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# Metal-Organic Frameworks Containing Frustrated Lewis Pairs for Hydrogen Storage at Ambient Temperature

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Project ID: **ST210**

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# Overview

## Timeline

- Project start: Dec. 2019
- Phase I end: Dec. 2020
- Phase II end: Sept. 2022

## Budget

- Total project requested: \$850K
  - DOE share: \$680K
  - Contractor share: \$170 K
- Funding received in FY2020 (Phase I)
  - \$ 300 K

## Barriers

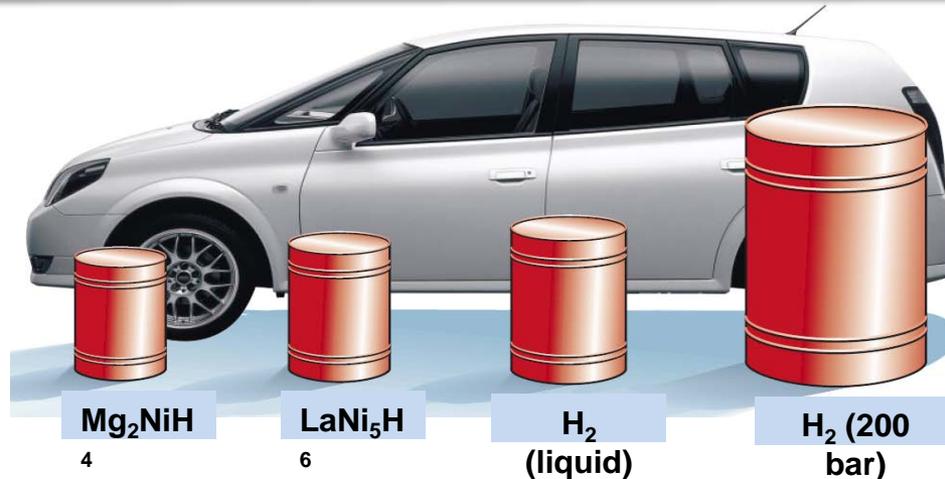
- Barriers to be addressed
  - A. Hydrogen binding energies
  - B. Hydrogen physisorption capacity at ambient temperature
  - C. New H<sub>2</sub> adsorption mechanism

## Partners

- Interactions/collaborations
  - University of South Florida (Lead)
  - Argonne National Laboratory (Subcontractor)
  - HyMARC



# Current Technology Options for on-board H<sub>2</sub> Storage



- Compressed hydrogen gas (**high pressure >700 bar**)
- Cryogenic storage of liquid hydrogen (**energy consuming**)
- Chemisorption using metal hydrides and chemical hydrides (**irreversible and poor kinetics of hydrogen recharging**)
- Physisorption using porous materials (**fast charge-recharge process but very low uptake capacity at room temperature**)

*To reach high storage capacity for porous materials at ambient temperature, H<sub>2</sub> binding energy needs to be in the range of 15 to 25 kJ/mol.*



# Objective – Relevance

Table 1. Technical System Targets: Onboard Hydrogen Storage for Light-Duty Fuel Cell Vehicles

Storage Parameter	Units	2025	Ultimate
<b>System Gravimetric Capacity:</b>	kWh/kg (kg H <sub>2</sub> /kg system)	1.8 (0.055)	2.2 (0.065)
<b>System Volumetric Capacity:</b>	kWh/L (kg H <sub>2</sub> /L system)	1.3 (0.040)	1.7 (0.050)

*Energy Environ. Sci.*, **2018**, 11, 2784—2812.



# Objective - Relevance

- **Phase I** – to demonstrate and deliver one FLP@MOF with reversible total gravimetric capacity  $\geq 1.5$  wt % and total volumetric capacity  $\geq 0.012$  kg H<sub>2</sub>/L at H<sub>2</sub> pressure of  $\leq 100$  bar at room temperature.
- **Overall** – to produce one or more FLP@MOF that that meets or exceeds the DOE's 2025 goal of H<sub>2</sub> storage gravimetric density (GD) of 5.5 wt.% and volumetric density (VD) of 0.040 kg H<sub>2</sub>/L.

