

2020 Hydrogen and Fuel Cell Annual Merit Review Meeting

# Ionomer Dispersion Impact on Advanced Fuel Cell and Electrolyzer Performance and Durability

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Newton, MA

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*Project ID*  
FC117

# Project Overview

## Timeline

- Project Start Date: 8/27/2018
- Project End Date: 2/26/2021

## Budget

- Total Project Value
  - Phase IIB: \$1 Million
  - Spent: \$734 K

## Barriers Addressed

- PEM fuel cell and electrolyzer performance and durability

## Contributors

- Giner: Natalie Macauley, Shirley Zhong, Magali Spinetta, and Fan Yang
- LANL: Dr. Yu-Seung Kim (sub.)
- NREL: Scott Mauger (sub.)
- UConn: Jasna Jankovic (collaborator)

## Technical Targets

- Elucidate how ionomer dispersions impact electrode structures and performance
- Create fuel cell MEAs that are mechanically and chemically stable
- Establish catalyst ink property- electrode structure-MEA performance correlation
- Develop processable and scalable MEA fabrication platforms

## Project Nature

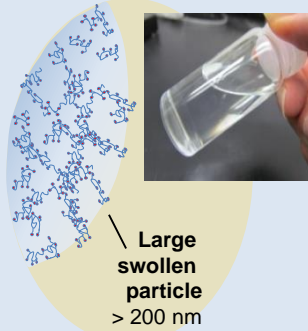
- DOE Technology Transfer Opportunity Project (SBIR-TTO)

# Relevance: Ionomer Dispersion Technology



## Conventional Ionomer Dispersion

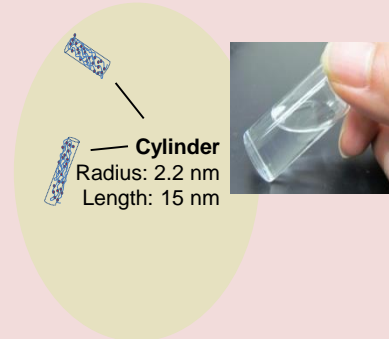
Dupont  
European Patent 0066369



- Water based **multiple** solvent system
- **Expensive** processing: requires high temperature ( $> 200^{\circ}\text{C}$ ) & pressure ( $> 1000$  psi)
- **Large** and **non-uniform** particle suspension: particle size (hydrodynamic radius: 200 – 400 nm)
- Produces **brittle** membrane: toughness  $\sim 0.001$  MPa
- Produces **less stable** electrode: cell voltage loss after durability test: 40-90 mV

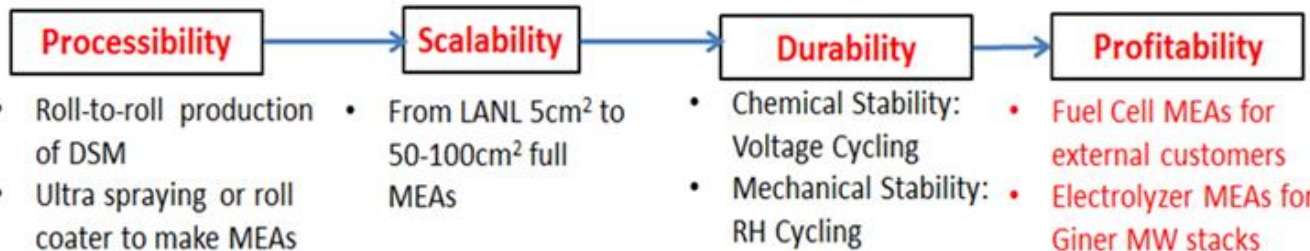
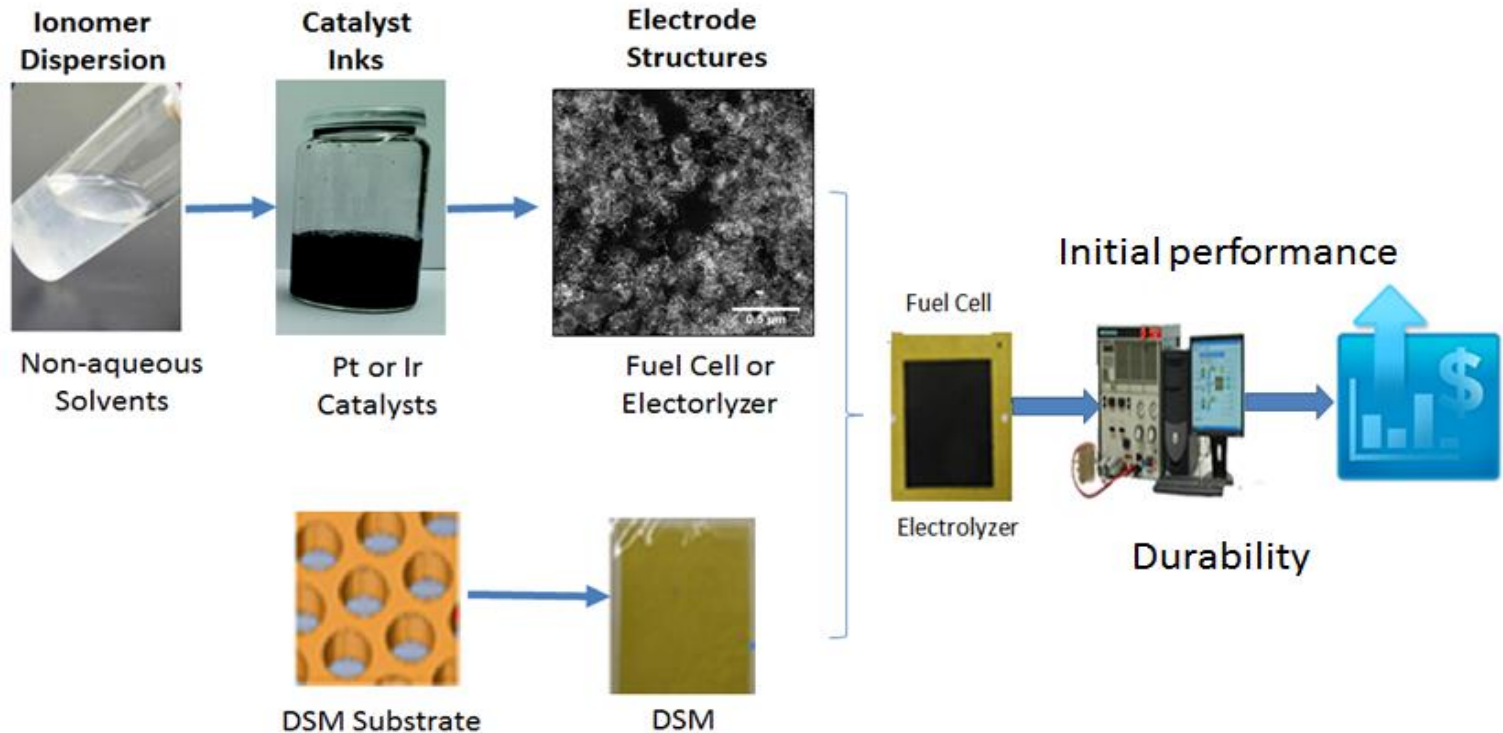
## LANL Ionomer Dispersion

