC) Water Splitting

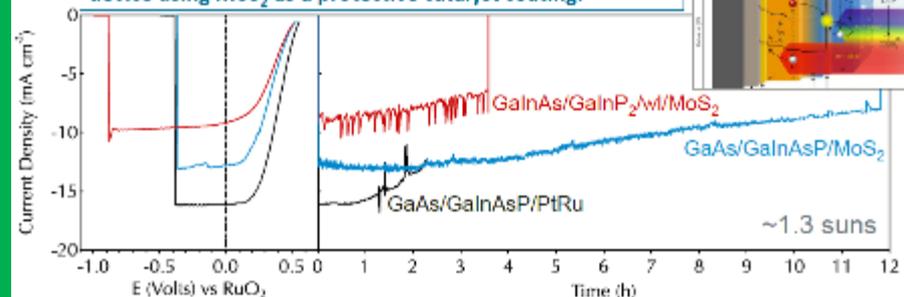
SU demonstrated **unassisted PEC water-splitting** with **PGM-free  $\text{MoS}_2$  HER catalyst** that achieved solar-to-hydrogen (STH) efficiency  $> 5\%$  under **1 sun**, providing a viable path for achieving low cost, stable and high (20%) STH PEC devices through earth-abundant protective catalysts, novel growth schemes, and new tandem III-V/Si system that has the potential to dramatically reduce  $\text{H}_2$  production cost. The project leverages the NREL III-V fabrication, PEC characterization, corrosion, and on-sun testing expertise.

Argonne  
NATIONAL LABORATORY



STANFORD  
UNIVERSITY

Unassisted water splitting for  $\sim 12$  hours and STH  $> 5\%$  with a III-V/III-V inverted metamorphic multijunction (IMM) PEC device using  $\text{MoS}_2$  as a protective catalyst coating.





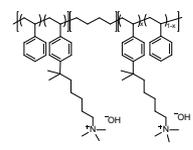
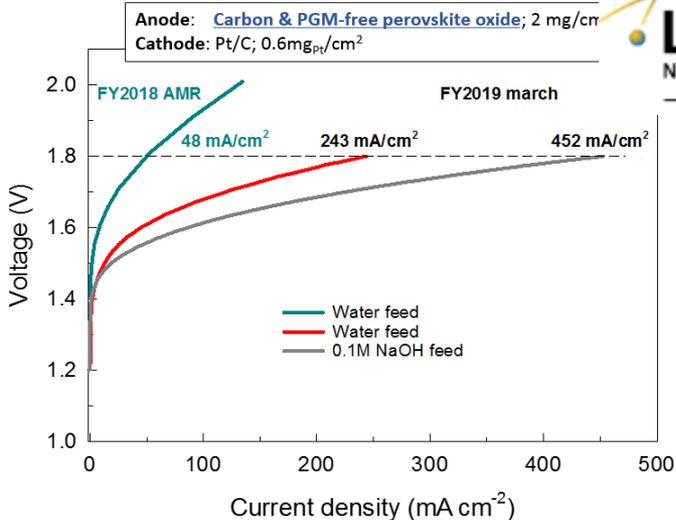
# Accomplishments/Collaborations: HydroGEN Collaborative R&D Technical Highlights

## Low Temperature Electrolysis (LTE)

LANL demonstrated **4x higher** in water-fed AEM electrolyzer performance (**0.243 A/cm<sup>2</sup> at 1.8 V**), compared to FY2018, using LANL developed **carbon-free and PGM-free perovskite OER catalyst** and SNL AEM membrane. The goal is to achieve low cost H<sub>2</sub> production via high performing, durable, and low cost PGM-free catalyst AEM electrolysis. NREL XPS and in-situ testing nodes helped LANL understand the OER catalyst surface composition and MEA electrode performance.

## Low Temperature Electrolysis (LTE)

LANL and RPI prepared **semi-crystalline AEM membranes**, by acid catalyzed polymerization & without using expensive metal catalyst, **that exceeded all of the project's (and state-of-the-art) membrane conductivity, alkaline stability, and mechanical strength targets**, as validated by LBNL characterization and modeling nodes. The goal is to synthesize and demonstrate high performing, durable, and economically-affordable AEMs in water-fed AEM electrolysis.



**Properties**  
IEC = 1.71 mequiv./g  
Water uptake = 144 %  
In-plane swelling = 30%



Properties	Target	Status
Hydroxide conductivity (mS cm <sup>-1</sup> ) at 30 °C	40	42 (30 °C) 54 (60 °C) 63 (80 °C)
Alkaline stability after 300 h in 1 M NaOH at 80 °C	< 5 % loss conductivity	0% loss
Mechanical toughness (mechanical strength (MPa) × % elongation) at 50 °C, 90% RH	> 1400	2091



# Accomplishments/Collaborations: HydroGEN Collaborative R&D Technical Highlights

## High Temperature Electrolysis (HTE)

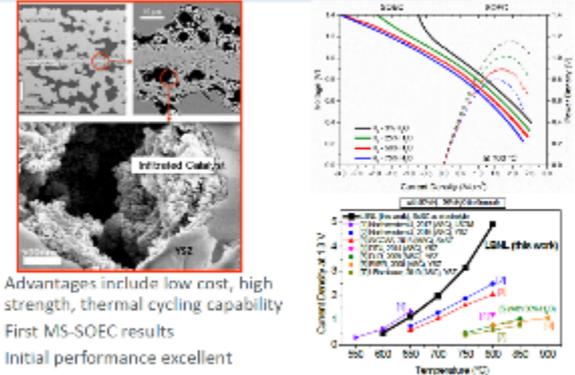
Using Northwestern University catalyst, YSZ electrolyte ( $(\text{ZrO}_2)_{0.92}(\text{Y}_2\text{O}_3)_{0.08}$ ), and LBNL Metal-Supported Solid Oxide Cell and INL Advanced HTE testing nodes, the collaboration demonstrated a metal-supported SOEC for the first time in electrolysis mode, with the highest performance for oxygen-conducting type electrolysis cells to-date and promising stability.