## Merits of FLP@MOFs & Their Impact on Technology Barriers

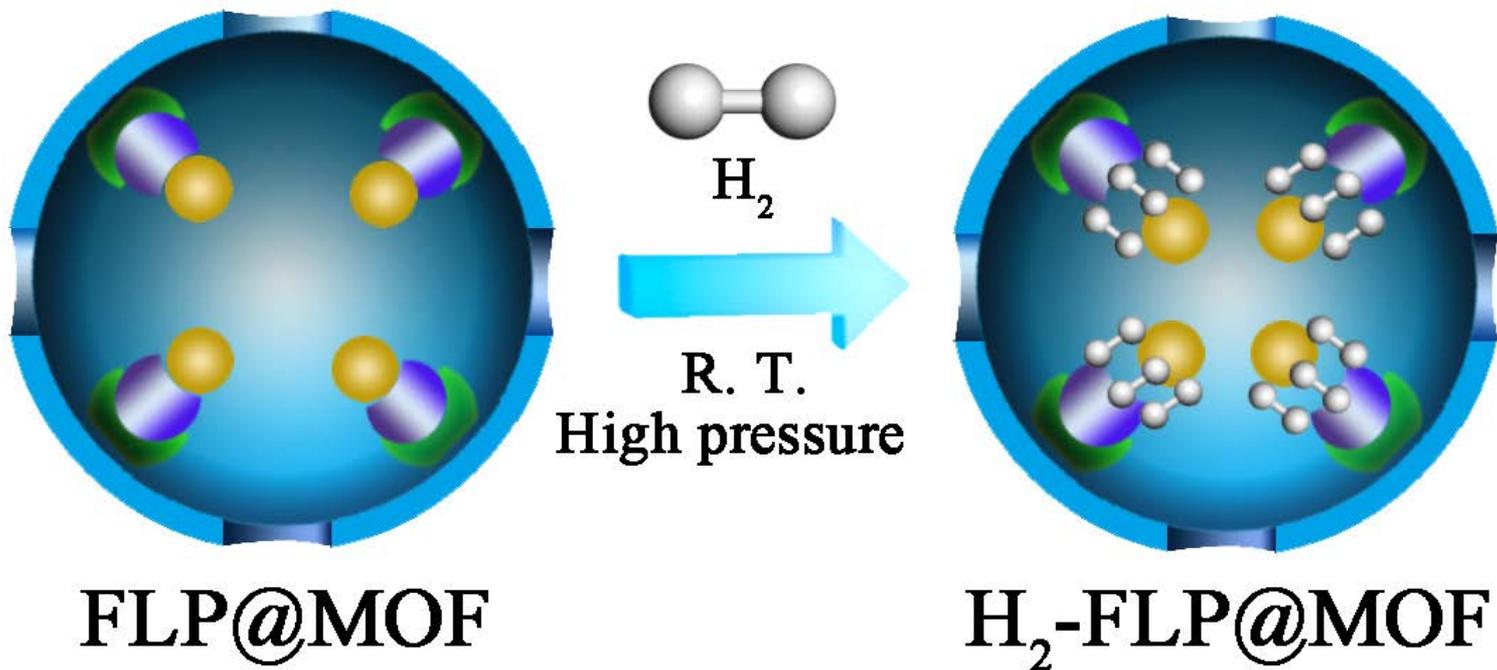
**Merits**

1. High H<sub>2</sub> adsorption working capacity
2. New H<sub>2</sub> adsorption mechanism
3. Strong H<sub>2</sub> binding energy
4. High H<sub>2</sub> uptake capacity at ambient temperature

The diagram illustrates the structure of FLP@MOF. It shows a central complex of a Ferrocene Ligand (FLP) and a Metal-Organic Framework (MOF). The FLP is represented by a purple and yellow structure, and the MOF is a blue and white porous structure. The text 'FLP@MOF' is written above the structure.



# Approach: Encapsulating FLP into MOF (FLP@MOF) for Hydrogen Storage at Ambient Temperature





# Approach – Development Strategy

## FLP@MOF Design and Synthesis (USF)

## Characterization & Modeling (USF/ANL/HyMARC)

## Optimization & Engineering (USF/ANL)

- Develop various approaches to synthesize new FLP@MOF
- Structure characterization of FLP@MOF
- High throughput synthesis of FLP@MOF

- H<sub>2</sub> storage capacity & reversibility measurements
- Advanced characterization of H<sub>2</sub>-FLP@MOF
- Computational modelling of H<sub>2</sub>-FLP@MOF

- Volumetric capacity enhancement of H<sub>2</sub> adsorption in FLP@MOF

- New FLP@MOFs with high H<sub>2</sub> storage capacities
- Enhancing H<sub>2</sub> hydrogen binding energy to 15-25 kJ/mol
- Unveiling possibly new H<sub>2</sub> adsorption mechanism
- Improving volumetric H<sub>2</sub> adsorption capacity by preparing monolithic FLP@MOF