LANL  
US Patent 7981319,  
8236207,  
8394298

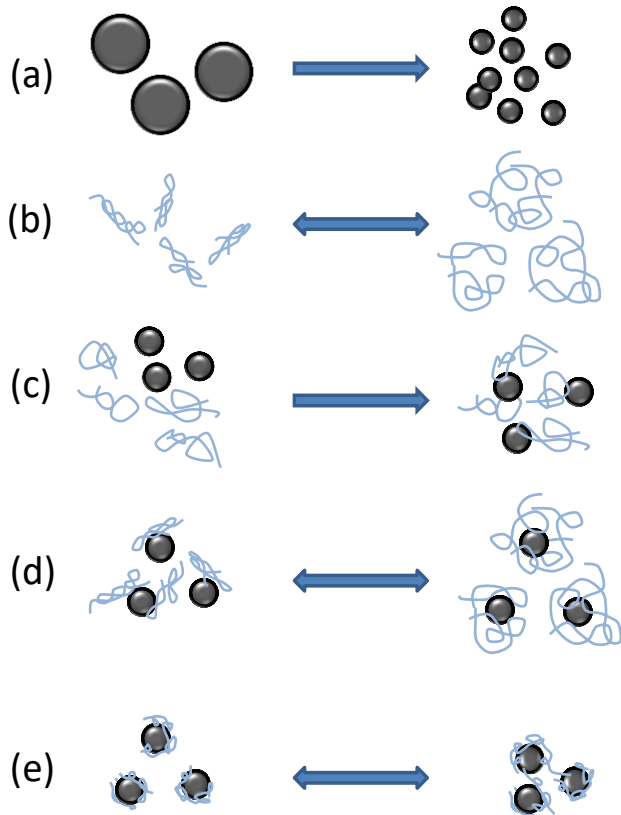


- **Single** solvent system
- **Cost effective** processing: requires lower temperature ( $< 120^{\circ}\text{C}$ ) & ambient pressure
- **Small** and uniform particle suspension: particle size (2.2 x 15 nm cylinder)
- Produces **tough** membrane: toughness 10 MPa ( $> 4$  orders of magnitude difference!!)
- Produces **stable** electrode: cell voltage loss after durability test: 0 mV

# Technical Approaches



# Background: Pt/C and Ionomer Interaction



- (a) Breakdown of core catalyst agglomeration
- (b) Ionomer re-conformation in various solvent blend
- (c) Ionomer adsorption onto catalyst particle surface
- (d) Ionomer re-conformation on particle surface
- (e) Formation and breaking-up of flocculation

# Phase IIB Project



- ❑ Correlate catalyst ink properties with electrode structure and fuel cell and electrolyzer performance
- ❑ Identify MEA improvement pathways toward roll-to-roll manufacturing methods and full MEA commercialization
  - Ink characterization: Rheology, Zeta potential, Particle size analysis
  - MEA Performance and Durability
  - Microstructure characterization: SEM & TEM
  - **Commercialization via Roll to Roll production**
    - TSA with NREL
    - Fuel cell GDE
    - Electrolyzer decal

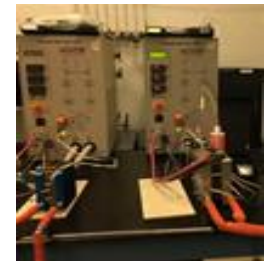
**Rheometer:  
Catalyst  
Inks**



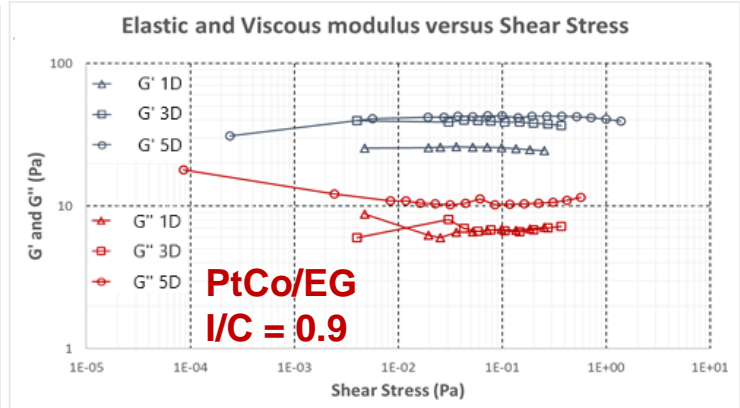
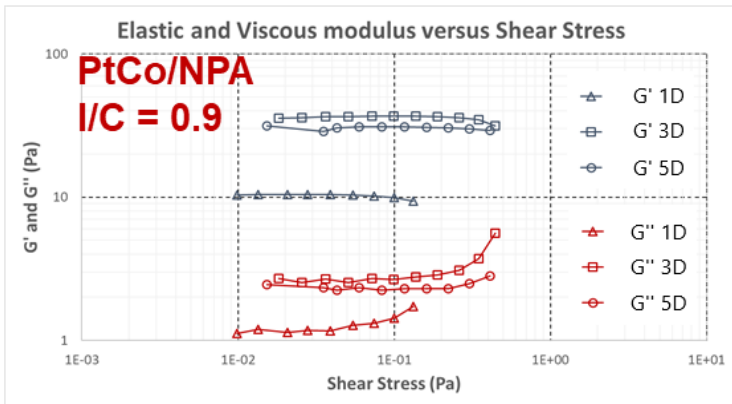
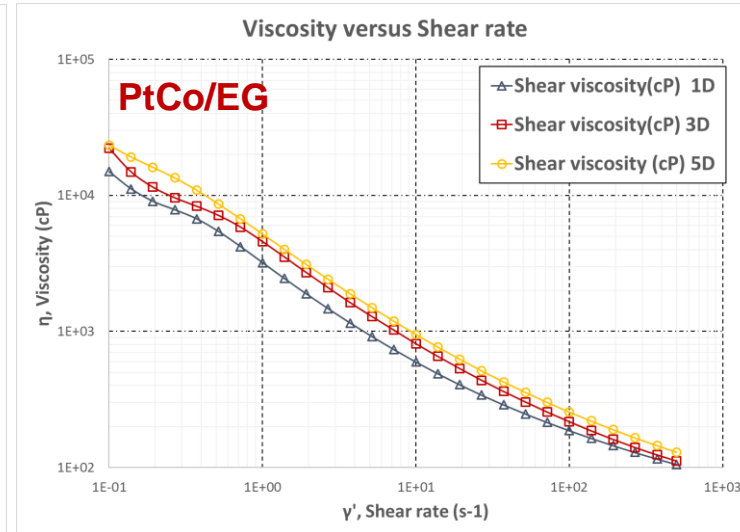
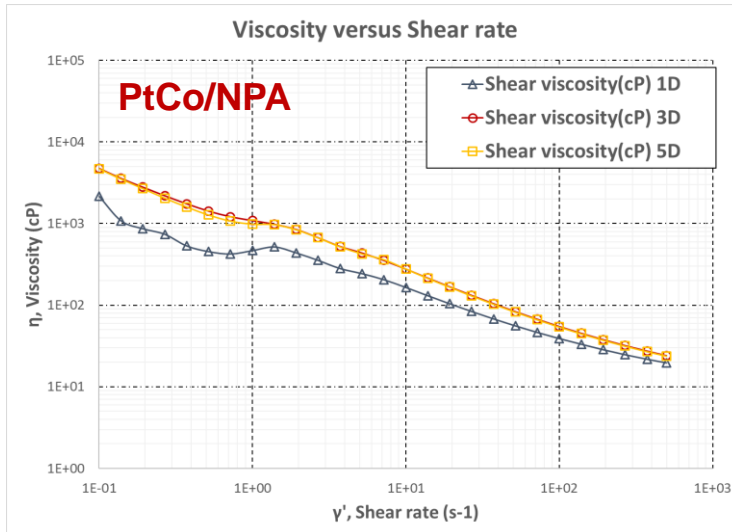
**Microscopy:  
Electrode  
Structures**