## Solar Thermochemical (STCH) Water Splitting

The University of Colorado Boulder, with NREL's DFT node, developed and applied machine learning (ML) to accelerate STCH materials discovery, **identifying several hundred stable STCH perovskites** from over 1.1 million possible candidates, with **92% accuracy**. SNL's stagnation flow reactor and High-Temperature XRD nodes were used to **experimentally validate** water splitting kinetics and crystal structures for a select number of materials, providing critical feedback to **develop rapid kinetic screening techniques of materials**. Four materials have also been demonstrated to have  $\text{H}_2$  productions **>200  $\mu\text{mol/g/cycle}$**  at  $T_{\text{red}} = 1450^\circ\text{C}$  and  $\Delta T = 250^\circ\text{C}$ , and results compared to computational predictions.

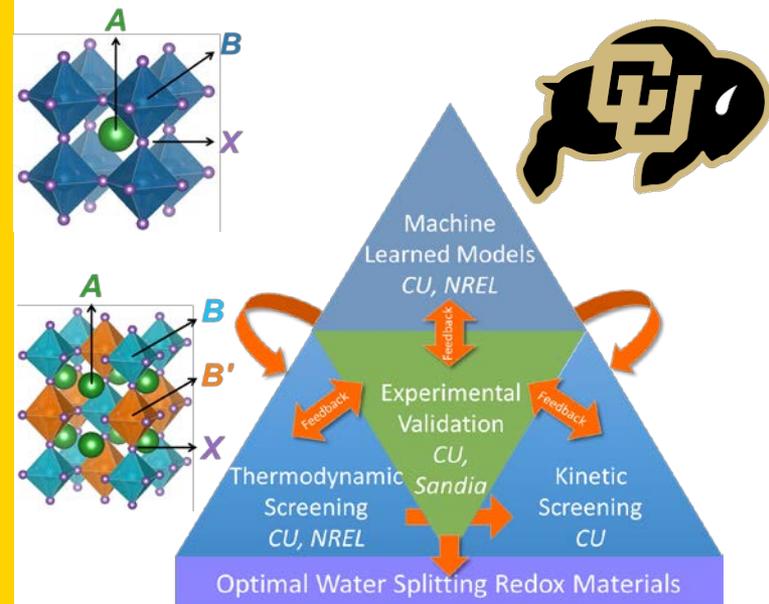
## Northwestern

Symmetric Structure Metal-Supported Solid Oxide Electrolysis Cells (MS-SOECs)



- Advantages include low cost, high strength, thermal cycling capability
- First MS-SOEC results
- Initial performance excellent
- Fast degradation

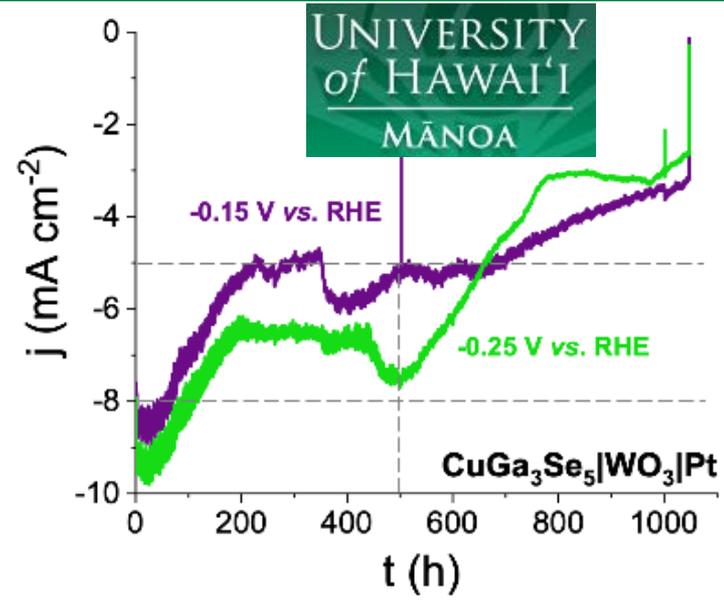
HydroGEN Advanced Water Splitting Variables





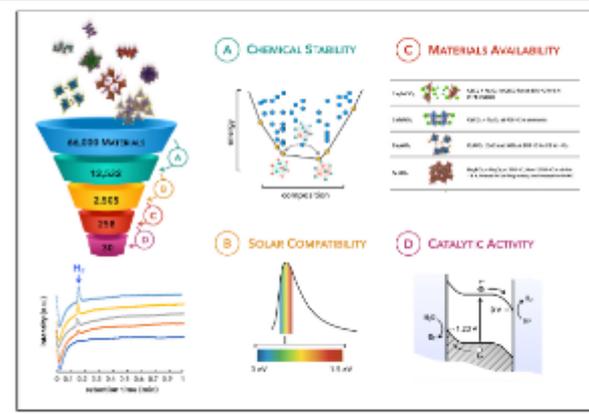
# Accomplishments/Collaborations: HydroGEN Collaborative R&D Technical Highlights

**Photoelectrochemical (PEC) Water Splitting**  
UH improved chalcopyrite stability in aqueous electrolyte by demonstrating **> 500 h of continuous operation at > 5 mA/cm<sup>2</sup>**, using a thin WO<sub>3</sub> coating on a high bandgap copper chalcopyrite, paving the way to creating a low cost (“printed”) chalcopyrite-based, semi-monolithic, tandem hybrid photoelectrode device prototype that can operate for at least 1,000 h with STH efficiency > 10%. This project is supported by NREL synthesis and advanced characterization and LLNL modeling expertise to accelerate the development of materials and interfaces.



Screening of photocatalysts for sustainable hydrogen generation  
H. D. Abruña<sup>1</sup>, I. Dabo<sup>2</sup>, T. G. Deutsch<sup>3</sup>, C. J. Fennie<sup>1</sup>, V. Gopalan<sup>2</sup>, R. E. Schaak<sup>2</sup>  
<sup>1</sup>Cornell University <sup>2</sup>The Pennsylvania State University <sup>3</sup>National Renewable Energy Laboratory  
Award DMREF-1728338

NSF-DMREF / DOE EERE HydroGEN EMN  
Example:  
**Photoelectrochemical (PEC) Water Splitting**



Starting from 66,000 materials from the Materials Project, 30 candidate photocatalysts have been screened by comparing their chemical stability, solar compatibility, and catalytic activity. The ability to synthesize these materials is ensured by a systematic survey of the literature. Experiment shows hydrogen generation for four of the nine candidates tested so far.