**Collaborating with HyMARC/others and leveraging existing experimental / theoretic supports are essential to the project success!**



# Approach: Phase I Milestone Status

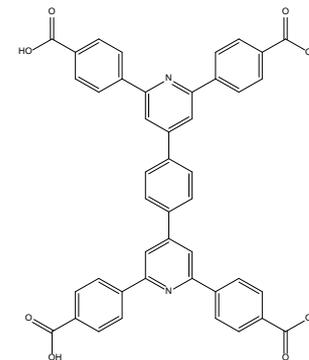
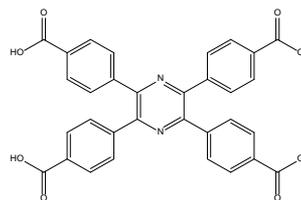
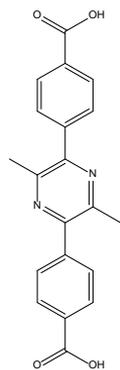
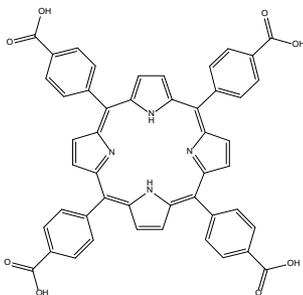
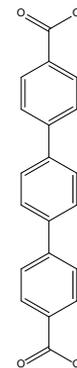
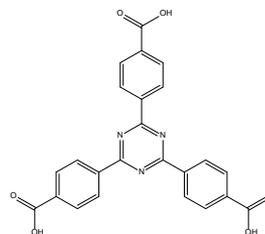
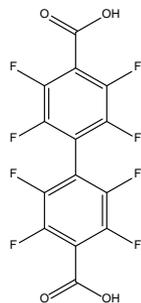
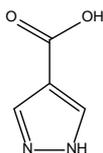
Milestone #	Milestones/Go-NoGo DP	Milestone Verification	Percent Complete
M 1.1.1	Complete synthesise at least six meso-MOFs for the incorporation of FLP	MOF synthesis will be carried out at USF based on literature reported method	100%
M 1.1.2	Design and synthesize a series of FLP@MOF using step-wise anchoring approach	FLP@MOF synthesis will be carried out at USF	20%
M 2.1.1	Complete high-pressure hydrogen storage measurements for stepwise synthesized FLP@MOFs	Isotherm of H <sub>2</sub> at different pressures will be measured by USF and ANL teams	0%
M 2.2.1	Complete structural studies of the first batch of FLP@MOF	Conventional analytic tools will be applied to study selected FLP@MOF at USF/ANL	10%
GNG 1.2.3	At least one FLP@MOF with reversible total gravimetric capacity $\geq 1.5$ wt % and total volumetric capacity $\geq 0.012$ kg H <sub>2</sub> /L at H <sub>2</sub> pressure of $\leq 100$ bar at room temperature	To be delivered to DOE designated lab for certification after initial measurement at USF/ANL	0%

**The focus of Phase I is to deliver one FLP@MOF with reversible total gravimetric capacity  $\geq 1.5$  wt %.**



# Accomplishments: Ligand Synthesis

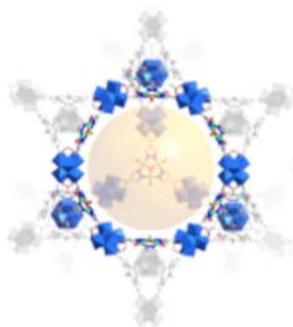
- The following ligands have been synthesized for the construction of MOFs.



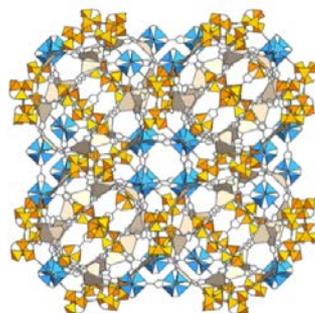


# Accomplishments: Preparation of meso-MOFs.

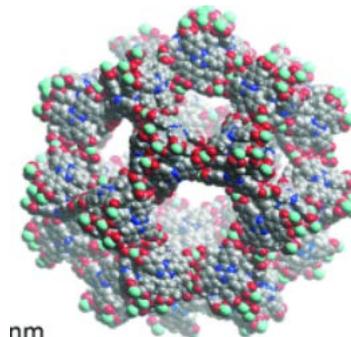
- Meso-MOFs: MOF818, FDM-3, Tb-TATB, Zr-UIO-68, Zr-UIO-67-8F, PCN-333(Fe), PCN-333(Cr), UIO-68, MIL-101(Cr), MIL-101-4F and MIL-101-Br(Cr) with the structures shown below have been prepared as planned.