**Fuel Cell and  
Electrolyzer  
Performance**



# Accomplishment: Rheology on Mixing PtCo Ink



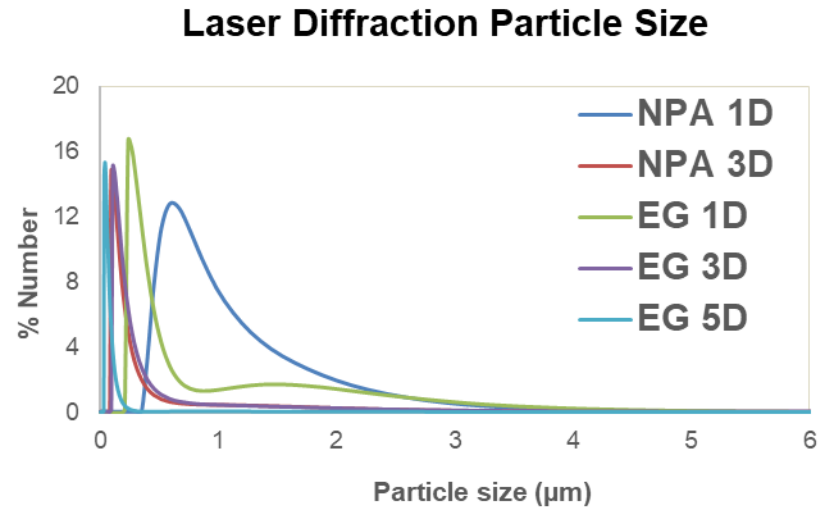
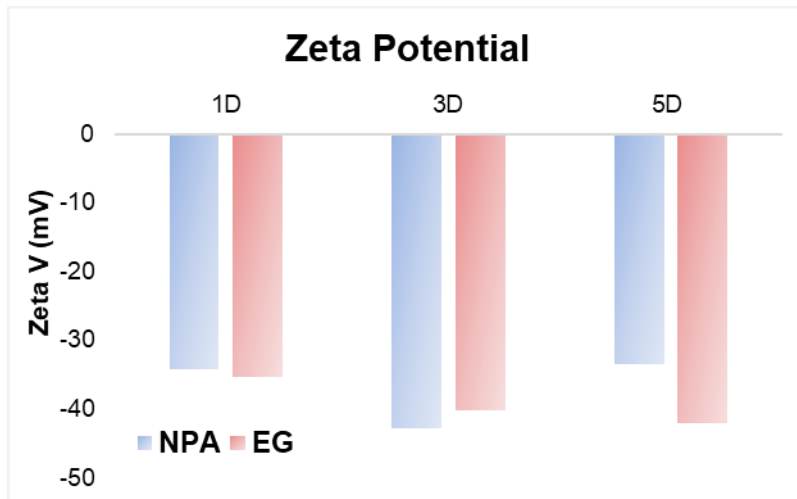
□ **Mixing time determined by stable viscosity**

**3 days for NPA and 5 days for EG**

- Stable viscosity and highest elastic modulus
- Elastic dominant inks

**G'** – Elastic modulus  
**G''** – Viscous modulus

## Mixing PtCo in NPA vs EG



### □ Zeta Potential

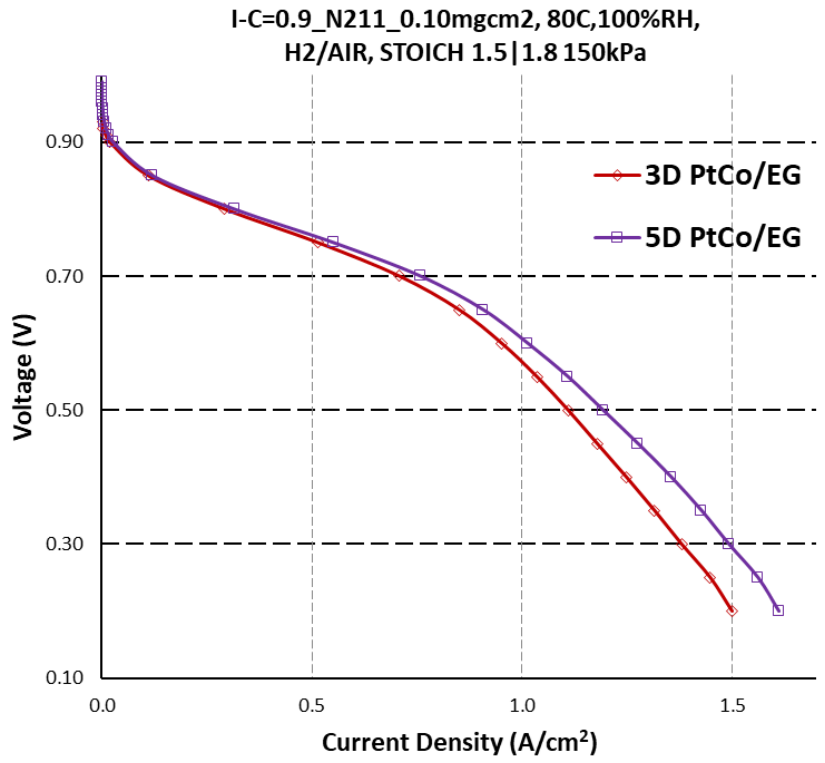
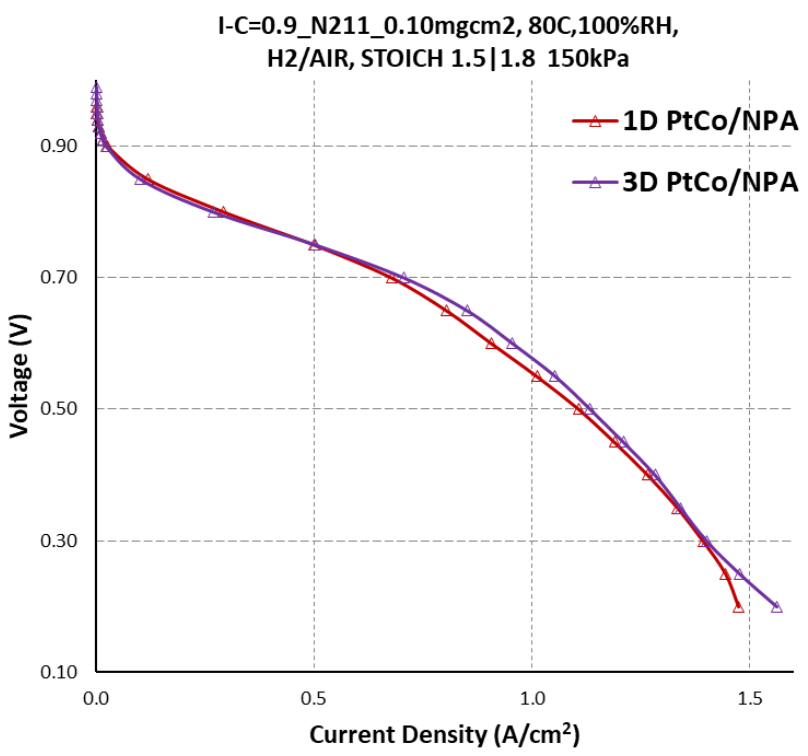
- Gradually decreases with mixing time for EG ink
- NPA ink has lowest value after 3 days

### □ Laser Diffraction Analysis:

- Carbon agglomerate size decreases with mixing time for EG ink
- Dx(50) NPA: 0.789 µm to 0.158 µm after 1 and 3 days
- Dx(50) EG: 0.369 µm, 0.178µm and 0.063 µm after 1,3 and 5 days