## Reviewer-Only Slides



# FY19 NREL AOP Milestones

Milestone Name/Description	End Date	Type	Status
Kick-start the three NSF DMREF projects that NREL is supporting (PSU PEC, PSU LTE, and CSM STCH) and integrate them into HydroGEN.	12/31/2018	Q1 QPM	<b>Complete</b>
Initiate the 5 supernode efforts including a joint virtual kick-off meeting for each one. (NREL, LBNL, SNL, LLNL, INL, SRNL).	3/31/2019	Q2 QPM	<b>Complete</b>
Integrate new services into the Data Hub, including centralized authorization and sample management functions. Much of the HydroGEN research hinges on aligning and associating characterization and modeling data to uniquely identified samples. We will implement a sample management service into the Data Hub, which will include a backend database, an API layer and a data security layer. For centralized authorization, we will implement the Cognito Authentication service into the Data Hub, enabling 2-factor authentication consistently across all EMN data hubs.	6/30/2019	Q3 QPM	<b>On Track</b> (central authorization is complete)
Quantify the relationship between the exchange current density, specific activity and electrochemical surface area of a standard LTE OER catalyst via RDE with in-situ performance for at least 3 different electrode processing conditions and/or compositions, with properties comparable to IrO <sub>2</sub> (mass exchange current density = 0.075 mA/mg, ECA = 28.7 m <sup>2</sup> /g and specific exchange current density = 0.26 μA/cm <sup>2</sup> ). (LTE/Hybrid Supernode Milestone)	9/30/2019	Q4 Annual Milestone	<b>On Track</b>



# FY19 Supernode Annual Milestones (AM)

Milestone Name/Description	Supernode/ Lab AOP	Type	Status
Compare the exchange current density, specific activity and electrochemical surface area of a standard Hybrid Sulfur Cycle anode electro-catalyst via RDE with in-situ performance for at least 3 different electrode processing conditions and/or compositions with the goal of approaching 50-60% agreement between ex-situ and in-situ test.	LTE/Hybrid SRNL	Q4 AM 9/30/2019	<b>On Track</b>
Multiscale model predicts dominant reaction pathway with overall reaction rates in agreement within 10% error with experimental data for OER on IrO <sub>2</sub> under acidic conditions at two different applied overpotentials.	OER LBNL & LLNL	Q4 AM 9/30/2019	<b>On Track</b>
The synthesized microstructures of 4-6 electrode/electrolyte pure phase material will be analyzed to confirm repeatability of microstructure and composition via microscopy and X-ray diffraction analysis. The materials will have no secondary phases and the microstructure will have porosity and pore size within 10% of each other.	HTE INL	Q4 AM 9/30/2019	<b>On Track</b>
Identify one or more Mn-O baseline systems sharing key characteristics with BCM, which produces 3x more hydrogen than CeO <sub>2</sub> when reduced at 1350°C and maintains higher water splitting favorability than SLMA at low oxygen partial pressure (>50 umole/g at H <sub>2</sub> O/H <sub>2</sub> ratio<500). Data from SNL nodes will support first principles model development and refinement activities.	STCH SNL	Q4 AM 9/30/2019	<b>On Track</b>
Measure efficiency and durability a device of 8 cm <sup>2</sup> or larger area using a flexible-platform photoreactor equipped with multiple in-situ diagnostic techniques on 2-axis tractor for 2+ days.	PEC LBNL & NREL (not in AOP)	Q4 AM 9/30/2019	<b>On Track</b>



# FY18 AOP Milestones

Milestone Name/Description	End Date	Type	Status
Organize and host a HydroGEN project kick-off meeting for the 19 new FOA awardees and the 6 core lab members to help integrate them into the EMN.	12/31/2017	QPM	<b>Complete Nov. 2017</b>
80 HydroGEN capabilities reviewed based on developed process and evaluation criteria (e.g., utilization across the 18 new FOA projects).	3/30/2018	Annual Milestone	<b>Complete</b>
Integrate a data publication process into the data hub, enabling methods for assigning DOIs to uniquely identify public datasets and processes for approving and sharing data with the public.	6/30/2018	QPM	<b>Complete (slide 10)</b>
Benchmark solar-to-hydrogen efficiency of best-in-class LiCoO <sub>2</sub> anode and Ni <sub>5</sub> P <sub>4</sub> cathode catalysts integrated on an upright GaInP <sub>2</sub> /GaAs tandem cell with a target of greater than 10%.	9/30/2018	QPM	<b>Complete (slide 39)</b>



# Patents and Patent Applications

1. “Nanofiber Electrocatalyst,” Di-Jia Liu and Lina Chong, US Patent Application Publication, US2019/0060888A1.
2. “Electrochemical cells for hydrogen gas production and electricity generation, and related structures, apparatuses, systems, and methods,” Dong Ding, Hanping Ding, Wei Wu and Chaojiang. US Provisional Patent Application 62/751, 969 (2018).
3. “Electrochemical cells for hydrogen gas production and electricity generation, and related structures, apparatuses, systems, and methods,” Dong Ding, Hanping Ding, Wei Wu and Chaojiang. US Provisional Patent Application 62/722, 151 (2018).