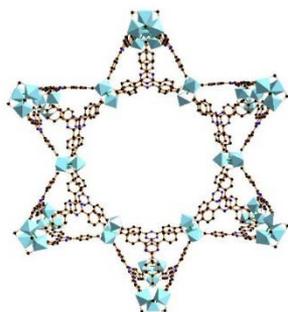
MOF-818



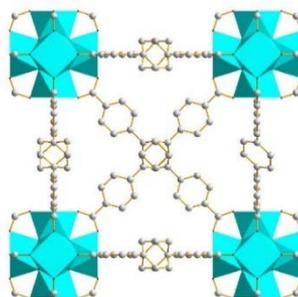
FDM-3



Tb-TATB



PCN-333



UIO-68

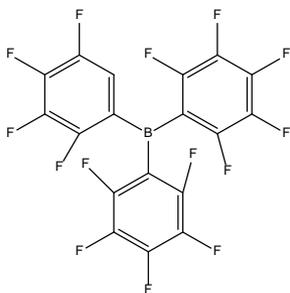


MIL-101

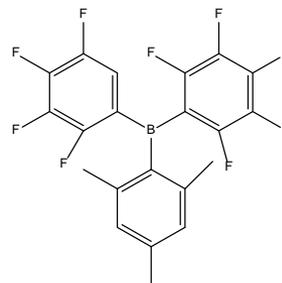


# Accomplishments: Synthesis of Lewis acids and Lewis bases for FLP

- The following two Lewis acids have been prepared.

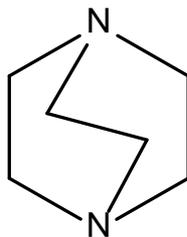


B(C<sub>6</sub>F<sub>5</sub>)<sub>3</sub>

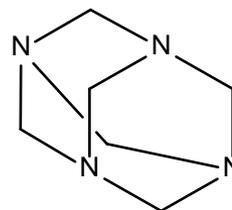


MesB(C<sub>6</sub>F<sub>5</sub>)<sub>2</sub>

- The following two Lewis bases have been prepared.