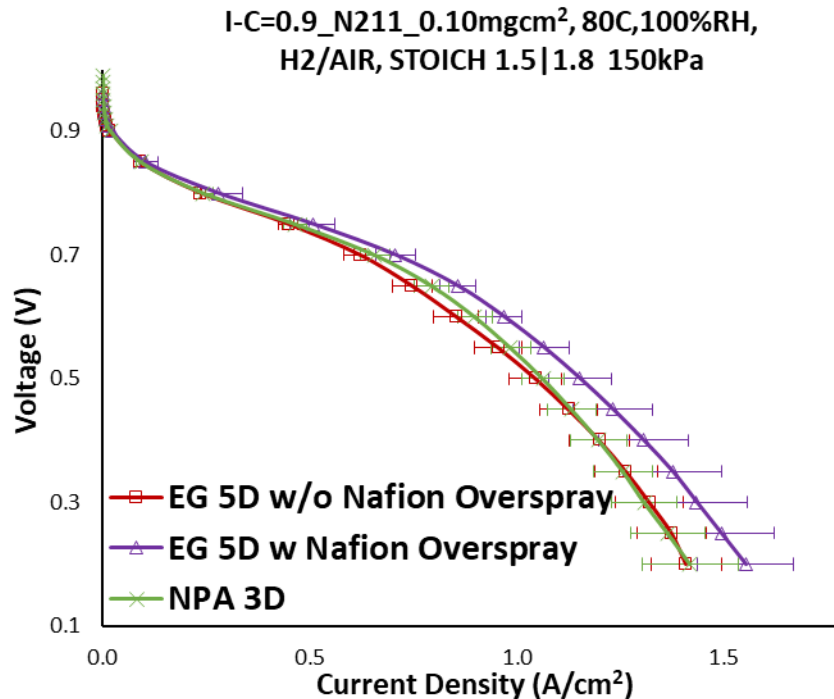
# Accomplishment: NPA vs EG Performance



**NPA:** 3 day mixing shows slightly better performance

**EG:** 5 day mixing shows better performance

# Decal Performance



## □ BOL Performance for Stoichiometric Flows

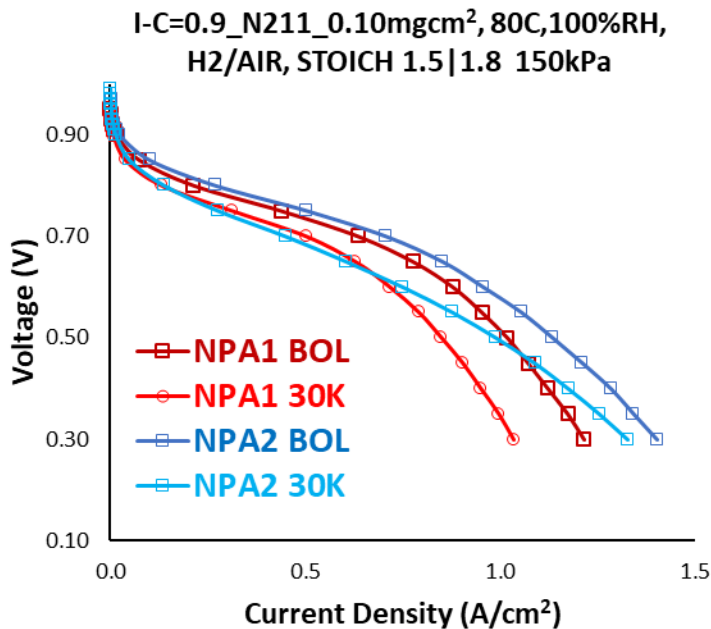
- EG MEAs with Nafion overspray performs better than NPA
- EG MEAs without overspray performs similar on average to NPA MEAs
- EG MEA has lower initial sheet resistance than NPA MEA

# Durability of PtCo in EG vs NPA 30K SW AST

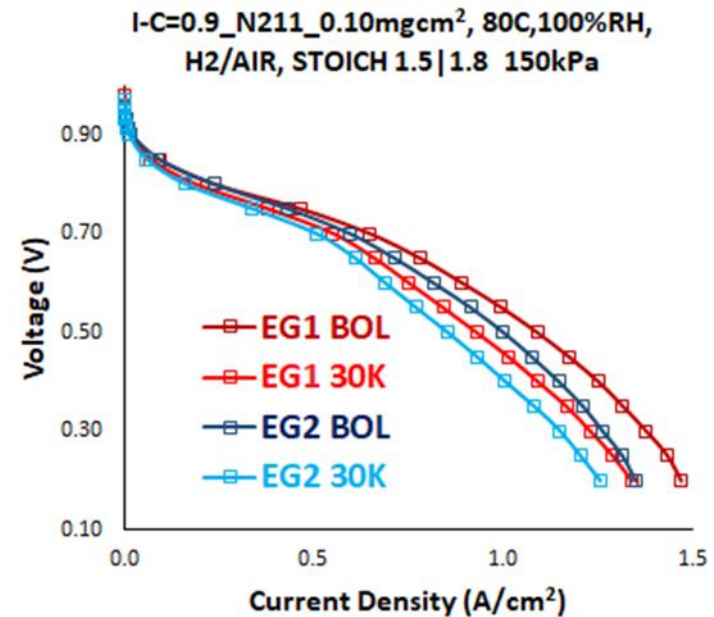


30,000 cycle 0.6-0.95 V Square Wave Accelerated Stress Test (30K SW AST)

## NPA



## EG

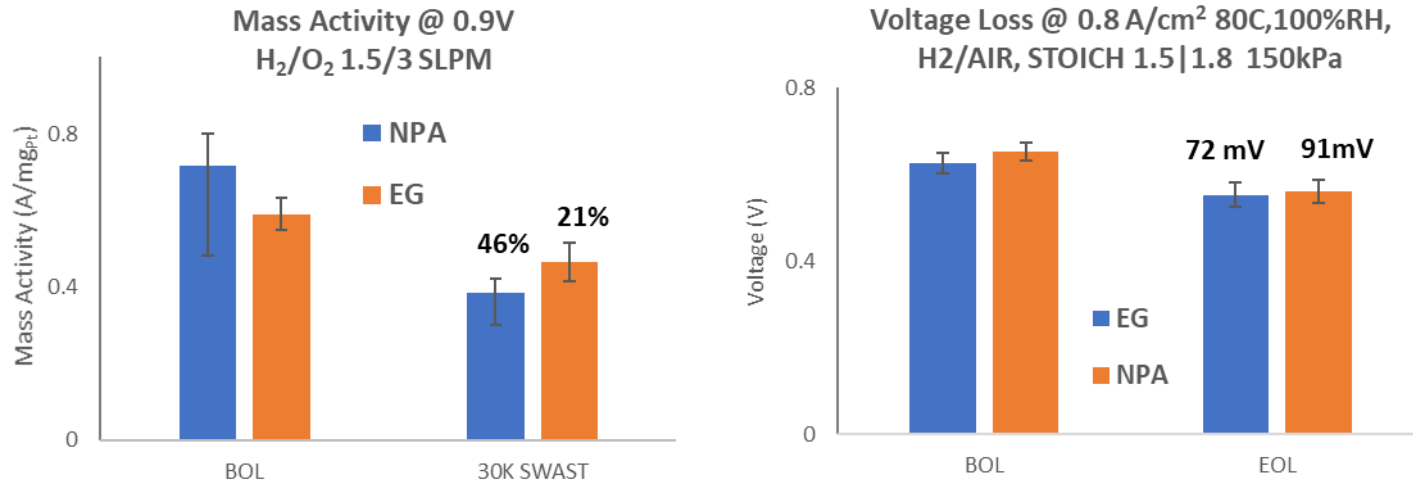


- NPA MEAs degrade faster than EG MEAs without overspray
  - NPA MEAs degrade in both regions
  - EG MEAs degrade more in high current density region

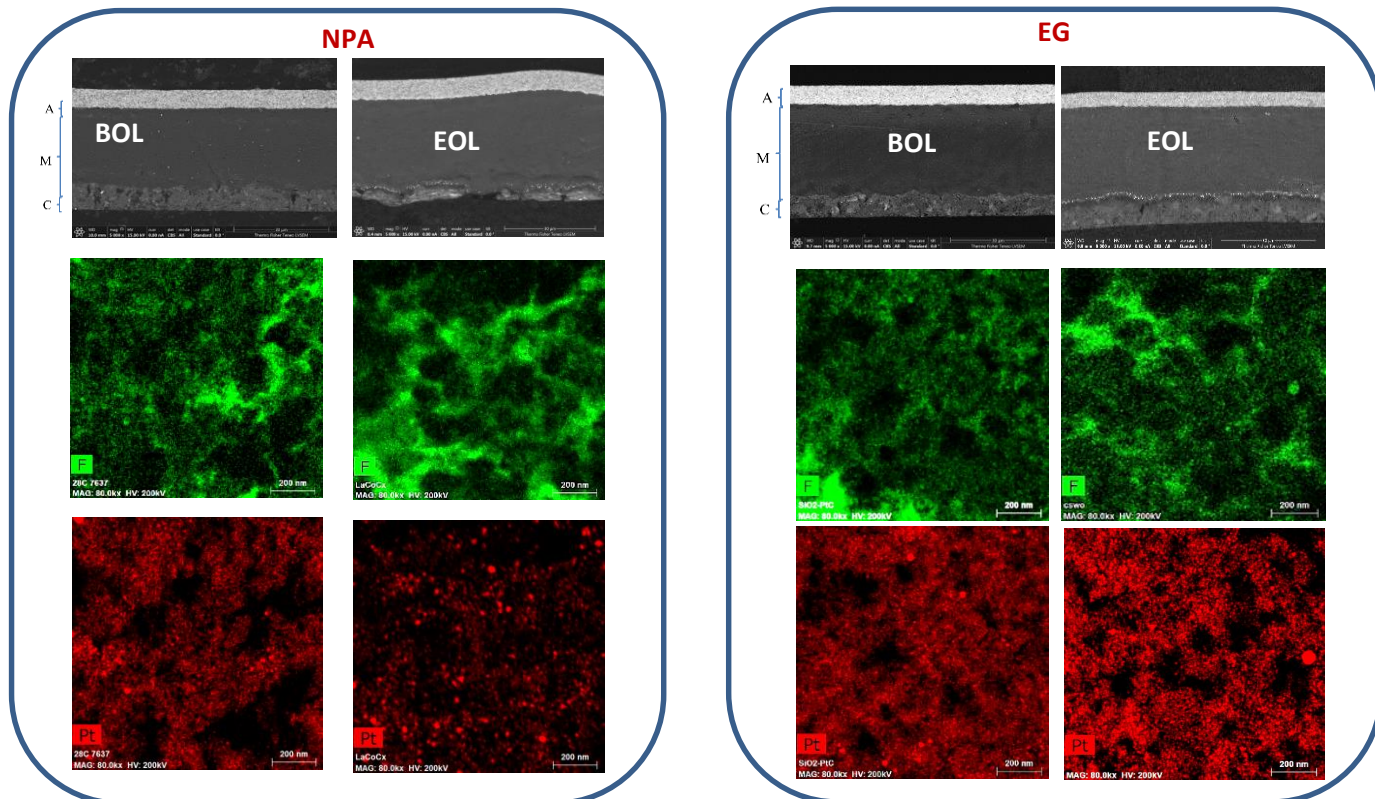
# Durability of PtCo in EG vs NPA 30K SW AST