# Publications

1. S.M. Alia and G.C. Anderson. “Iridium Oxygen Evolution Activity and Durability Baselines in Rotating Disk Electrode Half-Cells.” *Journal of The Electrochemical Society* 166, no. 4 (2019): F282–294.
2. S.M. Alia and B.S. Pivovar. “Evaluating Hydrogen Evolution and Oxidation in Alkaline Media to Establish Baselines.” *Journal of The Electrochemical Society* 165, no. 7 (2018): F441–55.  
<https://doi.org/10.1149/2.0361807jes>.
3. T.A. Kistler, D. Larson, K. Walczak, P. Agbo, I.D. Sharp, A.Z. Weber, and N. Danilovic. “Integrated Membrane-Electrode-Assembly Photoelectrochemical Cell under Various Feed Conditions for Solar Water Splitting.” *Journal of The Electrochemical Society* 166, no. 5 (2019): H3020–28.  
<https://doi.org/10.1149/2.0041905jes>.
4. C.P. Muzzillo, W.E. Klein, Z. Li, A.D. DeAngelis, K. Horsley, K. Zhu, and N. Gaillard. “Low-Cost, Efficient, and Durable H<sub>2</sub> Production by Photoelectrochemical Water Splitting with CuGa<sub>3</sub>Se<sub>5</sub> Photocathodes.” *ACS Applied Materials & Interfaces* 10, no. 23 (2018): 19573–79.  
<https://doi.org/10.1021/acsami.8b01447>.
5. T.R. Hellstern, D.W. Palm, J. Carter, A.D. DeAngelis, K. Horsley, L. Weinhardt, W. Yang, M. Blum, N. Gaillard, C. Heske, and T.F. Jaramillo. “Molybdenum Disulfide Catalytic Coatings via Atomic Layer Deposition for Solar Hydrogen Production from Copper Gallium Diselenide Photocathodes.” *ACS Appl. Energy Mater.* 2, no. 2 (2019): 1060.
6. A.D. Deangelis, K. Horsley, and N. Gaillard. “Wide Band Gap CuGa(S,Se)<sub>2</sub> Thin Films on Transparent Conductive Fluorinated Tin Oxide Substrates as Photocathode Candidates for Tandem Water Splitting Devices.” *The Journal of Physical Chemistry C* 122, no. 26 (2018): 14304.



# Publications

7. S. Hwang, S.H. Porter, A.B. Laursen, H. Yang, M. Li, V. Manichev, K.U.D. Calvinho, V. Amarasinghe, M. Greenblatt, E. Garfunkel, and G.C. Dismukes. “Creating stable interfaces between reactive materials: Titanium nitride protects photoabsorber-catalyst interface in water-splitting photocathodes.” *J. Mater. Chem. A* 7 (2019): 2400–2411.
8. D. Arifin and A.W. Weimer. “Kinetics and Mechanism of Solar-thermochemical H<sub>2</sub> and CO Production by Oxidation of Reduced CeO<sub>2</sub>.” *Solar Energy* 160 (2018): 178–185.  
<https://doi.org/10.1016/j.solener.2017.11.075>.
9. E. Liu, L. Jiao, H. Doan, Z. Liu, Y. Huang, K.M. Abraham, and S. Mukerjee. “Unifying Alkaline Hydrogen Evolution/Oxidation Reaction Kinetics by Identifying the Role of Hydroxy-Water-Cation Adducts.” *J. Amer. Chem. Soc.* 141 (2019): 3232–3239.
10. Q. Jia, E. Liu, L. Jiao, and S. Mukerjee. “Current Understanding of Sluggish Kinetics of Hydrogen Evolution and Oxidation reactions in Base.” *Current Opinion in Electrochemistry* (In Press).
11. J.Y. Jeon, S. Park, J. Han, S. Maurya, A.D. Mohanty, D. Tian, N. Saikia, M.A. Hickner, C.Y. Ryu, M.E. Tuckerman, S.J. Paddison, Y.S. Kim, and C. Bae. “Synthesis of Aromatic Anion Exchange Membranes by Friedel-Crafts Bromoalkylation and Cross-Linking of Polystyrene Block Copolymers.” *Macromolecules* (2019). <https://doi.org/10.1021/acs.macromol.8b02355>.
12. D. Li, I. Matanovic, A.S. Lee, E.J. Park, C. Fujimoto, H.T. Chung, and Y.S. Kim. “Phenyl Oxidation Impacts the Durability of Alkaline Membrane Water Electrolyzer,” *ACS Applied Materials and Interfaces* (2019). <https://doi.org/10.1021/acsami.9b00711>.



# Publications

13. F.A. Chowdhury, M.L. Trudeau, H. Guo and Z. Mi. “A Photochemical Diode Artificial Photosynthesis System for Unassisted High Efficiency Overall Pure Water Splitting.” *Nature Communications* 9 (2018): 1707. <https://doi.org/10.1038/s41467-018-04067DO>.
14. W. Wu, H. Ding, Y. Zhang, Y. Ding, P. Katiyar, P. Majumdar, T. He, and D. Ding. “3D Self-Architected Steam Electrode Enabled Efficient and Durable Hydrogen Production in A Proton Conducting Solid Oxide Electrolysis Cell at Temperatures Lower than 600°C.” *Advanced Science* 11 (2018): 1870166. Frontispiece Feature: <https://onlinelibrary.wiley.com/doi/10.1002/advs.201870070>.
15. R. Wang, C. Byrne, and M.C. Tucker. “Assessment of Co-Sintering as a Fabrication Approach for Metal-Supported Proton-Conducting Solid Oxide Cells.” *Solid State Ionics* 332 (2019): 25–33.
16. S.-L. Zhang, H. Wang, M.Y. Lu, A.-P. Zhang, L.V. Mogni, Q. Liu, C.-X. Li, C.-J. Li, and S.A. Barnett. “Cobalt-substituted SrTi<sub>0.3</sub>Fe<sub>0.7</sub>O<sub>3</sub>: a stable high-performance oxygen electrode material for intermediate-temperature solid oxide electrochemical cells.” *Energy and Environmental Science*, 11 (2018): 1870-1879. <https://doi.org/10.1039/C8EE00449H>.
17. D.R. Barcellos, M.D. Sanders, J. Tong, A.H. McDaniel, and R.P. O’Hayre. “BaCe<sub>0.25</sub>Mn<sub>0.75</sub>O<sub>3-δ</sub> — A Promising Perovskite-type Oxide for Solar Thermochemical Hydrogen Production.” *Energy and Environmental Science* 11 (2018): 3256-3265. <https://doi.org/10.1039/C8EE01989D>.
18. H.N. Dinh, R. Boardman, A.H. McDaniel, H. Colon-Mercado, T. Ogitsu, A.Z. Weber. “HydroGEN Overview: A Consortium on Advanced Water Splitting Materials (AWSM).” FY 2018 DOE Hydrogen and Fuel Cells Program Annual Progress Report.