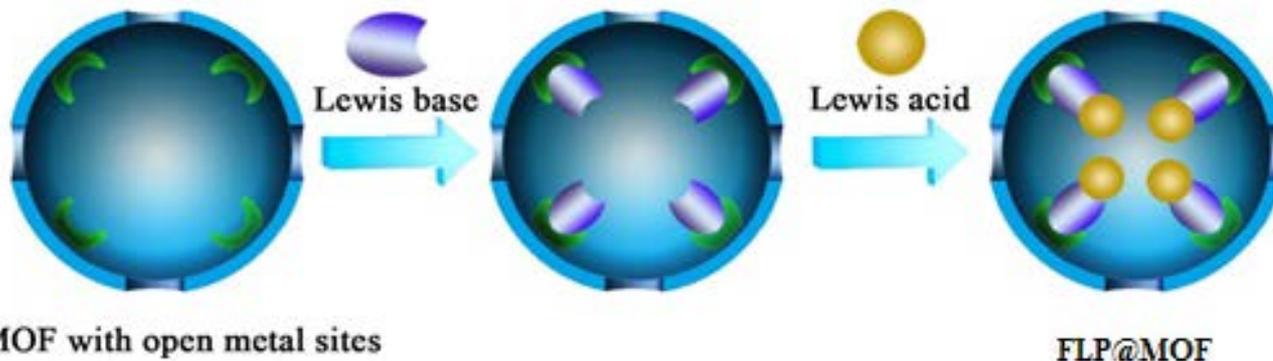
DABCO



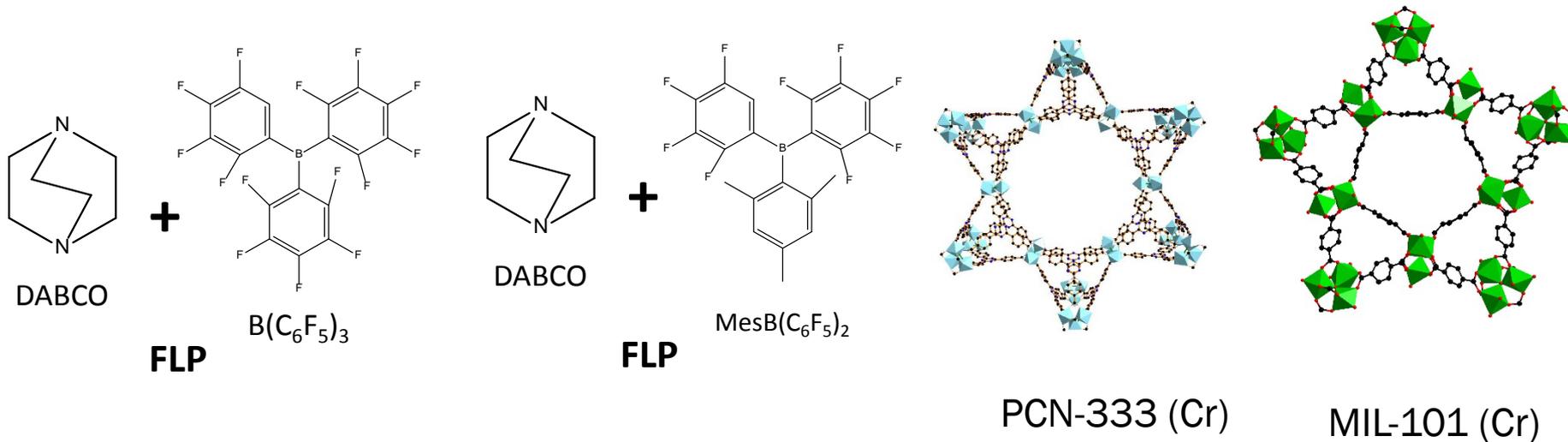
HMTA



# Accomplishments: Step-wise anchoring of FLP on MOF through coordination interaction



- The following two sets of FLPs have been encapsulated in PCN-333(Cr) and MIL-101(Cr) using step-wise anchoring approach.





# Collaborations: Working with HyMARC

- HyMARC – SNL
  - High pressure analysis
  - Surface characterization tools
- HyMARC – PNNL
  - In situ NMR study
- HyMARC – LLNL
  - Modeling/simulation of H<sub>2</sub>-FLP@MOF Surface XPS
- HyMARC - NREL
  - Measurements of Q<sub>st</sub> for H<sub>2</sub> adsorption in FLP@MOF
  - Capacity certification

**Looking forward to formulating detailed experimental plan through the discussion with HyMARC members!**



# Challenges and Barriers

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- The high-pressure sorption instrument at USF will need to be well calibrated to collect reliable high-pressure H<sub>2</sub> sorption isotherms particularly at pressure close to 100 bar.
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# Proposed Future Work

## ■ Remaining Phase I Activities

- Synthesis of a series of mesoporous MOFs
- Incorporation of FLPs into the synthesized mesoporous MOFs via step-wise anchoring approach
- Optimize FLP vs MOF formulation through high throughput synthesis
- Structural characterizations of the prepared FLP@MOFs with conventional and advanced tools
- High pressure H<sub>2</sub> storage capacity measurements

## ■ Planned Phase II Activities

- Developing other approaches to incorporate FLPs into MOFs
- Continue high throughput synthesis/screening of FLP@MOFs and optimization of FLP@MOFs
- High pressure hydrogen storage measurements
- Advanced characterizations and mechanistic studies of H<sub>2</sub> adsorption in FLP@MOFs
- Computational modeling/simulation support
- Process engineering of FLP@MOF

Any proposed future work is subject to change based on funding levels.



# Technology Transfer Activities

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- **A Provisional Patent Application entitled “FRUSTRATED LEWIS PAIR-IMPREGNATED POROUS MATERIALS AND USES THEREOF” has been filed.**



# Summary

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- Eight ligands have been synthesized for the construction of MOFs.
  - Six meso-MOFs: MOF818, FDM-3, Tb-TATB, Zr-UIO-68, Zr-UIO-67-8F, PCN-333(Fe), PCN-333(Cr), UIO-68, MIL-101(Cr), MIL-101-4F and MIL-101-Br(Cr) with the structures shown below have been prepared.
  - Two Lewis acids and two Lewis bases have been prepared for FLPs.
  - Two sets of FLPs have been encapsulated in PCN-333(Cr) and MIL-101(Cr) using step-wise anchoring approach.
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# Acknowledgements

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- This work is supported by US DOE, Office of Energy Efficiency and Renewable Energy, Fuel Cell Technologies Office
- Technology Manager: Zeric Hulvey
- Project Manager: Jesse Adams
- Hydrogen Fuel R&D Program Manager: Ned Stetson