30,000 cycle 0.6-0.95 V Square Wave Accelerated Stress Test (30K SW AST)



- ❑ EG with Nafion overspray lost 40% of performance after 30k SWAST similar to NPA durability – testing methanol based Nafion overspray
- ❑ EG without Nafion overspray meets DOE mass activity durability target of a loss < 40%
- ❑ Sheet resistance of the NPA cathode increases from 0.135  $\Omega\cdot\text{cm}^2$  to 0.162  $\Omega\cdot\text{cm}^2$
- ❑ EG cathode has lower sheet resistance than NPA cathode: increases from 0.088  $\Omega\cdot\text{cm}^2$  to 0.096  $\Omega\cdot\text{cm}^2$



- ❑ No thinning or structural damage observed in the EG catalyst layer
- ❑ NPA catalyst layer thins from  $7.02 \pm 2.52 \mu\text{m}$  to  $5.74 \pm 1.39 \mu\text{m}$ , i.e. 18% thinner
- ❑ Ionomer and platinum aggregation seen in both MEAs



# Microstructural Analysis



	<b>BOL Primary porosity</b>	<b>EOL Primary porosity</b>	<b>BOL Secondary porosity</b>	<b>EOL Secondary porosity</b>	<b>BOL PtCo Particle size (nm)</b>	<b>EOL PtCo Particle size (nm)</b>	<b>BOL Cobalt</b>	<b>EOL Cobalt</b>
PtCo/NPA	39%	39%	46%	44%	6.25±1.86	9.04±3.93	0.0326	0.0151
PtCo/EG, without spray	55%	29%	24%	62%	3.42±1.10	5.58±1.83	0.0187	0.0129
PtCo/EG, with spray	54%	24%	30%	67%	3.30±1.09	5.50±1.67	0.0252	0.0110





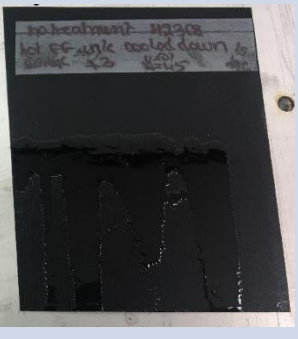
- ❑ EG cathode's lower secondary porosity may lead to higher mass transport losses
- ❑ Higher loss in primary porosity for EG cathode after durability test
- ❑ Significant porosity changes in EG cathodes vs no change in NPA cathode
- ❑ Similar particle size growth in all cathodes after durability test
- ❑ Lowest Cobalt loss of 30% in EG MEA without Nafion Overspray vs 53% for NPA MEA and 56% for EG MEA with Nafion Overspray

# Accomplishment: Coating GDL with EG Ink

- ❑ Shifting from CCM to GDE for commercialization
- ❑ Tested on various GDL: 29 BC SGL, 22BB SGL and Freudenberg H23C8
- ❑ Variability in GDL hydrophobicity
- ❑ Variability in ionomer batch
- ❑ Ways to improve GDL coating
  - Mix NPA:EG to improve coating – poor performance
  - Air plasma GDL surface modification
  - Heat ink to 60 °C
  - Treat GDL with NPA

# Coating Approaches on GDL H23C8



	NO treatment	GDL Plasma treatment	GDL NPA treatment	Heated ink (60 °C)	Cooled ink from 60 °C to 25 °C (ambient)
Comment	H23C8 is not treated and EG ink is coated	Plasma treatment is applied prior to coating	NPA deposit on the GDL, drying for a few minutes at 60°C prior to coating	Ink is heated at 60°C during a few minutes and then coated, <b>H23C8 is not treated</b>	Ink is heated to 60°C and then cooled down to 25°C prior to coating, <b>H23C8 is not treated</b>
Result	✗ Failed (no coating possible)	✓ worked	✓ worked	✓ worked	✓ worked
Pictures					

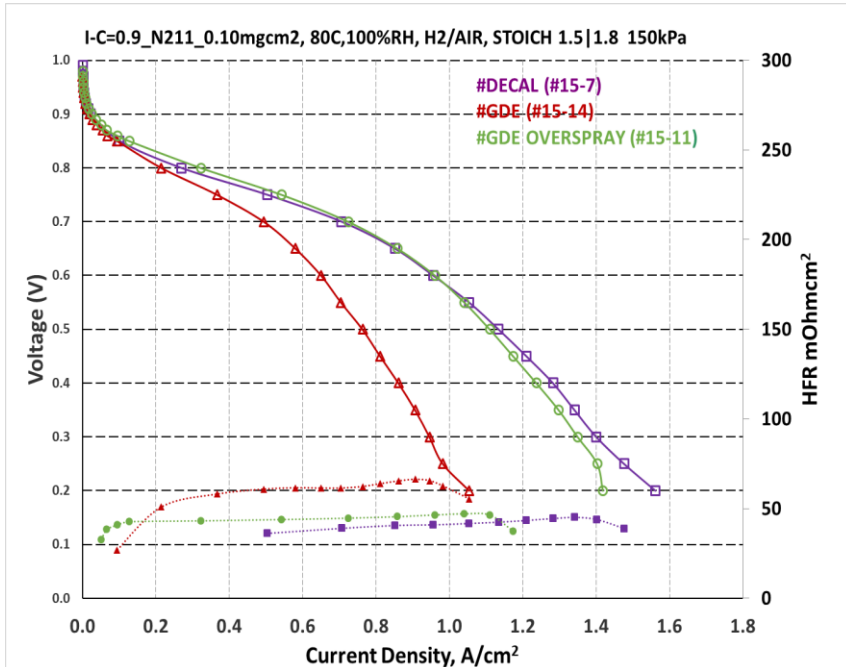


# CCM vs GDE Performance

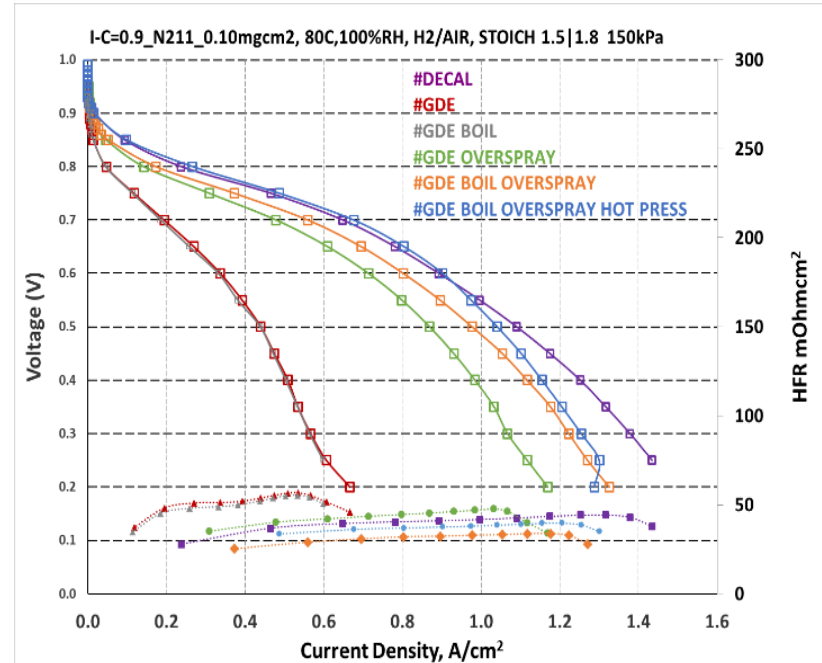


0.1 mg/cm<sup>2</sup> PtCo, 29 BC SGL, 25 cm<sup>2</sup>

## NPA



## EG NPA based Nafion Overspray



GDE proof of concept necessary for R2R production:

- ❑ NPA GDE performs similar to decal with just Nafion overspray
- ❑ EG GDE requires multi step process to match decal performance
  - Boiling + Nafion overspray + Hot pressing

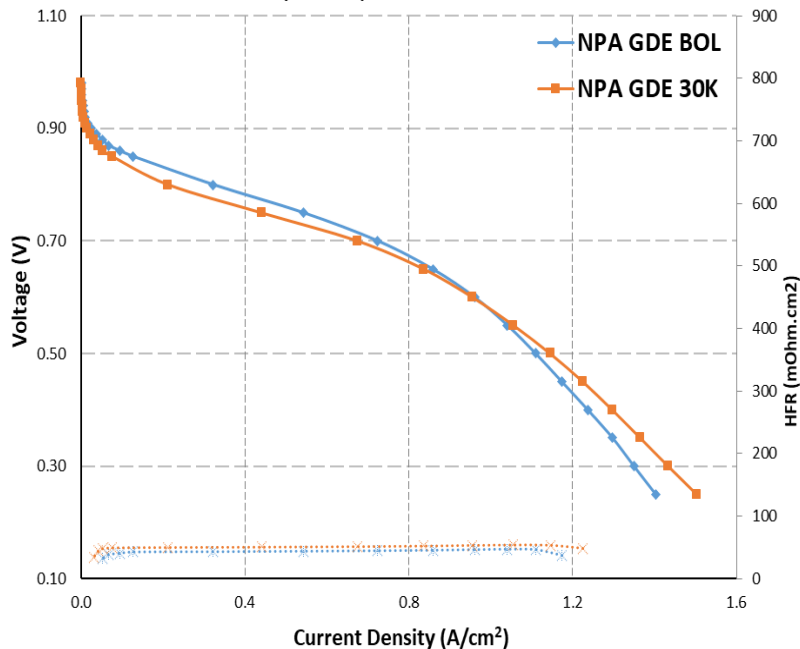
# R2R: GDE Durability



0.1 mg/cm<sup>2</sup> PtCo, 29 BC SGL, 25 cm<sup>2</sup>

## NPA

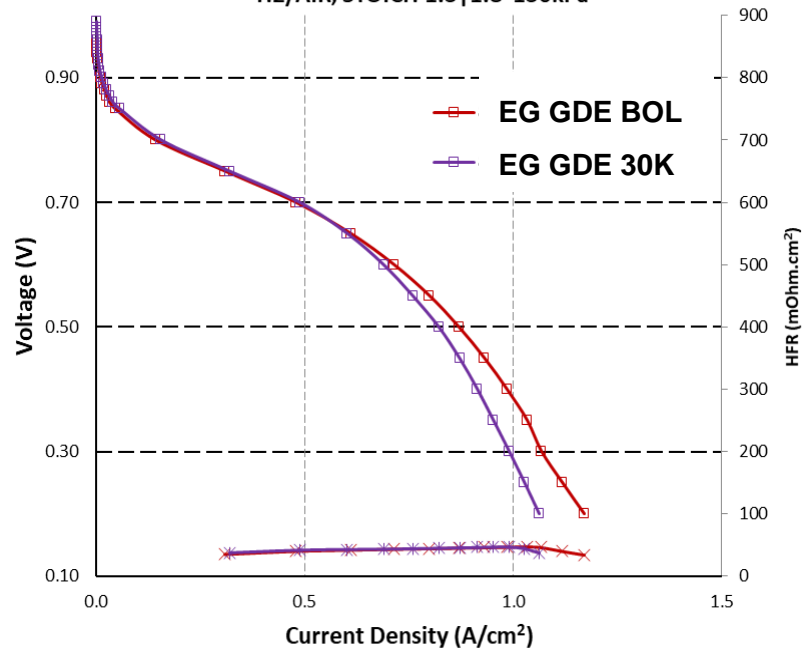
I-C=0.9\_N211\_0.1 mg/cm<sup>2</sup>, 80C, 100%RH,  
H<sub>2</sub>/Air, 1.5|1.8 Stoich, 150kPa, 29 BC SGL



## EG

### NPA based Nafion Overspray

I-C=0.9\_N211\_0.10mgcm2, 80C,100%RH,  
H<sub>2</sub>/AIR, STOICH 1.5|1.8 150kPa



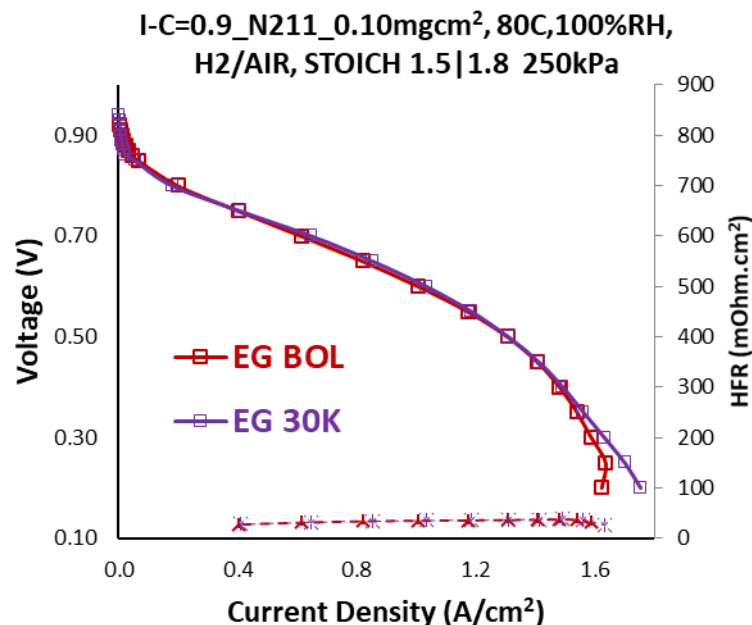
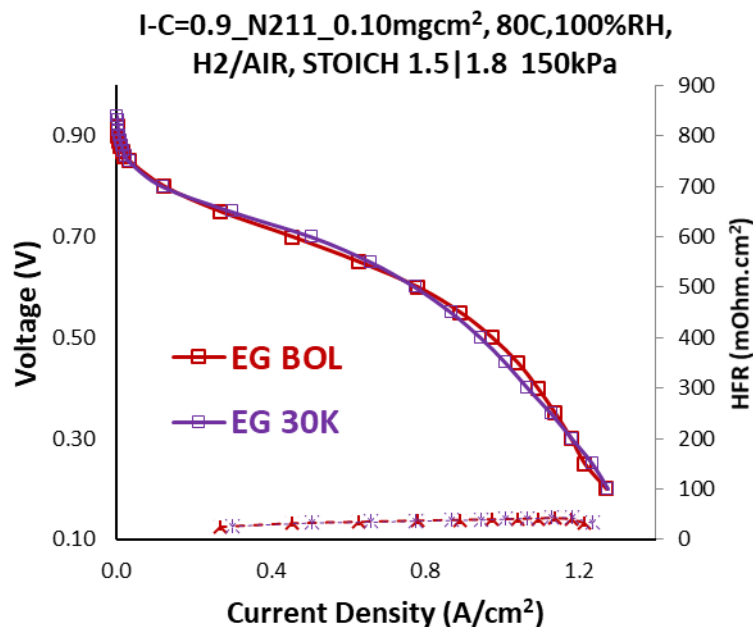
- ❑ GDE durability is different from decals: lower overall performance losses
- ❑ NPA MEA has higher losses at 0.8 V
- ❑ NPA GDE has higher sheet resistance than EG GDE, 0.26 Ω.cm<sup>2</sup> vs 0.21 Ω.cm<sup>2</sup>