# Publications

19. G.S. Gautam and E.A. Carter, “Evaluating transition metal oxides within DFT-SCAN and SCAN+U frameworks for solar thermochemical applications,” *Phys. Rev. Mater.* 2 (2018): 095401.
20. C. Bartel, A. Weimer, S. Lany, C. Musgrave, and A. Holder, “The role of decomposition reactions in assessing first-principles predictions of solid stability,” *npj Computational Materials* 5, no. 1 (2019): 4.
21. C. Bartel, C. Sutton, B. Goldsmith, R. Ouyang, C. Musgrave, L. Ghiringhelli, and M. Scheffler, “New tolerance factor to predict the stability of perovskite oxides and halides,” *Science Advances* 5, no. 2 (2019): eaav0693.
22. C. Bartel, S. Millican, A. Deml, J. Rumptz, W. Tumas, A. Weimer, S. Lany, V. Stevanovic, C. Musgrave, and A. Holder, “Physical descriptor for the Gibbs energy of inorganic crystalline solids and temperature-dependent materials chemistry,” *Nature Communications* 9, no. 1 (2018): 4168.
23. M. Carmo, C. Liu, G. Bender, A. Everwand, T. Lickert, T. Smolinka, D. Stolten, W. Lehnert, “Performance enhancement of PEM electrolyzers through Iridium-coated Titanium Porous Transport Layers,” *Electrochemistry Communications* 97 (2018): 96–99, <https://doi.org/10.1016/j.elecom.2018.10.021>.
24. G. Bender, M. Carmo, T. Smolinka, A. Gago, N. Danilovic, M. Mueller, F. Ganci, A. Fallisch, P. Lettenmeier, K.A. Friedrich, K. Ayers, B. Pivovar, J. Mergel, D. Stolten, “Initial Approaches in Benchmarking and Round Robin Testing for Proton Exchange Membrane Water Electrolyzers,” *International Journal of Hydrogen Energy* (accepted for publication).



# Presentations

1. **(Invited)** H.N. Dinh, A.H. McDaniel, A.Z. Weber, R. Boardman, T. Ogitsu, H. Colon-Mercado, “HydroGEN: A Consortium on Advanced Water Splitting Materials,” 9<sup>th</sup> IEA Annex 30 Electrolysis Workshop, NREL, Golden, CO, September 27, 2018.
2. **(Invited)** H.N. Dinh, A.H. McDaniel, A.Z. Weber, R. Boardman, T. Ogitsu, H. Colon-Mercado, “HydroGEN: A Consortium on Advanced Water Splitting Materials,” HydroGEN AWS Technology Pathways Benchmarking & Protocols Workshop, Arizona State University, Tempe, AZ, October 24, 2018
3. **(Invited)** H.N. Dinh, “FCTO’s HydroGEN AWSM Energy Materials Network Overview Webinar,” DOE Fuel Cell Technologies Office Webinar, February 7, 2019.
4. **(Invited)** H.N. Dinh, K. Randolph, A.Z. Weber, A.H. McDaniel, R. Boardman, T. Ogitsu, D.L. Anton, D. Peterson, E.L. Miller, “HydroGEN Overview, Projects, and the AWSM Node Capabilities,” Symposium ES11–Advanced Low Temperature Water Splitting for Renewable HydroGEN Production via Electrochemical and Photoelectrochemical Processes, Spring MRS Meeting, Phoenix, AZ, April 24, 2019.
5. **(Invited)** H.N. Dinh, D. Peterson, K. Randolph, A. Z. Weber, A.H. McDaniel, R. Boardman, T. Ogitsu, D.L. Anton, “HydroGEN Overview and AWSM Electrolysis Project Updates,” 235<sup>th</sup> ECS Meeting, Dallas, TX, May 24, 2019.



# Presentations

6. K.E. Ayers, C. Capuano, P. Mani, “High Efficiency PEM Electrolysis: Potential for H<sub>2</sub>@Scale,” 234<sup>th</sup> ECS Meeting, Cancun, Mexico, October 2, 2018.
7. R. Wang, M. Tucker, “Development of High Performance Metal-Supported Solid Oxide Electrolysis Cells,” 234<sup>th</sup> ECS Meeting (AiMES 2018), Cancun, Mexico, October 2018.
8. **(Invited)** S. Barnett, “High-Efficiency Electrical Energy Storage Using Reversible Solid Oxide Cells,” Boston University Materials Day—Materials for Electrochemical Energy Conversion & Storage, Boston, MA, October 26, 2018.
9. **(Invited)** N. Gaillard, A.D. DeAngelis, K. Horsley, “Wide Bandgap Copper Chalcopyrite Candidates for Renewable Hydrogen Generation,” 233<sup>rd</sup> ECS Meeting, Symposium I05, 1884, Seattle, WA, 2018.
10. **(Invited)** T. Ogitsu, J. Varley, A. Deangelis, K. Horsley, N. Gaillard, “Integrating Ab-Initio Simulations and Experimental Characterization Methods: Towards Accelerated Chalcopyrite Materials Development for Hydrogen Production,” 233<sup>rd</sup> ECS Meeting, Symposium I05, 1855, Seattle, WA, 2018.
11. K. Horlsey, A. Deangelis, N. Gaillard, “Cu(In,Ga)S<sub>2</sub> Photocathodes with Optical Bandgap Over 1.7 eV for Photoelectrochemical Water Splitting,” MRS Spring Meeting, Symposium EN18, EN18.15.05, Phoenix, AZ, 2018.
12. A. Deangelis, K. Horlsey, N. Gaillard, “Wide-Bandgap CuGa(S,Se)<sub>2</sub> As Top Cell Photocathodes for Tandem Water Splitting Devices,” 233<sup>rd</sup> ECS Meeting, Symposium I05, 1929, Seattle, WA, 2018.



# Presentations

13. S. Mukerjee and J. Qingying, “Fundamental aspects of regenerative hydrogen electrocatalysis in alkaline pH.” In *Abstracts of Papers of the American Chemical Society* 256 (2018).
14. I. Kendrick, M. Bates, Q. Jia, H. Doan, W. Liang, S. Mukerjee, “Tuning Ni Surfaces for Enhanced Oxygen Evolution Reaction.” In *Meeting Abstracts* 29 (The Electrochemical Society, 2018): 1702–1702.
15. D. Li, I. Matanovic, A.S. Lee, E.J. Park, C. Fujimoto, H.T. Chung, Y.S. Kim, “Phenyl Oxidation at Oxygen Evolution Potentials,” *Polymers for Fuel Cells, Energy Storage and Conversion*, Asilomar Conference Ground, Pacific Grove, CA, USA, Feb. 24–27, 2019.
16. E.J. Park, S. Maurya, M.R. Hibbs, C.H. Fujimoto, Y.S. Kim, “Caveat of High Temperature Accelerated Stability Test of Anion Exchange Membrane,” *Polymers for Fuel Cells, Energy Storage and Conversion*, Asilomar Conference Ground, Pacific Grove, CA, USA, Feb. 24–27, 2019.
17. H.T. Chung, “Carbon-free Perovskite Oxide OER Catalysts for AEM Electrolyzer,” Abstract number IO3-1687, 233<sup>rd</sup> ECS Meeting, Seattle, WA, May 13–17, 2018.
18. **(Invited)** L. Chong, H. Wang, D.-J. Liu, “PGM-free OER Catalysts for PEM Electrolyzer Application,” 235<sup>th</sup> ECS Meeting, Dallas, TX, May 26–31, 2019.
19. A.S. Lee, Y.S. Kim, P. Zelenay, C. Fujimoto, L.-W. Wang, G. Teeter, G. Bender, D.-J. Liu, G. Wu, H. Xu, “PGM-free OER Catalysts for PEM Electrolyzer,” Hydrogen Production Tech Team Meeting, Berkeley, CA, April 10, 2018.



# Presentations

20. G.C. Dismukes, “Bioinspired heterogeneous electrocatalysts for CO<sub>2</sub> reduction and water splitting: Energy-efficient C-C coupling rivaling enzymes,” Leiden University, Institute of Chemistry, Leiden, the Netherlands, Danish Technical University, Institute of Physics, Copenhagen, DK, Aarhus University, iNano, Aarhus, DK, October 2018.
21. E. Garfunkel, “Photoelectrochemical water splitting to form hydrogen,” 2018 Telluride Semiconductor Surface Chemistry Meeting, Telluride CO.
22. X. (Jessica) Luo, A. Kusoglu, “Structure-Transport Relationships of Anion Exchange Ionomers,” ACS Division of Polymer Chemistry, Pacific Grove, CA, 2019 (Poster).
23. A.Z. Weber, “Exploring Polymer-Electrolyte Fuel Cells using Physics-Based Modeling,” Max Planck Institute for Dynamics of Complex Technical Systems, Magdeburg, Germany, December 2018.
24. J.C. Fornaciari, J. Zhou, D. Primc, A.T. Bell, A.Z. Weber, “Electrocatalyst performance for selective methanol formation in methane electrolyzer,” ECS Fall Meeting, Cancun, Mexico (poster).
25. R. Wang, C. Byrne, M.C. Tucker, “Proton-Conducting Ceramics for Metal-Supported Solid Oxide Cells,” 19th International Conference on Solid State Protonic Conductors, Stowe, VT, Sept. 21, 2018.
26. D. Ding, W. Wu, H. Ding, T. He, “Development of Proton-Conducting Solid Oxide Electrolysis Cells at Intermediate Temperatures at Idaho National Laboratory,” 234<sup>th</sup> ECS Meeting (AiMES 2018), Cancun, Mexico, Sept. 30–Oct. 4, 2018.



# Presentations

27. D. Ding. “Development of Electrochemical Processing and Electrocatalysis at Intermediate Temperatures at Idaho National Laboratory (INL),” Invited Lecture for faculty and graduate students, Department of Mechanical and Aerospace, West Virginia University, Morgantown, WV, USA, Dec. 4, 2018.
28. B. Hu, M. Reiser, A. Aphale, S. Belko, O. Marina, J. Stevenson, D. Ding, P. Singh, “Barium Zirconate Based Electrolyte Densification Using Reactive Sintering Aids,” 43rd International Conference and Exposition on Advanced Ceramics and Composites (ICACC 2019), Daytona Beach, FL, USA, Jan 27–Feb 1, 2019.
29. H. Ding, W. Wu, D. Ding, “A Novel Triple Conducting Electrode for Fast Hydrogen Production in Protonic Ceramic Electrochemical Cells (H-SOECs).” 43rd International Conference and Exposition on Advanced Ceramics and Composites (ICACC 2019), Daytona Beach, FL, USA, Jan 27–Feb 1, 2019.
30. D. Ding, “Advancement of Intermediate Temperature Solid Oxide Energy Conversion Technologies at Idaho National Laboratory.” Invited Lecture for faculty and graduate students, Department of Chemical Engineering, University of Louisiana at Lafayette, Lafayette, LA, USA, April 1, 2019.
31. B. Hu, O.A. Marina, A.N. Aphale, D. Ding, H. Ding, A. Zakutayev, J. Stevenson, P. Singh, “Stable Proton-conducting Solid Oxide Electrolysis Cells for Pure Hydrogen Production at Intermediate Temperatures,” 2019 MRS Spring Symposia on Advanced Water Splitting, Phoenix, AZ, April 22–26, 2019.



# Presentations

32. D. Ding, “Advancement of reversible proton-conducting solid oxide cells at Idaho National Laboratory (INL),” 2<sup>nd</sup> International Conference on Electrolysis (ICE 2019), Loen, Norway, June 9–13, 2019.
33. C. Musgrave, C. Bartel, A. Holder, C. Sutton, B. Goldsmith, R. Ouyang, L. Ghiringhelli, M. Scheffler, “Ab Initio and Machine Learned Modeling for the Design and Discovery of New Materials for Energy Applications,” Air Force Research Laboratories, Dayton, OH, January 2019.
34. S. Millican, I. Androschuk, A. Weimer, C. Musgrave, “Computational discovery of materials for solar thermochemical hydrogen production” American Institute of Chemical Engineers, October 2018.
35. C. Bartel, C. Sutton, B. Goldsmith, R. Ouyang, C. Musgrave, L. Ghiringhelli, M. Scheffler, “New tolerance factor to predict the stability of perovskite oxides and halides,” American Institute of Chemical Engineers, October 2018.
36. C. Bartel, C. Sutton, B. Goldsmith, R. Ouyang, C. Musgrave, L. Ghiringhelli, M. Scheffler, “New tolerance factor to predict the stability of perovskite oxides and halides,” European Materials Research Society, September 2018.
37. S. Millican, I. Androschuk, A. Weimer, C. Musgrave. “Rapid Kinetic Profiling of Bulk Diffusion Barriers for Solar Thermal Water Splitting Materials,” 21st International Conference on Ternary and Multinary Compounds, September 2018.