# R2R: GDE Durability



0.1 mg/cm<sup>2</sup> PtCo, H23C8, 25 cm<sup>2</sup>

## EG Methanol based Nafion Overspray



## Freudenberg H23C8 with plasma treatment

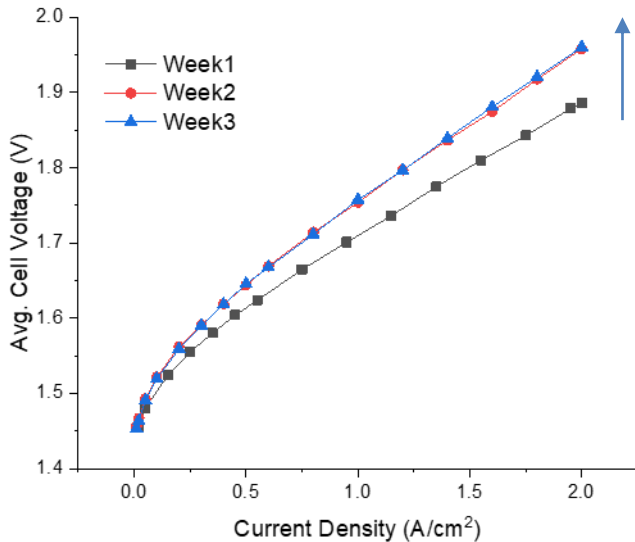
- GDE durability is better than with NPA based Nafion overspray
- Almost no performance degradation at all
- Hot pressed half CCM and then mild hot pressing of GDE to half CCM

# Accomplishment: Electrolyzer Performance

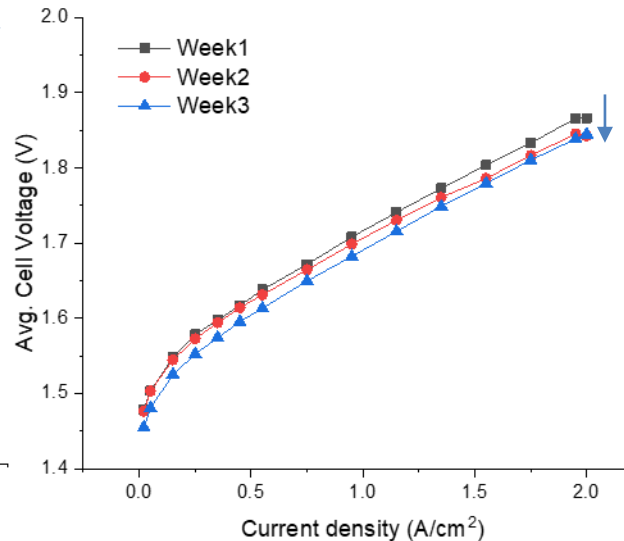
GINER

1 mg/cm<sup>2</sup> Ir Anode, 0.2 mg/cm<sup>2</sup> Pt/XC72 Cathode, 50cm<sup>2</sup>

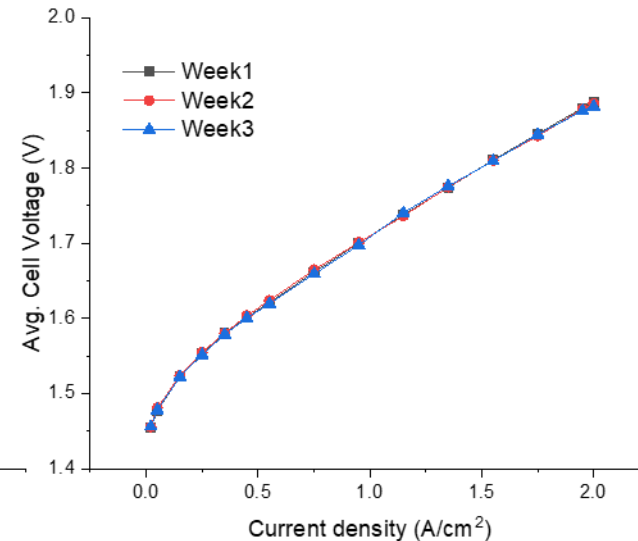
EG



H2O Rich



NPA Rich



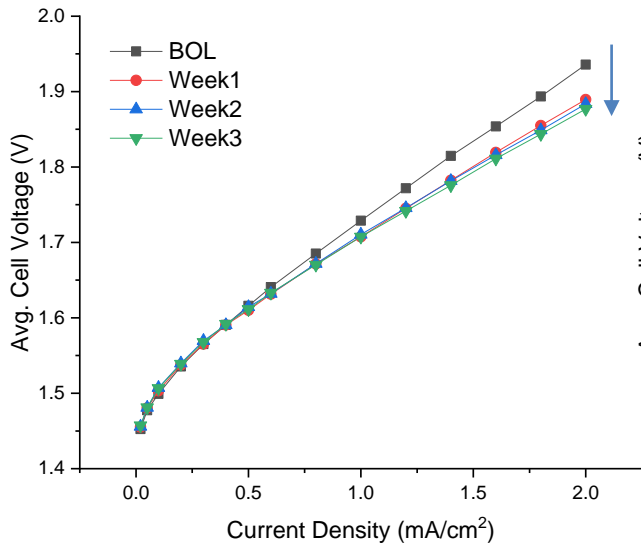
- ❑ 3 weeks at 3 A/cm<sup>2</sup> caused performance changes similar to conditioning
- ❑ No change was seen for the **NPA rich anode**
- ❑ Water rich anode improved; EG anode performance decreased

# Electrolyzer Performance

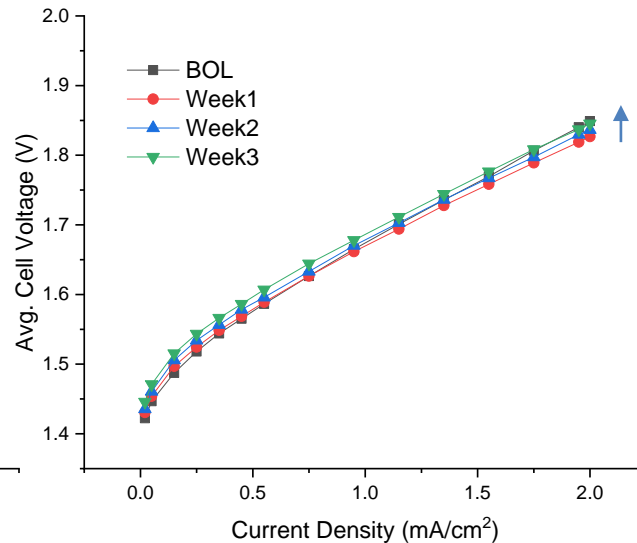


1 mg/cm<sup>2</sup> IrOx Anode, 0.2 mg/cm<sup>2</sup> Pt/XC72 Cathode, 50cm<sup>2</sup>