# Presentations

38. C. Bartel, C. Sutton, B. Goldsmith, R. Ouyang, C. Musgrave, L. Ghiringhelli, M. Scheffler, “New tolerance factor to predict the stability of perovskite oxides and halides,” 21st International Conference on Ternary and Multinary Compounds, September 2018.
39. C. Bartel, C. Sutton, B. Goldsmith, R. Ouyang, C. Musgrave, L. Ghiringhelli, M. Scheffler, “New tolerance factor to predict the stability of perovskite oxides and halides,” Application of Machine Learning and Data Analytics for Energy Materials Network Consortia 2018, May 2018.
40. S.L. Millican, I. Androshchuk, A.W. Weimer, C.B. Musgrave, “Ab-initio Modeling and Experimental Demonstration of Metal Oxides for Solar Thermochemical Water Splitting,” American Chemical Society Spring Meeting, March 2018.
41. C. Bartel, C. Sutton, B. Goldsmith, R. Ouyang, C. Musgrave, L. Ghiringhelli, M. Scheffler, “Improved tolerance factor for classifying the formability of perovskite oxides and halides,” American Physical Society Annual Meeting, March 2018.
42. C. Bartel, S. Millican, A. Deml, J. Rumpitz, W. Tumas, A. Weimer, S. Lany, V. Stevanovic, C. Musgrave, A. Holder, “Machine learning the Gibbs energies of inorganic crystalline solids,” American Physical Society Annual Meeting, March 2018.
43. Millican, S.L., I. Androshchuk, A.W. Weimer, and C.B. Musgrave, “Design and Discovery of Mixed Metal Oxides for Solar Thermochemical Water Splitting,” International Conference and Exposition on Advanced Ceramics and Composites, January 2018.



# Presentations

44. G. Bender, M. Carmo, S. Fischer, T. Lickert, T. Smolinka, J. Young, “IEA Annex 30 Overview,” Invited talk at Proton 2B Workshop: HydroGEN Benchmarking & Protocols Workshop, October 24, 2018, Tempe Arizona.
45. J.L. Young, F. Ganci, S. Madachy, S. Fischer, M. Carmo, G. Bender, “PEM Electrolyzer Characterization and Limitations When Using Carbon-Based Hardware and Material Sets,” Accepted ECS talk, 235th ECS Meeting, Dallas, TX, USA, May 26-31, 2019.
46. G. Bender, M. Carmo, S. Fischer, T. Lickert, T. Smolinka, J. Young, “Round Robin Testing for Polymer Electrolyte Membrane Water Electrolysis - Phase 2,” World Hydrogen Energy Conference, Rio de Janeiro, June 19, 2018.
47. G. Bender, H.N. Dinh, N. Danilovic, A. Weber, “HydroGEN: Low-Temperature Electrolysis,” Hydrogen and Fuel Cells Program 2018 AMR, Washington, DC, June 13, 2018.
48. J. Young, A. Kallen, E. Klein, S. Alia, G. Bender, “IEA Annex 30 Workshop – NREL LTE Activities Overview,” Invited Talk, IEA Annex 30 Workshop, September 2018, Golden, Colorado, USA.
49. K. Ayers, A. Motz, C. Capuano, P. Mani, “Development of Protocols and Standards for Low Temperature Electrolysis,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
50. C. Xiang, “Development of Protocols and Standards for Photoelectrochemical Water-Splitting,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.



# Presentations

51. O. Marina, “Framework and Test Protocols for High Temperature Electrolysis,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
52. E. Stechel, “Framework and Test Protocols for Solar Thermochemical Water Splitting,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
53. G. Dismukes, A.B. Laursen, S. Hwang, E. Garfunkel, T. Deutsch, D. Friedman, M. Steiner, “Electrochemical and Photoelectrochemical Water Splitting Using Bioinspired Catalysts That Out-Perform Nobel Metals,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
54. N. Danilovic, F. Houle, A. Kusoglu, F.M. Toma, M. Tucker, L.-W. Wang, A. Weber, “Low and High-Temperature Electrolysis, Photoelectrochemical and Solar Thermochemical Water Splitting Materials Characterization and Development at Berkeley Lab Under the HydroGEN Consortium,” Poster presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
55. I. Khan, K. Heinselman, C. Muzzillo, J. Young, T. Deutsch, A. Zakutayev, N. Gaillard, “CuGa<sub>3</sub>Se<sub>5</sub>/Zn<sub>1-x</sub>Mg<sub>x</sub>O Photocathodes for Photoelectrochemical Water Splitting,” Poster presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
56. T. Jaramillo, “Development of Catalytic Coatings for H<sub>2</sub>-Producing Photocathodes in Solar Water-Splitting,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.



# Presentations

57. N. Gaillard, “Wide Bandgap Chalcopyrites for Photoelectrochemical Water Splitting,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
58. J. Young, H. Doescher, J. Geisz, J. Turner, T. Deutsch, “Solar-to-Hydrogen Efficiency—Shining Light on Photoelectrochemical Device Performance,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
59. T. Deutsch, J. Young, C. Aldridge, C. Barraugh, M. Steiner, “Photoelectrochemical Water Splitting Durability Testing—What Can Half-Cell Results Can Tell Us About Full-Cell Performance?” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
60. W. Drisdell, A. Landers, M. Farmand, “Operando Synchrotron Characterization of Electrochemical Interfaces,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
61. N. Danilovic, T. Kistler, S. Alia, P. Agbo, A. Weber, “Benchmarking Water-Splitting Materials at the Intersection of Electrocatalysis and Photoelectrochemistry,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
62. G. Bender, S. Alia, M. Ulsh, S. Mauger, B. Pivovar, H. Dinh, A. Weber, N. Danilovic, A. Kusoglu, H. Colon-Mercado, “HydroGEN Supernode—Linking Low Temperature Electrolysis (LTE)/Hybrid Materials to Electrode Properties to Performance,” Poster presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.



# Presentations

63. J. Young, N. Danilovic, M. Steiner, F.M. Toma, G. Saur, J. Vidal, H. Breunig, D. Friedman, A. Weber, T. Deutsch, “HydroGEN PEC Supernode—Emergent Degradation Mechanisms with Integration and Scale Up of PEC Devices,” Poster presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
64. T. Ogitsu, J. Varley, A. Sharan, A. Janotti, N. Gaillard, A. DeAngelis, “Chalcopyrite Alloy Materials for PEC H<sub>2</sub> Production—Development of Theoretical Synthesis Support System for HydroGEN,” Poster presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
65. T. Ogitsu, B. Wood, T.A. Pham, J. Varley, J. Lee, M. Biener, C. Orme, Y. Han, “Photoelectrochemical and Low Temperature Water Splitting Materials Research at Lawrence Livermore National Laboratory Under HydroGEN Consortium,” Poster presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
66. R. Bell, D. Ginley, P. Parilla, S. Lany, E. Coker, A. Zakutayev, A. McDaniel, “Design, Synthesis, and Characterization of High Quality STCH Materials,” Poster presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
67. A. McDaniel, A. Ambrosini, E. Coker, J. Sugar, R. Bell, D. Ginley, S. Lany, P. Parilla, T. Ogitsu, S. Wan, B. Wood, “Developing an Atomistic Understanding of the Layered Perovskite Ba<sub>4</sub>CeMn<sub>3</sub>O<sub>12</sub> and Its Polytypes for Thermochemical Water Splitting—A HydroGEN Supernode,” Poster presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.



# Presentations

68. O. Marina, K. Meinhardt, G. Whyatt, D. Reed, K. Recknagle, B. Koepfel, J. Holladay, “High Temperature Electrolysis Capabilities at PNNL—Materials Development, Cell/Stack Manufacturing, Testing, Characterization and Modeling,” Poster presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
69. M. Gerhardt, L. Stanislaw, A. Weber, “Modeling of Anion-Exchange Membrane Electrolyzers to Guide Materials Development,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
70. S. Alia, S. Ghoshal, G. Anderson, M.-A. Ha, S. Stariha, C. Ngo, S. Pylypenko, R. Borup, R. Larsen, “Effects of Low Loading and Intermittency on Low Temperature Electrolysis from a Catalyst Perspective,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
71. H. Xu, “Alkaline Membrane Electrolysis—Challenges and Perspectives,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
72. Z. Mi, S. Vanka, “Monolithically Integrated InGaN/Si Tandem Photoelectrodes for Efficient and Stable Photoelectrochemical Water Splitting,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
73. H. Lim, J. Young, J. Geisz, D. Friedman, T. Deutsch, J. Yoon, “Surface-Tailored GaInP<sub>2</sub> Photocathodes for High Performance Solar Water Splitting,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.



# Presentations

74. T.A. Pham, Z. Mi, T. Ogitsu, “Probing the Surface Chemistry and Stability of III-V Photoelectrodes with First-Principles Simulations and In Situ Experiments,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
75. C. Musgrave, C. Bartel, S. Millican, B. Goldsmith, C. Sutton, S. Lany, A. Weimer, A. Holder, “Ab Initio and Machine Learned Modeling to Screen and Discover Materials for Solar Thermal Water Splitting,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
76. R. Bell, P. Parilla, E. Coker, E. Stechel, D. Ginley, “Developing Standard Materials for Solar Thermochemical Water Splitting Calibration,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
77. R. Bell, X. Qian, E. Coker, M. Rodriguez, P. Parilla, S. Haile, D. Ginley, “In-Situ Defect Mapping of High Temperature STCH Materials in Oxidizing and Reducing Environments,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
78. R. O’Hayre, M. Sanders, D. Barcellos, C. Duan, J. Huang, M. Papac, V. Stevanovic, N. Kumar, A. Zakutayev, S. Lany, A. Emery, C. Wolverton, C. Borg, A. McDaniel, “The ‘Perovskite Playground’—Engineering Defect Chemistry in Doped Perovskite and Perovskite-Related Oxides for High Temperature Redox-Active Chemical and Electrochemical Applications,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
79. C. Wolverton, “Oxygen Off-Stoichiometry and Defect Entropies in Solar Thermochemical Water Splitting Materials,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.



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80. S. Lany, “The Electronic Entropy of Charged Defect Formation and Its Impact on Thermochemical Redox Cycles,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
81. M. Sanders, A. Bergeson-Keller, N. Kumar, J. Pan, D. Barcellos, V. Stevanovic, S. Lany, R. O’Hayre, “The Effect of Structure on Oxygen Vacancy Formation Energy in Ce-Substituted Sr-Mn Oxides,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
82. S. Haile, X. Gian, “Thermochemical Trends in ABO<sub>3</sub>-Type Compounds for Solar Fuel Generation,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
83. O. Marina, C. Coyle, D. Edwards, J. Stevenson, “Durability Assessment of High Temperature Electrolysis Cells,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
84. A. Bergeson-Keller, D. Barcellos, M. Sanders, R. O’Hayre, “Study of the Reduction Thermodynamics of Sr<sub>1-x</sub>Ce<sub>x</sub>MnO<sub>3</sub> Perovskites for Solar Thermochemical Hydrogen Production,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
85. J. Miller, I. Ermanoski, A. Ambrosini, E. Stechel, “Ammonia Synthesis in Two Cyclic Steps—Basic Thermodynamic Considerations,” Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.



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86. L. Zhu, C. Cadigan, C. Duan, J. Huang, L. Bian, L. Le, N. Sullivan, R. O'Hayre, "High-Performance Reversible Proton-Conducting Ceramic Cells for Power Generation and Energy Storage Through Ammonia," Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
87. E. Stechel, I. Ermanoski, J. Miller, "Materials Thermodynamic Limits in Solar–Thermochemical Fuel Production," Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.
88. I. Ermanoski, E. Stechel, J. Miller, "Thermal–Driven Oxygen Pumping in Thermochemical Fuel Production," Oral presentation at the 2019 MRS Spring Meeting and Exhibit, Phoenix, AZ, April 22–26, 2019.