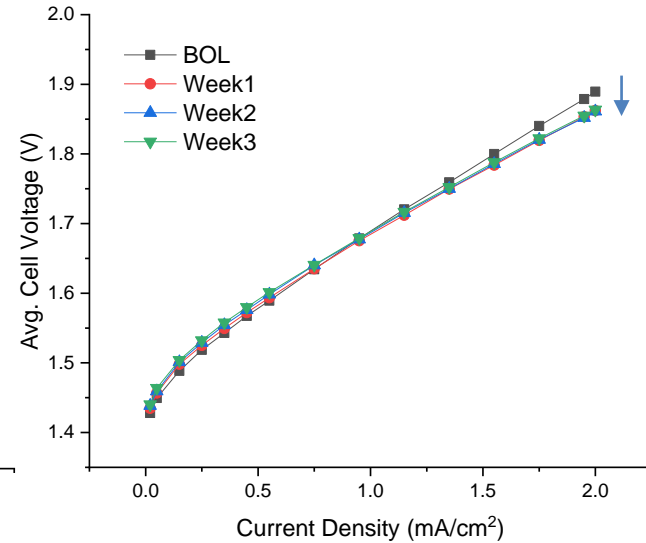
## EG



## H2O Rich



## NPA Rich

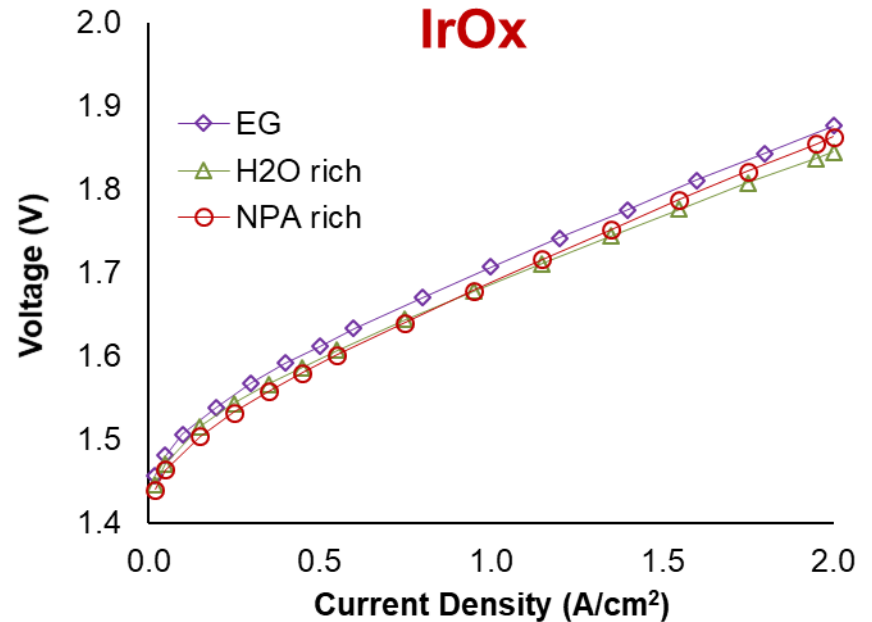
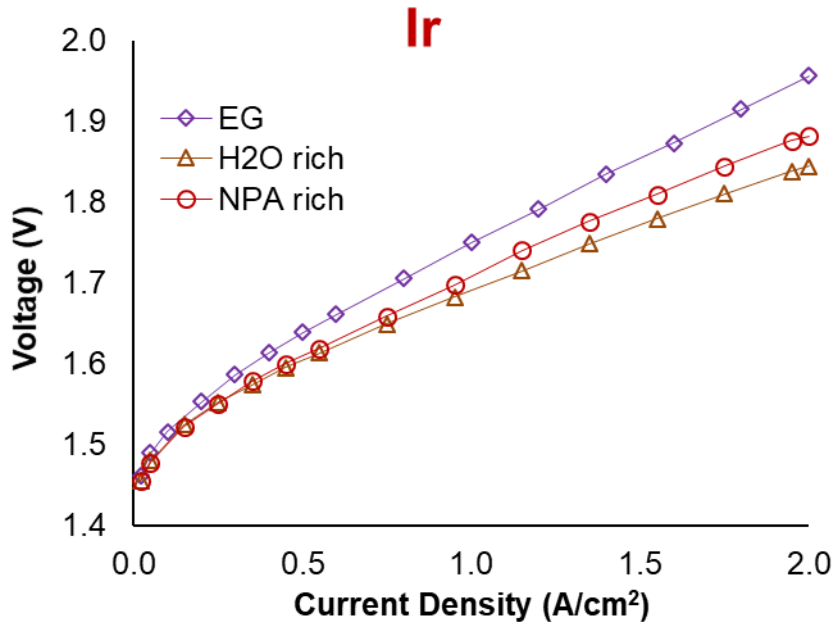


- ❑ 3 weeks at 3 A/cm<sup>2</sup> caused performance changes similar to conditioning
- ❑ NPA rich anode shows improvement
- ❑ Water rich anode improved but then decayed
- ❑ EG anode performance improves

# Ir vs. IrOx Comparison



1 mg/cm<sup>2</sup> IrOx Anode, 0.2 mg/cm<sup>2</sup> Pt/XC72 Cathode, 50cm<sup>2</sup>



- ❑ Initial performance of IrOx samples are better than Ir black
- ❑ Performance: H<sub>2</sub>O rich > NPA rich > EG
- ❑ For durability: move to lower loading of 0.1 mg/cm<sup>2</sup> Ir

# R2R Plan with NREL



- ❑ Address EG's high boiling point
- ❑ Identify R2R parameters for EG ink
- ❑ NREL's experience with coating battery anodes made with NMP will help guide EG ink R2R fabrication process
  - Similar boiling point to EG
  - Use of 2 ovens: Prebake + Bake

Ink Formulation (**Giner**): send premixed ink and substrate material to NREL

Small scale trials (**NREL**)

– rod coating, SEM

Roll-to-Roll Coating (**NREL**)

- Corona treatment of GDL
- Tuning of coating speed and flow rate
- Variations in ink formulation & drying process
- Inspection of coating quality – optical or SEM

Send **1 m<sup>2</sup>** coated GDE from NREL to Giner for in-situ testing

# Collaboration and Coordination

- **Giner (Lead):** Hui Xu, Natalia Macauley, Magali Spinetta, and Shirley Zhong. Oversee the entire project management, catalyst ink rheology, electrode design and scale-up, and MEA commercialization
- **LANL (Subcontractor):** Yu Seung Kim. Provide ionomer dispersions in EG and Methanol, and SANS measurements of Nafion
- **NREL (Subcontractor):** Scott Mauger. Optimize R2R conditions for GDE fabrication
- **UConn (Collaborator):** Jasna Jankovic. Perform microstructure analysis with SEM and TEM: ionomer, Pt, Co distributions and particle size changes from BOL to EOL



# Responses to Previous Year Reviewers' Comments



**This project was not reviewed in 2019**

# Conclusions

- ❑ Rheology, zeta potential and laser diffraction was used to characterize catalyst inks for ink quality
- ❑ Average performance of EG MEA without Nafion overspray is similar to NPA baseline MEA performance
- ❑ EG MEA with Nafion overspray outperforms NPA Baseline MEA
- ❑ Durability of EG MEA without Nafion overspray is better than that of NPA
- ❑ Better MEA durability of EG GDE compared to NPA GDE seen at low voltages
- ❑ Methanol based Nafion overspray results in highly stable EG MEA
- ❑ Catalyst microstructure agrees with observed performance in PtCo catalysts with NPA vs EG solvents
- ❑ NPA based electrolyzer anodes perform better than EG based anode
  - Non uniform microstructure of EG based Ir anode in good agreement with observed performance

# Future Work

## Fuel Cell

- ❑ Establish performance and durability of EG vs NPA GDEs on H23C8
- ❑ Finalize TSA with NREL and send EG ink for initial coating tests
- ❑ Local oxygen resistance of EG vs NPA GDEs
- ❑ Evaluate microstructure difference between GDEs in NPA vs EG solvent

## Electrolyzer

- ❑ Perform electrolyzer durability test on 0.1 mg/cm<sup>2</sup> Ir NPA rich and EG MEA to evaluate EG's durability advantage
  - 1.45 – 2.0 V square wave with 30s dwell time
- ❑ Microstructure analysis on degradation process of low loaded anodes
- ❑ Follow iridium dissolution during AST using ion chromatography

# Acknowledgements



- ❑ Financial support from DOE SBIR/STTR Program under award # DE-SC0012049
- ❑ Program Manager  
- Ms. Donna Ho
- ❑ Dr. John Kopasz for project suggestions
- ❑ Dr. Yu- Seung Kim at LANL (Subcontractor)
- ❑ Dr. Jasna Jankovic at UConn